



Insights from Fumarole Gas Geochemistry on the Recent Volcanic Unrest of Pico do Fogo, Cape Verde

Gladys V. Melián^{1,2,3}*, Pedro A. Hernández^{1,2,3}, Nemesio M. Pérez^{1,2,3}, María Asensio-Ramos¹, Eleazar Padrón^{1,2,3}, Mar Alonso^{1,2}, Germán D. Padilla^{1,2}, José Barrancos^{1,2}, Francesco Sortino⁴, Hirochicka Sumino⁵, Fátima Rodríguez¹, Cecilia Amonte¹, Sonia Silva⁶, Nadir Cardoso⁶ and José M. Pereira⁷

¹Instituto Volcanológico de Canarias (INVOLCAN), La Laguna, Spain, ²Instituto Tecnológico y de Energías Renovables (ITER), Granadilla de Abona, Spain, ³Agencia Insular de la Energía de Tenerife (AIET), Granadilla de Abona, Spain, ⁴Istituto Nazionale di Geofisica e Vulcanologia - Sezione Roma 2, Roma, Italy, ⁵Department of General Systems Studies, Graduate School of Arts and Sciences, The University of Tokyo, Komaba, Meguro-ku, Japan, ⁶Universidade de Cabo Verde (UNICV), Praia, Cape Verde, ⁷Laboratório de Engenharia Civil of Cape Verde (LEC) Tira - Chapéu, Praia, Cape Verde

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> *Correspondence: Gladys V. Melián gladys@iter.es

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Melián GV, Hernández PA, Pérez NM, Asensio-Ramos M, Padrón E, Alonso M, Padilla GD, Barrancos J, Sortino F, Sumino H, Rodríguez F, Amonte C, Silva S, Cardoso N and Pereira JM (2021) Insights from Fumarole Gas Geochemistry on the Recent Volcanic Unrest of Pico do Fogo, Cape Verde. Front. Earth Sci. 9:631190. doi: 10.3389/feart.2021.631190 We report the results of the geochemical monitoring of the fumarolic discharges at the Pico do Fogo volcano in Cape Verde from 2007 to 2016. During this period Pico do Fogo experienced a volcanic eruption (November 23, 2014) that lasted 77 days, from a new vent \sim 2.5 km from the fumaroles. Two fumaroles were sampled, a low (F1 \sim 100°C) and a medium (F2~300°C) temperature. The variations observed in the δ^{18} O and δ^{2} H in F1 and F2 suggest different fluid source contributions and/or fractionation processes. Although no significant changes were observed in the outlet fumarole temperatures, two clear increases were observed in the vapor fraction of fumarolic discharges during the periods November 2008–2010 and 2013–2014. Also, two sharp peaks were observed in CO₂/CH₄ ratios at both fumaroles, in November 2008 and November 2013. This confirms that gases with a strong magmatic component rose towards the surface within the Pico do Fogo system during 2008 and 2013. Further, F2 showed two CO₂/S_{total} peaks, the first in late 2010 and the second after eruption onset, suggesting the occurrence of magmatic pulses into the volcanic system. Time series of He/CO₂, H₂/CO₂ and CO/CO₂ ratios are low in 2008–2009, and high in 2013–2014 period, supporting the hypothesis of fluid input from a deeper magmatic source. Regarding to the isotopic composition, increases in air-corrected ³He/⁴He ratios are observed in both fumaroles; F1 showed a peak in 2010 from a minimum in 2009 during the first magmatic reactivation onset and another in late 2013, while F2 displayed a slower rise to its maximum in late 2013. The suite of geochemical species analyzed have considerably different reactivities, hence these integrated geochemical time-series can be used to detect the timing of magmatic arrivals to the base of the system, and importantly, indicate the typical time lags between gas release periods at depth and their arrival at the surface. The high ³He/⁴He ratios in both fumaroles in the range observed for mid-ocean ridge basalts, indicating that He is predominantly of upper mantle origin. This work supports that monitoring of the chemical and isotopic composition of the fumaroles of the Pico do Fogo volcano is a very important

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tool to understand the processes that take place in the magmatic-hydrothermal system and to be able to predict future episodes of volcanic unrest and to mitigate volcanic risk.

Keywords: geochemistry, volcanic gases, fumarolic emission, precursory signals, Pico do Fogo volcano

INTRODUCTION

Volcanoes are the main degassing windows of the planet, even, in a state of quiescence, are able to degas continuously. Volcanic gases are emitted to the atmosphere both as diffuse degassing (Chiodini et al., 1996; Hernández et al., 2001; Pérez et al., 2011; Melián et al., 2014, 2019; Cardellini et al., 2017; Padrón et al., 2021) and visible emanations (e.g., fumaroles, plumes; Giggenbach, 1975; Fisher et al., 1996; Hilton et al., 2002; Melián et al., 2012; Gresse et al., 2018). The composition of the fluids discharged by volcanoes are a reflection of 1) deep processes such as the injection of new magma, degassing of deep mafic magma in the lower crust etc. and 2) secondary processes, as for example gas re-equilibrium, interaction with meteoric water-fed, water-rock interaction etc. (Giggenbach, 1980, 1984, 1987, 1993; Chiodini and Marini 1998; Taran and Giggenbach, 2003; Oppenheimer et al., 2012). Increases of the volatile content in magma plays a fundamental role in pre-eruptive pressurization of volcanic systems (Wallace, 2001). Indeed, dissolved gases in magma are the driving force behind eruptions (Symonds et al., 1994).

Direct in situ sampling of fumaroles has been the commonly used technique for determining the chemical and isotopic composition of volcanic gases (Giggenbach 1975), due to its reliability and the possibility to extensively investigate the main, minor, trace and ultra-trace compounds to isotopic investigation (e.g., Symond et al., 1994; Giggenbach, 1992, Hilton et al., 2002; Chiodini, 2009; Melián et al., 2014). The main disadvantage of this sampling method is that continuous real-time monitoring is not yet feasible. Nevertheless, it is a valuable geochemical tool for evaluating the volcanic activity of those systems where there is no real-time monitoring, or remote sensing observation, of visible volcanic gas emissions (Giggenbach, 1987; Fischer et al., 1996; Melián et al., 2012; Padrón et al., 2012; Caliro et al., 2015). Variations in the relative content of magmatic fluids from fumarole discharges provide us with valuable information about processes that are affecting the volcanic-hydrothermal system at depths (Giggenbach, 1987; Fischer et al., 1996; Taran et al., 1998; Hilton et al., 2010). Likewise, the study of magmatic-hydrothermal fluids is crucial to understand and compare the origin and evolution of volcanoes worldwide (Javoy et al., 1986; Hilton et al., 2002; Aiuppa et al., 2005a, 2005b; Zelenski and Taran, 2011; Chiodini et al., 2012; Melián et al., 2014). The early detection of changes in volcanic activity are fundamental when establishing the necessary criteria to activate early warning protocols for civil protection (Tilling, 1995).

Pico do Fogo (2829 m a.s.l.), which is located on the island of Fogo, is the only currently active stratovolcano in Cabo Verde archipelago. The summit crater of Pico do Fogo is characterized by a wide field of active fumaroles and high emission rates of volcanic gases (Dionis et al., 2015a, 2015b; Aiuppa et al., 2020). During recent years, several volcanic gas studies have been carried

out in Pico do Fogo volcano, mainly focused on the diffuse degassing of carbon dioxide (CO₂), hydrogen sulphide (H₂S), and on thermal energy release (Dionis et al., 2015a, 2015b). More recently, Alonso et al. (2021) estimated the diffuse helium (He) degassing and thermal energy from the summit crater of Pico do Fogo during 2007-2018 period observed. Alonso et al. (2021) observes significant increases in ³He/⁴He isotopic ratio, ³He, mantle ⁴He emission rates and thermal energy in 2010 and 2013, and drawsa relationship between magmatic intrusions 4 years and 1 year before the eruptive period 2014-2015. However, very few studies of Fogo volcano fumarolic geochemistry have been reported. Recently, Aiuppa et al. (2020) estimated the emission of volatiles from Pico do Fogo fumaroles by combining measurements of the fumarole composition (direct sample and portable Multi-GAS) and SO2 flux (near-vent UV Camera recording). Aiuppa et al. (2020) showed that the CO₂ emitted from fumarolic vents of Fogo volcano $(1,060 \pm 340 \text{ t} \cdot \text{d}^{-1})$ is larger than current diffuse CO_2 degassing (828 ± 5 t · d⁻¹; Dionis et al., 2015a) of Fogo Island, evidencing the continuous contribution of deep magmatic gases to the system.

