



## Mineralization of the Tuwu Porphyry Cu Deposit in Eastern Tianshan, NW China: Insights From *In Situ* Trace Elements of Chlorite and Pyrite

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#### **OPEN ACCESS**

#### Edited by:

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#### Reviewed by:

Wei Zhang, China University of Geosciences Wuhan, China Saeed Alirezaei, Shahid Beheshti University, Iran

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#### Specialty section:

This article was submitted to Economic Geology, a section of the journal Frontiers in Earth Science

Received: 31 December 2020 Accepted: 10 May 2021 Published: 16 June 2021

#### Citation:

Tan W, Mao Q, Yu M, Sun Y and Lv X (2021) Mineralization of the Tuwu Porphyry Cu Deposit in Eastern Tianshan, NW China: Insights From In Situ Trace Elements of Chlorite and Pyrite. Front. Earth Sci. 9:648177. doi: 10.3389/feart.2021.648177 The Tuwu porphyry copper deposit is located on the Dananhu-Haerlik island arc in eastern Tianshan, NW China. Based on geology, petrology, and in situ trace element studies of pyrite and chlorite, we redefined the characteristics of hydrothermal fluids and the following three mineralization stages: premineralization stage (stage I), porphyry metallogenic stage (stage II), and superimposed transformation stage (stage III). Pyrite stage I (Py-I) has the highest Co/Ni ratios, and the precipitation crystallization of chlorite (Chl-I<sub>2</sub>) has the similar rare earth element distribution patterns with those of volcanic rocks Carboniferous Qieshan (CQ), indicating intense volcanic hydrothermal activity. The Co/Ni ratios of Py-II<sub>1</sub> and Pv-II<sub>2</sub> (stage II) tend to decrease over time. Moreover, the rare earth element (REE) distribution patterns of Chl-II have similar LREE enrichment, and the Eu anomalies in Chl-II1, Chl-II2, and Chl-II3 range from positive to negative. The initial ore-forming fluid was mainly magmatic hydrothermal fluid, and with the late-stage addition of meteoric water and continuous sulfide precipitation, the trace element composition of the fluid changed, and the whole system became more oxidizing. Py-III (stage III) has the lowest Co/Ni ratios, and the REE distribution pattern of ChI-III is characterized by LREE enrichment. Moreover, the Chl-III shows obvious shear deformation characteristics. The results indicate that the host rocks experienced intensely superimposed reformation. By combining and integrating our results with the regional evolution processes in the eastern Tianshan, we propose that the Tuwu porphyry deposit has undergone magmatic hydrothermal and metamorphic hydrothermal processes. Volcanism (stage I) provided the space and initial conditions for the emplacement of the metallogenic body. With the emplacement of the plagiogranite porphyry (stage II), the main copper mineralization occurred in the porphyry and surrounding rocks. After porphyry mineralization (stage III), regional ductile shearing and collisional compression led to a copper reaction, and its accumulation along the faults formed an ore shoot.

Keywords: porphyry deposits, mineralization, pyrite, chlorite, LA-ICP-MS

## INTRODUCTION

The eastern Tianshan orogen, which represents an important part of the tectonic evolution of the Central Asian orogenic belt, has undergone a complicated evolutionary process of stretching, multiblock merging, subduction accretion, collisional orogenesis, and postcollision processes (Figure 1A) (Chen et al., 2007; 2012; Pirajno, 2009, 2013; Zhai et al., 2011). Along with the multiple processes of continental accretion, crust-mantle interactions, and transformation in the eastern Tianshan orogen, numerous polygenetic mineral deposits have been formed, and most have been superimposed by late complex ore-forming processes. These deposits include the Honghai (Huangtupo) Cu-Zn deposit (Deng et al., 2016; Mao et al., 2019), Xiaorequanzi Cu-Zn deposit (Long et al., 2019; Mao et al., 2020), Yamaisu Cu-Fe deposit, Hongyuntan Cu-Fe deposit (Sun et al., 2017; Han et al., 2018; Zhang et al., 2018), and Tuwu-Yandong Cu deposit (Wang Y. F. et al., 2016; Xiao et al., 2017). Research on superimposed metallogenic systems is very important for recognizing the regional metallogenic history and guiding mineral exploration activities in the eastern Tianshan.

The Tuwu porphyry copper deposit is located at the southern part of the Dananhu-Haerlik island arc belt (Figure 1B). Since the Tuwu copper deposit was discovered by the No. 1 geological party of the Xinjiang Bureau of Geological Exploration in 1994, the Yandong, Fuxing, Yanxi, Linglong, and Chihu copper deposits were consecutively discovered afterward. All of these deposits formed the Tuwu-Yandong porphyry copper metallogenic belt, which has attracted extensive attention. Studies have focused on its geological characteristics (Wang et al., 2001; Rui et al., 2001, 2002; Han et al., 2006; Shen et al., 2014b), metallogenic background, metallogenic characteristics, zoning classification (Shen et al., 2012; Pan et al., 2013; Xiao et al., 2018), geochemistry, geochronology (Rui et al., 2002; Zhang et al., 2006; Han et al., 2014; Shen et al., 2014a; Wang et al., 2014), metallogenic chronology (Rui et al., 2002; Wang et al., 2014; Xiao et al., 2017), and compositions and characteristics of oreforming fluids (Han et al., 2006; Gao et al., 2015; Wang et al., 2015, 2017). We have found that there are many differences between the porphyry metallogenic system of the Tuwu-Yandong deposit and the classical porphyry metallogenic model. For example, 1) the boundaries of the potassic zone are not obvious, 2) the morphology of the ore body is tabular rather than equiaxial, and 3) the temperature of the ore-forming fluid is obviously lower than that of a typical porphyry copper deposit. These findings indicate that the Cenozoic porphyry mineralization model may not be strictly suitable for Paleozoic porphyry deposits which experienced a multistage structural evolution.

Mineral microanalysis is an important method for defining ore-forming fluid characteristics, metallogenic physicochemical conditions, and ore-forming element migration and enrichment processes (Large et al., 2009; Zhou, et al., 2010; Ye et al., 2011; Zhang, 2011; Reich et al., 2013; Franchini et al., 2015; Zhang et al., 2020). The development of laser ablation inductively coupled plasma mass spectrometry (LA-ICP-MS) analysis technology has allowed for the measurement of most trace elements in sulfide and oxides. Pyrite is an important sulfide in many metallogenic systems, and its chemical composition is often used to trace the genesis of ore deposits and the metallogenic environment and conditions (Cook et al., 2009; Large et al., 2009, 2011; Reich et al., 2013; Franchini et al., 2015). Chlorite is a mineral that commonly forms during diagenesis, medium- to lowtemperature metamorphism and hydrothermal alterations. Due to the variability in its structure and the nonstoichiometry of its composition, chlorite and its paragenetic mineral assemblages are often used to constrain the physicochemical conditions and determine the diagenetic and metallogenic environment (Davidson et al., 2001; Bortnikov et al., 2008; Inoue et al., 2010; Fadda et al., 2012; Wilkinson et al., 2015; David et al., 2016). This study used LA-ICP-MS to analyze the composition of trace elements in pyrite and chlorite in the main alteration zone of the Tuwu copper deposit, and the combination of detailed geological surveys and petrographical studies was used to study the genesis of ore minerals and the characteristics of trace elements and further reveal characteristics of hydrothermal fluids and mineralization processes of the Tuwu porphyry deposit.

### **REGIONAL GEOLOGICAL BACKGROUND**

A review of the regional geology and the tectonic evolution processes of the accretionary orogenic belt from the Precambrian to Triassic are given in Xiao et al. (2004). From north to south, eastern Tianshan consists of the Dananhu-Haerlik arc, Kangguer Forearc, Yamansu arc, Central Tianshan arc, and the South Tianshan (**Figure 1B**).

The Dananhu-Haerlik Ordovician-Carboniferous island arc, as an important Ordovician to Permian metallogenic belt located between the Kalameili and the Kangguer faults (Figure 1B), includes the Paleozoic Dananhu-Haerlik arc, the Bogda Late Paleozoic intra-arc basin, and the Kanggur forearc (Xiao et al., 2004, 2015). The Dananhu-Haerlik arc consists of Ordovician to Permian tholeiite to calc-alkaline mafic-felsic volcanic rock, volcanoclastics, tuff, and flysch sediments (Ma et al., 1997; Li et al., 2004; Xiao et al., 2004; Hou et al., 2005; Tang et al., 2006; Mao et al., 2014b, Mao et al., 2016; Deng et al., 2016; Li et al., 2016). The Ordovician to Silurian marine volcanic and volcaniclastic rocks consist of tholeiite to calc-alkaline basalts, andesite-basalts, andesite, high-Mg andesite, and volcaniclastic rocks (Ma et al., 1997; Mao et al., 2014b, 2016; Deng et al., 2016; Li et al., 2016). The Devonian to Carboniferous rocks mainly consists of tholeiite to calc-alkaline basalts, andesite, dacite rhyolite, volcaniclastic rocks, and sedimentary (Ma et al., 1997; Li et al., 2004; Xiao et al., 2004; Hou et al., 2005; Tang et al., 2006; Mao et al., 2014b). The arc-related granitic intrusions are formed from Ordovician to Permian (Song et al., 2002; Li et al., 2004; Chen et al., 2005; Hou et al., 2005; Sun et al., 2005; Wu et al., 2006; Chao et al., 2006; Guo et al., 2006; Mao et al., 2010, 2019; Long et al., 2016; Zhang et al., 2017).

The Bogda intra-arc basin, which formed within the Dananhu-Haerlik arc from the Carboniferous to Early



FIGURE 1 | (A) Schematic tectonic map of Central Asia (Şengör et al., 1993; Xiao et al., 2015) showing the position of the eastern Tianshan. (B) Schematic geological map of the eastern Tianshan [modified after Mao et al. (2017) and Xiao et al. (2004)] showing the location of the Tuwu area. The major faults separating the southern Tianshan, Central Tianshan arc, Yamansu arc, and Dananhu-Haerlik arc.

Permian, is composed of basalt, dacite, rhyolite, volcanoclastics, minor interbedded andesite, and sedimentary rocks. The mafic to granitic intrusions are mainly Early Permian (Gu et al., 2001a, Gu et al., 2001b; Li et al., 2004; Xiao et al., 2004; Li et al., 2006; Wang et al., 2010; Shu et al., 2011; Chen et al., 2013; Gao et al., 2014).

The Kanggur forearc located at the south margin of the Dananhu-Haerlik arc is made up of structurally dismembered Devonian-Carboniferous volcanic-sedimentary rocks and ophiolitic slices (Yang and Scott, 1996; Ma et al., 1997; Zhou et al., 2001; Xiao et al., 2004). The Permian granitoids are extensive, and a zone of earlier Permian mafic-ultramafic complexes is located along the Kangguer forearc terrane and stretches several hundred kilometers (Ma et al., 1997; Mao et al., 2002; Han C. M. et al., 2004; Xiao et al., 2004; Zhou et al., 2004; Qin et al., 2011). Multiple types of mineral deposits developed in

the Dananhu-Haerlik arc, including porphyry Cu (Au/Mo) deposits (Rui et al., 2002; Chen et al., 2005; Wu et al., 2006; Zhang et al., 2006; Shen et al., 2014a; Wang et al., 2015; Mao et al., 2018), hydrothermal vein Cu (Au) deposits (Fang et al., 2002; Mao et al., 2014a; Yu et al., 2019), Cu-Ni sulfide deposits (Mao et al., 2002; Han B. F. et al., 2004; Xiao et al., 2004; Zhou et al., 2004; Mao et al., 2006; Han et al., 2010; Qin et al., 2011), VMS Cu-Zn-(Au/Ag) polymetallic deposits (Mao et al., 2010, 2019; Deng et al., 2016; Li et al., 2002), and Au deposits (Wang et al., 2006).

