



## Photonic Curing of Nickel Oxide Transport Layer and Perovskite Active Layer for Flexible Perovskite Solar Cells: A Path Towards High-Throughput Manufacturing

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High-throughput roll-to-roll (R2R) manufacturing of perovskite solar cells (PSCs) is currently limited by thermal processes that take tens of minutes each, translating to impractically long annealing tools at high web speeds. In addition, PSCs are usually made with metal oxide transport layer materials that require high temperatures for thermal annealing. Here, we demonstrate the fabrication of PSCs using photonic curing, instead of thermal annealing, to convert NiO<sub>x</sub> directly from sol-gel precursors for hole transport layers and to crystallize methylammonium lead iodide (MAPbl<sub>3</sub>) active layers on flexible Willow<sup>®</sup> Glass substrates. Photonic curing uses short, intense pulses of light to process materials at a high speed, hence it is compatible with R2R manufacturing. We achieved power conversion efficiencies (PCEs) of 11.7% in forward-scan and 10.9% in reverse-scan for PSCs made with photonic cured NiO<sub>x</sub> and MAPbl<sub>3</sub> films. Furthermore, both NiO<sub>x</sub> and MAPbl<sub>3</sub> films could be processed with a single photonic curing pulse, with a web speed of 5.7 m/min, and still produce PCEs comparable to thermally annealed control samples. Based on the single-pulse photonic curing condition for each film, we project a web speed of 26 m/min, laying a pathway to high-throughput production of perovskite solar modules.

Keywords: intense pulsed light, perovskite solar cell (PSC), nickel oxide, roll- to-roll (R2R) manufacturing, high throughout, photonic curing, hole transport layer (HTL)

### INTRODUCTION

Halide perovskite solar cell (PSC) efficiencies have increased dramatically in the past decade, with power conversion efficiencies (PCE) exceeding 25% for single-junction devices (NREL, 2019). Halide perovskites can be deposited with solution methods and processed at low temperatures, making them a good candidate for roll-to-roll (R2R) printing processes on flexible substrates (Li et al., 2018; Jena et al., 2019). With a 1.5 m wide web moving at a web speed of 30 m/min, R2R processing of PSC could provide up to 4 GW/year for a single R2R assembly line (EMC, 2020). Solution-deposition approaches, including printing (ink-jet, screen, gravure, flexography) and coating (blade, slot-die, spray) methods, have already been developed and can exceed 10 m/min, hence they are compatible with R2R manufacturing (Søndergaard et al., 2013; Abbel et al., 2018). However, metal oxide electron transport layers (ETLs) and hole transport layers (HTLs) for PSCs typically require a high temperature (>100°C) annealing step to fully achieve the desired electronic properties.

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Piper RT, Daunis TB, Xu W, Schroder KA and Hsu JWP (2021) Photonic Curing of Nickel Oxide Transport Layer and Perovskite Active Layer for Flexible Perovskite Solar Cells: A Path Towards High-Throughput Manufacturing. Front. Energy Res. 9:640960. doi: 10.3389/fenrg.2021.640960 High temperature annealing can cause substrate damages or mechanical failures, particularly when processing on plastic substrates that have low glass transition temperatures. In addition, plastic substrates typically have a coefficient of thermal expansion mismatch with the transparent conducting oxide layer, which may lead to cracking or delamination in these layers during processing. Recently, significant efforts have been made to reduce annealing temperatures of metal oxide materials to enable fabrication on flexible, plastic substrates (Shin et al., 2019). Reducing the thermal annealing temperature is a step in the right direction, but typical thermal annealing processes require samples to be at elevated temperatures for tens of minutes and are a major bottleneck in achieving a desired R2R speed. For example, an annealing time of 20 min at a web speed of 30 m/min would require a web length of 600 m within a furnace. A furnace this large would demand a huge physical space for the R2R system, emit lots of excess heat, and drive the manufacturing cost up.

One promising alternative to thermal annealing that has received a lot of recent attention is photonic curing, or intense pulsed light annealing (Schroder et al., 2006). In photonic curing, energy from a short (0.03-100 ms) pulse of broadband light (200-1,500 nm) is converted to heat selectively within lightabsorbing layers in thin-film samples, resulting in rapid transformations and reactions. While the intensity can be high, the total energy is low because of the short pulse duration. Two major advantages of photonic curing compared to thermal annealing are shorter processing time and compatibility with low heat-tolerant, transparent substrates. Photonic curing has been used to sinter printed metal nanoparticle inks to make conductive traces on flexible substrates (Schroder, 2011; Akhavan et al., 2017). It has also been demonstrated for fabricating ZrO2 (Daunis et al., 2020) and HfO<sub>2</sub> (Tetzner et al., 2014) dielectric films from sol-gel precursors on flexible and rigid substrates, respectively. For PSCs, a few research groups have demonstrated that photonic curing can convert perovskite active layers (Lavery et al., 2016; Troughton et al., 2016; Muydinov et al., 2018; Xu et al., 2020). Separately, photonic curing has also been reported for making mesoporous or compact TiO2 (Das et al., 2016; Feleki et al., 2017; Luo et al., 2017) or SnO<sub>2</sub> (Zhu et al., 2017) ETLs with thermally annealed perovskite active layers. Most recently, Ghahremani et al. (2020) have applied photonic curing to dry SnO2 nanoparticle ETLs and crystallize triple cation perovskite active layers on the same device, using multiple pulses for each layer. However, no work has been done so far on converting metal oxide semiconductors from sol-gel precursors with photonic curing for use as transport layers in PSCs.

