



Phenol Derivatives From the Sponge-Derived Fungus *Didymellaceae* sp. SCSIO F46

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Seven new phenol derivatives named coleophomones E and F (**1**, **2**), diorcinols L and M (**3**, **4**), 1-hydroxy-6-methyl-11-methoxy-8-hydroxymethylxanthone (**5**), porric acid E (**6**), and 7-(2-hydroxyphenyl) butane-7,8,9-triol (**7**), were isolated from the EtOAc extract of the marine sponge-derived fungus *Didymellaceae* sp. SCSIO F46, together with 10 known compounds. Their structures were determined by spectroscopic analyses, including NMR, MS, X-ray diffraction, and theoretical calculations. Each of **1** and **2** contains an unusual spiro [cyclohexane-1,2'-inden] moiety, which is relatively seldom in nature products. Cytotoxic and COX-2 inhibitory activities of all purified compounds were tested and evaluated. Compound **3** displayed obvious cytotoxicities against Huh-7, HeLa, DU145 and HL60 cells (IC₅₀ values 5.7–9.6 μM) and weak activities against other five cell lines, while **8** showed weak cytotoxicities against HeLa and HL7702 cells. Compound **6** displayed COX-2 inhibitory activity with IC₅₀ value of 3.3 μM.

Keywords: sponge-derived fungus, *Didymellaceae* sp., Phenol derivatives, cytotoxic, COX-2 inhibitory

INTRODUCTION

Natural products are still irreplaceable and continuing sources of novel drug leads, especially in the anti-infective area (Newman and Cragg, 2016). The marine ecosystem is one of the most complex and largest aquatic systems on earth, and host a huge microbial biodiversity (Agrawal et al., 2017; Corinaldesi et al., 2017). The unique and extreme characteristics of marine systems have driven a variety of biological adaptations, leading to the production of a large number of novel molecules for the treatment of many diseases (Gerwick and Fenner, 2013; Blunt et al., 2017). Marine sponges, a kind of precious marine organisms for new drug discovery, are hosts for a large community of microbes (up to 50–60% of the biomass of the sponge host) (Bergmann and Burke, 1955; Wang, 2006; Zhang et al., 2017). It was indicated that the symbiotic microbes of marine sponges might be the true producers of bioactive chemical defense substance of the sponge ecosystem (Richelle-Maurer et al., 2003; Thomas et al., 2010).

Sponge-derived fungi have been proven to be a treasure trove of novel biomolecules (Indraningrat et al., 2016; Blunt et al., 2017). During an ongoing search for new bioactive metabolites from the sponge-derived fungi (Tian et al., 2015a, 2018a,b), a strain of *Didymellaceae* sp. (SCSIO F46) isolated from a sponge *Callyspongia* sp. was subjected to chemical study. The EtOAc extract of rice fermentation of F46 showed toxicity against brine shrimp. Further isolation yielded seven new phenol derivatives, coleophomones E, F (**1**, **2**), diorcinols L, M (**3,4**), 1-hydroxy-6-methyl-11-methoxy-8-hydroxymethylxanthone (**5**), porric acid E (**6**), and 7-(2-hydroxyphenyl)

butane-7,8,9-triol (7), together with ten known compounds (Figure 1). The cytotoxic and COX-2 inhibitory activities of all compounds were evaluated. Details of the isolation structure elucidation, and bioactivity screening of these metabolites are reported herein.

MATERIALS AND METHODS

General Experimental Procedures

The NMR spectra were recorded on a Bruker AC 500 NMR (Bruker, Fällanden, Switzerland) spectrometer with TMS as an internal standard. HRESIMS data were measured on a Bruker micro TOF-QII mass spectrometer (Bruker, Fällanden, Switzerland). UV spectra were recorded on a Shimadzu UV-2600 UV-Vis spectrophotometer (Shimadzu, Kyoto, Japan). ECD spectra were performed on a Chirascan circular dichroism spectrometer (Applied Photophysics). X-ray diffraction intensity data were collected on a CrysAlis PRO charge-coupled device (CCD) area detector diffractometer with graphite monochromated Cu $K\alpha$ radiation ($\lambda = 1.54178 \text{ \AA}$). Semi-preparative reversed-phase HPLC (RP-HPLC) was performed on a YMC-Pack Pro C₁₈ RS column (5 μm , 250 \times 10 mm id; YMC, Kyoto, Japan) with a Agilent 1260 separation module equipped with a Photodiode Array (PDA) detector. Silica gel GF254 used for TLC were supplied by the Qingdao Marine Chemical Factory, Qingdao, China. Sephadex LH-20 gel (GE Healthcare, Uppsala, Sweden) was used. Spots were detected on TLC under UV light or by heating by spraying with 12% H₂SO₄ in H₂O.

Fungal Material

The fungal strain SCSIO F46 was isolated from a sponge *Callyspongia* sp., collected from the sea area near Xuwen County, Guangdong Province, China, during August 2013. The isolate was stored on MB agar (malt extract 15 g, sea salt 10 g, agar 15 g) slants at 4°C and then deposited at CAS Key Laboratory of Tropical Marine Bio-resources and Ecology. The fungus was identified using a molecular biological protocol by DNA amplification and sequencing of the ITS region. The nucleotide sequence of the ITS region reported in this article was assigned the GenBank accession number KU361223.

