

Magnesium ions moderate calcium-induced calcium release in cardiac calcium release sites by binding to ryanodine receptor activation and inhibition sites

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Supplementary Material

1 Supplementary Data

1.1 Derivation of the equation for RyR open probability calculations

The RyR channel was described as an MWC-type homotetrameric molecule with one activation site on each monomer, for which Ca^{2+} and Mg^{2+} ions compete, and with one common inhibition site that binds two Mg^{2+} ions (Figure 2 of the main text).

Equation 11 from the main text was derived using the statistical mechanics framework (Marzen et al., 2013). Each state of the RyR molecule (Figure 2 of the main text) can be denoted as $X_{i,j,k}$, where $X \in \{C, O\}$ is the macrostate of the channel (open or closed), *i* and *j* are the numbers of bound Ca²⁺ and Mg²⁺ ions to the RyR activation site, respectively, and *k* is the number of magnesium Mg²⁺ ions bound to the RyR inhibition site. The energy of a molecule can be decomposed into the conformational energy and the binding energies of ligands. In a homotetrameric molecule of the MWC type, the conformational energy of each monomer is the same and corresponds to the state of the molecule as a whole (in our case, open or closed). The binding energies depend only on the macrostate *X*. We considered the same binding energy for the RyR activation site of all monomers, which leads to the existence of several equivalent microstates, i.e., there are several combinations of the number of *i* calcium and *j* magnesium ions bound to the four equal activation binding sites. Disregarding the RyR interaction with Mg²⁺ at the RyR inhibition site, which will be added to the final equation later, we can write the energy of the molecule in the state $X_{i,j}$ as follows:

$$E_{X_{i,j}} = E_X + iE_{b_X}^{Ca^{2+}} + jE_{b_X}^{Mg^{2+}},$$
(S1)

where E_X is the conformational energy in the macrostate $X_{i,j}$, $E_{b_X}^{Ca^{2+}}$ and $E_{b_X}^{Mg^{2+}}$ are the binding energies of calcium and magnesium ions when the molecule is in the macrostate X. The weight $w_{X_{i,j}}$ of each state equals to $g(X_{i,j})e^{-\beta(E_{X_{i,j}}-(i\mu_{Ca^{2+}}+j\mu_{Mg^{2+}}))}$ where $\beta = \frac{1}{k_BT}$, μ is the chemical potential (the energy cost of moving a ligand molecule from the solution to the receptor) and $g(X_{i,j})$ is the number of microstates of the state $X_{i,j}$, which in our case is the number of ways to put *i* calcium and *j* magnesium ions on the binding sites of the four monomers. The value of $g(X_{i,j})$ is calculated as the product of the respective binomial coefficients:

$$g(X_{i,j}) = \binom{4}{k}\binom{4-k}{l}.$$
(S2)

The chemical potential μ depends on the ligand concentration as follows:

$$\mu_c = \mu_{c_{ref}} + \frac{1}{\beta} \ln \frac{c}{c_{ref}}, \qquad (S3)$$

where *c* is the concentration, c_{ref} is the reference concentration, and $\mu_{c_{ref}}$ is the reference chemical potential of the ligand. Combining equations S1-S3, the weight equals to:

$$w_{X_{i,j}} = \binom{4}{i} \binom{4-i}{j} \left(\frac{[Ca^{2+}]}{[Ca^{2+}]_{ref}}\right)^{i} \left(\frac{[Mg^{2+}]}{[Mg^{2+}]_{ref}}\right)^{j} e^{-\beta(E_{X}+i(E_{b_{X}}^{Ca^{2+}}-\mu_{Ca^{2+}_{ref}})+j(E_{b_{X}}^{Mg^{2+}}-\mu_{Mg^{2+}_{ref}}))}$$
(S4)

