



Suffocating Phytoplankton, Suffocating Waters—Red Tides and Anoxia

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The dynamics of O₂ depletion in exceptional dinoflagellate blooms, often referred to as red tides or harmful algal blooms (HABs), was investigated in St. Helena Bay in the southern Benguela upwelling system in 2013. The transition to bloom decay and anoxia was examined through determination of O2-based productivity and respiration rates. Changes in O2 concentrations in relation to bloom metabolism were tracked by fast response optical sensors following incubation of red tide waters in large volume light-and-dark polycarbonate carboys. Concurrent measurements of nutrients and nutrient uptake rates served to assess the role of nutrient stressors in community metabolism and bloom mortality. The estimates of community productivity and respiration are among the highest values recorded. Nutrient concentrations were found to be low and were unlikely to meet the demands of the bloom as dictated by the rates of nutrient uptake. Ratios of community respiration to gross production were particularly high ranging from 0.6 to 0.73 and are considered to be a function of the inherently high cellular respiration rates of dinoflagellates. Nighttime community respiration was shown to be capable of removing as much as $17.34 \text{ ml} \text{ O}_2 \text{ l}^{-1}$ from surface waters. These exceptional rates of O2 utilization are likely in some cases to exceed the rate of O₂ replenishment via air-sea exchange thereby leading overnight to conditions of anoxia. These conditions of nighttime anoxia and nutrient starvation are likely triggers of cell death and bloom mortality further fueling the microbial foodweb and consumption of O_2 .

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INTRODUCTION

In coastal oceans zones of oxygen (O_2) depletion, often referred to as dead zones, are spreading with serious consequences for ecosystem functioning (Diaz and Rosenberg, 2008; Gilbert et al., 2010; Rabalais et al., 2014). The term dead zone most often applies to near-bottom coastal waters where stratification is a key requirement in the formation of hypoxia or anoxia in that it serves to separate O_2 -rich upper layers from lower less-oxygenated layers. The elevated surface O_2 concentrations are attributed to phytoplankton production, while the consequent sinking of labile organic matter to bottom waters fuels microbial respiration and the consumption of O_2 . The trend of spreading dead zones has therefore been linked in many cases to coastal eutrophication and enhanced primary production, which adds to the flow of organic matter to the seabed (Kemp et al., 2009; Rabalais et al., 2014). Increasing temperatures as dictated by climate change are also likely to promote the

extension of dead zones as the capacity of coastal waters to hold O_2 will diminish through reduced solubility, while enhanced stratification of the water column will reduce the transport of O_2 to bottom waters (Keeling et al., 2010; Gruber, 2011).

Less well-studied events of O₂ depletion in the coastal zone include incidents of complete stripping of O₂ from the water column of nearshore environments following the decay of blooms of exceptional biomass, in some cases referred to as red tides or harmful algal blooms (HABs; e.g., Vicente et al., 2001; Pitcher and Probyn, 2011). Bloom mortality follows one or another death pathway thought to be triggered in many cases by nutrient stressors (Bidle, 2014). The consequent production of large amounts of phytodetritus is considered to fuel the microbial foodweb leading to high O₂ consumption within the confined environment of the nearshore. The resulting episodic hypoxia or anoxia is often associated with major mortalities of marine life (e.g., Cockcroft et al., 2000). Numerous examples world-wide of increases in red tides or HABs attributed to increased nutrient loading of the coastal environment indicate a likely global increase in these events of episodic O₂ depletion (Anderson et al., 2002; Heisler et al., 2008). The dynamics of O₂ depletion within these high biomass nearshore blooms nevertheless remains poorly established. This limited understanding is a likely consequence of the difficulties in studying these events due to their often local and transient character and owing to the influence of air-sea exchange in determining O₂ concentrations in shallow water environments. Specifically the causes and timing of bloom mortality that may lead to a very rapid shift from net autotrophy to net heterotrophy are unknown.

In St. Helena Bay in the Benguela eastern boundary upwelling system the regular appearance of red tides provides an opportunity for investigation of the processes that serve to mediate bloom-to-post-bloom transitions important in determining the biogeochemical fate of the bloom and the consequent onset of anoxia. Here dinoflagellate blooms develop typically during late summer and early autumn and red tides are formed following inshore accumulation of blooms under conditions of downwelling (Pitcher and Probyn, 2011; Pitcher et al., 2014). Under conditions of persistent downwelling bloom decay and the resultant depletion of O_2 is the regular cause of large mortalities of marine life with major impacts on economically important marine resources, specifically the west coast rock lobster *Jasus lalandii* (Matthews and Pitcher, 1996; Cockcroft et al., 2000, 2008).