Extraction and Isolation

Didymellaceae sp. SCSIO F46 was cultured on MB-agar plates at 25°C for 7 days. The seed medium (malt extract 15 g, sea salt 10 g, distilled water 1,000 mL, pH 7.4–7.8) was inoculated with strain F46 and incubated at 25°C for 72 h on a rotating shaker (170 rpm). Mass scale fermentation of F46 was carried out using solid rice medium in 1,000 mL flasks (rice 200 g, sea salt 2.5 g, distilled water 200 mL), and inoculated with 10 mL of seed solution. Flasks were incubated at 25°C under normal day night cycle. After 30 days, cultures from 30 flasks were harvested. The culture of solid rice medium was soaked in acetone and cut into small pieces and kept for 1 day. The content was filtered and evaporated under vacuum and extracted with EtOAc thrice. The extract was partitioned between petroleum ether, and 90% aqueous MeOH to obtain the crude extract (43.0 g). The crude extract was subjected to silica gel column chromatography (CC) eluted with petroleum

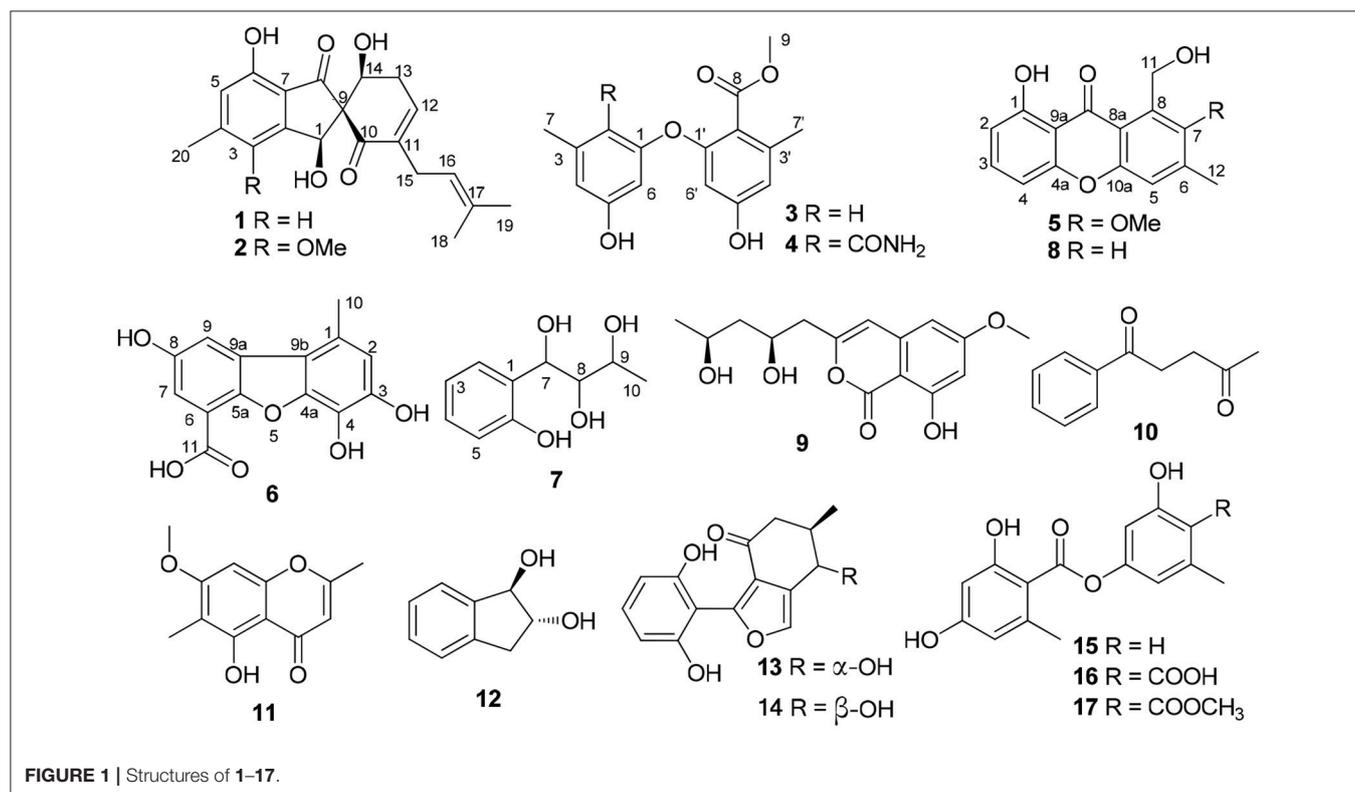


FIGURE 1 | Structures of 1–17.

