



# Effects of Three Types of Organic Fertilizers on Greenhouse Gas Emissions in a Grassland on Andosol in Southern Hokkaido, Japan

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Reduction of chemical fertilizers and effective use of livestock excrement are required for the realization of sustainable agriculture and reduction of greenhouse gas (GHG) emissions. The purpose of this study was to estimate the reduction rate of GHG emissions represented by comparing global warming potential (GWP) using organic fertilizers instead of chemical fertilizers. The study was conducted in a managed grassland on Andosol in southern Hokkaido for 3 years from May 2017 to April 2020. There were five treatment plots: no fertilizer, chemical fertilizer, manure, slurry, and digestive fluid. Organic fertilizers were applied such that the amount of NPK did not exceed the recommended application rate, and the shortage was supplemented with chemical fertilizers. Fluxes in CO<sub>2</sub> caused by heterotrophic respiration (RH), CH<sub>4</sub>, and N<sub>2</sub>O were measured using the closed chamber method. Net ecosystem carbon balance (NECB) was obtained as net primary production + organic fertilizer application – RH – harvest. The GWP was estimated by CO<sub>2</sub> equivalent NECB and CH<sub>4</sub> and N<sub>2</sub>O emissions in each treatment. Chemical fertilizer nitrogen application rates in the organic fertilizer treatments were reduced by 10% for manure, 19.7% for slurry and 29.7% for digestive fluid compared to chemical fertilizer only, but the grass yields were not significantly different among the fertilizer treatments. The 3-year NECB showed significantly smallest carbon loss in manure treatment, and smaller carbon loss in the organic fertilizer treatments than in the chemical fertilizer only. The reduction rate in the GWP with use of organic fertilizers relative to that of chemical fertilizer was 16.5% for slurry, 27.0% for digestive fluid, and 36.2% for manure. The NECB accounted for more than 90% of the GWP in all treatments. CH<sub>4</sub> emissions were < 0.1% of the GWP. On the other hand, N<sub>2</sub>O emissions accounted for more than 5% of the GWP, and was larger in the order of slurry > chemical fertilizer only > digestive fluid > manure. As a conclusion, these organic fertilizers can be used without no reduction of crop yield instead of chemical fertilizer, however, manure is the best way to increase soil carbon and to decrease GWP, followed by digestive fluid.

**Keywords:** CH<sub>4</sub>, global warming potential, manure, methane fermentation digestive fluid, slurry, N<sub>2</sub>O, soil carbon sequestration

## INTRODUCTION

The anthropogenic impact on the climatic system has increased annually, and greenhouse gas (GHG) emissions in 2018 reached a record high of 55.3 Gt CO<sub>2</sub> eq yr<sup>-1</sup> (UNEP, 2019). Approximately 24% of the GHG emissions come from agriculture, forestry and other land use (AFOLU) (IPCC, 2014). Mitigation in the AFOLU sector is urgently needed.

Soil is the largest carbon storage pool, approximately twice the amount of carbon in the atmosphere and three times the amount in terrestrial biomass (Schlesinger and Jeffrey, 2000). However, agricultural soil loses soil carbon because of organic matter decomposition and erosion, and its recovery is required (Lal, 2020). Furthermore, agriculture is the largest source of CH<sub>4</sub> and N<sub>2</sub>O (Blandford and Hassapoyannes, 2018). Therefore, improvement of carbon storage in farmland and reduction of CH<sub>4</sub> and N<sub>2</sub>O emissions from farmland are important as climate change mitigation measures in agriculture.

Grasslands are a very important ecosystem for the production of herbivorous livestock (Soussana et al., 2007). Because grasslands are not tilled for several years to several decades, the organic matter content in the surface soil increases because of plant residues, livestock excreta, or organic matter derived from applied manure (Ciais et al., 2013). A 3-year study in valley inland and coastal grasslands in California showed that manure application increased soil carbon by 26 and 37%, respectively (Ryals et al., 2014). These show the soil carbon sequestration reducing the concentration of CO<sub>2</sub> in the atmosphere (Paustian et al., 1997). Therefore, when organic matter application and no-tillage are adapted continuously, grasslands are expected to exhibit climate change mitigation effects as a carbon storage in agricultural soil.

Evaluation of carbon storage in agricultural land includes carbon output from the harvest system and carbon input into the system by the application of organic matter in addition to carbon cycling in the ecosystem through the atmosphere, plants, and soil (Shimizu et al., 2009). Studies on the net ecosystem carbon balance (NECB) and GHG balance caused by manure application in southern Hokkaido, Japan showed that although CO<sub>2</sub> emissions increased because of manure application, there were no differences in CH<sub>4</sub> and N<sub>2</sub>O emissions, and the carbon input from manure application reduced the global warming potential (GWP) (Mukumbuta et al., 2017a). However, N<sub>2</sub>O emissions in the manure and chemical fertilizer combinedly applied grasslands tended to be higher than the chemical fertilizer only applied grasslands (Shimizu et al., 2013). A study comparing the difference in CH<sub>4</sub> and N<sub>2</sub>O emissions from soil with manure or slurry application in grasslands in northern Tochigi Prefecture, Japan, showed no significant difference between the two organic fertilizers (Mori and Hojito, 2015). Research on the environmental factors controlling CO<sub>2</sub>, N<sub>2</sub>O, and CH<sub>4</sub> fluxes has shown that CO<sub>2</sub> flux has a significant relationship with soil temperature (Shimizu et al., 2009), N<sub>2</sub>O flux increased from 60% water-filled pore space (WFPS) peaking at 80% WFPS (Katayanagi et al., 2008), and CH<sub>4</sub> was normally absorbed by soil but the CH<sub>4</sub> uptake decreased with nitrogen application

(Hu et al., 2002), and CH<sub>4</sub> emitted from poorly drained soil (Shimizu et al., 2013).

In recent years, the use of livestock manure for methane fermentation has increased from the perspective of treating livestock excrement (Holm-Nielsen et al., 2009). Biogas energy can be obtained by fermenting livestock manure. Utilizing the methane fermentation digestive fluid, which is the fermentation residue, as liquid fertilizer not only prevents the outflow of pollutants into rivers, but also provides a supply of nutrients to farmlands and resource-recycling for livestock farming. Despite being fermented, the methane fermentation digestive fluid can be used as a liquid fertilizer with the same components as the raw slurry material (Matsunaka et al., 2003) and has no foul odor as compared with the slurry (Immovilli et al., 2008).

Different organic fertilizers show different physicochemical properties (Harada et al., 1993; Mori and Hojito, 2015). In particular, methane fermentation digestive fluid tends to have a higher pH and a higher ammonium nitrogen concentration than slurries (Yuyama et al., 2007). Therefore, application of digestive fluid makes soil nutrient status and can reduce the application of chemical fertilizer. Furthermore, increase of soil pH decreased N<sub>2</sub>O emissions (Mukumbuta et al., 2018). However, digestive fluid had lower C/N ratio than raw slurry (Holly et al., 2017), which can increase N<sub>2</sub>O emission (Toma and Hatano, 2007). On the other hand, a study in Wisconsin showed there was no significant difference of N<sub>2</sub>O production between digestive fluid and raw slurry applications (Holly et al., 2017).

Effect of organic fertilizers on soil moisture is also an important factor. Digestive fluid has a high water content and increases soil moisture just after the application. Soil moisture is a significant factor influencing nitrogen mineralization, nitrification and denitrification, which strongly influence N<sub>2</sub>O production in soil (Linn and Doran, 1984). Also increase of soil moisture may increase CH<sub>4</sub> in upland fields, and the CH<sub>4</sub> emission remaining in the digestive fluid during the fermentation reaction can occur after application to upland fields (Nakamura et al., 2008).

Since organic fertilizers do not always contain NPK in the best balance for crop growth, it is necessary for farmers to properly manage nutrients for the application of organic fertilizer. For this, for example in Hokkaido, the local government suggests an upper limit of the application rate of organic fertilizer to prevent excessive nutrients being applied and recommends that insufficient nutrients induced by this is supplemented using chemical fertilizers (Hokkaido Government Agricultural Department, 2015). Therefore, NPK composition of organic fertilizers influence the application rate of organic fertilizer and reduction rate of chemical fertilizer, which influence NECB and GHG balance.

Therefore, in this study, influences of three organic fertilizers (manure, slurry and digestive fluid) treatments on NECB and GHG balance in grassland are compared with chemical fertilizer only and control (no fertilizer) treatments. The GHG emissions from grassland soil, crop growth and harvest and organic matter application with the five treatments for 3 years in a grassland southern Hokkaido, Japan were measured. The emission factor of N<sub>2</sub>O in managed upland soil which is used in the IPCC

guideline for the National GHG Inventory Report (IPCC, 2006) was also calculated.

In this study, following results were expected: (1) The three types of organic fertilizers have the similar effect of fertilization and can reduce the amount of chemical fertilizer application rate; (2)  $N_2O$  emissions are lower in organic fertilizer treatments than in chemical fertilizer only treatment due to the reduction in chemical fertilizer nitrogen application rate; (3) The contribution of  $CH_4$  emissions to total GHG emissions is small; (4) NECB becomes manure > slurry > digestive fluid treatments, which is significantly larger than that in chemical fertilizer only treatment.

## MATERIALS AND METHODS

### Study Site

This study was conducted in a grassland cultivating reed canary grass in the Shizunai Experimental Livestock Farm, Field Science Center for the Northern Biosphere of Hokkaido University in Southern Hokkaido, Japan (Shizunai) ( $42^{\circ}26'05.4''N$ ,  $142^{\circ}28'52.1''E$ ) from May 2017 to April 2020. The study site has a humid continental climate, with cold winters and cool summers. The average temperature over the past 10 years (2007–2016) was  $8.4^{\circ}C$ , annual rainfall was 1,273 mm, deepest monthly snow was 1 to 22 cm, and snowfall of 10 cm or more was observed from December to March.

