



Insights on Alterations to the Rumen Ecosystem by Nitrate and Nitrocompounds

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Nitrate and certain short chain nitrocompounds and nitro-oxy compounds are being investigated as dietary supplements to reduce economic and environmental costs associated with ruminal methane emissions. Thermodynamically, nitrate is a preferred electron acceptor in the rumen that consumes electrons at the expense of methanogenesis during dissimilatory reduction to an intermediate, nitrite, which is primarily reduced to ammonia although small quantities of nitrous oxide may also be produced. Short chain nitrocompounds act as direct inhibitors of methanogenic bacteria although certain of these compounds may also consume electrons at the expense of methanogenesis and are effective inhibitors of important foodborne pathogens. Microbial and nutritional consequences of incorporating nitrate into ruminant diets typically results in increased acetate production. Unlike most other methane-inhibiting supplements, nitrate decreases or has no effect on propionate production. The type of nitrate salt added influences rates of nitrate reduction, rates of nitrite accumulation and efficacy of methane reduction, with sodium and potassium salts being more potent than calcium nitrate salts. Digestive consequences of adding nitrocompounds to ruminant diets are more variable and may in some cases increase propionate production. Concerns about the toxicity of nitrate's intermediate product, nitrite, to ruminants necessitate management, as animal poisoning may occur via methemoglobinemia. Certain of the naturally occurring nitrocompounds, such as 3-nitro-1-propionate or 3-nitro-1-propanol also cause poisoning but via inhibition of succinate dehydrogenase. Typical risk management procedures to avoid nitrite toxicity involve gradually adapting the animals to higher concentrations of nitrate and nitrite, which could possibly be used with the nitrocompounds as well. A number of organisms responsible for nitrate metabolism in the rumen have been characterized. To date a single rumen bacterium is identified as contributing appreciably to nitrocompound metabolism. Appropriate doses of the nitrocompounds and nitrate, singly or in combination with probiotic bacteria selected for nitrite and nitrocompound detoxification activity promise to alleviate risks of toxicity. Further studies are needed to more clearly define benefits and risk of these technologies to make them saleable for livestock producers.

Keywords: nitrate, nitrocompounds, rumen, methane reduction, nitrate toxicity

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INTRODUCTION

Nitrate and other oxidized nitrocompounds are scientifically pursued due to their toxicity and ability to reduce enteric methane emissions by ruminants. Methane production by microbes within the rumen is recognized as a fermentative inefficiency resulting in the loss of 2-12% of the gross energy consumed by the host (Johnson and Johnson, 1995). Environmentally, methaneis a significant greenhouse gas and strategies are sought to reduce its emission from livestock, which in the United States accounts for 95% of anthropogenic methane emissions arising from enteric fermentation (US EPA, 2012). However, despite its negative association with energy retention and greenhouse gas emissions, methanogenesis plays an important ecological role in the rumen. Archaea consume hydrogen emitted by bacterial and protozoal hydrogenases functioning to reoxidize reduced nucleotides produced during glycolysis and other catabolic pathways (Ellis et al., 1990; Miller, 1995). Methanogenesis functions to maintain a low partial pressure of hydrogen, which at partial pressures above 1 kPa promote end-product inhibition of NADH oxidoreductase thereby disrupting the oxidation of NADH and depleting concentrations of NAD to levels that inhibit fermentation (Miller, 1995; Van Nevel and Demeyer, 1996; Hegarty and Gerdes, 1999).

It is generally recognized that the most effective methaneinhibiting interventions provide alternative mechanisms for maintaining low partial pressures of dihydrogen within the rumen. A variety of electron accepting substrates are available for use as alternative electron acceptors for anaerobic respiration in the rumen, including unsaturated fatty acids, nitrate, sulfate or fumarate (Leng, 2014). From a thermodynamic perspective, however, the use of nitrate is particularly attractive because the dissimilatory reduction to ammonia is energetically more favorable ($\Delta G^{0'} = -600$ kJ/mol) than the reduction of carbon dioxide to methane ($\Delta G^{0'} = -136$ kJ/mol) and the reduction of other electron acceptors available in an anaerobic environment (Thauer et al., 1977; Table 1). Moreover, supplemental nitrate could under certain dietary conditions serve and perhaps even replace nonprotein nitrogen sources such as urea to support microbial protein synthesis in the rumen (Carver and Pfander, 1974; Sophea and Preston, 2011; Li et al., 2012; Silivong et al., 2012; Thanh et al., 2012). Theoretically, the consumption of four electrons with the reduction of nitrate to nitrous oxide, and potential consumption of an additional electron for the reduction of nitrous oxide to dinitrogen (N₂) via denitrification, could also serve as a metabolic route for electron disposal to the reduction of nitrate. However, from an energetic perspective, the reduction of nitrite to ammonia is slightly more favorable thermodynamically than the reduction of nitrite to nitrous oxide (436 kJ/mol vs. 453 kJ/mol), and thus ammonia is the prevalent reduction product in the rumen (Table 1).

Early work by Anderson et al. (1993, 1997) revealed that like nitrate and nitrite, the naturally-occurring nitrocompounds 3-nitro-1-propionate and 3-nitro-1-propanol and industriallyproduced nitroethane may also serve as electron acceptors within rumen microbial populations. However, in addition to serving as alternative electron acceptors these nitrocompounds TABLE 1 | Standard molar Gibbs Free energy for reductive processes.

Reaction ^a	Electrons	<i>E</i> ° (mV)	–∆G° (kJ/mol)
Carbon dioxide reduction to methane	-8	-244	131
Fumarate to succinate	-2	33	86
Oxygen to water	-2	818	228
Nitrate reduction to ammonia	-8	-	599.6
Nitrate reduction to nitrite	-2	433	163.2
Nitrite reduction to ammonia	-6	363	436.4
Nitrate reduction to nitrogen gas	-5	-	1120
Nitrate reduction to nitric oxide	-3	350	147
Nitric oxide reduction to nitrous oxide	-1	1175	306.1
Nitrous oxide reduction to nitrogen gas	-1	1355	341.4

^aAdapted from Thauer et al. (1977).

also exert a direct inhibition of ruminal methanogenesis (Anderson and Rasmussen, 1998; Gutierrez-Bañuelos et al., 2008). Consumption of electrons, at least with the reduction of naturally occurring nitrocompounds and nitroethane, occurs more slowly than the direct inhibition mechanism and requires in situ enrichment of competent nitro-reducing bacteria that are normally present at low numbers. The biological processes involved in the direct chemical inhibition of methane production by the short chain nitrocompounds are ill-defined. It has been speculated that this could occur via inhibition of electron transfer reactions like the nitroethanol-caused inhibition of electron transfer between ferredoxin and hydrogenase (Angermaier and Simon, 1983; Anderson et al., 2008). A number of other short chain nitrocompounds have been tested in vitro and while most if not all have been found to effectively inhibit ruminal methane production at present only a few have been found to be suitable electron acceptors for supporting growth of nitro-reducing bacteria. Alternatively, inhibition of methyl-coenzyme M reductase of methanogenic bacteria has been postulated for the recently identified inhibitor, 3nitrooxypropanol, as well as some other nitro-oxy-compounds (Martínez-Fernández et al., 2014; Prakash, 2014). Thus, these nitro-oxy compounds, which possess an oxygen atom binding the nitro-group at the number 3 carbon, not only differ structurally from the short chain nitrocompounds discussed above, but probably in their mode of action as well.

