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SPECIALTY SECTION  
This article was submitted to Carbon  
Capture, Utilization and Storage,  
a section of the journal  
Frontiers in Energy Research

RECEIVED 06 January 2022  
ACCEPTED 08 August 2022  
PUBLISHED 07 September 2022

CITATION  
Xu C-G and Zhang W (2022), Editorial:  
Gas hydrate and hydrate technology for  
greenhouse gas mitigation.  
*Front. Energy Res.* 10:849490.  
doi: 10.3389/fenrg.2022.849490

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# Editorial: Gas hydrate and hydrate technology for greenhouse gas mitigation

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## KEYWORDS

CCUS, Hydrate, CO<sub>2</sub> capture, NGH, CH<sub>4</sub>-CO<sub>2</sub> conversions

## Editorial on the Research Topic

### Gas hydrate and hydrate technology for greenhouse gas mitigation

Hydrate-based carbon dioxide separation, capture, utilization and storage (CCUS) is a new carbon emission reduction technology. Compared with the traditional technologies, it has the advantages of high energy efficiency, large separation capacity, simple process and environment-friendly. Since the 1980s, the research on gas hydrate and hydrate-based technology has gradually attracted more and more attention and become one of the focus of research (Li et al., 2016). Presently, the research is mainly focused on solving the two aspects restricting the industrial application of hydrate technology, low gas consumption and slow hydrate formation rate, including thermodynamics, kinetics, micro mechanism of hydrate-based CO<sub>2</sub> separation and capture and hydrate-based utilization and storage of CO<sub>2</sub>, such as CO<sub>2</sub>-CH<sub>4</sub> replacement and exploitation of natural gas hydrate (NGH) (Xu et al., 2019).

Promoting gas hydrate formation and enhancing CO<sub>2</sub> consumption during the hydrate formation under a certain temperature and pressure are the two key issues to transform hydrate-based technology from experiment to commercial application. This is also the top priority of current related research. The formation of gas hydrate is essentially the transformation of gas-liquid solid phases, involving heat and mass transfer and interfacial reaction. Increasing the gas-liquid contact area and promoting the mass and heat transfer in the process of hydrate formation are considered to be the two directions that researchers focus on to solve the commercial application of the hydrate technology. Therefore, researchers have proposed many physical/chemical methods such as mechanical stirring, bubbling, spraying, porous media and emulsification to promote gas-liquid contact during hydrate formation (Ma et al., 2016). However, despite years of research, the bottleneck restricting the commercial application of the hydrate technology has not been solved. Therefore, researchers return the research focus from dynamics to thermodynamics. It is recognized that the problem can be solved fundamentally only after

the affecting mechanism of the gas hydrate formation is deeply acknowledged from the micro level. For example, by adding surfactant to water, reducing the solution interfacial tension and making more gas evenly dispersed in the liquid phase, the gas-liquid contact area is promoted from the mesoscopic and even microscopic. This way is proven to not only significantly improve the hydrate formation rate, but also improve the gas consumption. For another example, by using the chemical polarity and charging characteristics of gas and solution, an electric field is introduced into the hydrate formation process to make the gas-liquid miscible phase move each other under the action of the electric field, promote the gas-water contact from the micro level, and also effectively improve the formation rate of natural gas hydrate [Zhao et al.](#) However, this does not mean that the stronger the electric field, the better the effect of promoting the gas hydrate formation, because the higher Joule heat brought by the strong electric field and the ionization of metal ions with higher concentration are not conducive to the formation of gas hydrate. However, through this study, we can also determine that promoting sufficient and uniform gas-liquid contact between gas and water at the mesoscopic or micro level can more effectively solve the problems of slow hydrate formation rate and low gas consumption caused by difficult gas diffusion in the liquid phase during the hydrate formation. Therefore, we have got such a revelation: for future industrial applications, the increase of CO<sub>2</sub> hydrate formation rate will also be mainly based on micro water vapor contact, supplemented by macro stirring or spray.