The Dananhu-Haerlik island arc, situated north of the Kanggur fault, comprises Paleozoic volcano-sedimentary rocks and Mesozoic to Cenozoic sedimentary rocks, intercalated with volcanic and pyroclastic rocks (Xiao et al., 2004; Hou et al., 2006; Qin et al., 2011). Paleozoic mafic-ultramafic complexes and granites intruded the sedimentary and volcanic rocks and are

associated with some Cu and Au mineralization (Deng et al., 2016; Gao et al., 2015). The Carboniferous rocks are mostly foliated and mylonitized and influenced by ductile shearing processes. Carboniferous acidic igneous rocks and a Permian mafic–ultramafic complex are well developed in this belt and mainly occur in the form of stocks (Han et al., 2006; Pan et al., 2013; Shen et al., 2014b). Late Paleozoic mineral deposits are developed at the southern zone of the Dananhu-Haerlik island arc. These deposits include shear zone-type gold deposits, such as the Tianmu Au deposit and Hongtan Au deposit (Yao, 2006), Cu-Ni sulfide deposits, which are associated with Permian mafic-ultramafic rocks (Wang Y. L. et al., 2016), and Carboniferous-Permian porphyry Cu (Mo) deposits, such as the Tuwu-Yandong Cu deposits (Pirajno, 2013; Shen et al., 2014b) and Sanchakou Cu deposit (Qin et al., 2009).

#### **GEOLOGY OF THE TUWU DEPOSITS**

The Tuwu copper deposit is hosted in the Carboniferous Qieshan (CQ) group, diorite porphyrite and plagiogranite porphyry at the northeastern part of the Qieshan anticline (Figure 2A). According to previous studies (Han et al., 2006; Shen et al., 2012; Pan et al., 2013; Shen et al., 2014b), the CQ group can be divided into five lithologic members (Figure 2A). The first lithologic member (CQ<sup>1</sup>) is volcanic effusive facies distributed symmetrically around the volcanic crater. The lithology is mainly basalt, followed by andesite, tuff, dacite, breccia lava, volcanic breccia, and other volcanics. Additionally, dioritic porphyrite, which is the main ore-hosting rock, is located at the contact between the basalt in CQ<sup>1</sup> and the pebble-rich lithic sandstone in  $CQ^3$ . The second lithologic member ( $CQ^2$ ), which is an explosive volcanic facies, is located in the north-central part of the mining area near the center of volcanism. The lithology is agglomerate breccia lava and minor andesite and basalt. The second lithologic group can be divided into three lithologic members. The third lithologic member (CQ<sup>3</sup>) is distributed in a nearly east-west direction in the northern part of the mining area. The lithology is mainly volcanic-sedimentary facies. CQ<sup>3</sup> is a pebble-rich lithic sandstone that contains basalt and andesite layers. The fourth lithologic member (CQ<sup>4</sup>) is distributed in the southern part of the mining area in a nearly east-west direction along the northern side of the Kangguer fault. The lithology is mainly pebble-rich sandstone and minor basalt and volcanic breccia at the top. The fifth lithologic member (CQ<sup>5</sup>) is in angular unconformable contact with the underlying breccia lava of CQ<sup>2</sup>. The lithology is mainly heterolithic conglomerate with lenticular feldspathic lithic sandstone. The volcanic rocks of the CQ Group constitute a relatively complete volcanic eruption cycle and resulted from Carboniferous magmatism in this area. The CQ group strikes nearly east-west and dips to the south at angles of approximately 40–65°. The upper Carboniferous Gandun formation ( $C_2g$ ) is distributed in the southern part of the mining area in a nearly east-west direction along the southern side of the Kangguer fault. The lithology is phyllonite, carbonaceous phyllonite and tuff, which have been altered by regional ductile shear deformation metamorphism. The Jurassic Xishanyao formation (Jx) is covered in cap rock and exposed in the northern part of the mining area. The lithology is mainly sandstone, siltstone, mudstone, and conglomerate.

The magmatism was developed quite intensely in the Tuwu-Yandong ore district. The intrusive rocks mainly include granite, plagiogranite porphyry, diorite and diorite porphyry. The Carboniferous metallogenic plagiogranite porphyry of the Tuwu deposit occurs as a small stock (Cheng et al., 2005; Wang et al., 2014; Xiao et al., 2018). The plagiogranite porphyry cut through the dioritic porphyrite and intrude into the contemporaneous volcanism-related fault with an E-W trend (Pan et al., 2013). The dioritic porphyrite is the main ore-hosting rock and belongs to subvolcanic rocks (Hou et al., 2005, 2006; Zhao et al., 2014). It cuts through the basalt in CQ<sup>1</sup> and the pebble-rich lithic sandstone in CQ<sup>3</sup>, thus forming an east-west-trending stock or dike (Shen et al., 2012) (**Figure 3**).

The Tuwu deposit is located in the northeastern part of the Qieshan anticlinorium (Figure 2A). The direction of fracture development is mainly east-west, followed by northwest-southeast. The copper ore bodies mainly occur in the dilated portion of the east-west fault. The intensity of the ductile deformation zone in the Tuwu-Yandong mining area is stronger in the south and weaker in the north. The southern part of the ore field features a continuous strong ductile zone, whereas the northern part is a weakly ductile deformation zone composed of several discretely distributed small lenticular and zonal ductile deformation zones. The copper ore body is distributed in an en echelon shape both in cross section and in map view. In general, the orecontrolling brittle-ductile shear zone is a ENE-trending tectonic belt that dips southward (Ma et al., 2003). The exploration data show that the schistosity obviously cuts the strata and ore body, indicating that the ore body suffered late brittle-ductile deformation. However, distinguishing radial and annular contemporaneous faults from the effect of brittle-ductile deformation is difficult.

The ore mainly has mediumto fine-grained hypidiomorphic-xenomorphic granular textures (Figure 4; Figure 5). The ore structure mainly includes veins and dissemination. The metallic minerals are mainly chalcopyrite, followed by pyrite and bornite, with minor chalcocite, covellite, magnetite, hematite, and molybdenite. Pyrite is mainly developed at the top and bottom of the ore body and it is negatively related to copper minerals. The gangue minerals are mainly quartz, sericite and chlorite, followed by biotite, epidote, actinolite, gypsum, feldspar, and calcite. The supergene minerals are malachite, atacamite, and limonite.

Propylitization is extensively developed in the volcanicpyroclastic rocks of the CQ Group. According to systematic alteration mapping and a comprehensive laboratory study, we find that weak propylitization related to volcanic hydrothermal activity in the late volcanic activity is concentrated in the range of 1.2–3 km from the ore body while strong propylitization obviously superimposed by late hydrothermal activity is concentrated in the range of 0–1.2 km. The alteration intensity differs significantly between these zones, with the F2 fault as the dividing line. The altered minerals associated with weak



propylitization are chlorite, epidote, quartz, zoisite, calcite, and pyrite, and the rock color is dark green in general (**Figures 4A,F**; **Figure 5A**). Chlorite is mainly in the form of interstitial mafic minerals developed between feldspars. The formation of chlorite is related to the late hydrothermal alteration of dark minerals, such as amphibole. Epidote mainly exists discontinuously in the form of veins in faulted structures. Pyrite grains with particle sizes of 0.01–0.05 nm exhibit a stellate distribution.

The concentric alteration zones from the interior outward in succession are the potassic zone, the quartz–sericite zone, and the propylitic zone.

The potassic zone is mainly distributed in plagiogranite porphyry and dioritic porphyrite. The alteration mineral

Ag	e Lithology	Description		Tectonic setting
Jurassic		sandstone, siltstone, mudstone and coal seam		intraplate
triassic	/	Porphyry Molybdenum Deposit		collision
Permian	_	tectonic altered rock type gold deposit (Late Permian) copper-nickel sulfide deposits(Early Perm	nian)	post-collisional extension
Carboniferous		heterolithic conglomerate with lenticular feldspathic lithic sandstone granite(255Ma) pebble-rich lithic sandstone that contains basalt and andesite layers diorite(308.5Ma),granite(255Ma) agglomerate breccia lava and minor andesite and basalt diorite(308.5Ma) mainly basalt, followed by andesite, tuff, dacite, breccia lava plagiogranite porphyry(334Ma) diorite porphyry(335Ma)	strong volcanism weak	collision(arc)
		phyllonite, carbonaceous phyllonite and tuff		collision (forearc basin)
tratigraphic	chart of geological events	s in Tuwu copper ore district.		

assemblage of disseminated and vein types includes potassium feldspar, biotite, and magnetite. The fine-grained and scaly biotite originated from biotitization of hornblende and recrystallization of primary biotite. The hydrothermal biotite is poorly preserved and often altered into chlorite, ilmenite, and rutile. Identifying the potassic zone is difficult because it has been altered by later hydrothermal activity. Consequently, the boundary of the potassic zone is not clear. Pyritization, sericitization, and silicification are well developed in plagiogranite porphyry (Figures 4C,H). Plagioclase has been almost completely altered to sericite and fine-grained quartz. Pyrite is also widespread (Figures 5B,C). The alteration assemblage is mainly sericite, pyrite, quartz, chalcopyrite, and calcite with net veins and disseminated forms. The dioritic porphyrite is locally developed and sericitized. Plagioclase has been altered to sericite and minor epidote and albite. The alteration assemblage is sericite, chlorite, pyrite, epidote, and quartz (Figures 4D,I,J; Figures 5D-F).

The strongly propylitized rocks are bright green in color and exhibit strong silicification. The alteration assemblage is mainly epidote, chlorite, quartz, calcite, and pyrite. Quartz, epidote, and chlorite are mainly distributed in dense disseminated and vein forms. The disseminated quartz is mainly associated with propylitic alteration, while the vein quartz that appears to have a hydrothermal filling genesis is an obviously younger overprinting feature. Radial and granular epidote aggregates can be seen in some veins. The basalt near the ore is altered and contains minor pyrite, epidote, chlorite, quartz, calcite, and chalcopyrite veins (**Figures 4B,G**).

Alterations related to ductile shear deformation are developed in the local area and characterized by intense sericitization and chloritization. The altered minerals show obvious crumpled and directional arrangement. The rocks have augen and cataclastic structures. The plagioclase in the dioritic porphyrite has been altered to scaly sericite and fine-grained quartz. Chlorite is present in the form of blocks and stockwork and well



FIGURE 4 | Photomicrographs of the chlorite and specimens of typical ore samples from different mineralization stages of Tuwu porphyry Cu deposit in eastern Tianshan, Xinjiang, NW China. (A) Weakly propylitized basalt; (B) strongly propylitized basalt; (C) pyrite + chalcopyrite + chlorite + sericite in diorite porphyrite; and (D) pyrite + chalcopyrite + silicification development in plagiogranite porphyry; (E) mylonitized diorite porphyrite; (f) weak chlorite development in basalt, Chl-II; (G) strong epidote and chlorite development in basalt, Chl-II1; (H) pyritization, sericitization, silicification, and chloritization of plagiogranite porphyry, Chl-II2; (I) sericite, silicification, and chlorite development in plagiogranite porphyry, Chl-II2; (J) chlorite, epidote, pyrite, and calcite development in plagiogranite porphyry, Chl-II2; (K) chlorite, chalcopyrite, and sericite in diorite porphyry (chlorite, sericite, and pyrite grains have an oriented arrangement) Chl-III; and (L) chlorite, chalcopyrite, and sericite in diorite porphyry rock with an augen structure, Chl-III. Ccp, chalcopyrite; Py, pyrite; PI, plagioclase; Chl, chlorite; Ep, epidote; Qtz, quartz; Cal, calcite; Ser, sericite.

developed in the local ductile shear zone (**Figures 4E,K,L**). The altered mineral assemblage is sericite, chlorite, and quartz. The metallic minerals are mainly chalcopyrite and pyrite, which are distributed in disseminated and vein types (**Figures 5G,H,I**). The alteration zone is related to the rich ore body in this area.

