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Variation in soil nitrous oxide emission with nitrogen application rates under reclaimed water irrigation

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Nitrous oxide (N₂O) is a potent greenhouse gas primarily emitted from agricultural soils through microbial nitrogen transformation processes. Different nitrogen application rates and fertilizer types influence soil nitrogen transformation pathways, thereby affecting N₂O production and emissions. Reclaimed water (RW), due to its chemical composition, may further modulate these processes. In this study, a disturbed soil incubation experiment was conducted using two irrigation water types [RW and deionized water (CW)], three nitrogen fertilizer forms [ammonium sulfate (NH₄⁺), potassium nitrate (NO₃⁻), and sodium nitrite (NO₂⁻)], and two nitrogen application rates (200 and 400 mg N kg⁻¹) to examine the dynamics of soil N₂O emissions. The study found that, compared to CW, high fertilization levels (400 mg N kg⁻¹) of NH₄⁺ under RW treatment significantly increased cumulative soil N₂O emissions by 25.04%, primarily by enhancing the abundance of the ammonia monooxygenase gene in ammonia-oxidizing archaea (*AOA-amoA*), the ammonia monooxygenase gene in ammonia-oxidizing bacteria (*AOB-amoA*), and the nitrite reductase gene (*nirS*). However, at low fertilization levels (200 mg N kg⁻¹) of NH₄⁺, there is no significant differences in cumulative N₂O emissions. Under NO₃⁻ treatment, although RW increased the abundance of *AOA-amoA* and *AOB-amoA*, it did not lead to higher soil NO or N₂O emissions at either high or low NO₃⁻ concentrations. In contrast, under NO₂⁻ treatment, RW increased the abundance of *AOA-amoA* and *AOB-amoA* compared to CW, significantly enhancing cumulative soil N₂O emissions by 27.56% and 39.25%, respectively. In conclusion, RW irrigation does not elevate soil N₂O emissions with nitrate-based fertilizers. However, careful management of nitrification rates is required with ammonium-based fertilizers, including the use of nitrification inhibitors and improved soil aeration, to minimize NO₂⁻ accumulation and related environmental risks.

KEYWORDS

water quality, fertilizer type, fertilizer level, soil nitrous oxide, soil nitrogen transformation-related genes

1 Introduction

With the increasing occurrence of extreme weather events, RW has emerged as a stable and reliable water source that can effectively alleviate agricultural water shortages (Maestre-Valero et al., 2019; Santos et al., 2023). However, RW contains dissolved organic nitrogen, dissolved organic carbon, inorganic nitrogen, and a diverse array of microorganisms, which can significantly alter the physical, chemical, and biological properties of the soil (Lyu and Chen, 2016), thereby influencing soil nitrogen transformation processes. N_2O is a potent greenhouse gas with a strong global warming potential and plays a critical role in stratospheric ozone depletion, with soil being a major source of its emissions (Allen et al., 2014; Tian H. et al., 2020). Understanding the mechanisms of soil N_2O emissions under RW irrigation is therefore of great environmental significance.

Soil N_2O emissions are a byproduct of nitrogen transformation processes and are highly influenced by changes in soil pH, electrical conductivity (EC), nutrient availability, and microbial community composition. As a key substrate for soil nitrogen transformation, including nitrification and denitrification, nitrogen fertilizers have long been a focal point of research (Janke et al., 2021; Chen et al., 2024; Lyu et al., 2024). Different types and application rates of nitrogen fertilizers can significantly alter soil physicochemical properties, affecting soil pH (Pareja-Sánchez et al., 2020) and nitrogen cycling genes (Ouyang et al., 2018). In dryland soils, nitrification is the predominant nitrogen transformation process, and increasing soil NH_4^+ levels has been shown to significantly enhance N_2O emissions (Li et al., 2017). Additionally, N_2O emissions tend to be lower with NO_3^- based fertilizers compared to NH_4^+ -based fertilizers (Tian D. et al., 2020), likely due to the fact that NH_4^+ -based fertilizers contribute to greater NH_3 volatilization (Bi et al., 2025). Moreover, AOA and AOB, the key microbial groups driving the nitrification process, are highly sensitive to nitrogen availability (Segal et al., 2017). Variations in nitrogen fertilizer application can lead to shifts in soil microbial communities and alter nitrification and denitrification pathways (RÜTting et al., 2010; Duan et al., 2019), further complicating efforts to mitigate N_2O emissions (Wang et al., 2018).

In agricultural production, numerous studies have confirmed that RW irrigation can effectively increase soil N_2O emissions (Chi et al., 2020b; Pareja-Sánchez et al., 2020). Given the complexity of fertilization environments and RW quality, RW may interact with nitrogen fertilizers, regulating nitrification and denitrification processes and further affecting N_2O emissions. Chi et al., 2020a found that slow-release fertilizers could effectively regulate soil N_2O emissions. Shang et al. (2016) demonstrated that applying nitrate-based fertilizers under RW irrigation could significantly reduce soil N_2O emissions. Additionally, Chi et al. (2020b) used the DNDC model to analyze the optimal fertilization rate of 225 kg N ha^{-1} under RW irrigation in North China, achieving a balance between crop yield and greenhouse gas emissions. These findings suggest that the behavior of different nitrogen ions may change under RW irrigation conditions. For example, a high NH_4^+ supply may enhance nitrification, while a high NO_3^- supply may not necessarily promote denitrification. Therefore, systematically studying the

transformation processes of different nitrogen ions under RW irrigation and their impact on N_2O emissions can help optimize fertilization management and water resource utilization to mitigate agricultural greenhouse gas emissions. However, systematic research on these transformation processes remains limited.

Therefore, this study investigates the effects of different nitrogen levels (NH_4^+ , NO_3^- , NO_2^-) on soil nitrogen transformation and soil N_2O and NO emission patterns under RW irrigation through indoor incubation experiments. The study analyzes the relationship between water quality changes and the abundance of soil nitrogen transformation-related genes under different nitrogen conditions. This aims to elucidate the impact of nitrogen types and fertilizer nitrogen levels on the pathways of soil N_2O production under RW irrigation.

2 Materials and methods

2.1 Experiment design

This study employed an indoor disturbed soil incubation experiment, considering three types of nitrogen fertilizers: ammonium sulfate- $(\text{NH}_4)_2\text{SO}_4$, potassium nitrate- KNO_3 , and sodium nitrite- NaNO_2 ; two nitrogen application gradients (200, 400 mg N kg^{-1}); and two water quality treatments (CW and RW), as detailed in Table 1. The soil samples for the RW treatment were taken from the 0–30 cm topsoil layer at China Agricultural University Tongzhou Experimental Station in 2020. The sampling area underwent RW and groundwater irrigation trials from 2013 to 2020, as documented in the literature (Chi et al., 2020b; 2020a; 2023). In short-term indoor incubation experiments, the high ion content in groundwater may interfere with the assessment of RW treatments and obscure the differences between treatments, thus affecting the accuracy and interpretability of the experimental results. Therefore, considering the differences in water quality during experimental design, CW should be used as a control treatment to avoid interference caused by complex water quality. Five sampling points were randomly selected and mixed in RW and groundwater irrigated area. Surface soil samples (0–30 cm) for the RW and CW treatments were collected after the final irrigation events with RW and groundwater, respectively, in 2020. The irrigation water quality and soil characteristics for the experiment are detailed in Table 2. According to Table 2, RW contains measurable levels of inorganic nitrogen. Furthermore, soil analysis reveals that long-term RW irrigation markedly elevates soil NO_3^- content and electrical conductivity (EC). The experiment is divided into two stages: pretreatment phase and experimental phase.