This work presents the first study of the temporal evolution (2007–2016) of fumarole chemistry from Pico do Fogo volcano, and describes the observed changes in the chemical and isotopic composition fumarolic discharges as geochemical precursory signals of the effusive volcanic eruption that commenced on 23 November 2014.

GEOLOGICAL AND GEOGRAPHICAL BACKGROUND OF FOGO ISLAND

The Cape Verde islands are located about 800 km west of Senegal, at 14-17° latitude and 21-25° longitude. The archipelago consists of a volcanic chain of 10 major islands and eight minor islands (Figure 1A). The Cape Verde islands are located over Mesozoic oceanic crust and their origin is related to the existence of a mantle plume (Courtney and White, 1986). Volcanic activity is driven by the existence of a hot spot beneath the oceanic crust (Fonseca et al., 2003). Pico do Fogo volcano is the most active volcano of the Cabo Verde Island, with approximately 30 recorded volcanic eruptions since 1500 A.D. (Day et al., 1999). During the 20th century, two eruptions occurred, one in 1951 and the other in 1995, while the last and most recent eruption commenced in late 2014. One of the most striking features of Fogo island is the Cha das Caldeiras (Figure 1B), a caldera with a diameter of ~9 km, and surrounded by a wall (Bordeira) that reaches 1000 m in height at some points. The caldera opens towards the sea on its eastern side, caused by a massive flank collapse at ~73 ka (Day et al., 1999; Ramalho et al., 2015). Pico do Fogo volcano is located in the center of the caldera, which gave rise to several eruptions from its summit between 1500 and 1750 A.D. (Torres et al., 1997). Magmatism is dominated by silica-undersaturated alkaline melts of basanitic through phonolitic



FIGURE 1 | (A) Geographical location of Fogo Island in Cape Verde. (B) Shaded relief map of Fogo Island with the location of Cha das Caldeiras (white square). (C) Shaded relief map of Cha das Caldeiras showing Pico do Fogo volcano. The white star indicates the location of the main vent of the 2014–2015 eruption. The white circle indicates the location of the earthquake (M = 2.9) occurred on April 12, 2012 (Pérez et al., 2015). (D) Aerial view of Pico do Fogo crater with the location of F1 (black star) and F2 (red star) fumaroles.

composition with the occurrence of carbonatites (Jørgensen and Holm 2002; Hildner et al., 2011; Mourão et al., 2012; Klügel et al., 2020). Escrig et al. (2005) reported that erupted lavas span a compositional spectrum from basanites to tephrites and nephelinites, with the 1995 eruption producing cogenetic basanites and phonotephrites (Hildner et al., 2011). The petrological study of lavas during the 2014–2015 volcanic eruption, showed somewhat evolved an alkaline composition ranging from tephrites to phonotephrites (Mata et al., 2017).

Fogo Island generally only experiences minor seismic activity (Leva et al., 2019). The Volcanological Institute of Cape Verde (OVCV), within the framework of the MAKAVOL research project in collaboration with Spanish institutions (INVOLCAN, ITER and University of Granada) and Cape Verdean institutions (University of Cape Verde, Civil Protection Service and Civil Engineering Laboratory), installed a seismic network on the island in April 2012 (Carrion et al., 2012). This seismic network was operational for 2 months and registered a significant seismic event of magnitude 2.9 on April 12, 2012 (Figure 1C; Pérez et al., 2015). The epicenter was located within Cha das Caldeira and was felt by the inhabitants of the island. Seismic activity is also recorded by the Fogo seismic network of the Cape Verde National Institute of Meteorology and Geophysics (INMG), who also alerted Civil Protection authorities one day before the start of the eruption (Cape Verde Civil Protection, personal communication).

Cape Verde archipelago has semi-arid climatic conditions, with the southern region being drier. Due to its geographical location, close to the equator, and thermal mass of the surrounding ocean, the average annual temperature of Fogo Island is about 25°C, with variation of the average daily temperature only 5°C in the year (Heilweil et al., 2009). However, the temperature can decrease to 0°C in Chã das Caldeiras (Mota Gomes, 2006). Rainfall varies throughout the year, with long dry spells, concentrating mainly between July and October (equatorial monsoon; Vailleux and Bourguet 1974), and ranging from $550 \text{ mm} \cdot \text{y}^{-1}$ for the northern regions to $150 \text{ mm} \cdot \text{y}^{-1}$ for the southern regions, being extremely variable from one year to another. As in other volcanic oceanic island, precipitation is lost primarily through evapotranspiration and runoff to the sea, as a consequence of warm weather, the thickness and type of soil cover, and steep topographic gradients (Langworthy and Finan, 1997). In general, most of infiltration and groundwater recharge occurs in the higher-altitude areas of the island, where precipitation rates are highest, high-permeability basalts, and closed-basin calderas are combined (Barmen et al., 1984; Kallrén and Schreiber 1988; Barmen et al., 1990; Heilweil et al., 2009). In contrast to other active oceanic hotspots, environmental tracers show that deep geothermal circulation does not strongly affect groundwater of Fogo Island (Heilweil et al., 2009). However, there are no comprehensive geochemical studies available in the literature for this hydrothermal system.

The 2014–2015 Volcanic Eruption

On November 23, 2014, a volcanic eruption began at the west flank of Pico do Fogo (González et al., 2015; Cappello et al., 2016; Richter et al., 2016; Calvari et al., 2018; Klügel et al., 2020). The eruption started from a fissure which opened along the southwest flank of the volcano, very close to the site of the previous 1995 eruptive fissure (white star in **Figure 1C**). The 2014 lava flows caused the evacuation of ~1000 residents living in the vicinity of the volcano, and the closure of a local airport. The 2014 event is well documented by a variety of field observations

Fumarole	Sample	Date	Temp	CO ₂	HCI	S _{total} (SO ₄)	H ₂ O	He	H ₂	02	N ₂	CH ₄	со			
	code		(°C)	(mmol/mol total gas)												
F1	1	April 30, 2007	113.0	443.4	0.0024	5.824	541.6	0.0066	0.8336	0.0842	8.216	0.0000938	0.007604			
F1	2	November 06, 2008	140.0	134.3	0.0718	0.758	863.8	0.0001	0.0052	0.1982	0.874	0.0000025	0.000217			
F1	3	February 21, 2010	68.6	455.5	0.0246	4.784	538.4	0.0034	0.3876	0.0563	0.856	0.0000087	0.000043			
F1	4	February 24, 2012	62.4	450.6	0.6046	0.752	544.6	0.0061	0.6083	0.1500	2.537	0.0001387	0.005356			
F1	5	April 27, 2013	89.6	115.3	0.0773	1.527	881.1	0.0021	0.2587	0.0273	1.656	0.0000535	0.001537			
F1	6	November 06, 2013	87.3	161.5	0.0691	1.468	830.2	0.0066	0.8334	0.8683	4.960	0.0000025	0.002648			
F1	7	March 22, 2014	77.1	123.6	0.0749	1.262	874.1	0.0021	0.2363	0.1387	0.501	0.0000486	0.001463			
F2	1	November 06, 2008	344.0	167.3	0.380	2.381	828.6	0.0004	0.0391	0.2294	1.069	0.0000098	0.002000			
F2	2	December 08, 2010	344.0	283.0	0.075	1.229	715.1	0.0030	0.4545	0.0082	0.175	0.0000684	0.012672			
F2	3	November 06, 2013	317.9	115.6	0.113	1.619	882.6	0.0030	0.6098	0.1803	1.671	0.000098	0.011084			
F2	4	March 22, 2014	266.0	95.0	0.151	1.177	902.6	0.0032	0.3894	0.1181	0.595	0.0000723	0.009940			
F2	5	January 03, 2015	306.2	388.5	0.170	2.384	606.2	0.0464	0.3318	0.0972	2.160	0.0001474	0.002920			
F2	6	February 16, 2015	331.6	483.7	0.178	2.190	511.3	0.0094	0.2473	0.0930	2.218	0.0001241	0.009202			
F2	7	October 30, 2016	354.0	380.6	0.109	1.638	614.7	0.0031	0.3999	0.2714	1.223	0.0000601	0.038168			

TABLE 1 | Outlet temperatures (in degree Celsius) and chemical composition (dry gas fraction in mmol/mol total gas) of fumarolic gases from Pico do Fogo volcano.

(Gonzalez et al., 2015; Cappello et al., 2016; Richter et al., 2016; Klügel et al., 2020). After 77 days of activity, the eruption ended on February 8, 2015.