In the autumn of 2013 we investigated the transition to bloom decay and anoxia in red tide waters collected from St. Helena Bay through determination of O_2 -based productivity and respiration rates. Changes in O_2 concentrations in relation to bloom metabolism were tracked by fast response optical sensors following incubation of red tide waters in large volume light-and-dark polycarbonate carboys. These methods served our aim of assessing the role of nighttime respiration in driving the transition from oxygen saturated to anoxic conditions. Concurrent measurements of nutrients and nutrient uptake rates served to assess the role of nutrient stressors in community metabolism and bloom mortality.

MATERIALS AND METHODS

In March 2013 surface samples of red tide were collected for incubation from four localities in St. Helena Bay [15th March (32 18.312 S; 18 19.307 E), 17th March (32 20.516 S; 18 18.327 E), 18th March (32 23.895 S; 18 19.569 E), and 19th March (32 22.732 S; 18 19.197 E); Figure 1]. Water temperatures were determined at each locality using a SBE-19 Seacat CTD. Water samples of 40 l were collected by NIO bottles and transported to the laboratory in darkened containers. Subsamples for Chlorophyll a (Chla) determination, corrected for phaeopigments, were analyzed fluorometrically following extraction in 90% acetone (Parsons et al., 1984). For the analysis of phytoplankton assemblages subsamples were fixed in buffered formalin and enumerated by the Utermöhl method (Hasle, 1978). Subsamples were also filtered through Whatman GF/F filters for manual analysis of nutrients. Ammonium (NH₄⁺) was analyzed according to the methods described in Koroleff (1983), nitrate (NO_3^-) , and nitrite (NO₂⁻) followed the procedure of Nydahl (1976) and phosphate (PO_4^{3-}) and silicate (SiO_4^{2-}) that of Grasshoff et al. (1983).

Uptake rates for NH_4^+ and NO_3^- were measured using ¹⁵N incorporation according to the protocol of Dugdale and Wilkerson (1986). Incubations were carried out in 250 ml tissue culture flasks spiked with either ¹⁵NH₄Cl or Na¹⁵NO₃. Final spike concentrations were $0.1-0.2 \,\mu$ mol l⁻¹ for NH₄⁺ and 0.012- $0.2 \,\mu$ mol l⁻¹ for NO₃⁻. Flasks were incubated submerged in water recirculated through chillers for temperature control. Experiments were terminated after 2.5-3.5 h by filtration onto 25 mm Whatman GF/F filters. Filters were thoroughly rinsed under vacuum with artificial seawater to flush dissolved isotopes and inorganic C from the filter matrix and were subsequently dried at 75°C for storage. Particulate ¹⁵N enrichment, C and N concentrations were measured on a Finnegan MAT mass spectrometer (Department of Archeometry, University of Cape Town). Uptake of NH_4^+ was not corrected for regeneration and is thus underestimated. However, isotope dilution is considered to play a relatively minor role given the short incubation periods employed here. PO_4^{3-} uptake was measured as a concentration decrease in ambient concentration over time.

For determination of O₂-based productivity and respiration rates samples were incubated for approximately 24 h in large volume (101) light-and-dark polycarbonate carboys. The carboys were incubated submerged, in full sunlight, with cooling water recirculated through chiller units to maintain temperature at in situ levels. Prior to incubation initial measurements of O2 were determined in triplicate by Winkler analysis as described by Carpenter (1965). Changes in O₂ concentrations in the carboys were logged every 10 min by means of fast response RINKO-1 optical sensors (Sasano et al., 2011; Figure 2). The increasing O₂ concentrations in the light carboy during the day provided estimates of hourly net community production (hourly-NCP). Decreasing O2 concentrations in the light carboy at night provided hourly estimates of respiratory losses (*hourly-R_n*). These estimates were considerably higher than the respiratory losses observed in the dark carboy (*hourly-R_d*). Estimates of hourly-NCP expressed in terms of carbon were determined through application of a photosynthetic quotient



of 1.2. Estimates of daily gross community production (*daily-GCP*) were calculated as the sum of the daily net community production (*daily-NCP*), as determined from the increment in O₂ concentration over 24 h, and daily estimates of respiration (*daily-R* = *hourly-R_n* × 24; respiration during the day was assumed to be equal to respiration during the night [*hourly-R_n*]).

