



A Brief Introduction on the Development of Ti-Based Metallic Glasses

M. Zhang^{1*}, Y.Q. Song¹, H.J. Lin¹, Z. Li^{2*} and W. Li¹

¹Institute of Advanced Wear and Corrosion Resistant and Functional Materials, Jinan University, Guangzhou, China, ²School of Physics and Materials Science, Guangzhou University, Guangzhou, China

OPEN ACCESS

Edited by:

Limin Wang,
Yanshan University, China

Reviewed by:

Pan Gong,
Huazhong University of Science and
Technology, China

Haizheng Tao,
Wuhan University of Technology,
China

*Correspondence:

M. Zhang
m.zhangjwrm@jnu.edu.cn
Z. Li
lizu2020@gzhu.edu.cn

Specialty section:

This article was submitted to
Ceramics and Glass,
a section of the journal
Frontiers in Materials

Received: 14 November 2021

Accepted: 03 December 2021

Published: 03 January 2022

Citation:

Zhang M, Song YQ, Lin HJ, Li Z and
Li W (2022) A Brief Introduction on the
Development of Ti-Based
Metallic Glasses.
Front. Mater. 8:814629.
doi: 10.3389/fmats.2021.814629

Ti-based metallic glasses (MGs) possess high specific strength, low elastic modulus, high elasticity, high wear and corrosion resistance, and excellent biocompatibility, which make them highly attractive as lightweight high-strength materials as well as biomaterials. However, the glass forming ability (GFA) of Ti-based MGs, particularly those bearing no toxic, noble, or heavy metals, that is, Be, Pd, or Cu alike, largely sets back their wide applications for the restricted critical glass forming size of these Ti-based MGs. In this review, the outlines in developing Ti-based MGs are delineated in order to provide an overall view on the efforts ever made to fabricate bulk size Ti-based MGs. The state of the art in the knowledge on the GFA of Ti-based MGs is briefly introduced, and possible directions for fabricating bulk size toxic and noble element free Ti-based MGs are discussed.

Keywords: Ti-based metallic glasses, glass forming ability, biomaterials, lightweight high-strength materials, composition design

BACKGROUND

The disordered atomic packing in metallic glasses (MGs) makes them drastically different from conventional crystalline alloys and exhibit novel physical, chemical, and mechanical properties (Wang et al., 2004; Wang, 2009; Zhang et al., 2019). For the high specific strength, corrosion resistance, and biocompatibility of Ti element, Ti-based MGs have been attracting intensive attentions for their potential applications as lightweight high-strength materials (Jiang et al., 2015) and biomaterials (Li and Zheng, 2016). Early observations of non-crystalline phases in titanium alloys were made in binary TiCu and TiNi alloys in the late 1960s *via* splat quenching (Polk et al., 1978), for the large difference in atomic size between early transition metals (Ti) and late transition metals (Cu, Ni). Later on, lightweight Ti-based MG with a nominal composition of $Ti_{50}Be_{40}Zr_{10}$ (i.e., METGLAS 2204) was first reported by Tanner and Ray in 1977 in the TiBeZr ternary system (Tanner and Ray, 1977). To date, hundreds of Ti-based MGs have been found (Gong et al., 2016) with improved glass forming ability (GFA) and intriguing mechanical and chemical properties, which effectively support the application of Ti-based MGs. The GFA of MGs generally refers to the lowest cooling rate required for the undercooled liquid to bypass crystallization and become a glass and is conveniently measured with the critical size of an MG sample (often the diameter of a rod) that could be cast into a fully amorphous state. However, those Ti-based MGs with excellent GFA (i.e., large critical diameter) (Gong et al., 2016) usually contain toxic element—Be, noble metal—Pd, or heavy late transition metals—Cu or Ni. The presence of these elements, to a certain extent, impairs the application prospects of Ti-based MGs. To summarize the achievements ever made and illuminate the directions for future research, the main development paths of Ti-based

MGs are sorted out in this review. Based on the development paths of Ti-based MGs, 4 series of Ti-based MGs could be categorized according to the main composing elements, that is, TiCuNi series MGs, TiZrBe series MGs, TiZrCu series MGs, and TiZrSi series MGs.

In the following part, the successful preparation of each series of Ti-based MGs will be introduced in more detail, respectively. Here, we would like to emphasize that this review focuses mainly on the history of the observation of Ti-based MGs and the effect of alloying elements on the critical diameter of Ti-based MGs. Statistical data on the criteria of GFA for Ti-based MGs, such as the width of the supercooled liquid region $\Delta T_x = T_x - T_g$, the reduced glass transition temperature $T_{rg} = T_g/T_l$, and the parameter $\gamma = T_x/(T_g + T_l)$, are not focused on, where T_g is the glass transition temperature, T_x is the crystallization temperature, and T_l is the liquidus temperature. Moreover, the mechanical, physical, and chemical properties of Ti-based MGs are not encompassed either. For details on these aspects of Ti-based MGs, the readers are referred to more comprehensive reviews (Jiang et al., 2015; Gong et al., 2016; Li and Zheng, 2016). In the end, based on a global view on the development of Ti-based MGs, potential routes are discussed for the fabrication of new toxic and noble element free Ti-based MGs with bulk size.

DEVELOPMENT PATHS OF TI-BASED MGS

TiCuNi Series

In early days, glass formation and their structures of Ti-based MGs were mainly focused on to recognize these new glass materials. For example, early research studies on Ti (Cu, Fe) binary MGs were made based on their hydrogen absorption behaviors (Maeland et al., 1978; Rodmacq et al., 1988). Crystallization kinetics in amorphous TiCu and TiNi binary systems were examined by Buschow in 1983 (Buschow, 1983a; Buschow, 1983b) and in TiPt by Gao and Wang (Yi Qun Gao and Whang, 1985) in 1985, and later on, those of $Ti_{74.8}Ni_{13.1}Cu_{12.1}$, $Ti_{50}Ni_5Cu_{45}$, $Ti_{70.8}Ni_{13.3}Cu_{12.3}Ge_{3.6}$, and $Ti_{66.6}Cu_{12.5}Ni_{13.6}Ge_{7.3}$ MGs were also analyzed (Šušić et al., 1986; Gao et al., 1989). Chemical short range orders in TiNi MGs (Fukunaga et al., 1983; Fukunaga et al., 1984) were found to be analogous to that in corresponding crystalline compounds. Coexistence of two glass phases was observed in $Ti_{61}Cu_{23}Ni_{16}$ and $Ti_{62.5}Cu_{12}Ni_{23}Si_{2.5}$ MGs (Duhaj et al., 1985). Electronic configurations in TiCu, TiCuAl, and TiCuSi MGs (Tanaka et al., 1988) were studied by Tanaka et al., and the results suggested that the chemical affinity of elements Si and Al for late and early transition metals is highly relevant to the GFA of these MGs.

Studies engaging in enhancing the GFA of Ti-based MGs rapidly emerged in the 1990s. Glass forming compositions in the ternary TiCuNi system with a wide supercooled liquid region (SLR) were reported in 1994 (Zhang et al., 1994). It was found that the glass forming composition range in the TiCuNi ternary system was with Ni from 0 to 50 at% and Cu from 0 to 75 at%, in which the $Ti_{50}Ni_{25}Cu_{25}$ MG showed the widest SLR of 55 K. Since a wide SLR indicates high thermal stability of supercooled liquid and possibly good GFA (Inoue, 2000), great efforts were

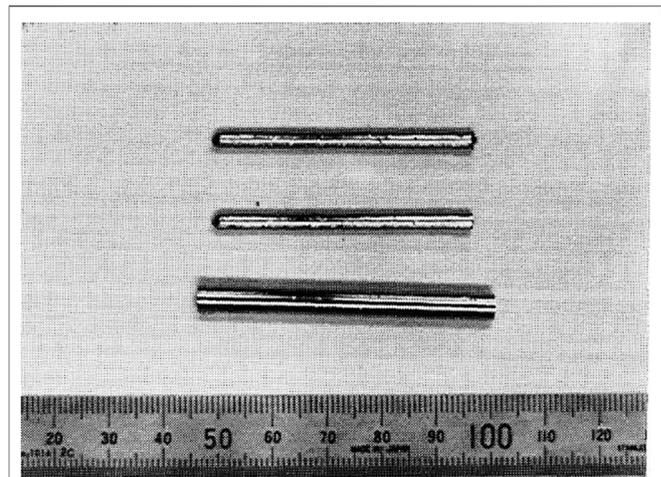
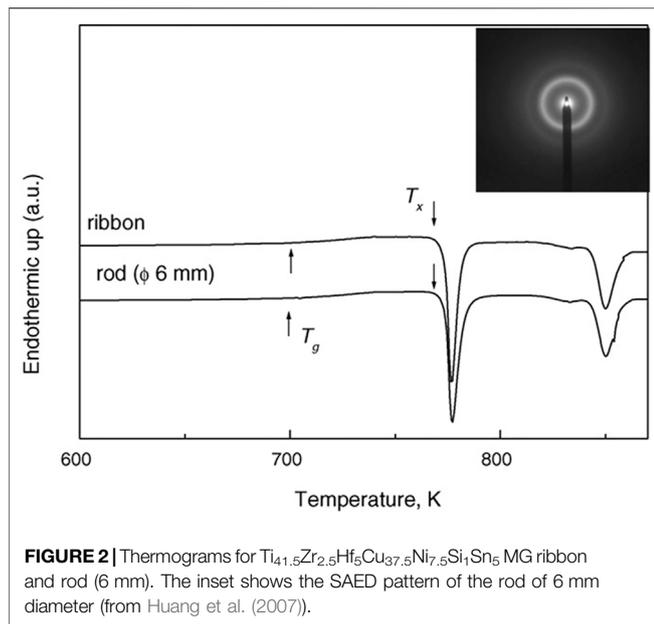


FIGURE 1 | $Ti_{50}Cu_{25}Ni_{15}Sn_5Zr_5$ MG rods of 3 and 6 mm diameter (from Zhang et al. (Zhang and Inoue, 1998)).

dedicated to the search for compositions with wider SLR. $Ti_{65}Ni_{25}Al_{10}$ and $Ti_{65}Cu_{25}Al_{10}$ MGs were also examined but to show no clear glass transition phenomenon before setting in of crystallization (Zhang et al., 1993). To widen the SLR of Ti-based MGs, different ternary systems Ti-Cu-(Ni, Co, Fe, and Si) were tried out (Inoue et al., 1994), whereas a quaternary $Ti_{50}Cu_{25}Ni_{20}Co_5$ MG was found to show an SLR of 90 K, but the GFA remained marginal. In 1998, the effects of Sn (and Zr) and Sb on the thermal stability of the ternary $Ti_{50}Cu_{25}Ni_{20}$ system were examined (Zhang and Inoue, 1998). With 5 at% replacement of Ni by Sn, $Ti_{50}Cu_{25}Ni_{20}Sn_5$ MG showed an SLR of 60 K. The authors argued that the similar size of Sn (atomic radius 0.141 nm) to Ti (0.147 nm) enabled Sn to replace Ti in its compounds and the replacement of Sn to Ti altered the crystallization mode and stabilized the supercooled liquid. However, with 3 at% replacement of Ni by Sb also of a similar size (0.145 nm) to Ti, the SLR was reduced to 45 K for unclear reasons. Intriguingly, as shown in **Figure 1**, it was also found that by further alloying with Zr, the $Ti_{50}Cu_{25}Ni_{15}Sn_5Zr_5$ MG showed a critical glass forming size of 6 mm rod diameter, indicating the crucial role of Zr in GFA. With slight composition variation, a $Ti_{50}Cu_{32}Ni_{15}Sn_3$ MG was found to show an SLR of 73 K and a critical diameter of 1 mm (Kim et al., 2001), suggesting the positive effect of Sn on GFA. In the meantime, for the smaller atomic sizes, metalloid elements Si (0.117 nm) and B (0.09 nm) were introduced to fabricate $Ti_{50}Cu_{20}Ni_{24}Sn_3B_1Si_2$ and $Ti_{50}Cu_{20}Ni_{24}B_2Si_4$ MGs (Zhang and Inoue, 1999; Zhang and Inoue, 2001) with SLRs of 74 and 65 K in width, respectively. $Ti_{50}Cu_{20}Ni_{24}Sn_3B_1Si_2$ MG could also be cast into rods of a diameter of 1 mm. The authors believed that this was because the addition of B and Si enabled the 2 MGs to satisfy the 3 empirical rules of Inoue (Inoue, 2000) for glass formation, that is, (1) multicomponent with more than 3 main elements, 2) significant differences in atomic size above 12% of main elements, and 3) negative mixing heats among main elements.



In 2002, based on $\text{Ti}_{50}\text{Cu}_{32}\text{Ni}_{15}\text{Sn}_3$ MG, Kim et al. (2002), by replacing Cu (0.128 nm) with Be (0.113 nm) which is of smaller size and has larger mixing heat with Ti, prepared the $\text{Ti}_{50}\text{Cu}_{25}\text{Ni}_{15}\text{Sn}_3\text{Be}_7$ MG with a critical diameter of 2 mm. The effects of Zr, Mo, and Ta on the glass formation of TiCuNiSn system were examined by He et al. (2003); He et al. (2004a); He et al. (2004b); He et al. (2006). However, mainly, amorphous ribbons could be fabricated. This was because the addition of Zr, Mo, and Ta lower the GFA by extending the melting interval and introducing nucleation sites for β -Ti solid solutions. Following Inoue's rule, Kim et al. (2003); Kim et al. (2004); Park et al. (2004); Kim et al. (2007) further proposed to add Be and Zr into the quaternary TiCuNiSn system, and $\text{Ti}_{45}\text{Cu}_{25}\text{Ni}_{15}\text{Sn}_3\text{Be}_7\text{Zr}_5$ and $\text{Ti}_{40}\text{Ni}_8\text{Cu}_9\text{Be}_{18}\text{Zr}_{25}$ MGs of critical diameters of 5 and 8 mm, but smaller SLR widths of 61 and 47 K were prepared. For the high content of Be and Zr, $\text{Ti}_{40}\text{Ni}_8\text{Cu}_9\text{Be}_{18}\text{Zr}_{25}$ MG actually enters the TiZrBe series and suggests the potential high GFA of TiZrBe series MGs. It also seems that GFA does not agree well with the width of SLR, indicating the decoupling between GFA and thermal stability of undercooled liquid.

