

Synergistic Heterojunction Effects in Ag₃PO₄/SnO₂ Nanocomposites: A Photocatalytic Study on Isoproturon Degradation

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S.I. 1. Synthesis of SnO₂:

The SnO₂ nanoparticles were synthesized using hydrothermal method. 0.11 mmol of SnCl₂•2H₂O was prepared in 40 mL of MeOH. Then about 20 mL of H₂O₂ was added slowly in the above solution. Then solution was stirred for 30 minutes. Then the solution was and heated in Teflon-lined stainless-steel autoclave at 150 °C for 15 hours. The synthesised nanoparticles were centrifuged, washed with excess methanol and dried in oven at 80 °C for 6 hours.

S.I. 2. Synthesis of Ag₃PO₄:

The Ag₃PO₄ nanoparticles were synthesized using hydrothermal method. 2.94 mmol of AgNO₃ were dissolved in 30 mL of distilled water. In another beaker 1 mmol of Na₂HPO₄ was dissolved in distilled water. After both the solutions were clear, Na₂HPO₄ solution was added slowly in AgNO₃ solution. Formation of yellow coloured precipitate was observed. Then the solution was put in Teflon-lined stainless-steel autoclave and heated at 150 °C for 15 hours. The yellow-coloured nanoparticles were centrifuged, washed with distilled water and dried in oven at 80 °C for 6 hours.

S.I. 3. Characterization details

For SEM images, Gemini SEM 500 scanning electron microscope (SEM) was used. A Jeol IT200 was used to record the element mapping. Thermo Scientific Nicolet iS50 FTIR tri-detector was used to observe FTIR. Rigaku MiniFlex instrument was used to generate X-ray diffraction (XRD) patterns having Cu filament of $K_{\alpha}=1.54\text{\AA}$ and scanning rate of 3 degree per minute at room temperature. BET adsorption-desorption isotherms were performed using Microtrac BELSORP Mix II at 77 K. UV-vis spectra were recorded by Michelson Interferometer FTIR Spectrophotometer. Shimadzu UV-1800 spectrophotometer which utilize a precision Czerny-Turner optical system was used for taking UV-visible diffuse reflectance spectrum. X-ray photoelectron spectroscopy (XPS) was performed by a Shimadzu AXIS Supra⁺ having Al K _{α} X-ray.

S.I. 4. Equations used for kinetic studies

$$-\frac{dC_P}{dt} = \frac{k_{deg}KC_P}{1 + KC_P} \quad (1)$$

$$\frac{-t}{C_P - C_{P,0}} = \frac{1}{k_{deg} * K} \times \frac{\ln(\frac{C_P}{C_{P,0}})}{C_P - C_{P,0}} + \frac{1}{k_{deg}} \quad (2)$$

$$\ln \frac{C_P}{C_{P,0}} = -k_1 t \quad (3)$$

Tables:

Table S1: FTIR bands of Ag₃PO₄/SnO₂ nanocomposite

Peak (cm ⁻¹)	About	Reference
1119	Antisymmetric streching of P-O bonds	(1)
949	PO ₄ ³⁻ stretching	(2)
663	P-O-P Streching	(3)
627	O-Sn-O Stretching	(4)
544	Asymmetric Bending of O=P-O bond	(2)
500	Sn-O Streching	(4)

Table S2: A comparison of synthesized nanocomposite with nano-catalysts reported in literature which have been used for photocatalytic IPU degradation

Catalyst	Synthesis method	Concentration of IPU	Amount of Catalyst	UV lamp power	Degradation efficiency	Time	Ref.
TiO ₂ functionalized silica nanofibrous membranes	Electrospinning and dip-coating method	0.5-1 mg/L	-----	300 Watt	100%	8 hours	(5)
SnS ₂ /RGO nanocomposite	Hydrothermal method	10mL solution of concentration 1ppm	2mg	65 Watt	Rate constant 0.0219 min ⁻¹	-----	(6)
GO-TiO ₂ catalyst	Hydrothermal method	5mg/L	200mg/L	15 Watt	About 100%	300 min	(7)
PAN/Ag-AgBr@Bi ₂₀ TiO ₃₂	Coaxial electrospinning method	15mg/L(50mL solution)	0.1g	500 Watt	87.9%	60 min	(8)
TiO ₂	Dip coating method	10mg/L	-----	Two 60 Watt lamps and four 15 watt lamps	57.07%	120 min	(9)

