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# Characterization of the evolution of thermal maturity and pore structure of continental organic-rich shales

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To clarify the evolution of thermal maturity and pore structure in continental organic-rich shales, calcareous shales of the Liaohe Basin (China) were pyrolyzed, and examined using Raman spectroscopy, Fourier transform infrared spectroscopy (FTIR), X-ray diffraction (XRD), X-ray photoelectron spectroscopy (XPS), scanning electron microscopy (SEM), nitrogen sorption isotherms, and nuclear magnetic resonance (NMR) spectroscopy. The increase in Raman G-D band separation and decrease in Raman  $I_D/I_G$  ratio with increasing thermal maturity indicate that these parameters provide superior thermal maturity indicators. This is also confirmed by the good linear correlation of G-D band shifts and  $I_D/I_G$  with vitrinite reflectance (VR) and maximum temperature ( $T_{max}$ ), respectively. The relative detection accuracy (DA), sampling requirements (SR), sample preparation (SP), detection time (DT), and equipment requirement (ER) of VR, G-D band shifts,  $I_D/I_G$ ,  $T_{max}$ , FTIR, and XPS indicate that Raman analysis is a simple, quick, and robust method to evaluate thermal maturity. The moderate SR, complex SP, and long DT suggest that VR and  $T_{max}$  are less widely applicable for characterizing thermal maturity. The FTIR and XPS techniques provide semi-quantitative maturity indicators with poor DA and high ER. Pores observed within organic matter and minerals under SEM indicate that an increase in thermal maturity not only affects the development of organic pores but is also beneficial for the formation of mineral dissolution pores, such as those formed during the transformation of  $Na_{0.95}Ca_{0.16}Al_{1.16}Si_{2.84}O_8$  to  $Na_{0.84}Ca_{0.02}Al_{1.02}Si_{2.98}O_8$ , a process confirmed by XRD. The BET and NMR data also indicate that the development of pore structure is closely related to the evolution of thermal maturity in calcareous shale. During the initial stage, primary pores are filled by bitumen generated from kerogen; this leads to a decrease in transition pores, mesopores, and shale porosity, and reduced pore connectivity. Then, secondary nanoscale pores, transition pores, and mesopores increase with increasing thermal maturity. The peak in secondary porosity is consistent with the liquid hydrocarbon production rate peak, a process that increases shale porosity and leads to improved pore connectivity. The dissolution of minerals induced by organic acids may also contribute to this secondary porosity. With a further increase in thermal maturity, secondary porosity at the microscale is further developed, while transition pores and mesopores collapse, resulting in reduced pore connectivity. The poor pore connectivity that occurs at both low and high VR values may be more conducive to the preservation of shale oil and gas. This study is significant for research into

the evolution of thermal maturity and pore structure in continental organic-rich shales.

#### KEYWORDS

continental organic-rich shales, evaluation, maturity, pore structures, SEM, NMR

## 1 Introduction

Shale gas is a type of unconventional petroleum, which is generated from, and hosted in, organic-rich shales. There are three states in which shale gas exists: 1) free state, distributed in micro-cracks and macropores; 2) adsorbed state, adhered on the surface of organic matter and clay mineral micropores; and 3) dissolved state, dispersed in water and oil. Among these, free gas and adsorbed gas are the main components of shale gas (Curtis, 2002; Ji et al., 2017; Zhu et al., 2021). Therefore, investigating the pore composition and development characteristics of shale and associated influencing factors is critical for shale oil and gas exploration and development.

According to the relationship between shale pores and mineral particles, shale pores can be divided into mineral pores, organic pores, and micro-fracture pores. In organic-rich shales, organic pores are the most important pore component because they can comprise up to 70% of the total pore volume of the shale, whilst also having good connectivity (Löhr et al., 2015; Ji et al., 2017; Borjigin et al., 2021; Zhang et al., 2022; Zheng et al., 2022). Investigating pore composition and evolution as organic-rich shales mature has important theoretical and practical significance for predicting shale gas content and evaluating shale gas resources.

Highly productive shale gas has been shown to be strongly associated with high-maturity shales (Wu et al., 2019; Chen et al., 2020; Song et al., 2021; Zheng et al., 2021; Cao et al., 2022), indicating that highly productive shale gas has inevitably experienced a high evolutionary stage and deep burial. In this long and complicated evolution process, the hydrocarbon generation of organic matter and the transformation of clay minerals in shale may have mutual influence. Such as, the organic acids formed in the process of hydrocarbon generation can accelerate the dissolution of minerals and affect the transformation of clay minerals; clay minerals also have catalytic effect on hydrocarbon generation and liquid hydrocarbon cracking. Previous investigators mainly focus on the change of organic pore structure with maturity, and few studies have investigated the effect of organic-inorganic interaction during shale thermal evolution on the formation and evolution of shale pores. At the same time, there are few experimental studies to discuss the influences of organic-inorganic interaction on mineral and organic pores with thermal maturity.

Generally, vitrinite reflectance (VR) and maximum temperature ( $T_{max}$ ) values provide good thermal maturity indicators for source rocks and organic matter (Hunt, 1996; Evenick, 2021; Al-Hajeri et al., 2021; Lohr et al., 2015; Waliczek et al., 2021). However, for some samples, VR and  $T_{max}$  cannot accurately reflect maturity due to a lack of vitrinite or low  $S_2$  values (Lewan and Kotarba, 2014; Katz and Lin, 2021), this is the case for highly mature marine shales in the Sichuan Basin, China, and for limestone source rocks with low

organic matter contents. Therefore, additional maturity indicators, which are easy to detect and provide high-performance, need to be developed to cover organic matter from a wide range of source rocks, deposited under different sedimentary settings, or at different stages of thermal evolution.

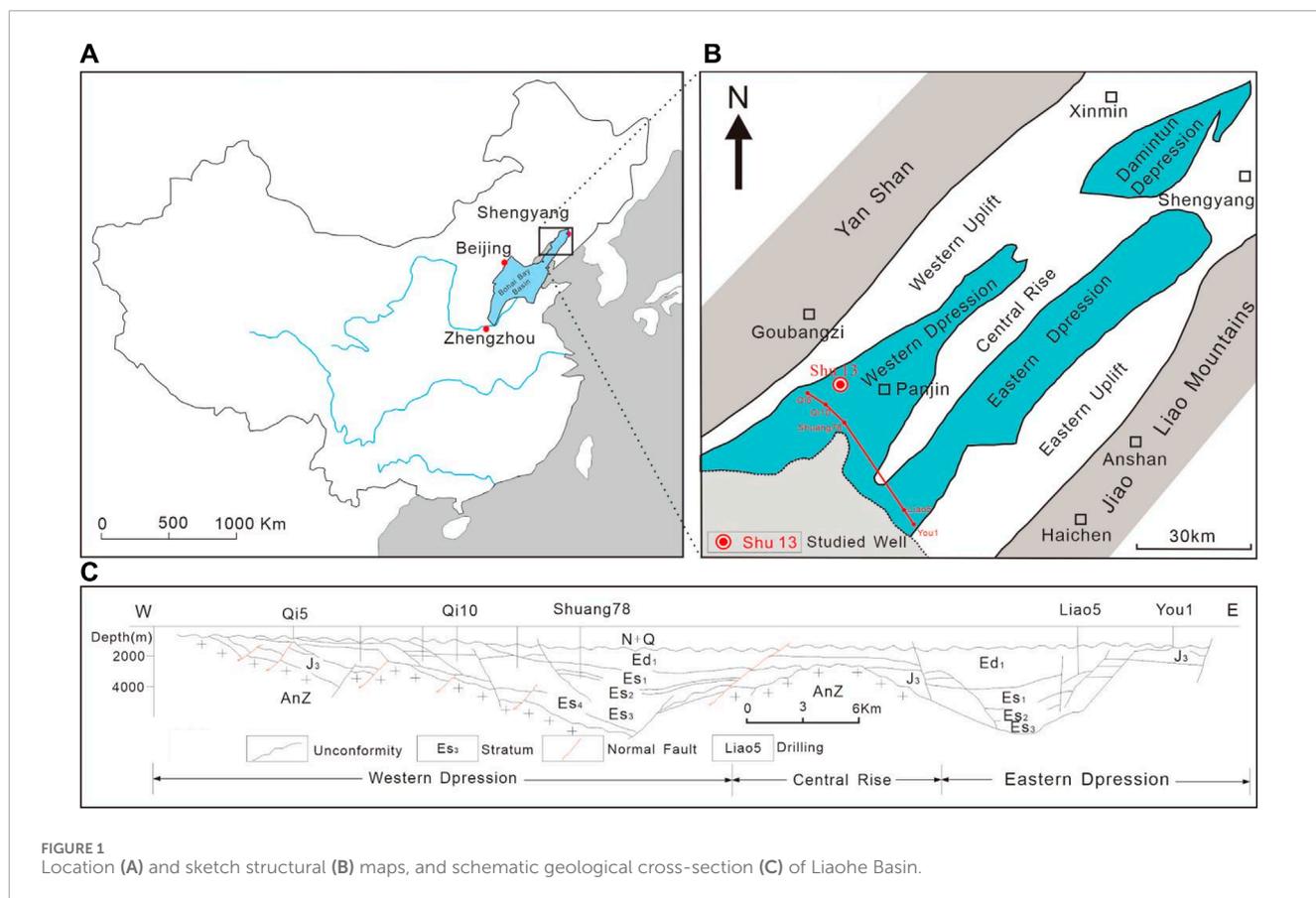
Owing to the success of marine shale gas exploration in both Permian shales and the Devonian-Carboniferous Bakken shale of North America, many investigators have focused on the pore structure and development of marine shales (Hao et al., 2013; Hao and Zou, 2013; Yang et al., 2017; Wu et al., 2018; Yang et al., 2018; Shi et al., 2021; Xu et al., 2021), while there have been relatively few investigations of the pore evolution of continental organic-rich shales. Continental shale gas has great potential, and accounts for one-third of the total shale gas resources of China (Fan et al., 1980; Fu et al., 1988; Grice et al., 1998). Therefore, understanding the evolution of pore textures in continental organic-rich shales can contribute meaningfully to global shale gas exploration and development, especially in China.