Multiple literature reviews on the toxicological aspects and methane reducing potential of feeding nitrate to ruminants have been published recently including excellent works by Lee and Beauchemin (2014) and Leng (2008). Consequently, the present work focuses our discussions on the microbiological response to nitrate and nitrocompound supplementation.

NITRATE AND NITRITE METABOLISM WITHIN THE RUMEN

Microbial reduction of nitrate can occur by dissimilatory and assimilatory processes. The genes involved, their regulation and the energetics of these pathways substantialy differ (**Table 2**). The assimilatory nitrate reduction pathway consumes energy to

Oxidation state		+5	+3	+2	+1	-3
Dissimilatory nitrate reduction	NO ₃		NO ₂			NH ₄
		nar/nap			nir/nrf	
Assimilatory nitrate reduction	NO ₃		NO ₂			NH4
		nar/nas			nir	
Denitrification	NO_3		NO ₂	NO	N ₂ O	
		nar/nap		Nir	nor	

TABLE 2 | Microbial nitrogen metabolism in the rumen^a.

^aNO₃, nitrate; NO₂, nitrite; NO, nitric oxide; N₂O, nitrous oxide; NH₄, ammonium.

reduce nitrate to ammonia as a nitrogen source for microbial protein synthesis and is repressed by ammonia (Moreno-Vivián et al., 1999). Consequently, the functional role of this process is largely unnecessary in environments like the rumen where the availability of ammonia may down regulate this activity. Dissimilatory nitrate reduction, on the other hand, is an energy generating process that is distributed widely amoung obligate and facultative anaerobic bacteria (Thauer et al., 1977). Within the rumen, dissimilatory nitrate reduction occurs primarily via a two-step pathway where nitrate is first reduced to nitrite, which can accumulate as an intermedite before it is ultimately reduced to ammonia. Enzymes involved in dissimilatory nitrate reduction include membrane bound and periplasmc nitrate reductases encoded by nar and nap genes and nitrite reductases encoded by nir and nrf genes (Thauer et al., 1977; Moreno-Vivián et al., 1999; Table 2). Dissimilatory nitritereduction occurs on the outer cytoplasmic membrane and depending on the organism, consumes electrons via the oxidation of reduced electron carriers such as NADH or FADH and upon subsequent transfer of these electrons to the respiratory electron transport system they can be used to reduce and thus detoxify nitrite (Thauer et al., 1977; Moreno-Vivián et al., 1999). Certain bacteria may lack a complete functional electron transport chain yet be able to reduce nitrate to nitrite and sometimes to ammonia via reductive reactions with the incomplete chain or other electron carriers without generating ATP. In such cases, these bacteria are thought to still gain an energetic benefit via more effective disposal of electrons and therefore more efficient recycling of NAD via oxidation of NADH produced during glycolysis (Hasan and Hall, 1975).

Denitrification is another pathway for dissimilatory nitrate reduction, yet despite evidence for the presence of the denitrifying genes (*nir, nor,* and *nos*) within the rumen eubacterial and archaeal metagenome (Zumft and Kroneck, 2007; Brulc et al., 2009), this process is not considered to contribute appreciably to ruminal nitrate reduction (Jones, 1972; Kaspar and Tiedje, 1981; Leng, 2008; **Table 2**). In a metagenomic study by Brulc et al. (2009), 85 occurances of genes associated with denifrication and nitrogen fixation were tagged in bovine rumen metagenomics samples compared to the occurance of 636 genes contributing to nitrate and nitrite ammonification and 1233 total genes contributing to nitrogen metabolism. Thus, the authors concluded that denitrification and nitrogen fixation activities

were likely inconsequential in the rumen. However, genetically this accounts for 7% of the genes involved in nitrogen metabolism and while clearly not dominant, their contribution cannot be completely ruled out.

Few studies have measured ruminal nitrous oxide accumulation in response to nitrate supplementation, however, in studies that have the amounts produced were found to vary considerably. Nitrous oxide concentrations equivalent to 0.3% the amount of added nitrate or nitrite were measured within nitrate- or nitrite-supplemented in vitro rumen fluid incubations and in vivo from the rumen of nitrate-supplemented [as 5 Ca(NO₃);NH₄NO;10 H₂O] sheep (Kaspar and Tiedje, 1981; de Raphélis-Soissan et al., 2014). However, Petersen et al. (2015) found nitrous oxide emissions account for as much as 3.4% of added nitrate (21 g NO3kg⁻¹ dry matter fed to adapted dairy cows, nitrate type unknown). The later evidence suggests that denitrification may be contributing to nitrous oxide production in the rumen rather than being produced simply as a nonspecific byproduct of dissimilatory nitrite reduction by the nir nitrite reductase as postulated by Kaspar and Tiedje (1981). Potential bacterial denitrifiers are Pseudomonas aeruginosa and certain species of Propionibacterium and Nitrosomonas, which in the absence of added nitrate may be considered transient or minor colonizers of the rumen (Bryant, 1959; Duncan et al., 1999; Mitsumori et al., 2002; Arai et al., 2003). Future metatranscriptomic analysis and/or RT-qPCR combined with nitrous oxide emission measurements could determine the relative abundance and gene expression of the denitrification pathways within the rumen of animals adapted to different types and amounts of nitrate.

MICROBIAL RESPONSE TO NITRATE

Microbial nitrate and nitrite metabolism in the rumen is a paradox in that it enables detoxification, but also results in the formation of toxicants (**Table 3**). Since these microbes are symbionts to the ruminant host, an understanding of both their response in terms of gene expression and community structure and their susceptibility to added nitrate and nitrite becomes exceedingly important. Exposure of unadapted microbial populations to high intakes of dietary nitrate results in the rapid induction of nitrate reducing activity, as evidenced by

TABLE 3 Summary of nitrate and nitrocompound toxicity.	TABLE 3
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Substrate	Source	Intoxication	Organisms involved	Transformation/ metabolism
Nitrate	Feeds	See nitrite	Many groups	Reduced to nitrite
Nitrite	Produced from nitrates in feeds	Methemoglobinemia	Many groups	Further broken down into ammonium
3-Nitro-1-propanol	Astragulus and many other plant species: hydrolysis of nitroglycosides in feed	Inhibits succinate dehydrogenase	Denitrobacterium detoxificans, Coprococcus sp., Megasphaera elsdenii, Selenomonas ruminantium	Metabolized to aminopropanol in the rumer and 3-nitro-1-propionate in the liver
3-Nitro-1- propionate	Astragulus and many other plant species: hydrolysis of nitroglycosides in feed	Inhibits succinate dehydrogenase	Denitrobacterium detoxificans, Coprococcus sp., Megasphaera elsdenii, Selenomonas ruminantium	Metabolized to β-alanine in the rumen which is futher metabolized
Nitroethane/ nitroethanol	Synthetic	Unknown (possible respiratory toxicosis)	Denitrobacterium detoxificans	Metabolized to ethylamine/possibly to ethanolamine