The main supergene alteration mineral assemblage is kaolinite, atacamite, malachite, and hematite.

Previous researchers have conducted a large amount of research on the alteration zonation and metallogenic timing of the Tuwu copper deposit based on the association of alteration minerals, mineral relationships, and crosscutting relationships (Shen et al., 2012; Pan et al., 2013; Wang Y. F. et al., 2016). After extensive research on the petrography of the main rock types exposed in the Tuwu mining area (**Figures 2B,C**), this study identified four stages of hydrothermal activity, namely, alterations related to volcanic activity (stage I), alterations related to porphyry mineralization (stage II), alterations related to supergenesis (stage IV) (**Figure 6**).



FIGURE 5 | Photomicrographs of pyrite from different mineralization stages in the Tuwu porphyry Cu deposit in eastern Tianshan, Xinjiang, NW China. (A) Disseminated pyrite in basalt with weak propylitization, Py-I; (B) Pyrite associated with chalcopyrite in plagiogranite porphyry, Py-II<sub>1</sub>; (C) pyrite growing along chalcopyrite fissure in plagiogranite porphyry, Py-II<sub>1</sub>; (D) pyrite associated with magnetite in plagiogranite porphyry, Py-II<sub>2</sub>; (E) veinlets of pyrite, chalcopyrite, and silicification, in plagiogranite porphyry, Py-II<sub>2</sub>; (F) chalcopyrite growing along pyrite fissure in plagiogranite porphyry in diorite porphyry, Py-III; (G) chlorite, sericite, and pyrite in diorite porphyrite, pyrite with crushed structure, Py-III; (H) sericite and pyrite in plagiogranite porphyry, pyrite with crushed structure, Py-III; and (I) chlorite and sericite in diorite porphyrite with pyrite and chalcopyrite vein. Ccp, chalcopyrite; Py, pyrite; Mag, magnetite; Qtz, quartz; Cal, calcite; Ser, sericite.

### SAMPLING AND ANALYSIS METHODS

This study carried out detailed geological observations and profile surveys of outcrops, troughs, open mining pits and drilling samples from the Tuwu mining area. Typical samples of various alteration zones were collected from the area, and thin sections with a thickness of 0.8 mm were prepared. Two to three typical samples were selected from each alteration zone, and 19 samples were selected for testing and analysis. The mineral types of the tested samples are mainly pyrite Py-I, chlorite Chl-I<sub>1</sub> (metasomatic crystallization), and Chl-I<sub>2</sub> (precipitation crystallization) related to volcanic activity, pyrite Py-II, and chlorite Chl-II related to porphyry mineralization, pyrite Py-III, and chlorite Chl-III related to ductile shear deformation.

The *in situ* analysis of chlorite composition was completed in the LA-ICP-MS laboratory of the State Key Laboratory of Ore Deposit Geochemistry, Institute of Geochemistry, Chinese Academy of Sciences. The LA-ICP-MS instrument is composed of an Agilent 7700cs quadrupole ICP-MS instrument connected to a GeoLasPro 193 nm LA system. The operating conditions of the instrument were as follows: the laser beam width was 44 nm, the frequency of the laser pulse was 6 Hz, and the pulse energy was 0.032-0.105. The analysis time was 90 s (30 s for the blank test and 60 s for the sample analysis). During testing, the LA spot coincided with the electron microprobe analysis spot. The internal standards NIST610 and NIST612 were analyzed once every nine chlorite samples to monitor the instrument deviation. The international standard samples ML3B-G and BCR-2G were used as unknown samples to monitor the data quality. Data processing was calibrated by Igor-pro software. The Si content was analyzed by electron microprobe as the internal standard to calibrate the Si content determined by LA-ICP-MS. The analysis results of GOR128-G, BIR-1G, BHVO-2G, ML3B-G, and BCR-2G were in good agreement with their reference values. The analytical error of most elements was less than 10%, although Cr, Y, and Zr had errors of 10-15%.



The analysis of pyrite was carried out in the State Key Laboratory of Ore Deposit Geochemistry, Institute of Geochemistry, Chinese Academy of Sciences. The standard sample of the LA system was NWR-213, and the experimental analysis instrument was an Agilent 7700x type quadrupole mass spectrometer. Helium (480 ml/min) was used as a transport medium for the ablated material, and it was mixed evenly with argon (900 ml/min) after leaving the ablation compartment. The mixture then entered the ICP-MS instrument, where the element contents were measured. The analysis time was 90 s (30 s for blank test and 60 s for sample analysis). The laser beam width was 44 nm, the frequency of the laser pulse was 10 Hz, and the pulse energy was 5 J/cm<sup>2</sup>. The analysis method was a multiple external standard–no internal standard method, and the external standards were GSE-1G, MASS, and Py. Data processing was performed with ICPMSDataCal software.

### **TEST RESULTS**

#### In Situ Trace Element Composition of Pyrite

The results of the *in situ* trace element composition of pyrite in the Tuwu copper mining area are shown in **Table 1**. Overall, the

contents of chalcophile elements (Cu, Zn, Pb, As, Se, Te, Ge, and Ag), siderophile elements (Cr, Co, and Ni), V, and Ba are all above the limit of detection. The contents of Sn, Cd, Sb, Au, Bi, and other elements were mostly lower than or near the detection limit (**Figures 7**, **8**).

The results showed that Py-I has the highest total amount of trace elements (average 4,351 ppm) and highest Co/Ni ratio (average 33). Py-I is rich in Co, Ni, and As and poor in Cu and Pb, and the Zn, Au, and Ag contents are below or near the detection limit. The total amount of trace elements of Py-II<sub>1</sub> (average 4,049 ppm) and Py-II<sub>2</sub> (average 544 ppm) displayed an obvious decrease with the occurrence of porphyry mineralization. Moreover, the Co/Ni ratio of Py-II<sub>1</sub> (average 3) and Py-II<sub>2</sub> (average 0.53) also showed a certain decrease. Py-II<sub>1</sub> and Py-II<sub>2</sub> are rich in Co, Ni, and Cu, and poor in As, Pb, Zn, Au, and Ag. The total amount of trace elements in Py-III ranges from 133 to 6,401 ppm (average 1982 ppm). Py-III has the lowest Co/Ni ratio (average 0.16). This phase features obviously higher mineralization element contents than Py-I, Py-II<sub>1</sub>, and Py-II<sub>2</sub>, and the contents of elements, such as Co, Ni, Cu, Pb, As, Pb, Zn, Au, and Ag, have extremely uneven distributions (Table 1).

Trace	Py-I	Py-II₁	Py-II <sub>2</sub>	Py-III n = 6 (total)	
element (ppm)	<i>n</i> = 3 (total)	<i>n</i> = 11 (total)	<i>n</i> = 10 (total)		
V	0.68–1.11; Av = 0.87 (n = 3)	0.06–32.17; Av = 7.34 (n = 11)	0.03–0.29; Av = 0.09 ( <i>n</i> = 10)	Mostly bdl; 0.53–0.53; Av = 0.53	
Cr	3.02–9.07; Av = 5.65 (n = 3)	1.21–27.41; Av = 7.28 (n = 11)	1.08–18.45; Av = 6.35 (n = 10)	2.25–20.96; Av = 7.56	
Co	288.5–6,417.17; Av = 3,027.09 (n = 3)	115.7–6,339.2; Av = 2,260.87 (n = 11)	5.73–318.06; Av = 91.40 (n = 10)	0.65–415.9; Av = 75.21	
Ni	25.38–105.89; Av = 71.9 (n = 3)	27.58–3,396.17; Av = 1,635.83 (n = 11)	11.19–1,623.02; Av = 325.23 (n = 10)	47.38–1,298.36; Av = 509.45	
Cu	8.95–50.68; Av = 23.16 (n = 3)	0.66–336.25; Av = 86.28 (n = 10)	2.96–43.6; Av = 16.38 (n = 10)	5.94–4,133.8; Av = 950.33	
Zn	0.1–0.7; Av = 0.4 (n = 2)	0.13–10.42; Av = 3.13 (n = 6)	0.29–2.61; Av = 1.17 (n = 8)	Mostly bdl; 8.84–8.84; Av = 8.84	
Ge	7.87–11.69; Av = 9.15 (n = 3)	6.75–12.88; Av = 9.28 (n = 11)	7.23–13.85; Av = 9.24 (n = 10)	6.17–10.85; Av = 8.38	
As	50.73–2,933.98; Av = 1,072.03 (n = 3)	0.27–21.14; Av = 5.03 (n = 10)	0.01–58.85; Av = 16.95 (n = 8)	10.1–1817.01; Av = 913.56	
Se	87.3–167.07; Av = 133.16 (n = 3)	21.15–82.59; Av = 39.62 (n = 11)	41.95–174.06; Av = 71.49 (n = 10)	19.65–137.42; Av = 68.10	
Ag	0.02–0.16; Av = 0.11 (n = 3)	0.02–0.57; Av = 0.13 (n = 9)	0.02–1.4; Av = 0.37 (n = 9)	0.02–20.22; Av = 5.21	
Cd	0.22–0.33; Av = 0.27 (n = 3)	0.1–0.44; Av = 0.24 (n = 6)	0.01–0.49; Av = 0.17 (n = 7)	0.39–1.41; Av = 0.69	
Sn	bdl	0.01–3.63; Av = 0.75 (n = 8)	0.01–0.49; Av = 0.26 (n = 6)	0.12–1.07; Av = 0.46	
Sb	0.22–0.22; Av = 0.22 (n = 2)	0.01–0.37; Av = 0.09 (n = 6)	0.06–4.8; Av = 0.73 (n = 8)	0.08–14.34; Av = 3.66	
Те	0.43–8.24; Av = 3.42 (n = 3)	0.02–0.74; Av = 0.42 (n = 8)	0.01–3.05; Av = 0.68 (n = 8)	0.61–12.15; Av = 5.57	
Ba	0.05–0.42; Av = 0.19 (n = 3)	0.03–8.12; Av = 2.18 (n = 8)	0.06–0.91; Av = 0.25 (n = 9)	0.15–0.3; Av = 0.21	
Au	0.03–0.03; Av = 0.03 (n = 2)	0.01–0.04; Av = 0.02 (n = 7)	0.01–0.1; Av = 0.03 (n = 7)	0.01–4.91; Av = 1.24	
Pb	1.93–6.86; Av = 3.93 (n = 3)	0.02–3.41; Av = 1.16 (n = 11)	0.04–27.01; Av = 5.99 (n = 10)	0.66–164.32; Av = 48.61	
Bi	0.09–0.19; Av = 0.12 (n = 3)	0.02–0.79; Av = 0.34 (n = 8)	0.01–6.2; Av = 1.28 (n = 8)	0.01–13.82; Av = 3.39	
Co/Ni	11.37–60.60; Av = 33.37 (n = 3)	0.09–9.94; Av = 2.71 (n = 11)	0.12–2.02; Av = 0.53 (n = 10)	0.003–0.57; Av = 0.16	
Total content	478.09–9,692.23; Av = 4,351.34	594-7,839.08; Av = 4,049.19	125.64–2028.18; Av = 543.68	132.52–6,401.02; Av =	
	(n = 3)	( <i>n</i> = 11)	(n = 10)	1982.41	

TABLE 1 | In situ trace element compositions of pyrite for different metallogenic stages of the Tuwu porphyry Cu deposit in eastern Tianshan, Xinjiang, NW China (ug/g).