In this study, we investigated the evolution of the thermal maturity and mineral/organic pore structure of continental organic-rich shales under semi-closed hydrous conditions to clarify the evolution of thermal maturity, and mineral/organic pores during shale gas generation. Calcareous shales from the Liaohe Basin (China) were pyrolyzed, and examined using laser Raman spectroscopy, Fourier transform infrared spectroscopy (FTIR), X-ray diffraction (XRD), X-ray photoelectron spectroscopy (XPS), scanning electron microscopy (SEM), nitrogen sorption isotherms (BET), and nuclear magnetic resonance (NMR) spectroscopy. Raman, FTIR, XRD, and XPS were used to detect changes in organic functional groups and mineral composition; SEM, BET, and NMR were used to analyze the evolution of pore texture and formation. This study has substantial implications for understanding the evolution of thermal maturity and pore structure in continental organic-rich shales, and for global shale gas exploration and development.

## 2 Samples and experiments

### 2.1 Samples

Calcareous shales were selected from the Shu 13 drill core from the Liaohe Basin (Shahejie Formation), as shown in Figure 1. Table 1 presents the geochemical parameters of these shales; details of the geochemical analysis methods were presented in a previous study (Wu et al., 2018). The geochemical data indicate that the kerogen in the calcareous shales is type-I kerogen and is immature ( $VR = 0.40\%R_o$ ;  $T_{max} = 434^\circ C$ ). Table 1 also shows that the hydrogen index (HI) of the calcareous shales is 792, which further suggests that they are oil-prone source rocks (Table 1; Wu et al., 2018).



## 2.2 Hydrous pyrolysis

A WYMN-3 HTHP instrument was used to simulate the thermal evolution process of continental organic-rich shales under geological conditions. The maximum simulation temperature of the WYMN-3 HTHP is 550°C, while the maximum values of water pressure and lithostatic pressure are 2,800 bar and 1,200 bar, respectively. A schematic diagram of the main structure of the instrument can be found in Zhang et al. (2020) and Wu et al. (2021). Hydrous pyrolysis was conducted at 350–520°C, with 400–800 bar water pressure, and 1,000–2000 bar lithostatic pressure. To simulate the entire evolution of the thermal maturity and pore structure of continental organic-rich shales, immature calcareous shale was used, and the pyrolysis temperatures were higher than classical hydrous pyrolysis (e.g., Hill et al., 2003; Xie et al., 2016). Detailed conditions of the pyrolysis experiments can be found in Wu et al. (2018).

## 2.3 Scanning electron microscopy (SEM)

The SEM analyses were performed at the South China University of Technology, using a Merlin Compact (Zeiss, Germany) instrument, equipped with an X-ray energy spectrometer (Oxford, England). The resolution of this instrument can reach 1.0 nm at 15 kV accelerating voltage and 2 mm working distance,

in high vacuum mode. The accelerating voltage used herein was 20 kV.

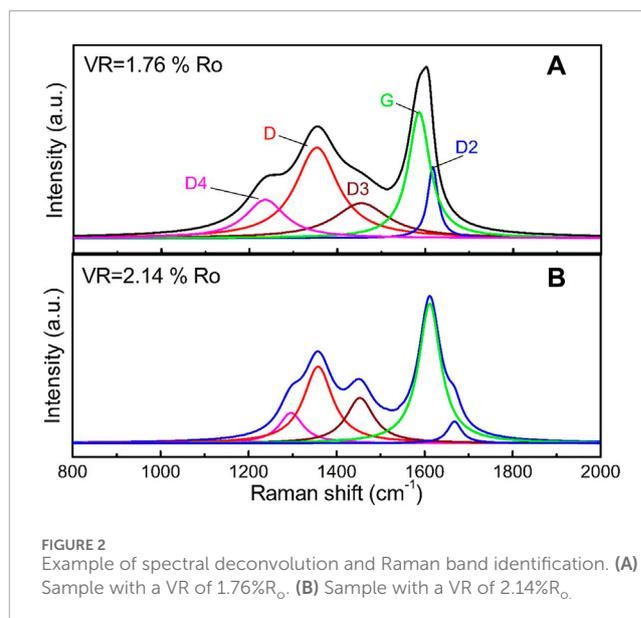
## 2.4 Raman spectroscopy

The source rocks, before and after pyrolysis, were examined by Raman spectroscopy using an Alpha 300M Raman spectrometer (WITec, Germany). The wavelengths of the laser source were 532 and 732 nm, and the power of the laser source was 5 mW.

Raman spectroscopy can record the entire spectrum of each sample. Generally, a Raman spectrum comprises a peak position, peak intensity, and peak area. To further resolve the sample information contained within a Raman spectrum, curve-fitting is usually performed. In this investigation, Gaussian-Lorentzian fitting was applied to the Raman spectra of all examined source rocks, before and after pyrolysis; at least 150 iterations were used until the best fit was achieved. Five peaks or bands, including D, G, D<sub>2</sub>, D<sub>3</sub>, and D<sub>4</sub>, can be fitted for organic matter evaluation in shale, as shown in Figure 2 (Khatibi et al., 2018; Khatibi et al., 2019; Birdwell et al., 2021; Craddock and Sauerer, 2022); all of these were fitted herein. However, only the D and G bands, which were most common and had the highest intensity, were considered further in this study (Table 2); the minor bands (D<sub>2</sub>, D<sub>3</sub>, and D<sub>4</sub>) were often unclear in both appearance and significance (cf. Beyssac et al., 2003).

TABLE 1 Geochemical parameters of the organic-rich shale study samples (after Wu et al., 2018).

Well	Stratum	Depth (m)	Lithology	TOC (%)	HI (mg/g)	OI (mg/g)	S <sub>1</sub> (mg/g)	S <sub>2</sub> (mg/g)	T <sub>max</sub> (°C)	VR (%R <sub>o</sub> )	Type of Kerogen
Shu13	Es3	1,387-1,390	Calcareous Shale	8.73	792	23	0.42	69.2	434	0.40	I



## 2.5 Infrared spectroscopy (IR)

The infrared spectra of source rocks, before and after pyrolysis, were detected using a Perkin Elmer Spectrum 100. Each spectrum was the result of 32 scans, and the resolution was 4 cm<sup>-1</sup> in the frequency range of 4,000–650 cm<sup>-1</sup>.

## 2.6 X-ray diffraction (XRD)

X-ray diffraction patterns of the source rocks, before and after pyrolysis, were obtained using a Rigaku D/Max 2550VB/PC diffractometer equipped with Cu K $\alpha$  radiation ( $\lambda = 1.5406 \text{ \AA}$ ). Duplicate analyses were performed to ensure reproducibility.

## 2.7 X-ray photoelectron spectroscopy (XPS)

Data from the source rocks, before and after pyrolysis, were obtained using a Kratos Axis ULTRA X-ray Photoelectron Spectrometer. Monochromatic Al K $\alpha$  X-rays (1,486.6 eV) at 150 W (15 kV, 10 mA) were used. The C 1s (286.8 eV) peak was used as an internal standard for correction. For full spectrum scanning, the transit energy was 100 eV, and the step size was 1 eV; for narrow spectrum scanning, the passage energy was 30 eV and the step size was 0.1 eV. To obtain good experimental results, powdered samples ground to  $\leq 0.074 \text{ mm}$  were made into compressed tablets for XPS analysis. In XPS C 1s spectra, the ordinate is electron counts, and the abscissa is electron binding energy (B.E.). Based on the different binding energies of carbon elements in different valence states, the XPS C 1s spectra of samples obtained from our experiments were fitted using Casa XPS software. The Shirley model was adopted for back-bottom deduction, and 70% Gaussian and 30% Lorentz hybrid

TABLE 2 Major D- and G-band shifts for the residue and original sample.

Sample no.	D band	G band	G-D band shifts	VR	T <sub>max</sub>
Origin Samples	1,358	1,574	216	0.4	434
311	1,356	1,576	220	0.77	446
312	1,348	1,583	235	1.46	455
313	1,346	1,583	237	1.76	467
314	1,346	1,586	240	2.14	568
315	1,341	1,586	245	2.23	605
321	1,336	1,569	233	0.81	449
322	1,341	1,576	235	1.50	461
323	1,348	1,588	240	1.95	474
324	1,338	1,583	245	2.29	573
325	1,336	1,583	247	2.44	607

lines were adopted. In the process of peak fitting, the peak position and half-width of all fitted peaks were fixed, and the half-width was 1.5 ( $\pm 0.1$ ) eV.

## 2.8 Low-temperature N<sub>2</sub> adsorption

An ASAP 2020 Plus HD88 SA analyzer (Micromeritics Instruments Corporation, USA) was used to measure the surface area (SA) and pore volume (PV) of the source rocks, before and after pyrolysis. All samples were crushed to 80 mesh and extracted with chloroform for 72 h, then degassed at 150 °C for 6 h under vacuum before low-temperature N<sub>2</sub> adsorption analysis. The N<sub>2</sub> adsorption-desorption isotherm measurements were conducted at the temperature of liquid nitrogen (77.3 K) with a relative pressure P/P<sub>0</sub> range of 0.01–0.998.