>14-fold increases in activity within 4 h of first exposure, as well as the gradual selection of highly competent nitrate and nitrite reducing rumen bacteria (Allison and Reddy, 1984). Induction of nitrate and nitrite metabolism by prominent fermentative bacteria in the rumen such as Selenomonas ruminantium as well as members belonging to Butyrivibrio, Clostridium, Peptostreptococcus, and Propionibacterium can readily and rapidly contribute enhanced nitrate reduction capabilities (Alaboudi, 1984; Iwamoto et al., 2002). Subsequent to induction of nitrate and nitrite reducing activity, these bacteria can be enriched in number via exposure to nitrate because of the greater energy yield from electron transport mediated phosphorylation of ADP to ATP or via achieval of more effective electron disposal (Hasan and Hall, 1975; Thauer et al., 1977; Moreno-Vivián et al., 1999). The contribution of propionibacteria to ruminal nitrate reduction is probably atypical, however, as most nitrate-reducing propionibacteria are denitrifiers that produce nitrous oxide as an end product (Kaspar, 1982). Wolinella succinogenes may be considered a specialist in that it is nonfermentative and exhibits highly active nitrate and nitrite reducing activity. Nitrate-utilizing Veillenolla parvula also contribute to ruminal nitrate and nitrite metabolism however the abundance of these populations appears to be dependent on nitrate concentrations (Asanuma et al., 2002; Iwamoto et al., 2002). A number of other nitrate-reducing bacteria, such as certain species of Desulfovibrio and members of family Enterobacteriaceae can inhabit the rumen, albeit at low abundance, and it is reasonable to suspect these bacteria could also be enriched during prolonged exposure to nitrate (Pfennig et al., 1981; Stewart, 1988). Certain members of Enterobacteriaceae, such as entertoxigenic and enterohemorrhagic Escherichia coli and numerous Salmonella enterica serovars, are important animal or foodborne pathogens and thus their enrichment due to nitrate feeding would be undesirable, although to our knowledge this has not yet been reported.

Evidence for enrichment of nitrate-reducing bacterial populations also comes from studies of Alaboudi and Jones (1985), who reported >3-fold increases in rates of ruminal

nitrate and nitrite metabolism in goats adapted to approximately 1.5 g nitrate (as KNO₃) kg⁻¹ body weight which coincided with a 3-fold increase in the proportion of nitrate-reducing bacteria. Unfortunately, they did not further characterize their isolated bacteria and the identity of nitrate-reducers was not reported (Alaboudi and Jones, 1985). More recently, Asanuma et al. (2015) reported 2.3–2.8-fold increases in nitrate and nitrite reducing activity following adaptation to a diet supplying approximately 0.18 g nitrate (as KNO₃) kg⁻¹ body weight per day to goats. Concomitant with this increase in nitrate and nitrite reducing activity were nearly equivalent increases in the relative abundance of *narG* and *nrfA* gene sequences specific for the nitrate reducing bacterium, *Selenomonas ruminantium*, as well this bacterium's 16S rRNA gene (Asanuma et al., 2015).

Conversely, Lin et al. (2013a) found no differences in abundance of nar or of 16S rRNA genes specific to S. ruminantium or to the less abundant nitrate-reducers V. parvula and W. succinogenes in rumen contents from nitrate nitrogen-fed steers (0.22–0.31 g nitrate kg⁻¹ body weight, fed as KNO₃) when compared to urea nitrogen-fed steers. Unfortunately, rates of nitrate and nitrite metabolism were not reported so comparison of abundance measurents to activity measurements is not possible. They did, however, observe increases in populations of the nitrate-reducers Campylobacter fetus, which was enriched in both liquid and solid fractions of ruminal contents, and Mannheimia succiniciproducens, which was enriched only in the liquid fraction collected from the rumen (Lin et al., 2013a). More recently, Zhao et al. (2015) proposed that Campylobacter and cyanobacteria were important nitrate-reducing taxa based on results from 16s rDNA sequencing. Campylobacter fetus is an important pathogen affecting ruminants and thus its enrichment would be undesirable. Conversely, M. succiniciproducens may be an attractive bacterium to enrich in the rumen because it has considerable potential fix carbon dioxide, via carboxylation of phospoenolpyruvate (Lee et al., 2006). Even after enrichment, however, C. fetus and M. succiniciproducens made up only a small proportion (<0.1%) of the total population (Lin et al., 2013a). Moreover, the study of Lin et al. (2013a) was a cross-over design,

however, with a 12 day wash-out period between nitrate-nitrogen and urea-nitrogen treatments which may have been insufficient to allow populations to re-achieve unperturbed densities. It was not stated if the animals had opportunity to physically contact one another during provision of the respective treatments, but if they had then induction or transfer of nitrate and nitrite reductive capacity could have occurred between groups of animals as has been as reported by Majak and Cheng (1984) and Cheng et al. (1985). In these earlier experiments, animals in treated groups received daily doses of nitrate intraruminally (0.1 g nitrate kg^{-1} body weight, salt unspecified) during the treatment period and were initially kept away from animals in the untreated control group (Majak and Cheng, 1984; Cheng et al., 1985). Then both groups were housed in adjacent pens while the treated group continued to receive the nitrate supplement. Despite receiving no nitrate, the control group of steers showed increased (80-200%) nitrate and nitrite reduction rates during the contact period. The authors hypothesized this transfer may have been mediated via horizontal gene transfer of plasmids containing nitrate reductase (Majak and Cheng, 1984; Cheng et al., 1985). However, this could also originate from other mobile genetic elements such as transposons or bacteriophages, oral transfer of microbes via animal licking, or some other signaling molecule that we are unaware of that would induce the upregulation of nitrate and nitrite reducing genes in the ruminal microbial population.

Exposure of ruminal populations of bacteria to nitrate not only selects for nitrate-reducing bacteria, but also acts as a selection mechanism for a different ecological makeup within the microbial community. Marais et al. (1988), using in vitro cultivation techniques, reported that nitrite accumulation resulting from reduction of added nitrate (approximately 26 g nitrate kg⁻¹ dry matter; added as KNO₃) decreased ruminal cellulolytic activity. They concluded that the decrease in cellulolytic activity was a result of 64, 25, and 57% decreases in numbers of cellulolytic, xylanolytic and total viable bacteria, respectively, as determined via viable cell count on selective media. Others also have observed toxic effects of nitrite, produced as an intermediate during the reduction of nitrate, on populations of cellulolytic bacteria as well as on other microbial populations including methanogens (Iwamoto et al., 2002; Zhou et al., 2011, 2012; Asanuma et al., 2015). The inhibition of cellulolytic organisms may explain the decreases in dry matter intake sometimes observed in animals feed nitrate-supplemented diets (Newbold et al., 2014; Lee et al., 2015a,b). It is recognized that decreased cellulolysis can decrease rates and extent of neutral detergent fiber digestion thus increasing rumen retention time of a forage and negatively affective rumen fill, both which can cause decreased dry matter intake (Allen, 2000). It is well known that nitrate and nitrite additions cause a shift in volatile fatty acid concentration, sometimes disproportionately against branched-chain volatile fatty acids essential for certain bacterial populations, and this has been suggested as a reason for decreases observed in total bacterial populations and particularly cellulolytic bacteria (Allison and Reddy, 1984) but this was discounted by Marais et al. (1988).