Open probability is the fraction of weights of open states:

$$P_0 = \frac{w_{open}}{w_{open} + w_{closed}} \,. \tag{S5}$$

The weight of a macrostate is the sum of weights of all corresponding microstates. These weights can be calculated as follows:

$$w_{open} = e^{-\beta E_O} \sum_{i=0}^{4} \sum_{j=0}^{4-i} {4 \choose i} {4-i \choose j} \left(\frac{[Ca^{2+}]}{[Ca^{2+}]_{ref}} \right)^i \left(\frac{[Mg^{2+}]}{[Mg^{2+}]_{ref}} \right)^j e^{-\beta (i(E_{b_O}^{Ca^{2+}} - \mu_{Ca_{ref}^{2+}}) + j(E_{b_O}^{Mg^{2+}} - \mu_{Mg_{ref}^{2+}}))}$$
$$= e^{-\beta E_O} \left(1 + \frac{[Ca^{2+}]}{[Ca^{2+}]_{ref}} e^{-\beta \left(E_{b_O}^{Ca^{2+}} - \mu_{Ca_{ref}^{2+}} \right)} + \frac{[Mg^{2+}]}{[Mg^{2+}]_{ref}} e^{-\beta \left(E_{b_O}^{Mg^{2+}} - \mu_{Mg_{ref}^{2+}} \right)} \right)^4, \tag{S6}$$

 W_{closed}

$$= e^{-\beta E_{C}} \sum_{i=0}^{4} \sum_{j=0}^{4-i} {4 \choose i} {4-i \choose j} \left(\frac{[Ca^{2+}]}{[Ca^{2+}]_{ref}} \right)^{i} \left(\frac{[Mg^{2+}]}{[Mg^{2+}]_{ref}} \right)^{j} e^{-\beta (i(E_{b_{C}}^{Ca^{2+}} - \mu_{Ca_{ref}^{2+}}) + j(E_{b_{C}}^{Mg^{2+}} - \mu_{Mg_{ref}^{2+}}))}$$
$$= e^{-\beta E_{C}} \left(1 + \frac{[Ca^{2+}]}{[Ca^{2+}]_{ref}} e^{-\beta \left(E_{b_{C}}^{Ca^{2+}} - \mu_{Ca_{ref}^{2+}} \right) + \frac{[Mg^{2+}]}{[Mg^{2+}]_{ref}}} e^{-\beta \left(E_{b_{C}}^{Mg^{2+}} - \mu_{Mg_{ref}^{2+}} \right)} \right)^{4}.$$
(S7)

From these equations, the channel open probability P_0 at different concentrations of Ca^{2+} and Mg^{2+} in the absence of the RyR inhibition site can be calculated as:

$$P_{O}([Ca^{2+}], [Mg^{2+}]) = \frac{e^{-\beta E_{O}\left(1 + \frac{[Ca^{2+}]}{[Ca^{2+}]_{ref}}e^{-\beta\left(E_{bO}^{Ca^{2+}} - \mu_{Ca^{2+}_{ref}}\right) + \frac{[Mg^{2+}]}{[Mg^{2+}]_{ref}}e^{-\beta\left(E_{bO}^{Mg^{2+}} - \mu_{Mg^{2+}_{ref}}\right)\right)^{4}}}{e^{-\beta E_{O}\left(1 + \frac{[Ca^{2+}]}{[Ca^{2+}]_{ref}}e^{-\beta\left(E_{bO}^{Ca^{2+}} - \mu_{Ca^{2+}_{ref}}\right) + \frac{[Mg^{2+}]}{[Mg^{2+}]_{ref}}e^{-\beta\left(E_{bO}^{Mg^{2+}} - \mu_{Mg^{2+}_{ref}}\right)\right)^{4}} + e^{-\beta E_{O}\left(1 + \frac{[Ca^{2+}]}{[Ca^{2+}]_{ref}}e^{-\beta\left(E_{bO}^{Ca^{2+}} - \mu_{Ca^{2+}_{ref}}\right) + \frac{[Mg^{2+}]}{[Mg^{2+}]_{ref}}e^{-\beta\left(E_{bO}^{Mg^{2+}} - \mu_{Mg^{2+}_{ref}}\right)\right)^{4}}}}$$
(S5)

