

## Supplementary Information

### Electrodeposition of Nanocrystalline $\text{Fe}_x\text{Co}_{1-x}$ Thin Films from Choline Chloride-Urea Deep Eutectic Solvents

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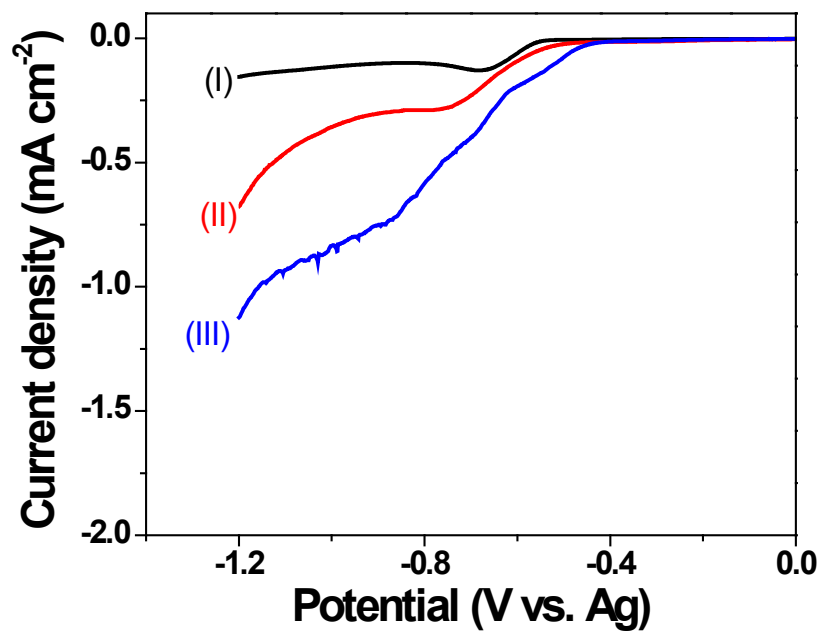
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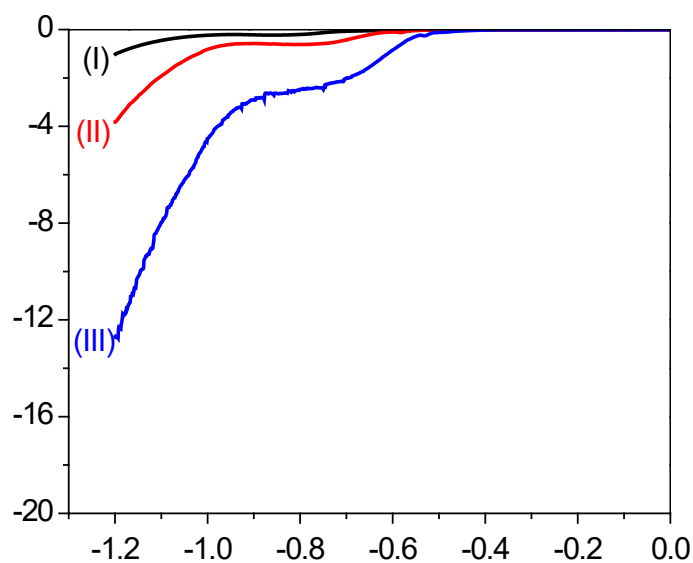
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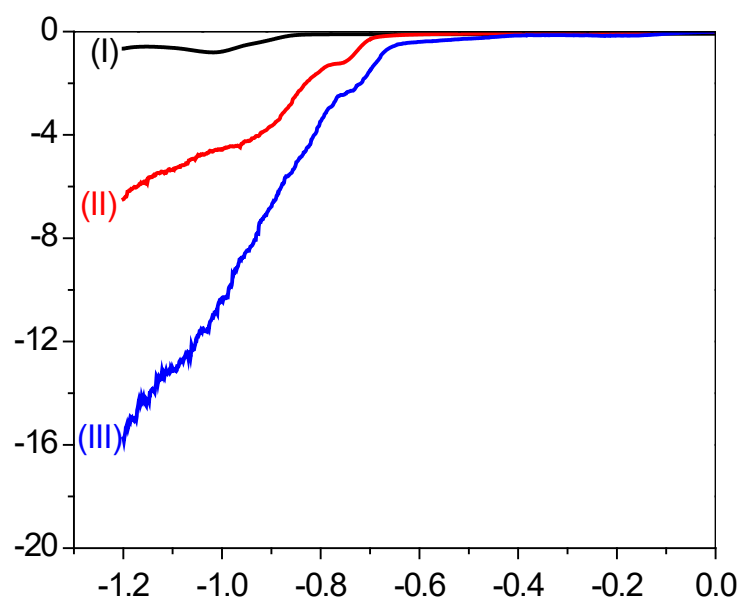
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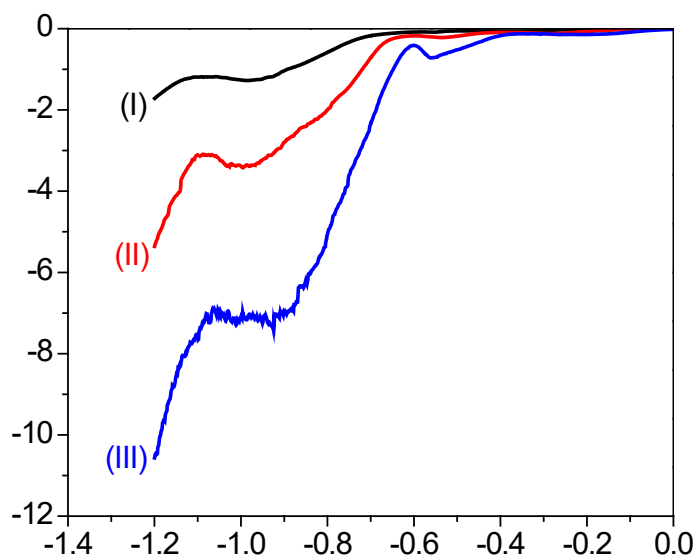
**Figure S1.** LSV curves with DES only at different temperatures (I): 70, (II): 100, (III): 130 °C. The scan rate was fixed at 1 mV/s.