In the case of methanogens, the reduction of nitrate preferentially consumes electrons at the expense of

methanogenesis. However, methanogens also appear to be particularly sensitive to the toxic effects of nitrite, with 50% inhibition in cell growth occurring with as little as 0.5 mM nitrite (Iwamoto et al., 2002). Asanuma et al. (2015) similarly found methanogens, as well as total populations of rumen bacteria, protozoa and fungi to be greatly decreased *in vivo* after goats were fed a high nitrate diet (5.4 g nitrateday⁻¹, as KNO₃) for 2 weeks. This total community depression may result from the oxidizing nature of nitrite as it relates to its antimicrobial properties, attributed to inactivation or inhibition of sulfur containing constituents involved in energy metabolism, DNA replication or maintenance of cell wall integrity (Marais et al., 1988; Cammack et al., 1999).

The high reactivity of nitrite could also disrupt the low E°/mV within the rumen as evidenced in the study of Jamieson (1959), who found that sheep intraruminally dosed with 25 g nitrate (as KNO₃) had a higher $E_{\rm h}$ value pre-dose than 2 h post dose (-225 vs. -70 mV, respectively). As reported by Kalachniuk et al. (1978) and discussed by Zhou et al. (2012), an increase in reduction potential (E_h) has been reported to be inhibitory to some rumen bacteria, notably S. ruminantium, Bacteroides amylophilus, Fibrobacter (Bacteroides) succinogenes and Streptococcus bovis. More recently, however, an increased $E_{\rm h}$ was found to be not particularly inhibitory per se to these same bacterial species (Marounek and Wallace, 1984), but it could potentially perturb thermodynamic control of important oxidation/reduction reactions such as those involved in electron transfer. For instance, Marais et al. (1988) reported that nitratecaused inhibition of forage digestibility could not be overcome by using the reducing agent cysteine to decrease the nitritecaused increase in culture $E_{\rm h}$. Thus, they proposed a direct effect of nitrite on inhibiting bacterial growth, and while they at the time suggested this inhibition appeared to be most potent against bacteria with electron transport linked phosphorylation capabilities, more recent evidence suggest the toxic effect may not be so specific. In the study of Asanuma et al. (2015) for instance, they found that populations of major the cellulolytic bacteria F. succinogenes, Ruminococcus flavefaciens, which contain electron transport capabilities, as well Ruminococus albus, which does not contain electron transport capabilities, all decreased in the rumen of goats fed nitrate. Zhou et al. (2012) also reported reductions in abundance of F. succinogenes, R. flavefaciens, and *R. albus* as well as in archaeal, but not total bacterial populations when measured by real-time PCR during in vitro incubations of mixed populations of ruminal microbes with =12 mM sodium nitrate. They further found evidence of adaption or acquisition of tolerance by populations of R. albus as well as F. succinogenes, but not R. flavefaciens or the archaeal populations following up to 6 consecutive cultures with 12 mM added sodium nitrate (Zhou et al., 2012). Broad activity of nitrite was found in the studies of Iwamoto et al. (2002), who tested 15 different bacterial species and found that growth of all except four were inhibited by concentrations of 3-5 mM nitrite, with many being producers of hydrogen, formate or lactate that can contribute reductants for nitrate and nitrite reduction or methanogenesis (Russell and Rychlik, 2001). Decreases in populations of hydrogen, formate or lactate-producing microbes could potentially limit the availability of reductant for methanogenesis or nitrate respiration. However, Asanuma et al. (2015) observed significant increases in genetic abundance of the highly efficient sugar-fermenting bacterium, *S. bovis*. It is reasonable to speculate that the sugar-fermentation by *S. bovis* may contribute a pool of electron donating substrates for the nitrate-reducing population thus potentally compensating for the decreased reductant that would be expected by inhibition of the hydrogen, formate or lactate-producing microbes observed by Iwamoto et al. (2002).

In direct opposition to the findings discussed above, Zhao et al. (2015) found that added nitrate *in vivo* (1–2% dry matter, nitrate type unknown) was associated with an increase in many cellulolytic bacterial species including *R. flavefaciens*, *R. albus*, and *F. succinogenes*. The authors attributed this selection to nitrate-caused increase production and thus availability of branch chain fatty acids required by these bacteria. The authors did not specifically discuss why branched chain fatty acids were increased but it is reasonable to speculate that this may have occurred due to changes in microbial diversity in the rumen population.

The protozoa, which reduce nitrate to ammonia for assimilatory purposes, play an unclear role in the total nitrate metabolism in the rumen. The protozoal fractions showed similar rates of nitrate reduction with less nitrite accumulation without any adaptation period as compared to the whole rumen fraction or bacterial fraction (Yoshida et al., 1982; Iwamoto et al., 2001; Lin et al., 2011). In contrast, Allison and Reddy (1984) reported that the nitrate and nitrite reducing activity was not found in their protozoal fraction. Moreover, they also reported that the nitrate reducing activity in rumen contents was membrane bound and was inhibited by azide and hydroxyl guinolone-Noxide, each inhibitors of electron carrier mediated respiration. The absence of nitrate-reducing activity when dissimilatory (respiratory) nitrate reduction was inhibited suggests that the assimilatory nitrate-reducing pathway, such as that by protozoa, was inoperative or contributed little to overall nitrate reduction. Future studies are needed to detect changes in the protozoal populations in the rumen from nitrate-fed ruminants as well as additional research exploring their potential, if any, to enhance nitrite removal and reduce methane production.

OTHER EFFECTS OF FEEDING NITRATE

Additions of nitrate to *in vitro* incubations of mixed populations of ruminal microbes have generally resulted in decreased production of methane and propionate while sometimes increasing production of total volatile fatty acids; however, this varies considerably depending on experimental conditions (Bozic et al., 2009; Shi et al., 2012; Zhou et al., 2012; Patra and Yu, 2014). The decreased production of more reduced fatty acids reflects a lesser need for this route of electron disposal because electrons are instead consumed for the reduction of nitrate to nitrite and ultimately to ammonia (Sutherland, 1977; Russell, 2002). Increased acetate production at the expense of butyrate can sometimes occur due to nitrate supplementation (Farra and Satter, 1971; Anderson and Rasmussen, 1998) and this could be due to the stimulation of more thermodynamically favorable pathways for electron disposal (Ungerfeld and Kohn, 2006).

