

## New Sesquiterpenoids From Plant-Associated *Irpex lacteus*

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Bacteria produce a large number of virulence factors through the quorum sensing (QS) mechanism. Inhibiting such QS system of the pathogens without disturbing their growth is a potential strategy to control multi-drug-resistant pathogens. To accomplish this, two new tremulane-type sesquiterpenoids, irpexolaceus H (1) and I (2), along with two known furan compounds, irpexlacte B (3) and C (4), were isolated from *Orychophragmus violaceus* (L.) OE Schulz endophytic fungus *Irpex lacteus* (Fr.) Fr. Their structures were elucidated by detailed spectroscopic data (NMR, HRESIMS, IR, and UV), single-crystal X-ray diffraction, and electronic circular dichroism (ECD) analysis. Furthermore, those compounds were evaluated for anti-quorum sensing (anti-QS) activity, and compound **3** was found contributing to the potential QS inhibitory activity.

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### INTRODUCTION

Bacterial quorum sensing (QS) is a cell-density-dependent communication process by which cells measure population density and trigger appropriate responses and conduct behavioral regulation, such as luminescence, motility, secretion of virulence factors, and formation of biofilms (Papenfort and Bassler, 2016). QS inhibitor (QSI) inhibits the QS system without affecting bacteria's growth and reduces its virulence production and biofilm formation; thus, the bacteria are in a low or non-toxicity state, the growth is not inhibited, and therefore it is difficult to cause drug resistance (Jiang and Li, 2013). However, the purpose of traditional antibacterial agents is to kill or inhibit the growth of bacteria, and it is difficult to avoid bacterial resistance (Kalia, 2013). Therefore, finding new QSIs to replace traditional antibacterial agents has become a new strategy in the antibacterial field.

*Irpex lacteus* (Fr.) Fr. (Phanerochaetaceae) is a basidiomycete that usually colonizes on the deadwood white (i.e., white rot), and it is often used as a traditional Chinese medicine for the treatment of chronic glomerulonephritis. In our previous study on this subject, seven sesquiterpenoids, irpexolaceus A–G, and two new furan derivatives, irpexonjust A–B, were isolated from *I. lacteus* OV38 (Luo et al., 2022). Furthermore, two new tremulane-type sesquiterpenoids, irpexolaceus H (1) and I (2), were isolated in this study from fungi, as well as two furan compounds, such as irpexlacte B (3) and C (4), were also obtained (Figure 1). These compounds were screened for QS inhibitory activity, and compound 3 exhibited the highest QS inhibitory activity among them. The details of the isolation, structure assignment, and QS inhibitory activities of 1-4 are presented.

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#### MATERIALS AND METHODS

#### **General Experimental Procedures**

Nuclear magnetic resonance (NMR) was obtained on a Bruker AV-400 spectrometer (Bruker Corporation, Karlsruhe, Germany). HRMSESI data were recorded on a Q-Exactive Orbitrap MS system (Thermo Fisher Scientific, Bremen, Germany). UV data were obtained on an Evolution 220 UVvis spectrophotometer (Thermo Fisher Scientific, Madison, United States). Infrared spectroscopy (IR) spectra were obtained on a Nicolet<sup>™</sup> iS10 FTIR spectrometer (Thermo Fisher Scientific, Madison, United States). Optical rotations were recorded on an Autopol VI automatic polarimeter (PerkinElmer, Waltham, MA, United States). The silica gel (100-200 and 200-300 mesh, Qingdao Marine Chemical Factory, Qingdao, China) and Sephadex LH-20 column (GE Healthcare Bio-Sciences AB, Uppsala, Sweden) were used for open column chromatography (CC). Fractions were monitored by TLC (HSGF 254, Yantai Jiangyou Silica Gel Development Co., Yantai, China), and spots were visualized by heating silica gel plates after soaking in methanol supplemented with 10% H<sub>2</sub>SO<sub>4</sub>. The preparative HPLC was performed with UltiMate 300 HPLC (Thermo Fisher Scientific, Madison, United States) equipped with a YMC-Pack ODS-A column (250  $\times$  10 mmI.D, S-5  $\mu$ m, 12 nm, and flow speed = 2-3 ml/min, YMC Co., Ltd., Kyoto, Japan).