2.1.1 Pretreatment phase

30 g of dry soil was placed into a 250 mL glass bottle. A syringe was used to slowly and evenly apply CW and RW into the soil in a clockwise manner to ensure uniform water distribution, maintaining a soil water-filled pore space (WFPS) of 40%. The bottles were weighed daily, and water was added as needed to maintain the WFPS. They were then placed into a completely dark incubator with ventilation for 7 days to deplete soil nitrogen and activate soil microbial activity.

TABLE 1 The treatment details and corresponding code.

Nov	Treatments	Fertilizer level (mg N kg ⁻¹)/code	Code
1	CW+(NH ₄) ₂ SO ₄	200/AN-2	CAN2
2	CW + KNO ₃	200/KN-2	CKN2
3	CW + NaNO ₂	200/NI-2	CNI2
4	CW +(NH ₄) ₂ SO ₄	400/AN-4	CAN4
5	CW + KNO ₃	400/KN-4	CKN4
6	CW + NaNO ₂	400/NI-4	CNI4
7	RW+(NH ₄) ₂ SO ₄	200/AN-2	RAN2
8	RW + KNO ₃	200/KN-2	RKN2
9	RW + NaNO ₂	200/NI-2	RNI2
10	RW +(NH ₄) ₂ SO ₄	400/AN-4	RAN4
11	RW + KNO ₃	400/KN-4	RKN4
12	RW + NaNO ₂	400/NI-4	RNI4

Note: RW, means reclaimed water; CW, means deionized water; AN, means (NH₄)₂SO₄; KN, means KNO₃; NI, means NaNO₂.

TABLE 2 The properties of water quality and soil.

Index	RW	DW	Soil-CW	Soil-RW
COD _{Cr} (mg L ⁻¹)	41.23 ± 2.23	0	—	—
BOD ₅ (mg L ⁻¹)	8.23 ± 5.23	0	—	—
NH ₄ ⁺ -N (mg kg ⁻¹)	7.21 ± 3.62	0	1.21 ± 0.12	0.81 ± 0.25
NO ₃ ⁻ -N (mg kg ⁻¹)	13.64 ± 4.12	0	9.12 ± 0.74	13.23 ± 0.37
SS (mg L ⁻¹)	11.23 ± 3.35	0	—	—
TN (mg kg ⁻¹)	0.03 ± 4.61	0	3,332.12 ± 140.34	4,012.23 ± 1,222.34
NO ₂ ⁻ -N (mg kg ⁻¹)	4.2 ± 0.42	0	0.56 ± 0.02	0.85 ± 0.14
pH	7.10 ± 0.2	7.10 ± 0.00	7.82 ± 0.31	7.73 ± 0.11
EC (μS cm ⁻¹)	812.57 ± 21.12	23.74 ± 21.12	621.34 ± 32.32	812.34 ± 17.11

Note: Soil-DW, means that Soil samples irrigated with groundwater for a long time; soil-RW, means that Soil samples irrigated with RW, for a long time; COD_{Cr}, means chemical oxygen demand; BOD₅ means biochemical oxygen demand; SS, means suspended solids; EC, means electrical conductivity; SOM, means soil organic matter. pH is measured using a 1:2.5 soil-to-water ratio (10 g soil: 25 mL deionized water).

2.1.2 Experimental phase

For each treatment, nitrogen fertilizer was dissolved in 3.03 g of water (either CW or RW) to achieve a nitrogen application rate of 200 mg N kg⁻¹. The solution was evenly applied to the soil samples using a syringe, adjusting the WFPS to 70%. The bottles were then placed in an incubator at 28°C. Every 12 h, the bottles were removed, weighed, and their water content adjusted as needed before returning them to the incubator.

2.2 The related index

2.2.1 Soil N₂O and NO

Headspace gas samples of N₂O and NO were collected on days 1, 2, 3, 4, 5, 7, 13, 18, 23, and 28 during the incubation period. Prior to sampling, each sample bottle was ventilated for

2 h. An initial 20 mL of headspace gas was drawn before sealing the bottles, and a subsequent sample is taken after the bottles remained sealed for 24 h. For NO concentration assays, a 20 mL sample of the detection gas was mixed with 980 mL of helium in a 1 L sampling bag to ensure the passage of a full liter of gas through the analysis system. N₂O concentrations were measured using gas chromatography with an Agilent GC-6820 system (Agilent Technologies Inc., Santa Clara, CA, United States), while NO levels were assessed using a 42i chemiluminescence NO-NO_x analyzer (Thermo Environmental Instruments Inc., Franklin, MA, United States). Each treatment was replicated three times.

Soil N₂O and NO emission fluxes were calculated based on [Formula 1](#). The cumulative gas was determined as the sum of the daily concentrations, and the unmonitored values were estimated from the adjacent differences, referred to [Wei et al., 2019](#).

$$F = \frac{M_0 \times 273 \times P_0 \times V}{22.4 \times (237 + T) \times P \times M \times 24} \quad (1)$$

M_0 is the mass fraction of N in N_2O or NO ($g\ mol^{-1}$), P_0 is the standard atmospheric pressure (kPa), P represents actual pressure inside the sampling vial (kPa), V is the volume increment of N_2O or NO in the sampling vial (mL), M is the quantity of the soil (kg), and T is the temperature of the bottle ($^{\circ}C$).

2.2.2 Soil inorganic nitrogen

Soil samples were collected on days 1, 3, 5, 7, 13, 18, 23, and 28 during the incubation period to determine concentrations of NH_4^+ , NO_3^- , NO_2^- concentrations. Soil inorganic nitrogen was extracted using 1 mol L^{-1} potassium chloride (KCl) solution at a 1:10 soil-to-solution ratio. The mixtures were shaken at room temperature ($25^{\circ}C$) for 30–60 min, either manually or on a mechanical shaker, to ensure sufficient extraction. After shaking, the suspensions were filtered through Whatman No. 42 filter paper or centrifuged at 4,000 rpm for 10 min to obtain the supernatant. The concentrations of NH_4^+ , NO_3^- , NO_2^- in the extracts were then determined using a continuous flow analyzer (AA3, SEAL Analytical, Germany). NO_2^- accumulative intensity was the sum of the changes in NO_2^- concentration in the soil in the soil during the incubation phase.