The soil was derived from Tarumae (b) volcanic ash, and the mottled upper end of the layer appears within the 0–50 cm soil horizon, and consequently was classified as Wet Andosols (The Fifth Committee for Soil Classification and Nomenclature of the Japanese Society of Pedology, 2017). The soil properties of the 0–7 cm surface layer (Ap1) were pH ( $H_2O$ )  $5.64 \pm 0.04$ , total carbon  $36.7 \pm 1.74 g kg^{-1}$ , total nitrogen  $2.7 \pm 0.05 g kg^{-1}$ , and C/N ratio 13.4. Before 2017, when this research began, the study site had been used as a grassland since 2009, and fertilization with chemical fertilizer and harvest were conducted twice a year. From 2009 to 2016, nitrogen, phosphorus, and potassium were applied as chemical fertilizer at an average of  $86 kg T-N ha^{-1} yr^{-1}$ ,  $71 kg P_2O_5 ha^{-1} yr^{-1}$ , and  $104 K_2O ha^{-1} yr^{-1}$ , respectively. Except for 2010,  $10 Mg FM ha^{-1} yr^{-1}$  of manure was applied every year after September when the second grass was harvested, and liquid urine fertilizer was also applied in 2012.

### Fertilization Treatments and Field Management

The study period consisted of 3 years from May 14, 2017 to April 26, 2020, including May 14, 2017 to April 26, 2018 (348 d), April 27, 2018 to April 26, 2019 (365 d), April 27, 2019, to April 26, 2020 (366 d). Fertilization was conducted twice a year with a base fertilizer (spring) and supplement fertilizer (summer). Harvest was performed twice a year for the first and second grasses. In this study, five treatments of fertilization were tested: no fertilizer (N), chemical fertilizer (F), manure (M), slurry (S), and digestive fluid (D). Fifteen subplots of  $5 \times 10 m$  were set up in five treatments  $\times$  three replicates in a random block design. In each subplot, a  $5 \times 8 m$  vegetation survey area, a  $5 \times 2 m$  gas sampling area, and a  $50 \times 50 cm$  bare area, excluding roots, was set up in the gas sampling area. In the bare area, a root permeable sheet (BKS9812,

TOYOBO CO. Ltd., OSAKA, Japan) was inserted at a depth of  $\sim 30 cm$  at the boundary with the planting area to prevent the entry of roots. Plants growing in the bare area during the survey period were regularly removed by hand.

### Organic Fertilizer Used

Manure, slurry, and digestive fluid were used as organic fertilizers. Every year, the manure used was from Shizunai, and the digestive fluid was from Sapporo Experimental Farm, Field Science Center for the Northern Biosphere of Hokkaido University (Sapporo) ( $43^{\circ}04'41.1''N$ ,  $141^{\circ}20'03.6''E$ ). The manure was made from a mixture of cow excreta, horse excreta, and bedding litter and turned over once every 10 days during winter. The slurry used was from Shizunai in 2017. However, the Shizunai slurry had a high water content and a low nitrogen content because it was mixed with rainwater. Due to this, in 2018 and 2019, Sapporo slurry was used. The slurry in Shizunai was from cow excreta, horse excreta, and rainwater and was stored in a slurry reservoir in the barn until use. The slurry in Sapporo was made from cattle, pig, and chicken excreta, and water was added as appropriate to increase fluidity. The digestive fluid used was from Sapporo, which was made from methane fermentation of the slurry in Sapporo. Each organic fertilizer was collected 1 month before application, and water content, pH, TN,  $NH_4^+-N$ , P, K, and TC were analyzed. The components of each organic fertilizer are shown in **Table 1**.

### Design of Fertilization

**Table 2** shows the application rates of organic and chemical fertilizers for each year. The nitrogen application rate depended on the legume rate (Hokkaido Government Agricultural Department, 2015). In 2017 and 2019, because the legume rate was 5–15%, nitrogen application rate was  $100 kg ha^{-1}$ . On the other hand, in 2018, the nitrogen application rate increased to  $160 kg ha^{-1}$  because the legume rate decreased to  $< 5\%$ . The organic fertilizer application rate was determined such that the organic fertilizer N, P, or K application rates did not exceed the recommended application rate of N, P, or K, and any shortage in N, P, or K was made up by chemical fertilizers. Chemical fertilizer was applied at the ratio of application at the base: supplement of 2:1, whereas organic fertilizer was applied only used as a base application. Both chemical and organic fertilizers were applied by top dressing.

## Measurements

### Environmental Factors

Daily air temperature and precipitation were obtained from the close Automated Meteorological Data Acquisition System (AMeDAS) station of the Japan Meteorological Agency, which are located about 14 km from the study site for air temperature and about 100 m from the study site for precipitation. Soil temperature at a 5 cm depth was measured at the same time as the gas flux measurements using a thermistor thermometer (CT-414WR, CUSTOM, Tokyo, Japan), and volumetric soil moisture content at 0–6 cm depth was measured using the frequency domain reflectometry (FDR) method (DIK-311A; Daiki, Saitama,

**TABLE 1** | Chemical components of organic fertilizer used.

		Water content	pH	TN	NH <sub>4</sub> <sup>+</sup> -N	P <sub>2</sub> O <sub>5</sub>	K <sub>2</sub> O	TC	C/N
		%		%FM	%FM	%FM	%FM	%FM	
2017	M	70.0 ± 0.49	8.39 ± 0.06	0.34 ± 0.17	0.02 ± 0.00	1.27 ± 0.38	3.81 ± 0.47	9.49 ± 0.50	28.07
	S	96.8 ± 0.05	7.93 ± 0.01	0.11 ± 0.02	0.07 ± 0.00	0.07 ± 0.02	1.53 ± 0.06	0.79 ± 0.01	7.15
	D	95.7 ± 0.05	7.85 ± 0.01	0.19 ± 0.03	0.13 ± 0.00	0.27 ± 0.10	0.43 ± 0.04	1.56 ± 0.06	8.36
2018	M	78.4 ± 0.88	8.52 ± 0.32	0.62 ± 0.02	0.09 ± 0.03	3.43 ± 0.22	2.45 ± 0.31	8.95 ± 0.50	14.33
	S	92.9 ± 0.10	7.22 ± 0.03	0.21 ± 0.01	0.19 ± 0.00	0.53 ± 0.00	0.58 ± 0.01	2.93 ± 0.09	13.77
	D	94.9 ± 0.14	7.79 ± 0.03	0.19 ± 0.00	0.18 ± 0.01	0.49 ± 0.02	0.53 ± 0.09	1.95 ± 0.04	9.88
2019	M	73.6 ± 0.88	7.91 ± 0.07	0.58 ± 0.04	0.10 ± 0.00	1.73 ± 0.39	3.08 ± 0.39	11.1 ± 0.23	19.17
	S	93.1 ± 0.09	6.05 ± 0.00	0.24 ± 0.01	0.12 ± 0.00	0.37 ± 0.12	0.36 ± 0.04	2.98 ± 0.17	12.22
	D	95.1 ± 0.08	7.56 ± 0.05	0.24 ± 0.00	0.13 ± 0.00	0.24 ± 0.12	0.36 ± 0.12	2.08 ± 0.04	8.81

Values represent mean ± standard deviation. FM is the fresh weight, M is the manure plot, S is the slurry plot, and D is the digestive fluid plot. n = 3.

**TABLE 2** | Annual organic fertilizer application rate each year, chemical fertilizer application to chemical fertilizer plots, and chemical fertilizer supply to organic fertilizer plots.

		Organic fertilizer					Chemical fertilizer			
		Application rate	TC	TN	P <sub>2</sub> O <sub>5</sub>	K <sub>2</sub> O	C/N	TN	P <sub>2</sub> O <sub>5</sub>	K <sub>2</sub> O
		Mg FM ha <sup>-1</sup>	kg ha <sup>-1</sup>					kg ha <sup>-1</sup>		
2017	N	0	0	0	0	0	–	0	0	0
	F	0	0	0	0	0	–	100	80	180
	M	8.4	797	28	106	320	28.1	95	59	0
	S	19	152	21	14	293	7.2	93	64	0
	D	58	914	109	154	247	8.4	65	38	0
2018	N	0	0	0	0	0	–	0	0	0
	F	0	0	0	0	0	–	160	80	180
	M	9.0	806	56	309	221	14.3	139	5.2	0
	S	38	1,115	81	204	221	13.8	127	0	0
	D	26	508	51	130	140	9.9	123	27	60
2019	N	0	0	0	0	0	–	0	0	0
	F	0	0	0	0	0	–	100	80	180
	M	7.4	828	43	128	228	19.2	90	23	0
	S	62	1,849	151	232	227	12.2	69	0	0
	D	42	874	99	101	153	8.8	66	40	58

FM is the fresh weight, N is the no fertilizer plot, F is the chemical fertilizer plot, M is the manure plot, S is the slurry plot, and D is the digestive fluid plot.

Japan). The water-filled pore space (WFPS,%) was calculated as:

$$WFPS = \left( \frac{\theta}{p} \right) \times 100 \quad (1)$$

where  $\theta$  is volumetric soil water content ( $\text{m}^3 \text{m}^{-3}$ ) and  $p$  is soil porosity ( $\text{m}^3 \text{m}^{-3}$ );  $p$  was measured by a three-phase meter (DIK-1150; Daiki, Saitama, Japan).