## Fungal Material and Fermentation Conditions

The strain *Irpex lacteus* was isolated from the healthy flowers and stems of *Orychophragmus violaceus* (L.) OE Schulz collected from Nanjing University of Science and Technology (NJUST) (Luo et al., 2021). The fungus *I. lacteus* was fermented in 250-ml Erlenmeyer flasks containing 100 ml Fungus No. 2 medium (2% sorbitol, 2% maltose, 1% glutamine, 1% glucose, 0.3% yeast extract, 0.05% tryptophan, 0.05% KH<sub>2</sub>PO<sub>4</sub>, and 0.03% MgSO<sub>4</sub>, pH 6.4) at 28°C and 140 rpm for 15 d (Zhou et al., 2017).

# Extraction and Purification of Secondary Metabolites

The supernatant (total 100 L) of I. lacteus fermentation was extracted three times by equal ethyl acetate (EtOAc) and concentrated under reduced pressure to give a crude extract (99.49 g), which was subjected to column chromatography (CC) over a silica gel (100-200 mesh) eluted with a gradient of dichloromethane/methyl alcohol (CH<sub>2</sub>Cl<sub>2</sub>/MeOH, 100:0-0:100) to obtain a fraction G (Fr. G) and other 13 fractions (Fr. A-F and H-M). Fr. A (19.61 g) was purified by CC over a silica gel (200-300 mesh) eluted with the gradient systems of petroleum ether/EtOAc (20:1-1:1) and EtOAc/ MeOH (100:1-10:1) to yield six sub-fractions Fr. A1-A6. Fr. A6 (4.79 g) was subjected to CC over a silica gel eluted with a gradient system of CH<sub>2</sub>Cl<sub>2</sub>/MeOH (80:1-0:100) to yield five sub-fractions, that is, Fr. A6.1-A6.5. Fr. A6.1 (97.4 mg) was further purified by HPLC to give 3 (40.1 mg, 45% MeOH, flow speed = 2.5 ml/min, and  $t_R =$ 31.0 min) and 4 (5.0 mg, 45% MeOH, flow speed = 2.5 ml/min, and  $t_{R} = min$ ). Fr. G (37.30 g) was continually separated by CC over a silica gel (200-300 mesh) eluted with a gradient system of CH<sub>2</sub>Cl<sub>2</sub>/ MeOH (100:0-0:100) to give nine sub-fractions Fr. G1-G9. Fr. G2 (7.76 g) was repeatedly separated using CC over a silica gel, Sephadex LH-20 (100% MeOH), and finally purified by HPLC to give 1 (30.3 mg, 50% MeOH, flow speed = 3.0 ml/min, and  $t_{\text{R}}$  = 14.1 min) and 2 (4.90 mg, 50% MeOH, flow speed = 3.0 ml/min, and  $t_{\rm R} = 25.6$  min).

Irpexolaceus H (1). white amorphous powder; [α]18 D + 17.2 (*c* 0.19, MeOH); UV (MeOH)  $\lambda_{max}$  (log  $\varepsilon$ ) 203 (3.99) nm and 290 (2.65) nm; IR (KBr):  $v_{max}$  3380.60, 2932.72, 2872.93, 1752.01, and 1027.87 cm<sup>-1</sup>; <sup>1</sup>H NMR (400 MHz in MeOD) and <sup>13</sup>C NMR (100 MHz in MeOD) data are shown in **Table 1**; and HRESIMS *m*/*z* 311.1502 [M + COOH]<sup>-</sup> (calcd for C<sub>16</sub>H<sub>23</sub>O<sub>6</sub>, 311.1500).