ether/EtOAc in a gradient eluent (v/v, 50:1, 30:1, 20:1, 10:1, 5:1, 1:1, 0:1) to obtain 8 fractions (fractions 1–8). Fr. 2 was subjected to ODS CC eluted with MeOH/H₂O in a gradient eluent (1:9, 2:3, 3:2, 4:1, 9:1), to give 3 sub fractions (fr. 2.1–2.3). Fr. 2.2 was further purified by HPLC eluting with MeOH/H₂O (60:40) to afford **9** (5.5 mg), **10** (4.7 mg), and **11** (3.8 mg). Fr. 3 was purified by Sephadex LH-20 (CH₃Cl/MeOH, 1:1) to give 3 sub fractions (fr. 3.1–3.3). Fr. 3.2 was further purified by silica gel CC eluted with petroleum ether/EtOAc in a gradient eluent (v/v, 10:1) to obtain **5** (20.1 mg) and **8** (10.3 mg). Fr. 4 was subjected to an ODS column (MeOH/H₂O: 10–100%) to give 4 sub fractions (fr. 4.1–4.4). Fr. 4.2 was further purified by HPLC eluting with MeOH/H₂O (50:50, 1% TFA) to afford **3** (20.8 mg). Compounds **16** (20.3 mg) and **17** (43.2 mg) were purified from Fr. 4.4 by HPLC (35% MeCN, 1% TFA). Fr.5 was purified by ODS CC (MeOH/H₂O: 10–100%) and HPLC (35% MeCN, 1% TFA) to afford **1** (7.4 mg), **2** (4.3 mg) and **6** (5.5 mg). Compound **12** (2.3 mg) was isolated from Fr.6 by an ODS CC and followed by HPLC using 10% MeCN. Fr.7 was subjected to Sephadex LH-20 CC (CH₃Cl/MeOH, 1:1) to provide 4 subfractions (Frs.7.1–7.4). Fr. 7.1 was further purified on HPLC by 25% MeCN (2.5 mL/min) to give **13** (32.3 mg), **14** (43.7 mg). Fr. 7.2 was purified on HPLC by 10% MeCN to give **7** (15 mg). Finally, compounds **15** (4.4 mg) and **4** (3.5 mg) were isolated from Fr.7.3 and Fr. 7.4 by HPLC using 25% MeCN and 30% MeCN, respectively.

Coleophomone E (**1**): Yellow amorphous solid; [α]_D²⁵ +14.1 (c 0.31, MeOH); UV (MeOH) λ_{\max} (log ϵ) 266 (3.16), 322 (2.25)

nm, HRESIMS m/z 343.1541 [M + H]⁺ (calcd for C₂₀H₂₃O₅, 343.1541); ¹H and ¹³C NMR data, **Table 1**.

Coleophomone F (**2**): Yellow amorphous solid; [α]_D²⁵ +13.2 (c 0.46, MeOH); UV (MeOH) λ_{\max} 220 (3.85), 266 (3.20), 313 (2.43) nm, HRESIMS m/z 373.1638 [M + H]⁺ (calcd for C₂₁H₂₅O₆, 373.1646); ¹H and ¹³C NMR data, **Table 1**.

Diorcinol L (**3**): Yellow amorphous solid; HRESIMS m/z 289.1070 [M + H]⁺ (calcd for C₁₆H₁₇O₅, 289.1071); ¹H and ¹³C NMR data, **Table 1**; The structure of **3** have been deposited in the Cambridge Crystallographic Data Centre as supplementary publication number CCDC 1502322.

Diorcinol M (**4**): Yellow amorphous solid; HRESIMS m/z 332.1127 [M + H]⁺ (calcd for C₁₇H₁₈NO₆, 332.1129). ¹H and ¹³C NMR data, **Table 1**.

1-Hydroxy-6-methyl-11-methoxy-8-hydroxymethylxanthone (**5**): Yellow needle crystals; HRESIMS m/z 287.0913 [M + H]⁺ (calcd for C₁₆H₁₅O₅, 287.0919); ¹H and ¹³C NMR data, **Table 2**.

Porric acid E (**6**): Colorless needle crystals; HRESIMS m/z 275.0544 [M + H]⁺ (calcd for C₁₄H₁₁O₆, 275.0550); ¹H and ¹³C NMR data, **Table 2**.

7-(2-Hydroxyphenyl) butane-7,8,9-triol (**7**): Yellow oil; HRESIMS m/z 221.0781 [M + Na]⁺ (calcd for C₁₀H₁₄NaO₄, 221.0784); ¹H and ¹³C NMR data, **Table 2**.

ECD Calculation

The eight possible stereoisomers (**a-h**) of **1** were initially performed using Confab (O'Boyle et al., 2011) with systematic search at MMFF94 force field. Room-temperature equilibrium

TABLE 1 | ¹H, ¹³C NMR data of **1–4** (500/125 MHz, in DMSO-*d*₆, δ ppm, and *J* in Hz).