Soil sampling was conducted at the same time as the gas flux measurement during from April to November when the soil was not frozen. The collected soil was sieved at 2 mm. Soil NO<sub>3</sub><sup>-</sup>-N content was determined by water extraction with the ratio of soil: deionized water = 1:5, and the NO<sub>3</sub><sup>-</sup>-N concentration in the water extraction was measured by ion chromatography (DIONEX ICS-1100; Thermo Fisher Scientific, MA, USA). Soil

NH<sub>4</sub><sup>+</sup>-N content was determined by KCl extraction with the ratio of soil:KCl (2 mol L<sup>-1</sup>) = 1:10, and the NH<sub>4</sub><sup>+</sup>-N concentration in the KCl extraction was measured by the indophenol blue method.

### Gas Fluxes

Soil CO<sub>2</sub>, CH<sub>4</sub>, and N<sub>2</sub>O fluxes were measured by the static closed chamber method (Toma and Hatano, 2007). In the gas sampling area of each treatment, chambers made of stainless steel were installed. Chambers with a diameter of 40 cm and a height of 30 cm were used to measure CH<sub>4</sub> and N<sub>2</sub>O fluxes in the planting area, and those with a diameter of 20 cm and a height of 25 cm were used to measure the CO<sub>2</sub> flux in the bare plot, which was assumed to correspond to microbial heterotrophic respiration (RH). The chambers were placed onto bases, which

were permanently installed during the measurement period. The chamber bases were inserted into the soil to a depth of 5 cm for at least 12 h before the first gas sampling. During the snowfall period, chamber bases were set up directly onto the snow (Katayanagi and Hatano, 2012). Gas flux measurements were performed between 8:00 and 13:00 for seven consecutive days after the application of the fertilizer, once a week during the plant growing season and once a month during winter. Changes in the concentrations of CO<sub>2</sub>, CH<sub>4</sub>, and N<sub>2</sub>O in the headspace of the chambers with time was measured according to a previously reported procedure (Nakano et al., 2004; Toma and Hatano, 2007; Shimizu et al., 2013), that is, gas samples in the chamber headspace were taken at 0 and 6 min for CO<sub>2</sub> and 0, 15, and 30 min for CH<sub>4</sub> and N<sub>2</sub>O after closing chambers by using a 25 mL gas tight syringe. A gas sample of 250 mL was injected into a 500 mL Tedlar bag for CO<sub>2</sub>. For CH<sub>4</sub> and N<sub>2</sub>O, each gas sample (20 mL) was placed into an evacuated glass vial (10 mL). The CO<sub>2</sub> concentration was determined using an infrared CO<sub>2</sub> analyzer (Model ZEP9GC11; Fuji Electric, Tokyo, Japan), and the CH<sub>4</sub> and N<sub>2</sub>O concentrations were determined using gas chromatography equipped with a flame ionization detector (GC-8A; Shimadzu, Kyoto, Japan) and an electron capture detector (GC-14B; Shimadzu, Kyoto, Japan), respectively.

The gas flux from the soil was calculated using the following linear regression equation (Toma et al., 2011):

$$F = \rho \times \left(\frac{V}{A}\right) \times \left(\frac{\Delta c}{\Delta t}\right) \times \left(\frac{273}{T}\right) \times \alpha \quad (2)$$

where F is the gas flux (mg C m<sup>-2</sup> h<sup>-1</sup> for CO<sub>2</sub> and CH<sub>4</sub>, mg N m<sup>-2</sup> h<sup>-1</sup> for N<sub>2</sub>O); ρ is the density of each gas under standard conditions (CO<sub>2</sub> = 1.997 × 10<sup>6</sup> mg m<sup>-3</sup>, CH<sub>4</sub> = 0.717 × 10<sup>6</sup> mg m<sup>-3</sup>, N<sub>2</sub>O = 1.978 × 10<sup>6</sup> mg m<sup>-3</sup>), V is the volume of the chamber (m<sup>3</sup>), A is the surface area of the chamber (m<sup>2</sup>), Δc/Δt is the rate of change in gas concentration in the head space of the chamber during the sampling time (10<sup>-6</sup> m<sup>3</sup> m<sup>-3</sup> h<sup>-1</sup>); T is the air temperature inside the chamber (°C); and α is the ratio of molar mass of carbon to the molecular weight of CO<sub>2</sub> and CH<sub>4</sub>, or of nitrogen to N<sub>2</sub>O. For N<sub>2</sub>O and CH<sub>4</sub>, the Δc/Δt of R more than 0.95 was used for the flux calculation. The flux of CO<sub>2</sub> was calculated at two points of 0 and 6 min based on the theoretical consideration that the increase in the CO<sub>2</sub> concentration in the chamber loses linearity after about 8 min (Nakano et al., 2004). However, the relationship between the multiple-times sampling and the two-points sampling is linear of 1: 1 (Mukumbuta et al., 2017b).

Cumulative gas emissions were calculated by linear interpolation between sampling events and numerical integration of the underlying area using the trapezoid rule as follows (Jin et al., 2010):

$$\text{Cumulative gas emission} = \sum_{i=1}^n (R_i \times 24 \times D_i) \quad (3)$$

where R<sub>i</sub> is the mean gas flux (mg m<sup>-2</sup> h<sup>-1</sup>) of the two successive sampling dates, D<sub>i</sub> is the number of days in the sampling interval and n is the number of sampling times.

## Plant Production and Harvest

Aboveground biomass, belowground biomass were measured four times a year including two times of harvest, mid-April (beginning of crop growing season), late June (the first crop harvest), early September (the second crop harvest), and early November (end of crop growing season), that is, total 13 times from April 2017 to April 2020. Grass samples were taken from the vegetation survey areas of each treatment plot. All the aboveground biomass, including green and dead biomass were collected from the 0.5 × 0.5 m quadrat in April and November. Aboveground biomass at the time of harvest was obtained as the sum of harvest and residue. The harvest was measured by clipping at 5 cm above the ground in the 1 × 1 m quadrat. The residue was measured by collecting the stubbles and dead biomass in a 0.5 × 0.5 m quadrat. Regarding the belowground biomass, the root samples were collected from the 0.5 × 0.5 m area × 0.3 m deep by collecting soil and passing through an 8 mm sieve in the field. Roots were washed in a 2 mm sieve in the laboratory. All the samples were oven-dried at 70°C for 72 h and weighed. Each dried sample was analyzed for total carbon content.

Net primary production (NPP) was estimated as the increments of aboveground and belowground biomass (Mu et al., 2006), that is, annual aboveground NPP (ANPP) and belowground NPP (BNPP) were estimated as follows:

$$\begin{aligned} ANPP = & H(1) + R(1) - ABb + H(2) + R(2) - R(1) \\ & + ABe - R(2) + ABb' - ABe \end{aligned} \quad (4)$$

where H and R are the harvest and the residue at crop harvest, respectively (1 and 2 in the parentheses mean the first and second crop harvest, respectively); ABb, ABe, and ABb' are the aboveground biomass at the beginning and the end of crop growing season and the beginning of crop growing season in next year, respectively.

Equation 4 can be shortened as follows:

$$ANPP = ABb' - ABb + H(2) + H(1) \quad (5)$$

As belowground biomass is not harvested, BNPP can be obtained as follows:

$$BNPP = BBb' - BBb \quad (6)$$

where BBb and BBb' are the belowground biomass at the beginning of crop growing season and in the next year, respectively.

## Calculations

### N<sub>2</sub>O Emission Factor

The N<sub>2</sub>O emission factor indicates the cumulative N<sub>2</sub>O emissions per unit applied nitrogen. According to the calculation method proposed by Shimizu et al. (2013), the N<sub>2</sub>O emission factors derived from chemical fertilizers and organic fertilizers were calculated as follows:

$$EF_{CF} = \frac{E_{CF} - E_{NF}}{N_{CF \text{ in } CF \text{ plot}}} \times 100 \quad (7)$$

$$EF_{OF} = \frac{\{E_{OF} - (N_{CFin\ OF\ plot} \times EF_{CF}) - E_{NF}\}}{N_{OFin\ OF\ plot}} \times 100 \quad (8)$$

where  $EF_{CF}$  is the  $N_2O$  emission factor for chemical fertilizer (%),  $EF_{OF}$  is the  $N_2O$  emission factor for organic fertilizer (%);  $E_{CF}$ ,  $E_{NF}$ , and  $E_{OF}$  are the  $N_2O$  emissions in the chemical fertilizer plot, no fertilizer plot, and organic fertilizer plot, respectively ( $kg\ N\ ha^{-1}$ );  $N_{CF}$  in CF plot and  $N_{OF}$  in OF plot were the chemical fertilizer N application rates in the chemical fertilizer plot and organic fertilizer plot ( $kg\ N\ ha^{-1}$ ), respectively, and  $N_{OF}$  in the OF plot was the organic fertilizer N application rate ( $kg\ N\ ha^{-1}$ ).