2.2.3 Soil nitrogen transformation-related genes

Soil samples were collected on the final day of the experiment to determine the abundance of genes related to nitrogen transformation. The absolute abundances of functional genes involved in soil nitrogen transformation—including *AOA-amoA*, *AOB-amoA*, *nirK*, *nirS*, and *nosZ*—were quantified using quantitative real-time PCR (qPCR) with fluorescence detection. DNA was extracted from 0.5 g of fresh soil using the FastDNA™ SPIN Kit for Soil (MP Biomedicals, United States) following the manufacturer's instructions. The quality and concentration of extracted DNA were assessed using a NanoDrop spectrophotometer. The primers and amplification protocols for each gene were consistent with those described by Zhu et al., 2023. Each sample was run in triplicate, and standard curves were generated using serial dilutions of plasmids containing the target gene fragments.

2.3 Statistical analysis

In the data processing stage, quality control was ensured primarily by verifying that repeated tested data did not show significant differences ($p > 0.05$) based on one-way ANOVA and exhibited a normal distribution trend. Normality tests (e.g., Shapiro-Wilk test) were conducted to confirm the data distribution, and standard deviation (SD) or coefficient of variation (CV%) was calculated to assess data consistency. The CV% values ranged from 2% to 10%, indicating an acceptable level of variability within the dataset.

The main effects analysis was performed using SPSS V25 for Windows (SPSS Inc., Chicago, IL, United States) to assess the effects of treatments. Further analysis of simple effects was conducted upon detecting significant interactions. To identify significant differences between treatment means, the least significant difference procedure

(LSD) was applied, with a significance threshold of 0.05. In this context, 'average' was defined as the arithmetic mean, and "correlation" was referred to as linear correlation (R^2).

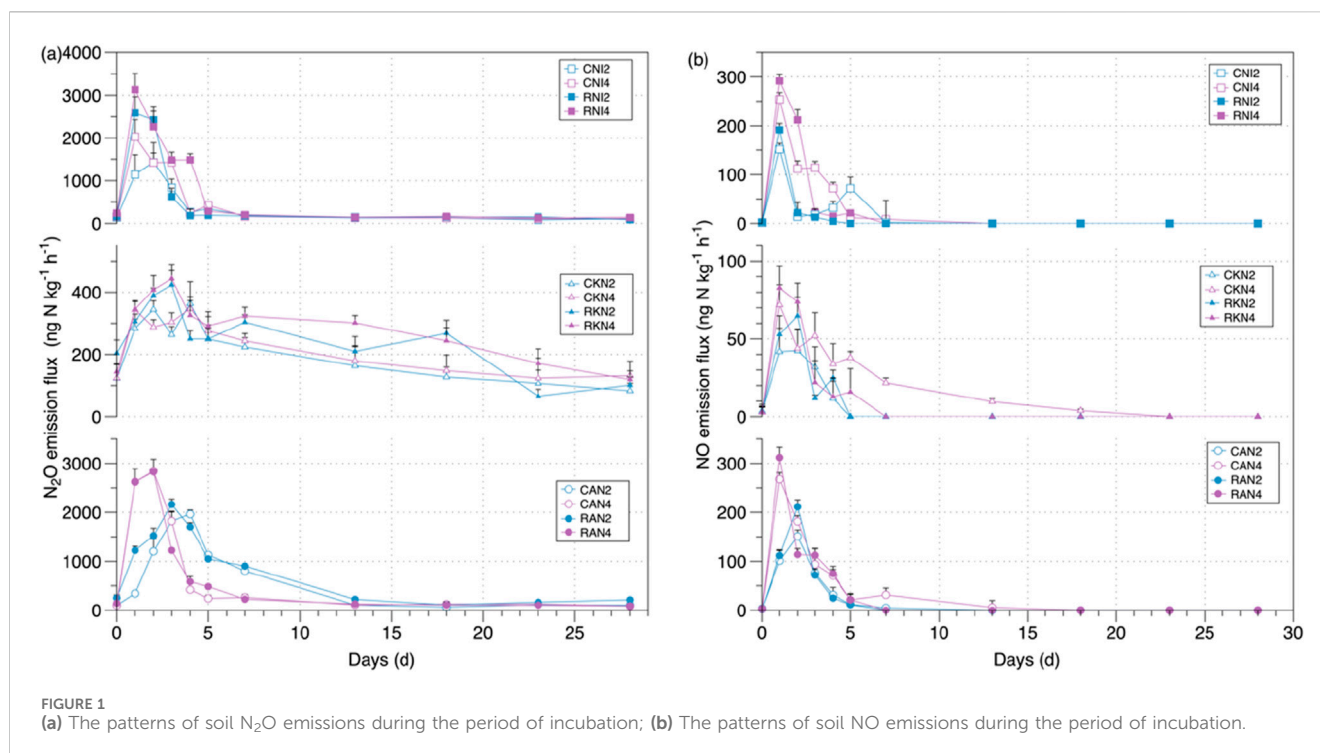
3 Results

3.1 Soil N_2O and NO emissions

After nitrogen application, N_2O emissions peak within 5 days (Figure 1a), while NO emissions reach their peak within 1–2 days after fertilization (Figure 1b), followed by a gradual decline and stabilization. Under high fertilization levels, the N_2O peak appears earlier than under low fertilization levels (Figure 1a), whereas for NO emissions, the peak timing is similar between high and low fertilization treatments (Figure 1b). As illustrated in Figure 1a, peak magnitude analysis showed that under AN and NI treatments, soil N_2O emission peaks increased with rising nitrogen application levels, with emissions consistently higher under RW irrigation compared to CW. Specifically, the peak values were 2,842.48 ng N kg^{-1} for RAN4, 3,129.30 ng N kg^{-1} for RNI4, 2,827.87 ng N kg^{-1} for CAN4, and 2027.72 ng N kg^{-1} for CNI4. For the KN treatment, peak values were similar between high and low fertilization levels, yet emissions remained higher under RW irrigation than under CW treatment. The specific peak values were 344.75 ng N kg^{-1} for CKN2, 343.73 ng N kg^{-1} for CKN4, 423.73 ng N kg^{-1} for RKN2, and 443.94 ng N kg^{-1} for RKN4.

Water quality significantly cumulative influenced soil N_2O emissions ($p < 0.05$), though it had no notable effect on NO cumulative emissions ($p > 0.05$), as shown in Table 3. The interaction between water quality and fertilizer type, water quality and fertilizer rate, as well as the three-way interaction among water quality, fertilizer type, and fertilizer rate, all had significant effects on soil N_2O emissions. In contrast, for soil NO emissions, only the interaction between water quality and fertilizer rate was significant. Simple effect revealed that under AN treatments, RW irrigation significantly increased soil N_2O and NO emissions ($p < 0.05$). In contrast, under KN treatments, RW had no significant effect on cumulative N_2O and NO emissions ($p > 0.05$). Under NI treatments, RW significantly affected only cumulative NO emissions ($p < 0.05$), with no significant impact on N_2O emissions. As illustrated in Figure 2, at a low nitrogen application rate of 200 mg N kg^{-1} , RW irrigation significantly ($p < 0.05$) increased cumulative soil NO emissions, with RAN2 showing a 25.04% increase compared to CAN2. Additionally, RNI2 significantly ($p < 0.05$) increased cumulative N_2O emissions by 27.58% compared to CNI2. At the higher nitrogen application rate of 400 mg N kg^{-1} , RAN4 also significantly ($p < 0.05$) increased cumulative N_2O emissions by 25.04% compared to CAN4.