Irpexolaceus I (**2**). Yellow oil; [α]18 D + 21.6 (*c* 0.12, MeOH); UV (MeOH)  $\lambda_{max}$  (log  $\varepsilon$ ) 205 (3.57) nm; CD (MeOH) 195 (Δ $\varepsilon$ -4.72) and 215 (Δ $\varepsilon$  + 8.61) nm; IR (KBr):  $\upsilon_{max}$  3359.87, 1564.95, 1397.66, and 1020.64 cm<sup>-1</sup>; <sup>1</sup>H NMR (400 MHz in MeOD) and <sup>13</sup>C NMR (100 MHz in MeOD) data are shown in **Table 1**; and HRESIMS *m/z*, 283.1551 [M-H]<sup>-</sup> (calcd for C<sub>15</sub>H<sub>23</sub>O<sub>5</sub> and 283.1551).

Νο	1		2	
	$\delta_{\rm C}$ , type	δ <sub>H</sub> (J in Hz)	$\delta_{\rm C}$ , type	δ <sub>H</sub> ( <i>J</i> in Hz)
1	126.8, C	_	148.0, C	-
2	139.9, C	-	134.1, C	-
3	38.7, CH	3.62, br d (14.4)	42.1, CH	3.31, m
4	30.9, CH <sub>2</sub>	1.99, m and 1.84, m	40.2, CH <sub>2</sub>	2.27, dd (14.0 and 7.3) and 2.11, br d (15.6)
5	73.3, CH	3.96, t (5.0)	181.5, C	
6	39.7, CH	1.87, m	143.6, CH	5.81, m
7	40.8, CH	3.39, m	46.3, CH	3.43, m
8	40.4, CH <sub>2</sub>	1.72, m; 1.40, dd (12.9 and 11.0)	43.8, CH <sub>2</sub>	1.93, ddd (13.2, 8.7, and 1.4) and 1.31, dd (13.2 and 7.4)
9	44.9, C	_	44.0, C	
10	42.1, CH <sub>2</sub>	2.17, d (16.2) and 1.81, m	41.6, CH <sub>2</sub>	2.57, br d (15.6) and 2.35, dd (14.0 and 7.3)
11	70.8, CH <sub>2</sub>	4.82, d (13.0) and 4.73, d (13.0)	60.2, CH <sub>2</sub>	4.10, d (11.6) and 4.00, d (11.6)
12	182.3, C	_	66.4, CH <sub>2</sub>	3.60, dd (10.8 and 6.6); 3.50, m
13	11.8, CH <sub>3</sub>	0.90, d (7.0)	114.0, CH <sub>2</sub>	5.05, dt (17.1 and 1.48) and 4.94, br d (10.2)
14	23.9, CH <sub>3</sub>	1.11, s	24.5, CH <sub>3</sub>	1.07, s
15	68.6, CH <sub>2</sub>	3.21, dd (12.2 and 11.4)	69.5, CH <sub>2</sub>	3.26, d (11.0) and 3.21, d (11.0)



#### **Crystallographic Data of Compound 1**

Crystal data for *Cu* **1**\_0*m*. C<sub>15</sub>H<sub>22</sub>O<sub>4</sub>, *M* = 266.33, *a* = 6.6684 (4) Å, *b* = 12.1018 (5) Å, *c* = 8.4219 (4) Å,  $\alpha = 90^{\circ}$ ,  $\beta = 93.164$  (4)°,  $\gamma = 90^{\circ}$ , *V* = 678.61 (6) Å<sup>3</sup>, *Z* = 2, *T* = 170.0 K, space group *P*2<sub>1</sub>2<sub>1</sub>2<sub>1</sub>,  $\mu$ (Cu K $\alpha$ ) = 0.760 mm<sup>-1</sup>, *Dcalc* = 1.303 g/cm<sup>3</sup>, 10,065 reflections measured (10.520° ≤ 2 $\theta$  ≤ 127.318°), and 2210 unique ( $R_{int} = 0.0504$ ,  $R_{sigma} = 0.0357$ ) which were used in all calculations. The final  $R_1$  was 0.0321 [*I* > 2 $\sigma$ (*I*)]. The final *w* $R_2$  was 0.0752 [*I* > 2 $\sigma$ (*I*)]. The final  $R_1$  was 0.0346 (all data). The final *w* $R_2$  was 0.0772 (all data). The goodness of fit on *F*<sup>2</sup> was 1.091. Flack parameter = -0.02 (12), which was determined using 1101 quotients [(I+)-(I-)]/[(I+)+(I-)]. CCDC: 2133065 (www.ccdc.cam.ac.uk).