### Net Ecosystem Carbon Balance

Net ecosystem carbon balance (NECB) was obtained as net biome production (Schulze et al., 2000). The NECB in agricultural land was obtained by adding carbon input from the application of organic fertilizer ( $C_{input}$ ) and carbon export via harvest ( $C_{output}$ ) for net ecosystem production (NEP). The NEP is estimated as the difference between net primary production (NPP) by photosynthesis of plants and heterotrophic respiration (RH) by decomposition of soil organic matter. Concerning carbon emission with  $CH_4$  flux, in uplands,  $CH_4$  flux is known to be very small compared to  $CO_2$  (Toma et al., 2011), therefore it was not included in the NECB calculation. Therefore, NECB ( $Mg\ C\ ha^{-1}\ yr^{-1}$ ) is calculated as follows:

$$NECB = C_{input} + ANPP + BNPP - C_{output} - RH \quad (9)$$

### GWP and GHG Balance

The NECB and cumulative emissions of  $CH_4$  and  $N_2O$  were converted to  $GWP_{CO_2}$ ,  $GWP_{CH_4}$ , and  $GWP_{N_2O}$ , respectively, using the  $CO_2$  conversion coefficient [ $CO_2$ :1,  $CH_4$ :28,  $N_2O$ : 265 (IPCC, 2014)]. The GHG balance ( $Mg\ CO_2\ eq\ ha^{-1}\ yr^{-1}$ ) was obtained as GWP, which are the sum of  $GWP_{CO_2}$ ,  $GWP_{CH_4}$ , and  $GWP_{N_2O}$  as follows:

$$GWP_{CO_2} = -NECB \times \frac{44}{12} \quad (10)$$

$$GWP_{CH_4} = CH_4 \times \frac{16}{12} \times 28 \quad (11)$$

$$GWP_{N_2O} = N_2O \times \frac{44}{28} \times 265 \quad (12)$$

$$GWP = GWP_{CO_2} + GWP_{CH_4} + GWP_{N_2O} \quad (13)$$

### Statistical Analysis

The Shapiro-Wilk test was performed on each GHG flux, environmental factors, cumulative GHG emissions, carbon balance, and GHG balance to confirm normality. If normality was not found, logarithmic conversion was performed and the test was performed again to confirm normality. Differences in GHG emissions among years and among treatments were tested using a two-way analysis of variance (ANOVA). Differences in 3-year total GHG emissions, net ecosystem carbon balance and GHG balance among treatments were tested using a one-way ANOVA. If a significant difference ( $p < 0.05$ ) occurred in the test, multiple comparisons were performed using the Tukey HSD method. In order to explain the relationship between the C/N and

$N_2O$  emission factors of organic fertilizers, the normality of each was confirmed by the Shapiro-Wilk test, and a simple regression analysis was performed. The analysis was performed using R (R Development Core Team, 2018; version 3.5.1).

## RESULTS

### Environmental Factors

Air temperature was highest in August and lowest in February (Figure 1A). The average annual temperatures during the study period in each year were 8.0, 8.4, and 8.9°C in 2017, 2018, and 2019, respectively, which were lower, similar, and higher than the average values for the last 10 years (8.4°C), respectively. Annual precipitation during the study period of each year was 1,227, 1,254, and 1,227 mm in 2017, 2018, and 2019, respectively, and the average annual precipitation for the past 10 years was 1,273 mm (Figure 1A).

Soil temperature tended to be similar to air temperature during the no-freeze period and ranged from 1.6 to 25.6°C, which was significantly lower in 2018 than in 2017 and 2019, although there was no significant difference among the treatments (Figure 1B).

WFPS ranged from 38 to 100%, tended to increase after heavy rainfall, and to decline when there was high temperature and no rainfall (Figure 1C). The WFPS was significantly lower in the no-fertilizer plot and the slurry plot in 2019, and there was no significant difference among the other treatment plots.

Soil  $NO_3^-$ -N content showed almost no peak in 2017, but several peaks after fertilization in 2018 and 2019 (Figure 1D). The highest mean  $NO_3^-$ -N was 82.6  $mg\ kg^{-1}$  in the slurry plot. Conversely, the lowest average  $NO_3^-$ -N was 75.3  $mg\ kg^{-1}$  in the non-fertilized plot.

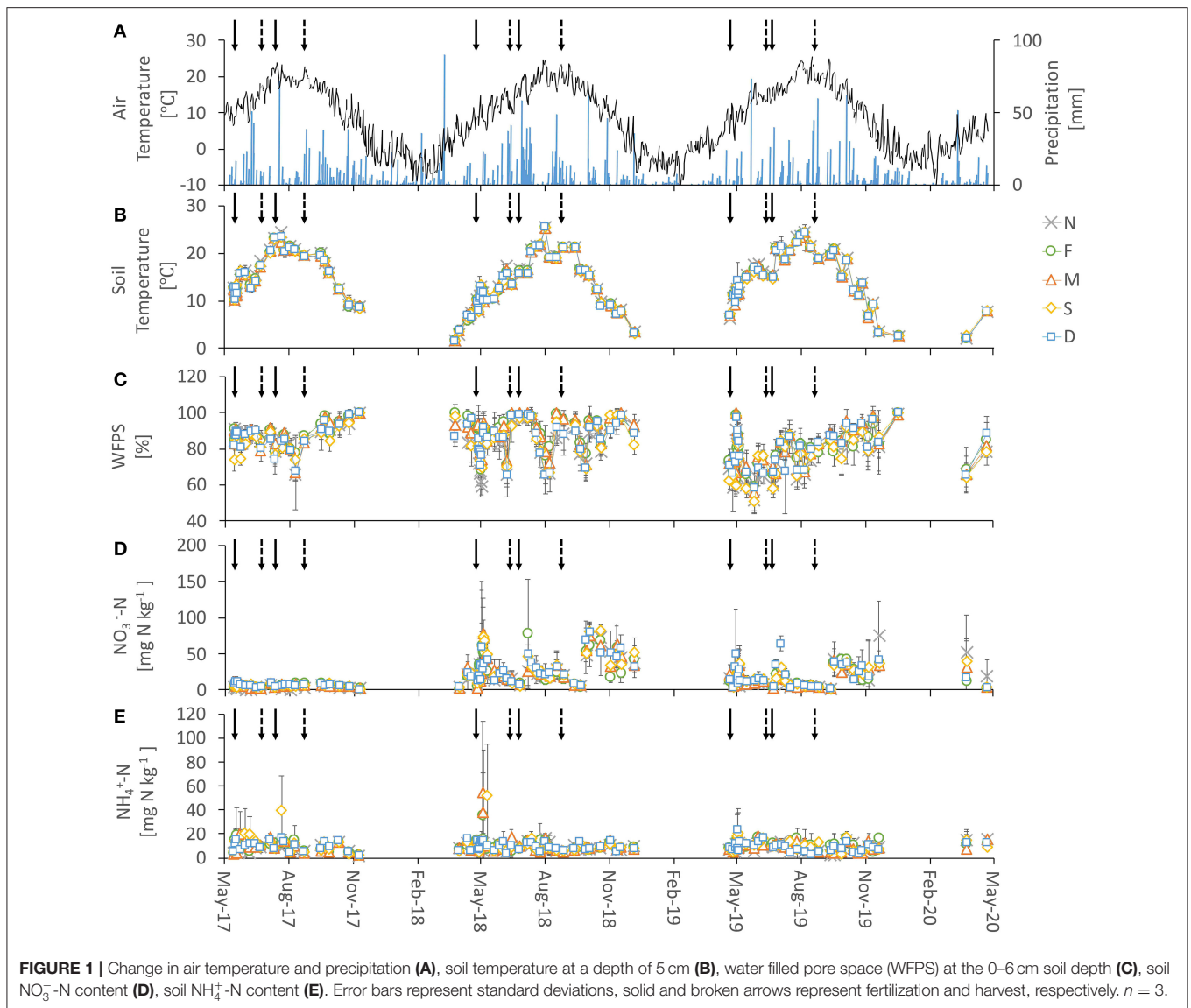
Soil  $NH_4^+$ -N content showed high peak in the slurry plot (39.7  $mg\ kg^{-1}$ ) after topdressing in 2017 (Figure 1E). In 2018, peaks were observed in the chemical fertilizer plot (36.1  $mg\ kg^{-1}$ ), manure plot (54.8  $mg\ kg^{-1}$ ), and slurry plot (52.3  $mg\ kg^{-1}$ ) immediately after the first fertilizer application. In 2019, no significant peak was observed in any treatment plot. The highest mean  $NH_4^+$ -N content was in the slurry plot at 10.8  $mg\ kg^{-1}$ . Conversely, the lowest mean of  $NH_4^+$ -N content was 9.0  $mg\ kg^{-1}$  in the non-fertilized plot.

### GHG Fluxes

The  $CO_2$  (RH) flux ranged from -49 (3 March, 2020) to 262  $mg\ C\ m^{-2}\ h^{-1}$  (22 August, 2017) and increased with increasing temperature (Figure 2A). Additionally, a decrease in  $CO_2$  (RH) flux was observed when WFPS was 100%.

The  $CH_4$  flux ranged from -401 to 357  $\mu g\ C\ m^{-2}\ h^{-1}$  and fluctuated highly (Figure 2B). The peaks of  $CH_4$  flux were observed after the application of organic fertilizers in 2017 and 2018, especially in the slurry plot in the winter of 2018. In 2019, there was a large daily fluctuation with high  $CH_4$  uptake by the soil.

The  $N_2O$  flux ranged from -115 to 839  $\mu g\ N\ m^{-2}\ h^{-1}$  just after the application of organic fertilizer (Figure 2C). A peak of  $N_2O$  flux occurred in the digestive fluid plot in 2017 (347.82  $\mu g\ N\ m^{-2}\ h^{-1}$ ). On the other hand, in 2018 and 2019, no peak in  $N_2O$



flux was observed just after the application of organic fertilizer. The peak of  $\text{N}_2\text{O}$  flux was smaller throughout the year in 2018 than in 2017 and 2019.

There was no significant correlation between  $\text{N}_2\text{O}$  flux and soil  $\text{NO}_3^-$ -N and  $\text{NH}_4^+$ -N contents (Figures 3A,B). However, the emission peaks of  $\text{N}_2\text{O}$  flux larger than  $191 \mu\text{g N m}^{-2} \text{h}^{-1}$  of top 5% were clearly observed and tended to increase when the soil  $\text{NO}_3^-$ -N content was  $2\text{--}12 \text{ mg N kg}^{-1}$  (except for one at the time of just after supplement fertilizer application for second crop in digestive fluid treatment in 2019). Concerning soil  $\text{NH}_4^+$ -N content, almost all  $\text{N}_2\text{O}$  fluxes including the high peaks were found in  $5\text{--}18 \text{ mg N kg}^{-1}$ .

