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# Evaluation of *Tetradesmus obliquus* and *Monoraphidium* sp. as potential feedstocks for sustainable biodiesel production

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The escalating global demand for sustainable energy has stimulated research into renewable alternatives, with microalgae emerging as a promising feedstock for biodiesel production. This study evaluates the potential of two microalgal species, Tetradesmus obliquus and Monoraphidium sp., for efficient and sustainable biodiesel synthesis. These species were selected based on their high lipid content, favorable fatty acid profiles, and adaptability to diverse cultivation conditions. Growth optimization experiments under varying nitrogen-to-phosphorus (N:P) ratios and light intensities revealed that balanced N:P ratios (1:1) and moderate light intensity (2.6 klux) significantly enhanced biomass and lipid yields in both species. Monoraphidium sp. achieved higher dry biomass productivity (0.43 g L<sup>-1</sup>) and lipid accumulation (50.6%) than T. obliguus (0.105 g  $L^{-1}$  and 41.8%, respectively). Gas chromatography-mass spectrometry (GC-MS) analysis of fatty acid methyl esters (FAMEs) confirmed the presence of key biodiesel components, including methyl palmitate and methyl oleate, in both species. Biodiesel quality predictions using specialized software indicated that T. obliguus exhibited better oxidative stability and cold flow properties, making it more versatile for varying climates. Monoraphidium sp., however, had higher cetane numbers but poorer cold flow performance. Collectively, these findings highlight the potential of T. obliquus and Monoraphidium sp. as viable candidates for biodiesel production, emphasizing their distinct advantages depending on application requirements. Further research into scalable cultivation and cost-effective extraction methods is essential for industrial implementation.

#### KEYWORDS

biodiesel production, fatty acid methyl esters (FAMEs), microalgae, *Monoraphidium* sp., *Tetradesmus obliquus* 

## **1** Introduction

Microalgae, a diverse subset of photosynthetic organisms that play a crucial role in various ecological and industrial applications (Mofeed and Mosleh, 2013; Gurau et al., 2025). Ranging from microscopic microalgae to large macroalgae, they are fundamental to aquatic ecosystems, forming the base of food webs and contributing significantly to global oxygen production (Naselli-Flores and Padisák, 2023). Beyond their ecological importance, algae have garnered attention for their potential in biotechnology, including applications in pharmaceuticals, nutraceuticals, wastewater treatment, and notably, energy production (El Basuini et al., 2023; Samoraj et al., 2024). Their rapid growth rates, high biomass yields, and ability to produce valuable metabolites make them promising candidates for sustainable energy solutions (Bora et al., 2024).

The escalating global demand for energy, coupled with concerns over environmental degradation and climate change, has intensified the search for renewable and environmentally friendly energy sources (Sharif et al., 2024). Traditional fossil fuels are finite and their combustion releases significant amounts of greenhouse gases (Wang and Azam, 2024). In contrast, renewable energy sources such as solar, wind, and biofuels offer sustainable alternatives that can mitigate environmental impacts (Karlilar Pata and Pata, 2025). Biofuels, particularly biodiesel, have attracted interest due to their compatibility with existing diesel engines and infrastructure, as well as their potential to reduce carbon emissions (Wan Osman et al., 2024). However, the sustainability of biodiesel production depends largely on the choice of feedstock.

Microalgae are superior feedstocks for biodiesel due to their high lipid content and rapid growth rates (Abdel-Aal and Mofeed, 2015; Sharma et al., 2025). Among the various microalgal species, *Tetradesmus obliquus* and *Monoraphidium* sp. have shown particular promise. *Tetradesmus obliquus* is a unicellular green alga known for its adaptability to diverse environmental conditions and its capacity for substantial lipid accumulation (He et al., 2017). Studies have demonstrated that under nitrogendeficient conditions, *T. obliquus* can achieve lipid contents up to 43% of its dry cell weight, with a lipid yield of 2.0 g/L, making it a viable candidate for biodiesel production (Mandal and Mallick, 2009). The fatty acid profile of *T. obliquus* is predominantly composed of C16 and C18 fatty acids, which are ideal for biodiesel synthesis (Bagchi and Mallick, 2016).

Complementarily, *Monoraphidium* sp., another microalga, has been identified as a potential feedstock for biodiesel (Khoo et al., 2023). Research indicates that this species possesses a favorable lipid profile, with a significant proportion of saturated fatty acids like palmitic acid (C16:0) and unsaturated fatty acids such as oleic acid (C18:1) and linolenic acid (C18:3). These fatty acids contribute to the production of biodiesel that meets established standards, highlighting the suitability of *Monoraphidium* sp. for biofuel applications (Díaz et al., 2015).

While numerous studies have characterized microalgae for biodiesel production, most have focused on either biomass productivity or lipid content alone, often neglecting a comprehensive evaluation of fuel quality parameters under uniform culture conditions. Moreover, cold flow properties such as cold filter plugging point (CFPP), which are critical for biodiesel usability in colder climates, remain underreported for these species. There is also limited comparative data systematically examines multiple microalgal strains under standardized protocols to draw meaningful conclusions about their suitability for scalable biofuel production.

The objective of this study is to address these gaps by evaluating and comparing the growth kinetics, lipid accumulation, and full fuel property profiles including cetane number, iodine value, and CFPP of *Tetradesmus obliquus* and *Monoraphidium* sp. cultivated under identical conditions. By integrating both physiological and biodiesel quality assessments, this research contributes to a more holistic understanding of microalgal biofuel potential and provides valuable insights for future biorefinery development and strain selection.

### 2 Materials and methods

### 2.1 Microalgae culture conditions

Pure strains of Tetradesmus obliquus and Monoraphidium sp. were isolated from two different locations in the Damietta branch of Nile River, Mansoura, Egypt (31°2'25"N 31°22'58"E). Tetradesmus obliquus was cultured in Bold's Basal Medium (BBM), while Monoraphidium sp. was cultivated in Mannose Binding Lectin (MBL) medium. The selection of BBM and MBL media was based on preliminary experiments conducted to determine optimal growth and lipid accumulation conditions for each strain. BBM supported higher lipid yields in T. obliquus, whereas MBL was more effective for Monoraphidium sp. Both media were prepared using distilled water and sterilized by autoclaving at 121°C for 15 minutes. Cultures were maintained in 5-liter flasks under standard conditions: 25 ± 1°C, a 16:08 h light-dark photoperiod, and illumination of approximately 1.2 klux at the flask surface, provided by cool white fluorescent lamps, over a 28-day cultivation period. To promote homogeneity and prevent sedimentation, cultures received continuous filtered aeration and were manually mixed twice daily. Growth was monitored by daily measurements of optical density at 640 nm using a microplate spectrophotometer (Infinite 200 PRO series, TECAN). The pH of the culture medium was continuously monitored using a pH sensor (DFRobot Gravity) throughout the entire cultivation period.

# 2.2 Morphological and molecular identification

Morphological identification of the selected microalgal species was performed using both light microscopy and scanning electron microscopy (SEM) at magnifications of 1000×. Light microscopy (ZEISS PRIMO STAR BINOCULAR) was conducted to examine the general cellular morphology, including cell shape, size, and arrangement. For more detailed surface and structural analysis, scanning electron microscopy was employed using a JEOL JSM 6510 system. Sample preparation for SEM followed the protocol described by Echlin (2009). Algal cells were initially fixed in 2.5% glutaraldehyde in phosphate buffer (pH 7.2) and stored at 4°C overnight. Following fixation, samples were rinsed three times with the same buffer to remove excess fixative. A graded ethanol dehydration series was then applied, involving successive treatments with 50%, 70%, 80%, 90%, and 100% ethanol, each for 15–20 minutes. The final dehydration step used absolute ethanol to ensure complete removal of water. A drop of the dehydrated sample was placed on a carbon-coated copper grid and air-dried using a Sputter Coating Evaporator (SPI Module – Sputter Carbon/Gold Coater) to create a conductive coating. The prepared specimens were then examined under the SEM to obtain high-resolution images for taxonomic comparison with established morphological descriptions.

For molecular identification, genomic DNA was extracted from the algal biomass using a modified CTAB protocol (Hossen et al., 2025). Amplification of the 18S rRNA gene was carried out using PCR with universal eukaryotic primers: the forward primer (5'-AACCTGGTTGATCCTGCCAGT-3') and the reverse primer (5'-CCTTGTTACGACTTCACCTTCC-3'). The PCR mixture (25 µL final volume) consisted of 2.5 µL of 10× Taq polymerase buffer containing 1 mM MgCl<sub>2</sub>, 2 µL of 2.5 mM dNTPs, 1 µL of each primer (10 pmol  $\mu$ L<sup>-1</sup>), 0.2  $\mu$ L of Taq DNA polymerase (5 U  $\mu$ L<sup>-1</sup>), and ultra-pure water. Amplification was performed using a PTC-100 thermal cycler under the following conditions: initial denaturation at 94°C for 5 minutes, followed by 35 cycles of 94°C for 30 seconds, 52°C for 30 seconds, and 72°C for 1 minute, with a final extension at 72°C for 10 minutes. PCR products were resolved on a 1.5% agarose gel in 0.5× TBE buffer at 120 V for 30 minutes, stained with 0.5 µg cm<sup>-3</sup> ethidium bromide, and visualized using a Syngene gel documentation system. Fragment sizes were assessed using a DNA molecular weight marker ranging from 100 to 5000 bp (Fisher, Canada). Amplified DNA fragments were purified using a PCR clean-up column kit (Maxim Biotech Inc., USA) according to the manufacturer's instructions, involving membrane binding, ethanol-based washing, and elution. The purified PCR products were stored at -20°C until further analysis.

Sequencing of the purified products was conducted using the forward primer and the BigDye<sup>®</sup> Terminator v3.1 Cycle Sequencing Kit (Applied Biosystems, Foster City, CA, USA) on a 3130xl Genetic Analyzer (Applied Biosystems). The resulting nucleotide sequences were subjected to Basic Local Alignment Search Tool (BLAST) analysis against the NCBI GenBank database to confirm the identity of the isolates. Multiple sequence alignments were carried out using CLUSTALW version 1.83 (Chenna et al., 2003), and phylogenetic relationships were inferred using the Unweighted Pair Group Method with Arithmetic Mean (UPGMA) implemented in MEGA version 12 (Kumar et al., 2018), allowing for the placement of the studied isolates within a broader evolutionary context.

# 2.3 Growth optimization and biomass determination

To determine the optimal conditions for biomass and lipid production, the growth performance of *Tetradesmus obliquus* and

*Monoraphidium* sp. was evaluated under different nitrogen-tophosphorus (N:P) mass ratios (1:1, 1:4, 1:0.5, and 4:1[control]) and varying light intensities (1.2 [control], 2.6, 3.0, and 4.2 klux). The microalgae were cultivated in 250 mL Erlenmeyer flasks containing 100 mL of Bold's Basal Medium (BBM) for *T obliquus* and Mannose Binding Lectin (MBL) medium for *Monoraphidium* sp., respectively. Each flask was inoculated with 20 mL of a twoweek-old algal culture with an initial dry biomass of approximately 0.05 g L<sup>-1</sup>. The cultures were maintained under controlled conditions ( $25 \pm 1^{\circ}$ C) with a 16:8 h light: dark photoperiod for a total duration of seven weeks. Growth was monitored daily through direct cell counting using a ZEISS PRIMO STAR binocular light microscope and by measuring optical density at 640 nm using a microplate spectrophotometer (Infinite<sup>®</sup> 200 PRO series, TECAN Group Ltd., Switzerland).

Biomass yield was quantified at the end of the cultivation period by centrifuging the cultures at 4000 rpm for 10 minutes, followed by drying the algal pellets at 60°C until a constant weight was achieved. Biomass was expressed as dry weight (mg DW L<sup>-1</sup>), and the specific growth rate ( $\mu$ , d<sup>-1</sup>) was calculated using the formula:

$$\mu = \frac{ln(W_t - W_0)}{d}$$

Where  $W_0$  is the initial biomass,  $W_t$  is the final biomass, and d is the incubation duration in days.