Sol-gel precursor films require high annealing temperatures (~250°C for NiO<sub>x</sub>) to form a dense semiconducting metal oxide film (Liu et al., 2018). So far, most reports of low-temperature processing of NiO<sub>x</sub> HTLs use pre-synthesized NiO<sub>x</sub> nanocrystal suspensions (Yin et al., 2016; Zhang et al., 2017). Pre-synthesis of nanocrystals requires an additional, time-consuming processing step which increases the manufacturing costs (Wang et al., 2017). Therefore, it is advantageous to develop a one-step process to convert the NiO<sub>x</sub> HTL from its sol-gel precursor. In this report,

we demonstrate the use of photonic curing for converting  $NiO_x$  HTLs from nickel nitrate sol-gel precursor films on flexible indium tin oxide (ITO) coated Corning Willow<sup>®</sup> Glass (WG) substrates. Nickel nitrate was chosen as the sol-gel precursor for the NiO<sub>x</sub> HTL because nickel nitrate sol-gel precursors exhibit a lower decomposition temperature than other commonly used NiO<sub>x</sub> precursors (**Supplementary Figure S1**; Liu et al., 2018; Sun et al., 2019). Therefore, less energy is required to convert nickel nitrate precursor to a dense metal oxide film using photonic curing.

For this study, we fabricated inverted planar *p-i-n* architecture PSCs with the following device structure, listed from bottom to top: WG/ITO/NiOx/methylammonium lead iodide (MAPbI3)/ Phenyl-C<sub>61</sub>-butyric acid methyl ester (PC<sub>61</sub>BM)/bathocuproine (BCP)/Al. We demonstrate photonic curing for both the NiO<sub>x</sub> HTL and MAPbI<sub>3</sub> active layer on the same device, i.e., a device without any thermal annealing step, and achieve performance comparable to a device with all layers processed by thermal annealing. We also uncover the source of observed performance reduction for PSCs made on WG/ITO substrates compared to on 1 mm Glass/ITO. By eliminating all thermal annealing steps from our process, we reduce the total annealing time from 55 min to 11 s. Furthermore, we show that NiO<sub>x</sub> HTLs can be converted with just a single photonic curing pulse, further reducing the total annealing time for both layers to 21 ms. Finally, we discuss pathways to reach desired R2R web speeds using commercially available industrial equipment. By applying photonic curing to convert two layers of PSCs from precursors to final products for the first time, this work demonstrates an approach to fabricate flexible PSCs conducive to upscaling to high-throughput manufacturing.

### MATERIALS AND METHODS

#### **Solar Cell Fabrication**

Unless otherwise stated, all chemicals are purchased from Sigma-Aldrich and used as purchased. ITO coated 100 µm thick WG (Fraunhofer FEP) or ITO coated 1 mm thick glass (Kintec) were used as substrates. The 1 mm Glass/ITO substrates were rinsed sequentially in soapy water, acetone, IPA, and DI water and dried with N2. The WG/ITO substrates were only blown with N2 without chemical cleaning. Both substrates were UV-ozone treated (Bioforce Nanosciences Procleaner Plus) for 20 min before film deposition. NiOx precursor solution was made, modified from Liu et al. (2018), by dissolving 0.1 M nickel nitrate hexahydrate and 0.1 M acetylacetone in 2methoxyethanol and stirring overnight at room temperature. The NiO<sub>x</sub> precursor solution was filtered through a 0.2 µm polytetrafluoroethylene (PTFE) filter before spin coating at 3,000 rpm for 30 s followed by drying on a hot plate at 70°C for 3 min. For thermal annealed (TA) NiO<sub>x</sub>, the dried precursor films were calcinated on a hot plate at 250°C for 45 min. For photonic cured (PC) NiO<sub>x</sub>, the samples were processed per Photonic Curing immediately after drying. The NiOx coated samples were then loaded into an N2-purged glovebox to deposit the active layer. MAPbI<sub>3</sub> precursor solution was made,

following Ahn et al. (2015), by dissolving lead (II) iodide (TCI America), methylammonium iodide (GreatCell Solar), and sulfoxide molar N,Ndimethyl (equal ratio) in dimethylformamide. The MAPbI3 precursor solution was filtered through a 0.2 µm PTFE filter before spin coating at 4,000 rpm for 25 s inside an N<sub>2</sub>-purged glovebox. After spinning for 10 s, 1 ml of diethyl ether was applied to the sample. For TA MAPbI<sub>3</sub>, the samples were immediately annealed on a hot plate in the same glovebox at 100°C for 10 min. For PC MAPbI<sub>3</sub>, the samples were removed from the glovebox immediately after spin coating and processed per Photonic Curing before returning to the glovebox to deposit the ETL. PC<sub>61</sub>BM (Solenne BV) in chlorobenzene (13 mg/ml) was spin coated at 1,200 rpm for 60 s followed by spin coating bathocuproine (BCP) in ethanol (0.5 mg/ml) at 4,000 rpm for 30 s to create the ETL. No annealing step was required for the ETL. The top contact was deposited by thermally evaporating 100 nm of Al through a shadow mask.

## Spectroscopic Ellipsometry

Spectroscopic ellipsometry measurements (M-2000DI, J. A. Woolam) were done at 55°, 65°, and 75° incident angles from 192 to 1,690 nm in wavelength to obtain the index of refraction (n) and extinction coefficient (k) for each material. Uncoated WG, WG/ITO, WG/ITO/NiO<sub>x</sub>-precursor and 1 mm Glass/ITO were measured with transparent tape adhered to the backside of the samples to eliminate back-side reflections. The NiO<sub>x</sub>, MAPbI<sub>3</sub>, PC<sub>61</sub>BM, and Al films were deposited on single side polished Si wafers, following the same deposition procedures as Solar Cell Fabrication, for the measurement. The data were fitted using Complete Ease software to determine the optical constants and thickness of each film. The fitting for WG was performed using a Cauchy model parameterized into a general oscillator. The ITO coated on WG was fitted using the previously determined WG optical constants to obtain the optical constants of the ITO film. The fitting for ITO on WG starts with a general oscillator function that was first fitted over a limited wavelength range then converted into a B-spline model. The B-spline model was then expanded over the full wavelength range and parameterized into a general oscillator. The fitting for NiOx-precursor on WG/ITO used a B-spline model and the optical constants of the WG/ITO. Fits for NiOx, MAPbI3, PC<sub>61</sub>BM, and Al were obtained using B-spline models and the optical constants of the Si substrates. A B-spline model was also used for the ITO on 1 mm Glass with the optical constants for the glass taken from Rubin (1985). All fits had a mean square error of less than 9.