To establish the likelihood of nighttime community respiration leading to either hypoxia or anoxia estimates of O₂ replenishment via air-sea exchange were determined from the equation: F = k ([O₂]w - [O₂]s), where F is the flux of oxygen across the air-sea interface (mmol m⁻² h⁻¹), k is the transfer velocity (cm h⁻¹) and [O₂]w and [O₂]s(mmol l⁻¹) are the measured oxygen concentrations in the surface layer and its corresponding saturation concentration. The gas transfer velocity was calculated using the updated wind speed gas exchange relationship of Wanninkhof (2014): $k = 0.251 < U^2 > (Sc/660)^{-0.5}$, where U the wind speed (m s⁻¹) and Sc the Schmidt number calculated from coefficients given in Wanninkhof (2014).

RESULTS

Bloom Characteristics

Surface water temperature in St. Helena Bay on the 15, 17, 18, and 19 March 2013 ranged from 15.65° to 17.21° C (**Table 1**). A Chl*a* ocean color image generated by the NOAA/NESDIS Center



FIGURE 2 | Changes in O_2 concentrations as determined from continuous measures (10-min intervals) of oxygen in large-volume (10 I) light- (green line) and dark- (purple line) polycarbonate carboys as logged by RINKO-1 sensors from (A) 15–16 March 2013, (B) 17–18 March 2013 (C) 18–19 March 2013, and (D) 19–20 March 2013. Hourly rates of (1) net community production (*hourly-NCP*) were determined from increases in O_2 in the light bottle during the day, of (2) community respiration (*hourly-R_n*) as determined from the decrease in O_2 concentration in the light bottle at night, and of (3) net community respiration (*hourly-R_n*) as determined from the decrease in oxygen in the dark bottle.

for Satellite Applications and Research provides an indication of the bloom distribution on the 19 March 2013 (**Figure 1**). The phytoplankton assemblages within these waters were dominated by dinoflagellates ranging in concentrations from 1.64×10^6 to 17.99×10^6 cells l⁻¹ (**Table 1**). The dinoflagellates *Ceratium furca* and *Prorocentrum micans* were the most abundant species. Several other dinoflagellates were prominent including the toxic species *Alexandrium catenella*. Phytoplankton biomass as represented by Chl-*a* concentrations ranged from 100.6 to 1208.8 mg m⁻³ (**Table 1**). There was a strong correlation between dinoflagellate concentration and the biomass parameters of Chl-*a*, POC, and PON (**Figure 3A**).

Nutrients and Their Uptake

Dissolved nutrient concentrations and uptake rates for the four sample dates are given in Table 2. Nutrient concentrations were particularly low in waters of highest phytoplankton biomass, except for SiO_4^{2-} . The excess SiO_4^{2-} in all samples is consistent with a largely dinoflagellate-dominated phytoplankton community. NO₃⁻ uptake tended also to be lowest in waters of higher biomass, NH₄⁺ uptake varied independently of biomass and PO_4^{3-} exhibited relatively similar rates of uptake on each sample date. Measured uptake rates in waters of highest biomass, i.e., on 15 and 18 March 2013, were sufficient to have removed all three nutrients in under 3 and 4 h, respectively. This conclusion ignores probable recycling of both NH_4^+ and PO_4^{3-} and diffusion of NO_3^{-} but does serve to illustrate the unlikelihood of observed nutrient concentrations meeting the nutrient requirements of the blooms of highest biomass. The fact that nutrient concentrations generally decreased over the time course of the uptake experiments further indicates that regeneration rates within the same water body could not keep pace with uptake demands. The f-ratio ranged broadly from 0.14 to 0.68 indicating a variable dependence of blooms on NO_3^- .

