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A field-scale evaluation of the removal of odorous gases by a plant material-based deodorant

Jun Wu¹, Jing Yang¹, Jian Wang², Liben Fang³ and Jingjing Fang³*

¹Center for Membrane and Water Science & Technology, Zhejiang University of Technology, Hangzhou, China, ²Shanghai Radio Equipment Research Institute, Shanghai, China, ³Navy Medical Centre, Shanghai, China

The composting process emits various volatile chemicals, including hydrocarbons, nitrogen compounds, sulfur compounds, and oxygenated compounds. These gases must be removed to avoid atmospheric pollution. Spraying extracted materials from plants, such as tea tree oil, onto the surface of a windrow is a novel odor-reduction technique for the treatment of gaseous emissions from composting facilities. In this study, a plant material-based deodorant from a mixture of coconut, lemon, and tea tree essential oils was sprayed directly onto the surface of a windrow with a spraying frequency of three times a day. Odorous compounds were measured before and after spraying to evaluate the performance of the plant material-based deodorant. After spraying, the ammonia concentration decreased from 10,000 to 7,000 μ g/m³. There was a large decline in alcohols, aromatics, and carbonyls from 30,612, 12,270, and 5,138 μ g/m³ to 4,965, 4,669, and 2,253 μ g/m³, achieving efficiencies of 83%, 62%, and 56%, respectively. The removal efficiencies for terpenes, volatile fatty acids, sulfur compounds, and ammonia were lower than the former, falling within the range of 24%-36%. The theoretical total odor concentration was decreased from 23,757 $\mu g/m^3$ at the beginning to 13,056 $\mu g/m^3$ in the end, revealing that most of the bad smell had been eliminated.

KEYWORDS

deodorant, plant extracted, odor gas, composting plant, odor abatement

1 Introduction

Solid waste management, particularly the organic fraction, is becoming a serious problem in developing countries. The common treatments include dumping, landfill, incineration, and composting, but each method has different negative impacts on the environment (Guo et al., 2021). Compared with these methods, composting is the most suitable treatment for organic waste. This technology uses microorganisms to transform organic waste into valuable fertilizer and soil amendment (Kundariya and Mohanty, 2021). Unfortunately, the composting process generates various volatile chemicals, including hydrocarbons, nitrogen compounds, sulfur compounds, and oxygenated compounds (Hwang et al., 2020). These gases have an irritating odor and are also hazardous to human health. Thus, they must be treated to avoid secondary atmospheric pollution (Gutiérrez, et al., 2015). Although the main objective of composting facilities is to reduce the environmental impact of organic solid wastes, there are unavoidable gaseous emissions and odor compounds responsible for nuisances and complaints (Tian, et al., 2013).

Odor abatement technologies are widely used and reliable methods for mitigating nuisance odors (Pillai et al., 2012; Alfonsín, 2015; Boraphech & Thiravetyan, 2015; Zaman & Rosli, 2015). The odor treatment technologies are classified mainly into hybrid (e.g., physical and chemical) and biological techniques. Physical and chemical methods have high abatement efficiency and

robustness when operated and maintained correctly, low empty bed retention time, and rapid start-up. These techniques are often based on the transfer of odorants from the gas emission to either a solid (adsorption) or liquid (absorption) phase (Ren and Zhao, 2019). On the other hand, biological techniques constitute a more costeffective and environmentally friendly alternative (Oliva et al., 2019), but they can present significant investment costs (Senatore et al., 2021). In recent decades, biological techniques have been increasingly widely used to efficiently treat pollutants with lower operating costs (Dubey and Shiwani, 2012; Taghizadeh et al., 2012).

In recent years, simple surface foam-spraying technology has been regarded as a promising method for physically suppressing odor and its subsequent degradation by bacteria. Bajagain uses durable surfactant foam, including bacteria that were sprayed on swine manure. The surfactant foam acted as a physical barrier, suppressing NH₃ release, and the aqueous-phase bacteria formed after the foam breaking infiltrated the waste and degraded NH₃ (Bajagain et al., 2022).