### **Photonic Curing**

Photonic curing was done using a NovaCentrix PulseForge<sup>®</sup> Invent equipped with one 500 V lamp driver, 1.5 kW power supply, and 20 mm (diameter)  $\times$  150 mm (length) xenon flash lamp (75 mm  $\times$  150 mm illumination area). All photonic curing was performed in ambient conditions (22–24°C, 30–50% RH). The processing stage height was adjusted to be as close to the lamp as the machine allows. The WG/ITO/NiO<sub>x</sub> samples were suspended between two pairs of magnetic steel slats such that

there was an air gap below the sample, between the WG substrate and the processing stage, during photonic curing.  $MAPbI_3$  samples were placed directly on the processing stage and held down by two magnetic steel slats for photonic curing.

The radiant energy at each PC processing condition was measured using a calibrated bolometer five separate times to get an average value. The bolometer was also used to determine the surface absorption of the WG/ITO/NiO<sub>x</sub>-precursor sample by measuring the radiant energy with and without the sample over the bolometer. The difference in the radiant energy measured with and without the sample, minus 8.8% substrate reflection, was used as the surface absorption of the sample for SimPulse simulations. The 8.8% reflection value was determined by performing the same measurement on an uncoated WG sample and assuming that the WG absorbs no light, which is quite accurate (Supplementary Figure S2). The WG/ITO/ NiOx-precursor and uncoated WG samples were pulsed with a lamp voltage of 200 V and a pulse length of 4 ms five separate times and the average radiant energy was used for calculations. The standard deviation in radiant energy for each set of bolometry measurements was less than 1%. This method estimates the surface absorption for WG/ITO/NiOx-precursor is 18%.

## Simulations

The percentage of WG/ITO/NiO<sub>x</sub>-precursor absorption can also be calculated with 1D transfer matrix method (TMM) (Burkhard et al., 2010) using the xenon lamp spectrum (NovaCentrix, 2020) as the source and n and k of WG/ITO/NiO<sub>x</sub>-precursor obtained from spectroscopic ellipsometry for the sample. We obtained 16% using the TMM method, which is consistent with the bolometry method described in *Photonic Curing*. The UV-vis spectrum of a dilute NiO<sub>x</sub>-precursor solution is shown in **Supplementary Figure S3**.

Simulations of film surface temperature vs. time were generated using NovaCentrix SimPulse software. The parameters input into the software were the PulseForge configuration (used to model the lamp output), the measured light absorption by the sample, the material properties (i.e., thermal conductivity, specific heat, and density) from literature, and the measured sample thickness. Note that SimPulse<sup>®</sup> only models the light absorption and thermal transfer processes and does not take into account any photochemical processes that may occur during photonic curing (Daunis et al., 2020). The WG/ITO/NiO<sub>x</sub>-precursor sample stack was modeled as 100 µm borosilicate glass and 150 nm ITO. The simulations were done in surface absorption mode using 18% absorption determined from the bolometry measurement as described in Photonic Curing. Using the TMM calculated 16% surface absorption value will only affect the simulated temperature values, but will not change the conclusions drawn from SimPulse results. Because the  $NiO_x$ -precursor layer is only ~30 nm thick, it is much thinner than the thermal length and is not included in the SimPulse" model stack. It is treated as a surface absorbing layer with its contribution measured by bolometry. SimPulse assumes that the light pulse is absorbed at the surface of the sample and the heat



generated there is either conducted down through the sample stack, raising the overall sample temperature, or removed from the sample via convective heat transfer to air from the top and bottom surfaces. The sample is suspended and has air above and below it. SimPulse<sup>®</sup> models this boundary condition using a convective heat transfer coefficient of 15 W/m<sup>2</sup>K at the top and bottom sample surfaces. The assumption that the heat is generated at the top surface is reasonable because most of the light is absorbed in the NiO<sub>x</sub>-precursor and ITO layers, which account for only 0.18% of the entire stack thickness.

Short-circuit current ( $J_{SC}$ ) calculations were performed by TMM with an AM 1.5G 100 mW/cm<sup>2</sup> light source. Two different material stacks were used (**Supplementary Figure S4A**) to model the different substrates: WG/ITO and 1 mm Glass/ ITO. The device layers, consisting of 10 nm NiO<sub>x</sub>, 450 nm MAPbI<sub>3</sub>, 50 nm PC<sub>61</sub>BM, and 100 nm Al, were the same for both substrates. The optical constants and thickness of the sample layers were determined from spectroscopic ellipsometry, except for 1 mm Glass, as described in *Spectroscopic Ellipsometry*. The TMM calculations assume the MAPbI<sub>3</sub> active layer has an internal quantum efficiency of 100%, which was reported in Yang et al. (2015).

## MATERIALS AND DEVICE CHARACTERIZATION

Current density-voltage (*J*-*V*) measurements were performed using an AM 1.5G 100 mW/cm<sup>2</sup> AAA solar simulator (Abet) and 2635A Keithley source-meter to conduct forward, -0.2 to 1.2 V, and reverse, 1.2 to -0.2 V, scans under illumination. The illumination area was 0.049 cm<sup>2</sup> and the voltage scan rate was 70 mV/s. *J*-V measurements were performed inside of an N<sub>2</sub>-filled glovebox after the samples had been stored in it overnight.

UV-vis absorption measurements were obtained using an Ocean Optics 4000 USB spectrometer. Film roughness was measured using an atomic force microscope (AFM, Asylum MFP-3D). Sheet resistance was measured with the four-point van der Pauw method with a Lake Shore 8400 HMS on at least three  $1 \text{ cm} \times 1 \text{ cm}$  films. Thermogravimetric analysis was conducted using a TA Instruments Q600 on 5-10 mg samples prepared by evaporating precursor solution. The heating rate is 5°C/min under 100 ml/min airflow. X-ray diffraction (XRD) was conducted using a Rigaku Ultima III diffractometer with Cu Ka (1.5418 Å) at a scan speed of 2°/min. Scanning electron microscope (SEM) images were taken with a Zeiss Supra 40 scanning electron microscope at an accelerating voltage of 10 kV. We analyzed MAPbI<sub>3</sub> grains in low magnification SEM images (Supplementary Figure S5) using ASTM E112-13 (ASTM International, 2013).