Note: Av stands for average and bdl stands for below the detection limit.



# *In Situ* Trace Element Composition of Chlorite

The results of the *in situ* trace element composition of chlorite in the Tuwu copper mining area are shown in **Table 2**. To ensure that the mineral in the experiment is chlorite, the chlorite are filtered based on the standard of  $\omega$  (Na<sub>2</sub>O + K<sub>2</sub>O + CaO) < 0.5%

(Foster, 1962; Zang and Fyfe, 1995). According to the Fe-Si classification of chlorite, the chlorite samples are mainly clinochlore and pyrochlore (unpublished data).

In this study, the data show that the total rare earth element (REE) content of chlorite from different stages is above the detection limit, thus, REEs are selected for research. The content of REEs in Chl-I<sub>1</sub> ranges from 0.03 to 17.62 ppm, and the REE patterns are characterized by obvious enrichment of light REEs over heavy REEs (LREE/HREE = 0.12–5.95), negative Eu anomalies ( $\delta$ Eu = 0.223–0.866), and no Ce anomalies ( $\delta$ Ce = 0.53–2.11). The results of (La/Sm)n (0.14–1.33) and (Gd/Yb)n (0.95–4.70) indicate that the internal fractionation of LREEs and HREEs is weak (**Figure 9A**).

The content of REEs in Chl-I<sub>2</sub> ranges from 0.02 to 0.71 ppm, and the REE patterns are mainly characterized by the enrichment of LREEs (LREE/HREE = 0.18–3.91), positive Eu anomalies ( $\delta$ Eu = 1.0–3.78), and weak Ce anomalies ( $\delta$ Ce = 0.54–2.01). The results of (La/Sm)n (0.07–1.81) and (Gd/Yb)n (0.22–5.61) indicate that the internal fractionation of LREEs and HREEs is relatively weak (**Figure 9B**).

The content of REEs in Chl-II<sub>1</sub> ranges from 0.056 to 0.77 ppm, and the REE patterns are characterized by the relative enrichment of HREEs (LREE/HREE = 0.07–0.59), positive Eu anomalies ( $\delta$ Eu = 1.08–1.62), and weakly negative Ce anomalies ( $\delta$ Ce = 0.43–1.52). The results of (La/Sm) n (0.11–0.497) and (Gd/Yb) n (1.1–12.02) indicate that the internal fractionation of LREEs and HREEs is stronger (**Figure 9C**).



The content of REEs in Chl-II<sub>2</sub> ranges from 0.02 to 3.42ppm, and the REE patterns are characterized by the relative enrichment of HREEs (LREE/HREE = 0.11–1.19), weak positive Eu anomalies ( $\delta$ Eu = 1.02–2.52), and positive Ce anomalies ( $\delta$ Ce = 0.52–1.95). The results of (La/Sm)n (0.07–0.71) and (Gd/Yb)n (0.502–11.60)

indicate that the internal fractionation of LREEs and HREEs is strong (Figure 9D).

The content of REEs in Chl-II<sub>3</sub> ranges from 0.05 to 3.34 ppm, and the REE patterns are characterized by the relative enrichment of HREEs (LREE/HREE = 0.08–0.85), negative Eu to weakly

TABLE 2 | In situ trace element composition and characteristic REE values of chlorite for different metallogenic stages of the Tuwu porphyry Cu deposit in eastern Tianshan, Xinjiang, NW China (µg/g).

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Trace	Chl-l₁	Chl-I <sub>2</sub>	Chl-II <sub>1</sub>	Chl-II <sub>2</sub>	Chl-II <sub>3</sub>	Chl-III	
element	n = 17 (total)	n = 15 (total)	n = 18 (total)	n = 10 (total)	n = 13 (total)	n = 6 (total)	
Na	76.46–1,050.02; Av = 498.1	76.75–1,413.61; Av = 562.06	269.87–1,274.41; Av = 637.36	24.09–1,346.2; Av = 512.28	20.07-864.1; Av = 202.91	42–96.48; Av = 66.36 (n = 6)	
K	(n = 17)	(n = 15)	(n = 18)	(n = 10)	(n = 13)	62 90 492 05 AV 144 47	
n	(p = 17)	(p = 15)	27.0-390.35; AV = 100.10 (p = 19)	12.36-604.36; AV = 196.13	(n - 12)	(n - 6)	
Ca	(7 = 17) 101.20.2.525.45: $\Delta y = 828.2$	(7 = 13) 186 82 2 104 71: Av = 581 74	(7 = 10) 210.25 1.550.54: $\Delta y = 455.25$	(7 = 10)	(7 = 13) 10.25 840.34 Av = 203.10	(7 = 0) 140 552 20: $\Delta y = 210.82$ (p	
0a	(n - 17)	(n - 15)	(n - 18)	(n - 9)	(n - 12)	- 6)	
Ti	(7 - 17) 12 22-159 33: $\Delta y = 29.64$	(7 - 13) 10.96-573.20: $\Delta y = 71.45$	(7 - 10) 11 07-176 09: $\Delta y = 37.54$	(7 - 0) 19-486 8: $\Delta y = 165.92$	(7 - 10) 21 64-1 055 86: $\Delta y = 334.43$	= 0 177 9–572 $4 \cdot \Delta v = 299.68$ (n	
	(n = 17)	(n = 15)	(n = 18)	(n = 10)	(n = 13)	= 6)	
V	(7 - 17) 44 63–210 72: Av = 115 75	35.38 - 188.80 Av = 99.53	3878 - 1816: Av = 10362	28 67–933 29: Av = 229 91	(1 - 10) 127 41–780 29: Av = 363 64	153 74–254 63 <sup>.</sup> Av = 214 70 (n	
•	(n = 17)	(n = 15)	(n = 18)	(n = 10)	(n = 13)	= 6)	
Mn	1946.41–3.466.17: Av =	1.376.12-2.973.11: Av =	2.368.23–3.662.86: Av =	380.31-2048.76: Av =	348.55–2.715.22: Av =	1951.67-2.853.68: Av =	
	2.275.41 (n = 17)	2099.84 (n = 15)	2.711.42 (n = 18)	1.505.32 (n = 10)	1.139.89 (n = 13)	2.247.81 (n = 6)	
Ni	153.95–387.96; Av = 309.24	147.21–397.03; Av = 301.59	251.76–371.36; Av = 329.01	80.73–477.43; Av = 325.56	117.74–382.95; Av = 249.15	262.68–313.44; Av = 284.64	
	(n = 17)	(n = 15)	(n = 18)	(n = 10)	(n = 13)	(n = 6)	
Cu	1.81–345.91; Av = 98.07	3.11–215.25; Av = 106.21	0.02–388.38; Av = 68.54 (n = 18)	0.05–102.99; Av = 37.96	0.01-1.03; Av = $0.288$ (n = 10)	0.04-1.2; Av = $0.64$ (n = 6)	
	(n = 17)	(n = 15)		(n = 7)			
Zn	190.46-411.24; Av = 257.89	185.61–293.06; Av = 256.57	157.33–597.87; Av = 251.98	40.31–395.41; Av = 192.57	57.63–154.24; Av = 106.17	154.04–188.16; Av = 168.21	
	(n = 17)	(n = 15)	( <i>n</i> = 18)	(n = 10)	( <i>n</i> = 13)	(n = 6)	
Ge	0.18–0.76; Av = 0.413 (n = 6)	0.02–1.11; Av = 0.506 (n = 7)	0.02–1.05; Av = 0.576 (n = 15)	2.98–3.66; Av = 3.33 (n = 3)	2.02–4.15; Av = 3 (n = 5)	bdl	
Sr	1.79–29.16; Av = 10.94 (n = 17)	0.86–7.49; Av = 4.79 (n = 15)	1.83–7.1; Av = 4.45 (n = 18)	0.51–58.69; Av = 12.79 (n = 10)	0.37–121.19; Av = 17.18 (n = 13)	0.71–2.02; Av = 1.42 (n = 6)	
Mo	0.01–0.159; Av = 0.088 (n = 16)	0.13–0.023; Av = 0.071 ( <i>n</i> = 15)	0.024–0.13; Av = 0.082 (n = 18)	0.01–0.09; Av = 0.056 (n = 9)	0.012–0.099; Av = 0.045 (n = 12)	0.04–0.34; Av = 0.12 (n = 6)	
Cd	0.053–0.107; Av = 0.062 ( <i>n</i> = 6)	0.02–0.09; Av = 0.052 (n = 5)	0.0064–0.065; Av = 0.037 (n = 14)	0.07–0.11; Av = 0.09 (n = 2)	0.022–0.15; Av = 0.075 (n = 5)	bdl	
In	0.002–0.025; Av = 0.012 ( <i>n</i> = 6)	0.005–0.02; Av = 0.011 (n = 6)	0.0012 - 0.056; Av = 0.022 (n = 15)	0.007–0.013; Av = 0.01 (n = 3)	0.004–0.027; Av = 0.013 (n = 6)	bdl	
Sb	0.002-0.456; Av = 0.085 (n = 14)	0.0004-0.62; Av = 0.193 ( $n = 12$ )	0.001-0.0186; Av = 0.067 ( $n = 14$ )	0.001-0.20; Av = 0.068 ( $n = 10$ )	(0.003-0.0302; Av = 0.072)	0.11–0.25; Av = 0.2 (n = 3)	
La	0.001-2.132 Av = 0.208	0.0009-0.043 Av = 0.012	0.0018-0.014 Av = 0.008	0.001-0.01 Av = 0.008	0.002 - 0.045 Av = 0.019	0 0004–0 0758 <sup>.</sup> Av = 0 03	
20	(n = 16)	(n = 15)	(n = 16)	(n = 10)	(n = 11)	(n = 6)	
Ce	0.003–6.37; Av = 0.566 (n = 17)	0.0008-0.18; Av = $0.037(n = 12)$	0.0038-0.067; Av = $0.027(n = 18)$	0.002–0.1; Av = 0.03 (n = 10)	0.006–0.22; Av = 0.056 (n = 12)	0.011–0.244; Av = 0.099 (n = 6)	
Pr	0.001–0.93: Av = 0.101 (n = 14)	0.0007–0.02: Av = 0.059	0.0008–0.016: Av = 0.007	0.001–0.01: Av = 0.005	0.005–0.03: Av = 0.011	0.001–0.052: Av = 0.022 (n = 5)	
	, , , , , , , , , , , , , , , , , , , ,	(n = 15)	(n = 15)	(n = 10)	(n = 10)		
Nd	0.008–4.56; Av = 0.506 (n = 14)	0.0056-0.048; Av = 0.026 (n = 12)	0.0086–0.11; Av = 0.038 (n = 16)	0.008–0.12; Av = 0.047 (n = 8)	0.008–0.14; Av = 0.07 (n = 10)	0.029–0.351; Av = 0.144 (n = 5)	
Sm	0.007–1.06; Av = 0.155 ( <i>n</i> = 12)	0.0005-0.026; Av = 0.013 (n = 11)	0.0039–0.04; Av = 0.024 (n = 16)	0.006–0.05; Av = 0.02 (n = 8)	0.006–0.1; Av = 0.034 (n = 11)	0.007–0.106; Av = 0.053 (n = 5)	
Eu	0.002–0.085; Av = 0.017 (n = 13)	0.002–0.021; Av = 0.01 (n = 12)	0.0033–0.02; Av = 0.011 ( <i>n</i> = 17)	0.004–0.042; Av = 0.017 (n = 8)	0.002–0.04; Av = 0.015 (n = 11)	0.011–0.047; Av = 0.019 (n = 5)	
Gd	0.007 - 0.847; Av = 0.153 (n = 12)	0.0022 - 0.064; Av = 0.026 ( $n = 15$ )	0.0093–0.084; Av = 0.048 (n = 18)	0.0001-0.125;  Av = 0.045 (n = 7)	0.005-0.18; Av = 0.065 ( $n = 11$ )	0.008–0.1071; Av = 0.048 (n = 5)	
Tb	0.0009-0.12;  Av = 0.017 ( <i>n</i> = 16)	0.0009-0.014; Av = 0.006 (n = 11)	0.0028–0.024; Av = 0.01 ( <i>n</i> = 16)	0.0008-0.045; Av = 0.016 (n = 9)	0.0003-0.04; Av = 0.015 ( <i>n</i> = 11)	0.0001-0.025;  Av = 0.0068 (n = 6)	
						(Continued on following page)	