The effects of nitrate feeding ultimately depend on nitrate concentration and its availability within the rumen which is not only affected by the amount fed but also by the type or source of nitrate fed which can markedly affect the ruminal availability of the free nitrate ion. Consequently, in vivo additions of nitrate have also resulted in variable effects on volatile fatty acid production and methane emissions as well as on methemoglobin development in studies measuring these responses (Table 4). For instance, encapsulated nitrate supplements used in the studies of Lee et al. (2015a,b) would be expected to release nitrate more slowly than the salt forms of nitrate, which themselves can differ in nitrate availability thus affecting their methane reducing potential and their toxicity. In studies using calcium nitrate, amounts consumed ranged from 0.26 to 0.94 g $Ca(NO_3)_2kg^{-1}$ body weight, an amount sufficient to cause toxicity if an equivalent dose of sodium or potassium nitrate were fed (van Zijderveld et al., 2010, 2011; Hulshof et al., 2012; Li et al., 2012; de Raphélis-Soissan et al., 2014; El-Zaiat et al., 2014; Velazco et al., 2014). Likewise, calcium nitratewas the least effective at decreasing methane, 3.5% reduction when expressed as mmol⁻¹ nitratekg⁻¹ body weight, as compared to sodium and potassium nitrate which caused a 14.5 and 9.6% decrease in methane per mmol nitratekg⁻¹ body weight respectively (Takahashi and Young, 1991; Sar et al., 2004, 2006; Nolan et al., 2010; van Zijderveld et al., 2010, 2011; Hulshof et al., 2012; Leng et al., 2012; Li et al., 2012; de Raphélis-Soissan et al., 2014; El-Zaiat et al., 2014; Velazco et al., 2014; Asanuma et al., 2015). Similarly, Phuong (2012) found that 24% less NaNO3 and 12% less KNO3 as compared to Ca(NO₃)₂ resulted in the same amount of methane production. This is most likely due to differences in solubility. Calcium has low solubility at pH above 5.5 and since the rumen pH seldom declines to below 5.5 it is likely that a significant portion of the calcium nitrate is unsolubilized and thus unavailable to exert the full effect of the nitrate. Additionally, the cation themselves (K, Na, and Ca) have important roles in the rumen including motility, osmolality, pH, acid base balance, and are known to influence the dry matter intake. As many of these nitrate salt feeding trials are feeding at level in excess of 2% of intake, there is the possibility that the accompanying cations may alter the microbiome and the animal physiology. Iso-cation diets controls may be necessary in future studies. Overall, it would be prudent to develop standards for nitrate salt feeding with stipulations for salt type, age of the animal, adaptation status, and previous exposure.

In global warming potential (GWP), carbon dioxide is 1, methane is 21, and nitrous oxide is 298 over a 100-year time scale. Therefore, even a small increase in nitrous oxide from nitrate reduction could have large effects on GWP. For example, nitrous oxide emission from sheep was higher when fed nitrate (calcinite at 0.625 g kg⁻¹ body weight), which in turn lowered the net benefit of methane mitigation on global warming potential by 18% despite the nitrous oxide being only 0.3% of added nitrate (de Raphélis-Soissan et al., 2014). Likewise, Petersen et al. (2015) found a 28–23% reduction in overall decrease in global warming mitigation effect and Neumeier et al. (2014) found no

Alaboudi and Jones, 1985 Asanuma et al., 2015 de Raphélis-Soissan et al., 2014 El-Zaiat et al., 2014 En-Zaiat et al., 2014 Farra and Satter, 1971 KN03:NaN03										:	ې ب
								MetHb (%)	effect (%)	matter intake	supplemented
		Total acids	Acetate	Propionate	Butyrate						
		No effect	Increased	Decreased	Decreased	Top dressed	Sheep	2.0	NMc	0.05	0.02480
		Decreased	Decreased	Decreased	Decreased	Top dressed	Goats	MN	-89	MN	0.00310
	2	No effect	Increased	Decreased	MM	Top dressed	Sheep	45.0	19	2.0	0.00767
	Ca(NO ₃) ₂ :NH ₃ NO ₃	Increased	Increased	No effect	Increased	Encapsulated	Lambs	1.0	-33	4.5	0.02110
	aNO ₃	No effect	Increased	Decreased	Decreased	Top dressed	Sheep	2.0	MN	6.0	0.00510
Hulshof et al., 2012 Ca(NO ₃) ₂	2	No effect	No effect	No effect	No effect	Top dressed	Beef	MN	-32	2.5	0.00600
Lee et al., 2015a,b Ca(NO ₃) ₂	2	No effect	No effect	No effect	No effect	Encapsulated	Beef	<2.0	18	2.5	0.00844
Leng et al., 2012 KNO ₃		MN	MM	MN	MM	Top dressed	Beef	<1.0	-29	6.0	0.01020
Li et al., 2012 Ca(NO ₃) ₂	2	No effect	Increased	No effect	Decreased	Top dressed	Sheep	<1.0	-35	3.0	0.01150
Lund et al., 2014 Ca(NO ₃)2	2	No effect	No effect	No effect	No effect	Top dressed	Dairy ^d	MN	-31	0.7	0.00120
Newbold et al., 2014 Ca(NO ₃)2	2	MN	MM	MN	MM	Top dressed	Beef	>20	-30	3.0	0.00266
Nolan et al., 2010 KNO3		Decreased	Increased	Decreased	Decreased	Top dressed	Sheep	<1.0	-23	4.0	0.00680
Pal et al., 2014 KNO3		No effect	No effect	Decreased	No effect	Top dressed	Sheep ^d	MN	MN	2.0	0.00600
Sar et al., 2004 NaNO3		Increased	Increased	Decreased	Decreased	Via cannula	Sheep	18.4	-50	n/a ^c	0.00580
Sar et al., 2005		No effect	Increased	No effect	Decreased	Via cannula	Sheep	2.0	-50	n/a	0.00580
Takahashi and Young, 1991 NaNO ₃		MN	MM	MN	MM	Via cannula	Sheep ^d	20.0	-86	n/a	0.00188
van Zijderveld et al., 2010 Ca(NO3)2	2	No effect	No effect	No effect	No effect	Top dressed	Lambs	7.0	-32	2.6	0.00805
van Zijderveld et al., 2011 Ca(NO3)3	e	MN	MM	MN	MM	Top dressed	Dairy	4.7	-16	1.0	0.00366
Velazco et al., 2014 Ca(NO ₃) ₂	2	MN	MM	MN	MM	Top dressed	Beef	3.0	-17	2.6	0.00960

^a KNU3, potassium nitrate; varivu3), varvum i muuvo, vuvi ^bmo/kg body weight. ^cNM, not measured; n/a, not applicable. ^dAnimal were not previously adapted to nitrate additions.

TABLE 4 | Effects of nitrate supplementation to ruminants on volatile fatty acid production, metHb, and methane reduction in vivo.

net reduction in greenhouse gases when considering the increase in nitrous oxide from feeding nitrate.

Considering the cost and risk to the producers of feeding nitrate, the decrease of greenhouse gases production may not justify the usage of nitrate feeding as a methane mitigation strategy. Toxicity aside, it has been suggested that the limiting factors for the adoption of supplemental-nitrate feeding in beef production are financial in nature (Callaghan et al., 2014).