TiO ₂ /HY composite	Solid state dispersion method	1.14×10^{-4} M(50mL)	50mg	intensity $\sim 75 \text{mWcm}^{-2}$	100%	120min	(10)
Fe-BTC MOF @ aramid fabric (Fe-BTC@AF) composite	layer-by-layer in situ self-assembly methods	5mg/L(50mL)	3cm×3cm	300 watt	90%	7 hours	(11)
Yb ³⁺ doped microspherical BiOI	Hydrothermal method	15mg/L(50mL)	50mg	500 watt	90.2%	-----	(12)
Bismuth modified porous silica (Bi ₂ SiO ₅)	Impregnating method	1.14×10^{-4} mol/L		250kW	100%	120 min	(13)
Ag ₃ PO ₄ /SnO ₂	Hydrothermal method	1ppm	0.1g	125 Watt	97%	120 min	Current work

Figures:

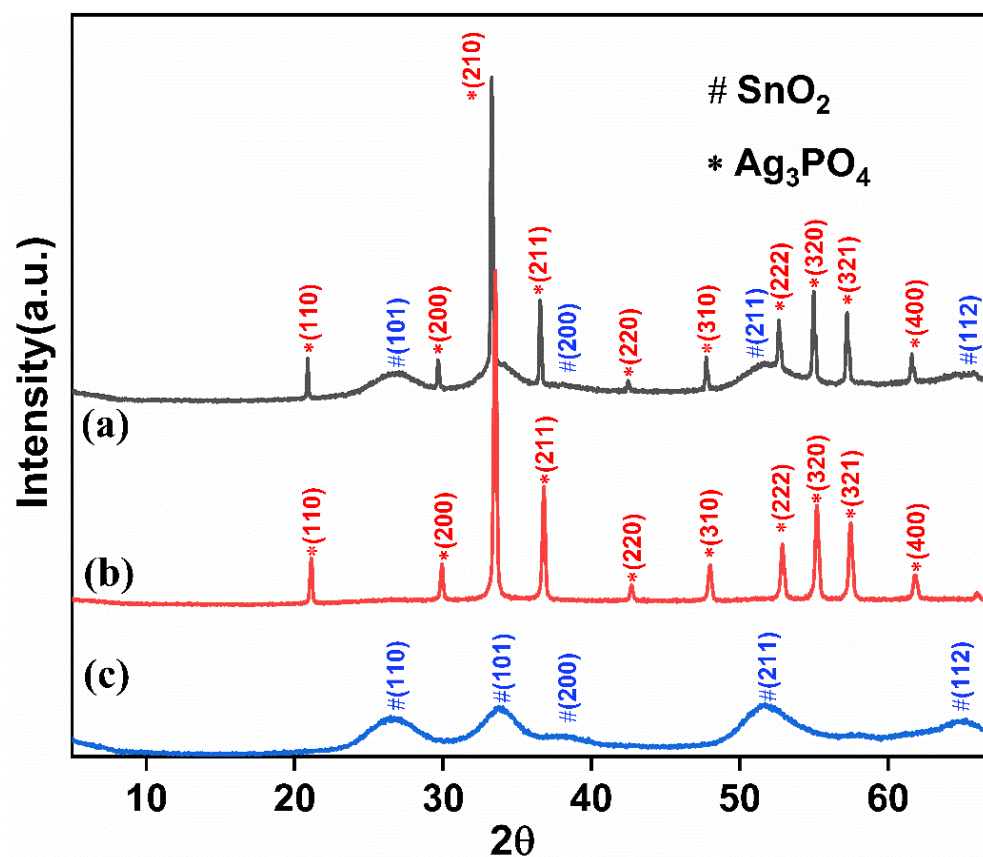


Figure S1: Powder XRD patterns comparison of $\text{Ag}_3\text{PO}_4/\text{SnO}_2$ nanocomposite, pure Ag_3PO_4 nanoparticles and pure SnO_2 nanoparticles.

