

Supplementary Material

“Enhanced Water Purification Performance of Ionic Liquid Impregnated MOF: Dye Removal by [BMIM][PF₆]/MIL-53(Al) Composite”

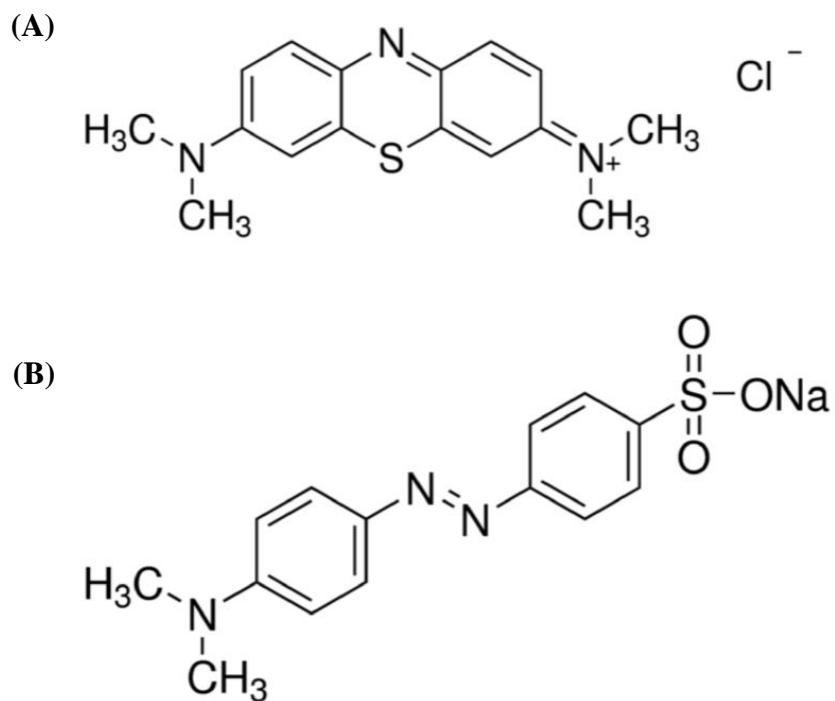


Figure S1. Structure of (A) cationic dye methylene blue (MB), (B) anionic dye methyl orange (MO)

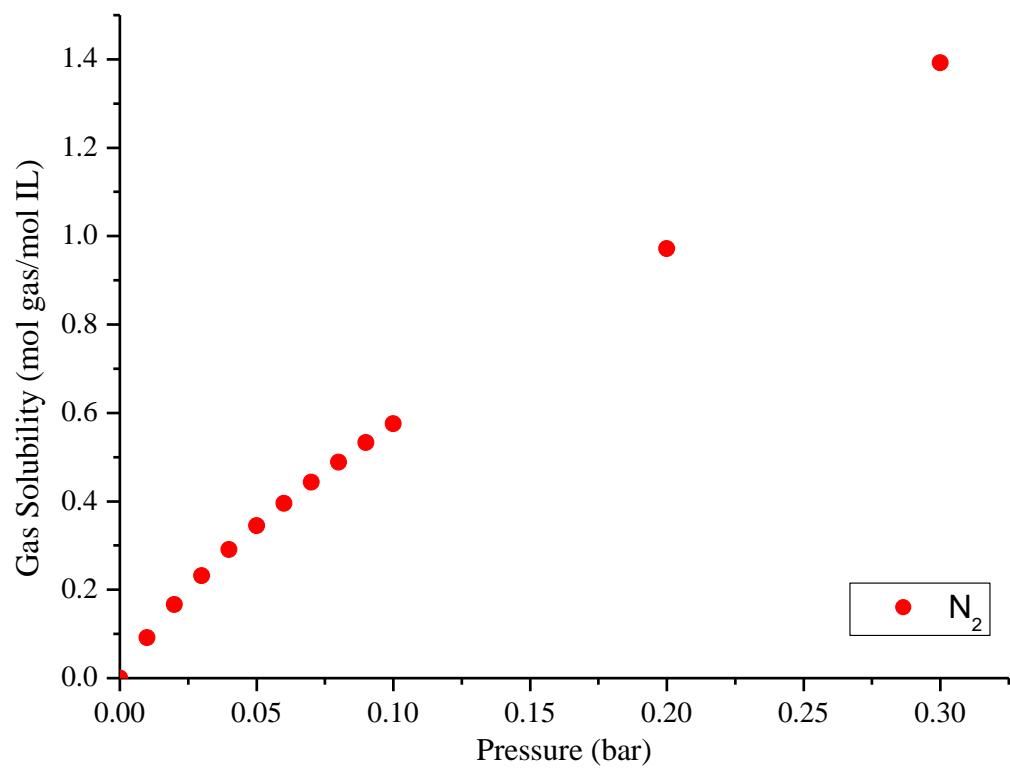


Figure S2. Solubility results of N_2 in bulk [BMIM][PF6] obtained by COSMO-RS calculations.