As shown in Table 3, both the type and amount of nitrogen applied significantly affected cumulative emissions of soil NO ($p < 0.05$). As illustrated in Figure 2, cumulative N_2O emissions in the AN and NI treatments were significantly ($p < 0.05$) higher than those in the KN treatments, with N_2O cumulative emissions increasing by 62.83%–115.49% and 39.08%–52.98%, respectively, and NO cumulative emissions increasing by 97.95%–213%–93% and 30.49%–182.47%, respectively. The nitrogen application rate also significantly ($p < 0.05$) influenced soil N_2O and NO cumulative emissions. Under the AN treatment, as the nitrogen application rate



increased, soil N_2O and NO cumulative emissions significantly rose ($p < 0.05$) (Figure 2). Specifically, with RW irrigation, N_2O and NO cumulative emissions increased by 45.56% and 48.08%, respectively, while with CW irrigation, they increased by 13.96% and 107.54%, respectively. For the KN treatment, increasing nitrogen application did not significantly affect soil N_2O and NO cumulative emissions (Table 2). In contrast, under the NI treatment, increasing nitrogen application significantly increased soil NO cumulative emissions ($p < 0.05$) but had no significant effect on N_2O cumulative emissions ($p > 0.05$) (Figure 2).

3.2 Soil nitrogen

During the incubation period, the changes in soil inorganic nitrogen were similar across all treatments, as shown in Figure 3. In the AN treatment, the NH_4^+ content gradually decreased over time. In the NI and AN treatments, the NO_3^- content gradually accumulated as the incubation time increased. In the AN, KN, and NI treatments, a peak in NO_2^- content was observed at the beginning of the incubation period, with the peak magnitude increasing with the amount of fertilizer applied.

As shown in Figure 4a, the soil NH_4^+ and NO_3^- contents directly affected the soil NO_2^- content. In KN treatments, the R^2 values for RW and CW treatments were both less than 0.3, and k is less than 0.05. In AN treatments, the R^2 values for RW and CW treatments were 0.50 and 0.48, respectively, with k greater than 0.1. This indicated that the effect of NO_3^- application on increasing NO_2^- content was smaller than that of AN treatment. As shown in Figure 4b, there was a significant linear relationship between NH_4^+ and NO_3^- application rates and NO_2^- accumulation intensity. The R^2 values for RW and CW treatments were

0.99 and 0.99 for NH_4^+ , and 0.81 and 0.93 for NO_3^- , respectively. Analysis of the differences between different nitrogen treatments showed that in AN treatments, the linear slopes for RW and CW treatments were 0.89 and 0.86, respectively, while in KN treatments, they were 0.05 and 0.06. Analyzing the impact of water quality on NO_2^- accumulation intensity, the R^2 and k values showed very small differences between treatments. This could be due to the experimental setup, where the nitrogen application gradient was relatively large, lacking low nitrogen application treatments. Consequently, the impact of water quality on NO_2^- accumulation intensity was minimal.

As shown in Figure 4c, there was a strong correlation between soil NO_2^- accumulation intensity and cumulative emissions of soil NO and N_2O during NH_4^+ treatments. Analyzing the differences in the impact of soil NO_2^- accumulation intensity on NO and N_2O emissions between different water qualities, the experiment suggested that the effect of NO_2^- accumulation intensity on soil NO and N_2O emissions was greater under RW treatment compared to CW treatment ($k_{\text{RW}} > k_{\text{CW}}$). The experiment found that the impact of NO_2^- accumulation intensity on cumulative NO emissions was greater than on cumulative N_2O emissions ($k_{\text{NO}} > k_{\text{N}_2\text{O}}$). In summary, the influence of soil NH_4^+ and NO_3^- on soil NO and N_2O emissions was likely primarily mediated through changes in NO_2^- ion content. This highlighted the importance of NO_2^- ions in the pathways of nitrogen transformations leading to the production of NO and N_2O in the soil.

3.3 Soil nitrogen transformation genes

As shown in Table 3, the main effect analysis indicated that RW significantly affected the gene copy numbers of *AOA-amoA* and

TABLE 3 Significance analysis of soil cumulative N₂O/NO and soil nitrogen related transformation microbial gene copy numbers.