## 2.9 Nuclear magnetic resonance (NMR) spectroscopy

The MicroMR series NMR analyzer used in this study was produced by Suzhou Newmag analytical instrument corporation. The NMR core analyzer can be used as free-induction decay (FID) and Carr-Purcell-Meiboom-Gill pulse sequence (CPMG). sequence. To obtain a more complete NMR response signal and improve the signal-to-noise ratio and resolution, CPMG-T<sub>2</sub> measurements were used. The pyrolyzed samples were dried under vacuum, then the dry columnar samples were saturated with deionized water. NMR T<sub>2</sub> distributions were measured using 0.3 ms echo spacing, 3,000 ms waiting time, 10,000 echo numbers, and 64 scan times (Zheng et al., 2019).

## 3 Results and discussion

### 3.1 The characterization of Raman spectroscopy, FTIR, XRD and XPS before and after pyrolysis of samples

Generally, VR and T<sub>max</sub> represent good thermal maturity indicators of source rocks and organic matter (Hunt, 1996; Evenick, 2021; Al-Hajeri et al., 2021; Lohr et al., 2015; Waliczek et al., 2021).

#### 3.1.1 Raman spectroscopy

Previous investigations have shown that with increasing kerogen maturity, aromaticity increases and there is a concurrent decrease in H, O, N, and S content (Hunt, 1996; Lewan, 1997; Sun et al., 2021). These changes may be reflected in the Raman spectra of kerogen at the molecular scale (Khatibi et al., 2019). Hence, Raman spectroscopy can be used to investigate the nature and degree of graphitization of kerogenous carbon (Charisiou et al., 2018). The Raman analysis results of pyrolyzed samples are presented in Figure 3. Both samples show Raman bands in the 800–2000 cm<sup>-1</sup> region of the spectrum; these are the carbon G- and D-bands, which appear between 1,500 and 1,600 cm<sup>-1</sup>, and between 1,300 and 1,400 cm<sup>-1</sup>, respectively. The former is induced by the presence of crystalline graphitic/sp<sup>2</sup> carbon atoms and involves out-of-phase intra-layer displacement in the graphene structure; the latter originates from sp<sup>3</sup> hybridization and relates to polycrystalline imperfect graphite (Li et al., 2017; Charisiou et al., 2018; Ghaffar et al., 2018). The degree of crystallinity of the carbon formed during the thermal evolution of organic matter can also be reflected by the intensity ratio of the D- and G-bands (I<sub>D</sub>/I<sub>G</sub>), where lower I<sub>D</sub>/I<sub>G</sub> values suggest higher crystallinity, because a higher quantity of graphitized carbon is generated during organic matter evolution (Hou et al., 2019). In this investigation, the I<sub>D</sub>/I<sub>G</sub>

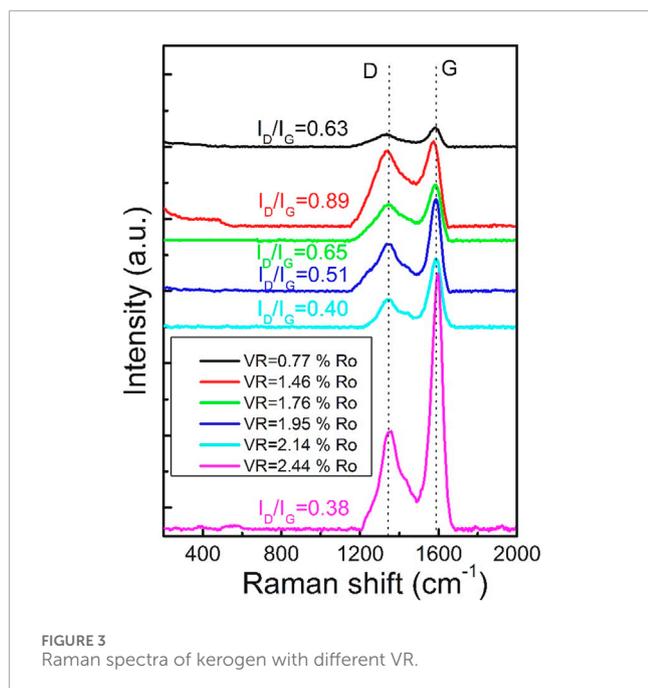


FIGURE 3  
Raman spectra of kerogen with different VR.

values for the pyrolyzed samples follow the order  $0.89 > 0.65 > 0.63 > 0.51 > 0.40 > 0.38$  corresponding to VRs of  $0.77\%R_o < 1.46\%R_o < 1.76\%R_o < 1.95\%R_o < 2.14\%R_o < 2.44\%R_o$ . A lower  $I_D/I_G$  ratio may illustrate higher thermal maturity (Muirhead et al., 2021; Zhang et al., 2021), the deduction of which matches the report of Sauerer et al. (2017). For the pyrolyzed continental shales in this investigation, the degree of crystallinity of kerogenous carbon increased with an increase in VR, and this trend was more pronounced for continental shales with  $VR > 1.46\%R_o$  (Figure 3); this is also consistent with previously noted increases in the aromaticity of kerogen (Beyssac et al., 2002; Lünsdorf, 2016). A decrease in the  $I_D/I_G$  ratio may indicate an increase in larger aromatic clusters and improved ordering of the existing organic compounds within the kerogen.

Previous investigations have reported that G- and D-band shifts increase with maturity until the dry gas window is reached (Khatibi et al., 2018; Delarue et al., 2021; Hackley et al., 2021); the D-band shifts towards lower wavenumbers and the G-band shifts towards higher wavenumbers (Kelemen and Fang, 2001; Khatibi et al., 2018; Hackley et al., 2021; Delarue et al., 2021). These increases in separation between the D- and G-bands may be caused by the increase in number of larger aromatic clusters and improved structural ordering of kerogen as maturity increases (Schito et al., 2017; 2021). Further investigations have speculated that these shifts may be attributed to the wavelength dependence of the D-band position and the independence of the excitation energy of the G-band position in the Raman spectra of carbonaceous materials (Marshall et al., 2010; Lünsdorf, 2016; Schito et al., 2017; Schito et al., 2021). The VR,  $T_{max}$ , and D- and G-band shifts of the original and pyrolyzed samples are listed in Table 3. Figure 4 shows a strong positive relationship between VR and D-G separation; statistical analysis shows that this is a linear relationship with  $R_2 = 0.8901$ .

Generally, the increasing carbon D-G band separation as organic matter maturity increases is attributed to the D-band shifting towards lower wavenumbers, while the G-band remains largely unchanged. During the initial stages of maturation, D-band shifting increases before gradually stabilizing at the stage of over-maturity, like the trend of coal. Kelemen and Fang (2001) found that the metamorphic grade, like maturity, evolves faster at the catagenesis stage than at the metagenesis stage, for both natural and laboratory matured coal. Moreover, some investigators have also indicated that there is good relationship between D-G band separation and  $T_{max}$  values, which represents a good thermal maturity index. However, previous research has suggested that  $T_{max}$  values may not be an appropriate parameter to quantify the thermal maturity of kerogen with high maturity ( $\geq 2.0\%R_o$ ) and low  $S_2$  values (Lewan and Kotarba, 2014; Wu et al., 2018). A cross-plot of  $T_{max}$  versus band separation showed good correlation using data from pyrolyzed samples with  $T_{max}$  ranging from 434°C to 474°C. Statistical analysis also shows that there was a good linear relationship between  $T_{max}$  and the D-G band separation for the pyrolyzed samples with  $T_{max}$  ranging from 434°C to 474°C (Figure 4). The band separation increases with increasing  $T_{max}$  further indicated that the organic compounds in kerogen evolved to larger aromatic clusters and a better-ordered structure.

### 3.1.2 FTIR

The FTIR spectra of the organic-rich shales, before and after pyrolysis, had well-resolved peaks and bands, and flat baselines (Figure 5), which indicated that the FTIR spectra were appropriate for further qualitative analysis. Generally, bands at 3,100–3,000  $cm^{-1}$  and 3,000–2,800  $cm^{-1}$  have been assigned to aromatic CH stretching vibrations and aliphatic CH stretching vibrations, respectively (Li et al., 2007; Okolo et al., 2015; Chen et al., 2019; Cincinelli et al., 2021; Mastalerz et al., 2021; Phan et al., 2021). As shown in Figure 5, the bands at 3,100–3,000  $cm^{-1}$  and 3,000–2,800  $cm^{-1}$  were indiscernible in our pyrolyzed samples; this may indicate that these bands are not suitable for identifying the hydrocarbon potential and maturity of kerogen where there is the influence of large quantities of inorganic minerals within the source rocks.

The band at  $\sim 2,361 cm^{-1}$  may be assigned to adsorbed  $CO_2$  and this band becomes broader and the center shifts to a lower wavenumber when  $^{18}O$  is incorporated into  $CO_2$  (Liao et al., 2002). The broad bands at  $\sim 2,361$  and  $\sim 2,347 cm^{-1}$  are due to band overlap of adsorbed  $C^{18}O_2$ ,  $C^{16}O^{18}O$ , and  $C^{16}O_2$  (Liao et al., 2002). The broad bands at  $\sim 2,361$  and  $\sim 2,347 cm^{-1}$  are due to band overlap of adsorbed  $C^{18}O_2$ ,  $C^{16}O^{18}O$ , and  $C^{16}O_2$ . Those can be confirmed in previous investigation in which  $\delta^{13}C$  of  $CO_2$  was +0.84 to +1.06‰, suggesting that the adsorbed  $CO_2$  was mainly generated from carbonate minerals (Wu et al., 2018), and the inorganic  $CO_2$  enrich  $^{18}O$ , comparing with organic  $CO_2$  under wet climate, low salinity environments, on the basis that the wet climate and the small mineralization of the shells of the lakes are composed of carbon and oxygen isotopes, which are lightly composed, with the climate drying, the salinity of the lake increases, and the shells are enriched with heavy carbon and oxygen isotopes (Ping, 1990).

A band at  $\sim 1,435 cm^{-1}$  was observed in the pyrolyzed samples, representing the aliphatic bending vibration due to  $CH_2$  and  $CH_3$ . This band exhibited more obvious absorbance in the FTIR

TABLE 3 Results of pore size distribution in NMR analyses.