XPS was performed on an Ulvac-PHI VersaProbe2 with a monochromated Al Ka source (1,486.8 eV) at an angle of 45° to the sample surface at 50 W and 15 kV with a 200  $\mu$ m beam size. Ni 2p<sub>3/2</sub> spectra were averaged over 20 scans with an energy step of 0.1 eV at 20 ms per step and a pass energy of 23.5 eV. N 1s spectra were averaged over five scans with an energy step of 0.8 eV at 20 ms per step and a pass energy of 187.85 eV. XPS data were analyzed using commercial software (MultiPak, Ulvac-PHI). The area under the N 1s curve was calculated from 401 to 413 eV with the background taken as a straight line between the two endpoints. The Ni 2p<sub>3/2</sub> data for each sample was fitted to seven peaks, with peak 1 at the lowest binding energy and peak seven at the highest binding energy. The position of peak 1 was fitted for each sample (ranging from 853.91 eV for the 2-pulse film to 854.71 eV for the precursor film). The remaining primary



peaks (peaks 2, 3, and 4) were set to a fixed binding energy separation from peak 1 of 0.87, 2.1, and 3.62 eV, respectively, based on the best fit for the TA sample. Similarly, the position of the first satellite peak, peak 5, was fitted for each sample while the separation of peaks 6 and 7 from peak 5 was fixed at 2.84 and 5.38 eV, respectively. The full width at half maximum (FWHM) and height for each peak was allowed to vary independently. Due to the non-Gaussian elongated tail observed at the low binding energy edge of the Ni  $2p_{3/2}$  spectrum, peak 1 was fitted using a mixed Lorentzian-Gaussian function with the Gaussian characteristic allowed to vary from 0 to 100%. All other peaks were restricted to 70–100% Gaussian characteristic.

#### RESULTS

# Photonic Curing $NiO_x$ HTL Films and MAPbI<sub>3</sub> Active Layers

**Scheme 1** depicts the different combinations of TA or PC processes to form NiO<sub>x</sub> HTLs and MAPbI<sub>3</sub> active layers on WG/ITO substrates. Since both layers can undergo TA or PC processes, the resulting PSCs can have four different configurations: TA NiO<sub>x</sub> + TA MAPbI<sub>3</sub> (TA/TA), TA NiO<sub>x</sub> + PC MAPbI<sub>3</sub> (TA/PC), PC NiO<sub>x</sub> + TA MAPbI<sub>3</sub> (PC/TA), and PC NiO<sub>x</sub> + PC MAPbI<sub>3</sub> (PC/PC). The PC<sub>61</sub>BM, BCP, and Al top layer deposition processes are kept constant. During the NiO<sub>x</sub> PC process, a xenon flash lamp delivers short, 0.45–10 ms, pulses of light (1–30 pulses) to the sample at repetition rates ranging from 0.1 to 1 Hz and a lamp voltage of 500 or 700 V.

All PC MAPbI<sub>3</sub> samples were processed using a single pulse with a lamp voltage of 300 V and a pulse length of 20 ms. This PC processing condition for MAPbI<sub>3</sub> was adapted from our previous work on 1 mm glass (Xu et al., 2020), with the lamp voltage reduced for the thinner WG substrates. To compare the PC MAPbI<sub>3</sub> conversion on top of TA or PC NiO<sub>x</sub> HTL, we first conducted XRD to confirm the crystallinity. **Figure 1A** shows that the underlying NiO<sub>x</sub> HTL processed by TA or PC has little impact on the crystalline phase of the PC MAPbI<sub>3</sub> films. Correspondingly, SEM shows that the PC MAPbI<sub>3</sub> films have similar grain size distribution and film morphology (**Figures 1B–D**).

#### **Solar Cell Device Performance**

**Figure 2A** shows that PSCs made in the PC/PC configuration exhibit similar champion *J*-*V* performance as TA/TA and TA/PC control samples. The PCE values of the champion devices were very similar for all three configurations: 11.5, 12.5, and 11.7%, for TA/TA, TA/PC, and PC/PC, respectively. It is worth noting that



**FIGURE 2 | (A)** Champion reverse (solid) and forward-scan (dashed) device *J*-*V* curves for PSCs on WG/ITO fabricated with TA NiO<sub>x</sub> + TA MAPbl<sub>3</sub> (TA/TA, black), TA NiO<sub>x</sub> + PC MAPbl<sub>3</sub> (TA/PC, blue), or PC NiO<sub>x</sub> + PC MAPbl<sub>3</sub> (PC/PC, red). PC NiO<sub>x</sub> was fabricated at 500 V lamp voltage, 10 ms pulse length, and 10 pulses at 0.2 Hz. The inset is a photograph of a flexible WG PSC device. **(B)** Box plots of  $V_{OC}$ ,  $J_{SC}$ , FF, and PCE variation measured in reverse-scan for PSCs made with the same conditions listed in **(A)**.