## Electronic Circular Dichroism Calculation of Compound 2

The experiment was performed as previously described (Luo et al., 2022). The conformers with Boltzmann population (over 1%) were initially optimized at B3LYP/6–31+G. The ECD spectra of all conformers were conducted by time-dependent density functional theory (TD-DFT) methodology at  $\omega$ B97X/def2-TZVP, CAM-B3LYP/TZVP, and M062X/def2-TZVP using IEFPCM in MeOH (Neuhaus and Loesgen, 2020). ECD spectra were generated using the program SpecDis 1.71 from dipole-length rotational strength by applying Gaussian band shapes with sigma = 0.2 eV (Bruhn et al., 2013).

#### **Anti-Quorum Sensing Activity**

The efficacy of purified compounds on inhibiting QS mechanisms was screened using indicator organisms Chromobacterium violaceum CV026 and Agrobacterium tumefaciens A136. Overnight grown C. violaceum CV026 culture (1 ml) (OD<sub>600</sub>  $\approx$ 0.1) was added into 100 ml LB agar medium supplemented with kanamycin (20 µg/ml) and C6-HSL (5 µM), mixed, and poured into the plates (Kumar et al., 2021). After the medium was solidified,  $2 \mu l$  of samples (1-4) (50 mg/ml) were dropped onto the plates and then incubated at 28°C for 24 h, recording the violacein changes in color. The absence of violacein production in CV026 represents compounds that inhibit the QS system. Furthermore, QS inhibitory activity-screened A. tumefaciens A136 was determined as described earlier (Zhu et al., 2018). X-gal (50  $\mu$ g/ml) and C10-HSL (5  $\mu$ M) were added into the LB agar medium supplemented with A. tumefaciens A136. Each experiment was repeated three times.





### **RESULTS AND DISCUSSION**

#### Structure Elucidation

Irpexolaceus H (1), a white amorphous powder, was assigned the molecular formula of C15H24O4 with five degrees of unsaturation based on HRESIMS data at m/z 311.1502 [M + COOH]<sup>-</sup> (calcd for  $C_{16}H_{23}O_6$ , 311.1500). The <sup>1</sup>H and <sup>13</sup>C NMR spectra of 1 (Table 1) revealed 15 carbon resonances, including two methyls at  $\delta_{\rm C}$  11.8 and 23.9, three methylene carbons at  $\delta_{\rm C}$  30.9, 40.4, and 42.1, two oxygenated methylene carbons at  $\delta_{\rm C}$  70.8 and 68.6, three methine carbons at  $\delta_{\rm C}$  38.7, 39.7, and 40.8, one oxygenated methine carbon at  $\delta_{\rm C}$  73.3, three sp<sup>2</sup> quaternary carbons at  $\delta_{\rm C}$  126.8, 139.9, and 182.3, and one sp<sup>3</sup> quaternary carbon at  $\delta_{\rm C}$  44.9. These data demonstrated high similarity to irpexolaceus F (Luo et al., 2022), with the main difference being that the low field at  $\delta_{\rm C}$  139.9 and 126.8 (1) shifted downfield to  $\delta_{\rm C}$  165.9 and 128.5 in irpexolaceus F, indicating differences in the position of the carbon-carbon double bonds, which was confirmed by the HMBC correlations from H<sub>2</sub>-8 ( $\delta_{\rm H}$  1.72, 1.40) and H<sub>2</sub>-10 ( $\delta_{\rm H}$ 2.17, 1.81) to C-1 ( $\delta_{\rm C}$  126.8) and from H<sub>2</sub>-4 ( $\delta_{\rm H}$  1.99, 1.84), H<sub>2</sub>-10  $(\delta_{\rm H} 2.17, 1.81)$ , and H<sub>2</sub>-11  $(\delta_{\rm H} 4.82, 4.73)$  to C-2  $(\delta_{\rm C} 139.9)$ , in combination with the  ${}^{1}H{-}^{1}H$  COSY correlations of H-3 ( $\delta_{H}$ 3.62)/H<sub>2</sub>-4 ( $\delta_{\rm H}$  1.99, 1.84)/H-5 ( $\delta_{\rm H}$  3.96)/H-6 ( $\delta_{\rm H}$  1.87)/H-7  $(\delta_{\rm H} \ 3.39)/{\rm H_2}$ -8  $(\delta_{\rm H} \ 1.72, \ 1.40)$  (Figure 2). The mentioned data combined with previously reported data (Luo et al., 2022) determined the planar structure of 1. In addition, the relative configuration of 1 was determined by the NOESY correlations of H-3 ( $\delta_{\rm H}$  3.62)/H-6 ( $\delta_{\rm H}$  1.87)/H-7 ( $\delta_{\rm H}$  3.39) and Me-13 ( $\delta_{\rm H}$  0.90)/ H-5 ( $\delta_{\rm H}$  3.96)/Me-14 ( $\delta_{\rm H}$  1.11) (Figure 2). Combined with the single-crystal X-ray diffraction analysis (Figure 3), the absolute configuration of 1 was confirmed as 3S,5S,6S,7R,9S.