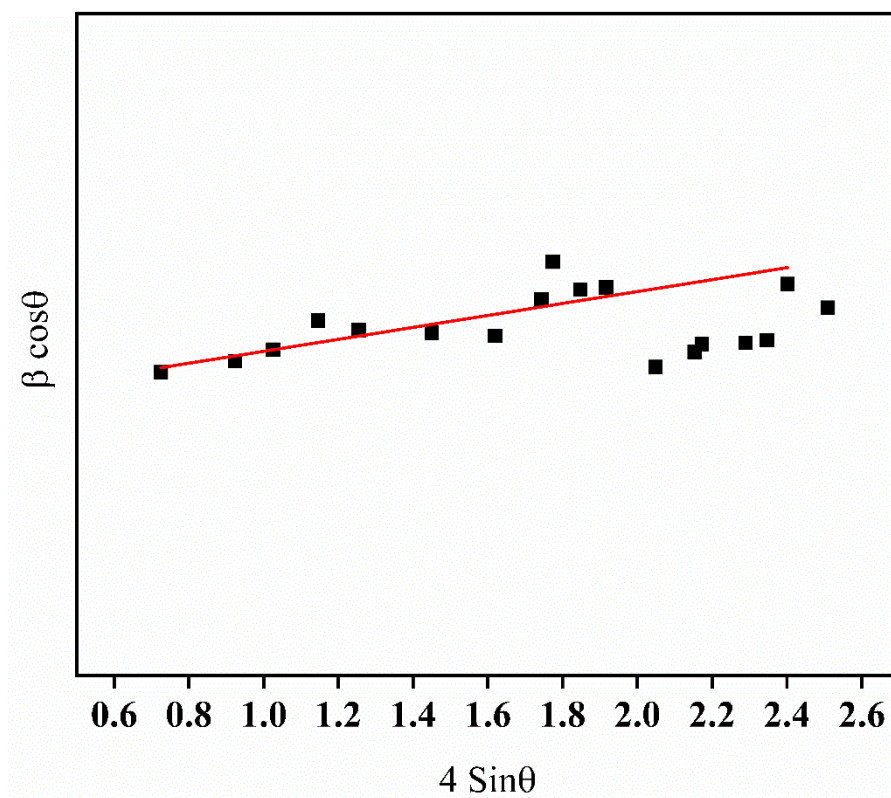


Figure S2: W-H plot of $\text{Ag}_3\text{PO}_4/\text{SnO}_2$ nanocomposite.

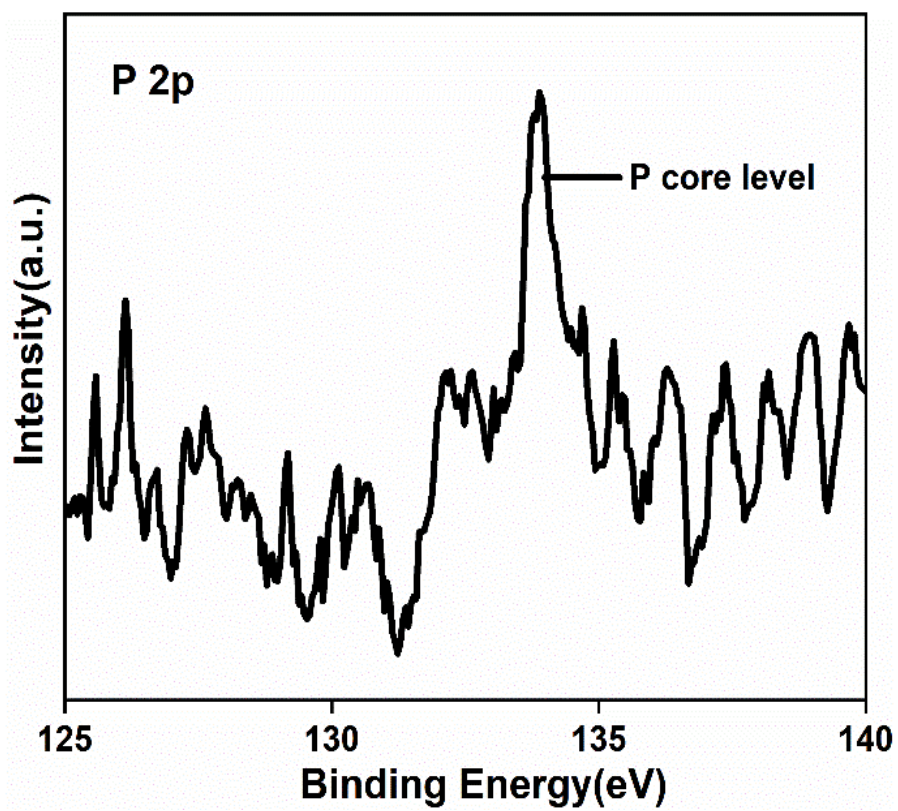


Figure S3: Core level XPS spectrum of Phosphorus.

