



Impacts of pH on the Fitness and Immune System of Pacific White Shrimp

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The atmospheric partial pressure of CO₂ (pCO₂) has been increasing dramatically since the beginning of the industrial revolution and about 30% of the CO₂ produced by anthropogenic activities was absorbed by the ocean. This led to a perturbation of the seawater carbonate chemistry resulting in a decrease of the average surface ocean pH by 0.1 and termed ocean acidification (OA). Projections suggest that pCO₂ may reach 900 µatm by the end of the twenty-first century lowering the average pH of the surface ocean by 0.4 units. The negative impacts of OA on many species of marine invertebrates such as mollusks, echinoderms, and crustaceans are well documented. However, less attention has been paid to the impacts of low pH on fitness and immune system in crustaceans. Here, we exposed Pacific white shrimps to 3 different pHs (nominal pH 8.0, 7.9, and 7.6) over a 100-days experiment. We found that, even though there were no significant effects on fitness parameters (survival, growth and allometries between length and weight), some immune markers were modified under low pH. A significant decrease in total hemocyte count and phenoloxidase activity was observed in shrimps exposed to pH 7.6 as compared to pH 8.0; and phagocytosis rate significantly decreased with decreasing pH. A significant increase in superoxide production was also observed at pH 7.6 as compared to pH 8.0. All these results suggest that a 100-days exposure to pH 7.6 did not have a direct effect on fitness but lead to a modulation of the immune response.

Keywords: pCO_2 , ocean acidification, seawater carbonate chemistry, growth, energy budget, dissolved inorganic carbon

INTRODUCTION

The atmospheric CO₂ partial pressure (pCO₂) has been increasing dramatically since the beginning of the industrial revolution (Feely et al., 2004) and about 30% of the CO₂ produced by anthropogenic activities was absorbed by the ocean (Sabine et al., 2004). This led to a drop of the average pH of the surface ocean by 0.1 units (Orr et al., 2005). It is projected that atmospheric CO₂ partial pressure may reach 900 μ atm by the end of the twenty-first century lowering the average pH of the surface ocean pH by 0.4 units from its pre-industrial value (Caldiera and Wickett, 2003; Feely et al., 2008; IPCC, 2014). This gradual lowering of pH in the ocean and associated changes in carbonate chemistry of seawater are termed ocean acidification (OA). As a consequence of the declining pH in seawater, relative abundance of dissolved inorganic carbon (DIC) species shift,

OPEN ACCESS

Edited by:

Guang Gao, Xiamen University, China

Reviewed by:

Meng-Chou Lee, National Taiwan Ocean University, Taiwan Pei-Jie Meng, National Dong Hwa University, Taiwan

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Specialty section:

This article was submitted to Global Change and the Future Ocean, a section of the journal Frontiers in Marine Science

> Received: 28 July 2021 Accepted: 13 September 2021 Published: 05 October 2021

Citation:

Weerathunga V, Huang W-J, Dupont S, Hsieh H-H, Piyawardhana N, Yuan F-L, Liao J-S, Lai C-Y, Chen W-M and Hung C-C (2021) Impacts of pH on the Fitness and Immune System of Pacific White Shrimp. Front. Mar. Sci. 8:748837. doi: 10.3389/fmars.2021.748837

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resulting in increased aqueous CO_2 and HCO_3^- ion concentration and decreased in CO_3^{2-} ion concentration and calcium carbonate saturation state (Orr et al., 2005).

A large body of evidence is documenting the biological impact of OA, ranging from negative to positive (Wittmann and Pörtner, 2013). Early works revealed inter- and even intra-species responses while other parameters such as lifehistory stages, exposure duration, tested level pH and other drivers were modulating the response. More recently, local adaptation to present natural variability in pH or pCO_2 has been identified as a key predictor of species sensitivity to ocean acidification (Vargas et al., 2017). This also highlighted the importance to consider environmental variability while designing and interpreting experimental work. While the main focus of the early work was on calcification, immune defense and energy budget appeared to play a key role in species response to OA. Immune suppression under OA was observed in many invertebrate species including Norway lobsters (Hernroth et al., 2012), common starfish (Hernroth et al., 2011), blue mussels (Ellis et al., 2015; Sun et al., 2017), blood clams (Liu et al., 2016) and Pacific oysters (Cao et al., 2018). However, recent studies showed enhanced immune response in east pacific red octopus (Culler-Juarez and Onthank, 2021) and in Chinese mitten crabs (Luo et al., 2021) exposed to OA. Migliaccio et al. (2019) found phenotypic plasticity in the immune system of purple sea urchins living nearby CO₂ vents (pH 7.8 \pm 0.2) as compared to the same species living in high pH coastal environments (pH 8.02). In addition, enhanced energetic costs or alterations in energy allocation to different metabolic functions with a consequence for the energy budget of marine invertebrates was observed (Bibby et al., 2008; Stumpp et al., 2011; Mackenzie et al., 2014; Liu et al., 2016; Meseck et al., 2016; Cao et al., 2018).