Sample no.	NMR pore size distribution frequency/%			VR	$T_{max}$
	<10 nm	10–100 nm	>100 nm		
Origin Samples	62.2	26.7	11.1	0.4	434
311	66.7	20.5	12.8	0.77	446
312	63.2	34.2	2.6	1.46	455
313	65.5	29.0	5.5	1.76	467
315	72.8	18.2	9.0	2.23	605
323	64.4	25.4	10.2	1.95	474
325	76.2	15.9	7.9	2.44	607

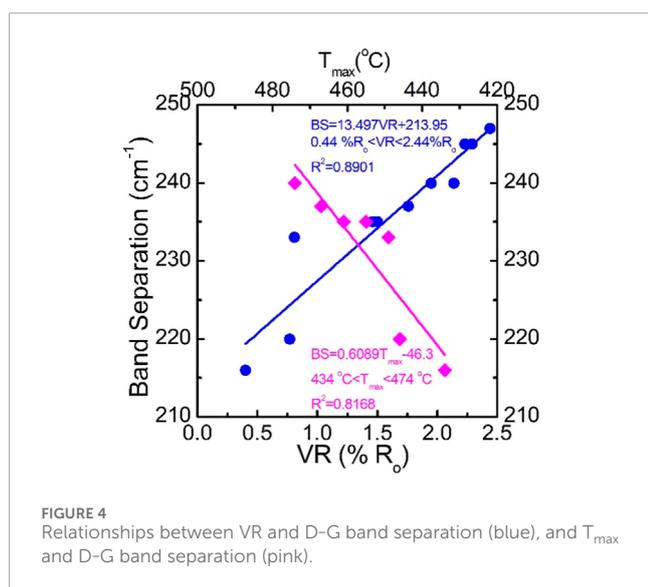


FIGURE 4 Relationships between VR and D-G band separation (blue), and  $T_{max}$  and D-G band separation (pink).

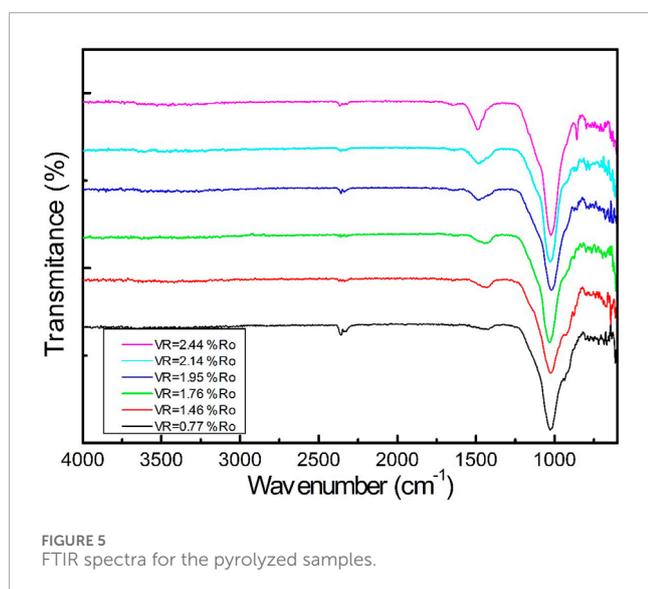
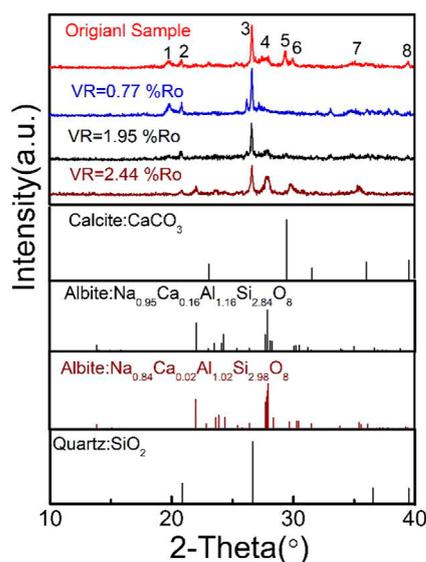


FIGURE 5 FTIR spectra for the pyrolyzed samples.

spectra of the pyrolyzed samples with  $VR_s$  lower than  $1.95\%R_0$ . A shoulder was observed on the left-hand side of the  $1,435\text{ cm}^{-1}$  band, at around  $1,460\text{ cm}^{-1}$ . This is generally attributed to the aromatic C=C stretching vibration (Coates, 2000; Van Niekerk et al., 2008; Ding et al., 2021; Wang et al., 2022). The intensity of this shoulder was found to be stronger for pyrolyzed samples with increased  $VR_s$  from  $0.77\%R_0$  to  $2.44\%R_0$  and was almost higher than the  $1,435\text{ cm}^{-1}$  peak for pyrolyzed samples with  $VR_s$  higher than  $1.95\%R_0$  (Figure 5). The  $1,435\text{ cm}^{-1}$  peak decreased with increasing maturity for  $VR_s$  ranging from  $0.77\%R_0$  to  $2.44\%R_0$  (Figure 5). The  $\sim 1,600\text{ cm}^{-1}$  band, assigned to stretching vibrations of aromatic rings (Okolo et al., 2015; Phan et al., 2021), became more obvious with increasing maturity for  $VR_s$  ranging from  $0.77\%R_0$  to  $2.44\%R_0$  (Figure 5), which was consistent with the behavior of the  $1,460\text{ cm}^{-1}$  peak.

Spectral bands at  $1,280\text{--}1,000\text{ cm}^{-1}$  can be assigned to saturated aliphatic skeletal C-C vibrations, C-O stretching vibrations of ether groups, and O-H bending vibrations in phenolic, phenoxy, and hydroxybenzene structures. These bands were also observed in the FTIR spectra of coals (Coates, 2000; Van Niekerk et al., 2008; Okolo et al., 2015; Phan et al., 2021; Wang et al., 2022). The band at  $1,030\text{ cm}^{-1}$  observed in Figure 5 indicates the intensity of C-C stretching vibrations within an aliphatic skeleton (Okolo et al., 2015; Jing et al., 2019; Razavian and Fatemi, 2021; Shang et al., 2021). The  $1,280\text{--}1,000\text{ cm}^{-1}$  peak heights of all pyrolyzed samples showed a steady increase with increasing VR, and the narrowing peak width as VR increased indicated that some organic functional groups (such as C-O and O-H) were rapidly pyrolyzed and hence sharply decreased; the remaining functional groups (such as C-C) were slowly pyrolyzed and hence showed less of a decrease.

The out-of-plane aromatic C-H bending vibrations indicated the presence of substituted aromatic species in aromatic rings; and trans- and cis- $\text{CH}_2$  in long, saturated aromatic CH-CH chains were observed in the  $900\text{--}700\text{ cm}^{-1}$  region of the spectra in the entirely pyrolyzed samples (Coates, 2000; Van Niekerk et al., 2008; Malumbazo et al., 2011; Okolo et al., 2015; Jing et al., 2019; Shang et al., 2021; Wang et al., 2022). The peaks in the  $900\text{--}700\text{ cm}^{-1}$  region of the spectra became more



**FIGURE 6**  
Identification of various mineral phases in pyrolyzed samples. The standard powder diffraction cards of quartz (PDF#85-0795), albite (PDF#70-3,752):  $\text{Na}_{0.95}\text{Ca}_{0.16}\text{Al}_{1.16}\text{Si}_{2.84}\text{O}_8$ , albite (PDF#76-0927):  $\text{Na}_{0.84}\text{Ca}_{0.02}\text{Al}_{1.02}\text{Si}_{2.98}\text{O}_8$ , and cal. (PDF#72-1937) are also presented.

obvious with increasing VR (Figure 5), indicating an increase in aromatic groups.

### 3.1.3 XRD with VR

In general, the brittleness and adsorption capacity of shale can be affected by its mineralogy. Hence, it is critical for shale evaluation to obtain accurate analysis of the mineral composition of the rocks. Therefore, XRD analysis was performed on the pyrolyzed samples. As can be seen in Figure 6, calcite, albite, and quartz were identified as the main mineral components of the pyrolyzed shale. According to the standard powder diffraction cards of quartz (PDF#85-0795), calcite (PDF#72-1937), albite:  $\text{Na}_{0.95}\text{Ca}_{0.16}\text{Al}_{1.16}\text{Si}_{2.84}\text{O}_8$  (PDF#70-3,752), and albite:  $\text{Na}_{0.84}\text{Ca}_{0.02}\text{Al}_{1.02}\text{Si}_{2.98}\text{O}_8$  (PDF#76-0927), the peaks at 20.8°, 26.7°, 36.6°, and 39.5° were assigned to quartz; peaks at 29.5° and 35.3° were assigned to calcite and  $\text{Na}_{0.84}\text{Ca}_{0.02}\text{Al}_{1.02}\text{Si}_{2.98}\text{O}_8$ , respectively; and peaks at 27.9° and 30.5° were assigned to  $\text{Na}_{0.95}\text{Ca}_{0.16}\text{Al}_{1.16}\text{Si}_{2.84}\text{O}_8$  and  $\text{Na}_{0.84}\text{Ca}_{0.02}\text{Al}_{1.02}\text{Si}_{2.98}\text{O}_8$ , respectively.