Trace	Chl-I <sub>1</sub>	Chl-I <sub>2</sub>	Chl-II <sub>1</sub>	Chl-II <sub>2</sub>	Chl-II <sub>3</sub>	ChI-III n = 6 (total)	
element	n = 17 (total)	n = 15 (total)	n = 18 (total)	<i>n</i> = 10 (total)	n = 13 (total)		
Dy	0.0007–0.672; Av = 0.115 (n = 14)	0.0008–0.088; Av = 0.044 (n = 12)	0.039–0.11; Av = 0.078 (n = 17)	0.003–0.454; Av = 0.11 (n = 7)	0.001–0.43; Av = 0.116 (n = 13)	0.016–0.109; Av = 0.054 (n = 50	
Но	0.002-0.139; Av = 0.028	0.0001 - 0.035; Av = 0.01	0.0085 - 0.042; Av = 0.021	0.001-0.134; Av = $0.027$	0.009-0.12; Av = 0.031	0.0076 - 0.033; Av = 0.013	
Er	(7 = 15) 0.002–0.43; Av = 0.085 (n = 16)	(n = 14) 0.0028–0.10; Av = 0.037 (n = 13)	(7 = 16) 0.0058–0.145; Av = 0.071 (n = 18)	(7 = 8) 0.005–0.796; Av = 0.191 (n = 10)	(7 = 13) 0.005–0.69; Av = 0.17 (n = 11)	(n = 5) 0.006–0.069; Av = 0.03 (n = 6)	
Tm	0.0003–0.045; Av = 0.012 (n = 16)	0.0008-0.017;  Av = 0.007 (n = 9)	0.0018–0.015; Av = 0.01 (n = 18)	0.0008-0.13; Av = 0.027 (n = 7)	0.0003–0.124; Av = 0.029 (n = 11)	0.001–0.009; Av = 0.004 (n = 6)	
Yb	0.005–0.25; Av = 0.073 (n = 17)	0.0059–0.21; Av = 0.054 (n = 12)	0.0236–0.269; Av = 0.087 (n = 18)	0.0001–1.175; Av = 0.206 (n = 9)	0.009–1.094; Av = 0.217 (=13)	0.006–0.027; Av = 0.018 (n = 5	
Lu	0.001–0.034; Av = 0.011 (n = 14)	0.0008–0.029; Av = 0.01 (n = 10)	0.0007–0.037; Av = 0.013 (n = 18)	0.002–0.23; Av = 0.048 (n = 7)	0.0009–0.198; Av = 0.042 (n = 12)	0.0003–0.01; Av = 0.0004 (n = 6)	
TI	0.0008-0.011; Av = 0.006 (n = 5)	0.0009–0.01; Av = 0.006 (n = 6)	0.0003-0.0089; Av = $0.0045(n = 10)$	0.005–0.05; Av = 0.02 (n = 3)	0.004–0.125; Av = 0.029 (n = 6)	bdl	
ΣREE	0.03–17.62; Av = 1.8	0.02–0.71; Av = 0.248	0.056–0.77; Av = 0.425	0.02–3.42; Av = 0.66	0.05–3.34; Av = 0.808	0.04–1.27; Av = 0.49	
LREE/HREE	0.12–5.95; Av = 1.32	0.18–3.91; Av = 0.96	0.07–0.59; Av = 0.296	0.11–1.19; Av = 0.64	0.08–0.85; Av = 0.55	1.26–3.17; Av = 1.75	
(La/Sm) n	0.14–1.33; Av = 0.70	0.07–1.81; Av = 0.64	0.11–0.497; Av = 0.268	0.07–0.71; Av = 0.298	0.124–0.78; Av = 0.37	0.2–0.88; Av = 0.52	
(Gd/Yb) n	0.95–4.70; Av = 1.90	0.22–5.61; Av = 2.28	1.1–12.02; Av = 3.33	0.502–11.60; Av = 3.28	0.51–7.38; Av = 2.99	1.8-8.08; Av = 4.84	
(La/Yb) n	0.066–6.16; Av = 1.09	0.09–1.09; Av = 0.31	0.01–0.09; Av = 0.06	0.01–8.11; Av = 1.05	0.02–1.93; Av = 0.26	0.57-12.08; Av = 3.89	
δΕυ	0.223–0.866; Av = 0.55	1–3.78; Av = 1.83	1.08–1.62; Av = 1.15	1.02–2.52; Av = 1.36	0.385–1.49; Av = 0.87	0.8–4.27; Av = 1.61	
δCe	0.53–2.11; Av = 0.915	0.54–2.01; Av = 0.99	0.43–1.52; Av = 1.03	0.52–1.95; Av = 1.24	0.64–1.5; Av = 1.06	0.73–3.92; Av = 1.53	

TABLE 2 | (Continued) In situ trace element composition and characteristic REE values of chlorite for different metallogenic stages of the Tuwu porphyry Cu deposit in eastern Tianshan, Xinjiang, NW China (µg/g).

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Age	Age Permian		Carboniferous					
Tectonic setting Stage Mineral structure		Post-collisional extension						
		Superimposed transformation stage	Porphyry metallogenic stage			Premineralization stage		
		Chl-III (disseminated and vein type with shear deformation characteristics)	Chl-II <sub>1</sub> (dense disseminated and stockwork type)	Chl-Il <sub>2</sub> (disseminated and vein type)	Chl-II <sub>3</sub> (mainly disseminated type)	Chl-i <sub>1</sub> (replacement remnant, pseudomorphic)	Chl-i <sub>2</sub> (disseminated type)	
		Py-III (disseminated and vein type with fragmentized)	Spotted (fine- grained)	Py-II <sub>1</sub> (disseminated and vein type)	Py-II <sub>2</sub> (disseminated and vein type)	Py-I (dissemina	ited type)	
Pyrite	Co/Ni	0.003–0.57; Av = 0.16	-	0.09–9.94; Av = 2.71	0.12-2.02; Av = 0.53	11.37–60.60; Av = 33.37		
(ppm)	Co + Ni	313.88–6,523.06;	_	143.28–9,735.37;	16.92–1941.08;	48.03-1714.26;		
		Av = 3,098.99		Av = 3,896.7	Av = 416.63	Av = 584.66		
	Total	478.09–9,692.23;	_	594-7,839.08;	125.64-2028.18;	132.52-6,401.02;		
	content	Av = 4,351.3		Av = 4,049.19	Av = 543.68	Av = 1982.41		
	Cu	8.95–50.68; Av = 23.16	_	0.66–336.25; Av = 86.28	2.96–43.6; Av = 16.38	5.94–4,133.8; Av = 950.33		
Chlorite (ppm)	REE	0.04–1.27; Av = 0.49	0.056–0.77; Av = 0.425	0.02–3.42; Av = 0.66	0.05–3.34; Av = 0.808	0.03–17.62; Av = 1.8	0.02–0.71; Av = 0.248	
	LREE/ HREE	1.26–3.17; Av = 1.75	0.07–0.59; Av = 0.296	0.11–1.19; Av = 0.64	0.08–0.85; Av = 0.55	0.12–5.95; Av = 1.32	0.18–3.91; Av = 0.96	
	δΕυ	0.8–4.27; Av = 1.61	1.08–1.62; Av = 1.15	1.02–2.52; Av = 1.36	0.385–1.49; Av = 0.87	0.223–0.866; Av = 0.55	1–3.78; Av = 1.83	
	δCe	0.73–3.92; Av = 1.53	0.43–1.52; Av = 1.03	0.52–1.95; Av = 1.24	0.64–1.5; Av = 1.06	0.53–2.11; Av = 0.915	0.54–2.01; Av = 0.99	
	Ti	177.9–572.4; Av = 299.68	11.07–176.09; Av = 37.54	19–486.8; Av = 165.92	21.64–1,055.86; Av = 334.43	12.22–159.33; Av = 29.64	10.96–573.20; Av = 71.45	
	Sr	0.71–2.02; Av = 1.42	1.83–7.1; Av = 4.45	0.51–58.69; Av = 12.79	0.37–121.19; Av = 17.18	1.79–29.16; Av = 10.94	0.86–7.49; Av = 4.79	

TABLE 3 | elationship table between the sequence of events and the chemistry of pyrite and chlorite.

positive anomalies ( $\delta Eu = 0.385-1.49$ ), and negative Ce to positive anomalies ( $\delta Ce = 0.64-1.5$ ) in general. The results of (La/Sm)n (0.124-0.78) and (Gd/Yb)n (0.51-7.38) indicate that the internal fractionation of LREEs and HREEs is strong (**Figure 9E**).

The content of REEs in Chl-III ranges from 0.04 to 1.27 ppm, and the REE patterns are characterized by the relative enrichment of LREEs (LREE/HREE = 1.26–3.17), positive to weakly negative Eu anomalies ( $\delta$ Eu = 0.80–4.27), and positive to weakly negative Ce anomalies ( $\delta$ Ce = 0.73–3.92). The results of (La/Sm)n (0.20–0.88) and (Gd/Yb)n (1.80–8.08) indicate that the internal fractionation of LREEs and HREEs is strong (**Figure 9F**).

The *in situ* trace element analysis of chlorite shows that the trace element with the highest content in Chl-I ~ Chl-III is Mn. The contents of Ge, Mo, Cd, In, Sb, and REEs are usually less than 1 ppm, and Mn (348.55–3,466 ppm), Ni (80.73–477.43 ppm), Cu (0.01–388.38 ppm), and Sr (0.37–121.19 ppm) vary in large ranges. The contents of V, Ni, Ti, and Zn mostly range from 100 to 500 ppm.