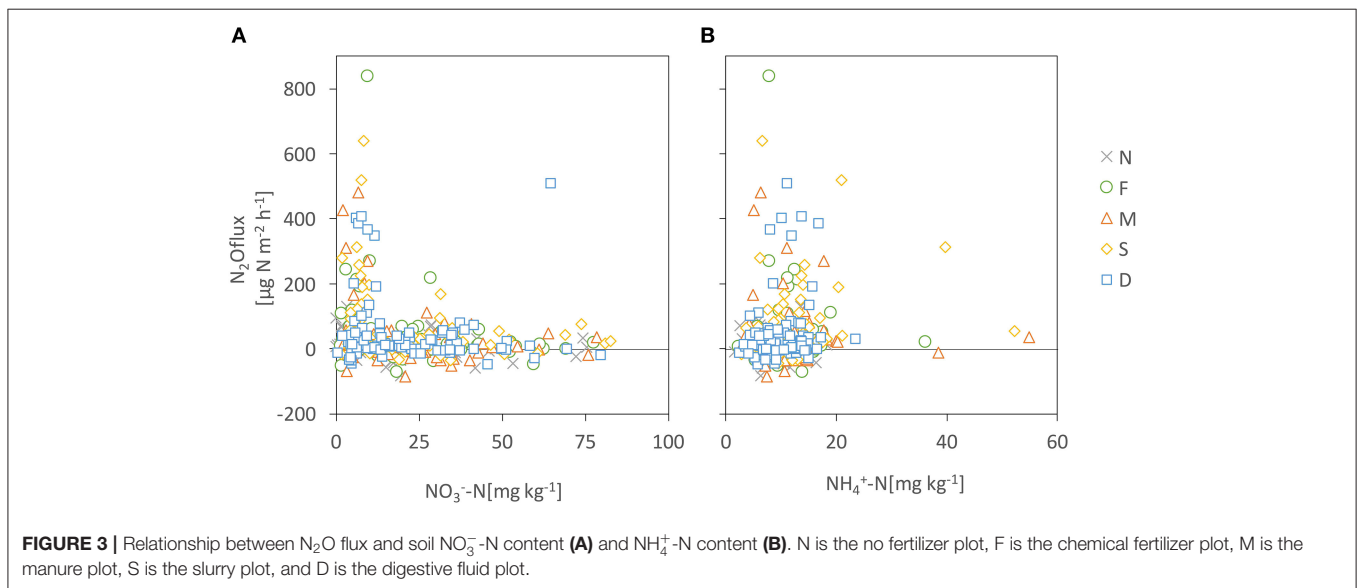
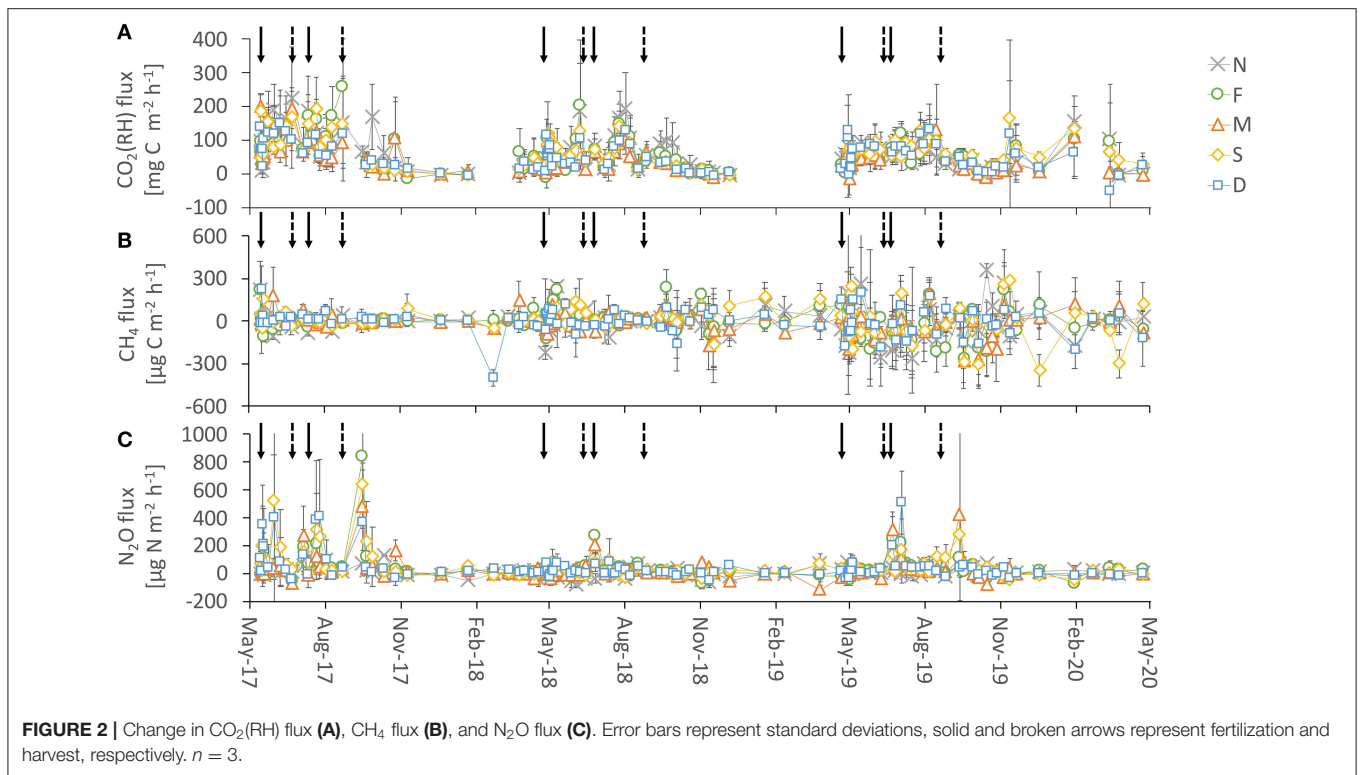
There was no difference in the relationship between gas flux and environmental factors caused by fertilization treatment (Figure 4). The  $\text{CO}_2$  flux increased with increasing soil temperature and decreasing WFPS (Figures 4A,B). There was no significant correlation between  $\text{CH}_4$  flux and soil temperature

and WFPS (Figures 4C,D). The emission peak of  $\text{N}_2\text{O}$  flux larger than  $191 \mu\text{g N m}^{-2} \text{h}^{-1}$  of top 5% was observed when the soil temperature was  $12\text{--}23^\circ\text{C}$  and the WFPS was 80–100%. However, at just after the harvest of first crop, all treatments showed high peaks in the range of 60–70% WFPS (Figures 4E,F).

## GHG Emissions

The result of two-way ANOVA shows that  $\text{CO}_2(\text{RH})$  emissions exhibited significant difference among years but no significant difference among the treatments (Table 3). It was maximum in the chemical fertilizer treatment in 2017, in no fertilizer treatment in 2018, and in the slurry treatment in 2019.

The result of ANOVA showed that  $\text{CH}_4$  emissions exhibited no significant differences among years and treatments (Table 3). However,  $\text{CH}_4$  uptake was observed with the no fertilizer or chemical fertilizer treatments in 2017, manure treatment in 2018, and all treatments in 2019 with lower precipitation.



The result of ANOVA showed that N<sub>2</sub>O emissions exhibited significant differences among years and treatments (Table 3). N<sub>2</sub>O emission was significantly higher in fertilizer treatments than in no fertilizer treatment, but there was no significant difference among the fertilizer treatments. However, the slurry treatment tended to be the highest N<sub>2</sub>O emission for all years.