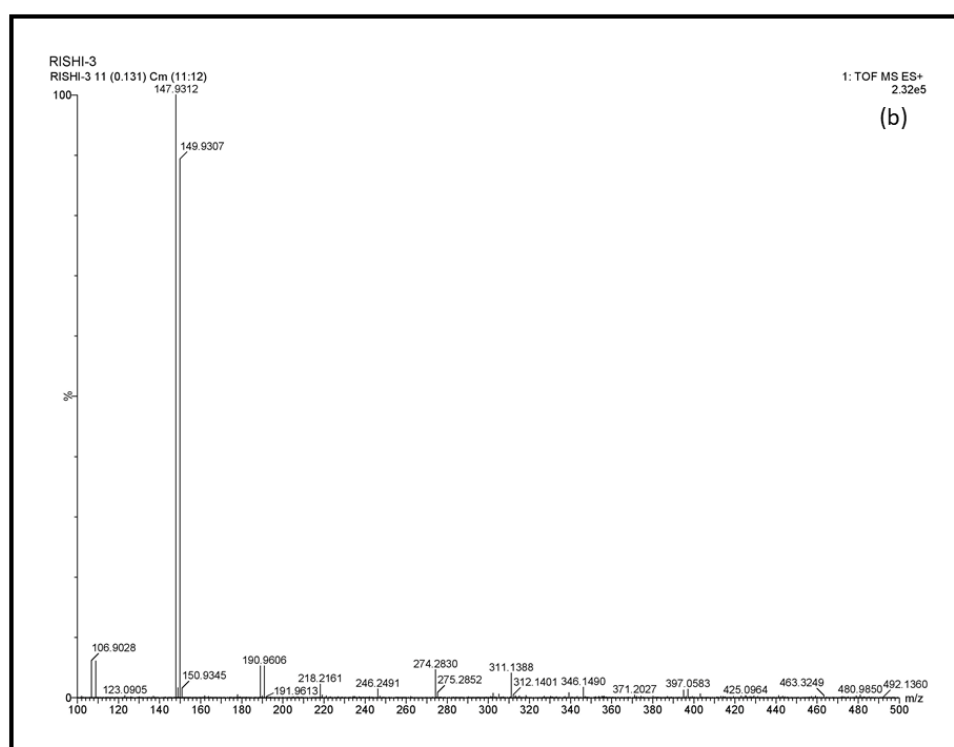
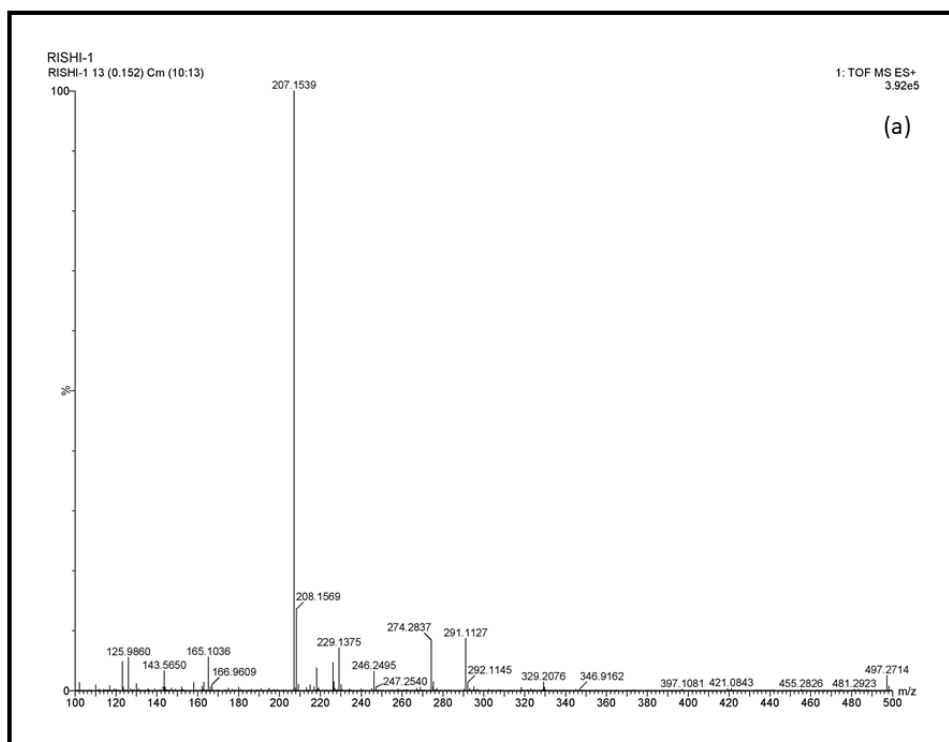


Figure S4: HRMS of isoproturon (a) 0 min and (b) 120 min of light irradiation.

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