Furthermore, deodorant is an effective way to eliminate odor pollution. Using plant material-based products, extracted from plants such as acacia or bamboo, in odor abatement is an effective technique for treating gaseous emissions from composting facilities (Zhou et al., 2012). Deodorant is an end-of-pipe treatment technology that combines several deodorization principles. Through oxidationreduction chemical reactions, decomposition reactions, neutralization reactions, etc., the deodorant transforms the fetid stench into an odorless gas, directly eliminates the odor, changes the adsorption odor molecule stereo configuration, weakens the molecular association bond, decreases the stability of odor molecules, or makes it easy to react with other molecules. For example, hydrogen sulfide reacts with deodorant to form sulfuric acid ions and water, while ammonia can form nitrogen and water (Dai et al., 2018; Yin and Lu, 2021). With the increasing awareness and demand for environmental protection, plant deodorants using plant extracts as the main ingredient have become a common theme in research. Zhou uses products derived from plants such as acacia or bamboo to eliminate odors from gas emissions from composting facilities (Zhou et al., 2012). Hu's research shows that 10 kinds of plant-based deodorants have a removal effect on hydrogen sulfide, with the dilution of a more significant number of deodorants, mainly essential oils of plants, to remove the highest amount-the highest rate being 84%-87% (Hu, 2022). The active compounds in the extracted materials, such as tea polyphenols and tannins, can react with odorous gases and eliminate them (Pang and Yu, 2019).

Sprays or atomizing equipment are efficient devices to atomize the liquids, which can enhance the deodorant to infiltrate and decompose odor molecules. The active compound complex odor molecules destroy bonds of odor molecules, decompose odor molecules, and generate non-toxic and odorless products (Jin and Meng, 2019). This achieves the effect of eliminating odors and exterminating the source of odors, which is an essential difference from the products that use a fragrance to cover abnormal and unpleasant odors. The nanometer-level effective molecules infiltrate and destroy the cell walls of bacteria and viruses, thus collapsing their living environment to realize effective sterilization and disinfection. Through the effective absorption, decomposition, and neutralization of suspended particles and dust, the problem of dirty air is eliminated, and the air quality is improved (Nechifor and Cotorcea, 2021). The odor abatement method by spraying deodorant does not need an odor

source sealing system and odor collection and transmission system to be installed; it only requires an odor treatment system. There is no need for civil works, collection systems, or overhead discharge pipelines, and there is no secondary pollution. The system is easy to install and use, without too much time and expense. It is not necessary to install any other accessories to handle the air (Zhang and Li, 2022). The system is easy to move, and the device can be moved and changed according to the odor source. The equipment for deodorizing natural plant extracts is simple and easy to use, requires no maintenance, and is fully automatic.

In this study, a composting facility was selected to observe the change in the odor gas concentration after applying the abatement method to evaluate the odor removal performance of plant extracts. The odorant removal efficiency was estimated by an on-site sampling campaign to observe the elimination or masking effect. The composting facility used in the study is located in a suburb of Shanghai, with the waste left there for 20 days for aerobic treatment after sorting. The plant extract was sprayed directly onto a windrow in the composting facility through pressure piping, with a spraying frequency of three times a day (9:00, 12:00, and 15: 00). Each spray lasted for 15 min. Before and after the spraying trial, odor compounds were sampled in Tedlar[®] bags or cartridges close to the respiratory zone using four methods.

2 Materials and methods

2.1 Deodorant

The deodorant was self-formulated. It was based on a commercially available plant essential oil. It was a mixture of coconut, lemon, and tea tree essential oils (Shanghai Zhenfeng Environmental Protection Technology Co., Ltd., China) with trimethylene glycol as the solvent. The ratio of trimethylene glycol to plant essential oil was 10:1. After mixing evenly, a certain proportion of emulsifiers and additives were added, and the final product was prepared with deionized water, 0.6% compound essential oil, 6% trimethylene glycol, 1% β -cyclodextrin, and 3% Tween-80.