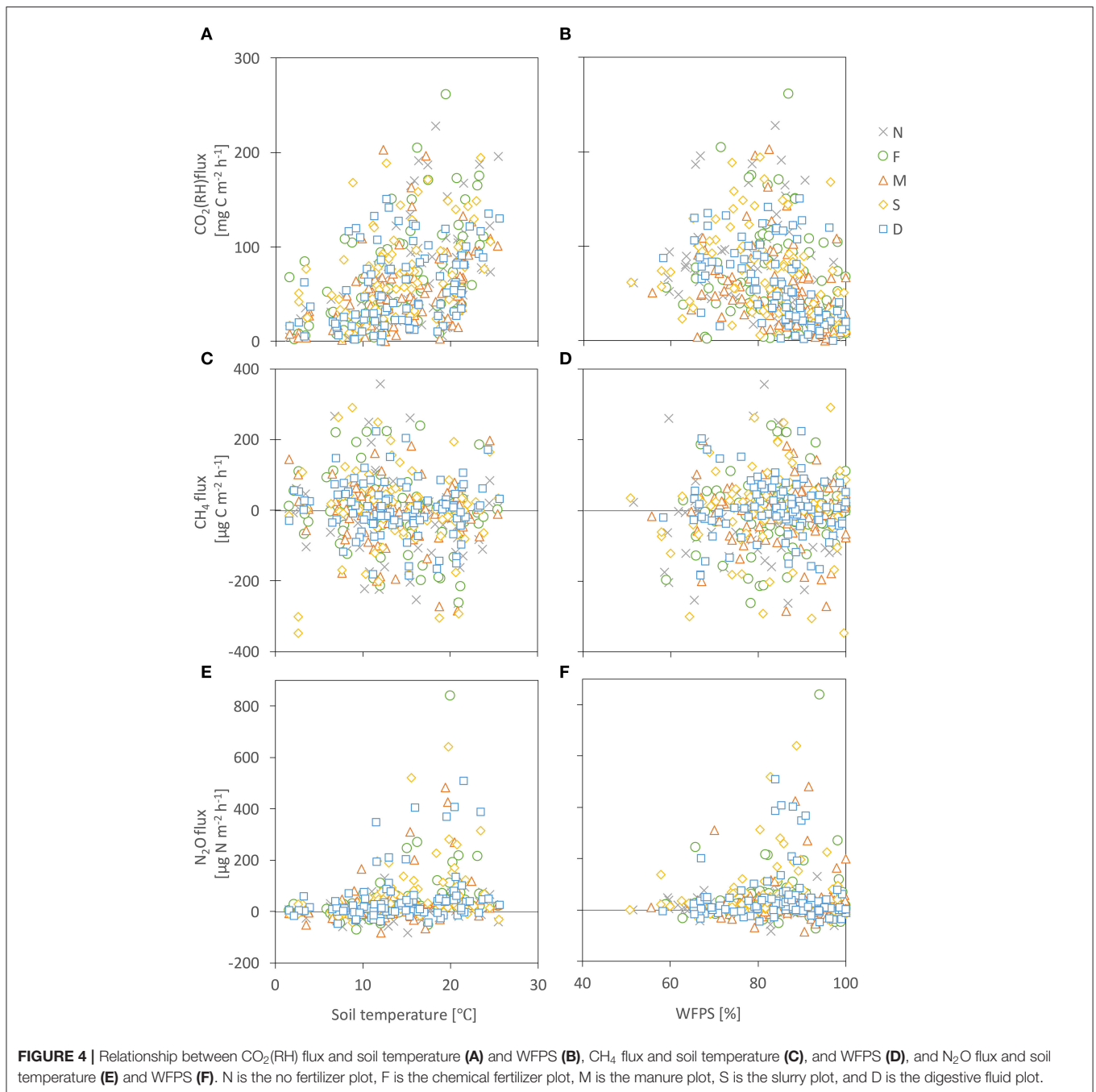
The result of ANOVA showed no significant difference in the 3-year total CO<sub>2</sub>(RH) among the treatments (Table 4). However, it tended to be the highest in the chemical fertilizer treatment

(13.79 Mg C ha<sup>-1</sup>), and among the organic fertilizer treatments, slurry > digestive fluid > manure.

There was no significant difference in the 3-year total CH<sub>4</sub> emission among the treatments (Table 4). However, 3-year total CH<sub>4</sub> emission tended to be highest in the slurry treatment (0.90 kg C ha<sup>-1</sup>) and lowest in the manure treatment (-1.19 kg C ha<sup>-1</sup>) among the organic fertilizer.

There was significant difference in 3-year total N<sub>2</sub>O emission among the treatments (Table 4). Three-year total N<sub>2</sub>O emission





was highest in the slurry treatments (10.8 kg N ha<sup>-1</sup>) and lowest in the manure treatment (6.21 kg N ha<sup>-1</sup>), although there was no significant difference among the fertilizer treatments.

### N<sub>2</sub>O Emission Factor

There was not a substantial variability on the N<sub>2</sub>O emission factor among years and treatments (Table 3). The 3-year average of the N<sub>2</sub>O emission factor was in the order chemical fertilizer > slurry > digestive fluid > manure (Table 4). However, among the organic fertilizer treatments, there was a significant negative

relationship between the C/N ratio and N<sub>2</sub>O emission factor (Figure 5).

### Grass Yield

The 3-year cumulative grass yield was not significantly different among fertilizer treatments and was significantly higher than that of the no fertilizer treatment (Figure 6). This was achieved despite of the reduction of chemical fertilizer for nitrogen by 10.0–29.4%, phosphorus by 56.3–73.3%, and potassium by 78.2–100% in 3 years as the concentrations of phosphorus

**TABLE 3** | Cumulative CO<sub>2</sub>(RH), CH<sub>4</sub>, and N<sub>2</sub>O emissions and the N<sub>2</sub>O emission factor (EF<sub>N2O</sub>) for each year.

		CO <sub>2</sub> (RH)		CH <sub>4</sub>		N <sub>2</sub> O		EF <sub>N2O</sub>		
		Mg C ha <sup>-1</sup>		kg C ha <sup>-1</sup>		kg N ha <sup>-1</sup>		%		
2017 (348 days)	N	4.91 ± 0.52	A	-0.16 ± 0.99	AB	1.10 ± 0.99	DE	-	-	-
	F	6.05 ± 2.44	A	-0.06 ± 1.12	AB	5.06 ± 1.12	AB	3.96 ± 1.57	A	
	M	4.08 ± 1.14	A	1.22 ± 1.06	AB	3.70 ± 1.06	ABCD	-4.09 ± 11.3	A	
	S	4.24 ± 0.70	A	1.23 ± 2.26	AB	6.29 ± 2.26	A	7.07 ± 13.8	A	
	D	3.61 ± 0.75	A	0.60 ± 1.21	AB	4.79 ± 1.21	ABC	1.02 ± 1.76	A	
2018 (365 days)	N	3.72 ± 1.73	A	1.10 ± 0.99	AB	0.13 ± 0.99	E	-	-	-
	F	3.18 ± 0.54	A	0.89 ± 0.35	AB	1.03 ± 0.35	DE	0.56 ± 0.51	A	
	M	2.30 ± 0.20	A	-1.36 ± 0.64	AB	-0.02 ± 0.64	E	-1.65 ± 0.61	A	
	S	2.99 ± 0.55	A	4.66 ± 0.87	A	1.82 ± 0.87	CDE	1.21 ± 0.57	A	
	D	2.30 ± 0.56	A	1.90 ± 0.57	AB	1.25 ± 0.57	DE	0.84 ± 0.56	A	
2019 (366 days)	N	4.40 ± 2.74	A	-2.03 ± 0.64	AB	1.27 ± 0.64	E	-	-	-
	F	4.56 ± 1.28	A	-2.78 ± 1.27	B	2.09 ± 1.27	BCDE	0.82 ± 1.67	A	
	M	3.25 ± 2.06	A	-1.05 ± 2.38	AB	2.53 ± 2.38	BCDE	1.22 ± 3.09	A	
	S	6.05 ± 1.10	A	-4.99 ± 0.34	B	2.70 ± 0.34	BCDE	0.57 ± 0.31	A	
	D	4.19 ± 1.47	A	-1.71 ± 0.94	AB	1.96 ± 0.94	CDE	0.16 ± 1.53	A	
ANOVA	<i>d.f.</i>	<i>F</i> -value	<i>p</i> -value	<i>F</i> -value	<i>p</i> -value	<i>F</i> -value	<i>p</i> -value	<i>F</i> -value	<i>p</i> -value	
year	2	6.85	0.004	12.3	<0.001	30.3	<0.001	0.49	0.62	
Treatment	4	1.93	0.13	0.31	0.87	6.66	0.001	1.60	0.22	
Treatment × year	8	0.78	0.62	1.97	0.09	1.70	0.09	1.01	0.44	

Values represent mean ± standard deviation. Values with the same letters are not significantly different ( $P < 0.05$ ). N is the no fertilizer plot, F is the chemical fertilizer plot, M is the manure plot, S is the slurry plot, and D is the digestive fluid plot. ANOVA is analysis of variance, and *d.f.* is degrees of freedom.  $n = 3$ .

**TABLE 4** | Cumulative CO<sub>2</sub>(RH), CH<sub>4</sub>, and N<sub>2</sub>O emissions and the N<sub>2</sub>O emission factor (EF<sub>N2O</sub>) for 3-year total.

		CO <sub>2</sub> (RH)		CH <sub>4</sub>		N <sub>2</sub> O		EF <sub>N2O</sub>		
		Mg C ha <sup>-1</sup>		kg C ha <sup>-1</sup>		kg N ha <sup>-1</sup>		%		
3 years (1,079 days)	N	13.0 ± 0.50	A	-1.09 ± 2.33	A	2.50 ± 1.67	B	-	-	-
	F	13.8 ± 3.27	A	-1.94 ± 4.48	A	8.17 ± 1.79	AB	1.58 ± 0.16	A	
	M	9.63 ± 1.18	A	-1.19 ± 4.00	A	6.21 ± 1.82	AB	-1.09 ± 1.40	A	
	S	13.3 ± 0.61	A	0.90 ± 3.32	A	10.8 ± 3.32	A	1.48 ± 1.08	A	
	D	10.1 ± 2.63	A	0.79 ± 2.09	A	8.00 ± 2.27	AB	0.58 ± 1.46	A	
ANOVA	<i>d.f.</i>	<i>F</i> -value	<i>p</i> -value	<i>F</i> -value	<i>p</i> -value	<i>F</i> -value	<i>p</i> -value	<i>F</i> -value	<i>p</i> -value	
Treatment	4	2.89	0.08	0.43	0.78	5.55	0.01	3.45	0.07	

Values represent mean ± standard deviation. Values with the same letters are not significantly different ( $P < 0.05$ ). N is the no fertilizer plot, F is the chemical fertilizer plot, M is the manure plot, S is the slurry plot, and D is the digestive fluid plot. ANOVA is analysis of variance, and *d.f.* is degrees of freedom.  $n = 3$ .

and potassium in organic fertilizers were high (Table 2). Thus, the fertilizer application design for each fertilizer area was appropriate.