In many invertebrates, hemocytes are the main cells that act as the defender of the organisms against pathogens and other foreign bodies (Rowley, 2016). There are three distinct hemocyte types in crustaceans: hyaline, granular, and semi granular cells (Söderhäll and Smith, 1983; Ratcliffe et al., 1985). Hyaline cells involve in immune defense act as phagocytic cells (Söderhäll and Smith, 1983), and produce reactive oxygen species (ROS) such as superoxide ions (Bell and Smith, 1993; Song and Hsieh, 1994; Muñoz et al., 2000), a type of microbicidal radical which can directly destruct the engulfed or encapsulated pathogen by the cell (Holmblad and Söderhäll, 1999). As these superoxide ions are toxic to living cells of the organism itself, an antioxidant system which is known as superoxide dismutase (SOD) scavenges the excess radicals to form hydrogen peroxide and oxygen (Holmblad and Söderhäll, 1999). On the other hand, granular and semi granular cells also play a role in the immune response by encapsulation, nodule formation, degranulation, and conversion of prophenoloxidase into melanin which shields off the infected region of the organism (Rowley, 2016).

We exposed pacific white shrimps to different nominal pHs (8.0, 7.9, and 7.6) relevant in the context of present and future pH variability over a 100-days experiment. To understand the impact on fitness, we measured survival and growth. Four basic hemolymph immune markers, playing a significant role

in the survival and health of the organism were measured: total hemocyte count (THC), superoxide ion production (SOP), phenoloxidase activity (PO activity), and phagocytosis (PGT).

MATERIALS AND METHODS

Organisms and Experimental System

Juvenile Pacific white shrimps, litopenaeus vannamei were obtained from a local shrimp farm in Taiwan and 200 of them having similar weights and lengths were kept for 2 weeks in a 1,000 L tank (inside the experiment premises) before starting the experiment. During that phase, shrimps were fed with commercially available shrimp's diet (major component: 38% of crude proteins) up to 4% of the total body weight per day. Salinity, temperature, and pH of the tank were 31.02 ± 0.21 , 25.93 ± 1.56 $^{\circ}$ C, 8.07 \pm 0.04, respectively. After measuring the initial weights (Shimadzu UniBloc, readability: 0.01g), nine random groups of 15 shrimps were moved into 9 glass 100-L tanks. There was no significant difference [ANOVA: $F_{(2, 8)} = 0.741$, *p*-value = 0.655] in initial weights of shrimps among 9 tanks. The initial average weight and length of shrimps were 9.3 ± 2.4 g and 10.1 ± 1.0 cm, respectively. Natural seawater pumped from the coast of the Sizihwan (passed through filters) was continuously flowed at a rate of 750 ml/min in each tank, separately and the system was maintained over 100-days (from 27th February 2020 to 05th of June, 2020). Shrimps were fed ad libitum with the previously mentioned diet four times per day. Three nominal pH values were selected (8.0, 7.9, and 7.6) covering present and near-future natural variability in the coastal zone. Seawater pH was decreased in a stepwise manner by 0.05 units per day until it reached the target pH to limit the stress of sudden increased acidity. Tank pH was maintained by diffusing pure CO₂ using a pH controller (Leilih P-LE-08, resolution 0.01 pH units, Taiwan) separately for each tank. pH probes of controllers were calibrated at weekly intervals using NIST buffer solutions (PanReac AppliChem, ITW Reagents, Germany).

In a preliminary experiment, a seawater flow rate of 750 ml per min was identified as optimal to maintain experimental conditions in the 100 L tanks (stable pH, pCO₂, dissolved oxygen, nitrite and ammonia). As the source water for each tank always had a pH value above 8.00, the pH of seawater in nominal pH 7.9 and 7.6 tanks were always in favor to increase. When the pH was above the target pH, the pH controller ensured diffusion of CO₂ into the tank until the desired pH was reached. Each tank was equipped with a separate water filter to maintain the water quality. Water was drawn from the bottom, passed through the filter, and released to the surface. This bottomup mixing ensured the homogeneity of pH inside the tank. Further, we cleaned each tank two times per day to avoid the accumulation of organic matter. Filtration, high enough flow rate, and frequent cleaning were also important to maintain high dissolved oxygen concentrations in our experimental system. Therefore, even without any external oxygenation to tanks, the dissolved oxygen concentration was closer to 5 mg/L (Table 1). Experiment was conducted in-door and covering tank walls in black to ensure stable pH controlling.