Similar to previous studies, in 2004, Ma et al. (2004a) examined the effect of Zr, Hf, and Si on the GFA of $\text{Ti}_{50}\text{Cu}_{42.5}\text{Ni}_{7.5}$ MG and found that $\text{Ti}_{41.5}\text{Cu}_{42.5}\text{Ni}_{7.5}\text{Zr}_{2.5}\text{Hf}_5\text{Si}_1$ MG exhibits a critical diameter of 5 mm. The increased GFA was mainly because the addition of Zr moved the composition closer to the eutectic "valley" (lower liquid temperature and possibly higher T_{rg} prone to glass formation) and the addition of Si stabilized the supercooled liquid, leading to the formation of local atomic structures. They (Ma et al., 2004b) also examined the effects of B, Al, and Zr on the glass formation in the TiCuNiSi system and reported $\text{Ti}_{53}\text{Cu}_{15}\text{Ni}_{18}\text{Al}_7\text{Zr}_3\text{Si}_3\text{B}_1$ MG and $\text{Ti}_{53}\text{Cu}_{15}\text{Ni}_{18.5}\text{Al}_7\text{Zr}_3\text{Si}_3\text{B}_{0.5}$ MG with a critical diameter of 2.5 mm. They found that small atom B was important in enhancing GFA for interactions with other elements and more efficient packing mode. Sun et al. also characterized the effect of

small atom C (0.077 nm) on the GFA of $\text{Ti}_{50}\text{Cu}_{23}\text{Ni}_{20}\text{Sn}_7$ MG (Sun et al., 2005) but fabricated only MG composites. Later then, according to the topological glass formation model (Egami et al., 2007), early transition metals M (= Hf (0.160 nm), Zr (0.162 nm), and Sc (0.165 nm)) with large atomic sizes were introduced into $\text{Ti}_{53}\text{Cu}_{15}\text{Ni}_{18.5}\text{Al}_7\text{Zr}_3\text{Si}_3\text{B}_{0.5}\text{M}_3$ (Xia et al., 2005a; Xia et al., 2005b; Xia et al., 2005c; Xia et al., 2005d) to prepare new Ti-based BMGs with critical diameters of 2.5 mm, indicating no prominent effect of Hf and Sc on GFA. In 2007, based on Ma et al.'s work (Ma et al., 2004a), Huang et al. (2007) found that $\text{Ti}_{41.5}\text{Cu}_{37.5}\text{Ni}_{7.5}\text{Zr}_{2.5}\text{Hf}_5\text{Si}_1\text{Sn}_5$ MG showed the best ever GFA in TiCuNi series MGs with a critical diameter of 6 mm, as shown in Figure 2, indicating the enhancement of GFA by Sn addition. Interestingly, with an equal atomic ratio, that is, assisted by the high entropy effect (Greer, 1993) and the coexistence of Ti, Zr, and Be, TiCuNiZrHfBe MG shows a much increased critical diameter of 15 mm (Gong et al., 2015). However, TiNi (Al, Si, and Sn) ternary systems were re-examined to show limited GFA (Lu and Xu, 2008; Lu et al., 2009a; Lu et al., 2009b). By "3D pinpointing approach," Wang et al. (Wang and Xu, 2008) found a series of quaternary (Ti, Zr) CuNi MGs with a critical diameter of 3 mm and proposed that the substitution of Zr for Ti increased GFA for its atomic size difference with Ti and negative mixing heat with Ni and Cu. Recently, Chen and Hsu also examined the effect of Sn on the GFA of a $\text{Ti}_{44}\text{Cu}_{40}\text{Ni}_{16}$ alloy and reported a best glass former of $\text{Ti}_{44}\text{Cu}_{40}\text{Ni}_{16}\text{Sn}_1$ (Chen and Hsu, 2016) with a critical diameter of around 4 mm.

The development of TiCuNi series MGs was aimed at increasing the GFA and enlarging the SLR, that is, alloying with elements of large atomic size ratios, and attractive bonding nature to increase the difficulty in the redistribution of constituent elements for crystallization. This principle later developed into the 3 rules of Inoue. To enhance the GFA, C, Si, Sn, Sb, B, Zr, Hf, Sc, Be, Fe, Mo, Al, and Ta have been added into the TiCuNi system separately. The elements playing a positive role could be identified as Sn, Zr, Be, B, Si, and Hf. More importantly, as summarized before, coexistence of 2 or more elements would be crucial in enhancing the GFA of TiCuNi series MGs, for example, Zr + Sn (Zhang and Inoue, 1998; Wang and Xu, 2008), Zr + Be (Kim et al., 2003; Kim et al., 2004; Park et al., 2004; Kim et al., 2007), and Zr + Si (Ma et al., 2004a). This fact is worthy of further investigation for improving the GFA. However, the GFA of TiCuNi series MGs remains way small when compared to Zr-based (73 mm) (Lou et al., 2011) and Pd-based (85 mm) MGs (Nishiyama et al., 2012). On the other hand, the large amount of Cu and Ni largely lowers the specific strength of Ti-based MGs. To this situation, the TiZrBe series MGs with high GFA and lower density were rapidly developed in the past decades.

TiZrBe Series

The fabrication of TiZrBe series MGs dates back to the synthesis of METGLAS 2204 (Tanner and Ray, 1977) by Tanner and Ray with high specific strength and a stable enough supercooled liquid state for the preparation of continuous ribbon. MG formation in binary TiBe and ZrBe systems was reported later in 1979, and it was found that glass formation compositions were near the



FIGURE 3 | $\text{Ti}_{32.8}\text{Zr}_{30.2}\text{Cu}_9\text{Fe}_{5.3}\text{Be}_{22.7}$ MG rod with a diameter of 50 mm (from Zhang et al. (2015)).

eutectic point and that ZrBe exhibited better GFA than TiBe (Tanner and Ray, 1979). The formation of TiBe (Si, Al) (Tanner, 1978; Tanner et al., 1988) MGs and phase separation in the TiZrBe (Tanner and Ray, 1980) system were also studied. In 1993, with the preparation of Vitreloy 1 ($\text{Zr}_{41.2}\text{Ti}_{13.8}\text{Cu}_{12.5}\text{Ni}_{10}\text{Be}_{22.5}$ MG) (Peker and Johnson, 1993), the first commercialized bulk metallic glass (BMG, referring to metallic glass that could be cast into a sample of over 1 mm in all 3 dimensions), the TiZrBe series MGs were under intensive study for bulk specimen forming compositions.

In 2002, for the small atomic size (0.113 nm) and negative mixing heat (-30 k J/mol) with Ti, Be was used in making $\text{Ti}_{50}\text{Cu}_{25}\text{Ni}_{15}\text{Sn}_3\text{Be}_7$ MG with a critical diameter of 2 mm (Kim et al., 2002). In 2005, Guo et al. (2005), by tuning the Cu/Ni ratio, developed $\text{Ti}_{40}\text{Zr}_{25}\text{Be}_{20}\text{Ni}_3\text{Cu}_{12}$ MG, with an enhanced glass forming critical diameter of 14 mm. This was because the high Cu content changed the competing phase in glass formation from quasi-crystalline to crystalline, indicating the key role of Cu in the GFA of TiZrBe series MGs. Then, Hao et al. (2006); Hao et al. (2009) examined the effect of Y on the GFA of $\text{Ti}_{40}\text{Zr}_{25}\text{Be}_{20}\text{Ni}_3\text{Cu}_{12}$ and found that with 0.5 at% Y addition, an MG rod of 5 mm in diameter could be fabricated with low-purity raw materials as Y scavenged oxygen and impurities, leading to the formation of the Laves phase. The effect of Nb on the GFA of $\text{Ti}_{40}\text{Zr}_{25}\text{Be}_{18}\text{Cu}_9\text{Ni}_8$ MG was investigated separately in Shan et al. (2008); Li et al., 2014), and no prominent enhancement was observed. Mei et al. examined the effect of Ta on the GFA of $\text{Ti}_{40}\text{Ni}_8\text{Cu}_9\text{Be}_{18}\text{Zr}_{25}$ MG but found deterioration in GFA (Mei et al., 2007). For applications such as lightweight high-strength materials, Duan et al. (2008) reported the fabrication of lightweight $\text{Ti}_{45}\text{Zr}_{20}\text{Be}_{35}$ (4.59 g/cm³) and $\text{Ti}_{40}\text{Zr}_{25}\text{Be}_{30}\text{Cr}_5$ (4.76 g/cm³) MGs containing no late transition metals with critical diameters of 6 and 8 mm. Following the path of TiZrBe series MGs, the best glass formers in Ti-based MG, $\text{Ti}_{32.8}\text{Zr}_{30.2}\text{Be}_{22.7}\text{Ni}_{5.3}\text{Cu}_9$ MG by Tang et al. (2010) and $\text{Ti}_{32.8}\text{Zr}_{30.2}\text{Cu}_9\text{Fe}_{5.3}\text{Be}_{22.7}$ MG by Zhang et al. (2015), were

reported with a diameter over 50 mm comparable to the GFA of Zr-based (Lou et al., 2011) and Pd-based MGs (Nishiyama et al., 2012). **Figure 3** shows the 50-mm rod of $\text{Ti}_{32.8}\text{Zr}_{30.2}\text{Cu}_9\text{Fe}_{5.3}\text{Be}_{22.7}$ MG. The superior GFA was considered to be resulting from obedience to Inoue's rules and the confusion effect induced by the 5 constituent elements (Greer, 1993).

Gong et al., 2012a; Gong et al., 2012b; Gong et al., 2012c; Gong et al., 2013a; Gong et al., 2013b; Gong et al., 2013c; Gong et al., 2018; Gong et al., 2020) developed the TiZrBe (Fe, Al, Fe + Cu, Al + Cu, Co.) MGs, Zhao et al. (2014); Zhao et al. (2015a); Zhao et al. (2015b); Zhao et al. (2016) fabricated TiZrBe (Cu, Ni, Ag) MGs, and Gu et al. (2017); Gu et al. (2019a); Gu et al. (2019b) developed the TiZrBeNi and TiZrBeNi (Fe, Cu) MGs with high GFA of a critical diameter of centimeters. The high GFA of these MGs are generally ascribed to obedience to Inoue's rules and the large electronegativity of the composing elements. These MG systems largely enrich the family of TiZrBe series MGs. Lin et al. found that by adjusting the Ti/Zr ratio, the critical glass forming diameter of $\text{Ti}_{32.8}\text{Zr}_{30.2}\text{Be}_{22.7}\text{Ni}_{5.3}\text{Cu}_9$ MG could be enhanced to 20 mm (Lin et al., 2018). Similar results were also observed in $\text{Ti}_{35}\text{Zr}_{30}\text{Be}_{27.5}\text{Cu}_7$ MG (Song et al., 2019). This was attributed to the fact that the addition of Zr increases the content of icosahedral quasi-crystalline clusters, which stabilize the undercooled melt.

With the inclusion of light metal Be of small atomic size (0.111 nm in diameter), TiZrBe series MGs usually possess excellent GFA and are readily developed for lightweight structural materials with high strength. More efficiently, with the addition of Cu and another late transition metal like Fe and Ni, that is, coexistence of Cu + Ni, Cu + Fe, the critical diameter of TiZrBe-series MG would be improved to several centimeters. As stated before, the competing phases in glass formation, the mixing heat, the confusion effect, electronegativity, the atomic size ratio, the short range order of melt, etc. have been adopted to rationalize the high GFA of TiZrBe series MGs. However, the fundamental physics underpinning the superior GFA of TiZrBe-based MGs remains less clear. On the other hand, for the biotoxicity of Be and Ni, despite the excellent GFA in TiZrBe series MGs, the application of these MGs encountered great difficulty in the field of biomaterials. Thereby, to meet the demand of biomaterials, Ti-based MGs bearing no Ni and no Be were developed.

TiZrCu Series

TiZrCu series MGs as brazing filler metals were reported early in 1991 (Rabinkin et al., 1991). In 1994, by introducing Zr into the TiCuNi ternary system, Amiya et al., 1994) found that the TiZrCu ternary system and the TiZrCuNi quaternary system showed much improved thermal stability. For example, the amorphous $\text{Ti}_{50}\text{Cu}_{40}\text{Zr}_{10}$ powders prepared by high-pressure gas atomization showed a clear supercooled liquid region of 47 K. This is due to the large atomic size of Zr and the negative mixing heat of Zr-Cu, which increased the difficulty in the redistribution of constituent elements for crystallization. By introducing Zr and Ni into the TiCu binary system, for the atomic size mismatch and strong interaction induced, Men et al. (2005) reported the formation of

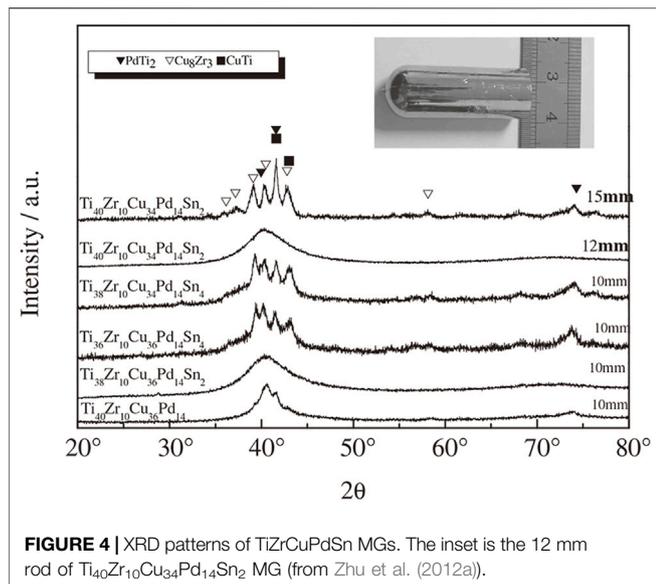


FIGURE 4 | XRD patterns of TiZrCuPdSn MGs. The inset is the 12 mm rod of $\text{Ti}_{40}\text{Zr}_{10}\text{Cu}_{34}\text{Pd}_{14}\text{Sn}_2$ MG (from Zhu et al. (2012a)).

$\text{Ti}_{45}\text{Zr}_5\text{Cu}_{45}\text{Ni}_5$ MG, with a critical diameter of 3 mm. However, $\text{Ti}_{45}\text{Zr}_5\text{Cu}_{45}\text{Ni}_5$ MG is still not Ni-free. As Pd and Ni belong to the same family, Zhu *et al.* proposed to replace Ni (0.125 nm) with Pd (0.137 nm) in the Ti–Cu–Ni–Zr system and prepared a Ni-free $\text{Ti}_{40}\text{Zr}_{10}\text{Cu}_{34}\text{Pd}_{16}$ MG, with a glass forming critical diameter of 6 mm (Zhu et al., 2007a; Zhu et al., 2007b). The improved GFA is due to the atomic size difference and the negative mixing heat between Pd and Ti, and Zr and Cu. By replacing Cu with Sn in $\text{Ti}_{40}\text{Zr}_{10}\text{Pd}_{14}\text{Cu}_{36}$ MG, as shown in **Figure 4**, Zhu *et al.* reported $\text{Ti}_{40}\text{Zr}_{10}\text{Cu}_{34}\text{Pd}_{14}\text{Sn}_2$ and $\text{Ti}_{40}\text{Zr}_{10}\text{Cu}_{32}\text{Pd}_{14}\text{Sn}_4$ MGs with maximized GFA of a critical diameter of 12 mm in conjunction with high plasticity and strength (Zhu et al., 2008; Zhu et al., 2012a). This was also due to the atomic size difference and the negative mixing heat induced by Sn, proving the positive effect of Sn on Ti-based MGs. For the atomic size and mixing heat, Xie et al. (2010a) examined the effect of large atom Sn (0.141 nm) and small atom Ni (0.125 nm) on the GFA of $\text{Ti}_{45}\text{Cu}_{37.8}\text{Zr}_{10}\text{Ni}_{7.2}$ MG and found ($\text{Ti}_{0.45}\text{Cu}_{0.378}\text{Zr}_{0.10}\text{Ni}_{0.072}$)₉₈ Sn_2 MG and $\text{Ti}_{43.15}\text{Zr}_{9.59}\text{Cu}_{36.24}\text{Ni}_{9.06}\text{Sn}_{1.96}$ MG (Xie et al., 2010b) with increased GFA of a critical diameter of 2–3 mm.