To determine the effect of nutrient availability, different N:P mass ratios (1:4, 1:1, 1:0.5) were tested against a control ratio of 4:1. Cultures were incubated for four weeks under these conditions, and biomass concentration and lipid content were evaluated. Additionally, the impact of light intensity on algal growth was investigated using a range of illumination levels:  $1.2 \pm 0.2$  klux, 2.6  $\pm 0.2$  klux (control),  $3.0 \pm 0.2$  klux, and  $4.2 \pm 0.2$  klux, provided by cool white fluorescent tubes. Following four weeks of cultivation, dry weight biomass and lipid productivity were measured to assess the influence of illumination. All growth experiments were conducted in triplicate to ensure reproducibility and statistical reliability.

# 2.4 Lipid content, extraction, and GC–MS analysis

Total lipid content was determined using the Soxhlet solvent extraction method. Dried algal biomass (0.5 g) was mixed with a chloroform-methanol (2:1 v/v) solution and sonicated at 40 kHz for 10 minutes to disrupt cell walls. The resulting mixture was centrifuged at 6,000 rpm for 10 minutes, and the organic phase containing lipids was carefully collected. Extracted lipids were weighed to calculate lipid yield as a percentage of dry biomass (Dayananda et al., 2005). For fatty acid analysis, lipid extracts were subjected to gas chromatography-mass spectrometry (GC-MS) to identify and quantify the fatty acid profile. Analysis was carried out using an Agilent GC-MS system equipped with a capillary column (e.g., HP-88), following standard temperature programming. Identification was based on retention times and mass spectra compared to FAME standards.

# 2.5 Saponification and esterification of lipid extracts

The extracted lipids were converted to fatty acid methyl esters (FAMEs) using a two-step transesterification process (Hartman and Lago, 1973). First, saponification was performed by heating 50 mg of lipid with 2 mL of 0.5 M KOH in methanol at 60°C for 1 hour. After cooling, esterification was carried out by adding 2 mL of 14% boron trifluoride in methanol and heating the mixture again at 60°C for 15 minutes. After the reaction, FAMEs were extracted using hexane, dried over anhydrous sodium sulfate, and stored at  $-20^{\circ}$ C until analysis.

# 2.6 Fatty acid profiling and biodiesel characterization

The composition of fatty acid methyl esters (FAMEs) derived from algal lipids was analyzed using GC-MS. Analysis was performed with a Thermo Scientific Trace GC Ultra system coupled with an ISO Single Quadrupole Mass Spectrometer. Separation was achieved using a TG-5MS fused silica capillary column (30 m  $\times$  0.25 mm  $\times$  0.1 µm film thickness), with helium as the carrier gas at a constant flow rate of 1.0 mL/min. The GC injector and MS transfer line were maintained at 280°C, and electron ionization was carried out at 70 eV. The oven temperature program initiated at 150°C, held for 4 minutes, and ramped to 280°C at a rate of 5°C/min, followed by a final hold for 4 minutes. Samples (1 µL) were injected in split mode. The identification of individual FAMEs, ranging from C14 to C22, was based on comparison of retention times and mass spectral data with entries in the NIST and WILLY libraries integrated into instrument software. Quantification was performed using the relative peak area percentage of each compound.

Following fatty acid profiling, the physicochemical properties of the produced biodiesel were predicted using the Biodiesel Analyzer software, a specialized computational tool developed by Talebi et al. (2014) and validated for assessing biodiesel fuel quality based on fatty acid composition. This software estimated key fuel parameters including density, kinematic viscosity, cetane number, iodine value, acid value, flash point, and pour point. The predicted values were further compared against international biodiesel standards such as ASTM D6751 and EN 14214 to evaluate the suitability of the algalderived biodiesel for practical fuel applications.

### 2.7 Statistical analysis

All experimental data were analyzed using one-way analysis of variance (ANOVA) to evaluate significant differences among treatments. Duncan's multiple range test was employed to separate means at a confidence level of p < 0.05. Statistical analyses were conducted using SPSS software (version 20, IBM, USA).

## **3** Results

# 3.1 Morphological and molecular identification

*Tetradesmus obliquus* exhibited a typical colonial morphology under light microscopy, forming coenobia of four to eight elongated cells with straight margins and rounded apices. The cells were arranged linearly or slightly curved, and the cell wall appeared smooth without visible ornamentation. SEM analysis confirmed the linear arrangement and revealed more defined ultrastructural details, including the smooth surface and compact coenobial structure (Figure 1).

*Monoraphidium* sp. displayed a unicellular form with a crescentshaped cell morphology and tapering ends. Light microscopy revealed individual cells dispersed in the medium, while SEM provided further details on cell curvature and surface texture. The cells lacked visible spines or ornamentation and had a smooth exterior, indicative of the genus (Figure 2).

Molecular identification based on 18S rRNA gene sequencing confirmed the taxonomic status of both isolates. The nucleotide sequence products of algal DNA were analyzed using NCBI-BLAST



for confirming the sequences. The obtained sequences of the two algal isolates were submitted to the GenBank and were given accession numbers, *Tetradesmus obliquus* (GenBank accession no. PV361776) and *Monoraphidium* sp. (GenBank accession no. PV300550). Phylogenetic analysis placed the isolates firmly within their respective clades, corroborating morphological findings (Figure 3).

### 3.2 Growth performance

The growth kinetics of *T. obliquus* and *Monoraphidium* sp. were evaluated under standard conditions  $(25 \pm 2^{\circ}C, 1.2 \text{ klux}, 16:8 \text{ h light/} dark cycle)$  are presented in Table 1. The comparative analysis of *Tetradesmus obliquus* and *Monoraphidium* sp. over a 28-day cultivation period revealed marked differences in biomass accumulation, lipid productivity, and growth kinetics.

*Monoraphidium* sp. demonstrated superior performance in dry biomass yield, reaching a maximum of  $0.177 \pm 0.0078$  g L<sup>-1</sup> at day 28, compared to  $0.047 \pm 0.0025$  g L<sup>-1</sup> recorded for *T. obliquus* at the same time point. Lipid content in *Monoraphidium* sp. also showed a significant increase over time, peaking at  $0.368 \pm 0.049$  g g<sup>-1</sup> on day 28, while *T. obliquus* achieved a maximum of  $0.213 \pm 0.028$  g g<sup>-1</sup>.

The specific growth rate ( $\mu$ ) was generally higher in *Monoraphidium* sp., with the highest rate observed on day 21 (0.132  $\pm$  0.019 d<sup>-1</sup>), slightly declining by day 28 (0.123  $\pm$  0.015 d<sup>-1</sup>), whereas *T. obliquus* exhibited its peak growth rate earlier, at day 14 (0.113  $\pm$  0.002 d<sup>-1</sup>). Correspondingly, the growth doubling per day (Dd<sup>-1</sup>) followed a similar pattern, being higher in *Monoraphidium* sp. (0.191  $\pm$  0.027 d<sup>-1</sup> on day 21) than in *T. obliquus* (0.163  $\pm$  0.003 d<sup>-1</sup> on day 14). Generation time (G) was shortest in *Monoraphidium* sp., particularly on day 21 (5.23  $\pm$  0.73 days), while *T. obliquus* required more time to double its population, especially on day 21 (10.50  $\pm$  1.75 days).







#### FIGURE 3

Phylogenetic tree of the two algal isolates identified using 18S rRNA gene universal primers, in comparison with related sequences retrieved from GenBank. The two characterized isolates were *Tetradesmus obliquus* (Accession No. PV361776) and *Monoraphidium* sp. (Accession No. PV300550).

Algae strain	Days	Biomass yield (g.L <sup>-1</sup> )	Lipid production $(g.g^{-1})$	μ	Dd⁻¹	G
	14	$0.017 \pm 0.0037$	$0.043 \pm 0.008$	0.113 ± 0.002	$0.163 \pm 0.003$	6.15 ± 0.12
Tetradesmus obliquus	21	$0.020 \pm 0.008$	$0.122 \pm 0.013$	0.066 ± 0.009	$0.095 \pm 0.014$	10.50 ± 1.75
	28	$0.047 \pm 0.0025$	0.213 ± 0.028	$0.085 \pm 0.006$	$0.122 \pm 0.008$	8.17 ± 0.58
	14	0.033 ± 0.0061	$0.061 \pm 0.007$	$0.127 \pm 0.042$	$0.184\pm0.061$	5.44 ± 1.78
Monoraphidium sp.	21	$0.086 \pm 0.0017$	$0.167 \pm 0.024$	0.132 ± 0.019	$0.191 \pm 0.027$	5.23 ± 0.73
	28	$0.177 \pm 0.0078$	0.368 ± 0.049	0.123 ± 0.015	$0.177 \pm 0.022$	5.65 ± 0.71

TABLE 1 Dry biomass yield(g/L), lipid production (g.g<sup>-1</sup>), specific growth rate ( $\mu$ ), growth doubling per day (Dd<sup>-1</sup>), and generation time (G) of *Tetradesmus obliquus* and *Monoraphidium* sp.

# 3.3 Optimization of biomass yield and lipid accumulation

The impact of varying nitrogen to phosphorus (N:P) ratios on dry biomass and lipid content was evaluated for *Tetradesmus obliquus* and *Monoraphidium* sp., as illustrated in Figure 4. In *Tetradesmus obliquus*, the dry biomass yield was significantly influenced by the N:P ratio (Figure 4A). The highest dry biomass (0.065 g L<sup>-1</sup>) was observed at an N:P ratio of 1:1, followed by 4:1 and 1:0.5, with the lowest biomass (0.038 g L<sup>-1</sup>) recorded at the 1:4 ratio. Statistically, the 1:1 ratio was significantly higher than all other treatments (p< 0.05), except 4:1, which showed intermediate values. For *Monoraphidium* sp., dry biomass followed a different trend (Figure 4B). The 4:1 N:P ratio resulted in the highest biomass



FIGURE 4

Effect of nitrogen to phosphorus (N:P) ratios on dry biomass and lipid content in *Tetradesmus obliquus* (A, C) and *Monoraphidium* sp. (B, D). Different letters indicate significant differences (p< 0.05).

production (0.085 g  $L^{-1}$ ), which was significantly greater than that of all other treatments (p < 0.05). The lowest biomass (0.054 g L<sup>-1</sup>) was found at the 1:4 N:P ratio. In terms of lipid content, Tetradesmus obliquus showed a significant increase at the 1:1 N:P ratio, reaching 34.8% (Figure 4C). All other N:P ratios (1:4, 1:0.5, and 4:1) resulted in significantly lower lipid percentages (19.2-22.5%) and did not differ significantly among each other (p > 0.05). Similarly, Monoraphidium sp. exhibited the highest lipid content (47.3%) at the 1:1 N:P ratio (Figure 4D), which was significantly greater than all other treatments (p < 0.05). The other ratios showed moderate lipid contents ranging between 34.6% and 37.2%, without statistically significant differences between them. These results demonstrate that optimal N:P ratios differ between species for biomass productivity, while a balanced 1:1 N:P ratio generally favors lipid accumulation in both Tetradesmus obliguus and Monoraphidium sp.