2.2 Site description and sampling location

The treatment capacity of the composting plant was 1,200 t/d, with a working area of 20,000 m², as shown in Figure 1. The waste was left in place for 20 days for aerobic treatment after sorting. Briefly, the municipal solid waste (MSW) was delivered to the facility, where the waste was sorted manually and then separated by sizes using trommel screens. The light fractions, such as plastics, were picked up by an air separator, and metals were picked up by a magnetic separator. After the pre-mechanical treatment, the waste was conveyed to the prebiotreatment facility for 20 days of aerobic treatment. Next, the waste was sorted again in a post-mechanical treatment facility, after which undersized material was delivered to a post-biotreatment facility for another 30-60 days of aerobic treatment. The spraying deodorant system and atomizing nozzles were arranged in the pipeline above the windrow in the dotted rectangle area, and the leachate was collected from the bottom of the pile for treatment using a biofilter, as shown in Figure 2. The operation time of the spraying system was generally 2 min/h, of which 10 nozzles require about 0.4 L liquid per hour. The monthly cost of the deodorant is about 5,000 RMB (renminbi). Before



FIGURE 1

Composting windrow and sampling operation in the composting plant.



the deodorant spraying trial, a sampling campaign was conducted as a control in the biotreatment and fermentation process in the facility at 9:00, 12:00, and 16:00, respectively. After spraying, the sampling

campaign was repeated to detect the concentration to evaluate removal efficiencies. The campaign was completed within 1 day on September 19th, 2021.

2.3 Gas sampling and analyses

Five analytical methods were used to measure the odor pollutant concentration before and after spraying.

- 1) Colorimetric tubes to detect ammonia.
- 2) Gas chromatography flame ionization detection (GC-FID) is preceded by a cold trap concentration for aromatics, terpenes, and alcohols.
- Gas chromatography-pulsed flame photometric detection (GC-PFPD) was preceded by the cold trap concentration for sulfur compounds.
- 4) Gas chromatography-flame ionization detection was preceded by a sorbent concentration to determine volatile fatty acids (VFAs).
- 5) High-performance liquid chromatography (HPLC) after the derivation by 2,4-dinitrophenylhydrazine (DNPH) to detect aldehydes.

2.3.1 Analyses of colorimetric tubes

Colorimetric tubes (Gastec., Japan) were used to detect ammonia in the field. A certain volume (1.5 L or 3 L) of gas was extracted with a hand pump attached to the tube. The tube color changed, and the length of the color change indicated the measured concentration.

2.3.2 Gas chromatography analyses after the cold pre-concentration method

Gas chromatography analysis after cold pre-concentration was determined for the most volatile compounds. Tedlar[®] bags were used to collect samples from the air. First, the bags were placed in a vacuum-generating container, then a vacuum pump was used to connect the container, and the samples were drawn directly into these bags *via* a Teflon tube. These samples were transported immediately (within 24 h) to the laboratory for analysis to minimize the loss of reduced sulfur compounds (RSCs) during storage. The pre-concentrated method was carried out by cryogenic liquid nitrogen according to the EPA TO-15 method. The experiment instrument was Entech 7100A (Entech Instruments Inc., United States), and the injection volume was 50–500 ml. The analysis conditions for GC-PFPD and GC-FID were reported by Fang et al. (2012, 2013).

2.3.3 Gas chromatography-flame ionization detection analysis after the sorbent concentration method

The sorbent concentration method was used to determine VFA concentrations in air samples. Commercial sorbent tubes (silica gel tube, SKC, United States) were used to collect the air samples. The airflow rate was 1,000 ml/min, and the collection time was 120 min. After sampling, the tubes were capped with end plugs and transported back to the laboratory for analysis. The silica gel was placed into a 5-ml volumetric flask and desorbed with 5 mL deionized water. After standing for 30 min in an ultrasonic instrument, the supernatant liquid was analyzed by GC-FID. The operating parameters for the FID detection system were the same as those described previously.