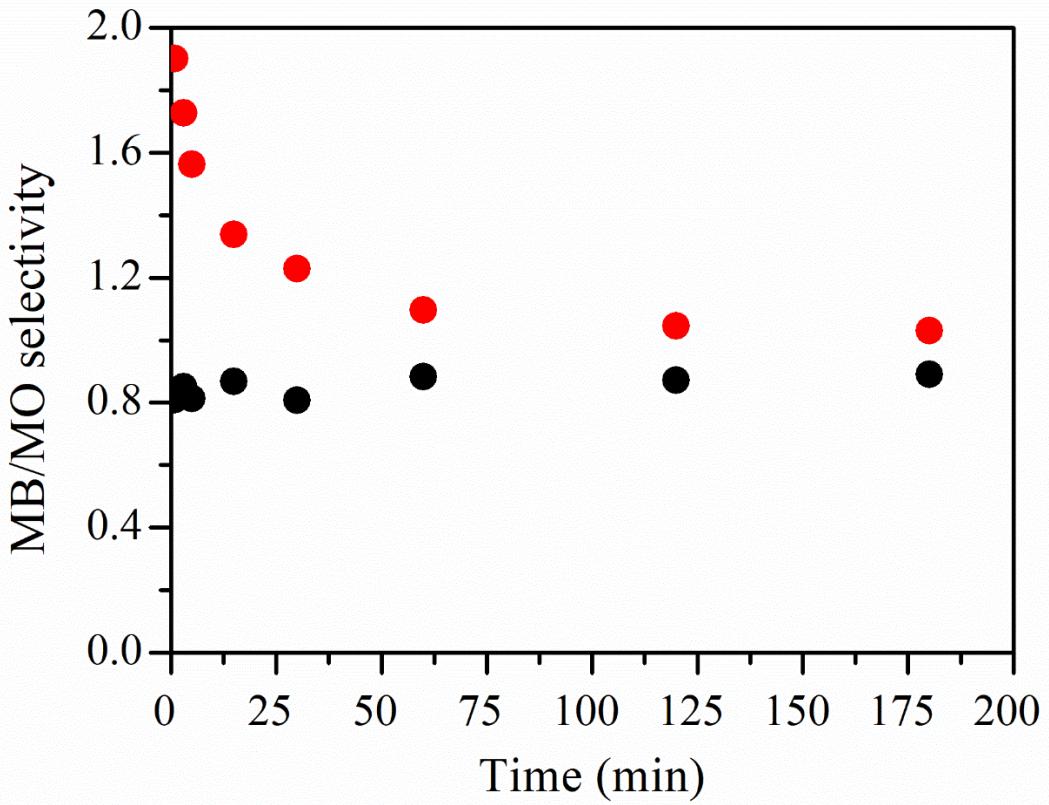


Figure S3. MB/MO removal selectivity of pristine MIL-53(Al) (black) and [BMIM][PF₆]/MIL-53(Al) composite (red).

Table S1. The pseudo-first-order kinetic constants (k_1) together with correlation coefficients (R^2) of MIL-53(Al), and [BMIM][PF₆]/MIL-53(Al).

Adsorbents	Dyes	$q_{e,exp}$ (mg/g)	k_1 (1/min)	$q_{e,cal}$ (mg/g)	R^2	RMSE
MIL-53(Al)	MB	123.22	1.94×10^{-2}	97.27	0.924	0.275
	MO	25.46	1.32×10^{-2}	14.45	0.830	0.412
[BMIM][PF₆]/	MB	162.88	1.99×10^{-2}	29.48	0.579	0.649
	MO	47.12	1.63×10^{-2}	20.14	0.850	0.387

Table S2. The pseudo-second-order kinetic constants (k_2) together with correlation coefficients (R^2) of MIL-53(Al), and [BMIM][PF₆]/MIL-53(Al).