Seawater Physico-Chemical Measurements

Weekly water samples were taken from each tank into 100 ml borosilicate bottles to determine the dissolved inorganic carbon (DIC) and pH. Samples were quickly poisoned with 60 µl of saturated HgCl₂ and kept in dark pending measurements to prevent any biological activities. Sample storage and carbonate chemistry parameters analysis were carried under room temperature of 24-26°C at National Sun-Yat sen University, Taiwan. DIC was analyzed using Apollo SciTech's AS-C3 DIC analyzer (Newark, DE, United States, precession $\pm 2 \,\mu$ mol/kg) connected LI-COR 7000 CO2/H2O analyzer (the United States, accuracy within 1% nominally) and the quality check was carried out during the measurements using certified reference materials (batch #185 and #186) from Dr. A. Dickson's laboratory at the Scripps Institution of Oceanography. Thermo Scientific Orion Star A211 benchtop pH meter was used to measure the pH of the sample and the pH sensor was calibrated by using three different buffer solutions, pH 4.01, pH 7.01, and pH 9.00 (HANNA, NIST traceable buffer solutions, the accuracy of \pm 0.01 pH units at 25°C). Temperature (WTW Cond 3110 SET 1, Germany, accuracy \pm 0.1°C), salinity (WTW Cond 3110 SET 1, Germany), and dissolved oxygen concentration (WTW Oxi 3310 Set 1, Germany, accuracy \pm 0.5% of reading) of each tank were recorded concurrently with seawater sampling. CO2 partial pressure (pCO_2) of water samples was calculated using CO2Sys_v2.1 (Pierrot et al., 2006) with dissociation constants (K1 and K2) from Mehrbach et al. (1973) refitted by Dickson and Millero (1987), the dissociation constant for hydrogen sulfate ion HSO₄⁻ from Dickson (1990) and [B]I value from Lee et al. (2010).

Determinations of dissolved inorganic nitrite (NO_2^-) and ammonia (NH_4^+) were done with the colorimetric methods as

described by Pai et al. (2001) and Grasshoff et al. (2009) on samples taken simultaneously using acid cleaned polyethylene bottles of 100 ml capacity.

Biological Measurements

Shrimp Growth Parameters

Survival (in %) was calculated for each replicate as the percentage of shrimps surviving the duration of the experiment. Wet weights (in g) and lengths (in cm) of each shrimp were measured every 2 weeks (including beginning and end) throughout the experiment. The wet weight of each shrimp was measured inside a water beaker (the same water from the tank to which the shrimp belongs) using a Shimadzu UniBloc balance. Lengths were measured outside the water using a ruler. Growth rate (in cm day⁻¹ and g day⁻¹) was calculated for each replicate as the coefficient of the significant linear relationship between length or weight and time (in days; see **Supplementary Tables 1, 2**). Length and weight were correlated following significant exponential relationships (see **Supplementary Table 3**).

Shrimp Immune System Parameters

Immune system parameters: total hemocyte count (THC), phenoloxidase (PO) activity, superoxide (O_2^-) production (SOP), and phagocytosis (PGT) were measured at the end of the experiment after withdrawing 0.3 ml volume of hemolymph from the ventral sinus (near the 1st abdominal segment) of each shrimp. Hemolymph samples were immediately transferred into falcon tubes filled with 2.7 ml of pre-cooled (~10°C) anticoagulant solution (0.01 M Tris-HCl, 0.25 M sucrose, 0.1 M sodium citrate; pH 7.6) as in Song and Hsieh (1994).

To determine THC, 10 μ l of a diluted blood sample was transferred into a hemocytometer (Neubauer, Germany), and the number of hemocytes in 5 large squares (1 \times 1 mm²) were

TABLE 1 | Physico-chemical parameters for each nominal pH treatment and results from respective analyses of variance testing the impact of pH treatment and difference between replicates.

	Nominal pH			ANOVA 2		
	8.0	7.9	7.6	Model [F _(8, 98) ; p)	pHnom [F ₍₂₎ ; <i>p</i>]	Replicate [<i>F</i> ₍₂₎ ; <i>p</i>]
pH _{NIST}	8.02 ± 0.02	7.87 ± 0.01	7.60 ± 0.02	F = 41.6 p < 0.0001	F = 165.4 p < 0.0001	F = 0.4 p = 0.90
DIC (μ mol kg ⁻¹)	2088 ± 12	2128 ± 7	2226 ± 9	F = 13.8 p < 0.0001	F = 53.6 p < 0.0001	F = 0.5 p = 0.79
Temperature (°C)	27.6 ± 0.2	27.6 ± 0.2	27.6 ± 0.2	F = 0.01 p = 0.99		
Salinity	32.7 ± 0.2	32.7 ± 0.2	32.7 ± 0.2	F = 0.01 p = 1		
DO (mg L ⁻¹)	4.98 ± 0.10	4.94 ± 0.09	4.81 ± 0.10	F = 0.70 p = 0.69		
NH ₄ (μM)	7.59 ± 0.48	7.08 ± 0.33	7.19 ± 0.41	F = 0.86 P = 0.55		
NO ₂ (μM)	1.44 ± 0.11	1.50 ± 0.11	1.56 ± 0.11	F = 0.15 P = 0.99		
ρCO ₂ (μatm)	654 ± 40	935 ± 22	1882 ± 70	F = 42.5 p < 0.0001	<i>F</i> = 169.1 <i>P</i> < 0.0001	F = 0.28 p = 0.94