The effects of different light intensities (1.2, 2.6, 3, and 4.2 klux) on dry biomass production and lipid content were examined in

Tetradesmus obliquus and Monoraphidium sp., as shown in Figure 5. For Tetradesmus obliguus, dry biomass was significantly affected by light intensity (Figure 5A). The highest dry biomass yield  $(0.105 \text{ g L}^{-1})$ was recorded at 2.6 klux, followed by 3 and 4.2 klux, which showed moderate but significantly lower values. The lowest biomass (0.056 g L<sup>-1</sup>) was observed at 1.2 klux. Statistical analysis indicated significant differences among all treatments (p < 0.05), with the exception of 3 and 4.2 klux, which were not significantly different from each other. In Monoraphidium sp., light intensity had a marked effect on biomass accumulation (Figure 5B). Maximum dry biomass  $(0.42-0.43 \text{ g L}^{-1})$ was observed at 3 and 4.2 klux, with no significant difference between these two intensities. A moderate biomass yield was recorded at 2.6 klux (0.32 g  $L^{-1}$ ), while the lowest value (0.14 g  $L^{-1}$ ) was at 1.2 klux, significantly different from all other treatments (p< 0.05). Lipid content in Tetradesmus obliquus also responded significantly to light intensity (Figure 5C). The highest lipid percentage (41.8%) was obtained at 2.6 klux, which was significantly higher than the 1.2 klux treatment (34.1%) and not significantly different from the 3 klux level. The



FIGURE 5

Effect of light intensity (klux) on dry biomass and lipid content in *Tetradesmus obliquus* (A, C) and *Monoraphidium* sp. (B, D). Different letters indicate significant differences (p< 0.05).

lowest lipid content was observed at 1.2 klux, while 4.2 klux produced a moderate lipid content (38.5%), significantly different from both the highest and lowest values. Similarly, *Monoraphidium* sp. showed the greatest lipid content (50.6%) at 2.6 klux (Figure 5D), significantly higher than the 1.2 klux condition (44.2%) and comparable to the 3 klux treatment. The lowest lipid yield was recorded at 1.2 klux, while 4.2 klux showed intermediate values (46.8%). Overall, 2.6 klux light intensity provided optimal conditions for lipid accumulation in both species, while biomass production was maximized at higher intensities (2.6–4.2 klux), with species-specific responses.

### 3.4 Biomass composition

GC-MS analysis of the chloroform-methanol extracts of *Tetradesmus obliquus* and *Monoraphidium* sp. revealed a diverse array of bioactive and lipid-derived compounds (Figure 6). In *Tetradesmus obliquus*, the dominant compounds included

hexadecanoic acid, methyl ester (methyl palmitate), 9-octadecenoic acid, methyl ester (methyl oleate), and octadecanoic acid, methyl ester (methyl stearate). Additionally, notable quantities of 1,2benzenedicarboxylic acid, diisooctyl ester, a phthalate derivative, and hydrocarbons such as 2,4-dimethylheptane were identified (Table 2). Similarly, *Monoraphidium* sp. exhibited a comparable compound profile, with methyl palmitate, methyl oleate, and methyl stearate again being the dominant constituents (Table 3). Other significant shared compounds included 1,2-benzenedicarboxylic acid, diisooctyl ester and branched alkanes like 2,6,10-trimethylpentadecane.

### 3.5 Biodiesel composition

GC-MS analyses of the biodiesel derived from *Tetradesmus obliquus* and *Monoraphidium* sp. revealed the presence of a range of fatty acid methyl esters (FAMEs), which are essential indicators of biodiesel quality (Figure 7).