Influencing factor			NO	N ₂ O	AOA	AOB	NOB	nirK	nirS	nosZ
WQ			1.11 ^{n.s}	20.12 [*]	17.77 [*]	11.22 ^{**}	1.66 ^{n.s}	0.45 ^{n.s}	15.84 [*]	0.62 ^{n.s}
FT			361.23 ^{***}	102.34 ^{**}	1.31 ^{n.s}	23.33 ^{**}	12.23 [*]	7.21 [*]	0.42 ^{n.s}	4.55 ^{n.s}
FL			59.33 ^{**}	7.43 [*]	18.22 [*]	14.21 [*]	8.71 [*]	4.23 ^{n.s}	10.26 [*]	3.21 ^{n.s}
WQ × FT			4.36 ^{n.s}	35.28 ^{**}	7.75 ^{n.s}	31.12 ^{**}	11.49 ^{**}	1.77 ^{n.s}	24.12 ^{**}	0.42 ^{n.s}
WQ × FL			37.13 ^{**}	12.12 [*]	61.23 ^{***}	40.12 ^{**}	1.02 ^{n.s}	2.11 ^{n.s}	35.22 ^{**}	0.73 ^{n.s}
FT × FL			61.94 ^{**}	2.12 ^{n.s}	4.11 ^{n.s}	8.01 ^{n.s}	1.42 ^{n.s}	9.33 ^{n.s}	22.45 ^{**}	2.01 ^{n.s}
WQ × FT × FL			7.92 ^{n.s}	21.87 [*]	43.43 ^{**}	63.44 ^{**}	9.11 ^{n.s}	1.21 ^{n.s}	20.33 ^{**}	2.16 ^{n.s}
WQ		200 mg N kg ⁻¹	10.22 [*]	32.23 ^{**}	13.23 [*]	10.12 [*]	32.45 ^{**}	2.22 ^{n.s}	11.56 [*]	0.13 ^{n.s}
		400 mg N kg ⁻¹	1.02 ^{n.s}	0.12 ^{n.s}	0.74 ^{n.s}	0.03 ^{n.s}	1.11 ^{n.s}	0.43 ^{n.s}	23.04 [*]	1.04 ^{n.s}
WQ		NH ₄ ⁺	16.22 [*]	32.71 [*]	42.21 ^{**}	13.31 [*]	1.13 ^{n.s}	1.21 ^{n.s}	17.23 [*]	0.23 ^{n.s}
		NO ₃ ⁻	4.11 ^{n.s}	3.05 ^{n.s}	21.23 [*]	25.23 [*]	0.12 ^{n.s}	0.13 ^{n.s}	5.83 ^{n.s}	1.21 ^{n.s}
		NO ₂ ⁻	28.15 [*]	7.12 ^{n.s}	45.22 ^{**}	32.04 ^{**}	1.23 ^{n.s}	2.14 ^{n.s}	42.45 ^{**}	2.34 [*]
FL		NH ₄ ⁺	46.12 ^{**}	24.23 [*]	27.32 [*]	0.87 ^{n.s}	1.44 ^{n.s}	0.47 ^{n.s}	7.22 [*]	0.67 ^{n.s}
		NO ₃ ⁻	2.33 ^{n.s}	2.23 ^{n.s}	0.91 ^{n.s}	2.74 ^{n.s}	5.18 ^{n.s}	0.25 ^{n.s}	7.52 [*]	0.04 ^{n.s}
		NO ₂ ⁻	12.34 [*]	6.34 ^{n.s}	14.52 [*]	13.22 [*]	0.06 ^{n.s}	0.03 ^{n.s}	0.14 ^{n.s}	0.72 ^{n.s}
WQ	NH ₄ ⁺	200 mg N kg ⁻¹	61.23 ^{**}	7.23 ^{n.s}	72.21 ^{**}	13.31 [*]	58.31 ^{***}	1.21 ^{n.s}	27.74 [*]	0.73 ^{n.s}
		400 mg N kg ⁻¹	10.11 ^{n.s}	84.11 ^{**}	1.01 ^{n.s}	0.03 ^{n.s}	0.23 ^{n.s}	0.09 ^{n.s}	32.1 [*]	2.22 ^{n.s}
	NO ₃ ⁻	200 mg N kg ⁻¹	2.34 ^{n.s}	1.34 ^{n.s}	63.21 ^{**}	324.11 ^{***}	5.29 ^{n.s}	3.11 ^{n.s}	5.83 ^{n.s}	0.06 ^{n.s}
		400 mg N kg ⁻¹	7.12 ^{n.s}	4.95 ^{n.s}	17.3 [*]	2.32 ^{n.s}	27.25 [*]	0.14 ^{n.s}	12.22 [*]	0.04 ^{n.s}
	NO ₂ ⁻	200 mg N kg ⁻¹	5.11 ^{n.s}	44.63 ^{**}	64.51 ^{**}	71.22 ^{**}	1.22 ^{n.s}	0.83 ^{n.s}	165.21 ^{**}	6.75 [*]
		400 mg N kg ⁻¹	4.12 ^{n.s}	13.23 [*]	33.68 [*]	52.91 ^{**}	3.03 ^{n.s}	2.01 ^{n.s}	10.12 ^{n.s}	0.05 ^{n.s}

Note: n = 36, The number is F values; n.s means $p > 0.05$; * means $p < 0.05$; ** means $p < 0.01$; *** means $p < 0.001$. FT, means fertilizer type; FL, means fertilizer levels; WQ, means water quality.

AOB-amoA ($p < 0.05$), but had no significant impact on the abundance of *NOB* genes. Similar effects were observed in the interactions between water quality and fertilizer rate, as well as among water quality, fertilizer type, and fertilizer rate. Further simple effect analysis and Figure 5 revealed that under all three nitrogen types, increasing nitrogen application significantly increased the gene copy numbers of *AOA-amoA*, *AOB-amoA* and *NOB* ($p < 0.01$). Under RW irrigation in AN and NI treatments, the gene copy numbers of *AOA-amoA*, *AOB-amoA*, and *NOB* significantly increased ($p < 0.05$). In KN treatments, RKN significantly increased the gene copy numbers of *AOA-amoA* and *NOB* compared to CKN ($p < 0.05$), but had no significant effect on *AOB-amoA* gene copy numbers ($p > 0.05$). For AN treatments, at a low nitrogen application rate (200 mg N kg⁻¹), RW significantly increased the gene copy numbers of *AOA-amoA*, *AOB-amoA*, and *NOB* compared to CW ($p < 0.05$). At a high nitrogen application rate (400 mg N kg⁻¹), water quality had no significant effect on *AOA-amoA* and *AOB-amoA* gene copy numbers ($p > 0.05$). For KN treatments, water quality significantly affected *AOA-amoA* and *NOB* gene copy numbers at both nitrogen application rates ($p < 0.05$), but had no significant effect on *AOB-amoA* ($p > 0.05$). For NI treatments, water quality significantly affected *AOA-amoA* and

AOB-amoA gene copy numbers at both nitrogen application rates ($p < 0.05$), but had no significant effect on *NOB* ($p > 0.05$). As shown in Figure 5, the differences in gene copy numbers of denitrification microorganisms *nirK* and *nosZ* between different water qualities are minimal. However, under the KN and AN treatments, RW significantly increased the *nirS* gene copy numbers ($p < 0.05$). As shown in Table 3, water quality significantly affected the gene copy numbers of *nirS* ($p < 0.05$), but had no significant impact on the abundance of *nirK* and *nosZ* genes ($p > 0.05$). Neither fertilizer type nor fertilizer rate significantly influenced the abundance of *nirK*, *nirS*, and *nosZ* genes ($p > 0.05$). The interaction between water quality and other factors was found to significantly affect the gene copy numbers of *nirS* ($p < 0.05$). However, simple effect analysis showed that under the KN-2 treatment, water quality changes had no significant effect on *nirS* gene copy numbers, and changes in fertilizer rate under the NI treatment also had no significant effect. For *nosZ*, simple effect analysis revealed that in NI treatments, water quality had no significant effect on *nosZ* gene copy numbers ($p > 0.05$). However, at a low nitrogen application rate (200 mg N kg⁻¹), RW significantly increased *nosZ* gene copy numbers ($p < 0.05$).