Oxygen Production and Consumption

Observed changes in O2 concentrations in the light-and-dark carboys as measured by the RINKO-1 sensors are plotted for each of the red tide incubations (Figure 2). Productivity and respiration rates determined from these changes in O₂ concentrations are provided in Table 3. O2 concentrations in surface waters at the start of the incubations were supersaturated ranging in concentration from 6.33 to 7.54 ml l⁻¹. Hourly-NCP determined from initial increases in O2 concentration in the light carboy ranged from 1.29 to 4.96 ml O_2 l⁻¹ h⁻¹ with values increasing in waters of higher biomass and increasing dinoflagellate concentrations (Figure 3B). Values of hourly-NCP expressed in terms of carbon ranged from 576 to 2215 mg C m⁻³ h⁻¹. Assimilation indices normalized to Chl-*a* ranged from 1.83 to 7.23 mg C [mg Chl-a]⁻¹ h⁻¹ and were lowest in waters of highest biomass. Hourly- R_n as determined from declining DO concentrations in the light carboy at night ranged from 0.44 to 1.71 ml $O_2 l^{-1} h^{-1}$ and were considerably higher than the respiratory losses observed in the dark carboy (hourly- R_d) that ranged from 0.08 to 0.34 ml O₂ l⁻¹ h⁻¹. As with estimates of production respiratory losses increased in waters of higher biomass and increasing dinoflagellate cell concentrations

Date Temperature (°C)	15 March 2013		17 March 2013		18 March 2013		19 March 2013	
	15.65		16.39		17.21		16.31	
Dominant spp. (x10 ⁶ cells I ⁻¹)	C. furca	9.00	P. micans	0.72	C. furca	3.97	C. furca	0.54
	P. micans	5.45	C. furca	0.49	C. lineatum	1.53	Katodinium sp.	0.49
	A. catenella	1.08	Katodinium sp.	0.42	P. micans	1.14	Gyrodinium sp.	0.22
	C. lineatum	1.02	C. lineatum	0.29	Gyrodinium sp.	0.27	P. micans	0.21
	Total	17.99	Total	2.15	Total	7.20	Total	1.64
Chl <i>a</i> (µg l ⁻¹)	1208.8		103.8		313.8		100.6	
PON (μ mol N I ⁻¹)	1096.8		87.4		282.9		57.6	
POC (μ mol C I ⁻¹)	7762.2		634.2		2135.3		415.4	

TABLE 1 | Date of surface water collection, surface water temperature, the dominant dinoflagellate species, and cell concentrations, and estimates of Chl-a, PON, and POC concentrations.

(Figure 3B). Cell-specific rates of *hourly-Rn* ranged from 4.22 to 12.20 pmol O₂ cell⁻¹ h⁻¹ and Chl-*a* specific rates of *hourly-Rn* ranged from 0.06 to 0.23 μ mol O₂ [μ g Chl-a]⁻¹ h⁻¹ and were lowest in waters of highest cell concentrations or biomass. The total nighttime utilization of DO attributed to community respiration ranged between 5.09 and 17.34 ml l⁻¹ and was strongly correlated with dinoflagellate concentration (Figure 3C). *Daily-NCP* varied between 4.54 and 27.21 ml O₂ l⁻¹ d⁻¹, *daily-R* between 10.56 and 41.04 ml O₂ l⁻¹ d⁻¹ and *daily-GCP* between 15.10 and 68.25 ml O₂ l⁻¹ d⁻¹ all of which increased with increasing biomass. The ratio of *daily-R: daily-GCP* varied between 0.60 and 0.73.

Nighttime Depletion of O₂

The maximum *hourly*- R_n measured on 15 March 2013 translates to an O_2 consumption rate of 381 mmol $O_2\ m^2\ h^{-1}$ for a 5 m upper mixed water column. At the supersaturated O2 concentrations measured on 15 March 2013, calculated O₂ transfer across the air-sea interface following the onset of darkness would initially constitute a loss to the atmosphere, but as respiration reduces ambient O_2 levels so the flux of O_2 into the water column will increase. However, the maximum flux of 5.46 mmol $O_2 m^2 h^{-1}$, calculated at zero ambient concentrations and conditions of low wind speed (3 m s^{-1}) , remains considerably less than consumption by respiration suggesting rapid progression toward anoxia. These calculations ignore horizontal and vertical exchange of O₂. In this highly simplified system nighttime respiration could reduce ambient O2 to zero in 3.5 h. Increased wind speed will delay the onset of anoxia but only marginally (Figure 3D). At the lower hourly- R_n measured on 19 March 2013, anoxia would not develop over the night period (10 h) and wind speed effects are more pronounced (Figure 3D).