NiO <sub>x</sub> anneal method	NiO <sub>x</sub> photonic curing process conditions					Average device parameters			
	Lamp voltage (V)	Pulse length (ms)	Radiant energy per pulse (J/cm <sup>2</sup> ) <sup>a</sup>	Number of pulses	NiO <sub>x</sub> processing time (s)	Open circuit voltage (V)	Short circuit current (mA/cm <sup>2</sup> )	Fill factor	Power conversion efficiency (%)
ТА	NA	NA	NA	NA	2,700	0.95 ± 0.06	15.2 ± 1.5	0.65 ± 0.07	9.4 ± 1.7
PC	500	3.5	10.5	1	0.0035	0.72 ± 0.02	$3.1 \pm 0.1$	$0.63 \pm 0.04$	$1.4 \pm 0.1$
PC	500	3.5	10.5	2	2.9	$0.93 \pm 0.02$	7.3 ± 1.4	0.62 ± 0.07	4.3 ± 1.2
PC	500	3.5	10.5	3	5.7	$0.87 \pm 0.03$	12.0 ± 0.8	$0.55 \pm 0.08$	$5.8 \pm 0.8$
PC	500	3.5	10.5	5	11.4	0.96 ± 0.02	14.1 ± 1.2	$0.60 \pm 0.05$	8.2 ± 1.3
PC	500	3.5	10.5	10	25.7	0.95 ± 0.02	14.6 ± 1.0	0.64 ± 0.04	8.8 ± 1.0
PC	500	3.5	10.5	20	54.3	0.93 ± 0.04	14.8 ± 0.5	0.64 ± 0.07	8.9 ± 1.5
PC	500	3.5	10.5	30	82.9	0.95 ± 0.03	15.1 ± 0.4	0.64 ± 0.03	9.1 ± 0.5
PC	500	1	4.2	5	4.0	$0.81 \pm 0.03$	$3.1 \pm 0.4$	0.58 ± 0.04	1.5 ± 0.2
PC	500	2	7.3	5	7.3	0.89 ± 0.10	11.8 ± 1.6	0.48 ± 0.10	5.0 ± 1.7
PC	500	3	9.6	5	10.3	0.92 ± 0.04	12.7 ± 0.5	0.65 ± 0.07	7.7 ± 1.4
PC	500	5	12.5	5	14.8	0.94 ± 0.01	13.8 ± 0.6	0.73 ± 0.01	$9.5 \pm 0.3$
PC	500	10	15.7	5	20.0	$0.95 \pm 0.02$	15.3 ± 0.4	0.61 ± 0.05	8.8 ± 0.7
PC	500	10	15.7	10	45.0	$0.97 \pm 0.04$	15.8 ± 0.6	$0.60 \pm 0.08$	9.3 ± 1.8
PC	700 <sup>b</sup>	0.45	7.7	1	0.00045	$0.99 \pm 0.01^{\circ}$	15.8 ± 0.3 <sup>c</sup>	$0.65 \pm 0.01^{\circ}$	$10.1 \pm 0.4^{\circ}$

TABLE 1   Average	reverse-scan device	J-V performance for	PSCs made with	PC NiO <sub>2</sub> + PC MAPbl <sub>2</sub>

<sup>a</sup>Radiant energy per pulse was measured by bolometry.

<sup>b</sup>NiO HTL was processed on a PulseForge 1300 equipped with four 950 V lamp drivers, a 4.75 kW power supply, and a 24 mm diameter xenon flash lamp.

<sup>c</sup>Devices were made with TA MAPbI<sub>3</sub> active layers.



FIGURE 3 | SimPulse<sup>®</sup> generated temperature at the top film surface vs. time of a WG/ITO/NiO<sub>x</sub>-precursor stack during the photonic curing process for two different pulse lengths and number of pulses at a fixed lamp voltage of 500 V: (A) A single 1 ms pulse. (B) A single 5 ms pulse. (C) Ten 1 ms pulses with a repetition rate of 1 Hz. (D) Ten 5 ms pulses with a repetition rate of 0.27 Hz.

the PC/PC devices show less hysteresis than the other two types of devices. **Figure 2B** shows the average and variation in reverse-scan open-circuit voltage ( $V_{OC}$ ),  $J_{SC}$ , fill factor (FF), and PCE for

all three configurations are similar. The NiO<sub>x</sub> HTLs in the PC/PC devices shown were photonic cured using 500 V lamp voltage, 10 ms pulse length, and 10 pulses at a repetition rate of 0.2 Hz.

The inset of **Figure 2A** shows an image of the flexible WG device. **Table 1** contains average reverse-scan PSC device parameters and total  $\text{NiO}_x$  processing time for each photonic curing condition used to process the  $\text{NiO}_x$  HTL.

To optimize the process for high-throughput manufacturing, we aim to find the PC processing conditions that can successfully convert NiO<sub>x</sub> with the shortest time within the PulseForge Invent's capability. The maximum lamp voltage of 500 V was selected to maximize the amount of power delivered per pulse. For each pulse condition, the maximum repetition rate allowed by the system was selected to minimize the processing time. The maximum repetition rate depends on the lamp voltage and pulse length, and is limited by how fast the capacitor bank can be recharged by the power supply. At a fixed lamp voltage, longer pulse lengths result in lower repetition rates. More energy is expended within a single pulse for longer pulses because the lamp is on for a longer time; thus, more time is needed to recharge the capacitor banks between pulses. Therefore, with a given tool and a fixed 500 V lamp voltage at the maximum repetition rate, the two independent variables are the pulse length and number of pulses.

At a fixed lamp voltage, the peak sample temperature  $(T_p)$  for a single-pulse process increases with pulse length up to a saturation value when the pulse length is longer than a threshold value  $(t_0)$ . For pulses longer than  $t_0$ ,  $T_p$  does not increase further with pulse length. Supplementary Figure S6 shows how T<sub>p</sub> changes with pulse length at 500 V lamp voltage. For our sample in our tool at the maximum 500 V lamp voltage,  $t_0$  equals ~3 ms.  $T_p$  increases with pulse length for pulses <3 ms until it reaches the single-pulse saturation value at  $\sim$  3 ms. T<sub>p</sub> is then equal to the single-pulse saturation value, independent of pulse length, for pulses >3 ms. to and the associated single-pulse saturation T<sub>p</sub> value depend on the photonic curing tool and parameters used particularly lamp voltage and number of lamp drivers as well as the sample properties and boundary conditions (Xu et al., 2020). Higher lamp voltages will produce higher T<sub>p</sub> values for a given pulse length because more power is being delivered to the sample.

