

Supplementary Material

Autotrophic Bacterial Production of Polyhydroxyalkanoates Using Carbon Dioxide as a Sustainable Carbon Source

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1 Supplementary Data

Methods for Figure 1

The evolutionary history was inferred using the Neighbor-Joining method (Saitou and Nei, 1987). The optimal tree is shown. The evolutionary distances were computed using the Maximum Composite Likelihood method (Tamura et al., 2004) and are in the units of the number of base substitutions per site. This analysis involved 32 nucleotide sequences. All ambiguous positions were removed for each sequence pair (pairwise deletion option). There were a total of 1623 positions in the final dataset. Evolutionary analyses were conducted in MEGA11 (Stecher et al., 2020; Tamura et al., 2021)

Methods for Figure 2

The evolutionary history was inferred using the Neighbor-Joining method (Saitou and Nei, 1987). The optimal tree is shown. The evolutionary distances were computed using the Maximum Composite Likelihood method (Tamura et al., 2004) and are in the units of the number of base substitutions per site. This analysis involved 68 nucleotide sequences. All ambiguous positions were removed for each sequence pair (pairwise deletion option). There were a total of 2128 positions in the final dataset. Evolutionary analyses were conducted in MEGA11 (Stecher et al., 2020; Tamura et al., 2021)

2 Supplementary Tables

Table S1. Photo-mixotrophic production of PHA from cyanobacteria

Cyanobacterial strain	Carbon source	PHA content in DCW (%)	Limiting factor	PHA composition	Bioreactor	Reference
<i>Arthrospira platensis</i> UMACC 161	CO ₂ (0.04%) + 0.5% Acetate	10	N	P (3HB)	Erlenmeyer flask	(Toh et al., 2008)
<i>Chlorogloea fritschii</i>	CO ₂ (5%) + 0.02M acetate	10	-	PHB	Erlenmeyer flask	(Carr, 1966)
<i>Chlorogloeopsis fritschii</i> PCC 6912	CO ₂ (0.04%) + 15 mM Acetate	6.2	N	PHB	Erlenmeyer flask	(Hai et al., 2001)
<i>Cyanothece</i> sp. PCC 7424	CO ₂ (0.04%) + 15 mM Acetate	1.1	N	PHB	Erlenmeyer flask	(Hai et al., 2001)
<i>Cyanothece</i> sp. PCC 8303	CO ₂ (0.04%) + 15 mM Acetate	2.7	N	PHB	Erlenmeyer flask	(Hai et al., 2001)
<i>Desmonostoc muscorum</i> SAG 1453-12b (<i>Nostoc muscorum</i> Agardh)	CO ₂ (10%) ^a	70	-	PHB-co-PHV	CSTR	(Bhati and Mallick, 2016)
<i>Desmonostoc muscorum</i> SAG 1453-12b (<i>Nostoc muscorum</i> Agardh)	CO ₂ (10%) ^a	65	-	PHB-co-PHV	CSTR	(Bhati and Mallick, 2016)
<i>Gloeocapsa</i> sp. PCC 7428	CO ₂ (0.04%) + 15 mM Acetate	1.5	N	PHB	Erlenmeyer flask	(Hai et al., 2001)
<i>Gloeotheca</i> sp. PCC 6501	CO ₂ (0.04%) + 15 mM Acetate	1.4	N	PHB	Erlenmeyer flask	(Hai et al., 2001)
<i>Gloeotheca</i> sp. PCC 6909	CO ₂ (0.04%) + 10 mM Acetate	6-9	-	PHB	Erlenmeyer flask	(Stal, 1992)
<i>Stanieria</i> sp. PCC 7437	CO ₂ (0.04%) + 15 mM Acetate	0.9	N	PHB	Erlenmeyer flask	(Hai et al., 2001)
<i>Synechococcus elongatus</i>	CO ₂ (0.04%) + 1% Sucrose	17.15	N	PHA ^b	Erlenmeyer flask	(Mendhulkar and Shetye, 2017)

<i>Synechococcus elongatus</i>	CO ₂ (0.04%) + 1% Fructose	7.02	P	PHA ^b	Erlenmeyer flask	(Mendhulkar and Shetye, 2017)
<i>Synechococcus</i> sp. MA19	CO ₂ (0.04%) + 15 mM Acetate	6.5	N	PHB	Erlenmeyer flask	(Hai et al., 2001)
<i>Synechocystis</i> PCC 6803	CO ₂ (0.04%) + 0.4% acetate	44–48	N	PHB	Erlenmeyer flask	(Dutt and Srivastava, 2018)
<i>Synechocystis</i> sp. UNIWG	CO ₂ (0.04%) + 0.5% Acetate	14	N	PHB	Erlenmeyer flask	(Toh et al., 2008)

^a Optimized substrates: 0.28 % acetate, 0.38 % glucose, and 0.30 % valerate.

^b not mentioned

Table S2. Mixotrophic production of PHA from *C. necator*.

Microorganism	Substrate used	Process type	Biopolymer Yield (g/L or % in DCW)	Limiting factor	PHA composition	Bioreactor	References
<i>C. necator</i> B-10646	CO ₂ :O ₂ :H ₂ (1:2:7) + Propionate, valerate, hexanoate, c-butyrolactone	Batch, Fed-batch	42.5 g/L (85%) -	N N, K, P, S, Mn	PHB 4HB, 3HV, 3HHx, 3HO	CSTR CSTR	(Volova et al., 2013)
<i>C. necator</i> DSM 545	H ₂ :O ₂ :CO ₂ (84:2.8:13.2 vol%) + Valerate	Fed-batch, semi-continuous	24.7 g/L (78%)	Valerate	PHB-co-PHV	CSTR	(Ghysels et al., 2018)
<i>C. necator</i> ATCC 17697	CO ₂ :H ₂ :O ₂ :N ₂ (1:7:1: 91%) + Valerate	Batch, Fed-batch	1.07 g/L	N	PHB-co-PHV	Erlenmeyer flask	(Park et al., 2014)
<i>C. necator</i> B-5786	CO ₂ :O ₂ :H ₂ (1:2:6) + Valerate	Two-stage batch	8-10 g/L (60%)	Valerate	PHB-co-PHV	CSTR	(Volova and Kalacheva, 2005)
<i>C. necator</i> B-5786	CO ₂ :O ₂ :H ₂ (1:2:6)	Batch	12 g/L (63%)	N	PHB	CSTR	(Volova et al., 2004)
<i>C. necator</i> DSM 545	H ₂ :O ₂ :CO ₂ (84:2.8:13.2 vol%) + Glycerol	Fed-batch	28 g/L	N, O ₂	PHB	CSTR	(Garcia-Gonzalez et al., 2015)
<i>C. necator</i> H16 and B5786	CO ₂	Fed-batch	63.0 & 61.4 %	N	PHB-co-PHV-co-PHHx	Erlenmeyer flask	(Volova et al., 2008)

<i>C. necator</i> ATCC 17697 ^T	H ₂ :O ₂ :CO ₂ (75:15:10)	Batch	36 g/L	H ₂ , O ₂ , N	PHB	CSTR	(Ishizaki and Tanaka, 1991)
<i>C. necator</i> DSM 545	H ₂ :O ₂ :CO ₂ (84:2.8:13.2) vol%	Two-phase batch	24 g/L (63%)	N	PHB	CSTR	(Garcia- Gonzalez and De Wever, 2017)

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