DISCUSSION

Bloom Magnitude, Composition, and Nutrient Uptake

The observed composition and cell concentrations of the red tides sampled during this study are typical of those observed in St. Helena Bay during the latter half of the upwelling season and also typify the blooms that lead to events of anoxia and consequent mortalities (Pitcher and Calder, 2000; Pitcher and Weeks, 2006). The high surface temperatures in St. Helena Bay at the time of sample collection (15.65° to 17.21°C) are indicative of the stratified conditions that develop at this time of the year when red tides are often formed by the shoreward advection and accumulation of dinoflagellate blooms under conditions of downwelling (Pitcher and Nelson, 2006). As expected under these stratified conditions nutrient concentrations were found to be low and nutrient uptake particularly during periods of very high biomass was shown to be capable of exhausting nutrients in a matter of hours. The relatively high f-ratios in some cases suggest that NO₃⁻ can be important under bloom conditions, although the bloom of highest biomass was largely NH⁺₄ dependent. Likewise Seeyave et al. (2009) have shown HABs in the southern Benguela to vary in their potential for reduced or oxidized N utilization, dependent on both species composition and state of upwelling. Somewhat contrary to expectations, most upwelling HAB taxa exhibit a comparable or greater capability to utilize NH_4^+ over NO_3^- (Kudela et al., 2010). However, nutrient utilization by HAB species remains poorly studied and clear patterns of N utilization strategies are not apparent.

Community Productivity

The estimates of productivity and respiration as determined in this study from changes in O₂ in red tide waters are among the highest values recorded. The conventional estimation of daily-GCP from initial and end point measurements in light-anddark BOD bottles as described by Gaarder and Gran (1927) assumes that respiration in the dark bottle is representative of that in the light bottle. The observations in this study of higher respiration at night in the light bottle (hourly-Rn) compared to that in the dark bottle (*hourly-Rd*) are supportive of the opinion that phytoplankton respiration is closely tied to photosynthesis and that respiratory losses are enhanced in the light bottle (e.g., Bender et al., 1987; Williams and del Giorgio, 2005). It has been shown that in the light bottle, respiration during the day is likely to be greater than that at night through stimulation of heterotrophic activity by autotrophic production (Pringault et al., 2007, 2009), while the photo-dependent degradation of organic matter during the day can also fuel enhanced bacterial



FIGURE 3 | The relationships between dinoflagellate concentration (DC) and (A) the bloom biomass parameters of Chl *a*, POC, and PON (Chl *a* = 68.7*DC-66.3 [r^2 = 0.98]; POC = 449.9*DC-522.4 [r^2 = 0.99]; PON = 63.7*DC-80.6 [r^2 = 0.98]), (B) *hourly-NCP* and *hourly-R_n* (*hourly-NCP* = 0.219*DC + 1.13 [r^2 = 0.99]; *hourly-R_n* = 0.076*DC + 0.31 [r^2 = 0.98]), and (C) nighttime utilization of O₂ (O₂ = 0.68*DC + 5.23 [r^2 = 0.97]) and (D) nighttime depletion of O₂ under the high and low respiration rates measure on the 15 and 19 March 2013 after accounting for air-sea exchange at varying wind speeds.

respiration (Gustavson et al., 2000). Furthermore, measurement of respiration during the night will result in the exclusion of the Mehler reaction and the RUBISCO oxidase function resulting in a further underestimate of respiration (Williams and del Giorgio, 2005). In the present study evidence of higher respiration during the day is provided by observations of an initial exponential decline in O_2 in the light bottle following the onset of darkness (**Figure 2**). This enhanced respiration near sundown is consistent with maximum utilization of recent photosynthate at the end of the day, and with declining substrate for respiration through the night (Markager and Sand-Jensen, 1989; Fukushima et al., 2000). Therefore, despite our use of *hourly-Rn* rather than *hourly-Rd* in the calculation of *daily-GCP* are likely to be underestimated.