Measured pH on the NIST scale (pH_{NIST}), dissolved inorganic carbon (DIC), temperature, salinity, dissolved oxygen (DO), NH₄ and NO₂; and calculated pCO₂.

counted using a Nikon microscope (Eclipse 80i) equipped with Olympus microscope digital camera (UC90). Cells located at lower and right-hand borders were not included in the count. As each cell has a capacity of 0.1 μ l, cell density (THC) can be calculated by using the equation (1), where χ cells = average cells per square,

$$THC = \chi \text{ cells} \times 10^4 \text{ cells/ml}$$
(1)

Samples were then diluted with the anticoagulant solution to ensure all blood samples having an equal cell density of 10^6 cells/ml.

PO activity was measured based on spectrophotometric methods described in Hernández-López et al. (1996) and Liu and Chen (2004) with a slight modification to the centrifugation force following Yeh and Chen (2009). Three replicates were tested for each shrimp to ensure the data quality. For each replicate, two samples were prepared: one for measuring PO activity and one for measuring the background PO activity. A 50 µl aliquot of diluted hemolymph sample was transferred into a sterilized Eppendorf tube and centrifuged at 800 \times g at 4°C for 20 min (Eppendorf centrifuge 5810R, Germany). The supernatant was carefully discarded using a micropipette, and pellets were again centrifuged (800 \times g at 4°C for 20 min) after adding 500 μl of cacodylate-citrate buffer (0.01 M sodium cacodylate, 0.45 M sodium chloride, 0.10 M trisodium citrate, pH 7.0). After removing the supernatant, pellets were re-suspended in 100 µl cacodylate buffer solution (0.01 M sodium cacodylate, 0.45 M sodium chloride, 0.01 M calcium chloride, 0.26 M magnesium chloride, pH 7.0). Subsequently, all samples were moved into a 96-well plate (Thermo Scientific, Denmark). For PO activity measurements, wells were treated with 50 μ l of trypsin (1 mg of trypsin in 1 ml of cacodylate buffer). For PO background activity measurements, wells were treated with 50 µl of cacodylate buffer. After keeping 10 min at 25°C, 50 µl of L-DOPA (3 mg of L-dihydroxyphenylalanine in 1 ml of cacodylate buffer) was added to each well and kept for 15 min at 25°C. Finally, optical density (OD) of produced dopachrome from L-DOPA was recorded at a wavelength of 492 nm using an ELISA reader (M965, Metertech, Taiwan, accuracy \pm 0.005 Abs). PO activity (in absorbance units) in shrimp hemolymph was calculated as follows,

PO activity = (measured OD_{492}) - (background OD_{492}) (2)

SOP was measured in diluted hemolymph samples according to the method from Song and Hsieh (1994) with some adjustments from Liu and Chen (2004). Three replicates were done for hemolymph from each shrimp. For each replicate, two samples were prepared: one for measuring induced SOP while the other for measuring the background SOP. Initially, 100 μ l aliquot of the diluted hemolymph sample was transferred into a sterilized Eppendorf tube and was centrifuged at 700 \times g at 4°C for 20 min. The supernatant was discarded and 100 μ l of 0.1% zymosan was added to samples measuring induced SOP while background samples were treated with 100 μ l of MCHBSS solution (modified complete Hanks's Balanced Salt solution having 10 mM CaCl₂, 3 mM MgCl₂, and 5 mM MgSO₄). After incubating samples at room temperature for 30 min, the supernatant was removed, and pellets were washed three times with 100 % methanol. All samples were stained with 100 μ l of 0.3 % NBT (nitroblue tetrazolium) solution and kept for 30 min at room temperature. Excess NBT solution was removed, and remnants were fixed with 100 % methanol. Subsequently, samples were rinsed three times with 100 μ l of 70 % methanol solution and air-dried for 30 min. One hundred and twenty microliter of 2 M KOH and 140 μ l DMSO (dimethyl sulphoxide) solutions were added to samples to dissolve formazan. After mixing (Thermolyne Maxi Mix II Vortex Mixer), solutions were transferred into a 96-well plate (Thermo Scientific, Denmark), and optical density was recorded at a wavelength of 630 nm using an ELISA reader (M965, Metertech, Taiwan, accuracy \pm 0.005 Abs). Superoxide (O₂⁻) production (in absorbance units) was expressed as a ratio of OD between the induced sample and background sample as follows,