#### DISCUSSION

#### Occurrence of Trace Elements in Pyrite

The distribution of trace elements of pyrite is obviously various in the different mineralization stages in the Tuwu copper deposit. Three main occurrence forms of trace elements in pyrite and other sulfides are observed: 1) in the mineral lattice in the form of isomorphism, 2) in visible micron-scale inclusions of other minerals, and 3) as invisible nanometer-scale mineral particles (Leng, 2017). LA-ICP-MS can not only accurately obtain the content of in situ trace elements but also determine the occurrence forms of elements in minerals according to the spatial variation trend of elements with depth during LA (Zhou et al., 2010; Keith et al., 2016). The trend of each element in the LA-ICP-MS time resolution profile shows that the signals of Sb, Se, Co, and Ni are generally stable and basically consistent with the signal distribution of Fe, indicating that Sb, Se, Co, and Ni may exist mainly in the form of isomorphism. Co and Ni replace Fe in pyrite, while Se and Sb can replace S. The ablation signal curves of Cu, Pb, Zn, Cu, and Au are variable and irregular, indicating that these elements mainly exist in the form of microinclusions in pyrite (Figure 8).

Moreover, most curves of Cu, Pb, and Ag appeared in local peaks (**Figure 8**), which indicates that Cu, Pb, and Ag elements mainly exist in pyrite in the form of microinclusions (chalcopyrite, galena, and native silver). In the correlation diagram (**Figure 10**), Cu shows a significant positive correlation with Pb and Ag. The degree of correlation was better in Py-II<sub>1</sub> and Py-II<sub>2</sub> than in Py-I and Py-III because Pb and Ag replacements of Fe are difficult by isomorphism, while Cu is known to occur as a substitution for Fe in many Fe-bearing sulfides (i.e., pyrite and pyrrhotite) and silicates (i.e., biotite and hornblende), indicating that more Cu exists in the form of



isomorphism in Py-I and Py-III than in Py-II<sub>1</sub> and Py-II<sub>2</sub>. A weak positive correlation occurs between Cu and As in pyrite. The signal distribution of As is consistent with that of Fe. A local peak

of As appears in a small number of samples and corresponds to the high value point of Cu, which indicates that As not only exists in arsenopyrite but also replaces S in pyrite *via* isomorphism. The



mass fraction of As fluctuates greatly, with obvious high values in Py-I and Py-III. Above all, these results indicate that the content of As was greatly affected by fluid in later stages, and its content may have a certain effect on the precipitation and occurrence of Cu. Zinc appeared in local peaks (**Figure 8**), which indicates that Zn may exist in the form of sphalerite microinclusions in pyrite, and the content of Zn appears to be highly variable in Py-III and Py-III. Moreover, considering that the ionic radius of Zn<sup>2+</sup> is 0.74 Å and Fe<sup>2+</sup> is 0.76 Å, Zn may replace Fe in the pyrite lattice *via* isomorphism.

### **Genesis of Pyrite**

Trace elements entering into the pyrite lattice in solid solution form are restricted by the chemical composition of the surrounding rock, mineral assemblage, and fluid properties. Because different hydrothermal fluids usually form under different physicochemical conditions, the contents of trace elements in pyrite vary significantly. Co and Ni are the most common trace elements in pyrite, and their contents and the Co/Ni ratio are often used to trace the genesis, deposition environment, and formation temperature of pyrite (Bralia et al., 1979; Campbell and Ethier, 1984; Bajwah et al., 1987; Cook, 1996). Different genetic types of pyrite usually have different Co/Ni ratios (Bralia et al., 1979; Brill, 1989; Large et al., 2011). The Co/Ni ratio of synsedimentary pyrite is usually less than 1, whereas the Co/Ni ratio of hydrothermal pyrite varies widely but is generally greater than 1; in particular, the Co/Ni ratio of volcanic hydrothermal pyrite is generally between 5 and 100 (Bralia et al., 1979; Craig et al., 1998; Franchini et al., 2015; Keith et al., 2016) because of the difference in the octahedral site preference energies of Co (7.4 cal/mol) and Ni (20.6 cal/mol) (Bralia et al., 1979). With the process of magma crystallization differentiation, Ni prefers to enter minerals with a high octahedral ratio (such as olivine and orthopyroxene) during early crystallization, while Co becomes relatively enriched in minerals forming in later-stage crystallization, thus causing the Co/Ni ratio of magmatic and hydrothermal pyrite to be greater than 1. Therefore, the Co/Ni ratio of pyrite that precipitated in response to magmatic or



hydrothermal activity is generally greater than 1 (Song and Zhang, 1986; Zhou et al., 2010; Leng, 2017).

The test results show that the Co and Ni contents of pyrite in the Tuwu mining area vary greatly and tend to decrease from Py-I, Py-II<sub>1</sub>, and Py-II<sub>2</sub> to Py-III, which reflects the fluid evolution characteristics of the Tuwu mining area to a certain (**Figure 11**). The Co/Ni ratio of the pyrite in the weakly propylitized zone (Py-I) ranges from 11.37 to 60.60. The siderophile elements Co and Ni are often enriched in basic or ultrabasic rocks. Thus, the anomalous enrichment of Co and Ni elements and high Co/Ni ratios in Py-I reflect a magmatic origin and are closely related to the formation of the CQ group volcanic rocks.

The Co/Ni ratio of the pyrite in the pyritization, sericitization, and silicification zone (Py-II1) ranges from 0.09 to 9.94 and is generally consistent with the characteristics of magmatic or hydrothermal pyrite. However, the lower the Co/Ni ratio of Py-II<sub>1</sub> is, the more the fragmentized pyrite is. Some studies have shown that using the Co/ Ni ratio as a criterion to distinguish the boundary of different types of ore deposits is not reliable, especially for hydrothermal deposits (Song and Zhang, 1986; Wang, 1989). Geochemistry characteristics of fluid inclusions and H, O, and S isotopes show that the ore-forming fluid mainly evolved from magmatic hydrothermal fluids, meteoric water, and metamorphic fluid (Liu et al., 2009; Wang et al., 2018). Therefore, Py-II<sub>1</sub> may have been affected by the addition of meteoric water or other fluids in the late stage of mineralization, and the Co/Ni ratio of the pyrite shows certain characteristics of sedimentary origin. Py-II<sub>1</sub> is depleted in low-temperature elements such as As, Se, and Te, which is consistent with the geochemical characteristics of high-temperature hydrothermal pyrite.

The Co/Ni ratio of the pyrite in the sericitization zone (Py-II<sub>2</sub>) ranges from 0.12 to 2.02 (average 0.53), and the corresponding data points mainly plot in the field of sedimentary-origin pyrite. The characteristics of trace elements in Py-II<sub>2</sub> indicate the complexity of pyrite genesis. The formation of this pyrite may

have involved the mixing of magmatic hydrothermal fluid and meteoric water. The meteoric water erased most of the magmatichydrothermal information. In addition, calcite, a typical lowtemperature mineral, is abundant in this stage.

The Co/Ni ratio of the pyrite in the chloritization-sericitization zone (Py-III) ranges from 0.003 to 0.57. Cobalt and Ni in pyrite can form CoS<sub>2</sub> and NiS<sub>2</sub> by replacing Fe via isomorphism. The unit cell parameters of pyrite are lengthening because the bond lengths of Co-S and Ni-S are greater than that of Fe-S. Pyrite can form a continuous solid solution series of FeS2-CoS2 and a discontinuous solid solution series of FeS2-NiS2 (Bralia et al., 1979; Song and Zhang, 1986; Craig et al., 1998). The content of FeS2-CoS2 is 2-3 orders of magnitude lower than that of FeS2-NiS2 in Py-III. The rocks in the chloritization-sericitization zone are characterized by shear deformation, which manifests as highly schistose local ore bodies and significant elongation of minerals. Overall, the formation of pyrite in this stage appears to involve superimposed transformation by metamorphic hydrothermal fluids, which is consistent with the geochemistry of fluid inclusions and isotopes (Liu et al., 2009; Wang et al., 2018).

The temperature controls the types and contents of trace elements in pyrite. Generally, a higher temperature of formation corresponds to more trace element types in pyrite and a higher content of trace elements (Craig et al., 1998; Large et al., 2007; Keith et al., 2016). Pyrite with different total trace element contents can be divided into three different stages. Comprehensive research on the contents of different trace elements and the Co/Ni ratio in pyrite shows that the total trace element content is highest in Py-I (4,351.34 ppm), which mainly formed in the later period of volcanic activity. The high copper content may reflect the high copper background value of the volcanic rocks in the CQ Group.

The total trace element contents in  $PyII_1$  and  $PyII_2$  are 4,049.19 and 543.68 ppm, respectively. Combined with previous studies of fluid inclusions (Liu et al., 2009; Han et al., 2010; Wang et al., 2018), these results reflect the variation characteristics of trace elements in pyrite from the main mineralizing stage to the late stage of the porphyry mineralization period. The total trace element content in Py-III is 1982.41 ppm. The temperature of the metamorphic hydrothermal fluid is between 150°C and 300°C, and the copper is obviously enriched by the superimposed transformation of the metamorphic hydrothermal solution. This finding explains why the temperature of the main metallogenic stage in the Tuwu copper deposit was approximately 200°C (Liu et al., 2009; Han et al., 2010), which is mainly because earlier high-temperature fluid inclusions were superimposed by ductile shearing.

# Rare Earth Element Geochemistry of Chlorite

The geochemical similarity of REEs allows them to be used in tracing applications, which is similar to isotopic values for the process of mineralization (Michard and Albarède, 1986; Maclean, 1988; Fulignati et al., 1999). By studying the REE composition characteristics of minerals from different metallogenic stages, we can determine the geochemical behavior of REEs during the processes of ore formation and alteration (Maclean, 1988;

Chen and Fu, 1991, 1997). Such work can effectively constrain the ore-forming material source, the metallogenic physicochemical environment, the fluid evolution process, and ore genesis (Chen and Fu, 1991; Bortnikov et al., 2008; Parsapoor et al., 2009; Tan et al., 2016, 2018). Research has shown that the geochemical behavior of REEs in minerals is mainly related to the secondary minerals formed by alteration, the content of REEs in hydrothermal fluids, the formation temperature of minerals, the pH and Eh of hydrothermal systems, the crystal chemical properties of minerals, and the partition coefficient of REEs between minerals and fluids (Michard and Albarède, 1986; Michard, 1989; Bortnikov et al., 2008). Previous studies on the REE characteristics of argillization zones (Karakaya et al., 2012), silicification zones (Terakado and Fujitani, 1998; Khalili and Malekmahmoudi, 2012), sericitization-silicification zones (Parsapoor et al., 2009), propylitization zones (Fulignati et al., 1999), and potassic zones (Taylor and Mclennan, 1985) show that the contents of REEs differ among different alteration zones.