While rates of production and respiration have been wellcharacterized in many coastal and oceanic communities this is not the case for the dinoflagellate-dominated blooms responsible for red tides. Of the few measurements most are based on the uptake of ¹⁴C which provides a metric close to net primary production attributed to autotrophic metabolism (Marra, 2009). Prior measurements of the productivity of red tides in St. Helena Bay include those of Walker and Pitcher (1991) and Mitchell-Innes et al. (2000) both of which assessed the productivity of blooms dominated by C. furca. Walker and Pitcher (1991) provided a productivity estimate of 517 mg C m⁻³ h⁻¹ as measured by ¹⁴C uptake in waters with C. furca concentrations of 1.3×10^6 cells l⁻¹. The Chl-*a* concentration of these waters was 139 mg m^{-3} and the corresponding assimilation index 3.7 mgC [mg Chl- a^{-1}] h⁻¹. Mitchell-Innes et al. (2000) estimated productivity by means of natural fluorescence measurements and reported values $>500 \text{ mg} \text{ C} \text{ m}^{-3} \text{ h}^{-1}$ in waters with Chl-a concentrations $> 175 \text{ mg m}^{-3}$. Other measurements of productivity of exceptional dinoflagellate blooms in the southern Benguela include those of Brown et al. (1979) for a bloom of an unknown species of Gymnodinium responsible for fish mortalities in False Bay. Their estimate of net production of $405 \text{ mg C m}^{-3} \text{ h}^{-1}$ represents the only prior measurement of the productivity of a local dinoflagellate bloom based on changes in O₂. The above estimates of productivity and biomass are similar to some of the measurements made during this study, but are considerably lower than the maximum values recorded on the 15 March 2013 (Table 1). Red tide forming dinoflagellate blooms of similar magnitude have also been reported for other eastern boundary upwelling systems (e.g., Blasco, 1975) and reported assimilation indices of 4.3 mg C [mg Chl-a]⁻¹ h⁻¹ for a bloom of Gonyaulax polyedra in the Californian upwelling system and of 5.0 mg C [mg Chl-a]⁻¹ h⁻¹ for a bloom of Gymnodinium splendens in the Peruvian upwelling system (Blasco, 1975) are similar to the assimilation indices recorded in the present study, which ranged from 1.83 to 7.23 mg C [mg Chl-a]⁻¹ h⁻¹. These measures of red tide productivity in eastern boundary upwelling systems are all considerably higher than the estimates of net community production made for Karenia-dominated blooms in coastal waters of West Florida (USA; Hitchcock et al., 2010).

Date	15 March 2013	17 March 2013	18 March 2013	19 March 2013			
NO_3^- (µmol I ⁻¹)	0.15	8.41	0.62	2.82			
NH_4^+ (µmol I ⁻¹)	0.50	13.88	0.45	0.88			
PO_4^{3-} (µmol I ⁻¹)	0.45	2.55	0.74	-			
NO_2^- (µmol I ⁻¹)	0.05	0.41	0.06	0.29			
SiO_4^{2-} (µmol I ⁻¹)	19.96	19.42	20.85	18.22			
NO_3^- uptake (µmol l ⁻¹ h ⁻¹)	0.0504	0.1497	0.1600	0.3399			
NH_4^+ uptake (μ mol l ⁻¹ h ⁻¹)	0.3057	0.8207	0.1551	0.1621			
PO_4^{3-} uptake (µmol l ⁻¹ h ⁻¹)	0.1802	0.1655	0.1584	-			
f-ratio	0.14	0.15	0.51	0.68			

TABLE 2 | Date of surface water collection, NO_3^- , NH_4^+ , PO_4^{3-} , NO_2^- , and SiO_4^{2-} concentrations, the rates of uptake of NO_3^- , NH_4^+ , and PO_4^{3-} , and the *f*-ratio.

TABLE 3 | Date of surface water collection, DO concentrations prior to incubation, hourly estimates of net community production (*hourly-NCP*), including estimates expressed in terms of carbon and normalized to Chl-*a*, hourly estimates of community respiration (*hourly-R_n*) as determined in the light carboy, hourly estimates of community respiration as determined in the dark carboy (*hourly-R_d*), cell, and Chl-*a* specific values of *hourly-R_n*, daily estimates of net community production (*daily-NCP*) as determined from the increment in DO concentration in the light bottle over 24 h, daily estimates of respiration (*daily-R_n* = *hourly-R_n* × 24; respiration during the day was assumed to be equal to respiration during the night), daily estimates of gross community production (*daily-GCP* = *daily-NCP* + *daily-R*), and the ratio of *daily-GCP*.