Calcite is a carbonate mineral, and its peak (29.5°) intensity dramatically decreased with increasing maturity (Figure 6); this could be ascribed to the thermal decomposition of carbonate under hydrous pyrolysis. It is notable that the quartz peak (26.7°) intensity decreased with increasing VR (Figure 6), while the albite peak (27.9°) intensity increased with increasing VR. This appears to indicate that feldspathization was enhanced with an increase in maturity. In addition, the elemental composition of albite changed with increasing VR; the 35.3° peak assigned to  $\text{Na}_{0.84}\text{Ca}_{0.02}\text{Al}_{1.02}\text{Si}_{2.98}\text{O}_8$  increased with increasing VR, indicating that  $\text{Na}_{0.95}\text{Ca}_{0.16}\text{Al}_{1.16}\text{Si}_{2.84}\text{O}_8$  converted to  $\text{Na}_{0.84}\text{Ca}_{0.02}\text{Al}_{1.02}\text{Si}_{2.98}\text{O}_8$  with increasing VR (Figure 6). Thus, the XRD results suggest that carbonates were broken down, and quartz and albite ( $\text{Na}_{0.95}\text{Ca}_{0.16}\text{Al}_{1.16}\text{Si}_{2.84}\text{O}_8$ ) were converted to  $\text{Na}_{0.84}\text{Ca}_{0.02}\text{Al}_{1.02}\text{Si}_{2.98}\text{O}_8$  as kerogen became more thermally mature. Such

inorganic mineral reactions may affect the brittleness and adsorption capacity of the shales.

### 3.1.4 XPS

To quantify the binding forms of oxygen, carbon, nitrogen, aluminum, and silicon during organic matter evolution, high resolution XPS spectra of O 1s, C 1s, N 1s, Al 2p, and Si 2p were de-convoluted using Gaussian-Lorentzian peaks (Wang et al., 2013; Jing et al., 2019; Liu et al., 2021; Wu and Xu, 2022). The XPS wide energy spectra of the original kerogen sample and the pyrolyzed samples are illustrated in Figure 7. The original sample was found to have low oxygen (12.12%) and high carbon (82.38%) contents. After pyrolysis, the oxygen content increased to 40.15% at VR = 1.46% $R_o$ , 52.01% at VR = 1.95% $R_o$ , and 52.18% at VR = 2.44% $R_o$ , accompanied by a decreasing trend in carbon content (Figure 7A, D, G, J). The trend of a decrease in C 1s suggests that the kerogen gradually broke down and hydrocarbons were generated as VR increased. The increase in O 1s may be attributed to the oxidation of organic carbon or reactions between inorganic minerals during hydrous pyrolysis.

The evolution of carbon content agree with that was found in the ultimate analysis, while evolution of oxygen content showed contrary tendency with that was found in the ultimate analysis of kerogen (Wu et al., 2018). This may indicate that the increase in oxygen content was mainly contained within inorganic material (e.g., albite and quartz) rather than organic matter (e.g., kerogen and retained petroleum). The data shown in Figure 7 directly confirmed the increase in silicon content within feldspar that occurred during feldspathization and organic matter evolution. The decrease in the 533.0 eV binding energy peak corresponding to Si-O in quartz, and the slight decrease in the 531.5 eV binding energy peak corresponding to Al-O in albite further confirmed the conversion of  $\text{Na}_{0.95}\text{Ca}_{0.16}\text{Al}_{1.16}\text{Si}_{2.84}\text{O}_8$  to  $\text{Na}_{0.84}\text{Ca}_{0.02}\text{Al}_{1.02}\text{Si}_{2.98}\text{O}_8$  (Figure 6C,F,I,L).

Figure 7B, E, H, K shows the high resolution XPS spectra of C 1s for calcareous shale. The peak at a binding energy of 284.6 eV corresponds to C-C or C-H groups, while that at 285.7 eV corresponds to a C-O group (alcohol, phenol, or ether) (Buckley and Lamb, 1996; Gong et al., 1998; Jing et al., 2019; Liu et al., 2021; Wu and Xu, 2022). The 288.9–289.5 eV binding energy peaks that appeared at VRs of 1.95% $R_o$  and 2.44% $R_o$  correspond to the COOH (carboxyl) group, which may indicate that water has some oxidability to organic carbon at a high stage of thermal evolution. This is also supported by the binding energy peaks at 531.5 eV in the O 1s XPS spectra increasing at VRs of 1.46% $R_o$ , 1.95% $R_o$ , and 2.44% $R_o$  (Figures 7I, L). This increase in O 1s may partly derive from the oxidation of organic carbon during hydrous pyrolysis.

## 3.2 Evolution of pore structure

### 3.2.1 SEM and SEM-EDS mapping

The Raman, XRD, and XPS results suggested changes in inorganic material and organic matter during hydrous pyrolysis. More information was then needed to directly demonstrate the processes of thermal decomposition of carbonates and organic matter, and conversion of  $\text{Na}_{0.95}\text{Ca}_{0.16}\text{Al}_{1.16}\text{Si}_{2.84}\text{O}_8$  to  $\text{Na}_{0.84}\text{Ca}_{0.02}\text{Al}_{1.02}\text{Si}_{2.98}\text{O}_8$ . As shown in Figure 8, the morphology

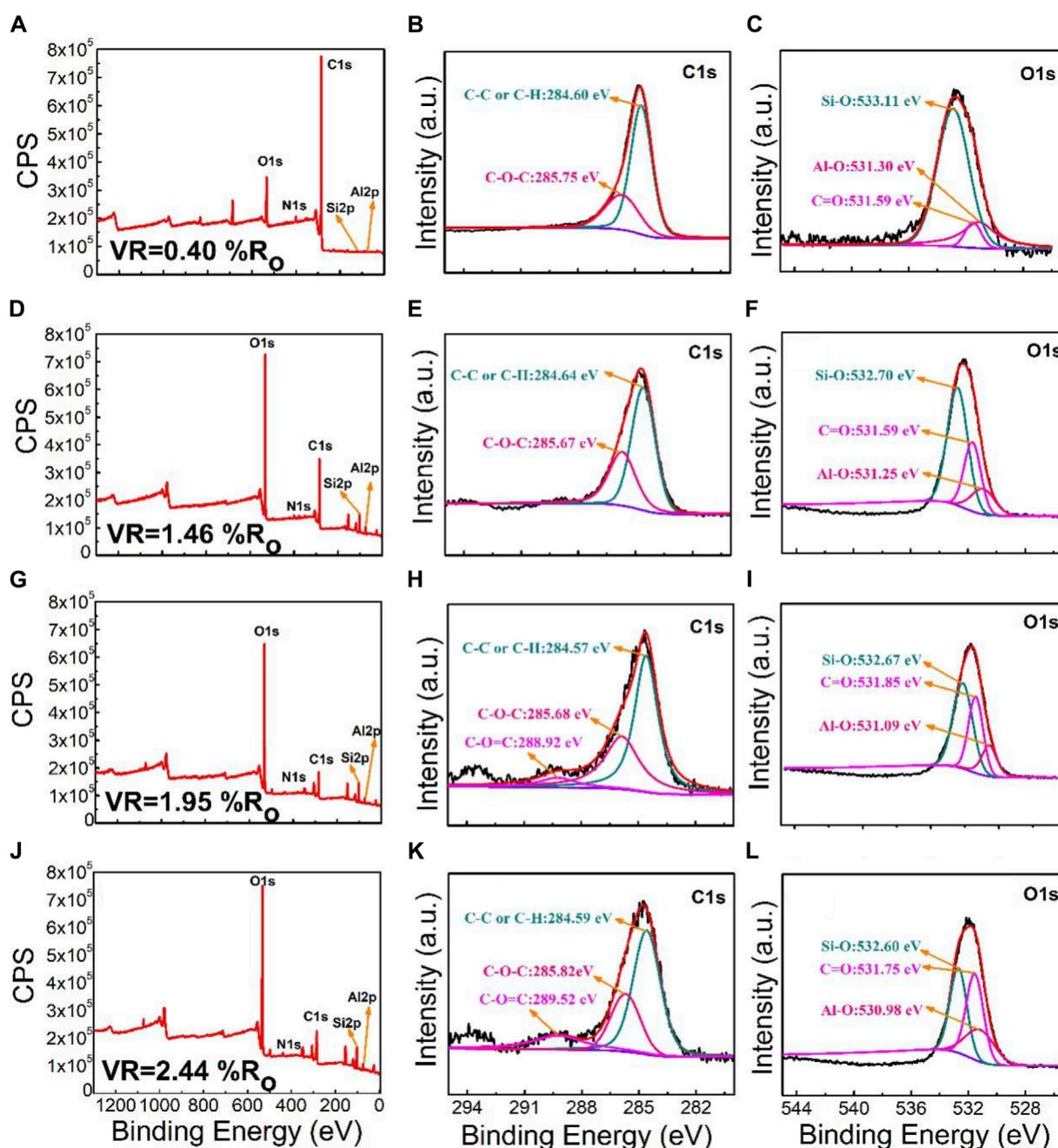


FIGURE 7  
XPS wide scan data from the original kerogen (A–C) and pyrolyzed samples (D–L).

and pore structure of inorganic and organic matter in the original samples and pyrolyzed residues were fully characterized using SEM imaging and SEM-EDS analysis.

In the original samples, organic matter and inorganic minerals were observed in SEM images, and C, Fe, and S were also detected by SEM-EDS (Figures 8A–D). The EDS map of carbon distribution (Figure 8B) indicates that the features labeled 1 and 3 in Figure 8A are organic matter. The distributions of Fe and S correspond with each other (Figures 8C, D), which suggests that the feature labeled 2 in Figure 8A is pyrite, a mineral associated

with a favorable reduction environment for oil-prone organic matter. The elements Si, O, and Al have distributions that correspond with one another (Figures 8F–H), suggesting that the mineral shown in Figure 8E is feldspar. The results of XRD analysis further indicate that this is albite (Figure 6). Carbonate can also be observed in SEM images (Figure 8I).