Adsorbents	Dyes	$q_{e,exp}$ (mg/g)	k_2 (g/(mg min))	$q_{e,cal}$ (mg/g)	R^2	RMSE
MIL-53(Al)	MB	123.22	6.90×10^{-4}	123.46	0.980	0.141
	MO	25.46	9.62×10^{-3}	25.32	0.997	0.055
[BMIM][PF₆]/	MB	162.88	7.02×10^{-3}	163.93	0.999	0.032
	MO	47.12	6.36×10^{-3}	47.62	0.999	0.032

Table S3. Langmuir and Freundlich isotherm parameters of MIL-53(Al), and [BMIM][PF₆]/MIL-53(Al).

Adsorbents	Dye	$q_{e,exp}$ (mg/g)	Langmuir Model				Freundlich Model			
			K_L (L/mg)	q_{max} (mg/g)	R^2	$RMSE$	K_F (mg/g)	n (g/L)	R^2	$RMSE$
MIL-53(Al)	MB	84.546	0.371	78.125	0.949	0.226	93.504	43.29 0	0.0102	0.995
	MO	44.089	1.421	46.296	0.991	0.095	21.909	5.120	0.676	0.569
[BMIM][PF ₆]/ MIL-53(Al)	MB	204.905	4.083	204.082	0.999	0.032	149.038	11.62 8	0.934	0.257
	MO	60.283	0.182	57.803	0.991	0.095	147.496	5.133	0.0817	0.958