2.3.4 High-performance liquid chromatography analysis after derivation by DNPH

High-performance liquid chromatography after derivation by 2,4dinitrophenylhydrazine (DNPH) was used to detect aldehydes. Commercial cartridges (Cleaner DNPH-Silica, Agela Technology, China) were used to collect air samples. The airflow rate was 1,000 ml/min, and the collection time was 30 min. After sampling, cartridges were connected to a clean syringe (Visiprep DL, Supelco Analytical, United States) and placed on a solid phase extraction (SPE) vacuum manifold (Visiprep, Supelco Analytical, United States). The compounds absorbed into the cartridges were then eluted into a 5-ml volumetric flask with 5 ml acetonitrile. The analytical column used was a C18 (4.6 mm ID \times 25 cm; 5 µm) stainless steel tube (Venusil XBP, Agela Technology, China). The conditions of the method were reported by Fang et al. (2012, 2013).

2.4 Quality assurance and control

Five levels (50, 200, 500, 1,000, and 2,000 μ g/m³) of standard gases (Air Liquid, France) were used to generate standard calibration curves. A good correlation (R² > 0.93) was found in the range of 0–2,000 μ g/m³. Instrumental detection limits were determined by linear extrapolation from the lowest standard in the calibration curve using the area of a peak having a signal-to-noise ratio of 3. The concentrations of blank samples were less than 1 μ g/m³, which implied that no contamination of the samples occurred during collection, transportation, and storage processes. Ten replicate standard gases were used to evaluate the reproducibility, and the relative standard deviations were less than 6%. The standard recovery rate of these methods was 80%–112%.

2.5 Theoretical odor concentrations

The theoretical odor concentration (C_{od}) was used to evaluate the odor abatement effect. It was calculated from the ratio of the analytical concentration of each odor compound to its odor threshold, according to Eqs 1, 2 (Fang et al., 2012). This method can be used to compare the odor elimination effect of spraying activities.

$$C_{od,i} = \frac{C_i}{OT_i} \tag{1}$$

$$C_{od} = \sum_{i=1}^{n} \frac{C_i}{\mathrm{OT}_i},\tag{2}$$

where C_i is the analytical concentration of the *i*th compound (μ g/m³), OT_i is the odor threshold value of the *i*th compound (μ g/m³), $C_{od,i}$ is defined as the theoretical odor concentration (dimensionless), *n* is the total number of odorous compounds, and C_{od} represents the sum of the theoretical odor concentrations of *n* compounds.

3 Results

3.1 The change in concentration of each compound in the trial

The concentration of every compound before and after spraying is shown in Figure 3–9, which show the average values of three measurements. In the carbonyl group, acetaldehyde, crotonaldehyde, and pentanone concentrations decreased, while the acetone and butyraldehyde concentrations increased to different degrees. There were no noticeable changes for the other compounds. For VFAs, the solution had only a slight effect on









most compounds except acetic acid. The most effective removal was achieved for butanoic acid. Substantial changes in the concentrations of alcohols were observed in this trial, especially propanol, which fell from 28,115 to $23 \ \mu g/m^3$. This was followed by 2-pentanol, which dropped from 1,439 $\ \mu g/m^3$ to $9 \ \mu g/m^3$. For the aromatics, the solution impacted toluene, m-xylene, and styrene. Their concentrations decreased from 2,850, 2,647, and 3,733 $\ \mu g/m^3$ to 244, 1,385, and 164 $\ \mu g/m^3$, respectively. For terpenes, the solution had a slight effect on α -pinene and d-limonene concentrations, with a decline from 816 and 741 $\ \mu g/m^3$ to 333 and 585 $\ \mu g/m^3$, respectively.