NITROCOMPOUND METABOLISM WITHIN THE RUMEN

Unlike nitrate and nitrite metabolism, less is known about the mechanisms of nitrocompound metabolism by ruminal microbes. Nishino et al. (2010) isolated 3-nitro-1-propionatedegrading species of strictly aerobic Cupriavidus and Pseudomonas bacteria from soil and water. Considering, however, that their degradation of 3-nitro-1-propionate proceeded to yield to propionate-3-nitronate and ultimately malonic semialdehyde, nitrate, nitrite, and hydrogen peroxide, it is unlikely these aerobic bacteria contribute to nitrocompound degradation in the anaerobic rumen. In the rumen, the naturally occurring 3-nitro-1-propionic acid and 3-nitro-1-propanol are first hydrolyzed from their glucose conjugates by microbial esterase or β-glycosidase activity thereby liberating the aglycones from their respective esters or ether glycosides (Table 3), the latter which are rather innocuous (Anderson et al., 2005). Once liberated, 3-nitro-1-propionic acid and 3-nitro-1-propanol are available to be absorbed or further metabolized by bacteria in the rumen. The nitroacid is absorbed less rapidly but metabolized more rapidly than the nitroalcohol which thus explains why 3-nitro-1-propionic acid is less toxic than 3-nitro-1-propanol to ruminants.

Presently, the ruminal bacterium Denitrobacterium detoxificans is the only known anaerobe to exhibit appreciable nitroalkane-reducing activity (Anderson et al., 1996, 1997, 2000). This bacterium conserves energy for growth exclusively via anaerobic respiration, oxidizing hydrogen, formate or lactate in the reduction of nitrate to ammonia and the reduction of 3-nitro-1-propionic acid, 3-nitro-1-propanol, and nitroethane to β-alanine, 3-amino-1-propanol and aminoethane, respectively (Anderson et al., 1993, 1997, 2000). β-alanine was rapidly metabolized to unknown products in mixed populations of ruminal microbes but 3-amino-1-propanol appeared to be a terminal product (Anderson et al., 1993). Early research by Looper et al. (1959) indicated that β -alanine was appreciably deaminated in the rumen but little other information is available on its fate as an endproduct. The reduction of the nitroalkanes to their amines is presumed to consume six moles electrons per mol of amino group reduced and is based on the stoichiometry reported for the reduction of nitroethanol to ethanolamine (Angermaier and Simon, 1983). Denitrobacterium detoxificans can also grow on dimethyl sulfoxide and trimethyl amine oxide, reducing these acceptors to dimethylsulfide and trimethyl amine, and can grow, albeit less readily, on nitroethanol, 2-nitro-4butanol, 1-nitropropane and 2-nitro-1-propanol (Anderson et al., 2000). The fate of these later four substrates has not been investigated. With respect to these nitrocompounds, all except 2-nitro-4-butanol and 1-nitropropane have been tested and found to be potent inhibitors of ruminal methanogenesis (Anderson and Rasmussen, 1998; Anderson et al., 2003, 2008, 2010). Evidence indicates that *D. detoxificans* can effectively contribute to detoxification of 3-nitro-1-propionic acid, 3-nitro-1-propanol, which exist as phytotoxins in certain leguminous plants, notably milkvetchs belonging to the genera *Astragalus* and *Coronilla* (Anderson et al., 2005). However, with little known about the physiology and nutritional requirements more research is needed to make growing this bacterium practical for large scale applications.

More recently, chemically synthesized dimethyl-2nitroglutarate and 2-nitro-methyl-propionate have also been tested and found to be similarly potent anti-methanogenic compounds (Anderson et al., 2010) as has ethyl nitroacetate (data not shown); however, the fate of these three nitrocompounds within incubations of mixed populations of rumen bacteria has not been determined.

From a thermodynamic perspective, Gibbs free energy values for the reduction of the nitrocompounds to their respective amines have not been determined. However, experimentally, nitrate was preferentially reduced by mixed populations of bovine ruminal microbes compared to 3-nitro-1-propionate or 3nitro-1-propanol when the compounds were incubated together (Anderson et al., 1997; Zhang et al., 2014). This suggests that the reduction of nitrate may be energetically more favorable than the reduction of the nitrocompounds although the contribution of other mechanisms, such as the presence of more active nitratereducing enzymes cannot be excluded.

The two different mechanisms of action of these nitroalkanes, primarily direct inhibition of methanogenesis and secondarily acting as an alternative electron acceptor, are not necessarily incompatible as the process that directly inhibits methanogenesis may promote the redirection of electrons and thereby facilitate the reduction of the nitroalkanes. Subsequent in vitro studies have reported significant decreases in methane production by ruminal microbial populations treated with 4-13 µmol nitroethane/mL with modest affects on volatile fatty acid production and with recovery of electrons, expressed as hydrogen recoveries, accounting for 37-52% of the decrease in methane production (Gutierrez-Bañuelos et al., 2008; Bozic et al., 2009). The fate of the remaining hydrogen was undetermined but the authors speculated that other unmeasured sinks such as ethanol, formate or anabolic processes, such as those contributing to cell growth could be functional sinks (Gutierrez-Bañuelos et al., 2008; Bozic et al., 2009) as could reduced products of nitrocompound metabolism or the production of extracellular or intracellular polysaccharide production. Several other studies using a variety of other short chain nitrocompound have supported these findings, with only a few exceptions, the latter possibly resulting from a lesser or even an inability to dispose of electrons onto the nitrocompounds or attributable to differences in cultural conditions (Table 5). For instance, dimethyl-2-nitroglutarate and 2-nitro-methyl-propionate were found to effectively decrease ruminal methane production during an in vitro batch culture

Nitrocompound	Author	Concentration	Initial culture	Methane	Effects on volatile fatty acids produced ($P < 0.05$)			
		(µmol/mL) ^a	gas phase	reduction	Total acids	Acetate	Propionate	Butyrate
3-Nitro-1- propionate	Anderson and Rasmussen, 1998	5–20	50:50 H ₂ :CO ₂	19–69%	Increased	No effect	Increased	Increased
Nitroethane	Gutierrez-Bañuelos et al., 2008	4–9	50:50 H ₂ :CO ₂	89–80%	No effect	No effect	No effect	Increased
Nitroethane	Bozic et al., 2009	13	50:50 H ₂ :CO ₂	99%	Decreased	Increased	Decreased	Increased
Nitroethane	Anderson et al., 2003	2–24	CO ₂	58-95%	Variable	Variable	Variable	Variable
Nitroethanol	Anderson et al., 2003	12	CO ₂	95%	No effect	No effect	No effect	No effect
2-Nitro-1-propanol	Anderson et al., 2003	12	CO ₂	91%	No effect	No effect	No effect	No effect
Nitroethane	Anderson et al., 2010	3–12	CO ₂	94–99%	No effect	No effect	Increased	Increased
Dimethyl-2- nitroglutarate	Anderson et al., 2010	3–12	CO ₂	92–97%	No effect	No effect	No effect	No effect
2-Nitro-methyl- propionate	Anderson et al., 2010	3–12	CO ₂	98%	No effect	No effect	No effect	No effect
Nitroethane	Zhou et al., 2011	12	10:15:85 H2:CO2:N2	None	No effect	No effect	No effect	No effect
Nitroethanol	Zhou et al., 2011	12	10:15:85 H ₂ :CO ₂ :N ₂	99%	Decreased	Decreased	Decreased	Decreased

TABLE 5 | Effects of nitrocompound supplementation on methane and volatile fatty acid production during in vitro rumen incubations.