Hydrate-based CO<sub>2</sub> capture technology can help to capture a large amount of CO<sub>2</sub> from power plant flue gas or other centralized emission sources of CO<sub>2</sub>. However, for carbon emission reduction, CO<sub>2</sub> capture is not the ultimate goal. The ultimate goal is to convert the captured CO<sub>2</sub> into other types of carbon containing products for reuse, or realize CO<sub>2</sub> permanent storage while using CO<sub>2</sub>. Hydrate-based CH<sub>4</sub>-CO<sub>2</sub> replacement extracting CH<sub>4</sub> from NGH is considered to be a technology with great potential and can achieve win-win results in carbon emission reduction and natural gas exploitation, and it is considered as a new way and technology to realize CCUS, because CO<sub>2</sub> is also stored in the form of CO<sub>2</sub> hydrate solid while realizing CH<sub>4</sub> exploitation. Currently, the biggest problem of CH<sub>4</sub> extraction from NGH by hydrate-based CH<sub>4</sub>-CO<sub>2</sub> replacement method is the low production efficiency. The fundamental reason is the difficulty of CO<sub>2</sub> gas diffusion in NGH reservoir ([Wang et al., 2021](#)). In fact, the basic condition for the replacement is that the concentration of CO<sub>2</sub> in the peripheral gas phase of CH<sub>4</sub> hydrate should exceed 20%, because the higher the CO<sub>2</sub> concentration, the more opportunities to promote the mass transfer reaction between CO<sub>2</sub> and CH<sub>4</sub> in hydrate cage. Due to the porous structure in NGH, the size of pores, capillary effect in the NGH and input pressure are the main factors affecting CO<sub>2</sub> diffusion. Without changing the physical structure of NGH

and increasing the compression cost, the gas molecules smaller than CO<sub>2</sub>, such as H<sub>2</sub> or N<sub>2</sub>, are introduced in the process of CH<sub>4</sub>-CO<sub>2</sub> replacement to enhance the CH<sub>4</sub>-CO<sub>2</sub> replacement efficiency. The research showed that due to the small size, H<sub>2</sub> or N<sub>2</sub> molecules can smoothly pass through the small pores of NGH and expand the pores in the NGH so that CO<sub>2</sub> can pass through relatively smoothly, and finally result in the improvement of the replacement efficiency by around 16% ([Ding et al.](#)). In fact, this method has been verified as early as last century. American researchers used CO<sub>2</sub>/N<sub>2</sub> mixture to extract CH<sub>4</sub> from NGH in permafrost region in 2007, but this method has not been tested in seafloor ([Oyama and Masutani, 2017](#)). Because the small molecule H<sub>2</sub> or N<sub>2</sub> can not be stable in the stratum in the form of hydrate like CO<sub>2</sub>, and it may lead to some other problems such as gas leakage, unless it can completely recovered after the extraction. Therefore, future research will focus on improving the efficiency of replacement and storage, while more energy needs to be invested in the recovery and reuse of small molecular gases.

Moreover, the application of the hydrate-based technology in NGH exploitation is of great significance to alleviate the fossil energy crisis faced by human sustainable development. Different from oil and gas reservoirs, the NGH exists in the stratum in the form of solid and mixed with sediments to support the stability of the stratum. Therefore, there is a risk of geological collapse in the NGH exploitation. In addition, there are various forms of NGH in the stratum. The process of NGH exploitation is accompanied by the production of a large amount of water and sand, which increases the difficulty of exploitation. At present, only Japan and China have carried out experimental NGH exploitations in the Nankai Trough of Japan and Shenhu sea area of China, but they have not lasted for more than 2 months because of the problem of producing a large amount of water and sand ([Yu et al., 2021](#)). Therefore, relevant research still focuses on the simulation and experimental optimization of the exploitation technologies ([Ruan et al., Yu et al., Yamamoto and Nagakubo](#)).

Gas hydrate technology serving CCUS is receiving more and more attention. As a new technology different from the traditional one, the hydrate technology is the only technology that can synchronously realize CO<sub>2</sub> separation, capture, utilization and storage, and has potential competitive advantages. However, the current research is still in the experimental stage, especially the research on the continuous process of CO<sub>2</sub> separation, capture, utilization and storage has not yet appeared, which needs to be strengthened in the later research.

## Author contributions

The Editorial is written by C-GX and WZ.

## Acknowledgments

The authors greatly appreciate financial support from the Key Program of National Natural Science Foundation of China (51736009), the Natural Science Foundation of Guangdong Province, China (2019A1515011490), Guangdong Special Support Program-Local innovation and entrepreneurship team project (2019BT02L278), Special Project for Marine Economy Development of Guangdong Province (GDME-2018D002), Fundamental Research and Applied Fundamental Research Major Project of Guangdong Province (2019B030302004, 2020B0301 03003), Science and Technology Apparatus Development Program of the Chinese Academy of Sciences (YZ201619), Frontier Sciences Key Research Program of the Chinese Academy of Sciences (QYZDJSSW-JSC033).

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