Superoxide production =
$$\frac{OD_{630} \text{ of induced sample}}{OD_{630} \text{ of backgroud sample}}$$
 (3)

Phagocytosis (PGT) activity was measured based on methods from Barracco et al. (1999); Ordás et al. (2000), and Chen et al. (2005) with some modifications. After slides were rinsed with 75% ethanol, 200 µl of diluted hemolymph samples were spread well on slides and kept for 1 h at room temperature. Likewise, triplicates were prepared from the hemolymph sample of each shrimp. Thereafter, slides were washed with MCHBSS buffer, and 200 μ l of latex beads (0.8 μ m, 3 \times 10⁷ beads/ml) was added to each slide and incubated for 1 h at room temperature. Slides were then washed with MCHBSS buffer and fixed with 100% alcohol. After 5 min, slides were washed twice with sterilized deionized water, stained with 5 % Giemsa solution (kept 20 min), and washed with the sterilized deionized water again. After drying for 24 h, the total number of hemocytes and the number of hemocytes containing beads were counted using a Nikon microscope (Eclipse 80i) equipped with Olympus microscope digital camera (UC90). Phagocytosis (in %) was evaluated using the following equation,

Phagocytic activity % = $\frac{\text{Total phagocytic hemocytes}}{\text{Total hemocytes}} \times 100$ (4)

Statistical Analyses

Statistical analyses were performed using the SAS software. Physico-chemical parameters (pH, DIC, temperature, salinity, DO, NH_4^+ , NO_2^- , and pCO_2) and immune markers (THC, SOP, PO, and Phagocytic activity) were analyzed with a twoway analysis of variance (ANOVA) to test for significant differences between pH treatments and replicates. Linear and exponential regressions were used to calculate growth rate and the relationship between weight and size, respectively. One-way ANOVAs were then used to test the effect of pH on survival, growth rates, and allometries between weight and size. Scheffe's *post-hoc* tests were used when relevant to discriminate between treatments. Data were normally distributed according to the Kolmogorov-Smirnov test. Equality of variance was tested using the Levene median test. All data are expressed as average and standard error of mean and the significance level α was set at 0.05.





RESULTS

Seawater Physico-Chemical Measurements

Measured and calculated physico-chemical parameters are summarized in **Table 1**. Target pHs were reached and significant differences (ANOVA 2; p < 0.0001; **Table 1**) between nominal pH were observed for pH_{NIST}, DIC, and pCO_2 with no significant difference (p > 0.05; **Table 1**) between replicates. Temporal variations have been observed in both temperature and salinity from the last phase of winter (end of February 2020) to the middle of the summer (beginning of June 2020). Temperature and salinity ranged from 24.8 to 30.3°C and from 30.3 to 33.9, respectively, over the study period. However, there were no significant differences (ANOVA 2; p > 0.05; **Table 1**) between pH treatments and replicates for temperature, salinity as well as other parameters of water quality (DO, NH₄⁺ and NO₂⁻).

Biological Measurements Fitness Parameters

By the end of the experiment, $35.5 \pm 2.7\%$ of the shrimps survived with no significant difference between pH treatments [ANOVA: $F_{(2, 8)} = 0.06$, p = 0.94; **Figure 1**]. Further, there was no considerable difference in shrimp survival with time between 3 nominal pH groups (**Supplementary Figure 1**). Pacific white shrimp's survival rates in our experiment were lower than those in larger cement tank culturing systems (Araneda et al., 2008). This may be due to transparent glass material, surrounding noise, confined space, and limited indoor light conditions. Growth rate (cm day⁻¹ and g day⁻¹) was calculated for each replicate as the coefficient of the significant linear relationship of length (cm) or weight (g) with time (day; **Supplementary Tables 1**, **2**). Shrimp grew at a rate of 0.055 ± 0.002 cm day⁻¹ throughout the experiment with no significant difference between pH treatments [ANOVA: $F_{(2, 8)} = 0.37$, p = 0.71; **Figure 2**]. Weight of the shrimps in pH 8.0, pH 7.9, and pH 7.6 treatments were increased at rates of 0.134 ± 0.030 , 0.138 ± 0.030 , and 0.143 ± 0.022 g day⁻¹, respectively, and there was no significant difference among 3 nominal pH treatments [ANOVA: $F_{(2, 8)} = 0.075$, p = 0.928]. Weight (g) and length (cm) were correlated following significant exponential relationships (**Figure 3** and **Supplementary Table 3**) with an overall relationship as followed:

Weight
$$(g) = 1.393 e^{0.179 \text{ Length (cm)}}$$

There was no significant effect of pH on this relationship [ANOVA: $F_{(2, 8)} = 0.05$, p = 0.96].