^aConcentrations have been rounded to nearest whole number.

with ruminal microbes (Anderson et al., 2010). However, while it was anticipated that hydrolysis of the methyl esters and reduction of the nitrocompounds would yield methionine and alanine as potential endproducts of use to the host, no such evidence was obtained (Anderson et al., 2010). The microbial population had no prior exposure to these nitrocompounds; therefore it would be worthwhile to determine if adapted populations may be able to metabolize the nitrocompounds (Anderson et al., 2010).

With respect to in vivo studies, doses of 24 and 72 mg nitroethane and 40 and 120 mg 2-nitro-1-propanol (per kg body weight⁻¹ day⁻¹) resulted in methane-reducing activity being 37-69% lower than that in untreated controls, with nitroethane being more potent than 2-nitro-1-propanol (Anderson et al., 2006a). Methane-producing activity was measured via in vitro incubation of freshly collected rumen contents with excess methanogenic substrate concentrations and thus conceptually represents an indirect measure of numbers of methanogens, assuming their numbers are correlated with quantities of methane-producing enzymes. When administered to cattle at doses of 30-120 mg nitroethane kg body weight⁻¹ day⁻¹, methane-producing activity was decreased by 6 to as much as 44% when compared to controls, although methane production in vivo, measured using the sulfur hexafluoride technique, were decreased by no more than 22% (Gutierrez-Bañuelos et al., 2007; Brown et al., 2011). Oral administration of 2-nitro-1-propanol decreased ruminal acetate concentrations from that of controls by about 15% and the high nitroethane dose (120 mg kg body weight⁻¹ day⁻¹) decreased rumen acetate concentrations by about 24% in the fed steers of the study of Gutierrez-Bañuelos et al. (2007). When measured, rates of ruminal nitroethanemetabolizing activity were increased more than 30% but these increases were not always significant (Gutierrez-Bañuelos et al., 2007; Brown et al., 2011) thus suggesting that onset and duration of enrichment of competent nitroethane-metabolizing bacteria is highly variable between animals and possibly reflects deenrichment in steers fed the lower doses due to depletion of ruminal nitroethane concentrations. Similarly, there was evidence of ruminal adaptation of the ruminal population to the lower but not the higher nitroethane dose, with methaneproducing activity approaching control levels by 7 days of treatment in steers fed 80 mg nitroethane kg body weight⁻¹ (Gutierrez-Bañuelos et al., 2007). This too possibly reflects a depletion of efficacious concentrations of nitroethane in rumen contents of low-dosed steers due to consumption by nitroethane-metabolizing bacteria. In the steers fed 120 mg nitroethane kg body weight⁻¹ day⁻¹, for instance, decreases in methane-producing activity persisted to the end of the 15 day administration. Thus, it is likely the higher dose not only allowed consumption of more reductant than the lower dose but also retained sufficient residual nitroethane concentration to maintain inhibitory pressure on methanogens and sustain growth of the nitroethane-metabolizing bacteria.

More recently, other researchers have observed 75-95% decreases in methane production during in vitro incubation of mixed populations of ruminal microbes treated with 0.05-0.66 µmol/mL ethyl-3-nitrooxy propionate or 3nitrooxypropanol that suggests a greater potency of these more oxidized nitrocompounds (Martínez-Fernández et al., 2014; Romero-Pérez et al., 2015a). In the Rusitec study of Romero-Pérez et al. (2015a), hydrogen recovery in volatile fatty acid products was increased approximately 13-14% due to 3-nitrooxypropanol treatment and hydrogen accumulations were increased more than 3-fold yet recovery of all of the reductant spared from methanogenesis was not achieved. These authors also speculated that formate could possibly be an unmeasured hydrogen sink, but the potential metabolism of the nitroxy compounds to reduced products has not been discussed. A number of studies have examined the effect of 3-nitrooxypropanol on ruminal methane emissions from dairy or beef cattle and results have yielded positive results, with methane emissions, based on dry matter intake, being decreased from controls by 6 to nearly 60% (Haisan et al., 2014; Reynolds et al., 2014; Romero-Perez et al., 2014, 2015b; Hristov et al., 2015). Negative effects on animal performance, if observed, were modest. Variability in efficacy between studies possibly reflects differences in administration protocols (intraruminal vs. in feed) and methane measurement techniques. In the study of Hristov et al. (2015), as much as a 64-fold increase in hydrogen emissions was observed in 3-nitrooxypropanol-treated cattle, which was still only about 3% of the hydrogen spared from methanogenesis.

Few studies, as of yet, have determined the microbial response to nitrocompounds. At approximately 4 mM concentration, the naturally occurring nitrocompounds, 3-nitro-1-propionate and 3-nitro-1-propanol, were found to be modestly inhibitory to total culturable anaerobes from the bovine rumen, but the inhibited populations were not characterized (Anderson et al., 1993). These compounds as well as nitroethane, nitroethanol and 2-nitro-1-propanol are also inhibitory to uric acid degrading bacteria of poultry origin (Kim et al., 2005, 2009). Additionally, nitrocompounds, similarly to nitrates, inhibit the growth of many foodborne pathogens including *Listeria*, *Salmonella*, and *Campylobacter* and enteropathogenic *E. coli* (Jung et al., 2003, 2004a,b; Anderson et al., 2006b; Dimitrijevic et al., 2006), a use that has been patented (Anderson et al., 2007).

FACTORS THAT INFLUENCE NITRATE AND NITROCOMPOUND METABOLISM

There is a significant amount of variation between animals with regard to their response to nitrate addition. Part of this response is due to the rumen ecosystem. The largest influencer would be the endogenous microbial populations and their enzymatic capacity. However, there are many other factors that influence the rate of reduction of nitrate, nitrite, and nitrocompounds.

First, the availability and amount of substrate supplying electrons used for the reductive nitrate reactions including carbohydrates, ethanol, mannitol, glycine, malate, citrate, lactate, succinate, pyruvate and formate (Lewis, 1950, 1951; Jones, 1972). Fatty acids, acetic, propionic, and butyric are not electron donors for nitrate reduction under this metabolic process. Thus, supplementing or provision of diets rich with these electron donors can potentially enhance nitrate reduction to ammonia, thereby further enhancing the decrease in methane and reducing the chances of nitrate poisoning as the animal will be exposed to less of the toxic intermediate nitrite which converts hemoglobin to methemoglobin also known as methemoglobinemia (Takahashi et al., 1980).