Irpexolaceus I (2), a yellow oil, was assigned the molecular formula of  $C_{15}H_{24}O_5$  with four degrees of unsaturation based on HRESIMS data at m/z 283.1551 [M-H]<sup>-</sup> (calcd for  $C_{15}H_{23}O_5$ , 283.1551). The <sup>1</sup>H and <sup>13</sup>C NMR spectra of **2** (**Table 1**) revealed 15 carbon resonances, including one methyl at  $\delta_C$  24.5, four



methylene carbons at  $\delta_{\rm C}$  40.2, 43.8, 41.6, and 114.0, three oxygenated methylene carbons at  $\delta_{\rm C}$  60.2, 66.4, and 69.5, three methine carbons at  $\delta_{\rm C}$  42.1, 143.6, and 46.3, three sp<sup>2</sup> quaternary carbons at  $\delta_{\rm C}$  148.0, 134.1, and 181.5, and one sp<sup>3</sup> quaternary carbon at  $\delta_{\rm C}$  44.0. There were two C=C and one carbonyl group, and the remaining one unsaturation was a ring, which was confirmed by the <sup>1</sup>H-<sup>1</sup>H COSY correlations of H<sub>2</sub>-8 ( $\delta_{\rm H}$  1.93, 1.31)/H-7 ( $\delta_{\rm H}$  3.43)/H-6 ( $\delta_{\rm H}$  5.81)/H<sub>2</sub>-13 ( $\delta_{\rm H}$  5.05, 4.94), and the HMBC correlations from H-7 ( $\delta_{\rm H}$  3.43) to C-1 ( $\delta_{\rm C}$  148.0) and C-10 ( $\delta_{\rm C}$  41.6), from H<sub>2</sub>-8 ( $\delta_{\rm H}$  1.93, 1.31) to C-1 ( $\delta_{\rm C}$  148.0), C-7 ( $\delta_{\rm C}$ 46.3), C-9 ( $\delta_{\rm C}$  44.0), and C-10 ( $\delta_{\rm C}$  41.6), and from H<sub>2</sub>-10 ( $\delta_{\rm H}$  2.57, 2.35) to C-1 ( $\delta_{\rm C}$  148.0), C-7 ( $\delta_{\rm C}$  46.3), C-8 ( $\delta_{\rm C}$  43.8), and C-9 ( $\delta_{\rm C}$ 44.0). Moreover, a high-field quaternary carbon, C-9 ( $\delta_{\rm C}$  44.0), was linked to C-14 ( $\delta_{\rm C}$  24.5) and C-15 ( $\delta_{\rm C}$  69.5). Combined with another set of <sup>1</sup>H–<sup>1</sup>H COSY correlations of H-3 ( $\delta_{\rm H}$  3.31)/H<sub>2</sub>-4  $(\delta_{\rm H} 2.27, 2.11)/{\rm H_2}$ -12 ( $\delta_{\rm H} 3.60, 3.50$ ), the planar structure of **2** was determined as shown in Figure 1, which was highly similar to ceriponol P (Ying et al., 2014), with a difference of a hydroxyl group binding to  $\delta_{\rm C}$  69.5. In addition, the relative configuration of 2 was confirmed by the NOESY correlations of H-6 ( $\delta_{\rm H}$  5.81)/H-3  $(\delta_{\rm H} 3.31)/{\rm Me}$ -14  $(\delta_{\rm H} 1.07)$  (Figure 2). The calculated and experimental ECD spectra of 2 showed excellent fit (Figure 4), indicating that the absolute configuration of 2 was 2S,7S,9R.