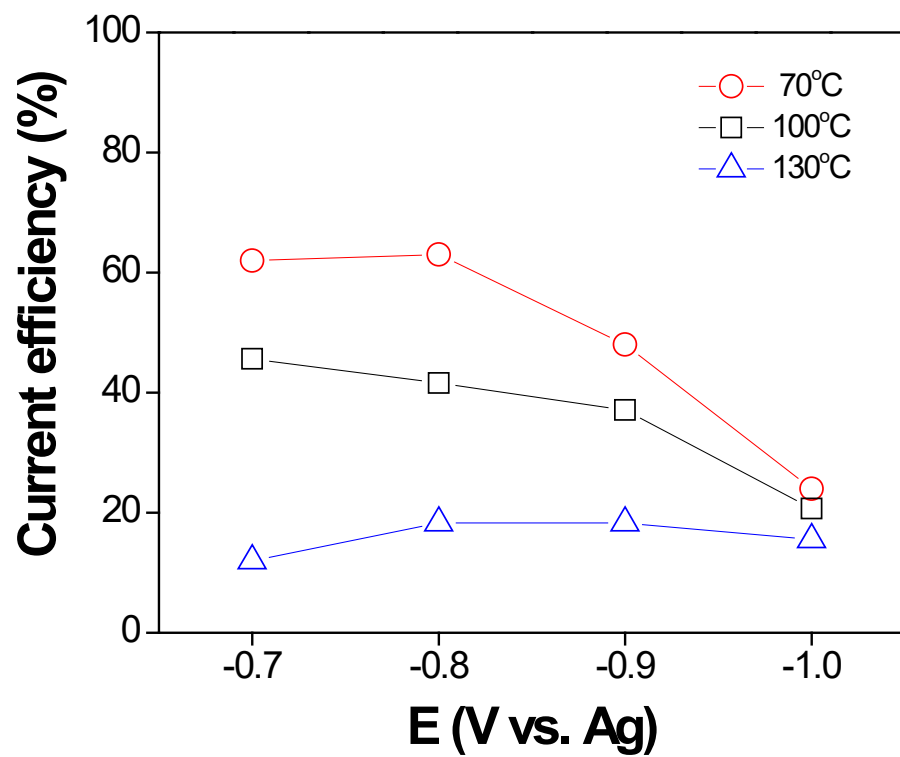
**Figure S2.** LSV curves for Co electrodepositions in DES at different temperatures (I): 70, (II): 100, (III): 130 °C, with 15 mM  $\text{CoCl}_2$ . The scan rate was fixed at 1 mV/s.



**Figure S3.** LSV curves for Fe electrodepositions in DES at different temperatures (I): 70, (II): 100, (III): 130 °C, with 85 mM FeCl<sub>3</sub>. The scan rate was fixed at 1 mV/s.



**Figure S4.** LSV curves CoFe electrodepositions in DES at different temperatures (I): 70, (II): 100, (III): 130 °C, with 15 mM CoCl<sub>2</sub> + 85 mM FeCl<sub>3</sub>. The scan rate was fixed at 1 mV/s.



**Figure S5.** Current efficiency of CoFe film electrodeposited at different applied potential. The electrolyte consisted of 15 mM  $\text{CoCl}_2$  + 85 mM  $\text{FeCl}_3$ .