Several studies have been published that characterize the 2014-2015 volcanic eruption. Mata et al. (2017) carried out a petrological study of lavas to investigate the magma plumbing system, reporting somewhat evolved alkaline compositions ranging from tephrites to phonotephrites. The lavas transported clinopyroxene and kaersutite phenocrysts and megacrysts to the surface, suggesting the main stages of magma evolution took place possibly at mantle depth (25.6 \pm 5.5 km b.s.l.). The presence of kaersutite that has been extensively replacement by rhönite indicates that, assuming these crystals are phenocrystic, magma stalled at a shallower depth (<1.5 km b.s.l.) for at least a month prior eruption (Mata et al., 2017). Should these crystals be related to older, more evolved, pockets of phonotephrite magma that were intersected by the tephrite (Klügel et al., 2020) stalling at these depths may not be a salient feature of the principle magmatic pathway. Clinopyroxene and olivine crystals have been examined (Klügel et al., 2020) and feature ubiquitous 10-50 µm wide rims on clinopyroxene crystals, implying they grew in a few days or weeks during the ongoing eruption at upper mantle pressures ~600 MPa (21 km depth), as a consequence of H₂O loss from the melt. According to those authors, the effect of the loss of H₂O could be due to magma degassing after its movement towards more superficial levels, or the exsolution of CO2 at depth. Ultimately, they concluded that rim zonation of olivine phenocrysts are indicative of magma rising to the surface in less than half a day after leaving this deep stall zone.

The eruption gave rise to a three-branch lava flow of rapid expansion that almost entirely destroyed the Portela and Bangeira villages within the Cha das Caldeiras (Calvari et al., 2018). Bagnardi et al. (2016) used high-resolution tri-stereo optical imagery acquired by the Pleiades-1 satellite constellation to generate a 1 m resolution DEM to quantify topographic changes associated with the 2014–2015 eruption at Fogo. They obtained a lava flow volume of $45.83 \pm 0.02 \times 106 \text{ m}^3$, emplaced over an area of 4.8 km^2 at a mean rate of $6.8 \text{ m}^3 \cdot \text{s}^{-1}$. Cappello et al. (2016) used the HOTSAT satellite thermal monitoring system and the MAGFLOW lava flow emplacement model to forecast lava flow hazards during the 2014–2015 Fogo eruption. González et al. (2015) observed deformation at the summit of Fogo Island associated with the last eruption occurred in Pico do Fogo by means of Bayesian inversion of Sentinel-1 TOPS interferometry.

METHODOLOGY

Sample Collection

Direct gas sampling was performed at two fumarole vents located in the northwest sector inside the summit crater of Pico do Fogo (F1 and F2; stars in **Figure 1D**). Fumarole gas samples were collected during the period 2007–2016, over 14 (F1) and 8 (F2) field surveys (**Figure 1D**). F1 fumarole was characterized by a lower outlet temperature (62–140°C) during the period 2007–2014, whereas the F2 fumarole was characterized by a higher outlet temperature (266–354°C) during the entire sampling period (2008–2016). After 2014, the gas flow rate from fumarole F1 decreased significantly.

During sampling, a glass funnel was introduced into the fumarolic vent and buried to prevent atmospheric air entering into the sampling system. Fumarolic gas was collected in a vacuum pre-evacuated glass flask filled with 50 ml of a 4N KOH solution (Giggenbach and Gougel, 1989a, 1989b). The H₂O vapour condenses, and the acidic gases (CO₂, H₂S, SO₂, HCl) are absorbed into the alkaline solution, while the non-condensable gases (N₂, O₂, He, H₂, CH₄, CO) are concentrated in the gas phase of the sampling flask. Different aliquots of the discharging of gases were also collected in separate high-vacuum

TABLE 2 | δ^{18} O and δ^{2} H isotope ratios (‰ vs.VSMOW) in fumarole condensates, air-corrected ³He/⁴He and ⁴He/²⁰Ne ratios, and fraction of He source (air, magma and crust) in fumarole gas samples. δ^{13} C-CO₂ (‰ vs.VPDB), CO₂/³He ratios and fraction of C source (L = limestone, S = organic sediment and M = mantle sources) in fumarole gas samples. M/(L + S) refers to the fraction of components of Pico do Fogo fumaroles are also present.

Fumarole	Date	δ ¹⁸ O (‰ vs VSMOW)	δD (‰ vs) VSMOW)		CO ₂ sources									
				(³ He/ ⁴ He) _{corr} R/R _A ^a	⁴ He/ ²⁰ Ne	Magma He (%)	Crust He (%)	Air He (%)	δ^{13} C–CO ₂ (‰ vs PDB)	CO₂/ ³ He (x10 ^{−9})	Mantle (M) (%)	Limestone (L) (%)	Sediment (S) (%)	M/ (L+S) (%)
F1	April 30, 2007	n.m.	n.m.	$8.09^{a} \pm 0.08$	123 ^a	91.6	8.2	0.2	n.m.	10.2 ^b	n.m.	n.m.	n.m.	n.m.
F1	November 06, 2008	n.m.	n.m.	8.22 ^a ± 0.09	362 ^a	93.2	6.7	0.1	n.m.	7.9	n.m.	n.m.	n.m.	n.m.
F1	May 10, 2009	n.m.	n.m.	7.81 ^a ± 0.09	23 ^a	87.4	11.3	1.3	n.m.	11.5 ^b	n.m.	n.m.	n.m.	n.m.
F1	February 21, 2010	-5.2 ± 0.0	-22 ± 3	$8.53^{a} \pm 0.09$	228 ^a	96.7	3.2	0.1	-4.07 ± 0.01	10.3	15	80	5	0.17
F1	December 08, 2010	n.m.	n.m.	8.23 ^a ± 0.07	94 ^a	93.1	6.6	0.3	-4.22 ± 0.01	10.7 ^b	14	80	6	0.16
F1	February 24, 2011	-9.6 ± 0.1	-25 ± 3	8.23 ^a ± 0.15	42 ^a	92.7	6.6	0.7	-4.10 ± 0.01	10.6	14	81	5	0.16
F1	February 24, 2012	-8.7 ± 0.1	-34 ± 2	7.93 ^a ± 0.12	194 ^a	89.8	10.1	0.1	-4.26 ± 0.01	12.7	12	82	6	0.13
F1	April 27, 2013	-8.0 ± 0.1	-40 ± 2	$8.20^{a} \pm 0.09$	18 ^a	91.4	6.8	1.8	-4.48 ± 0.02	11.9	13	80	7	0.14
F1	November 06, 2013	-6.0 ± 0.1	-24 ± 1	8.63 ^a ± 0.10	406 ^a	97.9	2.0	0.1	-4.37 ± 0.02	9.0	17	77	6	0.20
F1	March 22, 2014	-8.3 ± 0.1	-33 ± 6	$8.37^{a} \pm 0.09$	110 ^a	94.8	4.9	0.3	-4.62 ± 0.02	12.7	12	80	8	0.15
F1	January 03, 2015	-9.0 ± 0.1	-32 ± 3	n.m.	n.m.	n.m.	n.m.	n.m.	-3.61 ± 0.02	n.m.	n.m.	n.m.	n.m.	n.m.
F1	February 16, 2015	-11.8 ± 0.1	-30 ± 4	8.33 ^a ± 0.08	36ª	93.7	5.4	0.9	-3.88 ± 0.05	10.7	14	81	5	0.16
F1	November 22, 2015	n.m.	n.m.	n.m.	n.m.	n.m.	n.m.	n.m.	-4.62 ± 0.02	n.m.	n.m.	n.m.	n.m.	n.m.
F1	October 30, 2016	n.m.	n.m.	7.66 ^a ± 0.02	106 ^a	86.7	13.0	0.3	-4.29 ± 0.02	7.8 ^b	19	76	5	0.24
F2	May 10, 2009	n.m.	n.m.	7.73 ± 0.23	2	71.0	10.0	19.1	n.m.					
F2	December 08, 2010	n.m.	n.m.	8.33 ± 0.12	519	94.5	5.5	0.0	20 ± 0.02	10.5	14	80	6	0.17
F2	April 27, 2013	-1.1 ± 0.1	-26 ± 4	8.28 ± 0.57	1	60.7	3.9	35.4	-4.28 ± 0.03					
F2	November 06, 2013	-1.7 ± 0.1	-24 ± 3	8.82 ± 0.09	481	100.0	0.0	0.0	-4.06 ± 0.04	8.9	17	78	5	0.20
F2	March 22, 2014	-0.7 ± 0.1	-23 ± 2	8.36 ± 0.08	167	94.7	5.1	0.2	-4.54 ± 0.07	12.0	13	80	7	0.14
F2	January 03, 2015	-3.1 ± 0.1	-30 ± 5	n.m.	n.m.	n.m.	n.m.	n.m.	-3.78 ± 0.01	n.m.	n.m.	n.m.	n.m.	n.m.
F2	February 16, 2015	-3.1 ± 0.1	-18 ± 3	8.33 ± 0.08	36	93.7	5.4	0.9	-3.90 ± 0.04	10.7	14	81	5	0.16
F2	November 22, 2015	n.m.	n.m.	n.m.	n.m.	n.m.	n.m.	n.m.	-4.44 ± 0.03	n.m.	n.m.	n.m.	n.m.	n.m.

n.m.: not measured.