In addition to the aforementioned compounds, two furans (3 and 4) were obtained and identified as irpexlacte B and C according to the previously reported data (Luo et al., 2022).

#### **Proposed Biotransformation Pathway**

Tremulane (i) and (ii) were biosynthesized based on the relative structures of 1 and 2 by the cyclization and rearrangement of farnesyl pyrophosphate (**FPP**) (Ayer and Browne, 1981; Ayer and Cruz, 1993). After a multi-step reaction such as esterification, cyclization, and dehydration in microbes, tremulane (i) was possibly converted to irpexlaceus H (**Figure 5**). Moreover, lactarane skeletons **ii-1** and **ii-2** were produced by a series of methyl migrations of tremulane (ii)) (Ayer and Cruz, 1993), which differed from tremulane (i)) in the configuration of C-1 and C-7. Lactarane skeleton **ii-2** was





transformed to 5,6-secotremulane under the 5,6-cleavage (He et al., 2020) and then possibly converted to irpexlaceus I by a series of oxidation and dehydration (**Figure 5**).

#### **Biological Activity**

Compounds 1–4 were evaluated for their QS inhibitory activities at 50 mg/ml against biomarker strains *C. violaceum* CV026 (Figure 6A) and *A. tumefaciens* A136 (Figure 6B). We assessed the inhibitory effect of the compound by assaying the inhibition of C6-HSL-induced violacein production by *C. violaceum* CV026, and C10-HSL-induced  $\beta$ -galactosidase expression (blue pigment) by *A. tumefaciens* A136. Compounds 1–2 showed no inhibitory activities against both QS systems of biomarker strains. However, compound 3, in which the binding position of hydroxyl was closer to the furan ring than 4, exhibited stronger inhibition activity against the production of violacein in *C. violaceum* CV026 than that of the latter but weaker inhibitory activity against the production of blue pigment in *A. tumefaciens* A136. The results demonstrated that the binding position of hydroxyl was vital for QS inhibitory activity.

## CONCLUSION

Two new tremulane-type sesquiterpenoids, irpexolaceus H (1) and I (2), were isolated from the liquid fermentation of *I. lacteus*. Their

structures were established based on NMR, HRESIMS, IR, singlecrystal X-ray diffraction, and ECD analysis. These compounds (1–4) were evaluated for QS inhibitory activities against *C. violaceum* CV026 and *A. tumefaciens* A136 at 50 mg/ml. The results found that compound **3** exhibited a significant QS inhibitory activity against *C. violaceum* CV026, and compound **4** showed a weaker activity. In addition, compound **3** also showed a weak QS inhibitory activity against *A. tumefaciens* A136. But interestingly, the hydroxyl binding to  $\alpha$ -C in the furan ring showed a stronger QS inhibitory activity than that of **4** (hydroxyl binding to  $\beta$ -C in the furan ring), which suggested that the position of hydroxyl in the furan ring was possibly vital for QS inhibitory activity against *C. violaceum* CV026.

## DATA AVAILABILITY STATEMENT

The original contributions presented in the study are publicly available. These data can be found at: https://www.ccdc.cam.ac. uk/, 2133065, 2095205, 2095201, and 2095200.

### AUTHOR CONTRIBUTIONS

H-ZL contributed to the chemical and biological experiments and prepared the manuscript draft. HJ contributed to the spectra data analysis. X-SH contributed to the ECD calculation and analysis. A-QJ contributed to the manuscript revision and financial support. All authors approved this manuscript to submit and publish.

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#### SUPPLEMENTARY MATERIAL

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