Lately, Wang et al. (2013a) investigated the effect of Co. on the GFA of $\text{Ti}_{40}\text{Zr}_{10}\text{Pd}_{14}\text{Cu}_{36}$ MG and reported the $\text{Ti}_{39}\text{Zr}_{10}\text{Pd}_{14}\text{Cu}_{36}\text{Co}_1$ MG and the ($\text{Ti}_{0.39}\text{Zr}_{0.10}\text{Pd}_{0.14}\text{Cu}_{0.36}$)₉₉ Co_1 MG with critical diameters of 10 and 8 mm respectively. The increased GFA for the Co containing MG was attributed to the formation of network composed of Co, Zr, and Pd atoms where the Co atoms connect with Zr and Pd atoms, as illustrated in **Figure 5**.

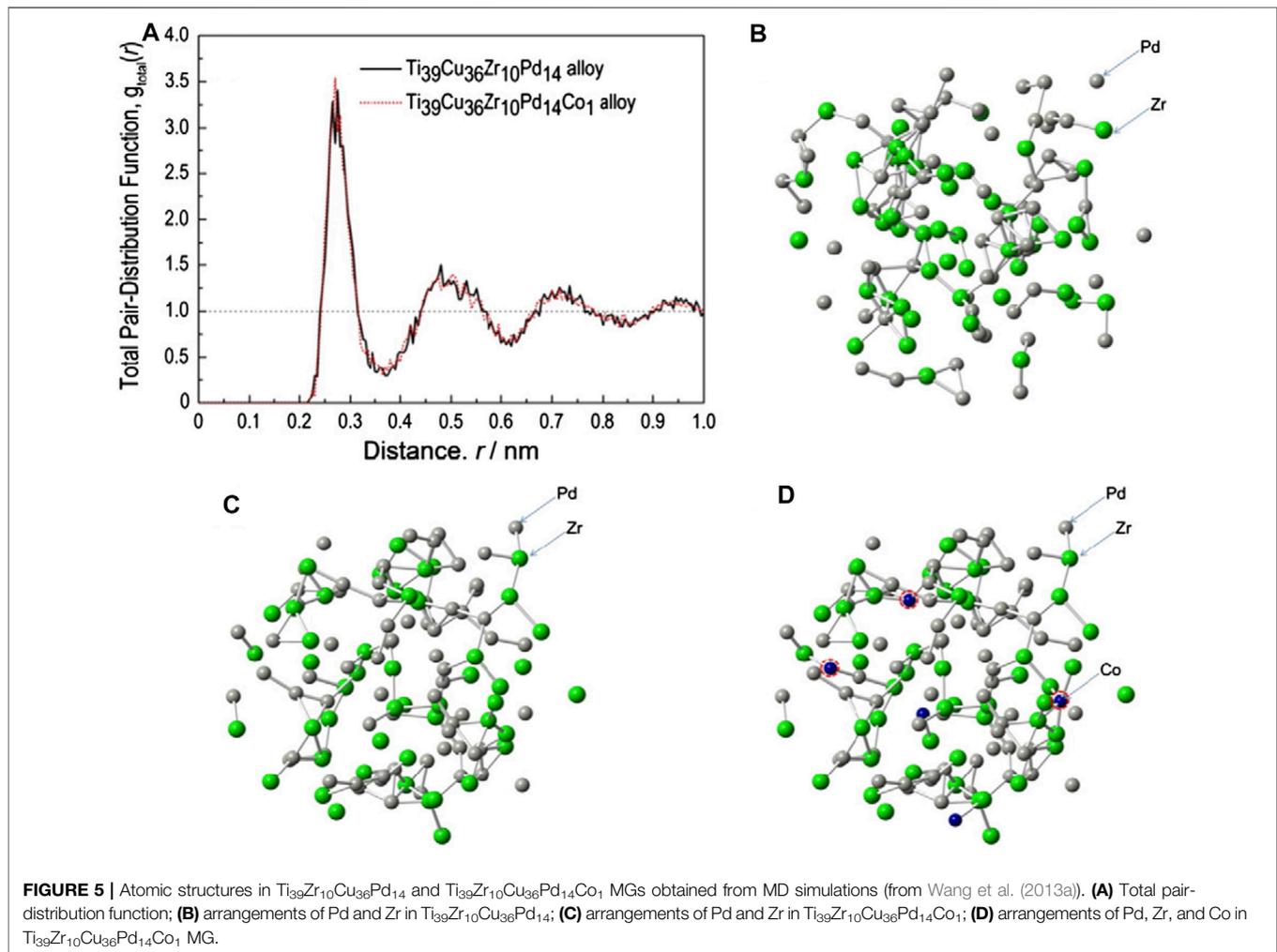
In 2010, Zhao *et al.* investigated the effect of Si addition on the GFA of $\text{Ti}_{43.89}\text{Cu}_{43.60}\text{Zr}_{6.75}\text{Ni}_{5.76}$ MG with a deep eutectic character (Zhao et al., 2010) and found that small addition (0.5–1 at%) would improve GFA for the negative mixing heat, the confusion effect, and possibly the scavenging effect of Si on O, and the destabilizing effect of Si on the TiO_2 cluster. By replacing Cu with Si, Zhu *et al.* found the formation of

$\text{Ti}_{40}\text{Cu}_{33}\text{Zr}_{10}\text{Pd}_{14}\text{Sn}_2\text{Si}_1$ MG with increased thermal stability of an SLR of 80 K but reduced GFA (Zhu et al., 2012b). This was possibly due to the increase in the liquidus temperature induced by Si addition. Yi *et al.* also reported the preparation of $\text{Ti}_{45.0}\text{Cu}_{40.1}\text{Zr}_{12.7}\text{Si}_{2.2}$ MG, with a critical diameter of 3 mm (Yi, 2012). However, Tsai *et al.* studied the small addition effect of Si on the GFA of $\text{Ti}_{40}\text{Zr}_{10}\text{Pd}_{14}\text{Cu}_{36}$ MG (Tsai et al., 2015) and reported only slight variation in GFA. It is inferred that the effect of Si on TiZrCu series MGs is more prone to the enhancement of thermal stability.

In 2008, Qin *et al.* found that micro-addition of Nb in $\text{Ti}_{40}\text{Zr}_{10}\text{Pd}_{14}\text{Cu}_{36}$ MG would lead to the formation of *in situ* Pd_3Ti nanoparticle MG composites (Qin et al., 2008). Oak *et al.* examined the effect of Nb and Ta addition on the GFA of $\text{Ti}_{45}\text{Zr}_{10}\text{Pd}_{10}\text{Cu}_{31}\text{Sn}_4$ MG (Oak et al., 2009; Oak et al., 2011) and also found decreased GFA. Qin *et al.* found that the addition of Ta reduced the GFA of $\text{Ti}_{40}\text{Zr}_{10}\text{Pd}_{14}\text{Cu}_{36}$ MG but increased the corrosion resistance and plasticity (Qin et al., 2012). Fornell *et al.* investigated the effect of Nb on another $\text{Ti}_{40}\text{Zr}_{10}\text{Cu}_{38}\text{Pd}_{12}$ MG and found the formation of CuTi_2 and CuTi nanoparticles (Fornell et al., 2013). Qin *et al.* (Qin et al., 2016) further explored the effects of Au, Pt, Nb, and Ta replacing Cu on the GFA of $\text{Ti}_{40}\text{Zr}_{10}\text{Pd}_{14}\text{Cu}_{34}\text{Sn}_2$ MG and found decayed GFA with the precipitation of Pd_3Ti nanoparticles. The authors attributed this fact to the decreased crystallization temperature after micro-addition of these elements. Yang *et al.* investigated the effect of Nb addition on the GFA of $\text{Ti}_{40}\text{Zr}_{10}\text{Cu}_{33}\text{Pd}_{14}\text{Sn}_3$ and found the precipitation of β -Ti and Ti_2Cu phases (Yang et al., 2017a). Wu et al. (2019) also observed decreased GFA in $\text{Ti}_{40}\text{Zr}_{10}\text{Cu}_{33}\text{Pd}_{14}\text{Sn}_3$ MG with the addition of Nb. It is seen that the addition of Nb is detrimental to the GFA, and the resulting glass formation competing phases are complicated. The effect of Nb on GFA might be due to the positive mixing heat of Nb–Ti, Nb–Zr, and Nb–Cu.

Different attempts for new Ti-based MGs were also made, but the reason for the reduced GFA was not clear. In 2017, Oak *et al.* reported the preparation of $\text{Ti}_{45}\text{Hf}_{10}\text{Pd}_{10}\text{Cu}_{30}\text{Sn}_5$ MG by replacing Zr with Hf in $\text{Ti}_{45}\text{Zr}_{10}\text{Pd}_{10}\text{Cu}_{30}\text{Sn}_5$ MG, however with reduced GFA (Oak et al., 2017). Jia et al. (2018) studied the effect of Ni and Pt replacing Pd on the GFA of $\text{Ti}_{41}\text{Cu}_{36}\text{Zr}_{10}\text{Pd}_{13}$ and also found the degradation of GFA. By replacing Cu with Ag, Nicoara *et al.* designed and prepared $\text{Ti}_{30}\text{Zr}_{32}\text{Ag}_7\text{Pd}_{24}\text{Sn}_7$ MG ribbons (Nicoara et al., 2018) with marginal GFA. Tantavisut *et al.* reported the fabrication of $\text{Ti}_{44}\text{Zr}_{10}\text{Pd}_{10}\text{Cu}_6\text{Co}_{23}\text{Ta}_7$, $\text{Ti}_{44}\text{Zr}_{10}\text{Pd}_{10}\text{Cu}_{10}\text{Co}_{19}\text{Ta}_7$, and $\text{Ti}_{44}\text{Zr}_{10}\text{Pd}_{10}\text{Cu}_{14}\text{Co}_{15}\text{Ta}_7$ MGs (Tantavisut et al., 2018). Bera *et al.* characterized the effect of Ga replacing Cu on the GFA of $\text{Ti}_{40}\text{Zr}_{10}\text{Pd}_{14}\text{Cu}_{36}$ MG and reported GFA in $\text{Ti}_{40}\text{Zr}_{10}\text{Cu}_{34}\text{Pd}_{14}\text{Ga}_2$ and $\text{Ti}_{40}\text{Zr}_{10}\text{Cu}_{32}\text{Pd}_{14}\text{Ga}_4$ MGs, with a diameter of 3 mm (Bera et al., 2019). They found that they could only obtain $\text{Ti}_{40}\text{Zr}_{10}\text{Pd}_{14}\text{Cu}_{36}$ MG of 2 mm diameter and attributed the GFA in $\text{Ti}_{40}\text{Zr}_{10}\text{Cu}_{32}\text{Pd}_{14}\text{Ga}_4$ MG to the larger atomic size and more negative mixing heat of Ga.

To remove Ni from $\text{Ti}_{47.4}\text{Cu}_{42}\text{Zr}_{5.3}\text{Ni}_{5.3}$ MG, Seki *et al.* prepared $\text{Ti}_{47.4}\text{Cu}_{42}\text{Zr}_{5.3}\text{TM}_{5.3}$ (TM = Co., Fe) MGs but with decreased thermal stability (Seki et al., 2008) due to the difference in the electronic state between Ni and Co or Fe. Yin *et al.*



examined the effect of Ni replacing Cu on $\text{Ti}_{45.8}\text{Zr}_{6.2}\text{Cu}_{45}\text{Sn}_2\text{Si}_1$ MG and that of Zr replacing Ti on $\text{Ti}_{52}\text{Cu}_{40.9}\text{Ni}_{5.1}\text{Sn}_2$ MG and prepared MGs, with a critical diameter of 4 mm (Yin et al., 2013). They proposed that the GFA of these MGs showing multiple crystallization events correlated strongly with the onset temperature of the last crystallization event. Using the cluster-plus-glue-atom model, Wang *et al.* designed 2 TiZrCuSn MGs (i.e., $(\text{Ti}_{7.2}\text{Zr}_{1.8})(\text{Cu}_{8.72}\text{Sn}_{0.28})$ and $(\text{Ti}_{7.2}\text{Zr}_{1.8})(\text{Cu}_{8.45}\text{Sn}_{0.55})$), with a critical glass forming diameter of 5 mm (Wang et al., 2013b), and later $\text{Ti}_{40}\text{Zr}_{10}\text{Cu}_{56.94}\text{Sn}_{3.06}$ and $\text{Ti}_{45.71}\text{Zr}_{11.43}\text{Cu}_{39.29}\text{Sn}_{3.57}$ MGs (Wang et al., 2016). However, for the high atomic fraction of Cu, they are more like Cu-based MGs, except for $\text{Ti}_{45.71}\text{Zr}_{11.43}\text{Cu}_{39.29}\text{Sn}_{3.57}$ MG. Based on Inoue's rules and the effective atomic radius, Shirasawa *et al.* prepared TiZrCo ternary MGs around a predicted near eutectic $\text{Ti}_{44}\text{Zr}_{30}\text{Co}_{26}$ MG (Shirasawa et al., 2014), as shown in **Figure 6**. These results provide possible routes for the design of new Ti-based MGs. Cao *et al.* studied the effect of nitrogen on the GFA of $\text{Ti}_{42.5}\text{Cu}_{40}\text{Zr}_{10}\text{Ni}_5\text{Sn}_{2.5}$ and reported that the addition of N of 0.1 at% could facilitate the formation of glass by suppressing the formation of the competing eutectic structure (Cao et al., 2016). Kuball *et al.* reported the fabrication of a new

class of sulfur-bearing Ti-based $\text{Ti}_{75}\text{Ni}_{17}\text{S}_8$ and $\text{Ti}_{40}\text{Zr}_{35}\text{Cu}_{17}\text{S}_8$ MGs (Kuball et al., 2018; Kuball et al., 2019). The effect of S on GFA was interpreted with its smaller size, higher mixing heat with other elements, low solubility in intermetallics, and the tendency for formation of covalent bonds. Introducing small metalloid atoms could be new paths for the development of Ti-based MGs.