No.	1		2		No.	3		4	
	δ_H	δ_C , type	δ_H	δ_C , type		δ_H	δ_C , type	δ_H	δ_C , type
1	5.50, s	67.7, CH	5.70, s	66.3, CH	1		158.0, C		155.7, C
2		154.7, C		146.2, C	2	6.22, s	110.3, CH		120.9, C
3	6.84, s	116.8, CH		148.5, C	3		140.6, C		137.4, C
4		147.4, C		141.0, C	4	6.35, s	111.9, CH	6.27, d (1.5)	110.3, CH
5	6.63, s	115.6, CH	6.68, s	118.0, CH	5		158.9, C		157.2, C
6		157.2, C		150.7, C	6	6.14, s	103.2, CH	6.22, d (1.5)	102.7, CH
7		121.7, C		123.1, C	7	2.18, s	21.5, CH ₃	2.20, s	19.5, CH ₃
8		200.1, C		201.2, C	8		167.7, C		167.1, C
9		71.8, C		71.6, C	9	3.67, s	52.2, CH ₃	3.69, s	51.9, CH ₃
10		196.9, C		196.6, C	10	-	-		169.2, C
11		138.4, C		138.3, C	1'		155.8, C		154.5, C
12	6.63, d (5.5)	141.9, CH	6.62, d (5.5)	141.8, CH	2'		117.1, C		116.9, C
13	2.61, m	32.4, CH ₂	2.66, m	32.3, CH ₂	3'		138.9, C		138.6, C
14	4.39, dd (9.2, 5.9)	67.4, CH	4.34, dd (9.5, 5.0)	67.8, CH	4'	6.41, s	112.8, CH	6.47, d (2.0)	112.9, CH
15	2.72, m	27.3, CH ₂	2.72, m	27.3, CH ₂	5'		159.7, C		159.4, C
16	5.03, t (6.8)	121.3, CH	5.06, t (7.0)	121.3, CH	6'	6.14, s	103.9, CH	6.16, d (2.0)	104.0, CH
17		132.2, C		132.3, C	7'	2.20, s	19.9, CH ₃	2.22, s	19.5, CH ₃
18	1.54, s	17.5, CH ₃	1.55, s	17.5, CH ₃	5OH	9.95, br.s		10.02, br.s	
19	1.65, s	25.5, CH ₃	1.66, s	25.5, CH ₃	5'OH	9.50, br.s		9.93, br.s	
20	2.31, s	21.7, CH ₃	2.22, s	16.2, CH ₃	NH ₂			7.45, 7.35, br.s	
21			3.72, s	60.6, CH ₃					

populations were calculated according to Boltzmann distribution law. The conformers with Boltzmann-population of over 1% were chosen for ECD calculations using Gaussian 09 (Frisch et al., 2009) software, and the stable conformers were initially optimized at the B3LYP/6-311G(d,p) in methanol using the IEFPCM model. Vibrational frequency analysis confirmed the stable structures. Under the same condition, the ECD calculation was conducted using Time-dependent Density functional theory (TD-DFT). Rotatory strengths for a total of 30 excited states were calculated. The ECD spectrum was simulated in SpecDis (University of Würzburg) with a half-bandwidth of 0.3–0.4 eV, according to the Boltzmann-calculated contribution of each conformer after UV correction.

NMR Calculation

The two stereoisomers **1e** and **1f** were delivered to geometry optimization at B3LYP/6-31+G(d,p) in gas phase. The theoretical calculation of NMR was conducted using the

Gauge-Independent Atomic Orbitals (GIAO) method at mPW1PW91/6-311+G(2d,p) in methanol by the IEFPCM model. Finally, the TMS-corrected NMR chemical shift values were averaged according to Boltzmann distribution for each conformer and fitting to the experimental values by linear regression. The calculated ^{13}C -NMR chemical shift values of TMS in methanol were 0 ppm.

Cytotoxicity Assay

The cytotoxic activities of **1-17** were screened against the growth panel of 10 tumor cell lines (K562, MCF-7, A549, Huh-7, H1975, HeLa, HL7702, HL60, MOLT-4, and DU145) (Bergeron et al., 1994) (**Supplementary Material**).

COX-2 Inhibitory Activity Assay

According to the manufacturer's instructions. The test compounds were dissolved in DMSO and the final concentration was set as 10 μM . The percentage inhibition has been calculated

TABLE 2 | ^1H , ^{13}C NMR data of **5-7** (500/125 MHz, in CDCl_3 , δ ppm, J in Hz).

No.	5		No.	6		No.	7	
	δ_{H}	δ_{C} , type		δ_{H}	δ_{C} , type		δ_{H}	δ_{C} , type
1		161.7, C	1		126.5, C	1		154.3, C
2	6.71, d (8.0)	110.4, CH	2	6.71, s	116.9, CH	2	6.72, d (8.0)	114.9, CH
3	7.49, t (8.0)	136.7, CH	3		146.9, C	3	7.04, td (8.0, 2.0)	127.4, CH
4	6.80, d (8.0)	106.5, CH	4		131.2, C	4	6.77, t (7.5)	118.6, CH
4a		155.4, C	4a		141.6, C	5	7.29, dd (7.5, 1.0)	128.7, CH
5	7.18, s	119.5, CH	5a		164.7, C	6		129.8, C
6		142.4, C	6		97.3, C	7	5.01, d (3.5)	67.5, CH
7		153.5, C	7	6.36, d (1.5)	101.0, CH	8	3.32, dd (6.0, 3.5)	78.0, CH
8		133.9, C	8		165.5, C	9	3.58, dq (6.0, 6.0)	66.8, CH
8a		117.8, C	9	7.21, d (1.5)	104.5, CH	10	1.11, d (6.0)	19.7, CH_3
9		184.2, C	9a		138.8, C	6-OH	9.33, br.s	
9a		108.9, C	9b		109.4, C			
10a		153.8, C	10	2.60, s	24.8, CH_3			
11	4.99, s	56.7, CH_2	11		164.2, C			
12	2.40, s	17.2, CH_3	COOH	11.84, s				
13	3.81, s	62.6, CH_3						
1-OH	12.53, s							
11-OH	4.37, br.s							