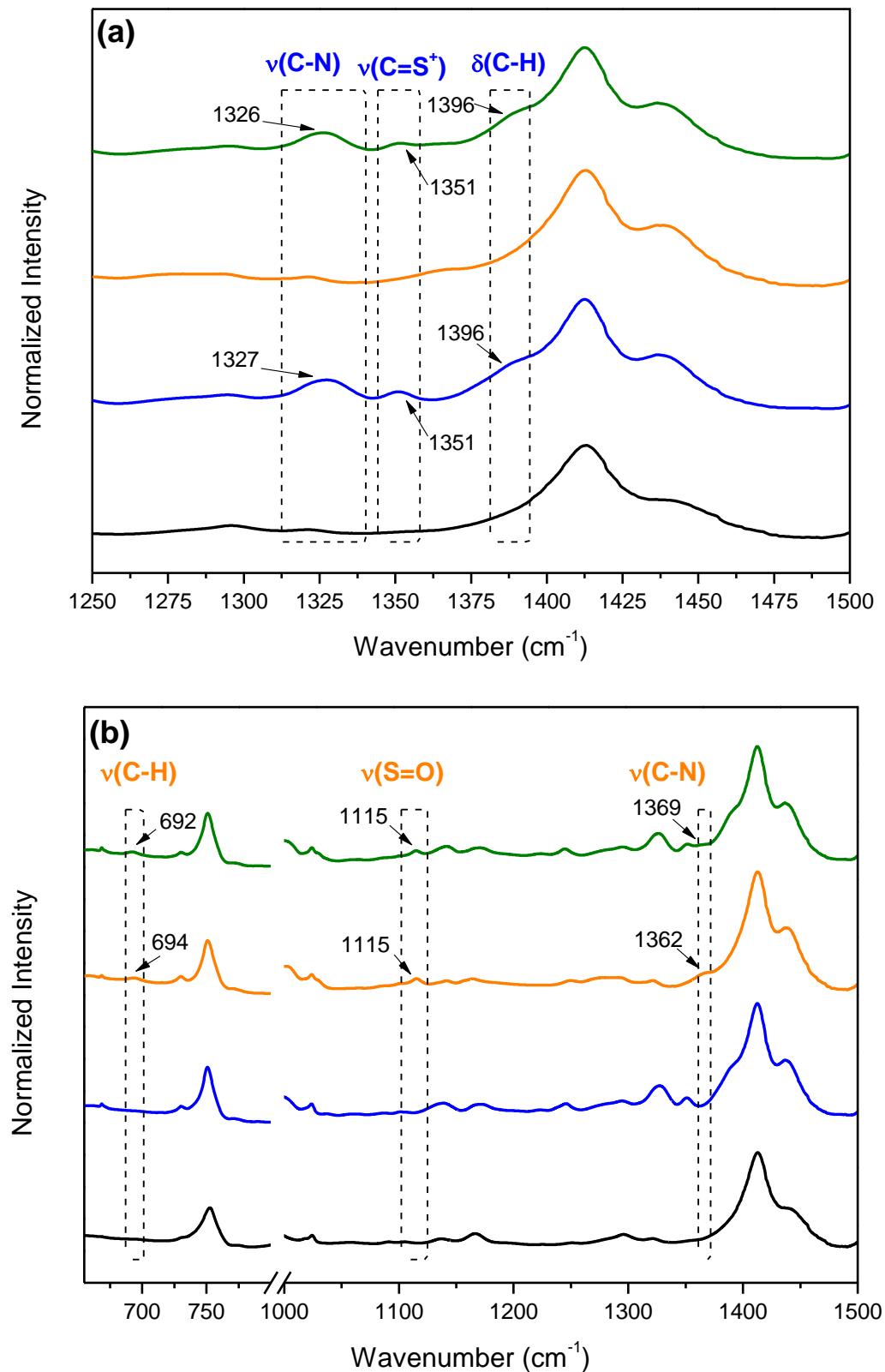


Figure S4. Characteristic FTIR peaks of (A) MB, and (B) MO dyes on MB adsorbed (blue), MO adsorbed (orange), and mixture of MB and MO adsorbed (olive) $[\text{BMIM}][\text{PF}_6]/\text{MIL-53}(\text{Al})$. Black line represents the FTIR spectrum of $[\text{BMIM}][\text{PF}_6]/\text{MIL-53}(\text{Al})$ before dye adsorption.

Table S4. Shifts in the characteristic FTIR peaks of MOF after dye adsorption (relative to the IL/MOF composite).

MOF Related Peaks (cm ⁻¹)	IL/MOF Composite (cm ⁻¹)	MB-adsorbed IL/MOF Composite (cm ⁻¹)	MO-adsorbed IL/MOF Composite (cm ⁻¹)	Mixture-adsorbed IL/MOF Composite (cm ⁻¹)	Assignment
658	654	+4	-	+4	v _s (Al-O-Al)
687	668	+1	-	+1	v _{as} (Al-O-Al)
3706	3700	-17	-	-3	μ ₂ (O-H)

Table S5. Changes in the characteristic FTIR peaks of IL after dye adsorption (relative to the IL/MOF composite).

IL Related Peaks (cm ⁻¹)	IL/MOF Composite (cm ⁻¹)	MB-adsorbed IL/MOF Composite (cm ⁻¹)	MO-adsorbed IL/MOF Composite (cm ⁻¹)	Mixture-adsorbed IL/MOF Composite (cm ⁻¹)	Assignment
737	731	+4	-4	+1	v _{as} (PF ₆) of IL's anion
837	837	+3	-	+1	v _{as} (PF ₆) of IL's anion
3124	3118	-1	+3	-	v(C(2)-H) of IL's cation
3170	3162	-3	+3	-	v _{ss} (C(4)HC(5)H) of IL's cation

Table S6. Changes in the characteristic FTIR peaks of MB after dye adsorption (relative to the IL/MOF composite).

MB Related Peaks (cm⁻¹)	IL/MOF Composite (cm⁻¹)	MB-adsorbed IL/MOF Composite (cm⁻¹)	MO-adsorbed IL/MOF Composite (cm⁻¹)	Mixture-adsorbed IL/MOF Composite (cm⁻¹)	Assignment
885	-	newly formed	-	newly formed	N _{het}HO
1334	-	-7	-	-7	v(C _{het} -N)
1355	-	-4	-	-4	v(C=S ⁺)
1390	-	+2	-	+2	δ(C _{het} -H)
1590	-	+5	-	+5	v _{ip} (C _{het} -H)
between 1690-1710	-	-4	-	-2	v(C _{het} =N ⁺ (CH ₃) ₂)

Table S7. Changes in the characteristic FTIR peaks of MO after dye adsorption (relative to the IL/MOF composite).

MO Related Peaks (cm⁻¹)	IL/MOF Composite (cm⁻¹)	MB-adsorbed IL/MOF Composite (cm⁻¹)	MO-adsorbed IL/MOF Composite (cm⁻¹)	Mixture-adsorbed IL/MOF Composite (cm⁻¹)	Assignment
1004	-	-	-2	-2	v(N-(CH ₃) ₂)
1036	-	-	-11	-10	v(SO ₃)
1113	-	-	+2	+2	v(C-SO ₃)
1195	-	-	-22	-23	v(S=O)
1420	-	-	-5	-6	v(N=N)
1444	-	-	-5	-6	v(N=N)
1518	-	-	-6	-3	v(C-N(CH ₃) ₂)