**Table S.1.** Comparison of the magnetic properties of electrodeposited FeCo thin films from both aqueous and deep eutectic solvents baths.

| <i>Bath Type</i>           | <i>Fe content (at%)</i> | <i>Saturation Magnetization (M<sub>s</sub>) (T)</i> | <i>Coercivity (H<sub>c</sub>) (Oe)</i> | <i>Reference</i>              |
|----------------------------|-------------------------|-----------------------------------------------------|----------------------------------------|-------------------------------|
| <i>Aqueous</i>             | <i>60 - 80</i>          | <i>1.8 – 2.4</i>                                    | <i>50 – 250</i>                        | <i>Zhou et al., 2012</i>      |
| <i>Aqueous (Nanowires)</i> | <i>70</i>               | <i>~2.3</i>                                         | <i>&lt;100</i>                         | <i>DTIC Report, 2025</i>      |
| <i>Aqueous</i>             | <i>10 - 90</i>          | <i>1.5 – 2.3</i>                                    | <i>60 – 300</i>                        | <i>Gonçalves et al., 2023</i> |
| <i>Aqueous</i>             | <i>65</i>               | <i>2.1</i>                                          | <i>80</i>                              | <i>Rahman et al., 2007</i>    |
| <i>Aqueous</i>             | <i>70</i>               | <i>2.2</i>                                          | <i>100</i>                             | <i>Ricq et al., 2001</i>      |
| <i>Aqueous</i>             | <i>0 - 10</i>           | <i>1.8 to 2.4</i>                                   | <i>8 to 12</i>                         | <i>Liao, 1987</i>             |
| <i>Aqueous</i>             | <i>0 - 80</i>           | <i>2.3 to 1.5</i>                                   | <i>20 - 62</i>                         | <i>Myung et al, 2004</i>      |
| <i>Aqueous</i>             | <i>55</i>               | <i>2.2</i>                                          | <i>30 – 60</i>                         | <i>Osaka et al., 1998</i>     |
| <i>Aqueous</i>             | <i>70</i>               | <i>2.0</i>                                          | <i>60 – 90</i>                         | <i>Lallemant et al., 2005</i> |
| <i>Non-Aqueous (DES)</i>   | <i>75</i>               | <i>1.9</i>                                          | <i>50 – 100</i>                        | <i>Doneux et al., 2024</i>    |
| <i>Non-Aqueous (DES)</i>   | <i>65-85</i>            | <i>0.8-1.74</i>                                     | <i>290-350</i>                         | <i>Side et al., 2019</i>      |
| <i>Non-Aqueous (DES)</i>   | <i>30-40</i>            | <i>0.2-1.6</i>                                      | <i>50 – 250</i>                        | <i>This work</i>              |

## **References:**

Doneux, T., Sorgho, A., Soma, F., Rayée, Q., & Bougouma, M. (2024). *Electrodeposition in Deep Eutectic Solvents: The “Obvious”, the “Unexpected” and the “Wonders”*. *Molecules*, 29(14), 3439.

DTIC Report. (2025). *Electrodeposition of FeCo nanowires and their magnetic properties*.

Gonçalves, S., Andrade, V., Sousa, C. T., et al. (2023). *Tunable Iron–Cobalt Thin Films Grown by Electrodeposition*. *Magnetochemistry*, 9(7), 161.

Lallemand, F., Ricq, L., Deschaseaux, E., et al. (2005). *Electrodeposition and magnetic properties of FeCo alloys*. *Surface and Coatings Technology*, 197(1), 10–16.

Liao, S. (1987). *High moment CoFe thin films by electrodeposition*. *IEEE Transactions on Magnetics*, 23(5), 2981–2983. <https://doi.org/10.1109/TMAG.1987.1065555>

Myung, N. V., Park, D.-Y., Urgiles, D. E., & George, T. (2004). *Electroformed iron and FeCo alloy*. *Electrochimica Acta*, 49(25), 4397–4404.

Osaka, T., Takai, M., Hayashi, K., et al. (1998). *Electrodeposition of high-density magnetic recording media*. *Nature*, 392(6679), 796–798.

Rahman, I. Z., et al. (2007). *Magnetic properties of FeCo alloys*. *Journal of Physics: Conference Series*, 61, 523.

Ricq, L., Lallemand, F., Gigandet, M. P., & Pagetti, J. (2001). *Electrodeposition of FeCo alloys and their magnetic behavior*. *Surface and Coatings Technology*, 138(2), 278–284.

Sides, W., Kassouf, N., & Huang, Q. (2019). *Electrodeposition of FeCo alloys in ionic liquids*. *Journal of The Electrochemical Society*, 166, D77.

Zhou, D., Zhou, M., Zhu, M., et al. (2012). *Electrodeposition and magnetic properties of FeCo alloy films*. *Journal of Applied Physics*, 111(7), 07A319.