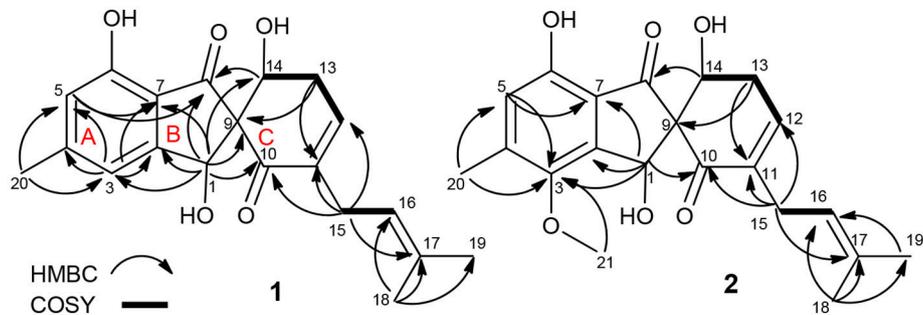


FIGURE 2 | Key COSY and HMBC correlations of **1** and **2**.

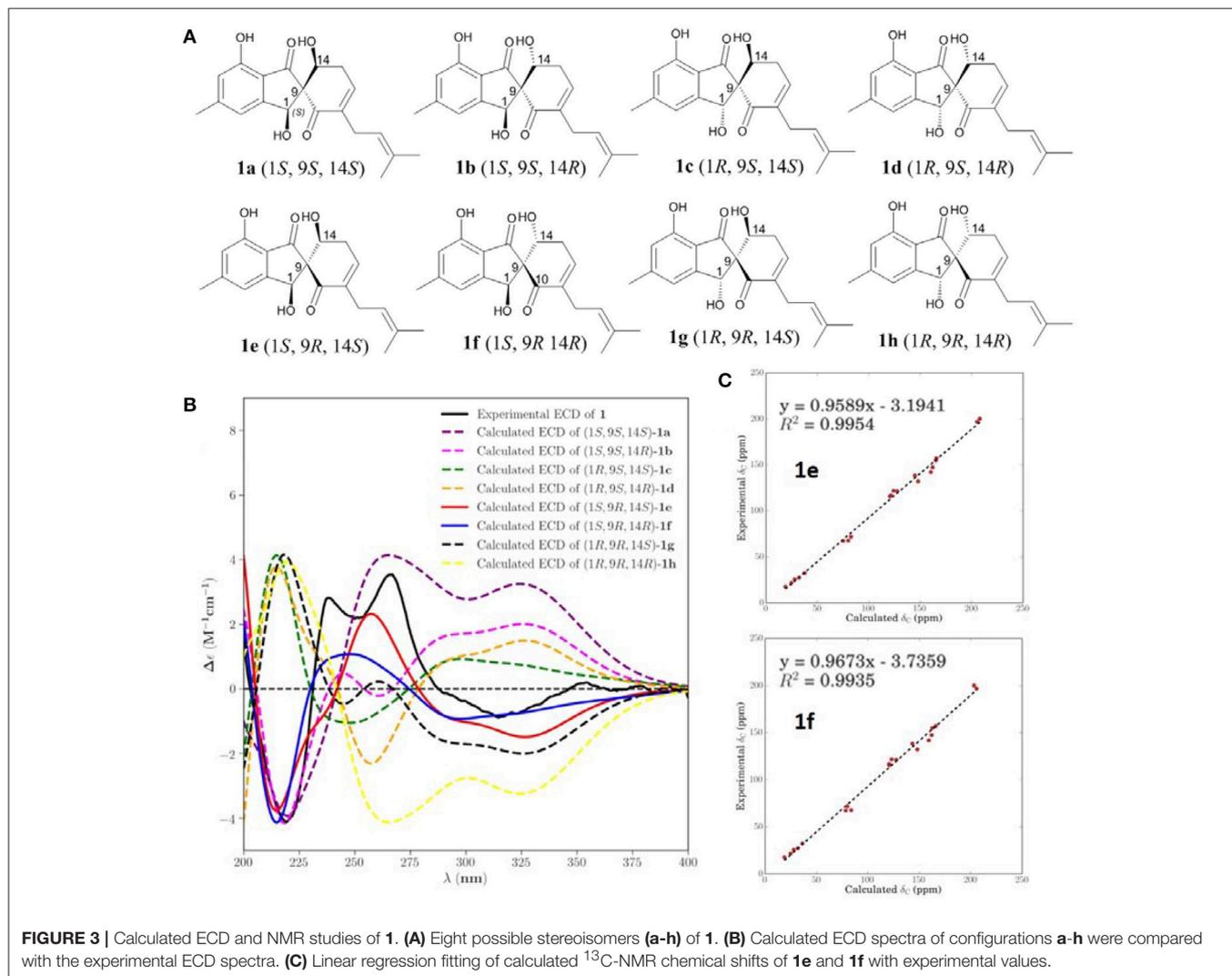
by comparison with control incubations (Tian et al., 2015b) (**Supplementary Material**).