^aData from Alonso et al. (2021).

 $^b\mbox{Measured}$ in sample for $\delta^{13}\mbox{C-CO}_2$ analysis.

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Aiuppa et al. (2020). Codes from **Table 1**.

Pb glass containers for the analysis of C and He isotopes and a sample of the vapor condensate was collected with an ice-cooled condenser in order to analyze O and H isotopes.

Analytical Techniques

The non-condensable gases were analyzed by a Varian 3800 gaschromatograph equipped with a thermal conductivity detector (TCD) for the analysis of N₂, O₂, He and H₂, and a flame ionization detector (FID) for the analysis of CH₄ and CO (converted to CH₄ at 400°C with a methanizer). Separation was carried out in a 20 m Molecular Sieve 5A packed column using argon as the carrier gas. Analytical error was estimated as <5% for the main gas components and <10% for minor gas compounds.

An aliquot of the alkaline liquid phase was oxidized with H_2O_2 for the determination of CO_2 , total sulfur (S_{total}) and HCl. The determination of CO_2 was accomplished through the analysis of CO_3^{-2} performed by titration with diluted HCl using an automatic titration system (Metrohm 716 DMS Titrino, Metrohm). S_{total} and HCl content were determined through the analysis of SO_4^{-2} and Cl⁻, respectively, by means of ion chromatography coupled to a conductivity detector (Dionex DX-500 system).



The isotopic signature of C in CO₂ (δ^{13} C-CO₂) from fumarolic gas was determined with an isotope-ratio mass spectrometer (IRMS) (Finnigan MAT 253, Thermo Fisher Scientific) equipped with a dual inlet system. The results are reported in δ units per mil vs. Vienna Pee Dee Belemnite standard (VPDB). The analytical error for δ^{13} C-CO₂ in the fumarolic gases was estimated in ±0.01‰.

Helium concentration and ³He/⁴He ratios were analyzed with a high-precision noble gas mass spectrometer (VG Isotech modified VG-5400) using the method described by Sumino et al. (2001). The analytical error for ³He/⁴He determination was <2%. The correction factor for the He isotope ratio was determined by measurements of inter-laboratory He standard named HESJ, with a recommended ³He/⁴He value of 20.63 \pm 0.10 R_A (Matsuda et al., 2002). The measured ³He/⁴He ratios were corrected for the addition of air on the basis of the ⁴He/²⁰Ne ratios measured by mass spectrometry, assuming that Ne has an atmospheric origin (Craig and Lupton, 1976). The error in the reported ⁴He/²⁰Ne ratios is 14%, which is estimated from stability of sensitivity of the mass spectrometer during repeated measurements of the air standard.

The isotopic signatures of hydrogen (${}^{2}H/{}^{1}H$, $\delta^{2}H$) and oxygen (${}^{18}O/{}^{16}O$, $\delta^{18}O$) in the fumarole condensate were determined using an IRMS (Finnigan MAT 253, Thermo Fisher Scientific) after a certain time of isotopic equilibrium with helium containing a known concentration of H₂ and CO₂, respectively, at 25°C (Epstein and Mayeda, 1953). In the case of ${}^{2}H/{}^{1}H$ ratio the use of platinum was necessary to catalyse the reaction (Coplen and Wildman, 1991). The results are reported in δ units per mil vs. Vienna standard mean ocean water (VSMOW) with a precision of ±0.1 and ±1‰ for $\delta^{18}O$ and $\delta^{2}H$, respectively.

RESULTS

The complete chemical composition of Pico do Fogo sampled fumaroles is reported in **Table 1**. Two groups of fumaroles, F1 and F2, can be distinguished on the basis of measured outlet temperatures and gas contents. F1 fumarolic gases show low ($<150^{\circ}$ C) outlet temperatures, whereas F2 shows intermediate higher temperatures ($>300^{\circ}$ C up to 354° C), suggesting active

magmatic degassing at the time of this study, and potentially more oxidizing conditions. Both fumaroles are water-dominated, with compositions ranging from 538.4 to 881.1 mmol/mol for F1 and 511.3–902.6 mmol/mol for F2. In the F1 fumarole, the concentration of CO₂ in the total gas reaches 455.5 mmol/mol, followed by N₂ (0.50–8.22 mmol/mol), H₂ (0.01–0.83 mmol/mol) and HCl (0.60 mmol/mol). The concentration of O₂, He, CO and CH₄ reached 0.86, 0.007, 0.007 and 1.4 × 10⁻⁴ mmol/mol, respectively. For the F2 fumarole, CO₂ concentration in the total gas reaches 483.7 mmol/mol, again followed by N₂ (0.18–2.22 mmol/mol), H₂ (0.04–0.61 mmol/mol) and HCl (0.38 mmol/mol) The concentration of O₂, He, CO and CH₄ concentrations are 0.27, 0.05, 0.038 and 1.5 × 10⁻⁴ mmol/mol, respectively.

Table 2 reports the oxygen and hydrogen isotopic composition of fumarole condensates, as well as carbon and helium isotopic composition of fumarole gas samples. The values obtained for water stable isotopes (δ units per mil vs. VSMOW) show much heavier δ^{18} O values at the F2 fumarole (-3.1 to -1.1‰; average -1.9‰) compared to fumarole F1 (-11.8 to -5.2‰; average -8.2‰). F1 fumarole presents a range of δ^2 H values from -40 to -22‰, with an average value of -30‰, while F2 fumarole δ^2 H values range from -30 to -18‰ with an average of -24‰. The carbon isotopic composition of Pico do Fogo fumarolic CO₂ (δ^{13} C vs. VPDB) varies from -4.62 ± 0.02 to -3.61 ± 0.02‰ (average -4.23‰) at the F1 fumarole, and from -4.54 ± 0.07 to -3.78 ± 0.01‰ (average -4.17‰) at the F2 fumarole.

The ${}^{3}\text{He}/{}^{4}\text{He}$ isotopic ratios of the fumarolic gases were measured for all samples collected in the study period. The air-corrected ${}^{3}\text{He}/{}^{4}\text{He}$ (R/R_A)_{corr} ranged from 7.66 to 8.63 (Alonso et al., 2021) and from 7.73 to 8.82 for F1 and F2, respectively. Based on three-component mixture of magmatic He (8.82 ± 0.09 R_A), crust He (0.01 R_A; Sano et al., 1985) and air He (1 R_A), the proportion of magmatic He at Pico do Fogo varies between 86.7 and 97.9% for F1 and 71.0 and 100%, for F2 (**Table 2**).

The relative contents of CO₂ and He (given as $CO_2/{}^{3}$ He in **Table 2**) are in the range 7.8–12.7 × 10⁹ and 8.9–12.0 × 10⁹ for F1 and F2, respectively, which are higher than mid-ocean ridge basalt (MORB) source mantle (~2 × 10⁹; Marty and Jambon, 1987). Based on a mass balance calculation proposed by Sano and Marty



FIGURE 4 | Ternary chemical composition diagrams for Pico do Fogo fumarole samples: (A) CO₂-N₂-He, (B) CO₂-H₂O-He, (C) CO₂-H₂O-St and (D) CO₂-St-HCl. The green circle shows the mean values of composition summit fumaroles of Pico do Fogo from Aiuppa et al. (2020). The orange star indicates composition of the 2014–2015 Fogo eruptive plume from Hernández et al. (2015). Codes from Table 1.

(1995), a significant contribution of mantle carbon (12–19% for F1, 12–17% for F2) can be estimated for Pico do Fogo volcano.