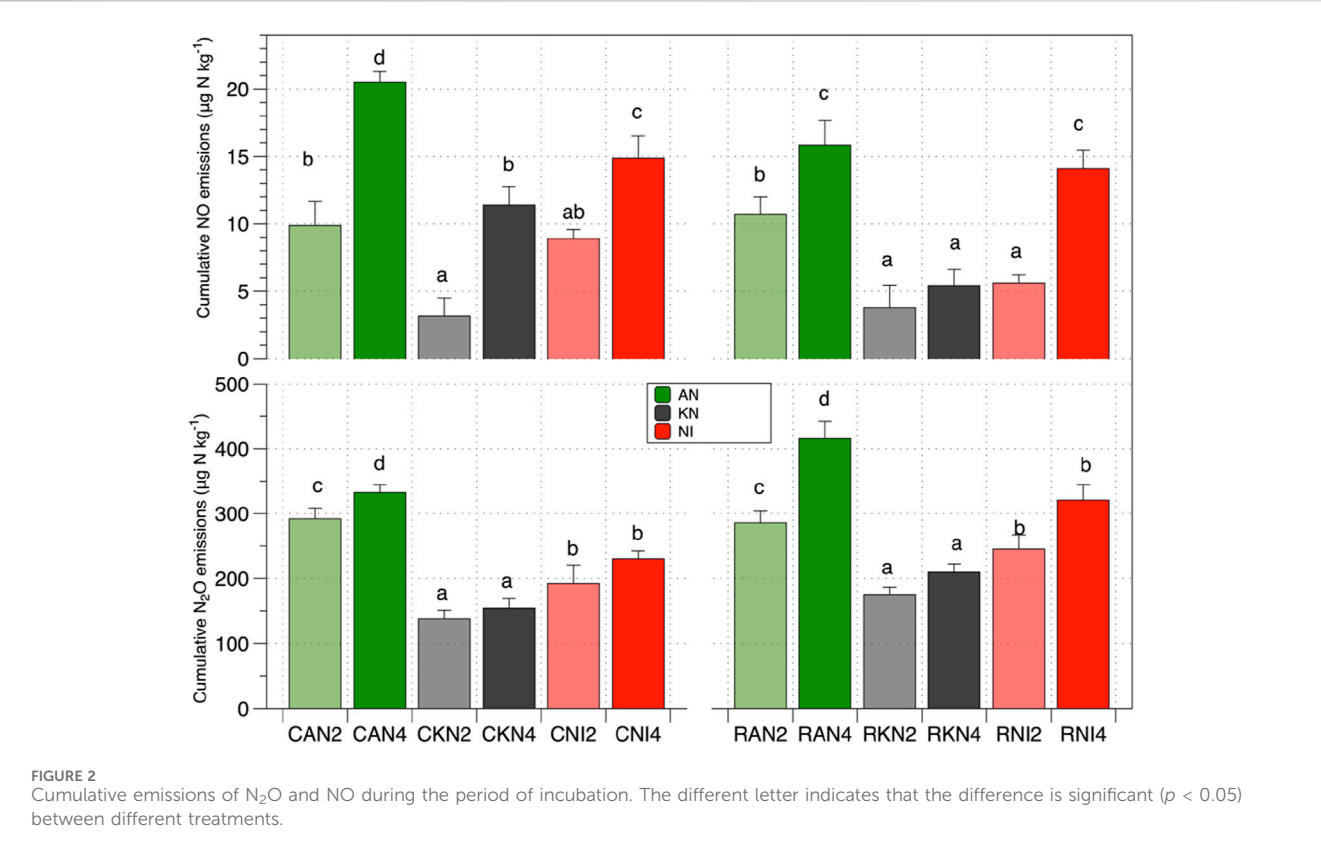


FIGURE 2 Cumulative emissions of N₂O and NO during the period of incubation. The different letter indicates that the difference is significant ($p < 0.05$) between different treatments.

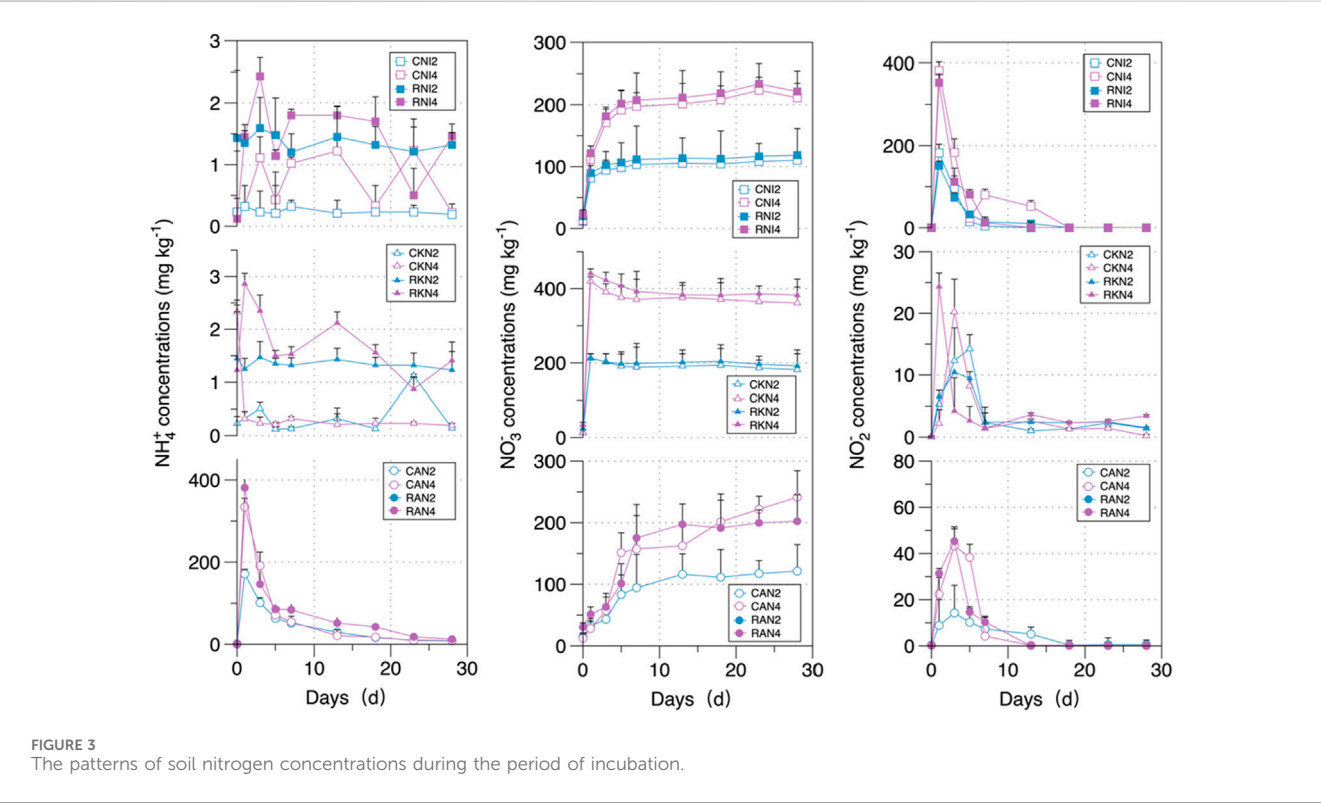


FIGURE 3 The patterns of soil nitrogen concentrations during the period of incubation.