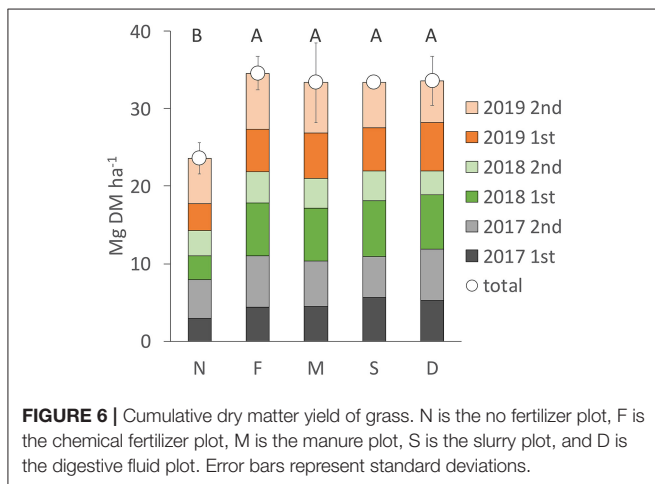
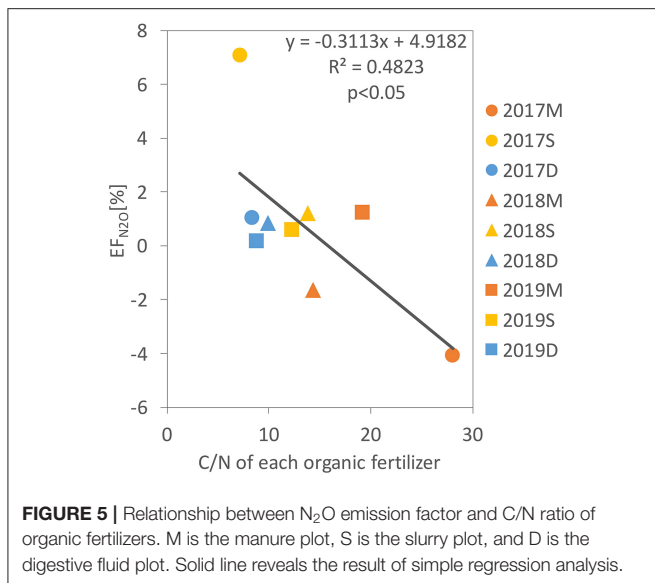
## Net Ecosystem Carbon Balance

NECB in all treatments was negative, that is, the ecosystem lost carbon (Table 5). NECB was significantly lower in no fertilizer and chemical fertilizer treatments than the manure treatment. Although there was no significant difference among the fertilizer treatments, NECB tended to be larger in organic fertilizer treatments than in the chemical fertilizer treatment. ANPP was significantly lower in the no fertilizer treatment than the fertilizer treatment, and there was no significant difference among the

chemical and organic fertilizer treatments. On the other hand, BNPP was negative, although there was no significant difference. Although ANPP + BNPP was positive, NEP was negative because of a larger RH than ANPP+BNPP. Therefore, a larger NECB in the organic fertilizer plot was caused by the contribution of C<sub>input</sub> with organic fertilizer application, that is, carbon input by organic fertilizer enhances soil carbon sequestration. NECB in organic fertilizer plots tended to be in the order manure > digested fluid > slurry.

## GHG Balance

The contribution of CH<sub>4</sub> emission to the GWP for 3 years was very small, which was < 0.1%. On the other hand, the



contribution of N<sub>2</sub>O emissions to the GWP was larger than 5% (Table 6). The 3-year GWP was significantly smaller in the manure plot ( $43.1 \pm 2.8 \text{ Mg CO}_2 \text{ eq ha}^{-1} \text{ yr}^{-1}$ ) than that in the no fertilizer ( $65.4 \pm 3.7 \text{ Mg CO}_2 \text{ eq ha}^{-1} \text{ yr}^{-1}$ ) and chemical fertilizer plots ( $67.4 \pm 13.6 \text{ Mg CO}_2 \text{ eq ha}^{-1} \text{ yr}^{-1}$ ), but there was no significant difference among the organic fertilizer treatments. The cumulative values for organic fertilizer treatments were slurry > digestive fluid > manure, and the GWP was 16.5, 27.0, and 36.2% smaller than those in the chemical fertilizer treatment, respectively.

## DISCUSSION

### Effect of Different Fertilizers on the Net Ecosystem Carbon Balance

In this study, the no-fertilizer treatment exhibited a significantly smaller NECB than did the fertilizer treatments. All organic fertilizers from manure, slurry, and digestive fluid tended to

have a larger NECB, although there was no significant difference compared to that of the chemical fertilizer treatment. This indicates that organic fertilizers have larger carbon storage than do chemical fertilizers. Previous studies conducted on grasslands also showed higher NECB in manure treatment than in chemical fertilizer treatment (Matsuura et al., 2014; Shimizu et al., 2015; Mukumbuta et al., 2017a). In this study, NECB was negative for 3 years and all treatment plots became carbon sources. On the other hand, in previous studies by Matsuura et al. (2014) and Shimizu et al. (2015), NECB was positive in the manure treatment. The manure application rate in the previous study was  $2.1\text{--}7.7 \text{ Mg C ha}^{-1} \text{ year}^{-1}$ , whereas that in this study was  $0.15\text{--}1.9 \text{ Mg C ha}^{-1} \text{ year}^{-1}$ . This depended on the raw material of the manure. The manure used in the previous study was a bark manure with a lower C/N ratio and lower potassium content.

Therefore, to increase soil organic carbon using organic fertilizer, the quality of organic fertilizer, especially the ratio of carbon to nutrients, should be taken into consideration.

### Effect of Different Fertilizer on the CH<sub>4</sub> and N<sub>2</sub>O Emissions

In this study, the slurry treatment showed the highest CH<sub>4</sub> emission in 3-year total, although CH<sub>4</sub> emissions were not significantly different (Table 4). However, CH<sub>4</sub> emission showed a large variation among the years. Additionally, a previous study showed that direct CH<sub>4</sub> emission from organic fertilizer often occurred just after the application of organic fertilizer to the soil (Mori and Hojito, 2015). It was also expected that the anaerobic conditions produced by the liquid fertilizer in the slurry and digestive fluid would promote CH<sub>4</sub> emission immediately after fertilization (da Silva Cardoso et al., 2020). However, in this study, no peak of CH<sub>4</sub> flux was observed immediately after fertilization. This was probably because the temperature was relatively low immediately after fertilization and no microbial degradation occurred (Ryals and Silver, 2013).

The relationship between N<sub>2</sub>O flux and environmental factors was not significantly different among fertilizer treatments. That is, N<sub>2</sub>O flux peaks were observed when the soil NO<sub>3</sub><sup>-</sup>-N content was 2 to 12 mg N kg<sup>-1</sup> (except for one plot of just after supplement fertilizer application for second crop in digestive fluid treatment in 2019) (Figure 3A), and the WFPS was 80–100% (Figure 4F). These suggest that the N<sub>2</sub>O emission occurred through denitrification in all fertilizer treatments (Takakai et al., 2006). However, at just after the first crop harvest in 2019, the large N<sub>2</sub>O peaks in the lower WFPS than 80%. This was probably because the stronger effect of the disturbance by harvest on N<sub>2</sub>O emission than the effect of WFPS, which was shown by Li et al. (2015).

The application of organic fertilizers increases nitrogen mineralization in the soil and, from a physical point of view, increased the water retention of the soil, which increase N<sub>2</sub>O emissions (Ryals and Silver, 2013). In particular, the application of slurry tends to promote denitrification to increase N<sub>2</sub>O production (Rochette et al., 2004). On the other hand, it has also been reported that there was no significant difference in annual N<sub>2</sub>O emission in a grassland in Hokkaido between chemical

**TABLE 5** | Net ecosystem carbon balance.

	C <sub>input</sub>		ANPP		BNPP		C <sub>output</sub>		RH		NEP		NECB	
	<b>Mg C ha<sup>-1</sup></b>													
N	0	8.65 ± 0.82	B	-4.17 ± 0.66	A	9.00 ± 0.92	B	13.0 ± 0.50	A	-8.55 ± 1.13	B	-17.6 ± 1.13	B	
F	0	12.9 ± 0.79	A	-3.31 ± 1.87	A	13.3 ± 0.92	A	13.8 ± 3.27	A	-4.17 ± 3.73	B	-17.5 ± 3.73	B	
M	2.43	12.5 ± 1.93	A	-3.51 ± 0.42	A	12.9 ± 2.01	A	9.63 ± 1.18	A	-0.63 ± 0.98	A	-11.1 ± 0.98	A	
S	3.12	12.6 ± 0.27	A	-3.60 ± 0.33	A	12.9 ± 0.35	A	13.3 ± 0.61	A	-4.30 ± 0.81	AB	-14.1 ± 0.81	AB	
D	2.30	13.0 ± 1.32	A	-4.46 ± 0.45	A	13.2 ± 1.25	A	10.1 ± 2.63	A	-1.57 ± 2.68	AB	-12.5 ± 2.68	AB	
ANOVA	<i>d.f.</i>	<i>F</i> -value	<i>p</i> -value	<i>F</i> -value	<i>p</i> -value	<i>F</i> -value	<i>p</i> -value	<i>F</i> -value	<i>p</i> -value	<i>F</i> -value	<i>p</i> -value	<i>F</i> -value	<i>p</i> -value	
Treatment	4	7.53	0.005	0.79	0.56	6.80	0.007	2.89	0.08	5.66	0.01	5.34	0.01	

Values represent mean ± standard deviation. Values with the same letters are not significantly different ( $P < 0.05$ ). N is the no fertilizer plot, F is the chemical fertilizer plot, M is the manure plot, S is the slurry plot, and D is the digestive fluid plot. C<sub>input</sub> is the carbon input from the application of organic fertilizer, ANPP is the aboveground net primary production, BNPP is the belowground net primary production, C<sub>output</sub> is the carbon output from the harvest, RH is organic matter decomposition, NEP is net ecosystem production, NECB is the net ecosystem carbon balance. ANOVA is analysis of variance, and *d.f.* is degrees of freedom.  $n = 3$ .