Date	15 March 2013	17 March 2013	18 March 2013	19 March 2013
 DO (ml I ⁻¹)	6.46	6.33	7.54	6.86
<i>hourly-NCP</i> (ml $O_2 ^{-1} h^{-1}$) [µmol $O_2 ^{-1} h^{-1}$]	4.96 [222]	1.68 [75]	2.86 [128]	1.29 [58]
<i>hourly-NCP</i> (mg C m ^{-3} h ^{-1})	2215	750	1277	576
Chl-specific <i>hourly-NCP</i> (mg C [mg Chl a] ^{-1} h ^{-1})	1.83	7.23	4.07	5.73
<i>hourly-R_n</i> (ml O ₂ $ ^{-1}$ h^{-1}) [µmol O ₂ $ ^{-1}$ h^{-1}]	1.71 [76]	0.53 [24]	0.79 [35]	0.44 [20]
<i>hourly-R</i> _d (ml O ₂ $ ^{-1}$ h^{-1}) [µmol O ₂ $ ^{-1}$ h^{-1}]	0.34 [15]	0.15 [7]	0.29 [13]	0.08 [4]
Cell-specific R_n (pmol O ₂ cell ⁻¹ h ⁻¹)	4.22	11.16	4.86	12.20
Chl-specific R_n (µmol O ₂ [µg Chl a] ⁻¹ h ⁻¹)	0.06	0.23	0.11	0.20
<i>daily-NCP</i> (ml O ₂ $ ^{-1} d^{-1}$) [µmol O ₂ $ ^{-1} d^{-1}$]	27.21 [1215]	5.96 [266]	7.17 [320]	4.54 [203]
<i>daily-R</i> (ml O ₂ $I^{-1} d^{-1}$) [µmol O ₂ $I^{-1} d^{-1}$]	41.04 [1833]	12.72 [568]	18.96 [847]	10.56 [472]
<i>daily-GCP</i> (ml $O_2 ^{-1} d^{-1}$) [µmol $O_2 ^{-1} d^{-1}$]	68.25 [3048]	18.68 [834]	26.13 [1167]	15.10 [674]
daily R: daily GCP	0.60	0.68	0.73	0.70

Community Respiration

The ratio of community respiration to gross community production widely taken in the past to be of the order 0.1 is now known to significantly underestimate the actual respiratory loss to water column production as this ratio does not typically take into account light respiration, which can be considerably higher than dark respiration (Langdon, 1993). The particularly high ratios of daily-R: daily GCP measured in the dinoflagellate-dominated waters of St. Helena Bay, which ranged from 0.6 to 0.73 are considered to be a function of the inherently high cellular respiration rates of dinoflagellates often attributed to the energetic cost of motility (Langdon, 1993; López-Sandoval et al., 2014). An additional factor influencing community respiration rates in dinoflagellate blooms is that many dinoflagellate species are mixotrophic, and utilize dissolved and particulate nutritional sources to supplement photoautotrophy (Stoecker, 1998). Also, unknown for the present study is the extent to which respiration by the phytoplankton community is augmented by bacterial decomposition of labile organic matter. Bacterioplankton are major respirers and the phasing of autotrophic and heterotrophic plankton metabolism is likely to be strongly influenced by the nature of the links between the phytoplankton and the bacterioplankton (Blight et al., 1995).

The exceptional rates of respiration measured in the present study of between 472 and $1833 \,\mu$ mol O₂ l⁻¹ d⁻¹ are placed in perspective through comparison with the reported mean respiration for surface coastal waters of 7.4 μ mol O₂ l⁻¹ d⁻¹ (Robinson and Williams, 2005) and the range of between 0.05 and 227 (mean 17.8) μ mol O₂ l⁻¹ d⁻¹ reported for 21 estuarine systems (Hopkinson and Smith, 2005). The Chl-a specific respiration rates reported for St. Helena Bay blooms of between 0.06 and 0.23 μ mol O₂ [μ g Chl-a]⁻¹ h⁻¹ are similar to those reported for dinoflagellates by Langdon (1993) of between 0.10 and 0.29 µmol O₂ [µg Chla]⁻¹ h⁻¹, which are notably higher than rates measured in other groups of phytoplankton. Cell specific rates of respiration as determined from the St. Helena Bay blooms ranged from 4.22 to 12.20 pmol O2 cell-1 h-1 and also showed good correspondence with previously reported values for

dinoflagellates including the value of 6.25 pmol O_2 cell⁻¹ h⁻¹ reported for *Ceratium tripos* in the New York Bight (Falkowski et al., 1980).