We used SimPulse" to calculate the thermal response of the WG/ITO/NiO<sub>x</sub>-precursor sample under different photonic curing conditions. Figures 3A,B show the simulated thermal response of the sample when processed with a single, 500 V pulse at pulse lengths of 1 ms (<t<sub>0</sub>) and 5 ms (>t<sub>0</sub>), respectively. The yellow shaded region indicates the time that the lamp is on. As soon as the lamp turns on, the sample temperature rapidly increases. The lamp intensity is highest when the lamp initially turns on and decreases as the charges in the capacitor bank deplete; therefore, the sample heating rate will decrease throughout the pulse duration. As the sample temperature increases, the cooling rate increases due to convection and conduction of heat away from the sample. When the lamp turns off, the heating rate goes to zero and sample cooling takes over, as evidenced by a steep decrease in temperature past the yellow shaded region. While the cooling rate is initially very fast when the sample surface is at high temperature, the rate decreases as the sample temperature decreases resulting in the temperature returning to ambient over a time period on the order of seconds, which is not shown in these plots. For pulse length shorter than t<sub>0</sub>, the lamp is turned off during the initial rapid increase in sample temperature, reaching the peak temperature  $T_p$  right at the time when the lamp is turned off, as seen in **Figure 3A**. This  $T_p$  value (200°C for 1 ms pulse length) is lower than the single-pulse saturation  $T_p$  value (234°C) shown in **Supplementary Figure S6** because the sample did not have time to absorb enough radiative energy from the lamp. When pulse length is longer than  $t_0$ , the cooling rate and heating rate balance each other out at  $t_0$  and the single-pulse saturation  $T_p$  value is attained. While the lamp is still on past  $t_0$ , the cooling rate becomes greater than the heating rate, resulting in the temperature declining. This is evident from the slow decrease in temperature between 3 and 5 ms in **Figure 3B**.

Because our tool is limited to a 500 V capacitor bank, multiple pulses are required to convert the nickel nitrate precursor films to NiO<sub>x</sub>. Figures 3C,D show the temperature profiles for 10 pulses at 1 ms (1 Hz) and 5 ms (0.27 Hz) pulse lengths, respectively. As previously discussed, the repetition rate is lower for longer pulse length conditions. After the first pulse is finished, and before the sample has time to cool down to ambient temperature, a second pulse is applied. Because the baseline temperature of the sample is now higher than ambient, the second pulse drives T<sub>p</sub> to a higher value. As this process is repeated, T<sub>p</sub> increases until a multi-pulse saturation value is reached. For a given lamp voltage, multi-pulse saturation T<sub>p</sub> is equal to or higher than the single-pulse saturation T<sub>p</sub> depending on whether the sample cools down to the ambient temperature in between pulses. The multi-pulse saturation temperature and the number of pulses required to reach it depend on several factors, including pulse length, repetition rate (determined by power supply output), lamp voltage, and number of lamp drivers. For example, Figure 3D shows a multi-pulse saturation  $T_p$  of 306°C after four pulses for the 5 ms pulse length condition. In contrast, T<sub>p</sub> does not saturate even after 10 pulses for the 1 ms pulse length condition (Figure 3C).

To ensure the maximum possible  $T_p$  is achieved, we first chose a pulse length of 3.5 ms and varied the number of pulses to fabricate the NiO<sub>x</sub> HTL. Figure 4A shows the average (dashed red curve, closed symbols) and champion device (solid red curve, open symbols) reverse-scan PCE results as a function of number of pulses for 3.5 ms pulse length. The PCE values increase with the number of pulses and reach a saturation value of ~9% above five pulses. Therefore, for fast processing speed while ensuring  $NiO_x$  is converted, the number of pulses can be reduced to five. Next, we fixed the number of pulses to five and varied the pulse lengths (Figure 4B). A similar trend is observed where PCE values increase significantly at short pulse lengths and saturate above ~3.5 ms. The black curve and grey shaded area in Figures 4A,B represent the average reverse-scan PCE and one standard deviation, respectively, of devices with TA NiO<sub>x</sub> HTL, i.e., TA/PC devices, made during the same time frame. Note that all PSCs in Figure 4 are made with PC MAPbI<sub>3</sub> active layers. Thus, we demonstrate that NiO<sub>x</sub> HTLs photonic cured using five 3.5 ms light pulses delivered by 500 V lamp voltage at a repetition rate of 0.35 Hz, translating to 11.4 s of total processing time, produce PSC device performance comparable to NiO<sub>x</sub> HTLs thermally annealed for 45 min.





To examine the chemical changes in the NiO<sub>x</sub> film during photonic curing, XPS measurements were performed on NiO<sub>x</sub> films after photonic curing with 500 V lamp voltage, 3.5 ms pulse

length, 0.35 Hz repetition rate, and varying numbers of pulses, as with the devices shown in Figure 4A. The precursor film and thermally annealed film were also measured for comparison. The precursor film has a large N 1s peak at 408 eV due to the nitrate in the precursor solution. The top plot in Figure 5 shows that the area under this peak decreases significantly with an increasing number of photonic curing pulses, and is no longer present after five pulses. The thermally annealed NiO film also shows no N 1s signal. The Ni 2p<sub>3/2</sub> measurements and fitted curves are shown in Supplementary Figure S7. The data for each measurement was fitted to seven peaks. The four lower energy peaks are the primary peaks, and the three higher energy peaks are satellite peaks. The energy of the lowest energy peak (peak 1), centered at ~854.1 eV (not present in the precursor or one pulse films), is attributed to the NiO chemical state (Mansour, 1994b). Peak 2 at ~855 eV contains contributions from several possible Ni hydroxide species including  $\alpha$  and  $\beta$ -Ni(OH)<sub>2</sub>, and NiOOH with Ni in the 2<sup>+</sup> and 3<sup>+</sup> oxidation states (Mansour, 1994a; Mansour and Melendres, 1994; Biesinger et al., 2009). Peak 3 (~856.5 eV) contains contributions from both the hydroxide species as well as NiO due to multiplet splitting of the primary NiO peak (Mansour, 1994b; Boyd et al., 2020). Peak 4 at highest binding energy (~858 eV) is attributed to Ni in the  $3^+$  and  $4^+$  oxidation states (Grosvenor et al., 2006). Because only NiO contributes to peak 1 and peak 2 arises from the various hydroxide species with no contributions from NiO, the relative areas of these two peaks, expressed as a percentage of the total area under the four primary peaks, are indicative of chemical transformations from the precursor film to various degree of conversion during photonic curing or after thermal annealing. The results are summarized in the bottom plot in Figure 5. The percent contribution of peak 1 is shown in blue (hashed bars) and that of peak 2 is shown in orange (solid bars). While the sum of the two peaks does not equal 100%, due to contributions from both species in the third and fourth peaks not represented in Figure 5, the percent contributions by two peaks are correlated with



**FIGURE 6** | (A) Champion forward (dashed) and reverse-scan (solid) device *J*-*V* curves for PSCs on WG/ITO fabricated with TA NiO<sub>x</sub> + TA MAPbl<sub>3</sub> (TA/TA, black) or PC NiO<sub>x</sub> + TA MAPbl<sub>3</sub> (PC700V/TA, green) with NiO<sub>x</sub> photonic cured with a single pulse at 700 V lamp voltage and 450  $\mu$ s pulse length. (B) Average reverse-scan device *V*<sub>OC</sub>, *J*<sub>SC</sub>, FF, and PCE for PSCs made with the same conditions listed in (A). (C) Thermal annealing vs. photonic curing processing times (in log scale) for NiO<sub>x</sub> films made with the same conditions used in (A).