3.2 The purification efficiency

The changes in the concentrations of the seven groups and total compounds are shown in Figure 8; Table 1. The figure shows that the extraction had a good efficiency for the total. After spraying, the ammonia concentration decreased from 10,000 μ g/m³ to 7,000 μ g/m³. There was a significant decline in alcohols, aromatics, and carbonyls from 30,612, 12,270, and 5,138 μ g/m³ to 4,965, 4,669, and 2,253 μ g/m³, respectively. The removal efficiency of terpenes, VFAs, and sulfur compounds were not as good. A decline of 40,703 μ g/m³ was observed





for the total compounds, indicating that the plant material-based solution played an important role in eliminating the odorous gases.

The purification efficiency is shown in Table 1. The results showed that the solution could efficiently remove alcohols, aromatics, and carbonyls, achieving removal efficiencies of 83%, 62%, and 56%, respectively. The removal efficiencies of terpenes, VFAs, and sulfur compounds were lower at only 36% and 25%, respectively. After atomization by deodorization equipment, the natural plant extract formed a mist, with liquid droplets' radius ≤ 0.04 mm. The droplets had a large specific surface area and large surface energy, with an average kinetic energy per mole of several tens of kcal. Energy levels of



TABLE 1 Purification efficiency of seven groups of compounds.

Odor group	Concentration (µg/m ³)		Efficiency (%)
	Before	After	
Ammonia	10,000	7,000	30.0
Carbonyls	5,138	2,253	56.2
VFAs	1,985	1,280	35.5
Alcohols	30,612	4,965	83.8
Aromatics	12,270	4,669	61.9
Terpenes	1,837	1,179	35.8
Sulfur compounds	845	638	24.5
Total	62,688	21,984	64.9

this magnitude are 1/3 to 1/2 of the bond energy of many elements (Kharat et al., 2020). Researchers found that monophenols, thymol, and carvacrol in essential oils exhibited odor-reducing activities (Nakasugi et al., 2017). The surface of the solution could, therefore, not only effectively adsorb odor molecules in the air but also change the three-dimensional configuration of the adsorbed odor molecules, weaken chemical bonds in odor molecules, and increase the instability of odor molecules.

3.3 The odor abatement effect

The odor abatement effect was evaluated by the theoretical odor concentration (C_{od}). The *total* C_{od} decreased from 23,757 at the beginning to 13,056 at the end, as shown in Figure 10. The decline order of C_{od} of gas groups was VFA > sulfur compounds > alcohols > aromatic hydrocarbons > terpenes > ammonia, showing that most of



the foul smell had been eliminated. The C_{od} of carbonyls increased exceptionally, mainly due to the increase of n-butyraldehyde after spraying. The cause may be that the deodorant promotes the release of n-butyraldehyde, but it may also have been caused by the interference of aldehydes in the deodorant.

4 Discussion

The composition of plant essential oils is complex, and it is challenging to determine the components; for example Artemisia argyi has more than 100 components. The deodorant detected many compounds that are reactive and antibacterial; a-terpineol, dihydrolaurenol, a-hexyl cinnamaldehyde, hexyl salicylate, lily aldehyde, and citral are volatile chemical components that are commonly found in plant essential oils. Methyl dihydrojasmonate, existing in jasmine oil and black tea, has a deodorizing and refreshing effect (Zhang et al., 2019; Jaouadi and Hajjikasmi, 2020; Huang et al., 2022). α-Hexyl cinnamaldehyde is present in black tea and orchid; it can sterilize, act as an antiseptic, and deodorize, and can replace the traditional fungicide glutaraldehyde. Cedrylene methyl ketone is found in cedar, which is used in daily chemical, sanitary, and sanitary products such as disinfectants, cosmetics, and washing products in the healthcare industry. An active substance with an unsaturated structure in these plant essential oils reacted with the malodorous substances, turning them into odorless compounds, and the foul smell further reduced.