InSchulp and schule	No	Chemical molecule		Retention Time (Min)	Peak Area %	Chemical group
1222300000133	1	2,4,6,8,10-Tetradecapentaenoic acid, 9a-(acetyloxy)-1a,1b,4, 4a,5,7a,7b,8,9,9a-deca hydro- 4a,7b-dihydroxy3-(hydroxymethyl)-1,1,6,8-tetramethyl-5-oxo-1 H-cyclopropa[3,4]benz[ 1,2-e] azulen-9-yl ester	$C_{36}H_{46}O_8$	5. 47	0.12	Carboxylic Acid Ester
1SpeakenCarlageG. 4.O. 4.More4Acaucal status Algorithm and algorithm	2	(2,2-Dibenzyloxy-3-nit ro-5,10,15,20-tetraphen yl-2,3-dihydroporphyri nato)copper(II)	C <sub>58</sub> H <sub>40</sub> CuN <sub>5</sub> O <sub>2</sub>	5. 57	0.08	Ether
4CalcalactionCallactionCallactionCallor	3	Spherodenon	$C_{41}H_{58}O_2$	6. 64	0.12	Alcohol Keton
s1SchulpschultschultergenergiescherSchultschultergenergiescher<	4	(2-Acetamido-3-nitro-5,10,15,20-tetraphenylpo	C46H30CuN6O3	7.60	0.10	Nitro Group
6CMACHALACIMENTALSTERTANCmmmm	5	(2,3-Dihydro-5,10,15,2 0-tetraphenyl[2-(2)H1] prophyrinato)copper(II)	C <sub>44</sub> H <sub>29</sub> DCuN <sub>4</sub>	8. 76	0.10	porphyrin ring
7SkalessyngenSkal	6	GLYCOCHOLIC ACID METHYL ESTER TMS	C <sub>36</sub> H <sub>69</sub> NO <sub>6</sub> Si <sub>3</sub>	8. 80	0.08	Ester
8InterfaceInte	7	2-bis(ethoxycarbonyl)m ethyl-9(2,3,5-tri-O-(2-m ethylprop-2-yl)dimethyl silyloxy-á-D-ribofuran osyl)purine	$C_{35}H_{64}N_4O_8Si_3$	19.57	0.08	Ester
9Si-Bisloid-eiterburg 4-scale-bisloid	8	Lucenin 2	$C_{27}H_{30}O_{16}$	27.89	0.11	Alcohol Keton
10Independent of the second secon	9	5,5'-Bis(3,5-di-tert-buty l-4-oxo-2,5-cyclohexad ien-1-ylidene)-5,5'-dihy dro-2,2'-biselenophene	$\mathrm{C_{36}H_{44}O_2Se_2}$	28.05	0.09	Keton Hydrocarbon
11Docume (AS)0.0140.0140.0140.0140.014120.010000000000000000000000000000000000	10	Dimethyl 2,anti-4,anti-9,12,anti14-pentabromodecacyc lo[9.9.0.0(1,8).0(2,12). 0(6,10).0(11,18)0 (13,17).0(16,20)] icosane-syn-4,syn-9-di carboxylate	C <sub>24</sub> H <sub>23</sub> Br <sub>5</sub> O <sub>4</sub>	28.95	0.11	Ester
125000000000000000000000000000000000000	11	Docosane (CAS)	$C_{22}H_{46}$	29.73	0.31	Hydrocarbon
11Diadeane, 24, 91, 11-heamently (CAS)ClasselClasselOlisionOlisionOlisionOlisionOlisionOlisionOlision149-Cricosen, CO, CAS)Coladean-12-Jocia cial, methyl extenderClasselClasselClasselOlisionOlisi	12	5-(Dibromomethyl)-1,3 -bis(tribromomethyl)benzene	$C_9H_4Br_8$	30.18	0.08	benzene ring
149riconence, (CASM)0.1780.07010400159riconence, (CASM)0.13180.1510.1510169riconence, (CASM)0.12400.1310.1310.131179riconence, (CASM)0.20400.20400.20400.20400.2040189riconence, (CASM)0.20400.20400.20400.20400.20400.2040109riconence, (CASM)0.20400.20400.20400.20400.20400.2040109riconence, (CASM)0.20400.20400.20400.20400.20400.2040109riconence, (CASM)0.20400.20400.20400.20400.20400.2040119riconence, (CASM)0.20400.20400.20400.20400.20400.2040129riconence, (CASM)0.20400.20400.20400.20400.20400.2040139riconence, (CASM)0.20400.20400.20400.20400.20400.2040149riconence, (CASM)0.20400.20400.20400.20400.20400.2040159riconence, (CASM)0.20400.20400.20400.20400.20400.20400.2040159riconence, (CASM)9riconence, (CASM)0.2040	13	Dodecane, 2,2,4,9,11,11-hexamet hyl- (CAS)	$C_{18}H_{38}$	31.66	0.33	Hydrocarbon
159-Octadecen-1-project aid, methyl ester16-list16-list9-Ister169-Methyl-Z-10-terradec an-1-ol acetateC2,H3,Q013.4710.13Faster179-EcosyneC2,gH3,a32.690.10Hydrocarbon1816-anicoaan, 11-(1-ethylproph)-C2,gH3,a32.690.15Hydrocarbon105-Ali-1-terramethyl-2-hexadecen-1-olC2,gH4,a32.690.15Hydrocarbon1116-anicoaci ad, methyl ester (CAS)C2,gH4,a33.800.15Garbonylic12s-5.8,11,14,17-Eicosa pentenoic acidC2,gH3,0035.810.15Garbonylic12s-6xdecnoic acid (CAS)CarbonylicC3,gH3,0035.810.15Garbonylic13s-6xdecnoic acid (CAS)CarbonylicC1,gH3,0235.810.15Garbonylic14s-6xdecnoic acid (CAS)Carbonylic (CAS)C3,gH3,0035.810.09Schorylic15s-6xdecnoic acid (CAS)CAS,GH2,GH2,H2,H2,H2,H3C3,gH3,0035.810.09Schorylic (CAS)15s-6xdecnoic acid, methyl ester (CAS)C3,gH3,00C3,gH3,0035.810.09Schorylic (CAS)16s-6xdecnoic acid, methyl ester (CAS)C3,gH3,00C3,gH4,0036.8036.80Schorylic (CAS)16s-6xdecnoic acid, methyl ester (CAS)C3,gH4,00C3,gH4,0036.8036.80Schorylic (CAS)17s-6xdecnoic acid, C1,Gh9,GH1,GH3,GH2,GH3,GH3,GH3,GH3,GH3,GH3,GH3,GH3,GH3,GH3	14	9-Tricosene, (Z)- (CAS)	$C_{23}H_{46}$	31.78	0.67	Hydrocarbon
169.Methyl-Z-10-tetradec en-1-ol acetateCigCigS2.470.13Ester179.Eicosyne6.03910.0310.0310.0310.0310.0310.0318Hencicosane, 11-(1-ethylpropyl)-5.2µH <sub>40</sub> 3.2µH <sub>40</sub> 3.3.181.0.1310.0110.01195.7.115-Tetramethyl-2-hexadecen-1-01C.2µH <sub>40</sub> 3.1.801.0.431.0.131.0.1320Headecanoic acid, methyl ester (CAS)C.2µH <sub>40</sub> 3.2.423.5.240.7.7Carboxylic Acid21s-5.8,11,14,17-Eicosa pentaenoic acidC.2µH <sub>40</sub> 3.5.340.3.5Carboxylic Acid22s-0ctadecenoic acid (Z)- (CAS)Carboxylic AcidCarboxylic AcidCarboxylic AcidCarboxylic Acid23headecanoic acid (Z)- (CAS)S.2Carboxylic AcidCarboxylic AcidCarboxylic Acid24b.2ctexyz-2hydroxyl,26,66,9,9,122-hept methyl-1,34,56,66,05,7,8,8,9,10,11,21,21 b.13,14b-octadecahyd ro-2H-picere-4a-carbox ylic acid, methyl esterCa <sub>10</sub> H <sub>30</sub> O3.8.181.0.9Stere256,12-Cutadeatrienoic acid, methyl ester (CAS)Carboxylic Carboxylic acidCarboxylic Carboxylic3.8.101.0.9Alcohol26Hexadecen-1-0, 3,7,11,51-tetramethyl-1,64(S)CarboxCarboxylic3.9.143.9.143.9.143.9.143.9.1427Hexadecartinoic acid, methyl ester (CAS)CarboxCarboxCarboxylic3.9.143.10.14Alcohol28Hexadecartinoic acid, methyl-1,54(S)Carbo	15	9-Octadecen-12-ynoic acid, methyl ester	C19H32O2	31.89	0.15	Ester
179-Eicosyne12.096.09Hydrocarbon18Heneicosane, 11-(1-ethylpropt)-C2abH3432.900.15Hydrocarbon193,71,1.5-Tertamethyl-2-hexadecen-1-olC2abH3033.181.54Alcohol20Hexadecanoi acid, methyl ester (CAS)C1,74,3Q235.240.15Carboxylic21si-5,8,1,1,4,1.7-Eicosa pentaenoic acidC2abH300C2abH30035.360.35Carboxylic22-0-ctadecenoic acid (Z)- (CAS)CarboxylicC1,4H340235.360.35Carboxylic23hexadecanoic acid (CAS)Carboxylic acid, methyl-1.3,4,5,6,6,6,6,7,8,8,8,0,11,12,12,01)S1,4H340231.800.09Ester240.Acctoxy-2-hydroxyl,2,6,a,6b,9,9,12a-hepta methyl-1.3,4,5,6,6,6,6,7,8,8,0,11,12,12,01S1,4H340231.800.09Ester256,9,12-Octadecatrienoic acid, methyl ester (CAS)CarboxylicC3,3H52031.800.08Ester256,9,12-Octadecatrienoic acid, methyl-1.3,4,5,6,6,6,6,7,8,8,0,11,12,12,01S1,4H340231.800.08Ester2610,21-Octadecatrienoic acid, methyl-1.3,4,5,6,6,6,6,5,7,8,8,0,11,12,12,01S1,3H32031.800.08Ester2610,21-Octadecatrienoic acid, methyl-1.3,4,5,6,6,6,6,5,7,8,8,0,11,12,12,01S1,4H340231.801.01Alcohol2710,21-Octadecatrienoic acid, methyl-1.6,16,17,16,16C2,0H4001.011.01Alcohol286,12-Octadecatrienoic acid, methyl-1.6,16,17,16,16C2,0H40031.9031.801.02Alcohol	16	9-Methyl-Z-10-tetradec en-1-ol acetate	C <sub>17</sub> H <sub>32</sub> O <sub>2</sub>	32.47	0.13	Ester
18Henicosane, 11-(1-ethylpropyl)-KeyderationC2cBH3433.2900.151Hydrocarbon1937,11,15-Tetramethyl-2-hexadecen-1-olC2abH400S3.181.54Alcohol20Hexadecanoic acid, methyl ster (CAS)C1,11C2abH3002S3.240.35Carboxylic Acid21is-58,11,14,17-Eicosa pentaenoic acidC2abH3002S3.240.35Carboxylic Acid22o-Octadecenoic acid (Z)- (CAS)Carboxylic AcidS3.840.35Carboxylic Acid23Hexadecanoic acid (CAS)Carboxylic AcidS3.840.35Carboxylic Acid2410-Acetoxy-2-hydroxyl,26a,6b,9,9,12a-hept methyl-1,34,56,6a,6b,7,88,89,10,11,12,12a,11 b,13,14b-octadecahyl ro-2H-picene-4a-carbox ylic acid, methyl esterC3,148,302S3.811.05Ester Acid256.91.2-Octadecatrienoic acid, methyl ester (CAS)C2abH40C2abH40S3.811.05Alcohol2619.4-Octadecatrienoic acid, methyl-1,16-(R*,R*-(E)]- (CAS)C2abH40S3.811.03Alcohol27Benzene, 13-(2-cyclohexylethyl)-6-cyclopentylhexyl-1,CASC2abH40S3.811.04Henzen ring28Hexadecarinoic acid, (2-phenyl-1,3-dioxolan 4-yl)methyl ester, cisC2abH40S9.401.04Henzen ring29Octadecenoic acid, (2-phenyl-1,3-dioxolan 4-yl)methyl ester, cisC2abH40S9.401.04Henzen ring29Octadecenoic acid, (2-phenyl-1,3-dioxolan 4-yl)methyl ester, cisC2abH40S9.401.04Henzen ring20Solotane	17	9-Eicosyne	C <sub>20</sub> H <sub>38</sub>	32.69	6.09	Hydrocarbon
193,7,11,15-Tetramethyl-2 -hexadecen-1-olC2,00H40O33.181.54Alcohol20Hexadecanoic acid, methyl ester (CAS)C1,7H3xO234.620.43Ester21cis-5,8,11,14,17-Eicosa pentaenoic acidC2,0H3yO235.240.77Carboxylic Acid229-Octadecenoic acid (Z)- (CAS)C1,8H3xO235.360.35Carboxylic Acid23Hexadecanoic acid (CAS)C1,8H3xO235.360.35Carboxylic Acid24b-Octadecenoic acid (CAS)C1,8H3xO235.811.75Carboxylic Acid256,912-Octadecathyl - 0.2H-picene-4a-carbox ylic acid, methyl esterC3,3H52O537.680.09Ester Alcohol269-Dexadecen-1-ol, 3,7,11,15-tetramethyl-1,R,R*-(E)]- (CAS)C1,9H32O238.000.68Ester269-Hexadecen-1-ol, 3,7,11,15-tetramethyl-1,GASC2,0H40018.151.03Alcohol27Benzene, [3-(2-cyclohexylethyl)-6-cyclopenylhexyl]- (CAS)C2,3H4039.400.08Ester28Hexadecarinoic acid, methyl ester (CAS)C2,3H4039.640.01Ester299-Octadecenoic acid, (2-phenyl-1,3-dioxalan -4-yl)methyl ester, cisC2,3H4039.690.08Ester30Cholestan-3-one, cyclic 1,2-ethanediyl atel, (5á)C2,9H30C2,9H3039.810.08Ketone30Cholestan-3-one, cyclic 1,2-ethanediyl atel, (5á)C2,2H40S9.9530.910.08Ketone31Cyclohexane, 1,4-dimethyl-2-octadecy 1C2,0H40C2,9H30<	18	Heneicosane, 11-(1-ethylpropyl)-	C <sub>26</sub> H <sub>54</sub>	32.90	0.15	Hydrocarbon
20 Hexadecanoic acid, methyl ester (CAS) CA C1, PH34O2 34.62 0.43 Ester   21 cis-5,8,11,14,17-Eicosa pentaenoic acid Ca <sup>2</sup> 0 35.24 0.77 Carboxylic   22 9-Octadecenoic acid (Z)- (CAS) ca <sup>1</sup> ca <sup>1</sup> 53.56 0.35 Carboxylic   23 Hexadecanoic acid (CAS) ca <sup>1</sup> ca <sup>1</sup> ca <sup>1</sup> ca <sup>2</sup> 24 Hexadecanoic acid (CAS) ca <sup>1</sup> ca <sup>2</sup> ca <sup>2</sup> ca <sup>2</sup> ca <sup>2</sup> 25 Hexadecanoic acid (CAS) ca <sup>1</sup> ca <sup>3</sup> ca <sup>3</sup> ca <sup>2</sup> ca <sup>2</sup> 26 10-Acctoxy-2-hydroxyl,2.6a,6b,9.9,12a-hepta methyl-1,3,4,5,6,6a,6b, 7,8,8a,9,10,11,2,12a,1 ca <sub>3</sub> si,81 1.75 ca <sup>2</sup> 27 10-Acctoxy-2-hydroxyl,2.6a,6b,9.9,12a-hepta methyl-1,2,4,56,6a,6b,7,8,8a,9,10,11,2,12a,1 ca <sub>3</sub> si,80 0.09 Ester   26 6,9,12-Octadecatrienoic acid, methyl ester (CAS) ca <sup>1</sup> ca <sup>1</sup> Alcohol   27 6,9,12-Octadecatrienoic acid, methyl-1,Re <sup>1</sup> ,R <sup>*</sup> ,(E)] <sup>1</sup> (CAS) ca <sup>2</sup> 38.00 0.68 Ester   26 2-Hexadecen-1-ol, 3,7,1,1,15-tetramethyl-, [Re <sup>1</sup> ,R <sup>*</sup> ,(E)] <sup>1</sup> (CAS) ca <sup>2</sup> <	19	3,7,11,15-Tetramethyl-2 -hexadecen-1-ol	C <sub>20</sub> H <sub>40</sub> O	33.18	1.54	Alcohol
11cis-5,8,11,14,17-Eicosa pentaenoic acidCarboxylic Acid229-Octadecenoic acid (Z)- (CAS)Carboxylic Acid23Hexadecanoic acid (CAS)Carboxylic Acid24Hexadecanoic acid (CAS)Carboxylic Acid2510-Acetoxy-2-hydroxyl,2,6a,6b,9,9,12a-hepta methyl-1,3,4,5,6,6a,6b,7,8,8a,9,10,11,12,12a,1 2b,13,14b-octadecahyd ro-2H-picene-4a-carbox ylic acid, methyl esterCarboxylic Acid2610-Acetoxy-2-hydroxyl,2,6a,6b,9,9,12a-hepta methyl-1,3,4,5,6,6a,6b,7,8,8a,9,10,11,12,12a,1 2b,13,14b-octadecahyd ro-2H-picene-4a-carbox ylic acid, methyl esterCarboxylic Acid266,912-Octadecatrienoic acid, methyl ester (CAS)Carboxylic CarboxylicSas.000.688Ester266,912-Octadecatrienoic acid, methyl ester (CAS)Carboxylic AlcoholSas.000.688Ester27Benzene, [3-(2-cyclohexylethyl)-6 -cyclopentylhexyl]- (CAS)Carboxylic CapHa0Sas.1031.03Alcohol28Hexadecanic acid, methyl ester (CAS)Carboxylic CapHa0Sas.1030.406.414Ester299-Octadecenoic acid, (2-phenyl-1,3-dioxolan -4-yl)methyl ester, cisCapHa0Sas.4030.690.088Ester30Cholestan-3-one, cyclic 1,2-ethanediyl aetal, (Sá)Capha0CapHa0Sas.4030.800.088Ketone31Oyclohexane, 1,4-dimethyl-2-octadecy 1CabhaCapHa0Sas.40Sas.4030.800.088Ketone32Sas.40Sas.40Sas.40Sas.40Sas.40Sas.40Sas.40Sas.40 <td>20</td> <td>Hexadecanoic acid, methyl ester (CAS)</td> <td>C<sub>17</sub>H<sub>34</sub>O<sub>2</sub></td> <td>34.62</td> <td>0.43</td> <td>Ester</td>	20	Hexadecanoic acid, methyl ester (CAS)	C <sub>17</sub> H <sub>34</sub> O <sub>2</sub>	34.62	0.43	Ester
229-Octadecenoic acid (Z)- (CAS)Carboxylic Acid23Hexadecanoic acid (CAS)C1BH34O235.360.35Carboxylic Acid240-Acetoxy-2-hydroxy1,2,6a,6b,9,9,12a-hepta methyl-1,3,4,5,6,6a,6b, 7,8,8a,9,10,11,12,12a,1 2b,13,14b-octadecahyd ro-2H-picene-4a-carbox ylic acid, methyl esterC33H52O537.680.09Ester Alcohol256,9,12-Octadecatrienoic acid, methyl ester (CAS)C19H32O238.000.68Ester262-Hexadecen-1-ol, 3,7,11,5-tetramethyl-, [R-[R*, E]]- (CAS)C20H40O38.151.03Alcohol27Benzene, [3-(2-cyclohexylethyl)-6 -cyclopentylhexyl]- (CAS)C22H4039.242.46benzene ring28Hexadecatrienoic acid, methyl ester (CAS)C17H28O239.460.01Ester299-Octadecenoic acid, (2-phenyl-1,3-dioxolan -4-yl)methyl ester, cisC22H4039.690.08Ester30Cholestan-3-one, cyclic 1,2-ethanediyl aetal, (5á)C29H50O239.830.08Hydrocarbon31Cyclohexane, 1,4-dimethyl-2-octadecy 1C28H40C28H5239.910.08Hydrocarbon	21	cis-5,8,11,14,17-Eicosa pentaenoic acid	$C_{20}H_{30}O_2$	35.24	0.77	Carboxylic Acid
23Hexadecanoic acid (CAS) $C_{16}H_{32}O_2$ $35.81$ $1.75$ $CarboxylicAcid2410-Acetoxy-2-hydroxyl,2,6a,6b,9,9,12a-hepta methyl-1,3,4,5,6,6a,6b, 7,8,8a,9,10,11,12,12a,1C_{33}H_{52}O_537.680.09EsterAlcohol256,9,12-Octadecatrienoic acid, methyl ester (CAS)C_{19}H_{32}O_238.000.68Ester262-Hexadecen-1-ol, 3,7,11,15-tetramethyl-, [R-[R*,R*-(E)]]- (CAS)C_{20}H_{40}O38.151.03Alcohol27Benzene, [3-(2-cyclohexylethyl)-6 -cyclopentylhexyl]- (CAS)C_{25}H_{40}39.242.46benzene ring28Hexadecatrienoic acid, methyl ester (CAS)C_{17}H_{28}O_239.460.41Ester299-Octadecenoic acid, (2-phenyl-1,3-dioxolan -4-yl)methyl ester, cisC_{28}H_{4}O_439.690.08Ester30Cholestan-3-one, cyclic 1,2-ethanediyl aetal, (5á)C_{26}H_{52}39.410.08Hydrocarbon31Cyclohexane, 1,4-dimethyl-2-octadecy l0.08HydrocarbonC_{26}H_{52}39.910.08Hydrocarbon$	22	9-Octadecenoic acid (Z)- (CAS)	C <sub>18</sub> H <sub>34</sub> O <sub>2</sub>	35.36	0.35	Carboxylic Acid
2410-Acetoxy-2-hydroxy1,2,6a,6b,9,9,12a-hepta methyl-1,3,4,5,6,6a,6b, 7,8,8a,9,10,11,12,12a,1 2b,13,14b-octadecahyd ro-2H-picer-4a-carbox ylic acid, methyl esterC_{33}H_{52}O_537,680.09Ester Alcohol256,9,12-Octadecatrienoic acid, methyl ester (CAS)C1_9H_{32}O_238.000.68Ester262-Hexadecen-1-ol, 3,7,11,15-tetramethyl-, [R-[R*,R*-(E)]]- (CAS)C2_0H_40O38.151.03Alcohol27Benzene, [3-(2-cyclohexylethyl)-6 -cyclopentylhexyl]- (CAS)C2_5H_4039.242.46benzene ring28Hexadecatrienoic acid, methyl ester (CAS)C1_7H28O239.460.01Ester299-Octadecenoic acid, (2-phenyl-1,3-dioxolan -4-yl)methyl ester, cisC2_9H44O439.690.08Ester30Cholestan-3-one, cyclic 1,2-ethanediyl actal, (5á)C2_9H50C2_9H50O239.830.08Ketone31Cyclohexane, 1,4-dimethyl-2-octadecy 1CyclohexaneC2_6H5239.910.08Hydrocarbon	23	Hexadecanoic acid (CAS)	C <sub>16</sub> H <sub>32</sub> O <sub>2</sub>	35.81	1.75	Carboxylic Acid
25 $6,9,12$ -Octadecatrienoic acid, methyl ester (CAS) $C_{19}H_{32}O_{2}$ $38.00$ $0.68$ Ester26 $2$ -Hexadecen-1-ol, $3,7,11,15$ -tetramethyl-, $[R-[R^*,R^*-(E)]]$ - (CAS) $C_{20}H_{40}O$ $38.15$ $1.03$ Alcohol27Benzene, $[3-(2-cyclohexylethyl)-6 - cyclopentylhexyl]$ - (CAS $C_{25}H_{40}$ $39.24$ $2.46$ benzene ring28Hexadecatrienoic acid, methyl ester (CAS) $C_{17}H_{28}O_{2}$ $39.46$ $0.41$ Ester299-Octadecenoic acid, $(2\text{-phenyl-1},3\text{-dioxolan -4-yl)methyl ester, cisC_{28}H_{44}O_{4}39.690.08Ester30Cholestan-3-one, cyclic 1,2-ethanediyl aetal, (5\dot{a})C_{29}H_{50}O_{2}39.810.08Ketone31Cyclohexane, 1,4-dimethyl-2-octadecy lC_{28}H_{52}39.910.08Hydrocarbon$	24	10-Acetoxy-2-hydroxy1,2,6a,6b,9,9,12a-hepta methyl-1,3,4,5,6,6a,6b, 7,8,8a,9,10,11,12,12a,1 2b,13,14b-octadecahyd ro-2H-picene-4a-carbox ylic acid, methyl ester	C <sub>33</sub> H <sub>52</sub> O <sub>5</sub>	37.68	0.09	Ester Alcohol
262-Hexadecen-1-ol, 3,7,11,15-tetramethyl-, [R-[R*,R*-(E)]]- (CAS) $C_{20}H_{40}O$ 38.151.03Alcohol27Benzene, [3-(2-cyclohexylethyl)-6 -cyclopentylhexyl]- (CAS $C_{25}H_{40}$ 39.242.46benzene ring28Hexadecatrienoic acid, methyl ester (CAS) $C_{17}H_{28}O_2$ 39.460.41Ester299-Octadecenoic acid, (2-phenyl-1,3-dioxolan -4-yl)methyl ester, cis $C_{28}H_{44}O_4$ 39.690.08Ester30Cholestan-3-one, cyclic 1,2-ethanediyl aetal, (5á) $C_{29}H_{50}O_2$ 39.830.08Ketone31Cyclohexane, 1,4-dimethyl-2-octadecy l $C_{20}H_{52}$ 39.910.08Hydrocarbon	25	6,9,12-Octadecatrienoic acid, methyl ester (CAS)	C19H32O2	38.00	0.68	Ester
27Benzene, [3-(2-cyclohexylethyl)-6 -cyclopentylhexyl]- (CAS $C_{25}H_{40}$ 39.242.46benzene ring28Hexadecatrienoic acid, methyl ester (CAS) $C_{17}H_{28}O_2$ 39.460.41Ester299-Octadecenoic acid, (2-phenyl-1,3-dioxolan -4-yl)methyl ester, cis $C_{28}H_{44}O_4$ 39.690.08Ester30Cholestan-3-one, cyclic 1,2-ethanediyl aetal, (5á) $C_{29}H_{50}O_2$ 39.830.08Ketone31Cyclohexane, 1,4-dimethyl-2-octadecy l $C_{29}H_{50}O_2$ 39.910.08Hydrocarbon	26	2-Hexadecen-1-ol, 3,7,11,15-tetramethyl-, [R-[R*,R*-(E)]]- (CAS)	C <sub>20</sub> H <sub>40</sub> O	38.15	1.03	Alcohol
28 Hexadecatrienoic acid, methyl ester (CAS) $C_{17}H_{28}O_2$ 39.46 0.41 Ester   29 9-Octadecenoic acid, (2-phenyl-1,3-dioxolan -4-yl)methyl ester, cis $C_{28}H_{44}O_4$ 39.69 0.08 Ester   30 Cholestan-3-one, cyclic 1,2-ethanediyl aetal, (5á) $C_{29}H_{50}O_2$ 39.83 0.08 Ketone   31 Cyclohexane, 1,4-dimethyl-2-octadecy l $C_{20}H_{52}$ 39.91 0.08 Hydrocarbon	27	Benzene, [3-(2-cyclohexylethyl)-6 -cyclopentylhexyl]- (CAS	C25H40	39.24	2.46	benzene ring
29   9-Octadecenoic acid, (2-phenyl-1,3-dioxolan -4-yl)methyl ester, cis   C28H44O4   39.69   0.08   Ester     30   Cholestan-3-one, cyclic 1,2-ethanediyl aetal, (5á)   C29H50O2   39.83   0.08   Ketone     31   Cyclohexane, 1,4-dimethyl-2-octadecy l   C20H52   39.91   0.08   Hydrocarbon	28	Hexadecatrienoic acid, methyl ester (CAS)	C <sub>17</sub> H <sub>28</sub> O <sub>2</sub>	39.46	0.41	Ester
30   Cholestan-3-one, cyclic 1,2-ethanediyl aetal, (5á)   C <sub>29</sub> H <sub>50</sub> O <sub>2</sub> 39.83   0.08   Ketone     31   Cyclohexane, 1,4-dimethyl-2-octadecy l   C <sub>20</sub> H <sub>50</sub> 39.91   0.08   Hydrocarbon	29	9-Octadecenoic acid, (2-phenyl-1,3-dioxolan -4-yl)methyl ester, cis	C <sub>28</sub> H <sub>44</sub> O <sub>4</sub>	39.69	0.08	Ester
31   Cyclohexane, 1,4-dimethyl-2-octadecy l   C <sub>26</sub> H <sub>52</sub> 39.91   0.08   Hydrocarbon	30	Cholestan-3-one, cyclic 1,2-ethanediyl aetal, (5á)	C <sub>29</sub> H <sub>50</sub> O <sub>2</sub>	39.83	0.08	Ketone
	31	Cyclohexane, 1,4-dimethyl-2-octadecy l	C <sub>26</sub> H <sub>52</sub>	39.91	0.08	Hydrocarbon