## Rare Earth Element Characteristics of Chlorite in the Premineralization Period

Microstructure research has shown that there are two main formation mechanisms of chlorite in the Tuwu copper deposit: 1. metasomatic crystallization, which is chlorite formation via the metasomatism of biotite and hornblende, and in those rocks by late-stage hydrothermal fluids, 2. precipitation crystallization, which is the chlorite precipitated from hydrothermal fluids rich in Fe and Mg in the cracks or gaps between minerals. Metasomatic crystallization chlorite (Chl-I<sub>1</sub>) in the weakly propylitized zone mainly formed by late hydrothermal alteration of hornblende, pyroxene, and biotite in basalt. This chlorite generally inherited the REE distribution pattern of the volcanic rocks in the CQ group in the Tuwu mining area (Hou et al., 2005, 2006; Zhao et al., 2014). The precipitation crystallization chlorite (Chl-I2) is mainly associated with quartz, epidote, and minor pyrite. The REE patterns of Chl-I2 are characterized by the relative enrichment of LREEs and positive Eu anomalies, which may be due to the Eu in minerals that were in equilibrium with the hydrothermal solution entering the fluid in the form of Eu<sup>2+</sup> during the rock-water interaction and separated from the other REEs. As a result, Chl-I1 has stronger negative Eu anomalies than the volcanic rocks. In contrast, Eu is enriched in the fluid, and Chl-I<sub>2</sub>, which precipitated from the fluid, has positive Eu anomalies.

The enrichment degree of Eu in the fluid is closely related to temperature. Even in relatively oxidized environments, Eu in hydrothermal solutions mainly exists in the form of  $Eu^{2+}$  at relatively high temperatures (Bortnikov et al., 2008; Parsapoor et al., 2009; Fadda et al., 2012). Therefore, Chl-I<sub>2</sub> has a high crystallization temperature, and the total content of REEs in Chl-I<sub>1</sub> is generally 1-2 orders of magnitude higher than those in Chl-II and Chl-III, indicating the characteristics of the fluids related to Chl-I were obviously different from those of the fluids in the metallogenic epoch. The fluid likely was a volcanogenic hydrothermal solution that formed during late volcanic activity. The content of Cu in chlorite is generally high. The closer to the ore body, the higher the Cu content. This relationship reflects that the volcanic rocks in the CQ group have a high background value of Cu. In addition, the volcanic syngenetic hydrothermal activity primarily led to the enrichment in copper.

## Rare Earth Element Characteristics of Chlorite in Porphyry Metallogenic Stage

Similar LREE enrichment is observed in the REE distribution patterns of Chl-II<sub>1</sub>, which is associated with epidote, quartz, calcite, and zoisite in the strong propylitization zone. Chl-II<sub>2</sub> is associated with sericite, pyrite, and chalcopyrite in the pyritization–sericitization–silicification zone, and Chl-II<sub>3</sub> is associated with sericite, calcite, pyrite, and chalcopyrite in the sericitization–silicification zone. In addition, the total REE content in Chl-II is significantly lower than that of Chl-I and the Eu anomalies in Chl-II<sub>1</sub>, Chl-II<sub>2</sub>, and Chl-II<sub>3</sub> range from positive to negative. The reasons for the above REE characteristics in Chl-II are as follows.

Because chloritization occurs in neutral to alkaline pH conditions with relatively low temperatures (<300°C) (Reed, 1997), the REEs in the fluid mainly form complexes with SO<sub>4</sub><sup>2-</sup> and CO<sub>3</sub><sup>2-</sup>. While SO<sub>4</sub><sup>2-</sup> does not preferentially bind with REEs, CO<sub>3</sub><sup>2-</sup> does preferentially bind with REEs to form complexes (Foster, 1962; Bortnikov et al., 2008; Inoue et al., 2010). The secondary minerals formed from this fluid are relatively rich in heavy REEs.

LREEs have a larger ionic radius and smaller atomic weight than HREEs; therefore, LREEs are more easily leached from minerals (Bortnikov et al., 2008; Ye et al., 2011; Karakaya et al., 2012). The REE patterns of Chl-II are characterized by the relative enrichment of HREEs, indicating that it experienced leaching action of late hydrothermal fluids. Additionally, the similarity in REE distribution patterns indicates that Chl-II<sub>1</sub>, Chl-II<sub>2</sub>, and Chl-II<sub>3</sub> have the same material source and geochemical behavior.

- 2) Although accessory minerals exist in only very small quantities in most rocks, the REE distribution coefficients in accessory minerals are very high; therefore, they play a great role in controlling the REE content in rocks. The content of REEs in plagiogranite porphyry (24.87–55.88 ppm) is lower than that in volcanic rock (50.64–117.73 ppm) (Wang et al., 2014, 2015; Zhao et al., 2014), and the accessory minerals in plagiogranite porphyry are more than that in the volcanic rock, which causes the content of REEs in Chl-II to be lower than that in Chl-I.
- 3) The stability of Eu<sup>2+</sup> is controlled not only by the temperature, pH, Eh, and oxygen fugacity of the open system but also by the supply of alteration minerals. Previous studies have found that Eu<sup>3+</sup> tends to occur under low pH and high oxygen fugacity (Sverjensky, 1984; Bau, 1991; Hedenquist, 1995). The valence state of Eu will transform from Eu<sup>3+</sup> to Eu<sup>2+</sup> and fractionate from other REEs<sup>3+</sup> in a reducing environment, and Eu<sup>2+</sup> will remain in a stable state under a low oxygen fugacity environment. The Chl-II<sub>1</sub> in the strong propylitization zone and the Chl-II<sub>2</sub> in the

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sericitization-silicification zone exhibit positive Eu anomalies, indicating that the hydrothermal system featured a higher temperature (>250°C), a low pH, and relatively reducing conditions and the chlorite generally inherited the REE distribution pattern of the ore-forming fluid. The stronger positive Eu anomalies in Chl-II2 may is consistent with the large amount of sulfide that crystallized from the ore-forming fluid. With the development of mineralization and the addition of meteoric water, the metallogenic environment will transform from a relatively reducing environment to a relatively oxidizing environment. The occurrence of a large number of lowtemperature minerals, such as calcite, indicates that the mineralization temperature has obviously decreased. Chl-II<sub>3</sub> formed at this stage shows obviously negative Eu anomalies. Eu<sup>2+</sup> has an ionic radius similar to that of Ca<sup>2+</sup> and often occurs in calcite via isomorphism. Therefore, a large amount of calcite precipitation also reduces the content of Eu in the oreforming fluid.

## Rare Earth Element Characteristics of Chlorite in the Superimposed Transformation Period

The chlorite in the rich ore bodies (Chl-III) is characterized by obvious shear deformation characteristics. The REE distribution pattern exhibits LREE enrichment, and the fractionation of HREEs and LREEs is not obvious. This result suggests that the REE distribution pattern of Chl-III is apparently different from that of Chl-I and Chl-II. The formation depth of the Tuwu copper deposit is approximately 2-3 km (Wang et al., 2018). Combined with the significant deformation characteristics of the minerals in the chloritization-sericitization zone, we can infer that the Tuwu copper deposit was located in the brittle-ductile transitional zone. As the water/rock ratio increased, the diffusion pattern of the fluids shifted from percolation to infilling. The REE distribution pattern of Chl-III was mainly controlled by the fluid properties. In addition, the fluid source is also an important factor in determining the composition of REEs. The REE distribution patterns of the plagiogranite porphyry and dioritic porphyrite are generally consistent with that of Chl-III. However, the fractionation of HREEs and LREEs in Chl-III is obviously low, which indicates that the physical and chemical conditions of the Chl-III formation are apparently different from those of Chl-I and Chl-II. The positive Eu anomalies of Chl-III may indicate that Chl-III formed in a relatively reducing environment at relatively high temperatures.

# Spatial Variation Pattern of Trace Elements in Chlorite

In the formation process of porphyry deposits, a large alteration halo usually occurs around the center of the ore body. In this alteration halo, the mineral association and chemical composition may show certain zonation (Lowell and Guilbert, 1970). The classical metallogenic model of porphyry copper provides theoretical support for exploring porphyry deposits on the regional scale. However, ore bodies of economic value tend to be concentrated in the range of 1 km, and there are many alterations that are irrelevant to the ore-forming process (Inoue et al., 2010; Fadda et al., 2012). Therefore, effective identification of metallogenic and nonmetallogenic alteration systems is of great significance for the exploration of porphyry deposits. In recent years, the geochemical characteristics of altered minerals have attracted the widespread attention (Inoue et al., 2010; Cooke et al., 2014; Wilkinson et al., 2015; Xiao et al., 2018). Understanding these characteristics not only helps reveal the genesis of ore deposits but also provides technical guidance for prospecting exploration. Currently, based on electron probe and LA-ICP-MS analyses of altered minerals in porphyry systems, scholars have documented spatial variations in trace elements in altered minerals (Cooke et al., 2014; Wilkinson et al., 2015).

These achievements have been applied and verified many times in the field of exploration in China and elsewhere. They are commonly used to assist exploration. The systematic electron probe and LA-ICP-MS analysis of chlorite in the propylitization zone of the Batu Hijau superlarge porphyry copper (gold) deposit in Indonesia has shown that K, Li, Mg, Ca, Sr, Ba, Ti, V, Mn, Co, Ni, Zn, and Pb can enter the crystal lattice of chlorite. With distance from the ore body, the contents of Ti, V, and Mg decrease exponentially, while the contents of K, Li, Ca, Sr, Ba, Mn, Co, Ni, Zn, and Pb increase exponentially (Cooke et al., 2014; Wilkinson et al., 2015; Xiao et al., 2018).

The spatial variations in the trace element compositions in chlorite in the Tuwu mining area are similar to those in the Batu Hijau superlarge porphyry copper (gold) deposit. For example, the contents of high field strength elements, such as Ti and V, are higher in Chl-II and Chl-III near the ore body than in Chl-I far from the ore body. The degree of isomorphism of Ti in chlorite is mainly controlled by the temperature of mineral crystallization (Wilkinson et al., 2015). Therefore, the spatial variation in Ti content in chlorite also provides information on the thermal anomaly around the mineralization center. The variation range of Ti contents in Chl-II and Chl-III can reach 2-3 orders of magnitude within 300 m. However, the migration ability of high field strength elements, such as Ti and V, in hydrothermal fluid is quite limited. The significant differences in Ti content in chlorite in the porphyry mineralization stage in the Tuwu mining area indicate that the migration and diffusion of Ti were affected not only by temperature but also by the repeated action of hydrothermal. Moreover, the Ti contents in Chl-I vary little within a range of 1 km, which suggests that Chl-I in the volcanic rocks mainly formed in the volcanism and was scarcely affected by late hydrothermal activity.

With increasing distance from the ore body, the contents of K, Ca, Sr, Ba, and other large ion lithophile elements decrease but the degree of spatial variation is not as obvious as in a previous study (Wilkinson et al., 2015). This finding shows that the large ion lithophile elements migrated more easily in the late-stage hydrothermal solution. The content ranges of K, Ca, Sr, and Ba in Chl-I are obviously less than those in Chl-II and Chl-III, which suggests that Chl-II and Chl-III were affected by multiple phases of hydrothermal activity. The contents of Mn and Zn in chlorite gradually increase with distance from the center of the ore body, with a maximum value at a distance of approximately 1.2 km; beyond this distance, they gradually decrease until they stabilize. This pattern reflects the horizontal movement of Mn and Zn with the magmatic hydrothermal fluid in the surrounding



rock. The high contents of Mn and Zn in Chl-I reflect the high background values of Mn and Zn in the volcanic rocks in the CQ group (**Figure 12**).

# Influence of Ductile Shearing on the Tuwu Deposit

Although the Tuwu deposit experienced intense superimposed reformation, there are many features and characteristics that match those of a porphyry system. 1) Although most of the Cu in the Tuwu deposit is hosted in the surrounding rock, the ore-bearing plagiogranite porphyries are almost wholly mineralized. 2) Although the shape of the orebodies is mainly tabular, the ore structures are disseminated and veinletdisseminated and form stockwork. 3) Although the boundaries of the alteration zone are not obvious, the concentric alteration zones from the interior outward in succession are the potassic zone, the quartz-sericite zone, and the propylitic zone in general.