In 2014, Hu *et al.* reported $\text{Ti}_{47}\text{Zr}_{7.5}\text{Cu}_{40}\text{Co}_{2.5}\text{Sn}_2\text{Si}_1$ and $\text{Ti}_{45}\text{Zr}_{7.5}\text{Cu}_{42}\text{Co}_{2.5}\text{Sn}_2\text{Si}_1$ MGs, with a critical diameter of 3 mm (Hu et al., 2014). Coincidentally, Wang *et al.* also studied the effect of Zr and Si on the GFA of $\text{Ti}_{46}\text{Cu}_{44}\text{Co}_7\text{Sn}_3$ MG and reported the $\text{Ti}_{46}\text{Zr}_{11.5}\text{Cu}_{31.5}\text{Co}_7\text{Sn}_3\text{Si}_1$ MG, with a critical diameter of 3 mm (Wang et al., 2015a). The improved GFA was due to the atomic size difference and mixing heat induced by Zr and Si. Later, they examined the effect of Ag replacing Cu on the GFA of $\text{Ti}_{46}\text{Cu}_{31.5}\text{Zr}_{11.5}\text{Co}_7\text{Sn}_3\text{Si}_1$ MG and found that $\text{Ti}_{46}\text{Cu}_{27.5}\text{Zr}_{11.5}\text{Co}_7\text{Sn}_3\text{Si}_1\text{Ag}_4$ MG showed a critical diameter of 4 mm (Wang et al., 2015b). This was inspired by the fact that Pang et al. (2015) reported the preparation of $\text{Ti}_{47}\text{Cu}_{38}\text{Zr}_{7.5}\text{Fe}_{2.5}\text{Sn}_2\text{Si}_1\text{Ag}_2$ MG with a diameter of 7 mm, as shown in **Figure 7**, the largest Ni-, Be-, and Pd-free Ti-based MG as the addition of silver promoted the formation of icosahedral clusters in the undercooled melt, which suppressed the nucleation

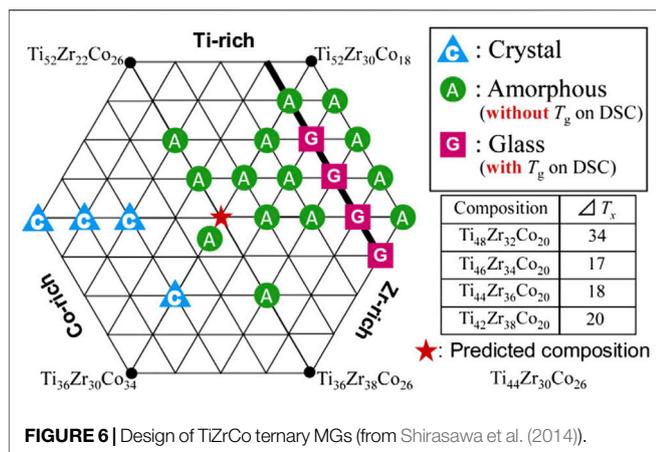


FIGURE 6 | Design of TiZrCo ternary MGs (from Shirasawa et al. (2014)).

of competing crystalline phases. Yan *et al.* investigated the effect of Nb replacing Cu on the GFA of $Ti_{47}Cu_{38}Zr_{7.5}Fe_{2.5}Sn_2Si_1Ag_2$ MG, but observed reduced GFA of a critical diameter of 3 mm (Yan et al., 2018). This result was ascribed to the positive mixing heat of Nb–Ti, Nb–Zr, and Nb–Cu. The effect of (Ti + Zr)/Cu ratio on the glass formation in $(TiZrCu)_{95}Fe_{2.5}Sn_2Si_1$ MG was also studied and was found that $Ti_{47}Zr_{7.5}Cu_{40}Fe_{2.5}Sn_2Si_1$ MG exhibited good GFA, with a critical diameter of 3 mm (Liu et al., 2016a). Later, they examined the effect of Sc replacing Ti on the GFA of $Ti_{47}Cu_{40}Zr_{7.5}Fe_{2.5}Sn_2Si_1$ MG and reported best glass formers $Ti_{45}Cu_{40}Zr_{7.5}Fe_{2.5}Sn_2Si_1Sc_2$ and $Ti_{44}Cu_{40}Zr_{7.5}Fe_{2.5}Sn_2Si_1Sc_3$ MGs with a critical diameter of 6 mm (Liu et al., 2016b), for the scavenging effect of Sc and its negative mixing heat with Si, Sn, Zr, and Cu. They also found that Ta replacing Cu reduced the GFA of $Ti_{47}Cu_{38}Zr_{7.5}Fe_{2.5}Sn_2Si_1Ag_2$ MG (Liu et al., 2020a) for the positive mixing heat of Ta–(Ti, Zr, Cu). Yang *et al.* from the same group also examined the GFA in (Ti, Cu, Zr) $_{92.5}Fe_{2.5}Sn_2Si_1Ag_2$ alloys with composition adjustment between Ti, Cu, and Zr (Yang et al., 2021). Wang *et al.* examined the effect of Pd replacing Cu on the GFA of $Ti_{47}Cu_{38}Zr_{7.5}Fe_{2.5}Sn_2Si_1Ag_2$ MG (Wang et al., 2021) and found reduced GFA for the increased liquidus temperature with Pd addition. However, the critical diameter was still no more than 7 mm.

The search for Be-, Ni-, and Pd-free TiZrCu series MG with high GFA has been a hot topic up to now. In the past decades, great efforts have been made to reduce the content of Pd in $Ti_{40}Zr_{10}Cu_{34}Pd_{14}Sn_2$ MG and increase the GFA. For this purpose, vastly different elements have been alloyed with TiZrCu series MGs, for example, Pd, Sn, Si, Ni, Nb, Hf, Ta, Fe, Ag, Co, N, S, and Sc. Also, coexistences of Pd + Sn and Pd + Co. would benefit glass formation. However, just like the situation in TiZrBe and TiCuNi series MGs, the fundamental physics underlying glass formation in Ti-based MGs remains elusive as vastly different concepts have been adopted to rationalize the intractable variation of GFA with the addition of alloying elements. Particularly, for the septenary and octonary alloys, understanding the GFA becomes increasingly difficult. On the other hand, the large atomic fraction of copper which would lead to adverse tissue reactions (Long and Rack, 1998)

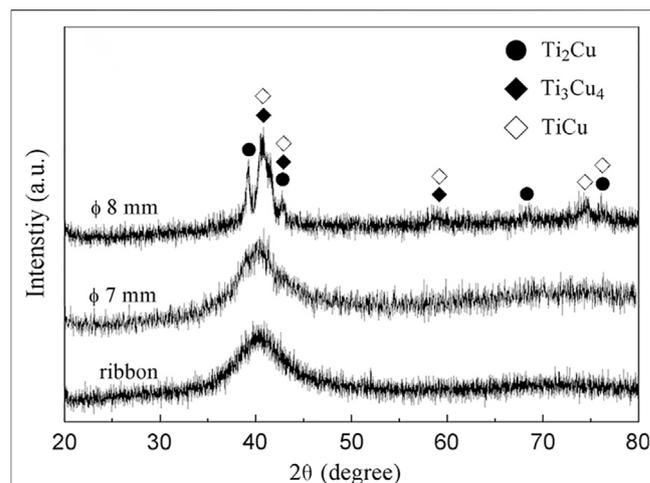


FIGURE 7 | XRD patterns of $Ti_{47}Cu_{38}Zr_{7.5}Fe_{2.5}Sn_2Si_1Ag_2$ MG (from Pang et al. (2015)).

and also poses a disadvantage for the application of TiZrCu series MGs as biomaterials.

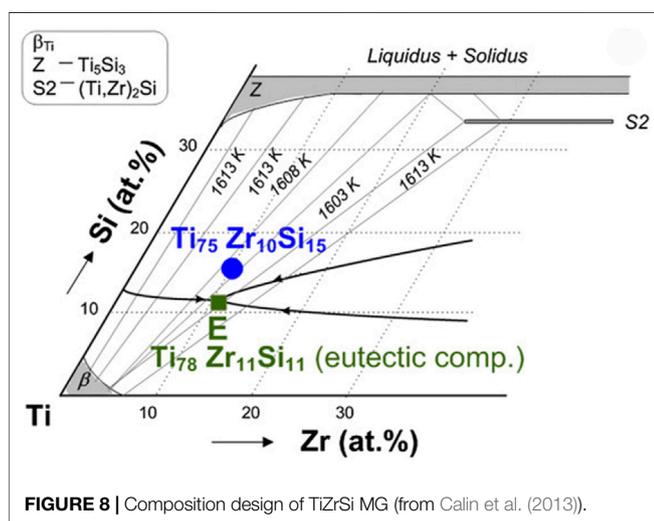
TiZrSi Series

MGs composed merely of non-toxic elements are highly desirable for biomedical applications. The TiZrSi system (Whang et al., 1985), an important system in Ti-based crystalline alloys, was focused on to this motive. However, compared to other 3 series of Ti-based MGs, the TiZrSi series MGs are highly vulnerable in glass formation. Polk et al. (1978) examined the glass formation behaviors in TiSi and TiNi binary and TiNiSi ternary systems and reported a typical $Ti_{60}Ni_{30}Si_{10}$ MG processable by melt spinning but with considerable Ni content. They proposed that low liquidus temperature, chemical difference, and size difference were critical to glass formation. In 1980, Suryanarayana *et al.* reported the formation of binary TiSi MG containing 15–20 at% Si (Suryanarayana et al., 1980). Inoue *et al.* reported the preparation of TiNb(Ta)Si(Mo, Rh, Ru, Ir, B, C, Ge) MG for the research on superconducting in MGs (Inoue et al., 1980a; Inoue et al., 1980b; Inoue et al., 1981). In 1992, Tarasova investigated the short range order of $Ti_{64}Zr_{22}Si_{14}$ MG (Tarasova et al., 1992). Oak *et al.* prepared thin ribbon $(Ti_xZr_yTa_z)_{85}Si_{15}$ MGs and found that the $Ti_{60}Zr_{10}Ta_{15}Si_{15}$ MG ribbon of 20 μm in thickness exhibited both high ductility and ultimate strength (Oak and Inoue, 2007). In 2008, Oak *et al.* not only examined the glass formation in TiZrPdSi system but also prepared amorphous ribbons of 20 μm thickness (Oak and Inoue, 2008), despite the fact that the TiZrPdSi system obeyed Inoue's 3 rules. The authors argued that the TiZrPdSi system would be a possible seed system for the development of Ti-based MGs bearing no toxic elements. Bai *et al.* prepared the biocompatible $Ti_{70}Zr_6Fe_7Si_{17}$ MG and $Ti_{64}Zr_5Fe_6Si_{17}Mo_6Nb_2$ MG and predicted that the addition of Mo and Nb should increase the GFA for the obedience to Inoue's rules, but the specimen remained in the ribbon shape with a thickness of 80 μm (Bai et al., 2008). As shown in Figure 8, Calin

TABLE 1 | Best glass formers of several metallic glasses.

Composition	Critical diameter (mm)	Method	Year	Ref
Hf ₄₃ Zr ₄ Cu _{31.25} Ni _{9.75} Al ₁₂	18	Water quenching	2021	Saini et al. (2021)
Hf ₄₀ Zr ₉ Cu ₃₁ Ni ₆ Al ₁₂				
Ni ₆₀ Pd ₂₀ P ₁₄ Si ₂ B ₄	25	Fluxing + water quenching	2020	Zeng et al. (2020)
Cu ₄₆ Zr _{33.5} Hf _{13.5} Al ₇	28.5	Water quenching	2020	Saini et al. (2020)
Gd ₅₅ Co _{17.5} Al _{27.5}	8	Copper mold casting	2019	Shao et al. (2019)
Al ₈₆ Ni _{6.75} Co _{2.25} Y _{3.25} La _{1.75}	2.5	Fluxing + copper mold casting	2017	Yang et al. (2017b)
Ce ₇₀ Ga ₈ Cu ₂₂	20	Copper mold casting	2015	Zhou et al. (2015)
Cr ₄₅ Fe ₁₁ Co ₇ Mo ₁₄ C ₁₅ B ₆ Y ₂	8	Copper mold casting	2015	Xu et al. (2015)
Cr ₃₀ Fe ₂₆ Co ₇ Mo ₁₄ C ₁₅ B ₆ Y ₂				
(Fe _{0.8} Co _{0.2}) ₄₇ Cr ₁₅ Mo ₁₄ C ₁₅ B ₆ Tm ₃	18	Copper mold casting	2013	Suryanarayana and Inoue (2013)
Pd _{42.5} Cu ₃₀ Ni _{7.5} P ₂₀	80	Fluxing + water quenching	2012	Nishiyama et al. (2012)
Ag _{38.5} Mg _{30.8} Ca _{23.1} Cu _{7.7}	4	Inverted injection casting	2012	Laws et al. (2012)
Zr ₄₆ Cu _{30.14} Ag _{8.36} Al ₆ Be _{7.5}	73	Copper mold casting	2011	Lou et al. (2011)
Co ₄₃ Fe ₅ Cr ₁₅ Mo ₁₄ C ₁₅ B ₆ Er ₂	15	Copper mold casting	2011	Zhang et al. (2011)
Au ₄₀ Si _{17.20} Cu _{28.30} Ag _{5.7} Pd ₅	6	Copper mold casting	2009	Guo et al. (2009)
Mg _{59.5} Cu _{22.9} Ag _{6.6} Gd ₁₁	27	Copper mold casting	2007	Zheng et al. (2007)
La ₆₅ Al ₁₄ (Cu ₅ /eAg _{1/6}) ₁₁ Ni ₅ Co ₅	30	Water quenching	2007	Jiang et al. (2007)
Pt _{42.5} Cu ₂₇ Ni _{9.5} P ₂₁	20	Fluxing + water quenching	2004	Schroers and Johnson (2004)
Ca ₆₅ Mg ₁₅ Zn ₂₀	15	Copper mold casting	2004	Park and Kim (2004)
Y ₃₆ Sc ₂₀ Al ₂₄ Co ₂₀	25	Water quenching	2003	Guo et al. (2003)

The numbers in the compositions are subscripts.

**FIGURE 8** | Composition design of TiZrSi MG (from Calin et al. (2013)).

et al. developed Ti₇₅Zr₁₀Si₁₅ and Ti₆₀Nb₁₅Zr₁₀Si₁₅ MG ribbons of 30 μm in thickness with better biocompatibility and found that the addition of Nb increases the GFA for atomic mismatch and negative mixing heat of Nb-Si (Calin et al., 2013).