Dietary interventions such as high starch and cereal grains have been proposed to help mitigate nitrate toxicity (Burrows et al., 1987; Hibberd et al., 1994) purportedly by stimulating the metabolism of ruminal microorganisms thereby promoting production and subsequent availability of electron-donating substrates for nitrate and nitrite metabolism (Lewis, 1950, 1951; Allison and Reddy, 1984). High starch diets, within a pH range of about 6.6-6.8, have been shown to increase hydrogen production by mixed populations of ruminal microbes in vitro (Lin et al., 2013b). However, high starch diets are also well established to decrease rumen pH, which could potentially create an environment inhibitory to ruminal nitrate and nitrite metabolism. While enzymatic analysis indicates nitrate and nitrite reductases work optimally at pH of 6.5 or 5.6, respectively (Lewis, 1951; Tillman et al., 1965), nitrate- and nitrite-reducing activity within mixed populations of ruminal microbes was more rapid at neutral pH than at pH 6.0 or lower (Iwamoto et al., 2001). The inhibitory activity of the lower pH on nitrate- and nitrite-reducing activity within mixed rumen populations was attributed to lower availability of electron-donating substrates such as hydrogen, formate, or lactate resulting from an inhibition of fermentation caused by the low pH (Iwamoto et al., 2001). This would suggest that nitrate supplemented to concentrate fedcattle may be ineffective or at least less effective in decreasing ruminal methane emissions than when supplemented to forage fed cattle. From a methane mitigation perspective, this may be of little consequence as the low ruminal pH in high concentratefed ruminants also inhibits methane production (Van Kessel and Russell, 1996; Lana et al., 1998; Russell, 1998) and thus concentrate-fed ruminants emit significantly less methane than do forage-fed cattle (Johnson and Johnson, 1995). From a toxicity perspective, however, it seems likely that supplementing nitrate to animals expected to have low rumen pH may increase risk for nitrite accumulation and subsequent toxicosis due to methemoglobinemia.

The next consideration is adaptation of the rumen microbial population to high nitrate diets. As has been alluded to earlier, the ruminal microbial population can be selected for increased nitrate- and nitrite-reducing activity by gradual feeding of increasing amounts of nitrate. However, little is known as to what might happen to the microbial population if an animal is de-adapted and selective pressure is removed. For instance, it is reasonable to suspect that should an animal go off feed and refuse its meals for a certain period of time a disbalance between rates of nitrate reduction and nitrite reduction may occur, which if favoring high nitrite accumulation may place the animal at greater risk to nitrite intoxication. Establishing or dosing the animal with a population of probiotic bacteria capable of reducing nitrite to non-toxic forms may also reduce the concentration of nitrite in the rumen. Currently, there is one patent, which is sold as a directfed microbial additive for the prevention of nitrite toxicity in ruminants fed nitrate. It employs a denitrifying strain of P. acidipropionici; however, its nitrate and especially its nitrite reductive capacity have not been proven to remediate nitrate toxicity nor enhance methane reduction (de Raphélis-Soissan et al., 2014). There is also a genetically modified E. coli strain that has been developed with enhanced nitrite utilization that has been proven to work in vitro and in vivo (Sar et al., 2005, 2006). However, its status as a GMO and as a strain of a

pathogenic genus may create a barrier too large for its usage in production.

Adaptation can occur at the animal, or systemic, level as well. Ruminants increase the amount of hemoglobin, red blood cells, and blood volume, thereby increasing the capacity to deliver oxygen and counter the effect of increasing methemoglobin. This process occurs over months (Jainudeen et al., 1964). In addition, NADPH reductase will convert methemoglobin back to hemoglobin (Hibberd et al., 1994). It is reasonable to hypothesize that animals adapted to nitrate have an elevated concentration of NADH reductase, but this has not been documented.

The rate of nitrate reduction also appears to be influenced by sulfur availability and concentrations. The reduction of sulfate to sulphite and then sulfide is less energetically favorable than nitrate to nitrite to ammonia. Opportunistically the enzymes for each reaction generally work for both reductive processes due to the molecule's similarity in structure and charge (Thauer et al., 1977). In addition, sulfur stimulates the growth of sulfidereducing bacteria. Therefore, the addition of sulfur should hypothetically increase the amount of nitrate and nitrite reduced and the speed of the reaction (Leng, 2008). Experimentally, Lcysteine, which is rich in sulfur, in conjunction with nitrate has been shown to suppress the formation of methemoglobin (Takahashi et al., 1998).

Less is known regarding factors affecting nitrocompound metabolism within the rumen, although protein supplementation enhances *in vivo* rates of ruminal 3-nitro-1-propanol metabolism in cattle (Majak, 1992) and the reducing substrates, hydrogen gas and formate, are known to be stimulatory to reduction of the nitrocompounds *in vitro* (Anderson et al., 1993). Moreover, supplementing ruminal populations with ferrous and sulfide ions markedly increased rates of 3-nitro-1-propionic acid and 3-nitro-1-propanol metabolism (Anderson et al., 1993), but had little if any effect on the metabolism of these nitrocompounds in populations of equine cecal microbes (Zhang et al., 2014). Mechanistically, the ferrous and sulfide ion additions were thought to promote ferredoxin-hydrogenase mediated electron transfer reactions contributing to the reduction of the nitrocompounds.

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CONCLUSIONS

Nitrate and nitrite reduction to ammonia in the rumen is a more thermodynamically favorable reaction than the formation of methane with carbon dioxide as an electron acceptor. The effectiveness and risks of toxicity of this strategy are dependent on the nitrate salt type with sodium nitrate being the most biologically available and calcium the least in the rumen environment. A portion, albeit small, of the introducted nitrate appears to be metabolized to nitrous oxide either via dissimilatory nitrate reduction or more likely via incomplete denitrification, which may lessen its net greenhouse gas mitigation. A variety of certain short chain nitrocompounds as well as some nitro-oxy compounds have also been shown to decrease ruminal methane production. The naturally occurring nitrocompound, 3-nitro-1-propionate, is known to be metabolized by ruminal microbes to β-alanine, a non-essential amino acid that may be metabolized by the host and potentially used as a source of carbon, nitrogen and energy making it an attractive candidate. Safe and successful feeding of supplemental nitrate and nitrocompounds requires careful adaptation of the ruminal microbes to prevent risks of toxic intermediates and this practice could benefit via concurrent feeding of appropriate nitrite-reducing or nitrocompound degrading probiotic bacteria. Additionally, combination feeding of 3-nitro-1-propionate, or another appropriate nitrocompound, with subtoxic amounts of nitrate may yield synergistic advantages in inhibition of rumen methanogensis and electron capture in non-methane products.

More research is needed, however, on the pathways involved in nitrate and nitrocompound metabolism, the organisms involved, and the regulation of their enzyme activities in order to mitigate concerns that persist over the risks of toxicities and realize the full potential of these methane-decreasing strategies.

AUTHOR CONTRIBUTIONS

EL: nitrate section and editing, RA: nitrocompound and editing, WP: editing, DN: supervision.

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Conflict of Interest Statement: The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

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