Plant material-based deodorants require the addition of surfactants for emulsification and form an oil-in-water structure that facilitates the hydration of plant active ingredients in scattered water (Jarzębski et al., 2020). Emulsifiers are divided into anionic surfactants, cationic surfactants, amphoteric surfactants, and nonionic surfactants (Dammak et al., 2020). When anionic surfactants

dissolve in water and ionize, they are connected with the lipophilic groups that are negatively charged, including carboxylate, sulfate and sulfonate groups, acid and phosphate anionic surfactants, and representative compounds, such as sodium dodecylbenzenesulfonate and stearic acid. Cationic surfactants dissolve in water and ionize; the hydrophilic base band is linked to lipophilic groups that are positively charged. The representative compounds are ammonium salts and quaternary ammonium salts, such as benzalkonium chloride and benzalkonium bromide. There are many types of amphoteric surfactants, for example, lecithin, amino acid type, and betaine. BS12 is a more common amphoteric surfactant. Non-ionic surfactants include fatty acid glycerides, AEO-7, OP-10, and Tween-80. Most of the ammonium and quaternary ammonium compounds are highly toxic and contain halogen elements; we do not consider the use of such surfactants. Tween-80 has good compatibility and stability; there are more hydrophilic groups in Tween-80 molecules (polycarbonate base), which have strong hydrophilicity and can emulsify and dissolve essential oils in water and improve the deodorizing capacity.

Due to the different sources of various odors, their components are also different. Using one of the chemical deodorants makes it challenging to meet the requirements of environmental deodorization. Many plant material-based deodorants are extracted from roots, stems, and leaves of plants. The extract protects the remaining natural bioactive components that can synergistically produce various effects on odor molecules. Research on the formulation of compound deodorants will be an important task in the future. The direction of development should receive more attention. The deodorant function of the deodorant designed by the formulation system could be enhanced, and the applications would therefore be expanded. The formulation of deodorant mainly focuses on systematic ingredient selection, active ingredient active dosage determination, and auxiliary and stabilizer choices. The identification of deodorant active ingredients is a technical challenge, requiring a variety of techniques to isolate and identify the active ingredient, mainly solvent extraction with gas chromatography or liquid phase mass spectrometry (Shin et al., 2020). In the future, we may analyze the composition of active substances, component structures, and efficacy relationships, etc. In-depth research studies to optimize formulations and reduce product development costs are needed.

The non-ionic surfactant Tween-80 was used in the formulation of this study. It can reduce surface tension, form smaller droplets, and increase the contact area with odor molecules. β -cyclodextrin and its derivatives are cyclic oligosaccharides with nanoscale porous molecular structures, the cavities of which can combine some small molecules and adsorb and release a variety of substances (Liu et al., 2020; Tian et al., 2021; Chodankar et al., 2022; Wang et al., 2022). A new adsorption-type deodorant is more and more widely used; β -cyclodextrin is added to the deodorant as an adsorptive modifier to enhance and improve the performance of the deodorant. Still, the amount of β -cyclodextrin should not be too much. Since the solubility of β -cyclodextrin in an aqueous solution is slight, adding too much will cause precipitation. It will affect the dispersion of the system. Further research is required on the compound system to enhance the solubility of β -cyclodextrin in the future.

In this study, plant essential oils were used as active ingredients. The deodorant contains highly reactive functional groups, which can effectively trap, quickly adsorb, and assimilate odorous substances, and can also interact with odorous molecules through a chemical reaction. Then, it will turn into an odorless substance to inhibit odors. This plant material-based deodorant can be sprayed in spaces to effectively clean odorous gases from composting facilities.

A plant material-based solution consisting of plant extracts was effective for removing toluene, xylene, styrene, α -pinene, d-limonene, propanol, pentanol, and butanoic acid from the air. Since the main odorants produced in the composting plant could be effectively removed by applying the solution, the odor reduction mode was removing odorous compounds rather than masking.

Data availability statement

The original contributions presented in the study are included in the article/Supplementary Material; Further inquiries can be directed to the corresponding author.

Author contributions

JW was responsible for interpreting results, updating reference lists, and creating "Summary of findings" tables. JY was responsible for writing the manuscript. JF was responsible for conducting the research. JW was responsible for screening potentially eligible studies and extracting and analyzing data. LF was responsible for extracting and analyzing data and interpreting results.

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Conflict of interest

The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

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