#### TABLE 2 Chemical composition of the biomass extract of Tetradesmus obliquus.

(Continued)

#### TABLE 2 Continued

No	Chemical molecule	Chemical Formula	Retention Time (Min)	Peak Area %	Chemical group
32	2,2-DIDEUTERO OCTADECANAL	C <sub>18</sub> H <sub>34</sub> D <sub>2</sub> O	41.26	0.24	Aldehyde
33	2(1H)-Pyrimidinethione,	C <sub>8</sub> H <sub>16</sub> N <sub>2</sub> OS	41.56	0.10	Ketone
34	9,12,15-Octadecatrienoic acid, 2,3-bis[(trimethylsilyl) oxy]propyl ester, (Z,Z,Z)	$C_{27}H_{52}O_4Si_2$	42.06	0.08	Ester
35	4H-Cyclopropa[5',6']ben z[1',2':7,8]azuleno[5,6- b]oxiren-4-one, 8-(acetyloxy)- 1,1a,1b,1c,2a,3,3a,6a,6b,7,8,8a-d odecahydro-3a,6b,8a-tr ihydroxy-2a-(hydroxym ethyl)-1,1,5,7- tetrameth yl-, [1ar-(1aà,1bá,1cà,2aà,3 aá,6aà,6bà,7à,8á,8aà)]-		42.92	0.19	Ester Alcohol
36	3,18-Epoxyandrosta-5, 7-dien-17-ol, 4,4-dimethyl-3-methoxy- (13á)		43.64	0.16	Alcohol Ether
37	Hexadecanoic acid, 2,3-dihydroxypropyl ester (CAS)	C19H38O4	44.30	0.26	Ester
38	10-Heneicosene (c,t)	$C_{21}H_{42}$	44.53	1.23	Hydrocarbon
39	(22S)-21-Acetoxy-6à,1 1á-dihydroxy-16à,17à-p ropylmethylenedioxypre gna-1,4-diene-3,20- dio ne	C <sub>27</sub> H <sub>36</sub> O <sub>8</sub>	45.11	0.26	Ester Alcohol
40	1,2-Benzenedicarboxylic acid, isodecyl octyl ester	$C_{26}H_{42}O_4$	45.44	74.11	Ester
41	4-(4'-pentylbicyclohexy l)-1-(propylcyclohexyl)b enzene	$C_{32}H_{52}$	46.16	0.12	benzene ring
42	Milbemycin b, 13-chloro-5-demethoxy -28-deoxy-6,28-epoxy5-(hydroxyimino)-25-( 1-methylethyl)-, (6R,13R,25R)-	C <sub>33</sub> H <sub>46</sub> ClNO <sub>7</sub>	47.00	0.11	Alcohol Ketone
43	Cyclohexane, 1,1'-dodecylidenebis[4-m ethyl	C <sub>26</sub> H <sub>50</sub>	47.53	0.18	Hydrocarbon
44	Lucenin 2	$C_{27}H_{30}O_{16}$	48.45	0.44	Alcohol Keton
45	5,11,17 23-tetrakis(1,1-dimethy lethyl)-28-methoxypent acyclo[19.3.1.1(3,7).1(9,13).1(15,19)] octacosa1(25),3,5,7(28),9,11,13 (27),15,17,19(26),21,23 -dodecene-25,26,27-triol	C <sub>45</sub> H <sub>58</sub> O <sub>4</sub>	48.99	0.08	Alcohol
46	cis-10-Heptadecenoic acid	C <sub>17</sub> H <sub>32</sub> O <sub>2</sub>	49.23	0.22	Carboxylic Acid