Compared with young porphyry systems, ancient porphyry systems tend to be subjected to later superimposed transformations. However, few studies have focused on the deformation and mineralization enrichment of porphyry deposits. The following evidence is presented in this article to demonstrate that the Tuwu porphyry deposit has undergone strong deformation and transformation. 1) The Tuwu porphyry deposit has experienced regional ductile shear activity after mineralization. The mining area is located in the northern influence zone of the large Qiugemingtashi-Huangshan ductile shear zone. 2) The intensity of the ductile deformation zone in the Tuwu-Yandong mining area is stronger in the south and weaker in the north. The southern part features a continuous strong ductile deformation zone and the northern part features a weak ductile deformation zone composed of several discretely distributed small lenticular and zonal ductile deformation zones. The ore has mylonite, augen, and fragmented structures. The mylonitic foliation developed in the ore body provides the main fine structural space for ore minerals (Ma et al., 2002, 2003). These characteristics indicate that the Tuwu mining area formed in the brittle-ductile transition zone. 3) Considerable research has already been done on fluid inclusions in the Tuwu copper deposit. Previous studies have found that the temperatures of the main ore-forming periods are approximately 200°C (Liu et al., 2009; Han et al., 2010), while others found that the temperatures are approximately 350°C (Wang et al., 2015). This discrepancy may be related to the different fluid properties of ore bodies in different positions. 4) Oriented secondary fluid inclusions are well developed in the deformation zone (Liu et al., 2009; Wang et al., 2018), indicating that the shape of the orebody changes from a disseminated low-grade to a vein highgrade orebody by tectonism. 5) The Tuwu mining area can be divided into three tectonic facies according to the characteristics of deformation. The ductile compression zone belongs to the moderately deep ductile deformation facies; the brittle-ductile shear zone belongs to the moderately shallow transition facies between ductile and brittle deformation, and the brittle fault zone belongs to the shallow structural facies. The fact that the moderately deep, moderately shallow, and shallow structures appear on the same plane indicates that they did not form at the same time but are instead the result of superimposing different structures in ore-bearing geologic bodies that have been uplifted several times and that have undergone multiple phases of structural deformation at different depths (Ma et al., 2003). 6) Closer to the ductile deformation zone, the contents of copper, sericite, chlorite, and quartz increase, and the oriented arrangement characteristics of these mineral become more obvious. There are some chalcopyrite and pyrite distributed along foliations of mylonite in the ductile deformation zone, and the chalcopyrite veins filled the fractures in the pyrite. These characteristics indicate that the ore has experienced strong deformation and the copper remobilized. 7) In situ trace element analysis of chlorite and pyrite in different stages shows distinctly different geochemical compositions Table 3.

The mineralization of the Tuwu and Yandong copper deposits occurred in the early Carboniferous at the southern belt of Dananhu-Haerlik island arc, and these deposits are separated by only 5 km. The ore-forming bodies are all plagiogranite porphyry and have similar mineralization processes (Chen et al., 2005; Hu et al., 2011; Shen et al., 2012; Pan et al., 2013). However, 80% of the Cu in the Tuwu mining area is hosted in the diorite porphyry, while 90% of the Cu in the Yandong mining area is hosted in the plagiogranite porphyry (Hu et al., 2011). The superposition and transformation of late ductile shearing on the two deposits in different spatial positions may be one of the reasons for this difference. The plagiogranite porphyry brought metallogenic materials and formed the initial mineralization of the Tuwu copper deposit, while the late ductile shear process accompanied by strong hydrothermal

activity resulted in the activation and enrichment of copper in the mining area. Moreover, the morphology of the ore bodies is mainly tabular, which is quite different from that of typical porphyry deposits.

These findings may be related to changes in the tectonic environment of the Tuwu mining area caused by ductile shearing to brittle shearing after metallogenesis, which not only led to the activation and migration of copper but also the enrichment of copper in the narrow brittle fractures. Therefore, the ore bodies are characterized by obvious tectonic control.

#### Mineralization of the Tuwu Copper Deposit

The Tuwu deposit is located approximately 10 km north of the Kangguer fault. The southern part of the Tuwu mining area experienced complicated subduction, island arc growth, plate collision, and ductile shearing from the Carboniferous to Permian. Finally, a faulted structural zone with complex structures and properties developed (Han et al., 2006; Wang et al., 2006; Cai et al., 2012), and the obviously superimposed mineralization characteristics of the Tuwu copper deposit developed. By combining and integrating our results with previously published data on the eastern Tianshan orogens, we suggest the following new model to interpret the evolution of the Tuwu deposit in eastern Tianshan and then provide a short discussion. Stage I: In the early Carboniferous, the northward subduction of the North Tianshan Ocean formed the CQ group volcanic rocks in the southern belt of the Dananhu-Haerlik island arc (Hou et al., 2005, 2006). A series of circular structures and radial structures around the volcanic crater were formed in the Tuwu mining area. The intersections of these structures formed dilational zones that provided the space and initial conditions for the emplacement of the plagiogranite porphyry. Typical minerals in the weakly propylitized zone, such as chlorite, epidote, dolomite, and calcite in the CQ Group volcanic rocks, were formed by volcanic hydrothermal processes. The higher contents of trace elements in pyrite and chlorite show that the mixing action of meteoric water played only a relatively minor role, while the high copper contents reflect the high copper background values of the volcanic rocks and the initial enrichment of Cu by volcanic hydrothermal processes. The volcanic rocks that were far from the ore bodies mainly suffered weakly volcanic hydrothermal process because of the undeveloped fracture system, and the trace element content of chlorite does not change obviously with the spatial position and generally inherited the REE distribution pattern of dark minerals in volcanic rocks Table 3.

Stage II: In the early Carboniferous, with the subduction and partial melting of the northern Tianshan oceanic plate at depth under certain physical and chemical conditions, adakitic magmas intruded into the shallow crust and formed the plagiogranite porphyry (Wang et al., 2014, Han et al., 2014; Shen et al., 2014b). The contemporaneous faults of the CQ group provided the tectonic environment for porphyry intrusion. The pyrite that formed in the early stage of porphyry mineralization under high-temperature and high-salinity conditions has an obvious magmatic-hydrothermal origin. With the development of mineralization, the input of meteoric water, and the long-term metasomatism of fluid in the late stage, the Co/Ni ratio of pyrite continuously decreased. From the early to late stages of porphyry mineralization, the REEs in chlorite show similar distribution patterns, indicating that the fluid had a similar source of ore-forming material **Table 3**. With the addition of meteoric water in the late stage II, athe metallogenic system became more open, and the fluid mixing caused minerogenetic matter precipitation (Liu et al., 2009; Shen et al., 2014b; Wang et al., 2015; Gao et al., 2017; Wang et al., 2018). As a consequence, the REE composition of the fluid changed, and the metallogenic system became a relatively oxidizing environment.

Stage III: In the Late Carboniferous-Triassic, the North Tianshan oceanic basin subducted and closed along the Kangguer ductile shear zone (Xiao et al., 2004; Mao et al., 2014b; Chen et al., 2019). The Tuwu copper deposit, which underwent multiple phases of compression and shear deformation, is located in the influence zone of the Kangguer ductile deformation zone, and the ore body underwent a strong superimposed transformation. The ore body was obviously mylonitized, and the Co/Ni ratio of pyrite decreased continuously under the action of multiphase fluids. The trace element characteristics of Chl-III also show distinct differences from those of Chl-II in the porphyry metallogenic stage **Table 3**.

#### CONCLUSION

- 1) We redefined the altered zones and the following three mineralization stages: premineralization stage (stage I), porphyry metallogenic stage (stage II), and superimposed transformation stage (stage III). Pyrite and chlorite are widely distributed in the three mineralization stages and different altered zones. Elements in pyrite, such as Sb, Se, Co, and Ni, mainly occur in the pyrite crystal lattice *via* isomorphism. Cu has obvious positive correlations with Pb, Zn, Ag, and Au in different ore-forming stages, and these elements may occur in the form of microinclusions or nanometer-scale particles.
- 2) The trace element compositions of pyrites from three stages obviously changed, thus reflecting the evolution of fluid properties with the mineralization process. The Co/Ni ratios changed from Py-I (11.37–60.60) to Py-II<sub>1</sub> (0.09–9.94), Py-II<sub>2</sub> (0.12–2.02), and finally Py-III (0.003–0.57), indicating that the fluid evolved from a volcanic genesis to a magmatic hydrothermal genesis. The fluid was affected by meteoric water and other fluids in the late metallogenic period and experienced a superimposed complicated structural hydrothermal fluid.
- 3) The REE distribution pattern of chlorite has obviously different features in the three stages. Chl-I generally inherited the REE distribution pattern of the volcanic rocks in the CQ Group, which shows that the fluid exsolved from the products of late volcanic activity. The similar REE distribution pattern of Chl-II<sub>1</sub> to Chl-II<sub>3</sub> indicates that the fluid had a similar ore-forming material source. The different features from Chl-II<sub>1</sub> to Chl-II<sub>3</sub> show that the fluid

evolved from magmatic hydrothermal to the mixture of magmatic and meteoric fluid. The REE distribution pattern of Chl-III shows that the fluid in ductile deformation zones experienced strong water–rock interactions with the surrounding rocks.

- 4) The contents of Ti, V, and Mg in the chlorite decreased exponentially with increasing distance, whereas the contents of K, Li, Ca, Sr, Ba, Mn, Co, Ni, Zn, and Pb increased. The maximum value appears at approximately 1.2 km and the values then gradually stabilize.
- 5) The Tuwu porphyry deposit has undergone volcanic activity, porphyry metallogenesis, and structural deformation from the early Carboniferous to early Triassic.

### DATA AVAILABILITY STATEMENT

The raw data supporting the conclusion of this article will be made available by the authors, without undue reservation.

## **AUTHOR CONTRIBUTIONS**

We the undersigned declare that this manuscript entitled "Mineralization of the Tuwu porphyry Cu deposit in eastern Tianshan, NW China: Insights from in situ trace elements of chlorite and pyrite" is original, has not been published before, and is not currently being considered for publication elsewhere. We would like to draw the attention of the editor to the following publications of one or more of us that refer to aspects of the manuscript presently being submitted, where relevant copies of such publications are attached. We confirm that the manuscript has been read and approved by all named authors and that there are no other persons who satisfied the criteria for authorship but are not listed. We further confirm that the order of authors listed in the manuscript has been approved by all of us. We understand that the corresponding author is the sole contact for the editorial process. He is responsible for communicating with the other authors about the progress, submissions of revisions, and final approval of proofs. Signed by all authors as follows: WT, QM\*, MY, YS, and XL.

## FUNDING

This study was financially supported by the National Key R&D Program of China (No. 2017YFC0601201), Chinese National Basic Research 973 Program (No. 2014CB440803), Ministry of Land and Resources of China (No. 201411026), and the Chinese Geological Survey Project (No. DD20160071).

### ACKNOWLEDGMENTS

Han Runsheng of the Kunming University of Science and Technology and Researcher Chen Zhengle of the Institute of Geomechanics provided valuable revision suggestions that improved this paper.

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**Conflict of Interest:** WT, QM, MY, YS, and XL were employed by the company Beijing Institute of Geology for Mineral Resources Co., Ltd., and WT and XL were employed by the company Sino-Zijin Resources Ltd.

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