Substituting variables in equation (S5):

$$K_{00} = e^{-\beta(E_C - E_0)},$$
(S6)

$$K_{ACa} = [Ca^{2+}]_{ref} e^{\beta \left(E_{b_c}^{Ca^{2+}} - \mu_{Ca_{ref}}^{2+} \right)},$$
(S7)

$$K_{AMg} = [Mg^{2+}]_{ref} e^{\beta \left(E_{b_c}^{Mg^{2+}} - \mu_{Mg^{2+}_{ref}} \right)},$$
(S8)

$$f_{Ca} = e^{-\beta \left(E_{b_C}^{Ca^{2+}} - E_{b_O}^{Ca^{2+}} \right)},\tag{S9}$$

$$f_{Mg} = e^{-\beta \left(E_{b_C}^{Mg^{2+}} - E_{b_O}^{Mg^{2+}}\right)}.$$
(S10)

where K_{Ax} stays for the respective equilibrium constants and f_x for the respective allosteric factors of the ion *x*, we obtain P_O in the absence of the RyR inhibition site:

$$P_{O}([Ca^{2+}], [Mg^{2+}]) = \frac{\left(1 + \frac{[Ca^{2+}]}{f_{Ca}K_{ACa}} + \frac{[Mg^{2+}]}{f_{Mg}K_{AMg}}\right)^{4}}{\left(1 + \frac{[Ca^{2+}]}{f_{Ca}K_{ACa}} + \frac{[Mg^{2+}]}{f_{Mg}K_{AMg}}\right)^{4} + K_{O0}\left(1 + \frac{[Ca^{2+}]}{K_{ACa}} + \frac{[Mg^{2+}]}{K_{AMg}}\right)^{4}}$$
$$= \frac{\left([Ca^{2+}] + f_{Ca}K_{ca} + \frac{f_{Ca}K_{ACa}}{f_{Mg}K_{AMg}}[Mg^{2+}]\right)^{4}}{\left([Ca^{2+}] + f_{Ca}K_{ACa} + \frac{f_{Ca}K_{ACa}}{f_{Mg}K_{AMg}}[Mg^{2+}]\right)^{4} + K_{O0}f_{Ca}^{4}\left(K_{ACa} + Ca + \frac{K_{ACa}}{K_{AMg}}[Mg^{2+}]\right)^{4}}.$$
(S11)

Magnesium binding to the RyR inhibition site was treated as independent of the state of the RyR activation sites. This means that binding of Mg^{2+} to the inhibition site renders the RyR channel non-conductive regardless of the state of the activation sites. Considering experimental data (Laver et al., 1997; Zahradnikova et al., 2003), the respective rate constant of Mg^{2+} binding was set proportional to the square of magnesium concentration and the rate of Mg^{2+} unbinding was set independent of magnesium concentration. This allows us to write an equation for P_O of the full model in the form:

$$P_{O}([Ca^{2+}], [Mg^{2+}]) = \frac{K_{IMg}^{2}}{K_{IMg}^{2} + [Mg^{2+}]^{2}} \frac{\left([Ca^{2+}] + K_{ACa}f_{Ca}\left(1 + \frac{[Mg^{2+}]}{K_{AMg}f_{Mg}}\right)\right)^{4}}{\left([Ca^{2+}] + K_{ACa}f_{Ca}\left(1 + \frac{[Mg^{2+}]}{K_{AMg}f_{Mg}}\right)\right)^{4} + K_{O0}f_{Ca}^{4}\left(Ca + K_{ACa}\left(1 + \frac{[Mg^{2+}]}{K_{AMg}}\right)\right)^{4}}, \quad (S12)$$

where K_{IMg} is the dissociation constant of Mg²⁺ ions at the RyR inhibition site.