DISCUSSION

Fumarolic Fluid Geochemistry and Equilibrium Considerations

The fumarolic samples collected during this study provide important insights into the magmatic composition of gas discharge. In this section we focus on the major gaseous species, H₂O, CO₂, and other minority species such as CO and H₂, which are essential to understand the thermodynamic processes occurring within the magmatic-hydrothermal system of the volcano. Two fumaroles with different characteristics were sampled, low temperature (F1) representing a hydrothermal endmember and a medium temperature (F2) representing a magmatic member (higher CO₂/St, H₂/H₂O ratios and lower St/Cl ratios), giving rise to the possibility that equilibrium conditions of the hydrothermal gas in the H₂O-H₂-CO₂-CO-CH₄ system might not be the same in both cases. In **Figure 2A**, F1 gas samples (except sample 2F1) plot close to the rock buffer, indicating that the H2-H2O-H2S-SO2-CO-CO2 system at Fogo is controlled by reactions involving the rock buffer and not the H₂S/SO₂ gas buffer. The separate position of sample 2F1 may be due to the extra addition of H₂O which results in more negative R_H factor $[R_H = \log(f_{H2}/f_{H2O}) \sim \log(X_{H2}/f_{H2O})$ X_{H2O}); Giggenbach, 1987]. However, the samples from the F2 fumarole, although close to the rock buffer line (H₂-H₂O-H₂S-SO₂-CO-CO₂ system is controlled by reaction involving the rock buffer), are also close to the SO₂-H₂S buffer, which involves contribution of elemental sulfur and SO₂. Among the samples analyzed, 2F1 and 1F2 have the lowest R_H, implying more oxidizing conditions (Giggenbach, 1987). These samples were collected in 2008, when the first magmatic intrusion of the period studied is assumed to have occurred. Therefore, the chemical composition of the majority of the gases in the fumarole discharges of the Pico do Fogo volcano are dominantly controlled by the rock buffer (FeO-FeO_{1.5}) at 50°C<T<100°C, with a possible influence of gas phase reactions involving sulfur species at higher (>200°C) temperatures. These indices underscore that this system is characterized by extensive exchange of reactive chemical species between phases within the hydrothermal system,



and so strongly implies that deep changes in the system will be detected at the surface after some delay.

As in Figure 2A and Figure 2B shows the redox conditions in the gas equilibration zone are controlled by the FeO-FeO_{1.5} hydrothermal buffer (Giggenbach, 1987) at relatively high H₂/ H₂O ratios. Except for samples 3F1 and 4F2, all values plot between the vapor line and the composition resulting from singlestep vapor separation at two fixed Ts temperatures (100 and 200°C) starting from an initial temperature (dashed lines in Figure 2B), indicating the vapours are generated through a single-step steam separation from boiling liquids of different initial temperatures (Chiodini and Marini, 1998). The position of the points along the single-step vapor separation (SSVS) lines is indicative of the extent of the boiling process. Values located near the vapor line have relatively high separation temperatures and low fractions of separated vapor, whereas values close to the 100°C Ts line are related to more energetic boiling processes, i.e., higher fractions of separated vapor (Chiodini and Marini, 1998). Sample 3F1 can be considered representative of equilibrated vapors, while sample 4F2 represents a pure equilibrated liquid phase. In general terms, there are significant variations in equilibrium temperatures between samples and they plot along different SSVS lines. F2 samples in particular have a wide range of separation temperatures indicative of changes in the physico-chemical conditions e.g., changes in the emission rates of magmatic fluids from the magmatic reservoir.

Estimated equilibrium temperatures for Pico do Fogo fumaroles based on the slow-reaction CH_4 , $CO-CO_2$ (**Figure 3**; Giggenbach, 1987) thermometer show a range from 205 to 428°C at F1, and from 341 to 524°C at F2 in agree with temperatures estimated in Figure 2B. Two maximum values in F1 are observed: the first in 2007 and the second sustained over time (since 2010) and reaching its maximum value in November 2013, one year before the start of the most recent eruption. No data were estimated after 2014 for F1 fumarole. F2 fumarole, measured for the first time in 2009, records a constant increase in the equilibrium temperature until also reaching its maximum value in November 2013, before decreasing rapidly until January 2015, when equilibrium temperatures began increasing again. The changes observed in the equilibrium temperatures of F1 and F2 were likely due to an increase of volatile pressure in the magmatic-hydrothermal system of Pico do Fogo volcano, as is suggested by other geochemical and geophysical signals observed during the same period (González et al., 2015; Cappello et al., 2016; Richter et al., 2016; Calvari et al., 2018; Klügel et al., 2020; Alonso et al., 2021). Temporal variations of equilibria temperature estimated during the period of study showed similar behavior than CO/CO₂ ratio, with a decreasing trend towards the magmatic reactivation onset of 2009-2010, and an increasing trend towards the volcanic eruption onset. Injections of deep-seated CO₂ into the Fogo magmatic-hydrothermal system might have caused the observed variations in the fumarole CO/CO2 ratio. Magma ascended, decompression caused exsolution of vapor and magmatic gases, which were subsequently injected into the hydrothermal system, producing the observed increase in the temperature and He flux, and later the CO_2 flux (Alonso et al., 2021), of the system.

The Pico do Fogo Fumaroles present, in general, compositions that are less hydrous than the fumaroles of others volcanic systems of the Macaranosia (Aippa et al., 2020). This can be explained be the long periods without significant rainfall at Fogo



Island, and rainfall can be said to not play an important role in fumarole gas chemistry of Pico do Fogo (Vailleux and Bourguet 1974). This matches the observably weaker hydrothermal system of Pico do Fogo Volcano weak when compared to those present on geologically comparable, yet wetter, volcanic islands.

Figure 4 presents the main volatile compositions (CO₂, He, N₂, HCl, S_{total} and H₂O) measured in the F1 and F2 fumaroles. A CO₂-N₂-He ternary diagram (Figure 4A) shows that fumarolic gas samples are dominated by a magmatic component rich in CO_2 , except for one sample (5F2), which falls between the CO_2 and He end-member. The CO2-H2O-He ternary diagram (Figure 4B) is used to distinguish between the magmatic and hydrothermal components of volcanic gases, corroborating the above interpretation by showing a strong magmatic component for most fumarolic samples. Figure 4C (CO2-H2O-St) and Figure 4D (CO₂-St-HCl) provide further information about the chemical compositions of volcanic gases. Figure 4C evidences an extensive steam condensation process, probably reflecting the relatively low temperature conditions of the fumaroles' conduits (Aiuppa et al., 2020) and/or a relatively weak hydrothermal system present at Pico do Fogo volcano (discussed above). The condensation process favors the scrubbing of Stotal and HCl as the magmatic gases rise to the

surface as are observed in **Figures 4C,D**. The similar trends observed in the C:S:Cl ratios suggest that the sources for the major constituents and processes affecting their concentrations are similar. We note that F2 values seems to have a shallower component in 2008 during the first magmatic reactivation onset, and in 2013–2014, one year before the volcanic eruption onset.

Although only minor changes were observed in the outlet fumarole temperatures (**Figure 5A**), two clear increases were observed in the vapor fraction of fumarolic discharges during the periods 2008–2010 and 2013–2014 (**Figure 5B**). The increase in one order of magnitude in the steam/gas ratio supports two hypotheses: increasing temperature of the hydrothermal aquifer due to the input of magmatic gases and/or exolution of gases from the volcanic hydrothermal system as a consequence of depressurization of the system.

The steam/gas increases were accompanied by increases in concentrations of gas species that are poorly soluble in magma, such as He (see below), and are supported by the temporal evolution of equilibrium temperatures of the hydrothermal system (**Figure 3**), with two clearly differentiated peaks coinciding with the magmatic reactivation and pre-eruptive periods. An increase of



Podosek, 2002). Curves represent theoretical mixing lines between air and other end-members.



equilibrium temperature at depth (see Figure 3) may cause a decrease in the coefficients of vapor-liquid gas distribution, and consequently an increase in the steam fraction of the

fluids leaving the hydrothermal system (Cioni et al., 1984). The relative low values of the steam/gas ratio in the fumarolic discharges at the Pico do Fogo volcano compared to



those observed in other volcanic systems in Macaronesia (Caliro et al., 2015; Ferreira et al., 2005; Melián et al., 2012) suggest poor shallow boiling hydrothermal aquifer underneath Fogo summit and therefore a greater magmatic signature evidence with a lower CO_2/S_{total} ratios and higher SO_2/H_2S ratios (Aiuppa et al., 2020). The equilibrium temperatures estimated for Pico do Fogo magmatic-hydrothermal system reached their highest values during the initial magmatic reactivation episode and then prior to eruption (**Figure 3**). Following this hypothesis would suggest that the relatively lower steam/gas ratios observed in 2008–2010 suggest that at that time, fumaroles were fed by a deeper and non-degassed magma source.