Immune Markers

Overall, the total hemocyte count (THC) was 224.5 \pm 16.3 10^4 cells ml⁻¹. ANOVA 2 revealed an effect of pH [model: $F_{(8, 37)} = 3.53$, p = 0.006; pH: $F_{(2)} = 4.50$, p = 0.019] and replicate [$F_{(6)} = 2.85$, p = 0.026] on THC. The only significant difference showed by the *post-hoc* test was a higher THC at pH 8.0—replicate 2 (403.1 \pm 35.1 10^4 cells ml⁻¹) as compared to pH 7.6—replicate 3 (106.3 \pm 41.2 10^4 cells ml⁻¹) (**Figure 4**).

Superoxide production (SOP) was impacted by pH [model: $F_{(8, 38)} = 3.51$, p = 0.006; pH: $F_{(2)} = 13.21$, p < 0.0001] with no difference between replicate [$F_{(6)} = 0.71$, p = 0.64]. The *posthoc* test revealed a significant increase of SOP by 61% in shrimp exposed to pH 7.6 as compared to the two other pHs (**Figure 5**).

For phenoloxydase (PO) activity, pH had a significant effect [model: $F_{(8, 37)} = 3.66$, p = 0.005; pH: $F_{(2)} = 8.61$, p = 0.0012] but not replicate [$F_{(6)} = 1.24$, p = 0.32]. *Post-hoc* test revealed a 44% lower PO activity in shrimp exposed to pH 7.6 as compared to the two other pHs (**Figure 6**).

pH [model: $F_{(8, 35)} = 19.22$, p < 0.0001; pH: $F_{(2)} = 70.48$, p < 0.0001] but not replicate [$F_{(6)} = 1.79$, p = 0.14] had an effect on the phagocytic activity. The phagocytic activity significantly decreased with decreasing pH leading to a 20% decrease at pH 7.9 and a 40% decrease at pH 7.6 as compared to pH 8.0 (**Figure 7**).

DISCUSSION

Our results show no direct impact of the exposure to lower pH (7.9 and 7.6) on fitness parameters (survival, growth, and allometries). This is consistent with some previously published studies on crustaceans. Mantis shrimps did not show any changes in their survival or growth following exposure to reduced pH of 7.57 for 6 months (DeVries et al., 2016). Lowder et al. (2017) reported that California grass shrimps exposed to a pH of 7.48 survived 7 weeks without any significant difference in carapace length compared to the controller group (pH of 7.98). Other studies documented negative impacts at similar low pH levels.



Kurihara et al. (2008) found that Pacific grass shrimps raised under high pCO₂ (1,000 and 1,900 ppm) for 7 weeks experienced increased mortality while a decrease in growth rate was observed only at the extreme pCO₂ treatment. Decreased pH (7.8 and 7.5) both affected growth and increased mortality of both juvenile red king crabs and tanner crabs while compared to higher pH (8.04) (Long et al., 2013).

While pH had no significant effect on fitness in our experiment, effects of decreased pH were observed for immune markers. For the THC, only a mild trend was observed as only one replicate in pH 7.6 had significantly lower THC than one replicate on pH 8.0 (**Figure 4**). A decline in THC was also noted in Norway lobsters (Hernroth et al., 2012), Blue mussels (Sun et al., 2017), Blood clams (Liu et al., 2016), common starfish (Hernroth et al., 2011) and American lobsters (Harrington and Hamlin, 2019) exposed to low pH. A significant decrease of hemocyte numbers was observed in Pacific white shrimps exposed to pH 7.5 for 15 days compared to pH 8.0 (Pan et al., 2005) and the same species exposed to pH 6.5 for 120 h exhibited a significant decrease in THC as compared to shrimps raised at pH 8.2 (Li and Chen, 2008).