1.2 Construction of the Q-matrix of the RyR gating model

The presented model of RyR gating is a continuous-time Markov chain hence it is defined by a transition rate matrix Q. This section describes the equations used for the construction of the Q-matrix (see Methods in the main text).

For the construction of the Q-matrix, the states O_{ij0} are considered open states, and the nonconductive states C_{ijk} and O_{ij2} are considered closed states. The transition rates are written in Supplementary Table S1.

2 Supplementary Figures and Tables

2.1 Supplementary Tables

Supplementary Table S1. Transition rates within the RyR gating model.

Transition	Rate							
Ca ²⁺ binding to the activation site								
$C_{i,j,k} \to C_{i+1,j,k}, i+j < 4$	$(4 - i - j) \cdot k_{AonCa} \cdot [Ca^{2+}]$							
$O_{i,j,k} \to O_{i+1,j,k}, i+j < 4$	$(4 - i - j) \cdot k_{AonCa} \cdot \frac{[Ca^{2+}]}{f_{Ca}^{\lambda_{onCa}}}$							
Ca ²⁺ unbinding from the activation site								
$C_{i,j,k} ightarrow C_{i-1,j,k}$, $i > 0$	$i \cdot k_{AoffCa}$							
$O_{i,j,k} \rightarrow O_{i-1,j,k}, i > 0$	$i \cdot k_{AoffCa} \cdot f_{Ca}^{1-\lambda_{onCa}}$							
Mg ²⁺ binding to the activation site								
$C_{i,j,k} \rightarrow C_{i,j+1,k}, i+j < 4$	$(4 - i - j) \cdot k_{AonMg} \cdot [Mg^{2+}]$							
$O_{i,j,k} \to O_{i,j+1,k}, i+j < 4$	$(4 - i - j) \cdot k_{AonMg} \cdot \frac{[Mg^{2+}]}{f_{Mg}^{\lambda_{onMg}}}$							
Mg ²⁺ unbinding from the activation site								
$C_{i,j,k} \to C_{i,j-1,k}, j > 0$	$j \cdot k_{AoffMg}$							
$O_{i,j,k} \rightarrow O_{i,j-1,k}, j > 0$	$j \cdot k_{AoffMg} \cdot f_{Mg}^{1 - \lambda_{onMg}}$							
Transitions between O- and C-states								
$C_{i,j,k} \to O_{i,j,k}$	$\frac{k_{CO}}{(f_{Ca}^i f_{Mg}^j)^{\lambda_{CO}}}$							
$O_{i,j,k} \to C_{i,j,k}$	$k_{oc}(f_{ca}^i f_{Mg}^j)^{1-\lambda_{co}}$							
Mg ²⁺ binding/unbinding at the inhibition site								
$X_{i,j,0} \to X_{i,j,2}$	$k_{IonMg}[Mg^{2+}]^2$							
$X_{i,j,2} \to X_{i,j,0}$	k_{IoffMg}							

Supplementary Table S2. Mg-binding parameters used in this study. Numbers in boldface represent the reference values estimated by fitting the model to data from the literature (see Methods).

k_{AonMg} (μ M ⁻¹ ms ⁻¹)	k_{AoffMg} (ms ⁻¹)	f_{Mg}		k_{IonMg} (10 ⁻⁷ µM ⁻² ms ⁻¹)	k_{IoffMg} (ms ⁻¹)
0.000596	0.055	0.5	10	1.842	0.055
0.001107	0.102184	1	25	2.857	0.08529
0.002057	0.189846	1.56292535	40	3.871	0.11558
0.003822	0.352712	2.1258507	55	4.886	0.14587
0.0071	0.655298	2.68877605	70	5.9002	0.17616
0.009923	0.915828	3.2517014	85	8.6120	0.25712
0.013868	1.279939	4.93877605	100	11.323	0.33808
0.019381	1.788812	6.6258507	500	1.4035	0.41904
0.027087	2.5	8.31292535		1.6746	0.5

3 References

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