The increase of steam/gas ratio in Pico do Fogo volcano could be explained as a consequence of increase of permeability at depth due to increased fracturing, leading to more efficient gas escape and hence decompression. Both processes could have occurred in the Pico do Fogo hydrothermal volcanic system in the study period. The presence of a weak hydrothermal system and agreement between thermal and thermally sensitive geochemical indices suggests that the degree of magmatic contribution is the factor that better explains the observed variations. The steam/gas ratio therefore may be an effective parameter for the detection of changes in heat flow at depth at Fogo.

$\delta^2\text{H-H}_2\text{O},\,\delta^{18}\text{O-H}_2\text{O},\,\delta^{13}\text{C-CO}_2$ and ${}^3\text{He}/{}^4\text{He}$ in Fumarole Discharge

Oxygen and hydrogen isotopic ratios in fumarole condensates, together with carbon and helium isotopic ratios are given in **Table 2. Figure 6** reports $\delta^2 H$ and $\delta^{18}O$ values, the Local Meteoric Water Line (LMWL; $\delta^2 H = 7.8 \times \delta^{18}O + 6.8$; Sacchi et al., 1995), which is slightly lower than the Global Meteoric Water Line (GMWL; $\delta^2 H = 8 \times \delta^{18}O + 10$; Craig, 1961). Condensates have a wider range of $\delta^2 H$ isotopic values



(F1 from -40 to -22‰ and F2 from -30 to -18‰) than δ^{18} O (F1 from -11.8 to -5.2‰ and F2 from -3.1 to -1.1‰). F1 samples are depleted in δ^{18} O relative to the LMWL by up to -7‰, whereas F2 samples are enriched by up to +3.5‰. All samples are depleted in δ^{2} H relative to SMOW. Rainfall and well waters from Fogo Island (Heilweil et al., 2009) have a δ^{2} H range similar to that of fumarole condensates, but a much narrower range of δ^{18} O values (**Figure 6**).

Depletion/enrichment in δ^{18} O relative to the meteorological LMWL, without changes in δ^2 H, are caused by the isotopic exchange of oxygen with CO2 and bedrock minerals at low and high temperature (black arrows in Figure 6; Karolyte et al., 2017), or by the separation of water and steam under geothermal conditions (Giggenbach, 1992). In natural fluids where the vapor phase coexists with abundant CO₂, the exchange of δ^{18} O between H₂O and CO₂ can be significant as long as the isotopic exchange rate and/or the residence time of the fluids are sufficiently long (Chiodini et al., 2000). The variations observed between F1 and F2 fumaroles suggest their fluid source contributions and/or fractionation processes are different. Samples plotting close to the LMWL probably have a stronger component of heated local water. F2 gas condensates are shifted towards heavier δ^{18} O isotope values, indicating water-rock interaction and/or mixing with volcanic fluids. δ^{18} O exchange between steam and CO₂, the principal components of volcanic-hydrothermal vapors, can be mainly considered for fumarole F1 (Chiodini et al., 2000; Heilweil et al., 2009), since δ^{18} O exchange between both components have a significant bearing on the interpretation of δ^{18} O data (Chiodini et al., 2000).

The ³He/⁴He ratios are normalized to that in the atmosphere ($R_A = 1.4 \times 10^{-6}$; Ozima and Podosek 2002). ³He/⁴He and ⁴He/²⁰Ne ratios are displayed in **Figure 7** and listed in **Table 2**. The air-corrected ³He/⁴He ratios at Pico do Fogo volcano (**Table 2**) range from 7.66 to 8.63 R_A in Fumarole F1, and between 7.73 and 8.82 R_A for F2, while ⁴He/²⁰Ne ratios fluctuate significantly, ranging from 18 to 406 at F1 and 1 to 519 at F2. Fumarole F1 measurements are in agreement with those previously reported by Dionis et al. (2015b). All fumarolic samples fall on or close to a single mixing line between Air and MORB, with relatively constant ³He/⁴He indicating that variation in the ³He/⁴He ratios is dominated by the level of atmospheric contamination of the magmatic component (**Figure 7**) and indicating that magmatic He is predominantly of MORB-source like, upper mantle origin.

Based on **Figure 7**, mixing between three sources of He has been considered: magmatic, atmospheric, and crustal endmembers. We assume ${}^{3}\text{He}/{}^{4}\text{He}$ value of the magmatic endmember to be 8.82 R_A based on the highest air-corrected ${}^{3}\text{He}/{}^{4}\text{He}$ ratio observed in this study. Although this value is slightly higher than average ${}^{3}\text{He}/{}^{4}\text{He}$ ratio (8.5 ± 0.2 R_A) of olivine phenocrysts in 126 ka ankaramite lava from Fogo Island (Foeken et al., 2012), they are similarly in the range of MORBs (8 ± 1 R_A, Graham, 2002), indicating almost negligible contribution of a plume-type endmember associated with higher ${}^{3}\text{He}/{}^{4}\text{He}$ ratio than the MORB range. The He contribution from the crust is required, since some aircorrected ${}^{3}\text{He}/{}^{4}\text{He}$ data are significantly lower than the magmatic ${}^{3}\text{He}/{}^{4}\text{He}$ value. The proportion of these three reservoirs considered can be calculated using the following equations:

$$({}^{3}\text{He}/{}^{4}\text{He})_{i} = ({}^{3}\text{He}/{}^{4}\text{He})_{a} \times A + ({}^{3}\text{He}/{}^{4}\text{He})_{m} \times M + ({}^{3}\text{He}/{}^{4}\text{He})_{c} \times C$$
(1)

$$\frac{1}{(^{4}\text{He}/^{20}\text{Ne})_{i}} = A/(^{4}\text{He}/^{20}\text{Ne})_{a} + M/(^{4}\text{He}/^{20}\text{Ne})_{m} + C/(^{4}\text{He}/^{20}\text{Ne})_{C}$$
(2)

$$A + M + C = 1 \tag{3}$$

where subscripts "a", "m" and "c" indicate atmospheric, magmatic and crustal sources respectively, and A, M and C are the fraction of helium from atmospheric, magmatic and crustal respectively (Sano et al., 1985; Sano and Wakita, 1985). Using the following values: $({}^{3}\text{He}/{}^{4}\text{He})_{a} = 1 \text{ R}_{A}$, $({}^{4}\text{He}/{}^{20}\text{Ne})_{a} =$ 0.318, $({}^{3}\text{He}/{}^{4}\text{He})_{m} = 8.82 \text{ R}_{A}$, $({}^{4}\text{He}/{}^{20}\text{Ne})_{m} = 1000$, $({}^{3}\text{He}/{}^{4}\text{He})_{c} =$ 0.01 R_A, $({}^{4}\text{He}/{}^{20}\text{Ne})_{c} = 1000$, the fraction of atmospheric, magmatic and crustal components can be calculated (Sano and Wakita, 1985), and are given in Table 2. Fumarole F1 contains a magmatic component between 86.7 and 97.9%, while the atmospheric component ranges from 0.1 to 1.8%, and the crust component from 2.0 to 13.0%. Fumarole F2 has a magmatic component between 71.0 and 100%, and atmospheric and crust components from 0 to 35.4% and 0-10.0% respectively. The atmospheric contamination is significant in the F2 samples of May 2009 and April 2013 that present a low ⁴He/²⁰Ne ratio (19 and 35%, respectively).

In order to further explore potential sources for the fumarole gases, relationships among CO₂, ³He and ⁴He were investigated (**Figure 8**). F1 and F2 samples form a cluster of points, with a MORB-like composition due to high ³He/⁴He ratios (7.66–8.82 R_A, air-corrected) and CO₂/ ³He ratios (7.76–14.9 × 10⁹). The sample cluster shows a slightly linear trend, extending towards the CO₂ vertex, with an average ³He/⁴He ratio of 8.2 R_A. Two samples show a slightly higher contribution of crustal-derived ⁴He with respect to the main sample cluster.

Fumarole fluid contributions can be further differentiated using isotopic He, C, and CO₂ gas (Figure 9). F1 and F2 have δ^{13} C-CO₂ ranging from -4.62 ± 0.02 to -3.61 ± 0.02‰ (vs. VPDB) and $CO_2/^3$ He ratios between 7.8 and 12.7×10^9 , indicating a strong limestone contribution, as can be expected based on the previous studies (Jørgensen and Holm, 2002), but with different mantle contribution in the range 12–19%. The variation in $CO_2/{}^{3}He$ is likely linked to the addition of ³He-poor, CO₂-rich gases (i.e., limestonederived CO₂), and radiogenic He (Figure 9). Figure 9A confirms this interpretation that Pico do Fogo fumarole samples are a mixture of a mantle-derived (MORB-like) component and a crustal component enriched in ¹³C, with high $CO_2/{}^{3}He$ (addition of limestone-derived CO_2), and lower $CO_2/{}^4$ He at the same $CO_2/{}^3$ He ratio (because of the addition of crust-derived, radiogenic He; Figure 9B). F1 and F2 samples show a trend towards more positive δ^{13} C-CO₂ values and lower $CO_2/^4$ He ratios (Figure 9B), confirming an



increase in radiogenic 4 He and CO₂ due to the addition of fluids of crustal origin.