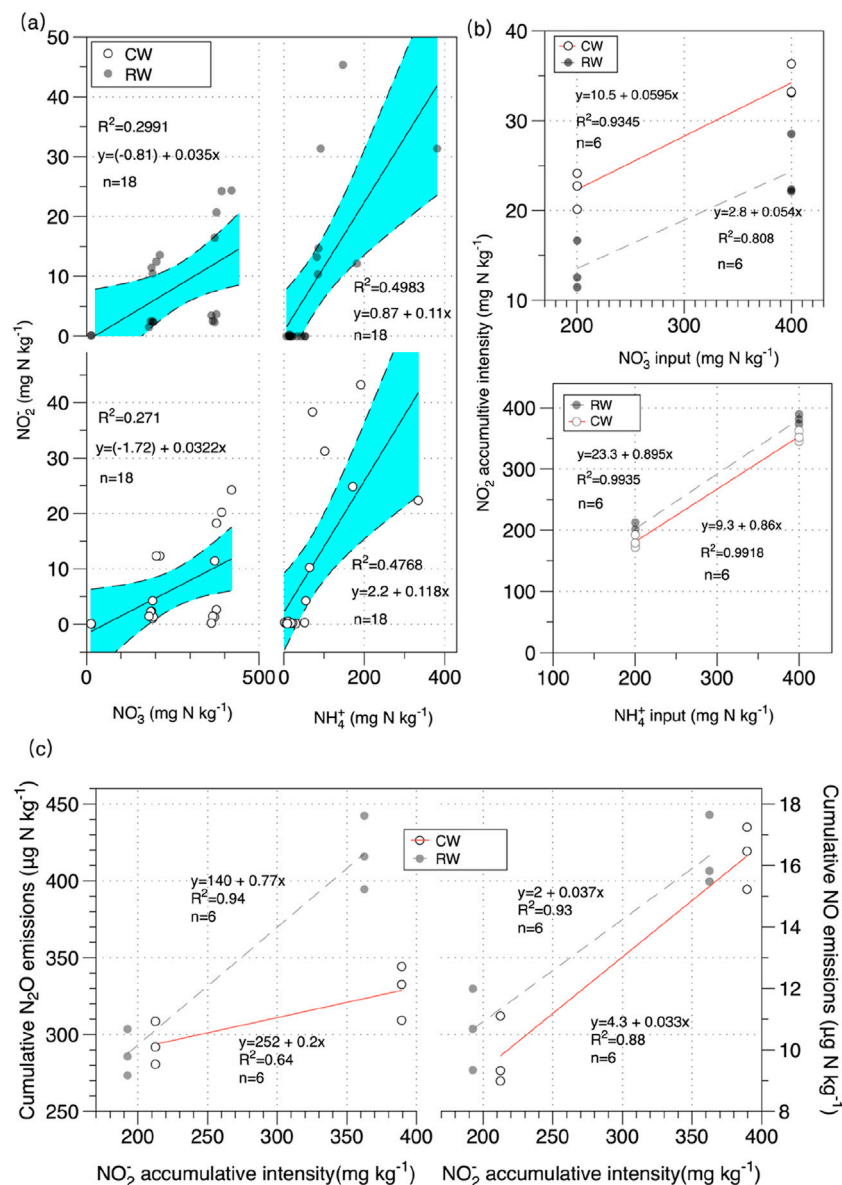


FIGURE 4

(a) The effect of NH₄⁺/NO₃⁻ concentrations on NO₂⁻ concentrations in soil under AN and KN treatments; (b) The effect of NH₄⁺ and NO₃⁻ input concentrations on NO₂⁻ accumulation intensity under different water quality conditions; (c) The effect of NO₂⁻ accumulation intensity on cumulative N₂O and NO emissions under NH₄⁺ treatments.

3.4 The relationship between nitrogen transformation genes and N₂O and NO

As shown in Figure 6, there was a significant strong correlation between soil N₂O and NO emissions, with an R^2 value greater than 0.6 ($p < 0.05$). Soil N₂O emissions were significantly correlated with the abundance of nitrifying microorganisms *AOA-amoA* and *AOB-amoA* ($p < 0.05$), with R^2 values of 0.48 and 0.71, respectively. Soil NO emissions were significantly correlated with *AOB*, with an R^2 value greater than 0.5 ($p < 0.05$). There was no significant correlation between soil N₂O and NO emissions and the abundance of denitrifying microorganisms *nirK*, *nirS*, and *nosZ*. Additionally, analysis shown in Figure 6 indicated no significant linear relationship between nitrifying and denitrifying genes, suggesting

no significant interaction between these two types of microorganisms.

4 Discussion

4.1 The effect of nitrogen ions on N₂O and NO emissions

Under conventional agricultural management, the amount of nitrogen fertilizer applied typically far exceeds the nitrogen content found in RW. In this experiment, it was observed that under the AN treatment, RW irrigation significantly increased soil N₂O emissions—a finding consistent with previous studies (Chi et al., 2023; 2024). As

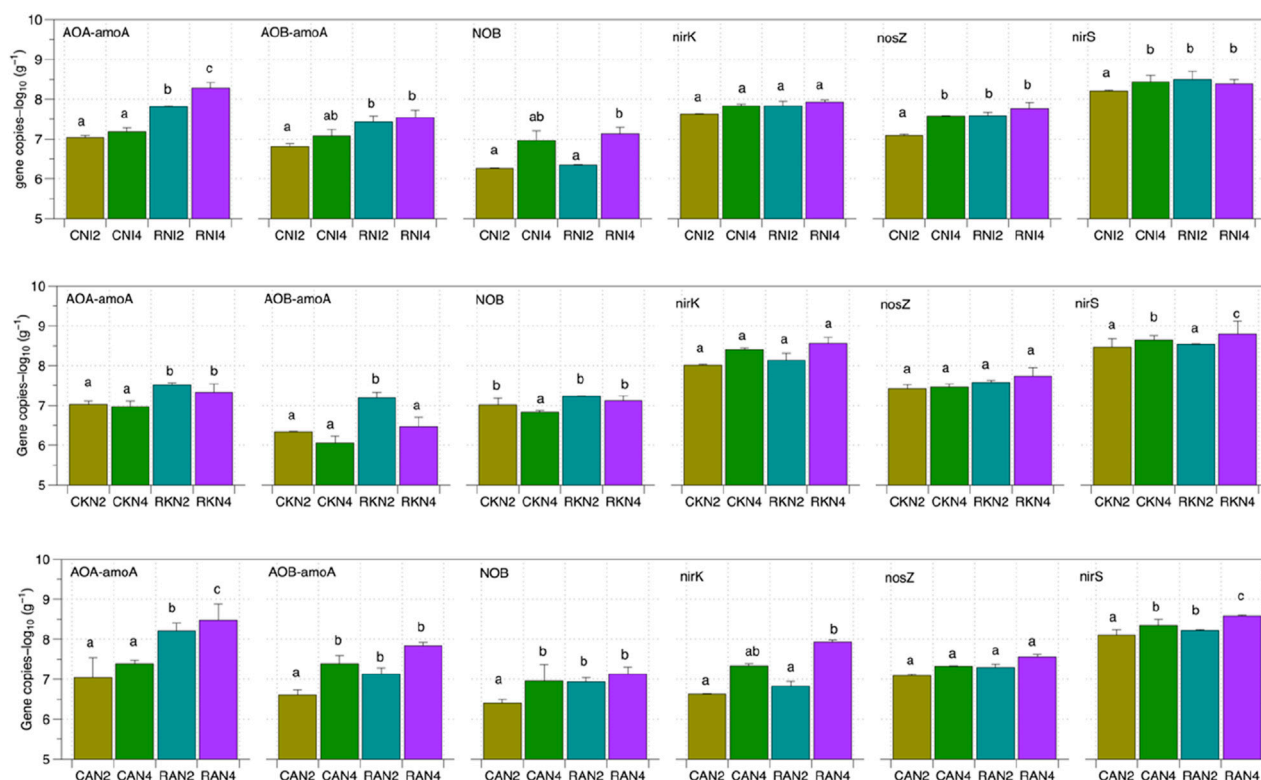


FIGURE 5

Log₁₀ values of the absolute abundance of soil nitrogen transformation genes under different nitrogen application gradients. The log₁₀-transformed gene copy numbers per g of dry soil of the targeted functional genes (AOA-amoA, AOB-amoA, NOB, nirS, nirK, and nosZ) under different treatments are shown; every treatment had six repetitions. The different letter indicates that the difference is significant ($p < 0.05$) between different treatments.