After hydrous pyrolysis, decomposition of carbonates was observed in SEM images; fractures formed on the surface of the carbonates, some of which broke down into similar layered blocks (Figure 8J). Many pores also formed on the surface of

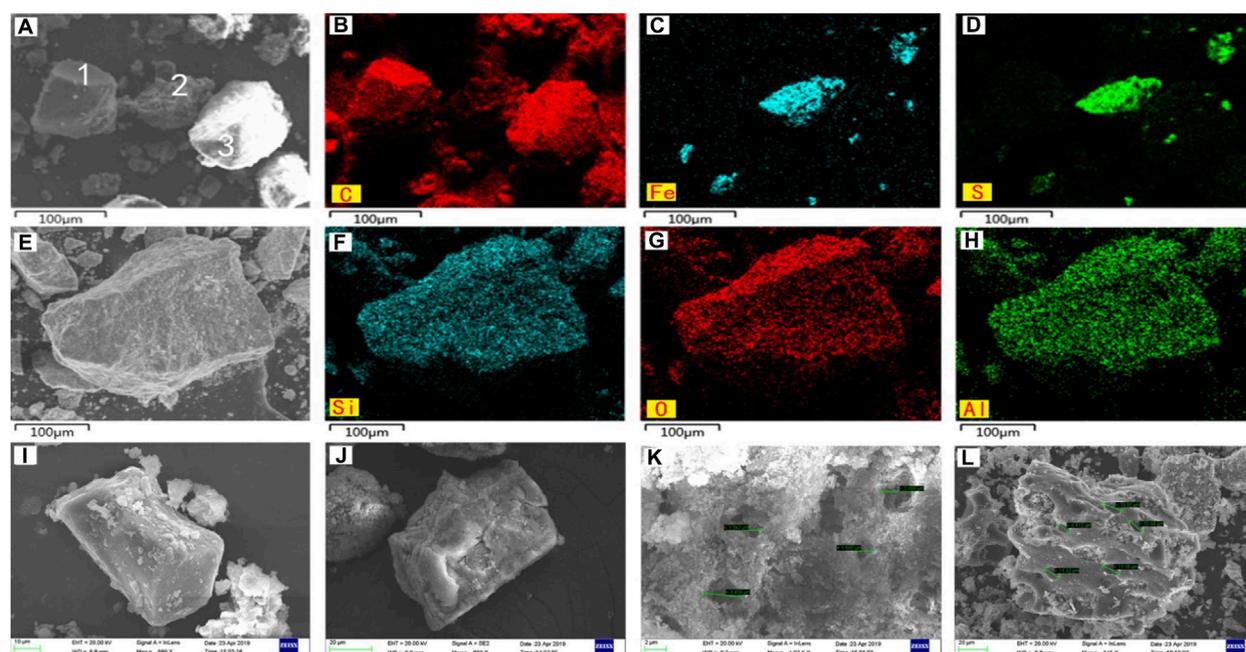


FIGURE 8

SEM images and SEM-EDS elemental maps showing the morphology and chemistry of inorganic minerals and organic matter. Organic matter (1, 3) and pyrite (2) in the original sample (A). Elemental maps of C (B), Fe (C), and S (D) within the organic matter and pyrite. Feldspar in the original sample (E). Elemental maps of Si (F), O (G), and Al (H) within feldspar. Carbonate before (I) and after (J) pyrolysis at VR=1.5. Morphology of pores in feldspar (K) and organic matter (L) after pyrolysis at VR=1.5.

albite, which may have developed during the transformation of  $\text{Na}_{0.95}\text{Ca}_{0.16}\text{Al}_{1.16}\text{Si}_{2.84}\text{O}_8$  to  $\text{Na}_{0.84}\text{Ca}_{0.02}\text{Al}_{1.02}\text{Si}_{2.98}\text{O}_8$  (Figure 6; Figure 9K). Pore sizes ranged from 1.505 to 3.430  $\mu\text{m}$  (Figure 8K). These pores may be the result of the leaching of  $\text{Ca}^{2+}$  and  $\text{Al}^{3+}$  during the transformation of albite. More pores developed around organic matter, and these had a size range of 8.412–13.970  $\mu\text{m}$  (Figure 8L). Similar pores were also observed on the surface of high-rank coals (Zheng et al., 2019; Zhang et al., 2021). Organic matter was also decomposed into many porous pieces by pyrolysis (Figure 8L).

According to the changes in mineral composition observed using XRD and SEM, the main reason for the change in porosity of inorganic minerals caused by temperature and pressure appears to be the dissolution of soluble minerals, such as plagioclase, and the fragmentation of refractory minerals, such as quartz. Owing to the influence of high temperature, the strength of quartz and other relatively strong particles will be reduced, so that they become more easily broken. With a rise in temperature, cracks gradually increase, and mineral particles gradually become broken. In addition, as temperature increases, feldspar and other easily plastically deformed minerals will not only dissolve, but also be transformed (e.g., into kaolinite). Some investigators have shown that the conversion of feldspar into clay minerals could markedly reduce the volume of feldspar and increase porosity (Worden and Morad, 1999; Yang et al., 2020; Sun et al., 2021). In the process of dissolution and deformation, minerals will be affected by compaction, resulting in closer particle contact and smaller pores. However, the sample pores are filled with high-pressure fluid, which counteracts much of the overlying rock pressure and inhibits compaction, thus protecting the existing and newly generated pores.

### 3.2.2 BET

SEM images show the micropores that directly reflect the changes in organic matter and inorganic minerals, and store free shale gas (Ji et al., 2017). However, some investigations have suggested that adsorbed gas, the main gas component of shales, is mostly stored in nano-pores (Ji et al., 2017; Memon et al., 2021; Zhu et al., 2021; Zhou et al., 2022). To analyze the nanoporous structure and surface chemistry of the origin samples and pyrolyzed residues, the BET isotherm measurements are carried out. The nitrogen isotherms at different VR are presented in Figure 9. The appearance of hysteresis loops appeared for the different capillary condensation and capillary evaporation under the same pressure (Kruk and Jaroniec, 2001; Sun et al., 2015; Qiu et al., 2021). Generally, the greater the separation between the desorption and adsorption isotherms the larger the pore content, under the same pressure (Fan et al., 2002; Wang et al., 2022). However, the degree of separation between the desorption and adsorption isotherms showed no obvious change in our pyrolyzed residues with increasing VR (Figure 9). As the VR increased, the quantity of adsorbed nitrogen gradually increased, compared with the unheated sample; the quantities adsorbed were 14.32, 21.70, 23.39, 25.84, 30.42, 30.89, 34.94, 35.47, 38.73, and 43.14  $\text{cm}^3/\text{g}$  for the unheated sample and the pyrolyzed samples with VRs of 0.77% $R_o$ , 0.81% $R_o$ , 1.16% $R_o$ , 1.76% $R_o$ , 1.95% $R_o$ , 2.14% $R_o$ , 2.23% $R_o$ , 2.29% $R_o$ , and 2.44% $R_o$ , respectively (Figure 8). There was a good relationship between the nitrogen quantity adsorbed and the VR, with a correlation coefficient ( $R^2$ ) of 0.93 (Figure 10). The cumulative BET surface areas also had a good correlation with VR, with an  $R^2$  of 0.86 (Figure 10). This may suggest that the maturation of

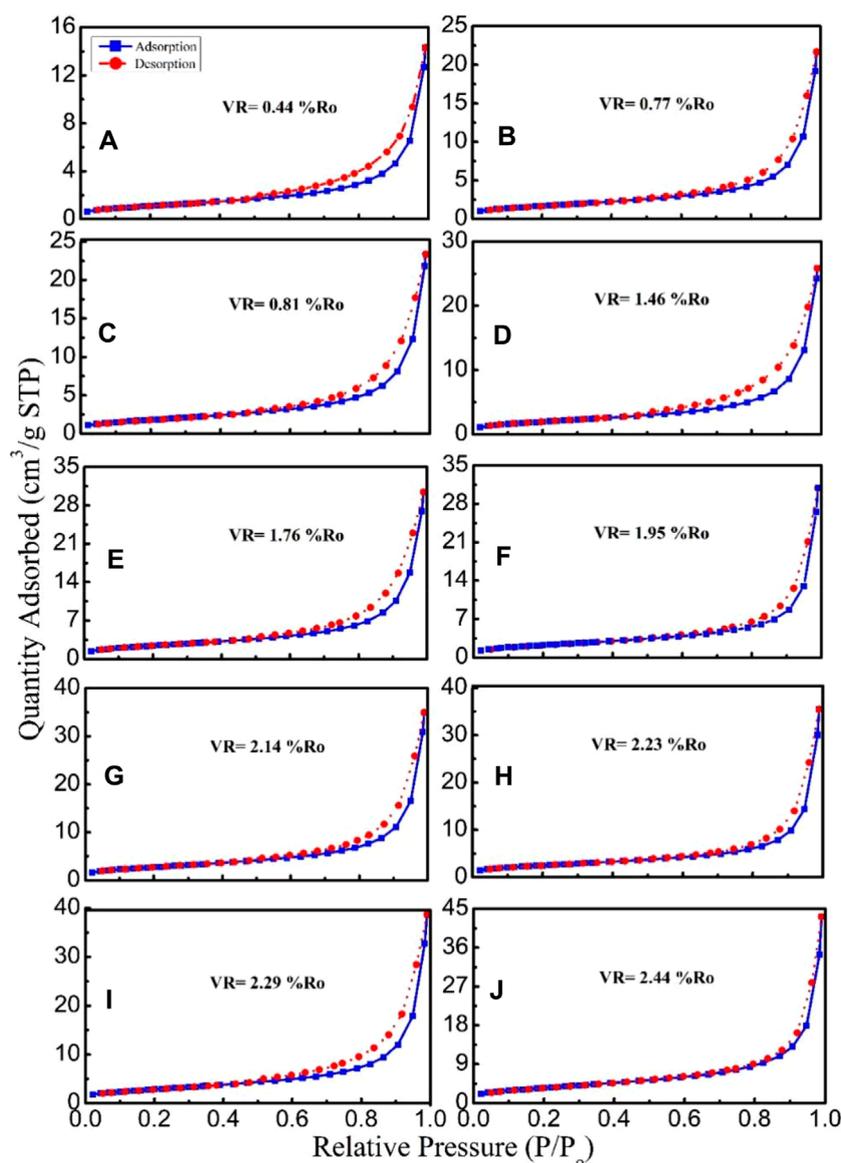


FIGURE 9 Nitrogen adsorption isotherms for the unheated sample (A) and pyrolyzed samples (B–J) with different VRs.

organic matter plays a positive role in the development of pores in shales. High pore content is helpful to enhance the self-storage capacity of shale gas. This is a possible reason why highly gas-productive shale is generally also highly mature shale.