Temporal Variations in Fumarolic Gas Compositions

Temporal variations recorded in the chemical and isotopic composition of the fumarolic discharges of the Pico do Fogo volcano can be used to understand the relationship between the magmatic-hydrothermal system, and the volcanic activity observed during the study period. The reactive species CH_4 and CO are excellent indicators of T-P redox conditions (Chiodini et al., 1993). **Figure 10A** shows the temporal evolution of CO/CO₂ ratios (a robust proxy for temperature variations at depth; Chiodini, 2009) measured in the Pico do Fogo fumarole discharges. A gradual increase in CO/CO₂ is observed starting in 2010, before the large earthquake on April 12, 2012, and continuing to increase up until the onset of the eruption on November 23, 2014.

Relative abundances of gases with a magmatic origin, such as He, increase during magmatic intrusion episodes. Helium is chemically inert, non-biogenic, highly mobile, and relatively insoluble in water (the opposite behavior to CO₂). H₂ is a well-known geochemical tracer in volcanic systems. H₂ plays a role in key redox reactions that occur in magmatic gases, because of its relationship to fO_2 via the water dissociation reaction (e.g., Giggenbach 1987; Oppenheimer et al., 2012). Time series of He/ CO_2 and $\mathrm{H}_2/\mathrm{CO}_2$ for the Pico do Fogo fumaroles are given in Figures 10B,C. Both ratios are low in the period 2008–2009, and high in the 2013-2014 period. The geochemical anomalies reported here in the chemical composition of the fumarolic gases in 2009 indicate fluid input from a deeper magmatic intrusion, and so an increase in the emission of He (and therefore He/CO_2) is expected at that time. However, He/CO_2 and H₂/CO₂ ratios were higher in 2007, before the magmatic reactivation period of 2009. A similar pattern is seen in the CO/ CO₂ ratio (Figure 10A). He degassed from new, volatile-rich magma ascends easily to the surface, giving rise to the observed increase (Padrón et al., 2013) before other geochemical markers



of magmatic reactivation in the system. This fact is confirmed by an obvious contribution of mantle-derived gases before 2009, for example ${}^{3}\text{He}/{}^{4}\text{He}$ (see following paragraph). The peaks observed in H₂/CO₂ are likely due to the relative increase of H₂ emission because water-rock reactions are enhanced once magma enters the Pico do Fogo volcanic system.

CH4 is an excellent indicator of T-P redox conditions (Chiodini et al., 1993) and it is widely accepted that an injection of magmatic fluids to a hydrothermal system causes an increase in the CO₂/CH₄ of fumarolic discharges (Chiodini, 2009), albeit with some time delay due to aforementioned interactions and exchanges. The CO2 concentration in magmatic fluids is much higher than the concentration of CH₄, which in turn is more abundant in hydrothermal fluids than in magmatic fluids. The exsolution of volcanic gases with a strong magmatic component rich in SO2 (due to higher oxidizing conditions) can dilute the CH4 content, increasing the CO_2/CH_4 ratio. Figure 10D shows the temporal evolution of CH₄/CO₂ ratios measured in the Pico do Fogo fumarole discharges. Pulses in the CO₂/CH₄ ratio are indicative of punctual of injection of magmatic fluids of deep origin (Chiodini, 2009). Two sharp peaks occur in CO₂/CH₄ ratios at both fumaroles, in November 2008 and November 2013.

The temporal evolution of the CO_2/S_{total} ratio in Fumaroles F1 and F2 of Pico do Fogo volcano is shown in **Figure 10E**. CO_2/S_{total} is commonly used to study variations in volcanic activity because CO_2 and S have different solubilities in magma (Carroll and Webster, 1994; Holloway and Blank, 1994). When magma rises towards the surface, the least soluble gases (i.e., CO_2) are first exsolved, producing an increase in the CO_2/S_t . The measured CO_2/S_{total} ratio in the fumarolic discharges at Pico do Fogo volcano are in general higher than the observed in other intraplate volcanoes (Aiuppa, 2015) and from gases emitted in the 2014 eruption (Hernández et al., 2015), suggesting intense scrubbing in the deep S compound (Aiuppa et al., 2020). At Fumarole F1, a peak in the CO_2/S_t ratio was observed 48 days before the 2.9 earthquake on April 12, 2012. In Fumarole F2, two clear peaks are observed in CO_2/St , the first in late 2010 and the second after eruption onset.

During the ascent of magma to shallower levels, the solubility of volatiles in the magma marks the chemical composition of the magmatic fluid, the exsolution of the most insoluble gases occurs in the initial depressurisation phases (CO₂ and He), followed by the most soluble (HCl, HF and H₂O; Gerlach, 1986; Aiuppa et al., 2007; Burton et al., 2007; López et al., 2013). The HCl/CO2 ratio for the F1 and F2 fumarole for the period 2013-2016 is shown in Figure 10E. At Fumarole F1, a peak in the HCl/CO₂ ratio was observed 48 days before the 2.9 earthquake on April 12, 2012. In Fumarole F2, a sharp increase in HCl/CO2 was observed in March 2014, which correlates with both the increase in CO/CO_2 (Figure 10A) and the maximum value of the steam/gas ratio (heat flux; Figure 5B). This peak in HCl/CO₂ corresponds to minima in the CO₂/S_{total} measured ratio (Figure 10D). The subsequent increase in the HCl/CO₂ ratio in 2014 supports the exsolution of magmatic HCl rich gases in the magmatic plumbing system of Pico do Fogo volcano.

In general, the CO/CO₂, He/CO₂ and H₂/CO₂ ratios showed the best correlation with the variations in magmatic activity, with increases before the magmatic intrusion occurred in 2010 and the onset of the volcanic eruption in November 2014. The CH₄/CO₂, St/CO₂ and HCl/CO₂ ratios showed different temporal variabilities, although they generally increased prior to the volcanic eruption.



FIGURE 13 | Schematic model for the Pico do Fogo volcano fumarolic gas discharge during the period 2007-2016 [modified from Klügel et al. (2020]].

Peaks in the air-corrected 3 He/ 4 He ratios are a strong indicator for the contribution of magmatic He from a deep magmatic system (Barry et al., 2013). The temporal evolution of air-corrected 3 He/ 4 He ratios in both F1 and F2 fumaroles is illustrated in **Figure 11A**. Clear increases in air-corrected 3 He/ 4 He ratios are visible in both fumaroles: F1 shows a peak in 2010, from a minimum in 2009 during the magmatic reactivation onset (2008–2009) and another in late 2013, while F2 displays a slower rise to its maximum in late 2013, one year before the eruption.

The temporal evolution of the isotopic composition of CO₂ (expressed as δ^{13} C–CO₂‰ vs. VPDB, **Figure 11B**) shows a decreasing trend towards lighter values from 2010 to 2014, with heavier ($^{13}C/^{12}$ C) δ^{13} C–CO₂ values measured just after the eruption (–4.62‰ and –4.54‰ vs. VPDB for F1 and F2, respectively). Once the eruption finished, δ^{13} C–CO₂ returned to lighter values. The temporal evolution indicates enrichment in carbon from a deeper source during the eruption onset, due to the volatile rich-magma associated to the extruded eruptive magma.

Observed variations in the air-corrected ${}^{3}\text{He}/{}^{4}\text{He}$ ratios at both fumaroles are in general well accompanied by coeval increases in $\text{CO}_{2}/{}^{3}\text{He}$ (**Figure 12**), with a peak in 2012. The observed changes in the $\text{CO}_{2}/{}^{3}\text{He}$ ratio coupled with variations in the isotopic composition of CO_{2} , as well as variations in the ${}^{3}\text{He}/{}^{4}\text{He}$ ratio, are consistent with injections of magmatic fluids from depth.