**FIGURE 7 | (A)** Forward (dashed) and reverse-scan (solid) *J-V* curves for PSCs made with TA NiO<sub>x</sub> exhibiting the highest measured  $J_{SC}$  on either 1 mm Glass/ITO or WG/ITO. **(B)** UV-vis transmission spectra for 1 mm Glass/ITO and WG/ITO. The inset table shows surface roughness and sheet resistance for both substrates. The sheet resistance for 1 mm Glass/ITO is from vendor data. **(C)** Highest measured  $J_{SC}$  (hashed), i.e., devices in **(A)**, vs. theoretical  $J_{SC}$  calculated from TMM (solid). Cyan and magenta represent 1 mm Glass/ITO and WG/ITO, respectively.

the concentration of NiO (blue) and hydroxides (orange) and are independent. Initially, there is no NiO present in the precursor film. After one pulse no NiO has formed (peak 1 remains at 0%), but the contribution of peak 2 to the total primary peak area decreases slightly. During this initial pulse, the nickel nitrate precursor is partially converted to nickel hydroxide, and the total concentration of all hydroxide species increases with their XPS signal shifted from peak 2 to peaks 3 and 4 (Supplementary Figure S7). After two pulses, there is a contribution from peak 1 and a large decrease in the area of peak 2 as the hydroxide species begin to convert to NiO. Additional pulses cause further conversion of hydroxides to NiO, and after 10 pulses the relative contributions from the two peaks are nearly identical to those in the TA sample. The XPS results provide an explanation for the device results in Figure 4A, which show that the PCE for the NiO<sub>x</sub> HTL processed with five pulses is slightly lower compared to PCE values for NiOx HTLs processed with 10, 20, and 30 pulses. Figure 5 shows that while the N 1s peak is not observed after five pulses, peaks 1 and 2 do not resemble the thermally annealed reference sample until 10 pulses. Therefore, the small difference in PCE values for NiO<sub>x</sub> HTLs processed with 5 and 10 pulses likely arises from the lower conversion of NiO<sub>x</sub> at five pulses, as evidenced by the lower peak 1 and higher peak 2 areas.

To explore whether the NiO<sub>x</sub> HTL can be converted in a single pulse, which will dramatically reduce the total processing time, we employed a PulseForge<sup>®</sup> 1300 with four 950 V lamp drivers, a 4.75 kW power supply, and a 24 mm diameter lamp. The WG/ITO/NiO<sub>x</sub> samples were held down by two magnetic steel slats and photonic cured directly on the processing stage. **Figures 6A,B** show champion device *J*-*V* curves and average reverse-scan *J*-*V* parameters for PSC devices made in the TA/TA and PC700V/TA configurations. The "700V" sample label denotes that the NiO<sub>x</sub> HTLs were photonic cured with a

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single pulse at 700V lamp voltage and 450  $\mu$ s pulse length. The NiO<sub>x</sub> samples photonic cured in 450  $\mu$ s give device performance comparable to the NiO<sub>x</sub> thermally annealed for 45 min at 250°C. **Figure 6C** shows a six-order-ofmagnitude reduction in processing time when the NiO<sub>x</sub> HTL was photonic cured with a single pulse.

# PSC Performance on Flexible vs Rigid Substrates

When transitioning from 1 mm Glass/ITO to flexible WG/ITO substrates, we observed a drop in performance for both champion and average device performance. Upon further examination of the J-V data, there is a ~13% decrease in  $J_{SC}$  when going to the WG/ITO substrate. Figure 7A shows the forward and reverse-scan J-V curves for PSCs made with TA NiOx on the two substrates with the highest J<sub>SC</sub> value we have measured on each substrate. To understand the discrepancy in J<sub>SC</sub>, we compared UV-vis transmission spectra, surface roughness, and sheet resistances of both substrates in Figure 7B. The AFM images are shown in Supplementary Figure S8. While the surface roughness and sheet resistance are both higher for WG/ITO compared to 1 mm Glass/ITO, they cannot explain the large JSC discrepancy. Because the NiOx is thermally annealed on a hot plate, a quasi-equilibrium heating process, any performance differences arising from heat transfer or substrate thickness effects during the NiOx annealing are negligible. UV-vis transmission spectra show a significant difference between the two substrates, with WG/ITO having a lower transmission, especially between 300 and 500 nm wavelengths. To determine whether the difference in optical transmission between the substrates is the reason for the decrease in  $J_{SC_2}$  we determined the ITO optical constants using ellipsometry and then used TMM to calculate the maximum JSC achievable on each substrate (Figure 7C).

Because both WG and 1 mm Glass are optically transparent and can be well fitted with a Cauchy model, the difference in the transmission spectra in Figure 7B must arise from the ITO layers. When fitting the spectroscopic ellipsometry data, the ITO films must be modeled as a graded layer where the optical constants, n and k, vary throughout the thickness of the ITO film. Fits for the ITO film on WG were achieved by modeling the ITO film as five, equally spaced, distinct layers (Supplementary Figure S4A). In contrast, the ITO film on 1 mm Glass only needed two layers in the grading model. The graded models showed good agreement with the measured spectroscopic ellipsometry parameters and optical transmission spectra. The optical constants for the ITO films on each substrate are shown in Supplementary Figures S4B,C. The extinction coefficient for the top-most ITO layer on WG is significantly higher throughout the 300-500 nm wavelength range, which contributes to the lower UV-vis transmission curve for WG/ITO (magenta) in Figure 7B. The n and k for the two layers in 1 mm Glass/ITO are similar, indicating a much more uniform film. The ITO layers on the two substrates are not deposited by the same manufacturer or the same method. The difference in the two materials likely originates from the ITO deposition processes. The WG/ITO was likely deposited using a R2R process where multiple layers of ITO were coated onto the WG sequentially (FraunhoferFEP, 2020).