Our results show that exposure to pH 7.6 increased superoxide production as compared to the two other pH treatments (**Figure 5**). This is consistent with the available literature, mostly on mollusks, showing that decreased pH can stimulate oxidative stress in the marine organism through increased ROS production (Tomanek et al., 2011; Wang et al., 2016; Sun et al., 2017; Cao et al., 2018; Liao et al., 2019). Enhanced production of ROS such as superoxide ions and hydrogen peroxide were observed under various environmental challenges in marine invertebrates (Liu and Chen, 2004; Hannam et al., 2010; Hernroth et al., 2012; Wang et al., 2016; Sun et al., 2017; Cao et al., 2018). However, the protective response of the antioxidant system (SOD, CAT, several peroxiredoxins) by scavenging excess radicals, can be varied under different situations. In some cases, the antioxidant



system was overwhelmed by the elevated ROS production or it failed due to the influence of the respective stressors (Liu and Chen, 2004; Hannam et al., 2010; Hernroth et al., 2012; Wang et al., 2016; Sun et al., 2017; Cao et al., 2018) while in the other cases the antioxidant system successfully scavenged the excess radicals to protect organisms cell structure from the oxidative

damage (Tomanek et al., 2011; Hu et al., 2015). PO activity of shrimps exposed to pH 7.6 was lower than for shrimps maintained at pH 8.0 (**Figure 6**). This is consistent with observations on Atlantic blue crabs (Tanner et al., 2006) and Pacific white shrimps (Kathyayani et al., 2019) under low pH of 7.0 and 6.0, respectively. A decrease of PO activity was also shown in Atlantic blue crabs (Tanner et al., 2006) and Pacific white shrimps (Kathyayani et al., 2019) under hypoxic and elevated ammonia concentration, respectively. In crustaceans, semi-granular and granular immune cells are the predominant sites that store and activate the pro-phenoloxidase system into melanin (Johansson and Soderhall, 1989; Rowley, 2016), shielding the infected region of the cuticle (Hernroth et al., 2012). Therefore, the reduction of these cells in the







hemolymph can be lead to suppression in the production of PO enzyme, which is the central enzyme of the pro-phenoloxidase activation cascade system (Rowley, 2016). A similar argument was presented by Meseck et al. (2016) for Tanner crabs exposed to pH 7.5, mentioning that observed alterations of normal functions in semi-granular and granular cells under low pH, could suppress the full functionality of phenoloxidase regulation. Hernroth et al. (2012) hypothesized that the conversion of pro-phenoloxidase to phenoloxidase could be impaired in Norway lobsters under low pH conditions. However, while the hemocyte count declined by 50%, they observed no significant difference in the mean PO activity. They assumed that neglecting the degranulation process of hemocytes during the *in vitro* measurements led to that result. In our study, following the method described in Hernández-López et al. (1996) and Liu and Chen (2004), we have used centrifugation to induce the degranulation of hemocytes and we did observe both a decrease in THC and PO activity.

Phagocytic (PGT) activity in Pacific white shrimps was the most sensitive parameter to pH. Shrimp exposure to pH 7.9 led to a 20% decrease in phagocytic activity as compared to pH 8.0 while exposure to pH 7.6 led to a 40% decrease. Similarly, a decrease in phagocytic activity was observed in Pacific ovsters (Cao et al., 2018), blood clams (Liu et al., 2016), blue mussels (Bibby et al., 2008; Sun et al., 2017), common starfish (Hernroth et al., 2011) and Norway lobsters (Hernroth et al., 2012) exposed to low pH conditions. Cao et al. (2018) hypothesized that energy suppression in Pacific oysters could be linked to the reduction of phagocytosis. Liu et al. (2016) suggested that suppressed phagocytic activity in blood clams may be associated with the observed reduction in THC. Bibby et al. (2008) and Li et al. (2015) proposed that changes in concentration of Ca²⁺ in intracellular structures may suppress the phagocytosis in blue mussels and Akoya pearl oysters, as it is an important messenger in cell signaling which controls the hemocytes functions such as phagocytosis. Hyaline cells are the type of immune cells involve in phagocytosis along with the formation of radicals such as superoxide ions (Johansson and Soderhall, 1989; Rowley, 2016).

Overall, the Pacific white shrimp only experienced sub-lethal effects within the tested pH range (8.0-7.6). This species live in tropical marine habitats. Adults live and spawn in the open ocean, while post-larvae migrate inshore to spend their juvenile and sub-adult stages in coastal estuaries, lagoons, or mangrove areas. As a consequence, the species experience large fluctuations in pH characteristics of such habitats and the tested scenarios in our experiment are well within the natural range of variability. Marine organisms are naturally exposed to a wide range of environmental conditions including pH that constitutes their ecological niche. As a consequence of evolution, an organism is often adapted to its niche and the concept of stress is then relative. Stress can then be defined as a condition evoked in an organism by one or more environmental and biological factors that bring the organism near or over the limits of its ecological niche (Van Straalen, 2003). Previous work has demonstrated that the pH effect can lead to different physiological and evolutionary processes depending on its intensity. Within the present range of natural variability, it leads to sub-lethal physiological plasticity while stress, selection, and mortality occur under pH levels deviating from the present natural range (Dorey et al., 2013; Thor and Dupont, 2015; De Wit et al., 2016; Ventura et al., 2016; Vargas et al., 2017). As a consequence, the absence of negative effect on fitness in our experiment is not surprising and sub-lethal effects can be interpreted as an expression of plasticity.