RESULTS AND DISCUSSION

Compound **1** was assigned a molecular formula of $C_{20}H_{22}O_5$ (10 degrees of unsaturation) by HRESIMS (m/z 343.1541 [$M + H$]⁺). Its NMR spectra showed resonances for three methyls, two methylenes, two oxygenated sp^3 methine, one sp^3 quaternary carbon, four sp^2 methine, six sp^2 quaternary carbon, and two ketocarbonyl carbons (δ_C 200.1 and 196.9) (**Table 1**). These data suggested a tricyclic skeleton of **1**. The 1H NMR spectrum exhibited aromatic signals at δ_H 6.84 (s, H-3), 6.63 (s, H-5) indicating the presence of a tetrasubstituted aromatic ring. The HMBC correlations of H-3/C-2, C-4, C-5, C-7, and C-20, H₃-20 (δ_H 2.31, s)/C-4, and C-5, and H-5/C-6, and C-7 suggested a methyl (C-20) and a hydroxy at C-4 and C-6, respectively. The HMBC correlations of H-1 (δ_H 5.50, s)/C-2, C-3, C-4, C-6, C-7, C-8 (δ_C 200.1) and C-9 (δ_C 71.8), and H-5/C-8 indicated linkage of ring B and the connection of C-1 to C-2 and C-7 to C-8.

The COSY cross-peaks of H-13 (δ_H 2.61, m) /H-14 (δ_H 4.39, dd, $J = 9.2, 5.9$ Hz), and H-12 (δ_H 6.33, d, $J = 5.9$ Hz) delineated the spin system C₁₂-C₁₃-C₁₄. Moreover, the HMBC correlations of H-12/C-10 and C-14, H-13/C-9, C-11, and C-12, and H-14/C-1, C-8 and C-9 indicated the presence of a α,β -unsaturated ketone (ring C) and rings B/C are connected by C-9. Finally, a 2-methyl-2 butene group was attached to C-11 (δ_C 138.4) by the evidence of the HMBC correlations of H₃-18 (δ_H 1.54, s)/C-16 (δ_C 121.3), C-17 (δ_C 132.2), and C-19 and H₂-15 (δ_H 2.72, s)/C-10, C-11, C-12, C-13, C-16, and C-17.

NOESY correlations and Mosher method were failure to determine the configurations of **1**, so theoretical calculations were used to solve it. There are eight possible stereoisomers (**a-h**) of **1**, as shown in **Figure 3A**. Computational studies of electron circular dichroism (ECD) were carried out. All stereoisomers (**a-h**) were selected for theoretical calculations using time dependent density functional theory (TDDFT) B3LYP/6-311G (d,p) level with the IEFPCM model in MeOH (**Tables S1, 2, 4**). A comparison of the experimental spectrum of **1** with the calculated ECD spectra of eight possible stereoisomers (**a-h**) were presented



in **Figure 3B**. The measured ECD curve exhibits two negative Cotton effects (CEs) at 219 and 315 nm, and two positive cotton effects at 238 and 267 nm, matching well with the calculated ECD curve of **1e** (1*S*, 9*R*, 14*S*) and **1f** (1*S*, 9*R*, 14*R*). Then, computed ^{13}C -NMR chemical shifts was carried out to define the stereochemistry of C-14. Computed ^{13}C -NMR chemical shifts of each conformer were first Boltzmann-weighted averaged, and then fitted to experimental values by Ordinary Least Squares (OLS) Linear Regression method in order to remove systematic error that results from the conformational search and random error from experimental conditions (**Tables S3**, **5**). As a result, the computed chemical shift for C-14 of **1e** is $\delta_{\text{C}} = 68.7$ ppm, with only a deviation of 1.3 ppm from the experimental value ($\delta_{\text{C}} = 67.4$ ppm) (**Table S8**). All in all, the computed chemical shifts of **1e** showed good agreement with the experimental values and has the higher R^2 and R^2_{adj} values than that of **1f**, which suggesting that **1e** (1*S*, 9*R*, 14*S*) be the true isomer of **1** (**Figure 3C**, **Table S7**).

The molecular formula of **2** was determined as $\text{C}_{21}\text{H}_{24}\text{O}_6$ by its HRESIMS (m/z 373.1638 $[\text{M} + \text{H}]^+$), corresponding to 10 units of unsaturation. Its UV and NMR date were similar to those of **1**, except for the presence of a methoxy (δ_{H} 3.72, δ_{C} 60.6) in **2** (**Table 1**). The extra methoxy (C-21) was located at C-3 by HMBC correlations from H_3 -21 to C-3 (δ_{C} 148.5) (**Figure 2**). The absolute configuration of **2** was suggested as (1*S*,9*R*,14*S*), as its chemical shift, coupling constant, optical

rotation and CD effect (**Table 1**, **Figure S20**) almost the same as those of **1**.