**TABLE 6** | GHG balance (CO<sub>2</sub> equivalent).

	CO <sub>2</sub>		CH <sub>4</sub>		N <sub>2</sub> O		GWP		
	<b>Mg CO<sub>2</sub> eq ha<sup>-1</sup> year<sup>-1</sup></b>								
N		64.4 ± 4.1	A	-0.04 ± 0.1	A	1.04 ± 0.7	B	65.4 ± 3.7	A
F		64.1 ± 13.7	A	-0.07 ± 0.2	A	3.40 ± 0.7	AB	67.4 ± 13.6	A
M		40.5 ± 3.6	B	-0.04 ± 0.1	A	2.59 ± 0.8	AB	43.1 ± 2.8	B
S		51.8 ± 3.0	AB	0.03 ± 0.1	A	4.50 ± 1.4	A	56.3 ± 3.6	AB
D		45.9 ± 9.8	AB	0.03 ± 0.1	A	3.33 ± 0.9	AB	49.2 ± 10.1	AB
ANOVA	<i>d.f.</i>	<i>F</i> -value	<i>p</i> -value	<i>F</i> -value	<i>p</i> -value	<i>F</i> -value	<i>p</i> -value	<i>F</i> -value	<i>p</i> -value
Treatment	4	5.34	0.01	0.43	0.78	5.55	0.01	5.01	0.02

Values represent mean ± standard deviation. Values with the same letters are not significantly different ( $P < 0.05$ ). N is the no fertilizer plot, F is the chemical fertilizer plot, M is the manure plot, S is the slurry plot, and D is the digestive fluid plot. GWP<sub>CO<sub>2</sub></sub> is the CO<sub>2</sub> equivalent NECB (the net ecosystem carbon balance), GWP<sub>CH<sub>4</sub></sub> is the CO<sub>2</sub> equivalent CH<sub>4</sub> emission, GWP<sub>N<sub>2</sub>O</sub> is the CO<sub>2</sub> equivalent N<sub>2</sub>O emission, GWP is the CO<sub>2</sub> equivalent GHG balance. ANOVA is analysis of variance, and *d.f.* is degrees of freedom.  $n = 3$ .

fertilizer treatment (0.6 kg N ha<sup>-1</sup> yr<sup>-1</sup>) and digestive fluid treatment (0.7 kg ha<sup>-1</sup> yr<sup>-1</sup>) (Sawamoto et al., 2010). In this study, N<sub>2</sub>O emissions from organic fertilizer treatments were higher than that of no fertilizer treatment (Tables 3, 4). The slurry treatment exhibited the highest N<sub>2</sub>O emission, and the manure treatment had the lowest N<sub>2</sub>O emissions among the fertilizer treatments. However, there was no significant difference among the organic fertilizer treatments and between organic and chemical fertilizer treatments. Similarly, in a previous study conducted in the same Andosols, no significant difference was found between the manure treatment and the chemical fertilizer treatment, although it tended to be higher in the manure treatment (Shimizu et al., 2010; Mukumbuta et al., 2017a). These results suggest that the high soil organic matter content of the Andosols reduce the effect of organic matter application.

## Importance of Fertilization Design on Crop Yield

The fact that there was no significant difference in the grass yield between the chemical fertilizer treatment and the organic fertilizer treatments (Figure 6) was that the fertilizer application design was correctly performed using a combination of chemical and organic fertilizers without significantly increasing nitrogen loss of the organic fertilizer compared to that of the chemical

fertilizer (Sawamoto et al., 2010; Mori and Hojito, 2015). It has been reported that field surplus nitrogen, which is calculated as the difference between nitrogen input by fertilizer application and nitrogen output by plant uptake is a good indicator of N<sub>2</sub>O emissions (Shimizu et al., 2010). Field surplus nitrogen did not correlate with plant nitrogen uptake but correlated with N<sub>2</sub>O emission and NO<sub>3</sub><sup>-</sup>-N leaching (Nagatake et al., 2018).

## Comparison of the N<sub>2</sub>O Emission Factors of Organic Fertilizer

The N<sub>2</sub>O emission factors in this study were 0.6–4.0%, -4.1–1.2%, 0.6–7.1%, and 0.2–1.0% for chemical fertilizer, manure, slurry, and digestive fluid, respectively. There was no significant difference among years or fertilizer treatments (Table 3), although chemical fertilizer treatment tended to have higher emission factor than organic fertilizer treatments. However, on average, there was a significant negative correlation with the C/N ratio of organic fertilizers (Figure 5). This result was consistent with the results on farmland where N<sub>2</sub>O emissions were measured using organic matter containing manure and plant residues (Akiyama and Tsuruta, 2003; Huang et al., 2004; Toma and Hatano, 2007; He et al., 2019). The C/N ratio of the microbes in soil is 5–10; that is, the synthesis of the cell requires nitrogen in an amount of 1/5–1/10 that of the carbon,

and some of the nitrogen mineralized through organic matter decomposition will be taken up by the microbes. The lower the C/N ratio of organic matter applied to the soil, the greater the amount of mineralized nitrogen released into the soil because there is more mineralized nitrogen than the microbes can uptake. On the contrary, the higher the C/N ratio of organic matter into the soil, the lower the release of mineral nitrogen into the soil because more mineralized nitrogen is taken up by the cells (Ruser et al., 2001). Soil mineral nitrogen ( $\text{NO}_3^-$ -N and  $\text{NH}_4^+$ -N) and easily decomposable organic carbon are substrates for nitrification and denitrification that cause soil  $\text{N}_2\text{O}$  emissions. In this study, soil  $\text{NO}_3^-$ -N and  $\text{NH}_4^+$ -N content tended to be larger in slurry with a low C/N ratio than in manure with a high C/N ratio (Figures 1E,F). Therefore, it is considered that the slurry releases more mineralized nitrogen into the soil, which enhances nitrification and denitrification, and promotes  $\text{N}_2\text{O}$  emission.

In this study,  $\text{N}_2\text{O}$  emission factor is obtained by subtracting  $\text{N}_2\text{O}$  emissions from chemical fertilizers and soil-derived emissions (Equations 7 and 8). Negative  $\text{N}_2\text{O}$  emission factor of organic fertilizer shown in Table 3 suggests that the application of organic fertilizer denitrifies  $\text{N}_2\text{O}$  derived from chemical fertilizer. Several reports show the lower  $\text{N}_2\text{O}$  emission factor of organic matter than that of chemical fertilizer only (Toma and Hatano, 2007; Toma et al., 2007; Jin et al., 2010; Mori and Hojito, 2012; Shimizu et al., 2013; De Rosa et al., 2018).

## Comparison of Global Warming Potential Among Fertilizers

The cumulative GWP for 1,079 days tended to be smaller for the organic fertilizer treatments than in the no fertilizer and chemical fertilizer treatments, especially the manure treatment, which had a significantly smaller GWP (Table 6). As suggested by Mukumbuta and Hatano (2020), estimates of the NECB showed that organic fertilizers tended to have a higher soil carbon sequestration effect than chemical fertilizers (Table 5), which is thought to reduce the GWP of organic fertilizers. Comparing organic fertilizer treatments, the cumulative GWP was the largest in the slurry, which showed the highest  $\text{N}_2\text{O}$  emission factor because of the lowest C/N ratio (Figure 5). Although manure application requires relatively higher chemical fertilizer nitrogen application rate among the organic fertilizer treatments (Table 2), manure application increased soil carbon sequestration and reduced  $\text{N}_2\text{O}$  emissions, resulted in the highest reduction of GWP. Digestive fluid application reduced chemical fertilizer nitrogen application most, and reduced  $\text{N}_2\text{O}$  emission next of manure.

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## CONCLUSION

The effects of application of manure, slurry, and digestive fluid on GHG emissions from a grassland on Andosol in a cold temperate climate were evaluated under fertilization management in accordance with regional recommendations. In the plots where organic fertilizer was applied, the amount of chemical fertilizer input could be reduced while maintaining yield. The relationships between  $\text{CO}_2$ ,  $\text{CH}_4$ , and  $\text{N}_2\text{O}$  emissions and the soil environmental factors (soil  $\text{NO}_3^-$ -N and  $\text{NH}_4^+$ -N contents, temperature, and WFPS) were not influenced by the type of fertilizer. The  $\text{N}_2\text{O}$  emission factor was highest in the slurry treatment and lowest in the manure treatment, showing a negative correlation with the C/N ratio of organic fertilizers. Additionally, the application of these organic fertilizers has been shown to improve ecosystem carbon balance and reduce the GHG balance. When the GWP for each fertilizer was evaluated based on the results of this 3-year study, it was suggested that manure is the best way to increase soil carbon and to decrease GHG emissions, followed by digestive fluid.

## DATA AVAILABILITY STATEMENT

The original contributions presented in the study are included in the article, further inquiries can be directed to the corresponding author.

## AUTHOR CONTRIBUTIONS

RH planned this study, compiled all the data, and edited the manuscript. RK, CS, and KY measured greenhouse gas emissions and environmental factors and wrote the results and discussion. AN designed the experimental field, measurement procedures, and sampling schedule. YY and JD measured greenhouse gases and soil environmental factors. NY and KT produced slurries and methane fermentation digestive fluid at the Sapporo Farm. MK produced manure used at the Shizunai livestock farm and managed the field work. All authors contributed to the article and approved the submitted version.

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**Conflict of Interest:** The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

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