The observed sub-lethal effects on immune parameters could be interpreted in the context of shifts in energy budget under low pH. Melzner et al. (2009) stated that many brachyuran crustaceans increase their metabolic demand under low pH for efficient ion- regulatory mechanisms by using powerful ion regulatory apparatus which are located in gill epithelia to maintain their acid-base balance. Han et al. (2018) showed that Pacific white shrimps achieve a new, balanced steady-state under low pH by maintaining higher osmoregulation rates. Activities of ion exchanging enzymes such as V-ATPase (Na⁺/H⁺) and HCO₃⁻-ATPase (Cl⁻/ HCO₃⁻) were enhanced under low pH conditions (pH 7.1 and 7.6) in Pacific white shrimps and those enzymes showed a negative correlation with pH (Pan et al., 2007). Further, they suggested that changing the H^+ and OH⁻ concentration in water can stimulate the activities of the V-ATPase enzyme in the gill epithelium membrane and increase the excretion of H⁺ to maintain the internal acid-base balance. Yet, this process does not directly enhance the energy demand as there are no ATPase linked to these enzymes, but this exchange is powered by the activities of Na⁺/K⁺ ATPase pumps which are the driving force as well as main energy consumer of the ion exchange process in marine ectothermic organisms including crustaceans (Morris, 2001; Freire et al., 2008; Melzner et al., 2009). Moreover, H^+ and HCO_3^- ions are accumulated in gill epithelial cells as carbonic anhydrase enzymes involved in the hydration of excretory CO₂ to form carbonic acids (H₂CO₃) which can further dissociate into H⁺ and HCO₃⁻ ions (Freire et al., 2008; Melzner et al., 2009). The aforementioned enhanced activities of the V-ATPase enzyme may drive into Cl- uptake, forcing enhanced functioning of HCO3⁻-ATPase, which can exchange HCO₃⁻ ions with Cl⁻ ions into the extracellular fluid (Pan et al., 2007; Melzner et al., 2009). As a consequence of the higher energetic cost for upregulated ion exchange, less energy may be invested into immune defense. However, how the observed changes in immune markers translate into health is difficult to assess. For example, Ellis et al. (2015) observed an apparent impairment in immune markers under low pH but hemolymph bactericidal activity was restored when mussels were exposed to the pathogenic bacteria Vibrio tubiaschii. This illustrates the possibility of reversible physiological trade-off rather than an irreversible impairment of immune function.

Our results are mostly in accordance with the published literature on the impact of low pH in shrimps. However, some differences in sensitivity between studies and species were observed. These can be partly attributed to methodological differences but also reflect differences in local adaptation. Species experiencing different pH variability in their natural environments have different sensitivities to low pH as a consequence of local adaptation (Vargas et al., 2017).

While the tested pHs in our experiment are within the present range of variability, the exposure to pH 7.6 highlights potential increased energy costs. In the context of ocean acidification, the Pacific white shrimp will be exposed more frequently to such pH values in the near future as well as to pH value deviating from the present range of variability.

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This will have negative sub-lethal but also lethal consequences for the species (Vargas et al., 2017) with potential impacts for natural populations as well as aquaculture. Identifying thresholds as well as solutions to minimize the impacts (e.g., adaptation of aquaculture practices, ecosystem-based management) should then be a priority.

DATA AVAILABILITY STATEMENT

The raw data supporting the conclusions of this article will be made available by the authors, without undue reservation.

AUTHOR CONTRIBUTIONS

C-CH and W-JH conceptualized and designed the experiment with contributions from SD and VW. Conducting the experiment, data collection, and sample analysis were done by VW with contributions from H-HH, NP, F-LY, J-SL, C-YL, and W-MC. VW and SD did the formal analyses. VW wrote the original draft of the manuscript with major reviewing and editing from SD, C-CH, and W-JH. All other authors read and edited the manuscript. All authors contributed to the article and approved the submitted version.

FUNDING

This work was supported by the Ministry of Education and the Ministry of Science and Technology (MOST), Taiwan, funded projects (MOST-108-2611-M-110-019-MY3, MOST-108-2621-M-110-001 and MOST-107-2911-I-110-301).

ACKNOWLEDGMENTS

We are glad to express our sincere thank to W.S. Weerakkody, S.Y. Huang, and M.H. Chuang for their support to maintain the experimental system. Further, we are very thankful to K.C. Huang, K.J. Kao, Cheau Chen, J.Y. Chen, and R.D.D.W. Kulathunga for their support during the sample analysis.

SUPPLEMENTARY MATERIAL

The Supplementary Material for this article can be found online at: https://www.frontiersin.org/articles/10.3389/fmars. 2021.748837/full#supplementary-material

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