Finally, we used the optical properties of each substrate to calculate the maximum  $J_{SC}$  possible using TMM. The WG/ITO substrate material stack was modeled as 100 µm WG and five 30 nm ITO layers. The 1 mm Glass/ITO substrate material stack was modeled as 1 mm soda lime glass and two 75 nm ITO layers. Both substrate materials stacks were modeled with the same top layers, as described in *Simulations*. Figure 7C shows an excellent agreement between our highest measured  $J_{SC}$  values (hashed bars) and TMM calculations (solid bars) using the models described above and shown in **Supplementary Figure S4A**. These results show that the optical properties of the ITO film on WG that is commercially available limit the achievable  $J_{SC}$  in PSCs. Higher  $J_{SC}$  and PCE may be achieved with a different ITO deposition process (Dou et al., 2017), but that is beyond this study.

### DISCUSSION

To achieve high-throughput manufacturing of PSCs, alternatives to thermal annealing must be developed. Typically, thermal conversion of metal oxides from sol-gel precursors requires long processing times (minutes to hours), making it incompatible with high-speed R2R manufacturing. R2R processing for our NiO<sub>x</sub> thermal annealing condition, 250°C for 45 min, and a web speed of 30 m/min would require a least a 1,350 m-long web length within a furnace for processing the NiO<sub>x</sub> HTL alone. In contrast, photonic curing provides much shorter processing times and lower energy consumption in a smaller physical space than thermal annealing.

We have shown that PSCs made on flexible WG/ITO substrates with photonic cured NiO<sub>x</sub> HTLs, starting from a sol-gel precursor, and photonic cured MAPbI<sub>3</sub> active layers yield an average PCE of ~10%, with champion devices reaching nearly 12%. These results are in good agreement with thermally annealed NiO<sub>x</sub> and thermally annealed MAPbI<sub>3</sub> control samples on WG/ITO. Thus, we can eliminate all TA steps from our fabrication and achieve a significant reduction in processing time.

For large-area manufacturing, two issues are of particular concern: uniformity and web speed. Xu et al. (2020) have already demonstrated ~5% variation across 5 inches in the cross-web direction for PC MAPbI<sub>3</sub> films. Down-web uniformity using photonic curing for processing is a concern because the light pulses are discrete while the substrate is conveyed at high speed underneath the lamp. To address this issue, stitching must be applied according to **Eq. 1**:

$$f = \frac{0.2 \cdot s \cdot o}{w} \tag{1}$$

where *f* is repetition rate of the light pulses in Hz, *s* is web speed in ft/min, *o* is overlap factor, and *w* is the down-web lamp width in inches (Schroder et al., 2013). For the PulseForge<sup>®</sup> Invent, the down-web lamp width is 75 mm (3 in). The repetition rate is calculated in SimPulse<sup>®</sup> considering the specific equipment and pulse conditions used (power supply output, lamp type, number of drivers, lamp voltage, and pulse length). The overlap factor determines the number of pulses received by a point on the sample. An overlap factor of one means that the lamp will only

pulse each point on the sample once. Because the light intensity at the edges of the illumination area is lower, it is recommended to use an overlap factor of 1.2 instead of 1. Using an overlap factor of 1.2 will cause subsequent pulses to overlap by 17% and reduce any non-uniformity in the down-web direction.

Using the 700 V single-pulse photonic curing conditions to convert NiOx described above, Eq. 1 yields a web speed of 5.7 m/ min using f = 1.5 Hz and o = 1.2. The web speed can be increased by using a larger power supply. For example, increasing the 5 A power supply used in this work to a 45 A power supply, the repetition rate can be increased to process material at a web speed of 13 m/min. Because the web is continuous in R2R processing, the slowest processing step dictates the overall web speed. Performing a similar calculation for the MAPbI<sub>3</sub> processing conditions yields a web speed of 18 m/min; hence, the NiOx processing step is the slowest. Faster web speeds can be achieved by machine design and layout considerations. For example, by rotating the 75 mm  $\times$ 150 mm lamp head by 90°, the processing area can be doubled allowing a web speed of 26 m/min. Placing additional lamps side by side can increase the processing width to meters. Thus, it is feasible to fabricate flexible PSCs using high-throughput photonic curing on both the NiOx HTLs and MAPbI3 active layers, with the potential to scale up to 26 m/min web speeds at arbitrary width. Thus, this work demonstrates the possibility to manufacture perovskite solar modules at a large scale, which will produce lower final product cost due to the economy of scale and reduced energy consumption.

#### DATA AVAILABILITY STATEMENT

The original contributions presented in the study are included in the article/**Supplementary Material**, further inquiries can be directed to the corresponding author.

### **AUTHOR CONTRIBUTIONS**

JH and TD conceived the idea. RP fabricated and measured devices, measured ellipsometry, performed TMM calculations,

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and wrote the first draft of the manuscript. W.X. participated in device fabrication, determined photonic curing conditions for perovskite on Willow<sup>®</sup> Glass, and performed four-point van der Pauw, XRD, SEM, TGA, and AFM measurements. TD measured XPS, wrote the corresponding XPS sections of the manuscript, and participated in ellipsometry and photonic curing experiments. TD and KS provided guidance on the design of photonic curing experiments and interpretation of the data. JH supervised the research and contributed to the design of experiments and analysis of results. All authors discussed the data and edited the manuscript.

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#### SUPPLEMENTARY MATERIAL

The Supplementary Material for this article can be found online at: https://www.frontiersin.org/articles/10.3389/fenrg.2021.640960/full#supplementary-material.

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