Figure 11C show the temporal evolution of magmatic He fraction (M) calculated based on the model proposed by Sano et al. (1985). The magmatic He fraction presents a similar behavior for F1 and F2 fumaroles with values of M slightly lower for F1 fumaroles than for F2 fumarole. The maximum M values were observed in 2010 and 2013–2014, when the first magmatic reactivation onset occurred and 1 year before of volcanic eruptions onset, respectively. These peaks are coincident with pulses in H₂/CO₂, He/CO₂, CO₂/S_{total} and CO/CO₂ ratios. These temporal variations of magmatic He fractions are indicative of the occurrence of magmatic fluid pulses from magma.

To estimate carbon sources for the Pico do Fogo fumaroles, the following mixing calculations are used (Sano and Marty, 1995):

$$\delta^{13}C_{obs} = \operatorname{Lim} \delta^{13}C_{L} + \operatorname{Morb} \delta^{13}C_{M} + \operatorname{Sed} \delta^{13}C_{S}$$
(4)

$$1/(CO_2/{}^{3}He)_{obs} = Lim/(CO_2/{}^{3}He)_L + Morb/(CO_2/{}^{3}He)_M$$

$$+\text{Sed}/(\text{CO}_2/^3\text{He})_{\text{S}}$$
 (5)

$$M + S + L = 1 \tag{6}$$

where obs, L, M, and S refer to the observed value, marine Limestone (L), MORB (M), and organic Sediments (S), respectively. Fumarole F1 shows: limestone 76–82%, organic Sediments 5–8% and MORB 12–19%; Fumarole F2 displays a

limestone contribution of 78–80%, organic Sediments 5–7% and MORB 12–19% (**Table 2**). The temporal variations of the M/(L + S) ratio are displayed in **Figure 12**. From the first measurement in 2009, fumarole F1 decreases until 2013, followed by a sharp peak at the end of 2013 before falling. F2 also peaks at the end of 2013 and diminishes thereafter. The peaks reflect an increase in the relative proportion of mantle-derived CO₂. The temporal agreement between ³He/⁴He ratios and the proportion of CO₂ of mantle derivation is consistent with release of volatiles from a mantle-derived component during the periods 2008–2009 and 2013–2014.

Proposed Conceptual Model

Several authors have reported that magmas from the 1951, 1995 and 2014-2015 eruptions of Pico do Fogo were stored prior to eruption within the uppermost mantle at 15-30 km depth and stalled during ascent within the lower crust at 8-13 km depth (Munhá et al., 1997; Hildner et al., 2011, 2012; Mata et al., 2017; Klügel et al., 2020). No evidence was reported for a very shallow crustal magma reservoir during these eruptions. Klügel et al. (2020) proposed a model for the storage and transport of magma erupted during the 2014-2015 eruption of Pico do Fogo volcano. The main magma storage reservoir was within the uppermost mantle, at ~24-30 km depth. The arrival of fresh and volatile-rich magma to this zone probably occurred during the onset of the first magmatic reactivation observed in 2008-2009 (Alonso et al., 2021). Around one year before the 2014 eruption, magma ascended to a shallower (~21 km) complex storage region with an interconnected network of magma pockets and dyke/sills (Klügel et al., 2020), and was stored along this zone in the lowermost crust.

Based on the present work and other previous findings (Aiuppa et al., 2020; Klügel et al., 2020) we reconstruct the evolution of the fluids at Pico do Fogo volcano from 2007 to 2016 in a conceptual model of five phases (**Figure 13**):

- Phase 1: April 2007–November 2008 (Figure 13A). This period corresponds to an inter-eruptive period. Fumarolic discharges were characterized by moderate ratios in the main volcanic gases, as well as a low steam/gas ratio and a relatively high magmatic He component. In general, the composition of Pico do Fogo fumarolic gas discharges were CO₂-richer with lower H₂O contents. This coincidence is supports the hypothesis that CO₂ flushing is responsible of H₂O loss in different magma pockets (Klüger et al., 2020). The source of magmatic gas at Pico do Fogo is located in the magma reservoir at the uppermost mantle at 24–30 km depth (Klüger et al., 2020).
- Phase 2: November 2008–2010 (Figure 13B). During this period, the chemical and isotopic composition of fumarolic discharges showed significant changes. In November 2008, the deep magmatic environment became slightly more oxidizing and a decrease in the equilibrium temperature with a minimum in 2010 was observed. This was probably due to injections of magmatic CO₂ into the Fogo magmatic-hydrothermal system that caused decreases in the fumarole CO/CO₂ ratio. In this period, a significant increase of ³He/⁴He occurred with a maximum value registered for F1 in 2010. Maximum values of magmatic He component were also observed for F1 and F2 fumarolic discharges. These changes were likely due to an increase of volatile pressure in the magmatic-hydrothermal

system. The geochemical data suggest the occurrence of a first magmatic reactivation due possibly to a deep magma intrusion with temporary stalling in a zone of magma accumulation at \sim 12–13 km depth, near the Moho, which is described by Klügel et al. (2015), such as the magma accumulation zone.

- Phase 3: 2011–2012 (Figure 13C). During this period, geochemical parameters generally returned to levels of Phase 1, which suggests that the intrusion of magma of Phase 2 ended. 48 days before a seismic event of magnitude 2.9 occurred in April 2012 (Pérez et al., 2015) an increase in the HCl/CO₂ and CO₂/S_{total} ratios were observed.
- Phase 4: 2014–June 2015 (Figure 13D). During this period, significant changes were recorded in the chemical and isotopic composition of the fumarolic discharges. Maximum values of H2/CO2, He/CO2, CO2/Stotal and CO/CO₂ ratios were observed in 2013–2014. These peaks are coincident with maximum values of ³He/⁴He and magmatic He component in both fumaroles. These changes were likely due to an increase of volatile pressure in the magmatic-hydrothermal system of Pico do Fogo volcano. The composition of Pico do Fogo fumarolic gas discharges were again CO₂-richer with lower H₂O contents. These changes were likely due to a new magmatic volatile injection and heat transfer in the magmatic-hydrothermal system as is suggested by other geochemical and geophysical signals observed during the same period (González et al., 2015; Cappello et al., 2016; Richter et al., 2016; Calvari et al., 2018; Klügel et al., 2020; Alonso et al., 2021). During the rise of magma to the surface, the flowing magma passed through а complex storage region characterized by an interconnected network of dikes/sills, where part of the magma passed through and part was stored (Klügel et al., 2020). The magma was again accumulated in the area near the Moho (~12-13 km depth). After breaking through the storage region, the magma rose rapidly through the crust and reached the surface on November 23, 2014 (Klügel et al., 2020). The presumably thin conduit and distance of ~2.5 km between the vent and the geochemical sample sites explains why close temporal trends are not observed during the magma's final ascent (days-weeks), and geochemical indices. Those indices are reflecting the broader scale and longer-term influences upon the hydrothermal-magmatic system as a whole rather than pre- or syn-eruptive release of gas closer to the eruption site.
- Phase 5: After the eruptive period and from July 2015 (**Figure 13E**), the situation seems to have reset. The steam/gas ratio, CH_4/CO_2 , H_2/CO_2 , He/CO_2 ratios, δC - CO_2 , ${}^{3}\text{He}/{}^{4}\text{He}$ and magmatic He fraction decreased quite dramatically. These drastic changes are attributed to efficient degassing of the magma responsible for the eruption.

CONCLUSION

The changes recorded in the chemical and isotopic composition of the fumarolic discharges of the Pico do Fogo volcano during the period 2007–2016 provide information of great value to understand the dynamics of this volcano and the relationship between these changes and the eruptive activity. The injection of hot magmatic fluids from rich-gas magma into the volcanic-hydrothermal system is reflected in the CO_2 , CH_4 , He, CO and H_2 contents, as well as the He and C isotopic magmatic signatures, exiting from the fumaroles.

The close relationship observed between the geochemical parameters studied in visible emissions (fumaroles) demonstrates the convenience of enhancing the geochemical program for volcanic monitoring of fumaroles as an important complement to understanding the evolution of volcanic activity in Pico do Fogo. Increasing the sampling frequency, at least every six months, would be of great importance to improve the surveillance of this volcano.

DATA AVAILABILITY STATEMENT

The raw data supporting the conclusions of this article will be made available by the authors, without undue reservation.

AUTHOR CONTRIBUTIONS

GM participated in sampling, chemical and isotopic analysis of samples, treatment of data and manuscript writing. PH, EP, and NP participated in sampling, treatment of data and manuscript writing. HS performed isotopic analysis of samples, treatment of data and manuscript writing. MA-R and CA contributed chemical and isotopic analysis of samples. MA, GP, JB, FR, SS,

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JP, FS, and NC participated in the collection of gas samples. MA and FS also participated in the chemical analysis of samples.

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