#### TABLE 3 Chemical composition of the biomass extract of Monoraphidium sp.

No	Chemical molecule	Chemical Formula	Retention Time (Min)	Peak Area %	Chemical group
1	Astaxanthin	$C_40H_{52}O_4$	8. 04	0.57	Ketone
2	Docosane (CAS)	C <sub>22</sub> H <sub>46</sub>	8. 30	0.56	Hydrocarbon
3	(2,3-Dihydro-2-nitro-5,1 0,15,20-tetraphenyl[3-( 2)H1]prophyrinato)copp er(II)	C44H28DCuN5O2	9.86	0.42	Nitro Group
4	(2,2-Dibenzyloxy-3-nit ro-5,10,15,20-tetraphen yl-2,3-dihydroporphyri nato)copper(II)	C58H40CuN5O2	10.06	0.42	Ether
5	2(4)-(1-Hydroxyethyl)- 4(2)-(1-isopropoxyethy l)deuteroporprophyrin Dimethyl Ester	C <sub>39</sub> H <sub>47</sub> DN <sub>4</sub> O <sub>6</sub>	11.25	0.42	Ester
6	PENITREM A	C <sub>37</sub> H <sub>44</sub> ClNO <sub>6</sub>	11.69	0.57	Alcohol Ketone
7	Cyclohexane, 1,1',1"'-(1,6-hexanedi ylidene)tetrakis- (CAS)	$C_{30}H_{54}$	11.87	0.51	Hydrocarbon
8	Nephthoside - 1,2',3',4'-Tetraacetate 4 3	$C_{40}H_{56}O_{10}$	11.94	0.47	Ester
9	Russuphelol	$C_{26}H_{16}C_{16}O_8$	12.35	0.76	Alcohol Ketone
10	Milbemycin B, 5-demethoxy-5-one-6,2 8-anhydro-25-ethyl-4- methyl-13-chloro-oxime	C <sub>32</sub> H <sub>44</sub> ClNO <sub>7</sub>	12.88	0.75	Alcohol Ether
11	PENITREM A	C <sub>37</sub> H <sub>44</sub> ClNO <sub>6</sub>	13.33	0.56	Alcohol Ketone
12	Tungsten, pentacarbonyl(4,5-dieth yl-2,2,3-trimethyl-1-phe nyl-1-phospha-2-sila-5- boracyclohex-3-ene-P1 )-, (oc-6-22)-	C <sub>21</sub> H <sub>26</sub> BO <sub>5</sub> PSiW	16.32	0.44	Organometallic

(Continued)

#### TABLE 3 Continued

No	Chemical molecule	Chemical Formula	Retention Time (Min)	Peak Area %	Chemical group
13	2-(16-Acetoxy-11-hydro xy-4,8,10,14-tetramethy l-3-oxohexadecahydroc yclopenta[a] phenanthre n-17-ylidene)-6-methyl -hept-5-enoic acid, methyl ester	$C_{32}H_{48}O_{6}$	24.63	0.57	Ester
14	2-OCTADEC-1"-ENY LOXY-1,1,2,2-TETRA DEUTERO ETHANOL	$C_{20}H_{36}D_4O_2$	27.36	0.77	alcohol
15	3-Pyridinecarboxylic acid,	C <sub>32</sub> H <sub>39</sub> NO <sub>10</sub>	28.68	0.63	Carboxylic Acid
16	Docosane (CAS)	C <sub>22</sub> H <sub>46</sub>	29.75	0.89	Hydrocarbon
17	ç-PICOLINE-à,ê-D5	$C_6H_2D_5N$	31.67	0.83	Amine
18	1,1-dichloro-2-dodecanol	$C_{12}H_{24}C_{12}O$	31.79	2.72	alcohol
19	2,4(1H)-Cyclo-3,4-seco akuammilanium, 3,17-dihydroxy-16-(me thoxycarbonyl)-4-meth yl-, (3á,16R)-	$C_{22}H_{29}N_2O_4$	31.88	0.83	Ester
20	Dodecachloroperylene	C200Cl12	32.04	0.54	Hydrocarbon
21	(2-hydroxy-5,10,15,20- tetraphenylporphinato)c opper(II)	C44H28CuN4O	32.42	0.47	alcohol
22	3,7,11,15-Tetramethyl-2 -hexadecen-1-ol	$C_{20}H_{40}O$	32.67	9.34	Alcohol Hydrocarbon
23	Phytol, acetate	$C_{22}H_{42}O_2$	33.18	3.10	Alcohol Ester
24	9-Octadecenoic acid (Z)-, octadecyl ester	$C_{36}H_{70}O_2$	33.32	0.42	Ester
25	2-Hexadecen-1-ol, 3,7,11,15-tetramethyl-, [R-[R*,R*-(E)]]- (CAS)	$C_{20}H_{40}O$	33.56	4.99	Alcohol Hydrocarbon
26	9-Octadecenoic acid (Z)-, 2-butoxyethyl ester	C <sub>24</sub> H <sub>46</sub> O <sub>3</sub>	34.55	0.53	Ester
27	Pentadecanoic acid, 14-methyl-, methyl ester (CAS)	$C_{17}H_{34}O_2$	34.63	3.04	Ester
28	Octadecanoic acid (CAS)	$C_{18}H_{36}O_2$	35.71	3.72	Carboxylic Acid
29	2,2-DIDEUTERO OCTADECANAL	$C_{18}H_{34}D_2O$	35.82	3.24	Aldehyde
30	Ethyl iso-allocholate	$C_{26}H_{44}O_5$	37.90	0.73	Ester
31	7,10,13-Hexadecatrienoic acid, methyl ester	$C_{17}H_{28}O_2$	38.00	2.80	Ester
32	(2-hydroxy-5,10,15,20- tetraphenylporphinato)c	$\mathrm{C}_{44}\mathrm{H}_{28}\mathrm{CuN}_{4}\mathrm{O}$	38.15	1.45	alcohol
33	9-Octadecenoic acid (Z)- (CAS)	$C_{18}H_{34}O_2$	39.04	5.74	Carboxylic Acid
34	Octadecanoic acid, 4-hydroxy-, methyl ester	C <sub>19</sub> H <sub>38</sub> O <sub>3</sub>	39.43	0.63	Ester alcohol
35	Pregn-4-ene-3,20-dion e, 17,21-dihydroxy-, bis(O-methyloxime)	$C_{23}H_{36}N_2O_4$	43.38	0.46	Ketone alcohol
36	Lycoxanthin	C <sub>40</sub> H <sub>56</sub> O	43.70	0.66	Alcohol Ketone
37	1,2-Benzenedicarboxylic acid, dioctyl ester (CAS)	C24H38O4	45.35	32.62	Ester
38	2,4,6-Decatrienoic acid, 1a,2,5,5a,6,9,10,10a-oc tahydro-5,5a-dihydroxy -4-(hydroxymethyl)- 1,7, 9-trimethyl-1-[[(2-meth yl-1-oxo-2-butenyl)oxy ]methyl]-11-oxo-1H-2, 8a- methanocyclopenta[a ]cyclopropa[e]cyclodece n-6-yl ester	$C_{35}H_{46}O_8$	45.90	0.80	Ester alcohol
39	Milbemycin B, 5-demethoxy-5-one-6,2 8-anhydro-25-ethyl-4- methyl-13-chloro-oxime	C <sub>32</sub> H <sub>44</sub> ClNO <sub>7</sub>	46.04	0.83	Alcohol Ether
40	Glucobrassicin	$C_{16}H_{20}N_2O_9S_2$	46.34	0.57	
41	TRISTRIMETHYLSILY L ETHER DERIVATIVE OF 1,25-DIHYDROXYVIT AMIN D2	C37H68O3Si3	46.42	0.43	Alcohol Ether

(Continued)

#### TABLE 3 Continued

No	Chemical molecule	Chemical Formula	Retention Time (Min)	Peak Area %	Chemical group
42	Perhydroindene-4-carbo xylic acid, 6-acetoxy-2,3-epoxy-1, 1-epoxymethyl-3a-hydro xy-5- isopropenyl-7a-m ethyl-7-oxo-, methyl ester	C <sub>18</sub> H <sub>22</sub> O <sub>8</sub>	46.98	0.69	Ester
43	ZINC CHLORIDE OXIDIZED	C <sub>28</sub> H <sub>37</sub> ClN <sub>4</sub> OZn	47.08	0.71	_
44	Cyclohexane, 1,1'-dodecylidenebis[4-m ethyl	$C_{26}H_{50}$	47.54	3.66	Hydrocarbon
45	Cinobufotalin	$C_{26}H_{34}O_7$	48.01	0.40	Alcohol Ketone
46	Methanesulfonic acid, 2-(3-hydroxy-4,4,10,13,14-pentamethyl-2,3,4,5,6,7,10,11,12,13,14,15, 16,17-tetradecahydro-1 H-cyclopenta[a]phenant hryl)-	$C_{26}H_{44}O_4S$	48.93	0.47	Acid alcohol
47	7,8-Epoxylanostan-11- ol, 3-acetoxy	$C_{32}H_{54}O_4$	49.01	0.57	Alcohol Ester
48	2,5-Dibromo-1,4-di-n-h exadecylbenzene	$\mathrm{C}_{38}\mathrm{H}_{68}\mathrm{Br}_{2}$	49.10	0.54	Hydrocarbon
49	Cyclohexane, 1,1',1"'-(1,6-hexanedi	$C_{30}H_{54}$	49.21	1.22	Hydrocarbon
50	Phenol, 2-methoxy-6-(3,7,11,1 5,19,23,27,31,35-nona methyl-2,6,10,14,18,22,26,30,34-hexatriacont anonaenyl)- (CAS)	$C_{52}H_{80}O_2$	49.30	0.96	Alcohol Ether

In *Tetradesmus obliquus*, hexadecanoic acid, methyl ester (methyl palmitate) was the most abundant component, accounting for 56.61% of the total peak area, followed by methyl oleate (9-octadecenoic acid, methyl ester) at 22.30%. Other notable compounds included methyl myristate and methyl linoleate, as well as lesser quantities of methyl linolenate and 2,3-dihydroxypropyl palmitate. Both saturated (C14:0, C16:0) and unsaturated (C18:1, C18:2, C18:3) FAMEs indicate a biodiesel profile with balanced oxidative stability and cold flow properties (Table 4).

Similarly, *Monoraphidium* sp. biodiesel was dominated by 1hexadecanol, a fatty alcohol derived from palmitic acid, with a peak area of 54.75%, and methyl palmitate at 12.09%. Other major constituents included methyl oleate (7.69%), methyl octacosanoate (4.39%), and methyl lignocerate (4.77%) (Table 5). Both species shared several key compounds, particularly methyl palmitate and methyl oleate.