Onset of Anoxia

Large diurnal fluctuations in O2 between daytime supersaturation and nighttime hypoxia have been observed in various shallow eutrophic waters (e.g., D'Avanzo and Kremer, 1994; Shen et al., 2008; Tyler et al., 2009). In such environments the level of O₂ surplus produced during the day is important in offsetting nighttime respiration. Observations during the present study show typical daytime O2 levels within the high biomass dinoflagellate blooms of St. Helena Bay to be supersaturated because of strong photosynthetic production of O₂. However, nighttime community respiration was shown to have the potential of removing as much as $17.34 \text{ ml O}_2 \text{ l}^{-1}$ from surface waters and it is considered likely that these exceptional rates of O₂ utilization may lead overnight to conditions of anoxia. For this to occur the biological consumption of O2 needs also to exceed O2 replenishment via air-sea exchange.

O₂ fluxes across the air-sea interface are governed by the disequilibrium between the concentrations of O2 in the ocean surface layer and the atmosphere, and a rate term known as the gas transfer coefficient, which refers to the rate at which the disequilibrium is removed (Wanninkhof et al., 2009). Several environmental conditions associated with the development of red tides are likely to reduce the gas transfer coefficient thereby retarding O2 replenishment and increasing the likelihood of anoxia. A reduction of near-surface turbulence owing to the reduced winds associated with the development of red tides is likely to be a primary cause of reduced nighttime oxygenation through reduction of the gas transfer coefficient. Further, the typical warming of surface waters during bloom events will decrease the solubility of O2 in the surface waters thereby driving O₂ from these waters. Also of likely importance are phytoplankton exudates produced within the bloom which serve as potent surfactants that further dampen turbulence and suppress gas exchange (Frew et al., 1990).

Our calculations show that the maximum respiration rate measured in St. Helena Bay far exceeds the potential for oxygenation through exchange with the atmosphere. Commonly used gas exchange models such as employed here are based on empirical relationships with wind speed despite the fact that processes other than wind speed (e.g., turbulence, bubbles, waves, etc.) are known to influence air-sea gas exchange (Garbe et al., 2014). A number of these interfacial processes, however, are intimately linked to the wind, which explains the success of many of these models. Much of the uncertainty in wind speed based models resorts in the high wind regime where turbulence and bubbles can impact gas transfer velocities significantly (Wanninkhof, 2014). However, the effect of wind speed on gas transfer velocities is likely to be small when red tides are present in St. Helena Bay owing to the low wind conditions associated with their development. Under these conditions diurnal heating cycles, by forcing daytime stratification and nighttime buoyancy fluxes, may be more important than wind in facilitating gas exchange at the sea surface (McGillis et al., 2004). Despite these considerations and despite the model shortcomings and assumptions applied here, such as the absence of vertical or horizontal mixing or advection, the extraordinary community respiration rates measured in the red tides present in St. Helena Bay are very likely to overwhelm oxygenation during the night thereby driving the water column to anoxia.

CONCLUSION

Shallow coastal waters tend to have high O_2 because they are well-ventilated, despite typically higher rates of O_2 utilization. However, within red tides the extreme levels of utilization associated initially with the high rates of respiration of the dinoflagellate-dominated blooms followed by the high rates of oxidation of organic matter following bloom death are likely to exceed O_2 supply despite the relative rapidity of air-sea gas exchange. The formation of red tides during periods of relative calm and increasing waters temperatures serves to decrease both the O_2 solubility in surface waters and the rate of O_2 transfer from the atmosphere thereby increasing the likelihood of hypoxic or anoxic conditions.

It is now known that phytoplankton die spontaneously upon encountering adverse abiotic or biotic environmental conditions and various forms of autocatalytic cellular self-destruction have been identified (Bidle, 2014). Also, viruses have been shown to be intimately locked mechanistically with these cell death pathways, fueling microbial food-webs through the lysis of phytoplankton host cells. A variety of nutrient stressors are considered a common trigger for cell death and as demonstrated in this study the low nutrients in red tides relative to the high demand are a likely cause of bloom demise. However, our study further serves to demonstrate the potential role of nighttime community respiration in the development of overnight anoxia, an environmental condition also likely to trigger bloom mortality. Further, understanding of respirationregulating factors within red tides will facilitate better prediction of nighttime declines in O₂.

AUTHOR CONTRIBUTIONS

GP contributed to experimental design, field work, data analysis, and was primarily responsible for writing the paper. TP contributed to experimental design, field work, data analysis, and contributed to writing the paper.

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Conflict of Interest Statement: The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

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