NH_4^+ application rates increased, N_2O emissions also rose (Figure 2; Hickman et al., 2014), primarily due to enhanced nitrification. Elevated NH_4^+ levels stimulated the activity of ammonia-oxidizing archaea and ammonia-oxidizing bacteria (Figure 5; Guo et al., 2017; Pei et al., 2022), thereby promoting N_2O production.

In contrast, under the KN treatment, increasing nitrogen application did not lead to a similar rise in N_2O emissions. This may be attributed to the disturbed soil incubation setup, where the availability of oxygen likely inhibited denitrification, even under high NO_3^- conditions. Among the key intermediates, NO_2^- plays a crucial role in regulating N_2O emissions. Some studies report an exponential relationship between NO_2^- concentration and N_2O emissions (Fan et al., 2020), while others indicate a strong linear relationship with NO_3^- (Chi et al., 2020b). In this study, a strong linear correlation was found between NO_2^- accumulation intensity and cumulative N_2O and NO emissions, aligning with the majority of previous findings.

Soil NO_2^- primarily originates from both nitrification and denitrification processes. The addition of NH_4^+ and NO_3^- significantly increased soil NO_2^- content (Heil et al., 2016). Our analysis showed that both AN and KN treatments enhanced soil NO_2^- levels, suggesting that the oxidation of NH_4^+ exceeded the reduction of NO_3^- . The production of N_2O and NO in soil mainly involves the formation of NH_2OH , its oxidation to NO_2^- , and subsequent reduction of NO_2^- to N_2 (Duan et al., 2019). In the NI treatment, where oxygen was abundant due to the disturbed

bottle setup, N_2O production under normal conditions would be minimal, as oxygen tends to oxidize NO_2^- to NO_3^- . However, the observed N_2O emissions may have originated from nitrifier denitrification, a process in which AOB partially oxidize NH_4^+ to NO_2^- and then reduce NO_2^- to N_2O or N_2 under high NO_2^- concentrations (Wrage et al., 2001), rather than fully oxidizing it to NO_3^- . Another possible mechanism is abiotic, non-biological N_2O formation, RW irrigation can enhance such chemical N_2O production in soils (Zhu et al., 2023). These findings underscore the importance of controlling soil NO_2^- concentrations in agricultural systems utilizing RW irrigation, as a strategy for mitigating N_2O emissions. The application of nitrification inhibitors or biochar is recommended, as both have been shown to suppress NO_2^- production and reduce N_2O emissions (Weiske et al., 2001; Di and Cameron, 2011; Fan et al., 2020).

4.2 Changes in soil nitrifying and denitrifying microorganisms under RW

Numerous studies have demonstrated that RW irrigation increases the abundance of soil nitrifying and denitrifying microorganisms (Saha et al., 2010; Ibekwe et al., 2018; Sun et al., 2018; Zhu et al., 2023). In this study, under low nitrogen application rates, RW treatments (RAN and RNI) significantly increased the abundance of microorganisms associated with AOA-amoA and nosZ



FIGURE 6

Linear correlation between nitrogen transformation genes and soil cumulative N_2O and NO emissions under RW. The \log_{10} -transformed gene copy numbers per g of dry soil of the targeted functional genes (*AOA-amoA*, *AOB-amoA*, *NOB*, *nirS*, *nirK*, and *nosZ*) under different treatments are shown. The values represent linear correlation coefficients (R^2). $n = 36$, * means $p < 0.05$; ** means $p < 0.01$; *** means $p < 0.001$.

genes. However, the influence of RW on soil microbial communities is strongly modulated by both the type and amount of nitrogen applied. At higher nitrogen application rates, the quantity of nitrogen becomes the dominant factor shaping the abundance of nitrogen transformation genes, rather than RW. Previous research supports this observation: Tian et al. (2014) found that nitrogen input significantly increased nitrifying microbial populations, while Zhu et al. (2016) reported that higher fertilizer rates enhanced both nitrifying and denitrifying bacteria. In our study, under the KN treatment with nitrogen application rates of 200 and 400 mg N kg^{-1} , RW irrigation did not significantly increase the abundance of denitrifying microbes compared to CW treatment. This may be

due to the already high NO_3^- concentrations, suggesting that shifts in NO_3^- availability have a more pronounced effect on the growth, reproduction, and community structure of denitrifying bacteria. Therefore, at 200 mg N kg^{-1} , nitrogen input itself becomes the primary driver of microbial proliferation, surpassing the influence of RW irrigation.

5 Conclusion

RW irrigation significantly increases soil N_2O and NO emissions; however, its impact is strongly influenced by the type

and amount of nitrogen fertilizer applied. When NH_4^+ fertilizers are used, RW irrigation enhances the abundance of AOA-*amoA*, AOB-*amoA*, and NOB genes in the soil, thereby accelerating the nitrification process. Notably, a significant increase in soil cumulative N_2O emissions under RW irrigation is observed only at high concentrations of NH_4^+ . In contrast, under NO_3^- fertilizers, although RW also increases the abundance of nitrification-related genes, it does not result in elevated N_2O or NO emissions. These differences between fertilizer types are closely related to the accumulation of NO_2^- . As the application rate of ammonium-based fertilizers increases, the accumulation of NO_2^- becomes more pronounced compared to nitrate-based fertilizers. Compared to CW irrigation, RW irrigation requires more careful management of ammonium fertilizer application. The use of nitrification inhibitors may be necessary under RW conditions to effectively suppress NO_2^- accumulation and its associated gaseous nitrogen emissions.

Given the temporal and environmental limitations of laboratory experiments, future studies should further explore the long-term effects of combined RW irrigation and different nitrogen sources on soil microbial community structure, functional gene expression, and nitrogen emission dynamics across various crop growth stages and soil types.

Data availability statement

The raw data supporting the conclusions of this article will be made available by the authors, without undue reservation.

Author contributions

YC: Data curation, Visualization, Writing – original draft, Writing – review and editing. CW: Data curation,

Writing – review and editing. PY: Conceptualization, Funding acquisition, Writing – original draft.

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Conflict of interest

The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

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