### 3.6 Biodiesel quality estimation

The predicted biodiesel properties of *Tetradesmus obliquus* and *Monoraphidium* sp., as estimated using Biodiesel Analyzer software are displayed in Table 6. *Monoraphidium* sp. showed a higher proportion of saturated fatty acids (SFA) at 74.65%, compared to 61.19% in *T. obliquus*, which contributed to its elevated cetane number (CN = 75.96) and reduced degree of unsaturation (DU = 7.69). Conversely, *T. obliquus* presented a more balanced profile with 22.84% monounsaturated fatty acids (MUFA) and 1.01% polyunsaturated fatty acids (PUFA), leading to a higher DU (24.86) and improved oxidative stability (OS = 119.35 hours), whereas *Monoraphidium* sp. recorded zero oxidative stability. The iodine value (IV), an indicator of unsaturation, was notably higher in *T. obliquus* (22.84) than in *Monoraphidium* sp. (6.92), correlating with its enhanced oxidative resistance. Despite the superior CN and

lower IV, Monoraphidium sp. exhibited significantly higher cold flow properties, including a long-chain saturated factor (LCSF = 17.50), cloud point (30.28°C), and cold filter plugging point (CFPP = 38.50°C), suggesting lower suitability for colder climates compared to T. obliquus, which recorded lower values in these parameters (LCSF = 5.85, CP = 25.78°C, CFPP = 1.90°C). Both species exhibited similar kinematic viscosities (1.15-1.16 mm<sup>2</sup>/s) and densities (0.71-0.74 g/cm<sup>3</sup>), with T. obliquus also displaying a slightly higher heating value (HHV = 33.41 MJ/kg). Collectively, these findings indicate that Tetradesmus obliquus offers a more favorable balance between fuel stability, cold flow properties, and energy content, making it a more versatile and climate-resilient biodiesel feedstock. While both microalgae strains are viable candidates for biodiesel production, T. obliquus is characterized by a higher content of conventional biodiesel esters, contributing to better oxidative stability and combustion quality, whereas Monoraphidium sp. is distinguished by a higher proportion of longer-chain saturated and alcohol-based lipids, which may enhance specific fuel attributes such as lubricity and viscosity but potentially limit cold flow performance.

# 4 Discussion

In recent years, extensive research has focused on exploring the biodiesel potential of microalgae, yet significant knowledge gaps remain regarding the comparative performance of different strains under standardized conditions. This study addresses these gaps by providing a comprehensive evaluation of *Tetradesmus obliquus* and *Monoraphidium* sp., incorporating morphological, molecular, physiological, biochemical, and fuel property analyses. Unlike most previous studies, which often examine single strains or focus on limited parameters, our work offers an integrated assessment, including the rarely reported cold flow properties of biodiesel



critical for determining real-world applicability. Furthermore, the investigation of species-specific responses to nutrient ratios and light intensities highlights the complex trade-offs between maximizing biomass and enhancing fuel quality.

Algae have emerged as a promising renewable resource for biodiesel production due to their high lipid content and rapid growth rates (Gaurav et al., 2024; Sharma et al., 2025). Among various species, *Tetradesmus obliquus* and *Monoraphidium* sp. have demonstrated significant potential (Holbrook et al., 2014; Bibi et al., 2022; Falfushynska, 2024). *T. obliquus* has been shown to accumulate substantial lipid content under various stress conditions, including heavy metal exposure, which enhances its suitability for biodiesel applications (Alwaleed et al., 2025). Additionally, the application of p-coumaric acid has been reported to significantly increase lipid accumulation in *T. obliquus*, further improving its biodiesel yield (Esakkimuthu et al., 2020). In Egypt, studies have identified *T. obliquus* as a high biomass-producing species, while *Monoraphidium minutum* exhibited the highest lipid productivity among tested microalgae, indicating their complementary roles in biodiesel feedstock development (Mohamed et al., 2022). These findings underscore the importance of selecting and optimizing specific algal strains to advance sustainable biodiesel technologies.

The observed morphological characteristics of *Tetradesmus* obliquus align well with previous descriptions of this species, which typically forms coenobia composed of four to eight elongated cells with smooth walls and rounded apices arranged in a linear or slightly curved pattern (Cho and Lee, 2024). The presence of these compact coenobial structures under both light and scanning electron microscopy (SEM) reflects the adaptive colonial nature of *T. obliquus*, which may contribute to its

No	Chemical molecule	Chemical Formula	Retention Time (Min)	Peak Area %	Chemical group	Carbon double bond
1	Acetic acid, 5-(2,2-dimethyl-6-oxocy clohexylidene)-3- methy l-pent-3-enyl ester	$C_{16}H_{24}O_3$	7. 79	0.28	methyl palmitate (palmitic acid)	C16:0
2	2H-2,4a-Methanonapht halen-8(5H)-one, 1,3,4,6,7,8a- hexahydro -1,1,5,5-tetramethyl- (CAS)	C <sub>15</sub> H <sub>24</sub> O	8. 07	1.15	Methyl myristate (myristic acid)	C14:0
3	9-Octadecenoic acid (Z)- (CAS)	$C_{18}H_{34}O_2$	12.61	0.54	oleic acid	C18:1
4	6-(1',1'-Dimethylethyl)- 2-methyl-2,4,5-decatrie n-1-ol	$C_{15}H_{26}O$	13.90	0.23	methyl tetradecanoate (myristic acid)	C14:0
5	7-Oxabicyclo[4.1.0]hep tane-3-carboxylic acid, 7- oxabicyclo[4.1.0]hep t-3-ylmethyl ester	$C_{14}H_{20}O_4$	17.30	1.31	Methyl myristate (myristic acid)	C14:0
6	Hexadecatrienoic acid, methyl ester (CAS)	$C_{17}H_{28}O_2$	17.64	0.51	Methyl linolenate (linolenic acid)	C18:3
7	Hexadecanoic acid, methyl ester (CAS)	$C_{17}H_{34}O_2$	18.21	56.61	Methyl palmitate (palmitic acid)	C16:0
8	Hexadecanoic acid, 2,3-dihydroxypropyl ester (CAS)	C <sub>19</sub> H <sub>38</sub> O <sub>4</sub>	21.73	0.24	2,3-dihydroxypropyl palmitate (palmitic acid)	C16:0
9	8,11-Octadecadienoic acid, methyl ester (CAS)	$C_{19}H_{34}O_2$	22.16	0.50	methyl linoleate (linoleic acid)	C18:2
10	9-Octadecenoic acid (Z)-, methyl ester (CAS)	C <sub>19</sub> H <sub>36</sub> O <sub>2</sub>	22.30	22.30	methyl oleate (oleic acid)	C18:1
11	Cyclopentanetridecanoic acid, methyl ester (CAS)	C <sub>19</sub> H <sub>36</sub> O <sub>2</sub>	22.86	1.37	methyl 16- cyclopentylpalmitate (palmitic acid)	C16:0

TABLE 4 Chemical composition of biodiesel derived from Tetradesmus obliguus.

resilience and high biomass yield in mass culture systems (Lürling and Van Donk, 2000; Cardon et al., 2018). The smooth cell wall surface observed under SEM supports its taxonomic classification and indicates minimal extracellular ornamentation, a feature commonly reported for this genus (do Carmo Cesário et al., 2022). In contrast, *Monoraphidium* sp. exhibited a distinctive unicellular, crescent-shaped morphology with tapering ends, as consistently reported in the literature for members of this genus (Fawley et al., 2006). The solitary cell arrangement and absence of spines or visible wall ornamentation under SEM further validate its identity and suggest potential advantages for ease of harvesting and processing in biotechnological applications.

Molecular identification through 18S rRNA gene sequencing effectively validated the taxonomic classification of the isolated microalgae strains, supporting their identification as *Tetradesmus obliquus* and *Monoraphidium* sp. The observed sequences displayed high similarity with GenBank reference sequences indicating a high level of genetic conservation within these taxa. This degree of similarity is consistent with previous studies that highlight the robustness of 18S rRNA markers for accurate phylogenetic placement and species-level discrimination among green microalgae (Hanan et al., 2022; Lortou et al., 2022). The phylogenetic analysis further reinforced these findings, as the isolates clustered within their respective clades with strong bootstrap support, aligning with morphological observations and demonstrating congruence between molecular and classical taxonomic approaches.

The comparative analysis of growth kinetics and lipid productivity between Tetradesmus obliguus and Monoraphidium sp. under standard cultivation conditions (25 ± 2 °C, 1.2 klux, 16:8 h light/dark cycle) revealed significant differences in their biofuel potential. Monoraphidium sp. exhibited a markedly higher dry biomass yield, reaching 0.177  $\pm$  0.0078 g L<sup>-1</sup> by day 28, compared to 0.047  $\pm$  0.0025 g L<sup>-1</sup> for *T. obliquus*. This superior biomass accumulation aligns with previous findings where Monoraphidium sp. achieved biomass levels up to 1.18 g L<sup>-1</sup> under optimized conditions (Dong et al., 2019). In terms of lipid content, Monoraphidium sp. peaked at 0.368  $\pm$  0.049 g g<sup>-1</sup> on day 28, surpassing *T. obliquus*, which reached  $0.213 \pm 0.028$  g g<sup>-1</sup>. Such lipid accumulation in Monoraphidium sp. is consistent with reported ranges of 19-35% of dry weight (Hawrot-Paw et al., 2020). The specific growth rate ( $\mu$ ) was generally higher in *Monoraphidium* sp., with a peak of 0.132  $\pm$  0.019 d<sup>-1</sup> on day 21, while T. obliquus reached its maximum  $\mu$  of 0.113  $\pm$  0.002 d<sup>-1</sup> earlier, on day 14. Correspondingly, the growth doubling per day (Dd<sup>-1</sup>) and generation time (G) metrics favored Monoraphidium sp., indicating a more efficient growth profile. These findings suggest that Monoraphidium sp. holds greater promise for biodiesel production under the tested conditions, although further optimization could enhance the performance of T. obliquus, as

No	Chemical molecule	Chemical Formula	Retention Time (Min)	Peak Area %	Chemical group	Carbon double bond
1	Dodecanoic acid, methyl ester (CAS)	$C_{13}H_{26}O_2$	8.19	0.39	Methyl laurate (lauric acid)	C12:0
2	Perhydroindene-4-carbo xylic acid, 6-acetoxy-2,3-epoxy- 1, 1-epoxymethyl-3a-hydro xy-5-isopropenyl-7a-m ethyl-7-oxo-, methyl ester	C <sub>18</sub> H <sub>22</sub> O <sub>8</sub>	8. 54	0.39	Methyl ester of 9,10-dihydroxy-12- octadecenoic acid, or Methyl 9,10- dihydroxy-octadecenoate.	
3	2-Propenoic acid, 2-methyl-, dodecyl ester (CAS)	$C_{16}H_{30}O_2$	8. 75	0.22	Palmitic acid	C16:0
4	1-Hexadecanol (CAS)	C <sub>16</sub> H <sub>34</sub> O	17.31	54.75	palmitic acid	C16:0
5	2,15-Heptadecadiene, 9-(ethoxymethyl)-	C <sub>20</sub> H <sub>38</sub> O	17.60	0.22	arachidate methyl ester arachidic acid	C20:0
6	Hexadecanoic acid, methyl ester (CAS)	C <sub>17</sub> H <sub>34</sub> O <sub>2</sub>	18.20	12.09	Methyl Palmitate palmitic acid	C16:0
7	7-Methyl-Z-tetradecene1-ol ace	C <sub>17</sub> H <sub>32</sub> O <sub>2</sub>	22.16	0.94	margaric acid (or heptadecanoic acid)	C17:0
8	16-Octadecenoic acid, methyl ester	$C_{19}H_{36}O_2$	22.32	7.69	methyl oleate (oleic acid)	C18:1
9	Pentadecanoic acid, 14-methyl-, methyl ester (CAS)	$C_{17}H_{34}O_2$	22.86	0.58	Iso-pentadecanoic acid methyl ester or 14-methylpentadecanoic acid methyl ester.	C15:0
10	Docosane (CAS)	C <sub>22</sub> H <sub>46</sub>	23.13	0.38	behenic acid methyl ester (Behenic acid)	C22:0
11	Docosane (CAS)	C <sub>22</sub> H <sub>46</sub>	30.44	0.31	behenic acid methyl ester (Behenic acid)	C22:0
12	Octacosanoic acid, methyl ester (CAS)	$C_{29}H_{58}O_2$	31.98	4.39	Methyl octacosanoate. octacosanoic acid	
13	(rac)-4,4',7,7'-Tetramet hoxy-5,5'-dimethyl-2H,2 'H-6,6'- bichromen-2,2'-	C <sub>24</sub> H <sub>22</sub> O <sub>8</sub>	32.09	4.77	Lignoceric acid methyl ester	C24:0

TABLE 5 Chemical composition of biodiesel derived from Monoraphidium sp.

studies have demonstrated improved biomass and lipid yields under modified cultivation strategies (Jin et al., 2024).