The pore size distributions of unheated and pyrolyzed samples are presented in Figure 11. In these samples, the pore sizes were distributed across two main size intervals of 4–30 nm and 30–100 nm, which can be generalized as mesopores (pore diameter: 2–50 nm) and macropores (pore diameter: >50 nm), respectively (Figure 11). The surface area of the unheated sample was relatively low, and the pores were mainly mesopores with 4–30 nm diameters (Figure 11). The pore volume of mesopores gradually increased with an increase in VR from 0.77% $R_0$  to 1.76% $R_0$  (Figure 11). This increase in mesopores may be a response to peak hydrocarbon generation, which appears at a VR of 1.5% $R_0$ .

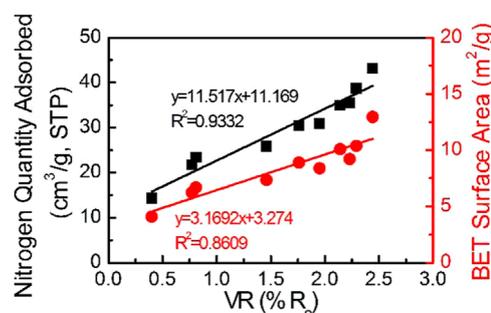


FIGURE 10 Relationships between VR and nitrogen quantity adsorbed (black), and between VR and BET surface area (red).

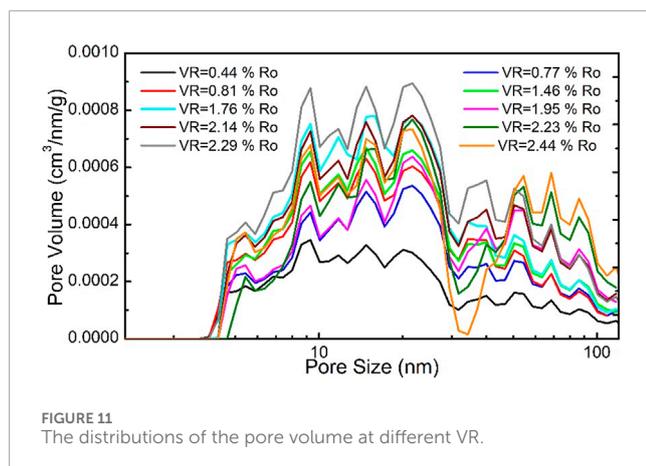


FIGURE 11  
The distributions of the pore volume at different VR.

(Wu et al., 2018a). At VRs beyond 1.76% $R_o$ , there was no obvious further increase in the pore volume of mesopores; instead, there were merely small fluctuations (Figure 11). For the macropores, pore volume gradually increased with an increase in VR from 0.77% $R_o$  to 2.44% $R_o$  (Figure 11). We suspect that these macropores are related to the decomposition of inorganic minerals during the pyrolysis experiments because they increased with increasing thermal maturity.

The changes in pore volume and surface area show that there were two evolutionary stages in the hydrous pyrolysis of the calcareous shale: 1) a rapid increase stage coinciding with the hydrocarbon generation peak; and 2) a relatively stable stage after the hydrocarbon generation peak. The results show that pore structure evolution in the calcareous shale resulted from two different pore type behaviors: the pore volume and surface area of pores over 30 nm increased gradually, while the volume of pores under 30 nm firstly increased and then continuously decreased after the hydrocarbon generation peak.

### 3.2.3 NMR $T_2$ distribution

Nuclear magnetic resonance is an effective method to characterize the pore size distribution (PSD) of source rocks in a non-destructive manner (Xin et al., 2019; Zheng et al., 2019; Nie et al., 2021; Ma et al., 2022). The transverse relaxation time ( $T_2$ ) and pore size ( $r$ ) are related by  $r = C * T_2$ , in which  $C$  is a constant ( $\mu\text{m}/\text{ms}$ ). Therefore, the  $T_2$  measured by NMR can directly reflect the pore size distribution of rock samples, and a higher  $T_2$  value corresponds to a larger pore radius (Xu et al., 2018; Meng et al., 2021). In addition, the connectivity among pores can also be obtained using NMR.

Figure 12 presents the NMR analysis results from the calcareous shale, before and after pyrolysis. Overall, NMR peaks occurred at approximately 1, 10–100, and 100–1,000 ms, which indicated that mesopores and micropores in both the calcareous shale and the pyrolyzed samples were more developed (Figure 12). This result is consistent with previous investigations, such as Zheng et al. (2019) and Xin et al. (2019). The  $T_2$  distribution of the original calcareous shale showed a trimodal distribution, with the left-hand peak being the highest, indicating a high micropore content and low content of medium-large pores before the pyrolysis experiment (Figure 12). After pyrolysis, the peak height of the micropores increased, while

that of the macropores decreased, and all three peaks gradually shifted to the left, representing an increase in micropores and a decrease in macropores (Xin et al., 2019). These changes may be due to the disappearance of primary pores and the further development of secondary pores and fractures. In subsequent experiments, with an increase in VR, the shape of the  $T_2$  spectrum did not change, but the peak height of the main peak kept rising, which confirmed the development of secondary pores and fractures.

In addition, the contribution percentages of pore sizes of different scales as VR increased were analyzed and are presented in Figure 14. The peaks at 10–100 ms almost vanished, which suggested that transition pores sharply reduced when VR increased from 0.40% $R_o$  to 0.77% $R_o$  (Figures 13, 14). The highest yields of bitumen (22.6 mg/g TOC; Wu et al., 2018) occur in this % $R_o$  range, which may indicate that the formation of a large quantity of bitumen blocked some micropores, and most of the transition pores and mesopores, ultimately resulting in low pore connectivity as VR increased from 0.40% $R_o$  to 0.77% $R_o$ . The peaks at 10–100 ms reached a maximum at a VR level of 1.46% $R_o$  (Figures 13, 14), indicating that a large quantity of transition pores was generated at this stage. Low bitumen yields (10.0 mg/g TOC) and high oil yields (96.8 mg/g TOC) suggest that bitumen was further pyrolyzed into liquid hydrocarbons with lower molecular weight and viscosity with this increase in maturity (Wu et al., 2018); here, transition pores and mesopores were flushed by hydrocarbon expulsion, which markedly improved pore connectivity. This process continued until the oil yield peak at a VR of 1.95% $R_o$  (Figures 13, 14). As maturity increased, the oil began to crack, which resulted in an increase in micropores. Higher maturity and deeper burial may result in the collapse of transition pores and mesopores, and reduced pore connectivity (Dong and Harris, 2020; Zhang et al., 2020; Liu et al., 2021). ; pore collapse may also be triggered by tectonic deformation (Liu et al., 2020). This process may provide a better shale oil and gas preservation condition.

The variation in pore size distribution of rock samples shows that the primary pores of rocks will gradually decrease with an increase in burial depth. On the other hand, owing to hydrocarbon generation and oil cracking, organic matter is pyrolyzed, minerals are corroded by organic acids generated during hydrocarbon generation, and secondary pores are constantly generated. The results of pore evolution characterized by NMR were essentially consistent with the geological conditions.

### 3.2.4 Evolution of hydrocarbon generation and pore structure

Thermal maturity has a marked impact on hydrocarbon generation (Tissot and Welt, 1984; Hunt, 1996; Li et al., 2022; Shao et al., 2022) and pore development (Sun et al., 2015; Sun et al., 2018; Song et al., 2019; Gao et al., 2021; Xu et al., 2021). Pore connectivity, pore size distribution, and pore volume are important for the methane adsorption performance of shale (Karacan et al., 2007; Zou et al., 2015; Xu et al., 2018; Kong et al., 2021). Previous investigations have reported that the formation and development of pore structure is closely related to hydrocarbon generation (Sun et al., 2015; Gao et al., 2021; Xu et al., 2021). Hence, the evolution of hydrocarbon generation and pore structure corresponding to different VR levels were analyzed; this is illustrated in Figure 14.

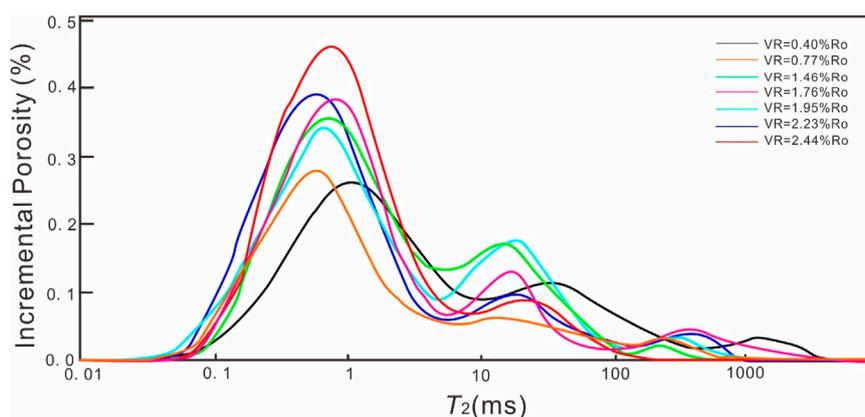


FIGURE 12  
NMR  $T_2$  distributions of calcareous shale and pyrolyzed samples.