The molecular formula of **3** was assigned as $\text{C}_{16}\text{H}_{16}\text{O}_5$ by its HRESIMS ion peak at m/z 289.1070 $[\text{M} + \text{H}]^+$ and NMR date. The ^1H NMR spectrum of **3** exhibited two hydroxyl proton at δ_{H} 9.95 and 9.50, five aromatic signals at δ_{H} 6.41, 6.35, 6.22, and 6.14×2 , one *O*-methyl at δ_{H} 3.67, and two single methyls at δ_{H} 2.18 and 2.20 (**Table 2**). Analysis of the ^{13}C and DEPT-135 NMR spectra of **3** indicated the presence of 16 carbons, including one carbonyl carbon (δ_{C} 167.7), 12 aromatic carbons (four oxygenated ones at δ_{C} 159.7, 158.9, 158.0 and 155.8), one methoxy and two methyls. These spectra of **3** were similar to those of diorcinol (Tian et al., 2015b), except for the presence of a methyle formate group (one carbonyl carbon at δ_{C} 167.7 and one methoxy at δ_{H} 3.67/ δ_{C} 52.2). However, relying solely on the NMR date, the location of methyle formate was more difficult to determine. In order to determine location of methyle formate of **3**, a single-crystal X-ray diffraction pattern was obtained using the anomalous scattering of Cu $K\alpha$ radiation shows an ORTEP drawing (**Figure 4**, **Table S9**) and unambiguously determined methyle formate at C-2'. Thus, the structure of **3** was determined, and named as diorcinol L.

Compound **4** was obtained as brown powder. The molecular formula of was determined as $\text{C}_{17}\text{H}_{17}\text{NO}_6$ by its HRESIMS (m/z 332.1131 $[\text{M} + \text{H}]^+$), which corresponded to ten units of unsaturation. The ^1H and ^{13}C NMR data of **4** were similar to those of **3**, except for the presence of one amide [δ_{C} 169.2 (CONH_2)/ δ_{H} 7.45, 7.35 (CONH_2)] (**Table 1**). The extra amide was determined at C-2 by the HMBC correlations of H-4, H-6/C-2, as well as H_3 -7/C-2, C-3, C-4, and C-10 (**Figure 5**). Hence, the structure of was elucidated and the trivial name diorcinol M was assigned.

Compound **5** was isolated as pale yellow needle-like crystals. Its molecular formula was determined to be $\text{C}_{16}\text{H}_{14}\text{O}_5$, by HR-ESI-MS, indicating 10 degrees of unsaturation. The 1D NMR data (**Table 3**) of **5** contained resonances for one carbonyl carbon, eight sp^2 quaternary carbons, four sp^2 methines, one sp^3 methylene, two sp^3 methyls. Comparison of UV-vis and NMR data with those of 1-Hydroxy-6-methyl-8-hydroxy-methylxanthone (**8**) revealed a high degree of similarity skeleton, where the only obvious differences is in the presence of one methoxy (δ_{H} 3.81; δ_{C} 62.6) and low-field chemical shift of C-7 (from δ_{C} 127.2 in **8** to 153.5 in **1**). Meanwhile, this difference can also be ascribed by the HMBC correlations (**Figure 5**) of δ_{H} (3.81,

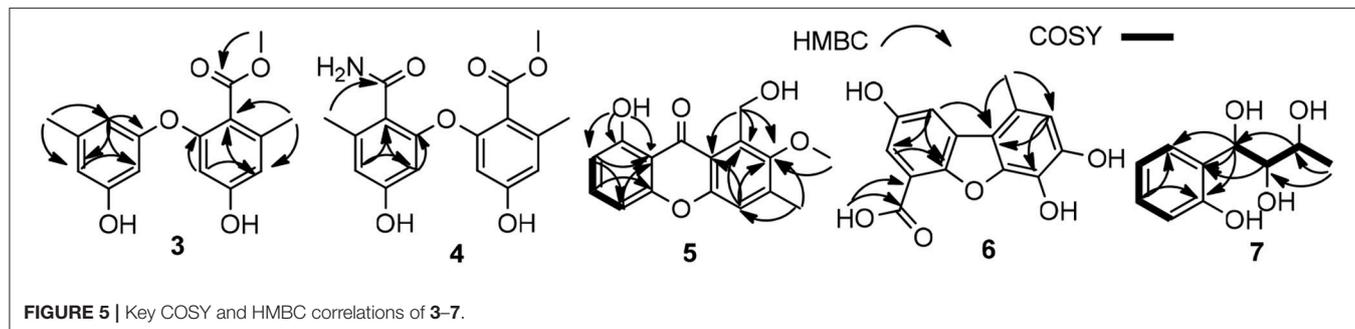
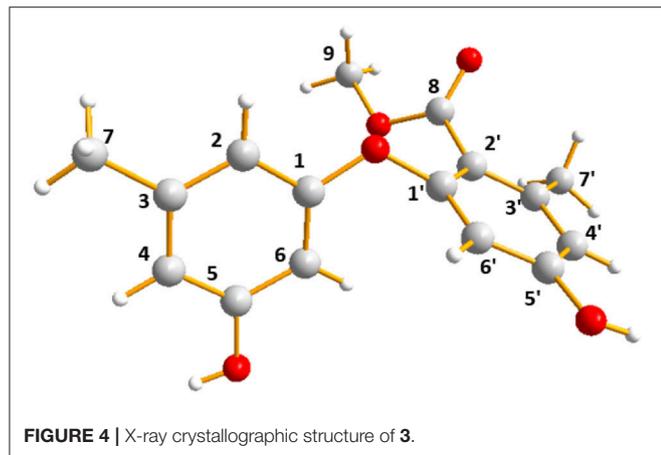


TABLE 3 | Cytotoxic results of the compounds (IC₅₀, μ M).