Other alloys, for example, Ti₆₄Zr₁₀Si₁₅Nb₁₁ and Ti₅₆Zr₁₀Si₁₅Nb₁₉, were also prepared in ribbons but containing nanoparticles (Gabor et al., 2019). Huang *et al.* reported the fabrication of thin ribbon Ti₄₂Zr₄₀Si₁₅Ta₃ and Ti₄₀Zr₄₀Si₁₅Cu₅ MGs of 80–100 μm thickness (Huang et al., 2014; Huang et al., 2016) and found that for using as biomaterials, the content of Cu should be less than 5 at%, although it benefited GFA. Wu *et al.* investigated the effect of Sn on the GFA of Ti₆₀Zr₁₀Ta₁₅Si₁₅ MG (Wu et al., 2014) and found that the addition of Sn reduced the GFA. The authors attributed this fact to the relative decrease in the Si content with the addition of Sn, which reduces the atom pairs (Ti-Si) of negative mixing heat and is of great atomic size

mismatch. Guo *et al.* fabricated the TiFeSi, TiFeSi(Zr,Pd,Ge) MGs (Guo et al., 2015) near the eutectic composition Ti₆₅Fe₃₀Si₅ and found that small additions of Zr will ease glass formation, but the presence of Ge and Pd promotes crystallization for the atomic size mismatch. Abdi *et al.* studied the Ti₇₅Zr₁₀Si₁₅ and Ti₆₀Zr₁₀Nb₁₅Si₁₅ alloys and found that by melt spinning only glassy matrix composites of 30–50 μm in thickness could be obtained (Abdi et al., 2016). These results suggest that the GFA of Ti₇₅Zr₁₀Si₁₅ and Ti₆₀Zr₁₀Nb₁₅Si₁₅ MGs is not robust. Although poor GFA, the Ti₆₀Nb₁₅Zr₁₀Si₁₅ MG thin film was found to exhibit excellent biocompatibility (Thanka Rajan et al., 2019). Based on a design approach for high entropy alloys, Calin et al. (2021) prepared Ti₂₀Zr₂₀Nb₂₀Hf₂₀Si₂₀, Ti₃₀Zr₂₅Nb₂₅Si₁₅Ga₃B₂, and Ti₂₀Zr₂₀Nb₂₀Hf₂₀Si₁₅Ga₃B₂ alloys to obtain better glass former; however, slight crystalline phases precipitated in the glassy matrix.

Compared to the other 3 series of MGs, the GFA of TiZrSi series MGs is quite poor, as evidenced by the fact that only ribbon samples could be fabricated in the discovered compositions. This is largely because aiming for biomaterials, the development of TiZrSi series MGs excludes the presence of most late transition metals which are either adverse to the human body or noble. According to the empirical rules for glass formation, the late transition metals of medium atomic sizes compared to large early transition metal atoms and small metalloid atoms are critical in the GFA of MGs. As evidence suggests, the best glass formers in other MG systems are given in **Table 1**. It is seen that these alloy systems generally consist of early transition metals (ETMs), late transition metals (LTM), metal elements, and metalloids. More importantly, late transition metals (Cu, Ni, Co., Zn, Ag, Pd, etc.) are a prerequisite for their excellent GFA either as the base element or as the main constituents. As also observed in Zhao et al. (2015b), compared to electronegativity and mixing enthalpy, larger atomic size difference generally leads to higher GFA and seems more dominant for Ti-based MGs, whereas the mixing enthalpy and the electronegativity are also important. Thereby, to

make a breakthrough and develop new TiZrSi series MGs, more advanced and instructive understandings on the GFA of Ti-based MGs is of top priority. On the other hand, as a compromise for the development of biomaterials, addition of expensive elements Pd, Hf, Ta, etc. would be tolerable on a condition that enough GFA could be achieved. Nevertheless, there is still a long journey before the fabrication of toxic element free Ti-based MGs.

CONCLUDING REMARKS

In retrospect, on the development paths of Ti-based MGs, it is seen that the basic method for the design of MG forming compositions mainly includes the following 3 steps: i) according to the 3 empirical rules of Inoue, usually 3 main composing elements are chosen by their atomic sizes and the mixing heat between them; ii) the eutectic compositions are usually taken as guides for the base composition; and iii) the addition of minor alloying elements chosen by Inoue's rules or substituting the main elements with similar elements by the confusion principle was performed to optimize the GFA of the base composition with trial and error." However, the factors that play a non-trivial role on GFA are far more than those covered by the 3 empirical rules of Inoue—the eutectic composition, and the confusion principle. This fact significantly reduces the efficiency of the basic method. In this sense, in order to develop novel Ti-based MGs, as presented before, of top priority is to advance current understandings on the GFA of MGs—for instance, the coexistence of elements on GFA. More recently, the novel strong correlation between GFA and the full width at half maximum of the first diffraction peak of MGs (Li et al., 2021a) would help rapidly identify best glass formers.

On the other hand, based on the advances in artificial intelligence and manufacture technology, there could be 2 other ways around the aforementioned difficulty to fabricate bulk size Ti-based MGs. First is the machine learning approach (Sun et al., 2017). As outlined in *Development paths of Ti-based MGs*, large quantity of data on the glass formation behaviors of Ti-based MGs have been accumulated in the past half century. For the large number of Ti-based MG compositions observed and the many different alloying elements adopted, it is pretty difficult to manually establish a self-consistent GFA model compatible with all the data. In recent years, machine learning models for materials design boom up because of its ability in analyzing large-volume and high-dimension data and have proved to be a promising approach for finding MG compositions with good GFA

(Sun et al., 2017; Liu et al., 2020b; Xiong et al., 2021). With machine learning models, based on the glass forming compositions and the non-glass forming compositions accumulated to date, potential Ti-based MGs with high GFA could be more conveniently found.

Second is the 3D printing technique (Zhang et al., 2021). It has been long noticed that the thermal stability and the GFA of MGs are usually decoupled. As the fundamental physics in glass formation is the suppression of crystal formation, a possible reason for the decoupling between GFA and thermal stability could arise from the discrepancy between the crystal nucleation activation energy and the crystal growth activation energy. We argue that those MGs with high thermal stability tend to have high crystal nucleation activation energy, which enables large undercooling of liquid, while those MGs with high GFA, besides high crystal nucleation activation energy, tend to have high crystal growth activation energy which suppresses the burst growth of crystal nuclei. This is partially validated by recent work on 3D printing of MGs (Ouyang et al., 2021). For the rapid cooling process in the 3D printing process, the thermal stability of MGs weighs much over the GFA. Thereby, by the 3D printing technique, with MG powders of excellent thermal stability, the size limit on the fabrication of Ti-based MGs would be ultimately relieved. A similar approach is the ultrasonic assisted forming (Ma et al., 2019), by which giant MGs are successfully prepared (Li et al., 2021b). This processing method also provides a potential way for the fabrication of bulk size Ti-based MGs containing no noble, no toxic, and no heavy metal elements as long as Ti-based MGs have high enough thermal stability to survive the processing period.

AUTHOR CONTRIBUTIONS

All authors have made substantial, direct, and intellectual contribution to the work and approved it for publication.

FUNDING

This work was financially supported by the National Nature and Science Foundation of China under grant No. 51701082, Guangzhou Science and Technology Plan Project under grant No.202102020821, and the Fundamental Research Funds for Central Universities.

REFERENCES

- Abdi, S., Oswald, S., Gostin, P. F., Helth, A., Sort, J., Baró, M. D., et al. (2016). Designing New Biocompatible Glass-Forming Ti75- X Zr10 Nb X Si15 (X = 0, 15) Alloys: Corrosion, Passivity, and Apatite Formation. *J. Biomed. Mater. Res.* 104, 27–38. doi:10.1002/jbm.b.33332
- Amiya, K., Nishiyama, N., Inoue, A., and Masumoto, T. (1994). Mechanical Strength and thermal Stability of Ti-Based Amorphous Alloys with Large Glass-Forming Ability. *Mater. Sci. Eng. A* 179–180, 692–696. doi:10.1016/0921-5093(94)90294-1
- Bai, L., Cui, C., Wang, Q., Bu, S., and Qi, Y. (2008). Ti-Zr-Fe-Si System Amorphous Alloys with Excellent Biocompatibility. *J. Non-Crystalline Sol.* 354, 3935–3938. doi:10.1016/j.jnoncrysol.2008.05.015
- Bera, S., Ramasamy, P., Şopu, D., Sarac, B., Zálešák, J., Gammer, C., et al. (2019). Tuning the Glass Forming Ability and Mechanical Properties of Ti-Based Bulk Metallic Glasses by Ga Additions. *J. Alloys Comp.* 793, 552–563. doi:10.1016/j.jallcom.2019.04.173
- Buschow, K. H. J. (1983). Stability and Electrical Transport Properties of Amorphous Ti1-xNixalloys. *J. Phys. F: Met. Phys.* 13, 563–571. doi:10.1088/0305-4608/13/3/006

- Buschow, K. H. J. (1983). Thermal Stability of Amorphous Ti-Cu Alloys. *Acta Metallurgica* 31, 155–160. doi:10.1016/0001-6160(83)90075-5
- Calin, M., Gebert, A., Ghinea, A. C., Gostin, P. F., Abdi, S., Mickel, C., et al. (2013). Designing Biocompatible Ti-Based Metallic Glasses for Implant Applications. *Mater. Sci. Eng. C* 33, 875–883. doi:10.1016/j.msec.2012.11.015
- Calin, M., Vishnu, J., Thirathipviwat, P., Popa, M.-M., Krautz, M., Manivasagam, G., et al. (2021). Tailoring Biocompatible Ti-Zr-Nb-Hf-Si Metallic Glasses Based on High-Entropy Alloys Design Approach. *Mater. Sci. Eng. C* 121, 111733. doi:10.1016/j.msec.2020.111733
- Cao, D., Wu, Y., Wang, H., Liu, X.-J., and Lu, Z. P. (2016). Effects of Nitrogen on the Glass Formation and Mechanical Properties of a Ti-Based Metallic Glass. *Acta Metall. Sin. (Engl. Lett.)* 29, 173–180. doi:10.1007/s40195-016-0374-5
- Chen, T.-H., and Hsu, Y.-K. (2016). Mechanical Properties and Microstructural of Biomedical Ti-Based Bulk Metallic Glass with Sn Addition. *Comput. Mater. Sci.* 117, 584–589. doi:10.1016/j.commatsci.2015.12.020
- Duan, G., Wiest, A., Lind, M. L., Kahl, A., and Johnson, W. L. (2008). Lightweight Ti-Based Bulk Metallic Glasses Excluding Late Transition Metals. *Scripta Materialia* 58, 465–468. doi:10.1016/j.scriptamat.2007.10.040
- Duhaj, P., Vlasák, G., and Svec, P. (1985). “The Study of Structural Changes in Ti-Based Metallic Glasses,” in *Rapidly Quenched Metals*. Editors S. Steeb and H. Warlimont (Elsevier), 767–770. doi:10.1016/b978-0-444-86939-5.50184-6
- Egami, T., Poon, S. J., Zhang, Z., and Keppens, V. (2007). Glass Transition in Metallic Glasses: A Microscopic Model of Topological Fluctuations in the Bonding Network. *Phys. Rev. B* 76, 024203. doi:10.1103/physrevb.76.024203
- Fornell, J., Pellicer, E., Van Steenberge, N., González, S., Gebert, A., Suriñach, S., et al. (2013). Improved Plasticity and Corrosion Behavior in Ti-Zr-Cu-Pd Metallic Glass with Minor Additions of Nb: An alloy Composition Intended for Biomedical Applications. *Mater. Sci. Eng. A* 559, 159–164. doi:10.1016/j.msea.2012.08.058
- Fukunaga, T., Hayashi, N., Kai, K., Watanabe, N., and Suzuki, K. (1983). Chemical Short-Range Structure of Ni_xTi_{1-x} (X = 0.26–0.40) alloy Glasses. *Physica B+C* 120, 352–356. doi:10.1016/0378-4363(83)90405-9
- Fukunaga, T., Watanabe, N., and Suzuki, K. (1984). Experimental Determination of Partial Structures in Ni₄₀Ti₆₀ Glass. *J. Non-Crystalline Sol.* 61–62, 343–348. doi:10.1016/0022-3093(84)90572-6
- Qin, F. X., Xie, G. Q., Zhu, S. L., and Dan, Z. H., 8th International Forum on Advanced Materials Science and Technology (IFAMST-8), Fukuoka Inst Technol, Fukuoka City, JAPAN, 2012, pp. 23–26.
- Gabor, C., Cristea, D., Velicu, I.-L., Bedo, T., Gatto, A., Bassoli, E., et al. (2019). Ti-Zr-Si-Nb Nanocrystalline Alloys and Metallic Glasses: Assessment on the Structure, Thermal Stability, Corrosion and Mechanical Properties. *Materials* 12, 1551. doi:10.3390/ma12091551
- Gao, Y.-Q., Liu, X., Wang, W., Lo, Y.-B., and Yang, X.-Q. (1989). Crystallization Behavior of Ti_{66.6}Ni_{13.6}Cu_{12.5}Ge_{7.3} Glass. *Mater. Sci. Eng. A* 108, 19–23. doi:10.1016/0921-5093(89)90401-2
- Gong, P., Deng, L., Jin, J., Wang, S., Wang, X., and Yao, K. (2016). Review on the Research and Development of Ti-Based Bulk Metallic Glasses. *Metals* 6, 264. doi:10.3390/met6110264
- Gong, P., Li, F., and Jin, J. (2020). Preparation, Characterization, and Properties of Novel Ti-Zr-Be-Co Bulk Metallic Glasses. *Materials* 13, 223. doi:10.3390/ma13010223
- Gong, P., Wang, S., Liu, Z., Chen, W., Li, N., Wang, X., et al. (2018). Lightweight Ti-Based Bulk Metallic Glasses with superior Thermoplastic Formability. *Intermetallics* 98, 54–59. doi:10.1016/j.intermet.2018.04.019
- Gong, P., Wang, X., Shao, Y., Chen, N., Liu, X., and Yao, K. F. (2013). A Ti-Zr-Be-Fe-Cu Bulk Metallic Glass with superior Glass-Forming Ability and High Specific Strength. *Intermetallics* 43, 177–181. doi:10.1016/j.intermet.2013.08.003
- Gong, P., Yao, K.-F., Wang, X., and Shao, Y. (2012). Centimeter-sized Ti-Based Bulk Metallic Glass with High Specific Strength. *Prog. Nat. Sci. Mater. Int.* 22, 401–406. doi:10.1016/j.pnsc.2012.10.007
- Gong, P., Yao, K., and Ding, H. (2013). Centimeter-Sized Ti-Based Quaternary Bulk Metallic Glass Prepared by Water Quenching. *Int. J. Mod. Phys. B* 27, 1350087. doi:10.1142/s0217979213500872
- Gong, P., Yao, K. F., and Ding, H. Y. (2015). Crystallization Kinetics of TiZrHfCuNiBe High Entropy Bulk Metallic Glass. *Mater. Lett.* 156, 146–149. doi:10.1016/j.matlet.2015.05.018
- Gong, P., Yao, K. F., and Shao, Y. (2012). Effects of Fe Addition on Glass-Forming Ability and Mechanical Properties of Ti-Zr-Be Bulk Metallic Glass. *J. Alloys Comp.* 536, 26–29. doi:10.1016/j.jallcom.2012.04.048
- Gong, P., Yao, K. F., and Shao, Y. (2012). Lightweight Ti-Zr-Be-Al Bulk Metallic Glasses with Improved Glass-Forming Ability and Compressive Plasticity. *J. Non-Crystalline Sol.* 358, 2620–2625. doi:10.1016/j.jnoncrysol.2012.06.011
- Gong, P., Yao, K., Wang, X., and Shao, Y. (2013). A New Centimeter-Sized Ti-Based Quaternary Bulk Metallic Glass with Good Mechanical Properties. *Adv. Eng. Mater.* 15, 691–696. doi:10.1002/adem.201200391
- Greer, A. L. (1993). Confusion by Design. *Nature* 366, 303–304. doi:10.1038/366303a0
- Gu, J.-L., Shao, Y., and Yao, K.-F. (2019). The Novel Ti-Based Metallic Glass with Excellent Glass Forming Ability and an Elastic Constant Dependent Glass Forming Criterion. *Materialia* 8, 100433. doi:10.1016/j.mtla.2019.100433
- Gu, J.-L., Shao, Y., Zhao, S.-F., Lu, S.-Y., Yang, G.-N., Chen, S.-Q., et al. (2017). Effects of Cu Addition on the Glass Forming Ability and Corrosion Resistance of Ti-Zr-Be-Ni Alloys. *J. Alloys Comp.* 725, 573–579. doi:10.1016/j.jallcom.2017.07.165
- Gu, J., Yang, X., Zhang, A., Shao, Y., Zhao, S., and Yao, K. (2019). Centimeter-sized Ti-Rich Bulk Metallic Glasses with superior Specific Strength and Corrosion Resistance. *J. Non-Crystalline Sol.* 512, 206–210. doi:10.1016/j.jnoncrysol.2018.10.034
- Guo, F., Poon, S. J., and Shiflet, G. J. (2003). Metallic Glass Ingots Based on Yttrium. *Appl. Phys. Lett.* 83, 2575–2577. doi:10.1063/1.1614420
- Guo, F., Wang, H.-J., Poon, S. J., and Shiflet, G. J. (2005). Ductile Titanium-Based Glassy alloy Ingots. *Appl. Phys. Lett.* 86, 091907. doi:10.1063/1.1872214
- Guo, H., Zhang, W., Qin, C., Qiang, J., Chen, M., and Inoue, A. (2009). Glass-Forming Ability and Properties of New Au-Based Glassy Alloys with Low Au Concentrations. *Mater. Trans.* 50, 1290–1293. doi:10.2320/matertrans.me200809
- Guo, Y., Bataev, I., Georganakakis, K., Jorge, A. M., Nogueira, R. P., Pons, M., et al. (2015). Ni- and Cu-free Ti-Based Metallic Glasses with Potential Biomedical Application. *Intermetallics* 63, 86–96. doi:10.1016/j.intermet.2015.04.004
- Hao, G. J., Zhang, Y., Lin, J. P., Wang, Y. L., Lin, Z., and Chen, G. L. (2006). Bulk Metallic Glass Formation of Ti-Based Alloys from Low Purity Elements. *Mater. Lett.* 60, 1256–1260. doi:10.1016/j.matlet.2005.11.011
- Hao, G., Ren, F., Zhang, Y., and Lin, J. (2009). Role of Yttrium in Glass Formation of Ti-Based Bulk Metallic Glasses. *Rare Met.* 28, 68–71. doi:10.1007/s12598-009-0013-7
- He, G., Eckert, J., and Hagiwara, M. (2004). Glass-forming Ability and Crystallization Behavior of Ti-Cu-Ni-Sn-M (M=Zr, Mo, and Ta) Metallic Glasses. *J. Appl. Phys.* 95, 1816–1821. doi:10.1063/1.1643776
- He, G., Eckert, J., and Hagiwara, M. (2006). Mechanical Properties and Fracture Behavior of the Modified Ti-Base Bulk Metallic Glass-Forming Alloys. *Mater. Lett.* 60, 656–661. doi:10.1016/j.matlet.2005.09.054
- He, G., Eckert, J., and Löser, W. (2003). Stability, Phase Transformation and Deformation Behavior of Ti-Base Metallic Glass and Composites. *Acta Materialia* 51, 1621–1631. doi:10.1016/s1359-6454(02)00563-3
- He, G., Löser, W., and Eckert, J. (2004). Devitrification and Phase Transformation of (Ti_{0.5}Cu_{0.25}Ni_{0.15}Sn_{0.05}Zr_{0.05})_{100-x}Mox Metallic Glasses. *Scripta Materialia* 50, 7–11. doi:10.1016/j.scriptamat.2003.09.049
- Hu, Q., Zhang, M., Li, H., Yin, E., Pang, S., and Zhang, T. (2014). Formation, Bio-Corrosion Behavior and Mechanical Properties of Ti-Zr-Cu-Co-Sn-Si Bulk Metallic Glasses. *J. Mater. Eng.*, 18–21. doi:10.11868/j.issn.1001-4381.2014.06.004
- Huang, C. H., Huang, Y. S., Lin, Y. S., Lin, C. H., Huang, J. C., Chen, C. H., et al. (2014). Electrochemical and Biocompatibility Response of Newly Developed TiZr-Based Metallic Glasses. *Mater. Sci. Eng. C* 43, 343–349. doi:10.1016/j.msec.2014.06.040
- Huang, C. H., Lai, J. J., Huang, J. C., Lin, C. H., and Jang, J. S. C. (2016). Effects of Cu Content on Electrochemical Response in Ti-Based Metallic Glasses under Simulated Body Fluid. *Mater. Sci. Eng. C* 62, 368–376. doi:10.1016/j.msec.2016.01.080