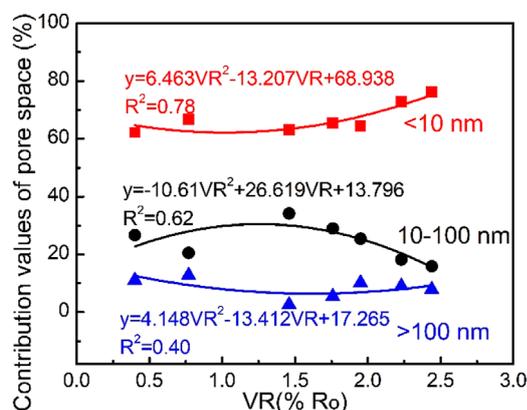


FIGURE 13  
Contribution percentages of pore sizes of different scales in relation to increasing VR.

With an increase in burial depth and thermal maturity, the evolution of hydrocarbon generation from organic matter mainly follows a pathway from bitumen generation to liquid hydrocarbon generation, and then to gaseous hydrocarbon generation, according to the classical theory of hydrocarbon generation from organic matter (Tissot and Welt, 1984; Hunt, 1996; Lewan, 1997; Hill et al., 2003; Ugunna et al., 2016). The yields of bitumen, oil, and hydrocarbon gas indicated that the calcareous shale experienced the above stages at VRs from 0.40% $R_o$  to 2.44% $R_o$ , under hydrous pyrolysis, and the yields of oil and gaseous hydrocarbons showed a good quadratic function relationship with VR values (Figure 14; Wu et al., 2018). The peak oil yield occurred at a VR of 1.46% $R_o$  to 1.95% $R_o$ . Gaseous hydrocarbon generation increased markedly between 0.5% $R_o$  and 2.0% $R_o$ , and then stabilized (Figure 14; Wu et al., 2018).

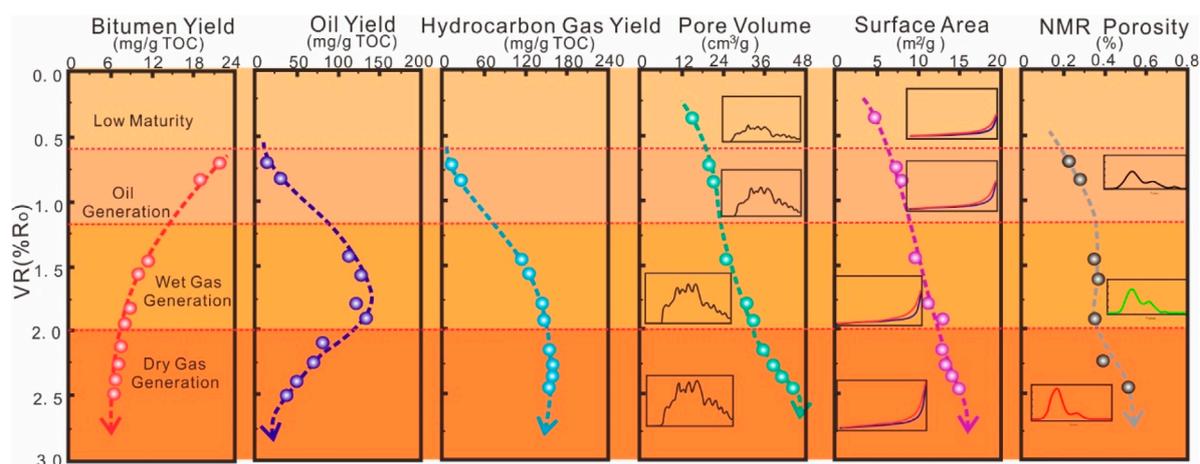
Because of the pyrolysis reactions that occurred during the generation of hydrocarbons, the products had different chemical compositions and physical properties. The bitumen mainly consisted of macromolecular hydrocarbons and heteroatomic functional

groups, and thus had high viscosity; oil had better fluidity due to it containing a high proportion of saturated hydrocarbons; gaseous hydrocarbons had the best diffusion ability and provided the largest contribution to pressure increase during pyrolysis for the gas phase. The transformations of kerogen to bitumen, bitumen to oil, and oil to gaseous hydrocarbons with increasing maturity determined the compositions and properties of the hydrocarbons, and markedly affected the evolution of pore structure. The good linear relationships between the nitrogen quantity adsorbed, BET surface area, and the VR indicated that the surface area and pore volume of shale increased with the prolongation of hydrocarbon generation (Figure 10). However, the distributions of the pore volume in BET and NMR at different VRs further suggested that mesopores, transition pores, and micropores in calcareous shale had different evolutionary trends during prolonged hydrocarbon generation (Figures 13, 14).

The highest yields of bitumen corresponded to the lowest mesopore and transition pore volumes, suggesting that some micropores and most of the transition pores and mesopores were plugged by the high-viscosity bitumen, ultimately resulting in poor pore connectivity. Then, mesopores, transition pores, and micropores increased gradually with a decrease in average pore size. This may indicate that the bitumen pyrolyzed into oil that had better fluidity, flushing the mesopores and transition pores; further pyrolysis of organic matter generated more pores and resulted in better pore connectivity. The pores observed on the surface of feldspar and organic matter in SEM images further confirmed the development of pores through organic matter pyrolysis and through the dissolution of feldspar induced by organic acids that formed during the pyrolysis of organic matter (Figures 8I, K). Oil cracking to gaseous hydrocarbons resulted in an increase in micropores, and poor pore connectivity due to the collapse of transition pores and mesopores under prolonged hydrocarbon generation.

### 3.3 Geological implications

Organic matter and mineral properties are the main factors determining the pore characteristics of shale because the



**FIGURE 14**  
Maturity, hydrocarbon generation, and pore evolution of calcareous shale during pyrolysis experiments. The evolution of hydrocarbon generation is based on Wu et al. (2018).

development of secondary porosity corresponds to hydrocarbon generation and the transformation/dissolution of minerals (Ji et al., 2017; Wu H. et al., 2021). Generally, the pore structures of shales have been determined by the total organic carbon content (TOC), type of organic matter, mineral composition, and thermal maturity (Ji et al., 2017; Wang et al., 2021; Xu et al., 2021; Guan et al., 2022). Among them, thermal maturity has been most widely used (Liang et al., 2014; Sun et al., 2015; Gao et al., 2021; Xu et al., 2021), because pores induced by the generation of hydrocarbons are the main contributors to secondary porosity in shale. Hence, highly productive shale gas has been strongly associated with high-maturity shales (Chatellier et al., 2011; Hao and Zou, 2013; Kotarba and Lewan, 2013; Ji et al., 2017; Wu et al., 2019; Chen et al., 2020; Song et al., 2021; Zheng et al., 2021; Cao et al., 2022). Rapid and effective characterization of maturity is important for shale gas exploration. Vitrinite reflectance is commonly used to characterize the maturity of organic matter. However, owing to differences in origin, some types of organic matter are amorphous and lack vitrinite, for example, the shales of the Wufeng-Longmaxi Formation (Chen et al., 2019). Such types of organic matter may not be suitable for VR analysis. Whole-rock FTIR, Raman, XRD, XPS, SEM, BET, SEM, and NMR results can provide thermal and pore evolution information, which is useful for understanding the association of pore development with hydrocarbon generation in shale. The evolution of pore structure can be divided into three stages: i) primary pores are filled by high-viscosity bitumen, resulting in poor pore connectivity; ii) development of secondary porosity corresponding to oil generation, with mineral dissolution induced by organic acids also providing some contribution to the generation of secondary pores; iii) further development of secondary porosity at the micropore scale, accompanied by the collapse of transition pores and mesopores, resulting in poor pore connectivity and an improved shale oil and gas preservation condition.

## 4 Conclusion

To interpret the evolution of thermal maturity and pore structure of continental organic-rich shales and understand the interaction between organic matter and minerals during thermal evolution, calcareous shale (before and after pyrolysis) was analyzed using Raman, FTIR, XRD, XPS, SEM, BET, and NMR. The DA, SR, SP, DT, and ER of VR, Raman G-D band separation, Raman  $I_D/I_G$ ,  $T_{max}$ , FTIR, and XPS were also estimated to develop and evaluate a comprehensive new set of maturity indicators. The results indicated that the Raman G-D band separation and  $I_D/I_G$  ratio, with good DA, low SR, low SP, and short DT, correlated well with VR and  $T_{max}$  in the pyrolyzed samples. This may indicate that Raman analysis provides good thermal maturity indicators. Vitrinite reflectance and  $T_{max}$  may be less widely applicable given their moderate SR, complex SP, and long DT. The FTIR and XPS techniques also provided semi-quantitative maturity indicators but had poor DA and high ER. Moreover, SEM allowed direct observation of pore formation accompanying the pyrolysis of organic matter, together with changes in minerals, such as the transformation of  $Na_{0.95}Ca_{0.16}Al_{1.16}Si_{2.84}O_8$  to  $Na_{0.84}Ca_{0.02}Al_{1.02}Si_{2.98}O_8$ , which was confirmed by XRD. The results of BET and NMR further suggested that three stages may exist in the evolution of pore structure, in which secondary porosity developed after primary pores were filled by bitumen generation, then this secondary porosity further developed during oil-cracking. The poor pore connectivity observed at both low and high VR values may be more conducive to the preservation of shale oil and gas. This work is significant for research into the interactions between organic matter and minerals during the thermal evolution of continental shales.

## Data availability statement

The original contributions presented in the study are included in the article/supplementary material, further inquiries can be directed to the corresponding author.

## Author contributions

YL: Conceptualization, Writing–review and editing, Project administration. YW: Methodology, Writing–original draft. RZ: Data curation, Investigation, Writing–review and editing. SJ: Methodology, Supervision, Writing–review and editing. XD: Validation, Writing–review and editing.

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## Conflict of interest

The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

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