Comp.	K562	MCF-7	A549	Huh-7	H1975	HeLa	HL7702	HL60	MOLT-4	DU145
3	43.5	10.5	17.7	5.7	15.3	7.1	68.2	9.6	NA	9.1
8	NA	141.0	128.0	122.0	NA	14.3	33.8	NA	NA	NA
TSA	0.1	0.7	0.3	0.08	0.09	0.08	0.09	0.04	0.03	0.04

NA, No active (IC₅₀ > 200 μ M).

s, H-13) with δ_C (153.5, C-7) and requirements of HRESIMS spectrum. Thus, the structure of **5** was determined and named as 1-hydroxy-6-methyl-11-methoxy-8-hydroxymethylxanthone.

Compound **6** showed a molecular ion peak at m/z 275.0544 [M + H]⁺ in the HR-ESI-MS, in accordance with the molecular formula C₁₄H₁₀O₆, which was also supported by NMR data. The ¹H NMR data displayed one sharp hydroxy proton singlet (11.84, s, COOH), two aromatic protons at δ_H 7.21 (1H, d, J = 1.5 Hz), 6.71 (1H, s), 6.36 (1H, d, J = 1.5 Hz), and a singlet methyl at δ_H 2.60 (3H, s). The ¹³C and DEPT-135 NMR spectra (Table 2) showed 14 carbons, including one carboxyl group (δ_C 164.2), three aromatic methine carbons (δ_C 116.9, 104.5, 101.0), 11 aromatic quaternary carbons (δ_C 165.5, 164.7, 146.9, 141.6, 138.8, 131.2, 126.5, 109.4, 97.3), and one methyl (δ_C 24.8). The aforementioned NMR data showed **1** was closely related structurally to the porric acid C (Carotenuto et al., 1998). The only difference was the substituent of C-4, the chemical shift to low field of C-4 (δ_C 131.2 in **4**; δ_C 101.0 in porric acid C), the HMBC correlation (Figure 5, Table 2) from H-2 (δ_H 6.71, s) to C-3 (δ_C 146.9) and C-4 (δ_C 131.2) requirements of HRESIMS spectrum suggested that a hydroxy proton group was connected to C-4. Thus, the structure of **6** was determined and named porric acid E.

Compound **7** was obtained as a yellow oil. Its HRESIMS gave the molecular formula C₁₀H₁₄O₄, requiring four degrees of unsaturation. The ¹H NMR spectrum (Table 2) exhibited aromatic signals at δ_H 7.29 (dd, J = 7.5, 1.0 Hz, H-5), 7.04 (td, J = 8.0, 2.0 Hz, H-3), 6.77 (t, J = 8.0, Hz, H-4), and 6.72 (d, J = 8.0 Hz, H-2), indicating the presence of a disubstituted aromatic ring. The substitutions of a trihydroxybutyl group (C-7—C-10) and a hydroxy (δ_H 9.33, br.s, OH-6) at C-1 (δ_C 154.3) and C-6 (δ_C 129.8), were assigned by the COSY and HMBC interactions (Figure 5, Table 2).

Biological Activity

Cytotoxic activities of the isolated compounds **1–17** and trichostatin A (TSA, positive control) against 10 tumor cell lines (K562, MCF-7, A549, Huh-7, H1975, HeLa, HL7702, HL60, MOLT-4, and DU145) were tested. Among all of them, **3** displayed a wide range of cytotoxicities with IC₅₀ values in the range of 5.7–68.2 μ M, and **8** showed weak selective cytotoxicities against HeLa and HL7702 cells (Table 3). Furthermore, compound **6** displayed COX-2 inhibitory activity with the prominent IC₅₀ values of 3.3 μ M. Celecoxib was used as the positive control with IC₅₀ values of 0.01 μ M.

CONCLUSION

In conclusion, 17 phenol derivatives were isolated from the EtOAc extract of a marine sponge-derived fungus *Didymellaceae* sp. SCSIO F46. Coleophomones E and F (**1** and **2**) possess unprecedented spiro [cyclohexane-1,2'-inden] moiety, which is relatively seldom in natural products. Other new compounds **3–7** represent common types of phenol derivatives, which are widely found in fungal metabolites. Amongst, compounds **3** and **8** displayed a wide range of cytotoxicities against several tumor cell lines. In addition, **6** displayed significant COX-2 inhibitory activity with the prominent IC₅₀ value of 3.3 μ M.

AUTHOR CONTRIBUTIONS

YT designed the experiments and performed the isolation and characterization of all the compounds and wrote the manuscript. XL performed the isolation and purification of the fungal strain. XZ designed the research work and revised the manuscript. YL contributed in project design and manuscript preparation. All authors reviewed the manuscript and approved for submission.

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SUPPLEMENTARY MATERIAL

The Supplementary Material for this article can be found online at: <https://www.frontiersin.org/articles/10.3389/fchem.2018.00536/full#supplementary-material>

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Conflict of Interest Statement: The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

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