- Huang, Y. J., Shen, J., Sun, J. F., and Yu, X. B. (2007). A New Ti-Zr-Hf-Cu-Ni-Si-Sn Bulk Amorphous alloy with High Glass-Forming Ability. *J. Alloys Comp.* 427, 171–175. doi:10.1016/j.jallcom.2006.03.006
- Inoue, A., Kimura, H. M., Masumoto, T., Suryanarayana, C., and Hoshi, A. (1980). Superconductivity of Ductile Ti-Nb-Si Amorphous Alloys. *J. Appl. Phys.* 51, 5475–5482. doi:10.1063/1.327506
- Inoue, A., Masumoto, T., Suryanarayana, C., and Hoshi, A. (1980). Superconductivity of Ductile Titanium-Niobium-Based Amorphous Alloys. *J. Phys. Colloques* 41, C8–758C758761. doi:10.1051/jphyscol:19808189
- Inoue, A., Nishiyama, N., Amiya, K., Zhang, T., and Masumoto, T. (1994). Ti-based Amorphous Alloys with a Wide Supercooled Liquid Region. *Mater. Lett.* 19, 131–135. doi:10.1016/0167-577x(94)90057-4
- Inoue, A. (2000). Stabilization of Metallic Supercooled Liquid and Bulk Amorphous Alloys. *Acta Materialia* 48, 279–306. doi:10.1016/s1359-6454(99)00300-6
- Inoue, A., Takahashi, Y., Suryanarayana, C., Hoshi, A., and Masumoto, T. (1981). Crystallization-induced Superconductivity in Amorphous Ti-Ta-Si Alloys. *J. Mater. Sci.* 16, 3077–3086. doi:10.1007/bf00540315
- Jia, H., Xie, X., Zhao, L., Wang, J., Gao, Y., Dahmen, K. A., et al. (2018). Effects of Similar-Element-Substitution on the Glass-Forming Ability and Mechanical Behaviors of Ti-Cu-Zr-Pd Bulk Metallic Glasses. *J. Mater. Res. Tech.* 7, 261–269. doi:10.1016/j.jmrt.2017.08.009
- Jiang, J.-Z., Hofmann, D., Jarvis, D. J., and Fecht, H.-J. (2015). Low-Density High-Strength Bulk Metallic Glasses and Their Composites: A Review. *Adv. Eng. Mater.* 17, 761–780. doi:10.1002/adem.201400252
- Jiang, Q. K., Zhang, G. Q., Yang, L., Wang, X. D., Saksl, K., Franz, H., et al. (2007). La-based Bulk Metallic Glasses with Critical Diameter up to 30mm. *Acta Materialia* 55, 4409–4418. doi:10.1016/j.actamat.2007.04.021
- Kim, Y.-C., Kim, W. n. T., Kim, D.-H., and Kim, D.-H. (2002). Glass Forming Ability and Crystallization Behavior in Amorphous Ti50Cu32-xNi15Sn3Bex (X=0, 1, 3, 7) Alloys. *Mater. Trans.* 43, 1243–1246. doi:10.2320/matertrans.43.1243
- Kim, Y.-C., Yi, S. H., Kim, W. T., and Kim, D. H. (2001). Glass Forming Ability and Crystallization Behaviors of the Ti-Cu-Ni-(Sn) Alloys with Large Supercooled Liquid Region. *Msf* 360-362, 67–72. doi:10.4028/www.scientific.net/msf.360-362.67
- Kim, Y. C., Bae, D. H., Kim, W. T., and Kim, D. H. (2003). Glass Forming Ability and Crystallization Behavior of Ti-Based Amorphous Alloys with High Specific Strength. *J. Non-Crystalline Sol.* 325, 242–250. doi:10.1016/s0022-3093(03)00327-2
- Kim, Y. C., Chang, H. J., Kim, D. H., Kim, W. T., and Cha, P. R. (2007). Unusual Glass-Forming Ability Induced by Changes in the Local Atomic Structure in Ti-Based Bulk Metallic Glasses. *J. Phys. Condens. Matter* 19, 196104. doi:10.1088/0953-8984/19/19/196104
- Kim, Y. C., Kim, W. T., and Kim, D. H. (2004). A Development of Ti-Based Bulk Metallic Glass. *Mater. Sci. Eng. A* 375-377, 127–135. doi:10.1016/j.msea.2003.10.115
- Kuball, A., Gross, O., Bochtler, B., Adam, B., Ruschel, L., Zamanzade, M., et al. (2019). Development and Characterization of Titanium-Based Bulk Metallic Glasses. *J. Alloys Comp.* 790, 337–346. doi:10.1016/j.jallcom.2019.03.001
- Kuball, A., Gross, O., Bochtler, B., and Busch, R. (2018). Sulfur-bearing Metallic Glasses: A New Family of Bulk Glass-Forming Alloys. *Scripta Materialia* 146, 73–76. doi:10.1016/j.scriptamat.2017.11.011
- Laws, K. J., Shamlaye, K. F., and Ferry, M. (2012). Synthesis of Ag-Based Bulk Metallic Glass in the Ag-Mg-Ca-[Cu] alloy System. *J. Alloys Comp.* 513, 10–13. doi:10.1016/j.jallcom.2011.10.097
- Li, H. F., and Zheng, Y. F. (2016). Recent Advances in Bulk Metallic Glasses for Biomedical Applications. *Acta Biomater.* 36, 1–20. doi:10.1016/j.actbio.2016.03.047
- Li, H., Li, Z., Yang, J., Ke, H. B., Sun, B., Yuan, C. C., et al. (2021). Interface Design Enabled Manufacture of Giant Metallic Glasses. *Sci. China Mater.* 64, 964–972. doi:10.1007/s40843-020-1561-x
- Li, L., Liu, R., Zhao, J., Cai, H., and Yang, Z. (2014). Effects of Nb Addition on Glass-Forming Ability, Thermal Stability and Mechanical Properties of Ti-Based Bulk Metallic Glasses. *Rare Metal Mater. Eng.* 43, 1835–1838. doi:10.1016/s1875-5372(14)60141-7
- Li, M.-X., Sun, Y.-T., Wang, C., Hu, L.-W., Sohn, S., Schroers, J., et al. (2021). Data-Driven Discovery of a Universal Indicator for Metallic Glass Forming Ability. *Nat. Mater.* doi:10.1038/s41563-021-01129-6
- Lin, S., Liu, D., Zhu, Z., Li, D., Fu, H., Zhuang, Y., et al. (2018). New Ti-Based Bulk Metallic Glasses with Exceptional Glass Forming Ability. *J. Non-Crystalline Sol.* 502, 71–75. doi:10.1016/j.jnoncrysol.2018.06.038
- Liu, X., Li, X., He, Q., Liang, D., Zhou, Z., Ma, J., et al. (2020). Machine Learning-Based Glass Formation Prediction in Multicomponent Alloys. *Acta Materialia* 201, 182–190. doi:10.1016/j.actamat.2020.09.081
- Liu, Y., Pang, S., Li, H., Hu, Q., Chen, B., and Zhang, T. (2016). Formation and Properties of Ti-Based Ti-Zr-Cu-Fe-Sn-Si Bulk Metallic Glasses with Different (Ti + Zr)/Cu Ratios for Biomedical Application. *Intermetallics* 72, 36–43. doi:10.1016/j.intermet.2016.01.007
- Liu, Y., Wang, G., Li, H., Pang, S., Chen, K., and Zhang, T. (2016). Ti Cu Zr Fe Sn Si Sc Bulk Metallic Glasses with Good Mechanical Properties for Biomedical Applications. *J. Alloys Comp.* 679, 341–349. doi:10.1016/j.jallcom.2016.03.224
- Liu, Y., Wang, H.-J., Pang, S.-J., and Zhang, T. (2020). Ti-Zr-Cu-Fe-Sn-Si-Ag-Ta Bulk Metallic Glasses with Good Corrosion Resistance as Potential Biomaterials. *Rare Met.* 39, 688–694. doi:10.1007/s12598-018-1124-9
- Long, M., and Rack, H. J. (1998). Titanium Alloys in Total Joint Replacement-A Materials Science Perspective. *Biomaterials* 19, 1621–1639. doi:10.1016/s0142-9612(97)00146-4
- Lou, H. B., Wang, X. D., Xu, F., Ding, S. Q., Cao, Q. P., Hono, K., et al. (2011). 73 Mm-Diameter Bulk Metallic Glass Rod by Copper Mould Casting. *Appl. Phys. Lett.* 99, 051910. doi:10.1063/1.3621862
- Lu, B.-c., Li, Y., and Xu, J. (2009). Optimal Glass-Forming Composition and its Correlation with Eutectic Reaction in the Ti-Ni-Al Ternary System. *J. Alloys Comp.* 467, 261–267. doi:10.1016/j.jallcom.2007.12.050
- Lu, B.-C., Wang, Y.-L., and Xu, J. (2009). Revisiting the Glass-Forming Ability of Ti-Ni-Si Ternary Alloys. *J. Alloys Comp.* 475, 157–164. doi:10.1016/j.jallcom.2008.07.055
- Lu, B.-c., and Xu, J. (2008). Glass Formation of Ti-Ni-Sn Ternary Alloys Correlated with TiNi-Ti3Sn Pseudo Binary Eutectics. *J. Non-Crystalline Sol.* 354, 5425–5431. doi:10.1016/j.jnoncrysol.2008.09.016
- Ma, C., Ishihara, S., Soejima, H., Nishiyama, N., and Inoue, A. (2004). Formation of New Ti-Based Metallic Glassy Alloys. *Mater. Trans.* 45, 1802–1806. doi:10.2320/matertrans.45.1802
- Ma, C., Soejima, H., Ishihara, S., Amiya, K., Nishiyama, N., and Inoue, A. (2004). New Ti-Based Bulk Glassy Alloys with High Glass-Forming Ability and Superior Mechanical Properties. *Mater. Trans.* 45, 3223–3227. doi:10.2320/matertrans.45.3223
- Ma, J., Yang, C., Liu, X., Shang, B., He, Q., Li, F., et al. (2019). *Sci. Adv.* 5, eaax7256. doi:10.1126/sciadv.aax7256
- Maeland, A. J. (1978). “Comparison of Hydrogen Absorption in Glassy and Crystalline Structures,” in *Hydrides for Energy Storage*. Editors A. F. Andresen and A. J. Maeland (Pergamon), 447–462. doi:10.1016/b978-0-08-022715-3.50040-0
- Mei, J., Li, J., Kou, H., Hu, R., Fu, H., and Zhou, L. (2007). Effects of Ta Addition on the Microstructure and Mechanical Properties of Ti40Zr25Ni8Cu9Be18 Amorphous alloy. *J. Univ. Sci. Tech. Beijing Mineral, Metall. Mater.* 14, 31–35. doi:10.1016/s1005-8850(07)60103-0
- Men, H., Pang, S., Inoue, A., and Zhang, T. (2005). New Ti-Based Bulk Metallic Glasses with Significant Plasticity. *Mater. Trans.* 46, 2218–2220. doi:10.2320/matertrans.46.2218
- Nicoara, M., Buzdugan, D., Locovei, C., Bena, T., and Stoica, M. (2018). About Thermostability of Biocompatible Ti-Zr-Ag-Pd-Sn Amorphous Alloys. *J. Therm. Anal. Calorim.* 133, 189–197. doi:10.1007/s10973-018-7031-3
- Nishiyama, N., Takenaka, K., Miura, H., Saidoh, N., Zeng, Y., and Inoue, A. (2012). The World’s Biggest Glassy alloy Ever Made. *Intermetallics* 30, 19–24. doi:10.1016/j.intermet.2012.03.020
- Oak, J.-J., and Inoue, A. (2007). Attempt to Develop Ti-Based Amorphous Alloys for Biomaterials. *Mater. Sci. Eng. A* 449-451, 220–224. doi:10.1016/j.msea.2006.02.307
- Oak, J.-J., and Inoue, A. (2008). Formation, Mechanical Properties and Corrosion Resistance of Ti-Pd Base Glassy Alloys. *J. Non-Crystalline Sol.* 354, 1828–1832. doi:10.1016/j.jnoncrysol.2007.10.025
- Oak, J.-J., Louzguine-Luzgin, D. V., and Inoue, A. (2011). Fabrication of Ni-free Ti-based Bulk-Metallic Glassy alloy Having Potential for Application as Biomaterial, and Investigation of its Mechanical Properties, Corrosion, and Crystallization Behavior. *J. Mater. Res.* 22, 1346–1353. doi:10.1557/jmr.2007.0154