The optimization of biomass yield and lipid accumulation in *Tetradesmus obliquus* and *Monoraphidium* sp. under varying nitrogen to phosphorus (N:P) ratios and light intensities reveals species-specific responses that are crucial for biodiesel production. In *T. obliquus*, the highest dry biomass yield (0.065 g L<sup>-1</sup>) was observed at an N:P ratio of 1:1, indicating that a balanced nutrient supply supports optimal growth. This aligns with previous findings that suggest balanced N:P ratios enhance biomass productivity in green microalgae (Dhup and Dhawan, 2014). Conversely, *Monoraphidium* sp. achieved its maximum biomass (0.085 g L<sup>-1</sup>) at a 4:1 N:P ratio, suggesting a preference for nitrogen-rich conditions, which corroborates studies highlighting the influence of nitrogen availability on microalgal growth and lipid productivity (Dhup and Dhawan, 2014).

Regarding lipid accumulation, both species exhibited the highest lipid content at the 1:1 N:P ratio, with *T. obliquus* reaching 34.8% and *Monoraphidium* sp. attaining 47.3%. These results are consistent with literature indicating that balanced nutrient conditions favor lipid biosynthesis in microalgae (Morales et al., 2021).

Light intensity also significantly impacted biomass and lipid production. For *T. obliquus*, the optimal biomass yield (0.105 g L<sup>-1</sup>) and lipid content (41.8%) were achieved at 2.6 klux, suggesting that moderate light intensities promote both growth and lipid accumulation. This observation is supported by studies demonstrating that specific light intensities can enhance lipid synthesis in microalgae (Mulgund, 2022). Similarly, *Monoraphidium* sp. showed maximum biomass (0.42–0.43 g L<sup>-1</sup>) at higher light intensities (3 and 4.2 klux), while the highest lipid content (50.6%) was recorded at 2.6 klux, indicating that optimal light conditions for lipid accumulation may differ from those for biomass production. These findings align with research highlighting the species-specific responses of microalgae to light intensity variations (Maltsev et al., 2021).

The GC-MS analysis of chloroform-methanol extracts from *Tetradesmus obliquus* and *Monoraphidium* sp. revealed a diverse array of bioactive and lipid-derived compounds, underscoring their potential in biodiesel production and biotechnological applications. In *T. obliquus*, the predominant fatty acid methyl esters (FAMEs) identified were methyl palmitate (hexadecanoic acid, methyl ester), methyl oleate (9-octadecenoic acid, methyl ester), and methyl stearate (octadecanoic acid, methyl ester). These saturated and

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ltem	Tetradesmus obliquus	Monoraphidium sp.
Saturated Fatty Acids (SFA)	61.19	74.65
Monounsaturated Fatty Adds (MUFA)	22.84	7.69
Polyunsaturated Fatty Acids (PUFA)	1.01	0.00
Degree of Unsaturation (DU)	24.86	7.69
Saponification Value (SV)	181.66	174.86
Iodine Value (IV)	22.84	6.92
Cetan Number (CN)	71.21	75.96
Long-Chain Saturated Factor (LCSF)	5.85	17.50
Cold Filter Plugging Point (CFPP)	1.90	38.50
Cloud point (CP)	25.78	30.28
Allylic Position Equivalents (APE)	24.86	7.69
Bis -Allylie Position Equivalents (BAPE)	1.52	0.44
Oxidation Stability (OS)	119.35	0
Higher Heating Value (HHV)	33.41	32.41
Kinematic Viscosity (u)	1.15	1.16
Density (p)	0.74	0.71

TABLE 6 Biodiesel properties of *Tetradesmus obliquus* and Monoraphidium sp. predicted using Biodiesel Analyzer software.

monounsaturated FAMEs are known to enhance biodiesel quality by improving oxidative stability and cetane number, aligning with previous findings that highlight *T. obliquus* as a favorable feedstock for biodiesel due to its high palmitic acid content (Ahiahonu et al., 2022). Additionally, the presence of hydrocarbons such as 2,4-dimethylheptane and phthalate derivatives like 1,2benzenedicarboxylic acid, diisooctyl ester, suggests potential applications in bioplastics and industrial solvents, although the latter's origin warrants further investigation to rule out contamination (Vladić et al., 2023).

Similarly, *Monoraphidium* sp. exhibited a comparable compound profile, with methyl palmitate, methyl oleate, and methyl stearate as dominant constituents. The detection of branched alkanes such as 2,6,10-trimethylpentadecane further indicates the species' capacity to produce hydrocarbons suitable for biofuel applications. The shared presence of these compounds in both microalgae underscores their potential as sustainable sources for biodiesel production. Moreover, the identification of bioactive compounds with known antioxidant and antimicrobial properties, such as certain fatty acid methyl esters, aligns with reports that microalgal metabolites can serve as functional food ingredients and therapeutic agents (Eze et al., 2023). These findings highlight the

multifaceted applications of *T. obliquus* and *Monoraphidium* sp., extending beyond energy production to include roles in health and industrial sectors.

The biodiesel profiles of Tetradesmus obliquus and Monoraphidium sp. reveal distinct compositions and properties, influencing their suitability for biodiesel applications (Ahiahonu et al., 2022). In T. obliquus, the predominant fatty acid methyl ester (FAME) is methyl palmitate (hexadecanoic acid methyl ester), comprising 56.61% of the total peak area. This aligns with findings by Ahiahonu et al. (2022), who reported palmitic acid as the most abundant fatty acid in T. obliquus, accounting for 34.26% of the total composition. The presence of both saturated (C14:0, C16:0) and unsaturated (C18:1, C18:2, C18:3) FAMEs suggest a biodiesel profile with balanced oxidative stability and cold flow properties. This balance is crucial, as higher unsaturation levels can lead to increased oxidative instability and lower cetane numbers. The calculated cetane number (CN) of 73 for T. obliquus biodiesel falls within acceptable standards, indicating reliable ignition quality. Moreover, the iodine value (IV) of 22.84 reflects a moderate degree of unsaturation, correlating with enhanced oxidative stability. These characteristics position T. obliquus as a favorable feedstock for biodiesel, offering a balance between performance and stability. The higher heating value (HHV) of Tetradesmus obliquus biodiesel was 33.41 MJ/kg, which is notably lower than that of conventional petrodiesel (~45 MJ/kg). This difference is expected, as biodiesels generally have lower energy content due to their oxygenated nature. However, the HHV of T. obliquus falls within the typical range reported for microalgal biodiesels, which often varies between 30 and 40 MJ/kg, depending on lipid composition and fatty acid profile. Although the energy density is lower than petrodiesel, T. obliquus still presents a viable alternative biofuel source, particularly when considering its renewable origin and the potential for optimizing cultivation and extraction conditions to improve fuel quality. This makes T. obliquus a viable candidate for biodiesel production, especially in regions where optimization of cultivation processes can lead to enhanced biofuel quality. Additionally, blending T. obliquus biodiesel with other biodiesels or petrodiesel could mitigate the lower HHV and improve its overall fuel performance.

Conversely, Monoraphidium sp. biodiesel is characterized by a higher proportion of saturated fatty acids (74.65%), leading to a higher CN of 75.96. However, this high saturation correlates with elevated cold flow properties, such as a cloud point of 30.28°C and a cold filter plugging point (CFPP) of 38.50°C, suggesting reduced suitability for colder climates. The absence of oxidative stability in Monoraphidium sp. biodiesel may be attributed to its high saturated and low unsaturated fatty acid content, which typically enhances oxidative stability. This finding contrasts with studies on other microalgae, such as Desmodesmus sp. (I-AU1), which, despite a high saturated fatty acid content, exhibited favorable oxidative stability, possibly due to different lipid compositions or cultivation conditions (Arguelles et al., 2018). The presence of 1-hexadecanol (a fatty alcohol derived from palmitic acid) as a major component in Monoraphidium sp. biodiesel is noteworthy. Fatty alcohols can influence biodiesel properties, potentially enhancing lubricity and

viscosity, which are beneficial for engine performance. However, excessive concentrations may adversely affect cold flow properties (Gurau et al., 2025).

Both species share key compounds, notably methyl palmitate and methyl oleate, highlighting their potential as biodiesel feedstocks. The distinct differences in their fatty acid compositions underscore the importance of strain selection and cultivation conditions in optimizing biodiesel quality. *T. obliquus* offers a more balanced biodiesel profile, suitable for diverse climatic conditions, while *Monoraphidium* sp. may be more appropriate for applications where high cetane numbers and specific fuel properties are desired, despite potential challenges with cold flow performance.

## **5** Conclusion

This study evaluated Tetradesmus obliquus and Monoraphidium sp. as potential feedstocks for sustainable biodiesel production. Both microalgae demonstrated high lipid content and favorable fatty acid profiles, with Monoraphidium sp. showing superior biomass productivity (0.43 g L<sup>-1</sup>) and lipid accumulation (50.6%) compared to T. obliquus (0.105 g  $L^{-1}$  and 41.8%, respectively). GC-MS analysis confirmed key biodiesel components like methyl palmitate and methyl oleate in both species. Predictive assessments indicated that T. obliquus offers better oxidative stability and cold flow properties, making it versatile for varying climates, while Monoraphidium sp. exhibited higher cetane numbers but limited cold flow performance. However, Monoraphidium sp. biodiesel may still be suitable for use in warmer regions where cold flow properties are less critical. Additionally, blending Monoraphidium biodiesel with other biodiesels, particularly those with better cold flow properties, could mitigate its performance limitations in cooler climates.

These findings highlight the distinct advantages of each species depending on application requirements. However, further research into scalable cultivation techniques and cost-effective extraction methods is essential for industrial implementation. By advancing these areas, we can harness the full potential of microalgae for sustainable energy solutions.

### Data availability statement

The original contributions presented in the study are included in the article/supplementary material. Further inquiries can be directed to the corresponding authors.

# Author contributions

JM: Visualization, Data curation, Resources, Writing – original draft, Conceptualization, Formal Analysis, Validation, Methodology, Investigation, Supervision, Writing – review & editing, Software. IM: Software, Resources, Formal Analysis, Data curation, Investigation, Writing – original draft, Writing – review & editing, Methodology, Validation. EE: Supervision, Validation, Conceptualization, Writing – original draft, Writing – review & editing, Methodology, Data curation, Investigation, Software. AS: Writing – review & editing, Data curation, Software, Investigation, Resources, Validation, Formal Analysis, Methodology, Writing – original draft. AE: Investigation, Writing – original draft, Writing – review & editing, Methodology, Software, Formal Analysis, Resources, Data curation, Validation. IT: Formal Analysis, Writing – original draft, Data curation, Software, Visualization, Resources, Supervision, Conceptualization, Investigation, Writing – review & editing, Methodology. EE: Conceptualization, Visualization, Validation, Resources, Methodology, Data curation, Writing – review & editing, Writing – original draft, Investigation, Formal Analysis, Supervision, Software.

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# **Conflict of interest**

The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

## **Generative AI statement**

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