- Oak, J.-J., Louzguine-Luzgin, D. V., and Inoue, A. (2009). Investigation of Glass-Forming Ability, Deformation and Corrosion Behavior of Ni-free Ti-Based BMG Alloys Designed for Application as Dental Implants. *Mater. Sci. Eng. C* 29, 322–327. doi:10.1016/j.msec.2008.07.009
- Oak, J. J., Kim, Y. H., Bae, K. C., and Park, Y. H. (2017). Effect of Hafnium and Zirconium to Glass Forming Ability, Thermal Stability, Plasticity Deformation and Crystallization of Ni-free Pentabasic Ti-Based Bulk Metallic Glasses. *Arch. Metall. Mater.* 62, 1081–1087. doi:10.1515/amm-2017-0158
- Ouyang, D., Zhang, P. C., Zhang, C., and Liu, L. (2021). Understanding of crystallization behaviors in laser 3D printing of bulk metallic glasses. *Appl. Mater. Today* 23, 100988. doi:10.1016/j.apmt.2021.100988
- Pang, S., Liu, Y., Li, H., Sun, L., Li, Y., and Zhang, T. (2015). New Ti-Based Ti-Cu-Zr-Fe-Sn-Si-Ag Bulk Metallic Glass for Biomedical Applications. *J. Alloys Comp.* 625, 323–327. doi:10.1016/j.jallcom.2014.07.021
- Park, E. S., and Kim, D. H. (2004). Formation of Ca-Mg-Zn Bulk Glassy alloy by Casting into Cone-Shaped Copper Mold. *J. Mater. Res.* 19, 685–688. doi:10.1557/jmr.2004.19.3.685
- Park, J. M., Kim, Y. C., Kim, W. T., and Kim, D. H. (2004). Ti-Based Bulk Metallic Glasses with High Specific Strength. *Mater. Trans.* 45, 595–598. doi:10.2320/matertrans.45.595
- Peker, A., and Johnson, W. L. (1993). A Highly Processable Metallic Glass: Zr_{41.2}Ti_{13.8}Cu_{12.5}Ni_{10.0}Be_{22.5}. *Appl. Phys. Lett.* 63, 2342–2344. doi:10.1063/1.110520
- Polk, D. E., Calka, A., and Giessen, B. C. (1978). The Preparation and thermal and Mechanical Properties of New Titanium Rich Metallic Glasses. *Acta Metallurgica* 26, 1097–1103. doi:10.1016/0001-6160(78)90137-2
- Qin, F.-X., Zhou, Y., Ji, C., Dan, Z.-H., Xie, G.-Q., and Yang, S. (2016). Enhanced Mechanical Properties, Corrosion Behavior and Bioactivity of Ti-Based Bulk Metallic Glasses with Minor Addition Elements. *Acta Metall. Sin. (Engl. Lett.* 29, 1011–1018. doi:10.1007/s40195-016-0468-0
- Qin, F. X., Wang, X. M., Xie, G. Q., and Inoue, A. (2008). Distinct Plastic Strain of Ni-free Ti-Zr-Cu-Pd-Nb Bulk Metallic Glasses with Potential for Biomedical Applications. *Intermetallics* 16, 1026–1030. doi:10.1016/j.intermet.2008.05.004
- Rabinkin, A., Liebermann, H., Pounds, S., Taylor, T., Reidinger, F., and Lui, S.-C. (1991). Amorphous TiZr - Base Metglas Brazing Filler Metals. *Scripta Metallurgica et Materialia* 25, 399–404. doi:10.1016/0956-716x(91)90200-k
- Rodmacq, B., Lançon, F., Chamberod, A., and Maret, M. (1988). Large-angle Neutron Scattering Studies of Amorphous FeTi_{1-x} Alloys and Hydrides. *Mater. Sci. Eng.* 97, 157–161. doi:10.1016/0025-5416(88)90032-8
- Saini, J. S., Miska, J. P., Lei, F., AuYeung, N., and Xu, D. (2021). Hafnium Based Metallic Glasses with High Density and High Glass-Forming Ability. *J. Alloys Comp.* 882, 160896. doi:10.1016/j.jallcom.2021.160896
- Saini, J. S., Palian, C., Lei, F., Dyal, A., AuYeung, N., McQuade, R., et al. (2020). Rare-earth and Precious-Metal Free Cu-Based Metallic Glasses with superior Glass-Forming Ability and Processability. *Appl. Phys. Lett.* 116, 011901. doi:10.1063/1.5131645
- Schroers, J., and Johnson, W. L. (2004). Highly Processable Bulk Metallic Glass-Forming Alloys in the Pt-Co-Ni-Cu-P System. *Appl. Phys. Lett.* 84, 3666–3668. doi:10.1063/1.1738945
- Seki, I., Kimura, H., and Inoue, A. (2008). Thermal Stability and Mechanical Properties of Ti_{47.4}Cu₄₂Zr_{5.3}TM_{5.3}(TM = Co, Fe) Metallic Glass Sheets Prepared by Twin-Roller Casting Method. *Mater. Trans.* 49, 498–501. doi:10.2320/matertrans.mbw200738
- Shan, S.-F., Zhan, Z.-J., Fan, C.-Z., Jia, Y.-Z., Zhang, B.-Q., Liu, R.-P., et al. (2008). *Chin. Phys. Lett.* 25, 4165–4167.
- Shao, L., Xue, L., Luo, Q., Wang, Q., and Shen, B. (2019). The Role of Co/Al Ratio in Glass-Forming GdCoAl Magnetocaloric Metallic Glasses. *Materialia* 7, 100419. doi:10.1016/j.mta.2019.100419
- Shirasawa, N., Ito, R., Takigawa, Y., Uesugi, T., and Higashi, K. (2014). Prediction and Fabrication of Ti-Zr-Co Ternary Metallic Glasses Based on Effective Atomic Radius in Ti Solid Solution from First-Principles Calculations. *J. Non-Crystalline Sol.* 400, 67–71. doi:10.1016/j.jnoncrysol.2014.05.005
- Song, X., Liu, X., Yang, Y., Feng, S., Lu, Y., and Kong, J. (2019). Glass Forming Ability and a Novel Method for Evaluating the Thermoplastic Formability of Zr Ti₆₅Be_{27.5}Cu_{7.5} Alloys. *Intermetallics* 114, 106600. doi:10.1016/j.intermet.2019.106600
- Sun, Y. F., Wang, Y. R., Wei, B. C., Li, W. H., and Shek, C. H. (2005). Effect of Quasicrystalline Phase on the Deformation Behavior of Zr₆₂Al_{9.5}Ni_{9.5}Cu₁₄Nb₅ Bulk Metallic Glass. *Trans. Nonferrous Met. Soc. China* 15, 727–732. doi:10.1016/j.msea.2005.01.040
- Sun, Y. T., Bai, H. Y., Li, M. Z., and Wang, W. H. (2017). Machine Learning Approach for Prediction and Understanding of Glass-Forming Ability. *J. Phys. Chem. Lett.* 8, 3434–3439. doi:10.1021/acs.jpclett.7b01046
- Suryanarayana, C., and Inoue, A. (2013). Iron-based Bulk Metallic Glasses. *Int. Mater. Rev.* 58, 131–166. doi:10.1179/1743280412y.0000000007
- Suryanarayana, C., Inoue, A., and Masumoto, T. (1980). Transformation Studies and Mechanical Properties of Melt-Quenched Amorphous Titanium-Silicon Alloys. *J. Mater. Sci.* 15, 1993–2000. doi:10.1007/bf00550625
- Šušić, M. V., Budberg, P. B., and Alisova, S. P. (1986). Kinetics of thermal Devitrification (Crystallization) of a Titanium Amorphous alloy. *Mater. Chem. Phys.* 15, 1297–1300. doi:10.1007/bf00553266
- Tanaka, K., Shibagaki, N., and Yamauchi, T. (1988). Electronic Structure and Local Environment of Non-transition Elements in Ternary Metallic Glasses. *Mater. Sci. Eng.* 99, 269–271. doi:10.1016/0025-5416(88)90337-0
- Tang, M. Q., Zhang, H. F., Zhu, Z. W., Fu, H. M., Wang, A. M., Li, H., et al. (2010). TiZr-base Bulk Metallic Glass with over 50 Mm in Diameter. *J. Mater. Sci. Tech.* 26, 481–486. doi:10.1016/s1005-0302(10)60077-1
- Tanner, L. E., Jacobson, L. A., and Wall, M. E. (1988). Formation and Crystallization of Amorphous Phases in the TiBeAl System. *Mater. Sci. Eng.* 97, 379–383. doi:10.1016/0025-5416(88)90078-x
- Tanner, L. E. (1978). Physical Properties of TiBeSi Glass Ribbons. *Scripta Metallurgica* 12, 703–708. doi:10.1016/0036-9748(78)90312-5
- Tanner, L. E., and Ray, R. (1979). Metallic Glass Formation and Properties in Zr and Ti Alloyed with Be-I the Binary Zr-Be and Ti-Be Systems. *Acta Metallurgica* 27, 1727–1747. doi:10.1016/0001-6160(79)90087-7
- Tanner, L. E., and Ray, R. (1980). Phase Separation in ZrTiBe Metallic Glasses. *Scripta Metallurgica* 14, 657–662. doi:10.1016/0036-9748(80)90018-6
- Tanner, L. E., and Ray, R. (1977). Physical Properties of Ti₅₀Be₄₀Zr₁₀ Glass. *Scripta Metallurgica* 11, 783–789. doi:10.1016/0036-9748(77)90076-x
- Tantavisit, S., Lohwongwatana, B., Khamkongkaeo, A., Tanavalee, A., Tangpornprasert, P., and Ittiravivong, P. (2018). The Novel Toxic Free Titanium-Based Amorphous alloy for Biomedical Application. *J. Mater. Res. Tech.* 7, 248–253. doi:10.1016/j.jmrt.2017.08.007
- Tarasova, O. B., Smirnov, L. S., and Fykin, L. E. (1992). A Study of Short-Range Order in Amorphous Ti-Zr-Si Alloys. *Phys. met. Metallogr* 73, 177–180.
- Thanka Rajan, S., Bendavid, A., and Subramanian, B. (2019). Cytocompatibility Assessment of Ti-Nb-Zr-Si Thin Film Metallic Glasses with Enhanced Osteoblast Differentiation for Biomedical Applications. *Colloids Surf. B: Biointerfaces* 173, 109–120. doi:10.1016/j.colsurfb.2018.09.041
- Tsai, P. H., Hsu, K. T., Ke, J. H., Lin, H. C., Jang, J. S. C., and Huang, J. C. (2015). *Mater. Tech.* 30, A161–A165. doi:10.1179/17535557a15y.000000003
- Wang, C., Hua, N., Liao, Z., Yang, W., Pang, S., Liaw, P. K., et al. (2021). Ti-Cu-Zr-Fe-Sn-Si-Ag-Pd Bulk Metallic Glasses with Potential for Biomedical Applications. *Metall. Mater. Trans. A* 52, 1559–1567. doi:10.1007/s11661-021-06183-y
- Wang, H., Park, E. S., Oak, J. J., Setyawan, A. D., Zhu, S. L., Wada, T., et al. (2013). Effect of Cobalt Microalloying on the Glass Forming Ability of Ti-Cu-Pd-Zr Metallic Glass. *J. Non-Crystalline Sol.* 379, 155–160. doi:10.1016/j.jnoncrysol.2013.08.001
- Wang, T., Wu, Y. D., Si, J. J., Cai, Y. H., Chen, X. H., and Hui, X. D. (2015). Novel Ti-Based Bulk Metallic Glasses with superior Plastic Yielding Strength and Corrosion Resistance. *Mater. Sci. Eng. A* 642, 297–303. doi:10.1016/j.msea.2015.05.060
- Wang, T., Wu, Y., Si, J., and Hui, X. (2015). Effects of Zr and Si on the Glass Forming Ability and Compressive Properties of Ti-Cu-Co-Sn Alloys. *Metall. Mat Trans. A* 46, 2381–2389. doi:10.1007/s11661-014-2484-x
- Wang, W. H. (2009). Bulk Metallic Glasses with Functional Physical Properties. *Adv. Mater.* 21, 4524–4544. doi:10.1002/adma.200901053
- Wang, W. H., Dong, C., and Shek, C. H. (2004). Bulk Metallic Glasses. *Mater. Sci. Eng. R Rep.* 44, 45–89. doi:10.1016/j.mser.2004.03.001
- Wang, Y.-L., and Xu, J. (2008). Ti (Zr)-Cu-Ni Bulk Metallic Glasses with Optimal Glass-Forming Ability and Their Compressive Properties. *Metall. Mat Trans. A* 39, 2990–2997. doi:10.1007/s11661-008-9647-6

- Wang, Z., Dong, D., Qiang, J., Wang, Q., Wang, Y., and Dong, C. (2013). Ti-based Glassy Alloys in Ti-Cu-Zr-Sn System. *Sci. China Phys. Mech. Astron.* 56, 1419–1422. doi:10.1007/s11433-013-5104-7
- Wang, Z. R., Qiang, J. B., Wang, Y. M., Wang, Q., Dong, D. D., and Dong, C. (2016). Composition Design Procedures of Ti-Based Bulk Metallic Glasses Using the Cluster-Plus-Glue-Atom Model. *Acta Materialia* 111, 366–376. doi:10.1016/j.actamat.2016.03.072
- Whang, S. H., Lu, Y. Z., and Kim, Y. W. (1985). Microstructures and Age Hardening of Rapidly Quenched Ti-Zr-Si Alloys. *J. Mater. Sci. Lett.* 4, 883–887. doi:10.1007/bf00720529
- Wu, X. Q., Wang, H. L., and Lin, J. G. (2014). Effects of Sn Content on thermal Stability and Mechanical Properties of the Ti60Zr10Ta15Si15 Amorphous alloy for Biomedical Use. *Mater. Des.* 63, 345–348. doi:10.1016/j.matdes.2014.06.045
- Wu, X., Zhou, Y., Chen, F., and Qin, F. (2019). Effect of Nb Addition on Corrosion Behavior and Mechanical Properties of Ti-Based Metallic Glasses. *Mater. Sci. Tech.* 27, 73–80. doi:10.11951/j.issn.1005-0299.20180094
- Xia, M.-x., Zheng, H.-x., Liu, J., Ma, C.-l., and Li, J.-g. (2005). Thermal Stability and Glass-Forming Ability of New Ti-Based Bulk Metallic Glasses. *J. Non-Crystalline Sol.* 351, 3747–3751. doi:10.1016/j.jnoncrysol.2005.09.033
- Xia, M. X., Ma, C. L., Zheng, H. X., and Li, J. G. (2005). Preparation and Crystallization of Ti₅₃Cu₂₇Ni₁₂Zr₃Al₇Si₃B₁ Bulk Metallic Glass with Wide Supercooled Liquid Region. *Mater. Sci. Eng. A* 390, 372–375. doi:10.1016/j.msea.2004.08.019
- Xia, M. X., Zheng, H. X., Ma, C. L., and Li, J. G. (2005). Formation and Thermal Properties of Ti-Based Bulk Amorphous Alloy. *Rare Metal Mater. Eng.* 34, 1235–1238.
- Xia, M. X., Zheng, H. X., Ma, C. L., and Li, J. G. (2005). Preparation and Properties of High Strength Bulk Metallic Glass Ti₅₃Cu₁₅Ni_{18.5}Al₇M₃Si₃B_{0.5} (M = Zr, Hf, Sc). *Acta Metall. Sin.* 41, 199–202.
- Xie, K.-F., Yao, K.-F., and Huang, T.-Y. (2010). A Ti-Based Bulk Glassy alloy with High Strength and Good Glass Forming Ability. *Intermetallics* 18, 1837–1841. doi:10.1016/j.intermet.2010.02.036
- Xie, K. F., Yao, K. F., and Huang, T. Y. (2010). Preparation of (Ti_{0.45}Cu_{0.378}Zr_{0.10}Ni_{0.072})_{100-x}Sn_x Bulk Metallic Glasses. *J. Alloys Comp.* 504, S22–S26. doi:10.1016/j.jallcom.2010.02.199
- Xiong, J., Shi, S.-Q., and Zhang, T.-Y. (2021). Machine Learning Prediction of Glass-Forming Ability in Bulk Metallic Glasses. *Comput. Mater. Sci.* 192, 110362. doi:10.1016/j.commatsci.2021.110362
- Xu, T., Pang, S., Li, H., and Zhang, T. (2015). Corrosion Resistant Cr-Based Bulk Metallic Glasses with High Strength and Hardness. *J. Non-Crystalline Sol.* 410, 20–25. doi:10.1016/j.jnoncrysol.2014.12.006
- Yan, H.-M., Liu, Y., Pang, S.-J., and Zhang, T. (2018). Glass Formation and Properties of Ti-Based Bulk Metallic Glasses as Potential Biomaterials with Nb Additions. *Rare Met.* 37, 831–837. doi:10.1007/s12598-015-0664-5
- Yang, B. J., Lu, W. Y., Zhang, J. L., Wang, J. Q., and Ma, E. (2017). Melt Fluxing to Elevate the Forming Ability of Al-Based Bulk Metallic Glasses. *Sci. Rep.* 7, 11053. doi:10.1038/s41598-017-11504-6
- Yang, S., Li, D., Li, X. C., Zhang, Z. Z., Zhang, S. F., and He, L. (2017). Composition Dependence of the Microstructure and Mechanical Behavior of Ti-Zr-Cu-Pd-Sn-Nb Bulk Metallic Glass Composites. *Intermetallics* 90, 1–8. doi:10.1016/j.intermet.2017.06.006
- Yang, W., Liu, Y., Hua, N., Pang, S., Li, Y., Liaw, P. K., et al. (2021). Formation and Properties of Biocompatible Ti-Based Bulk Metallic Glasses in the Ti-Cu-Zr-Fe-Sn-Si-Ag System. *J. Non-Crystalline Sol.* 571, 121060. doi:10.1016/j.jnoncrysol.2021.121060
- Yi Qun Gao, G., and Whang, S. H. (1985). Crystallization Behavior of Binary Metallic Glasses Containing Pt. *J. Non-Crystalline Sol.* 70, 85–92. doi:10.1016/0022-3093(85)90095-x
- Yi, S. (2012). 이철규. *J. Korea Foundry Soc.* 32, 177–180. doi:10.7777/jkfs.2012.32.4.177
- Yin, E., Pang, S., Hui, X., Zhang, M., Zhuo, L., Chen, C., et al. (2013). Correlation of Glass-Forming Ability to thermal Properties in Ti-Based Bulk Metallic Glasses. *J. Alloys Comp.* 546, 7–13. doi:10.1016/j.jallcom.2012.07.037
- Zeng, Y. Q., Yu, J. S., Tian, Y., Hirata, A., Fujita, T., Zhang, X. H., et al. (2020). Improving Glass Forming Ability of Off-Eutectic Metallic Glass Formers by Manipulating Primary Crystallization Reactions. *Acta Materialia* 200, 710–719. doi:10.1016/j.actamat.2020.09.042
- Zhang, C., Ouyang, D., Pauly, S., and Liu, L. (2021). 3D Printing of Bulk Metallic Glasses. *Mater. Sci. Eng. R: Rep.* 145, 100625. doi:10.1016/j.mser.2021.100625
- Zhang, L.-C., Jia, Z., Lyu, F., Liang, S.-X., and Lu, J. (2019). A Review of Catalytic Performance of Metallic Glasses in Wastewater Treatment: Recent Progress and Prospects. *Prog. Mater. Sci.* 105, 100576. doi:10.1016/j.pmatsci.2019.100576
- Zhang, L., Tang, M. Q., Zhu, Z. W., Fu, H. M., Zhang, H. W., Wang, A. M., et al. (2015). Compressive Plastic Metallic Glasses with Exceptional Glass Forming Ability in the Ti-Zr-Cu-Fe-Be alloy System. *J. Alloys Comp.* 638, 349–355. doi:10.1016/j.jallcom.2015.03.120
- Zhang, T., Inoue, A., and Masumoto, T. (1994). Amorphous (Ti,Zr, Hf)NiCu Ternary Alloys with a Wide Supercooled Liquid Region. *Mater. Sci. Eng. A* 181–182, 1423–1426. doi:10.1016/0921-5093(94)90877-x
- Zhang, T., Inoue, A., and Masumoto, T. (1993). The Effect of Atomic Size on the Stability of Supercooled Liquid for Amorphous (Ti, Zr, Hf)₆₅Ni₂₅Al₁₀ and (Ti, Zr, Hf)₆₅Cu₂₅Al₁₀ Alloys. *Mater. Lett.* 15, 379–382. doi:10.1016/0167-577x(93)90100-c
- Zhang, T., and Inoue, A. (1999). Preparation of Ti-Cu-Ni-Si-B Amorphous Alloys with a Large Supercooled Liquid Region. *Mater. Trans. JIM* 40, 301–306. doi:10.2320/matertrans1989.40.301
- Zhang, T., and Inoue, A. (1998). Thermal and Mechanical Properties of Ti-Ni-Cu-Sn Amorphous Alloys with a Wide Supercooled Liquid Region before Crystallization. *Mater. Trans. JIM* 39, 1001–1006. doi:10.2320/matertrans1989.39.1001
- Zhang, T., and Inoue, A. (2001). Ti-based Amorphous Alloys with a Large Supercooled Liquid Region. *Mater. Sci. Eng. A* 304–306, 771–774. doi:10.1016/s0921-5093(00)01592-6
- Zhang, T., Yang, Q., Ji, Y., Li, R., Pang, S., Wang, J., et al. (2011). Centimeter-scale-diameter Co-based Bulk Metallic Glasses with Fracture Strength Exceeding 5000 MPa. *Chin. Sci. Bull.* 56, 3972–3977. doi:10.1007/s11434-011-4765-8
- Zhao, L., Zhang, Z., Zhang, J., Pang, S., Ma, C., and Zhang, T. (2010). Composition Design and Glass-Forming Ability of Ti-Based Bulk Metallic Glasses. *Int. J. Mod. Phys. B* 24, 2326–2331. doi:10.1142/s0217979210064873
- Zhao, S., Chen, N., Gong, P., and Yao, K. (2016). Centimeter-Sized Quaternary Ti-Based Bulk Metallic Glasses with High Ti Content of 50 at%. *Adv. Eng. Mater.* 18, 231–235. doi:10.1002/adem.201500165
- Zhao, S. F., Chen, N., Gong, P., and Yao, K. F. (2015). New Centimeter-Sized Quaternary Ti-Zr-Be-Cu Bulk Metallic Glasses with Large Glass Forming Ability. *J. Alloys Comp.* 647, 533–538. doi:10.1016/j.jallcom.2015.05.214
- Zhao, S. F., Gong, P., Li, J. F., Chen, N., and Yao, K. F. (2015). Quaternary Ti-Zr-Be-Ni Bulk Metallic Glasses with Large Glass-Forming Ability. *Mater. Des.* 85, 564–573. doi:10.1016/j.matdes.2015.07.032
- Zhao, S. F., Shao, Y., Gong, P., and Yao, K. F. (2014). A Centimeter-Sized Quaternary Ti-Zr-Be-Ag Bulk Metallic Glass. *Adv. Mater. Sci. Eng.* 2014, 15. doi:10.1155/2014/192187
- Zheng, Q., Xu, J., and Ma, E. (2007). High Glass-Forming Ability Correlated with Fragility of Mg-Cu(Ag)-Gd Alloys. *J. Appl. Phys.* 102, 113519. doi:10.1063/1.2821755
- Zhou, Y., Zhao, Y., Qu, B. Y., Wang, L., Zhou, R. L., Wu, Y. C., et al. (2015). Remarkable Effect of Ce Base Element Purity upon Glass Forming Ability in Ce-Ga-Cu Bulk Metallic Glasses. *Intermetallics* 56, 56–62. doi:10.1016/j.intermet.2014.09.003
- Zhu, S. L., Wang, X. M., and Inoue, A. (2008). Glass-forming Ability and Mechanical Properties of Ti-Based Bulk Glassy Alloys with Large Diameters of up to 1cm. *Intermetallics* 16, 1031–1035. doi:10.1016/j.intermet.2008.05.006
- Zhu, S. L., Wang, X. M., Qin, F. X., Yoshimura, M., and Inoue, A. (2007). New TiZrCuPd Quaternary Bulk Glassy Alloys with Potential of Biomedical Applications. *Mater. Trans.* 48, 2445–2448. doi:10.2320/matertrans.mra2007086

- Zhu, S. L., Wang, X. M., Qin, F. X., and Inoue, A. (2007). A New Ti-Based Bulk Glassy alloy with Potential for Biomedical Application. *Mater. Sci. Eng. A* 459, 233–237. doi:10.1016/j.msea.2007.01.044
- Zhu, S. L., Xie, G. Q., Qin, F. X., and Wang, X. M. (2012). *8th International Forum on Advanced Materials Science and Technology (IFAMST-8)*. Fukuoka City, JAPAN: Trans Tech Publications Ltd, Fukuoka Inst Technol, 36–39.
- Zhu, S., Xie, G., Qin, F., Wang, X., and Inoue, A. (2012). Effect of Minor Sn Additions on the Formation and Properties of TiCuZrPd Bulk Glassy Alloy. *Mater. Trans.* 53, 500–503. doi:10.2320/matertrans.m2011281

Conflict of Interest: The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

Publisher's Note: All claims expressed in this article are solely those of the authors and do not necessarily represent those of their affiliated organizations, or those of the publisher, the editors, and the reviewers. Any product that may be evaluated in this article, or claim that may be made by its manufacturer, is not guaranteed or endorsed by the publisher.

Copyright © 2022 Zhang, Song, Lin, Li and Li. This is an open-access article distributed under the terms of the Creative Commons Attribution License (CC BY). The use, distribution or reproduction in other forums is permitted, provided the original author(s) and the copyright owner(s) are credited and that the original publication in this journal is cited, in accordance with accepted academic practice. No use, distribution or reproduction is permitted which does not comply with these terms.