



# Substrate pathways and mechanisms of inhibition in the sulfur oxygenase reductase of *Acidianus ambivalens*

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**Background:** The sulfur oxygenase reductase (SOR) is the initial enzyme of the sulfur oxidation pathway in the thermoacidophilic Archaeon *Acidianus ambivalens*. The SOR catalyzes an oxygen-dependent sulfur disproportionation to H<sub>2</sub>S, sulfite and thiosulfate. The spherical, hollow, cytoplasmic enzyme is composed of 24 identical subunits with an active site pocket each comprising a mononuclear non-heme iron site and a cysteine persulfide. Substrate access and product exit occur via apolar chimney-like protrusions at the fourfold symmetry axes, via narrow polar pores at the threefold symmetry axes and via narrow apolar pores within in each subunit. In order to investigate the function of the pores we performed site-directed mutagenesis and inhibitor studies. **Results:** Truncation of the chimney-like protrusions resulted in an up to sevenfold increase in specific enzyme activity compared to the wild type. Replacement of the salt bridge-forming Arg<sub>99</sub> residue by Ala at the threefold symmetry axes doubled the activity and introduced a bias toward reduced reaction products. Replacement of Met<sub>296</sub> and Met<sub>297</sub> which form the active site pore, lowered the specific activities by 25–55% with the exception of an M<sub>296</sub>V mutant. X-ray crystallography of SOR wild type crystals soaked with inhibitors showed that Hg<sup>2+</sup> and iodoacetamide (IAA) bind to cysteines within the active site, whereas Zn<sup>2+</sup> binds to a histidine in a side channel of the enzyme. The Zn<sup>2+</sup> inhibition was partially alleviated by mutation of the His residue. **Conclusions:** The expansion of the pores in the outer shell led to an increased enzyme activity while the integrity of the active site pore seems to be important. Hg<sup>2+</sup> and IAA block cysteines in the active site pocket, while Zn<sup>2+</sup> interferes over a distance, possibly by restriction of protein flexibility or substrate access or product exit.

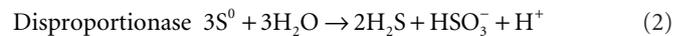
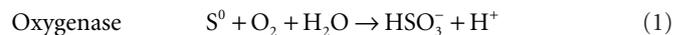
**Keywords:** Archaea, sulfur metabolism, site-directed mutagenesis, structural biology, X-ray crystallography

## INTRODUCTION

A large number of microorganisms oxidize sulfur and reduced inorganic sulfur compounds (ISC) for energy conservation (for review, see for example Friedrich et al., 2005; Kletzin, 2007; Frigaard and Dahl, 2009; Ghosh and Dam, 2009). Most studies on ISC oxidation were performed with soluble sulfur species like thiosulfate, sulfite and sulfide. Their mechanisms of activation and oxidation are reasonably well understood (Friedrich et al., 2005; Ghosh and Dam, 2009). Less is known how the barely soluble elemental sulfur is mobilized and oxidized (19–30 nmol/l α-S<sub>8</sub> at 25°C, 478 nmol/l at 80°C; Kamyshny, 2009). Different enzymes and enzyme activities were described but few were analyzed in molecular detail (Rohwerder and Sand, 2003; Kletzin, 2007; Ghosh and Dam, 2009; also see Protze et al., 2011, this volume).

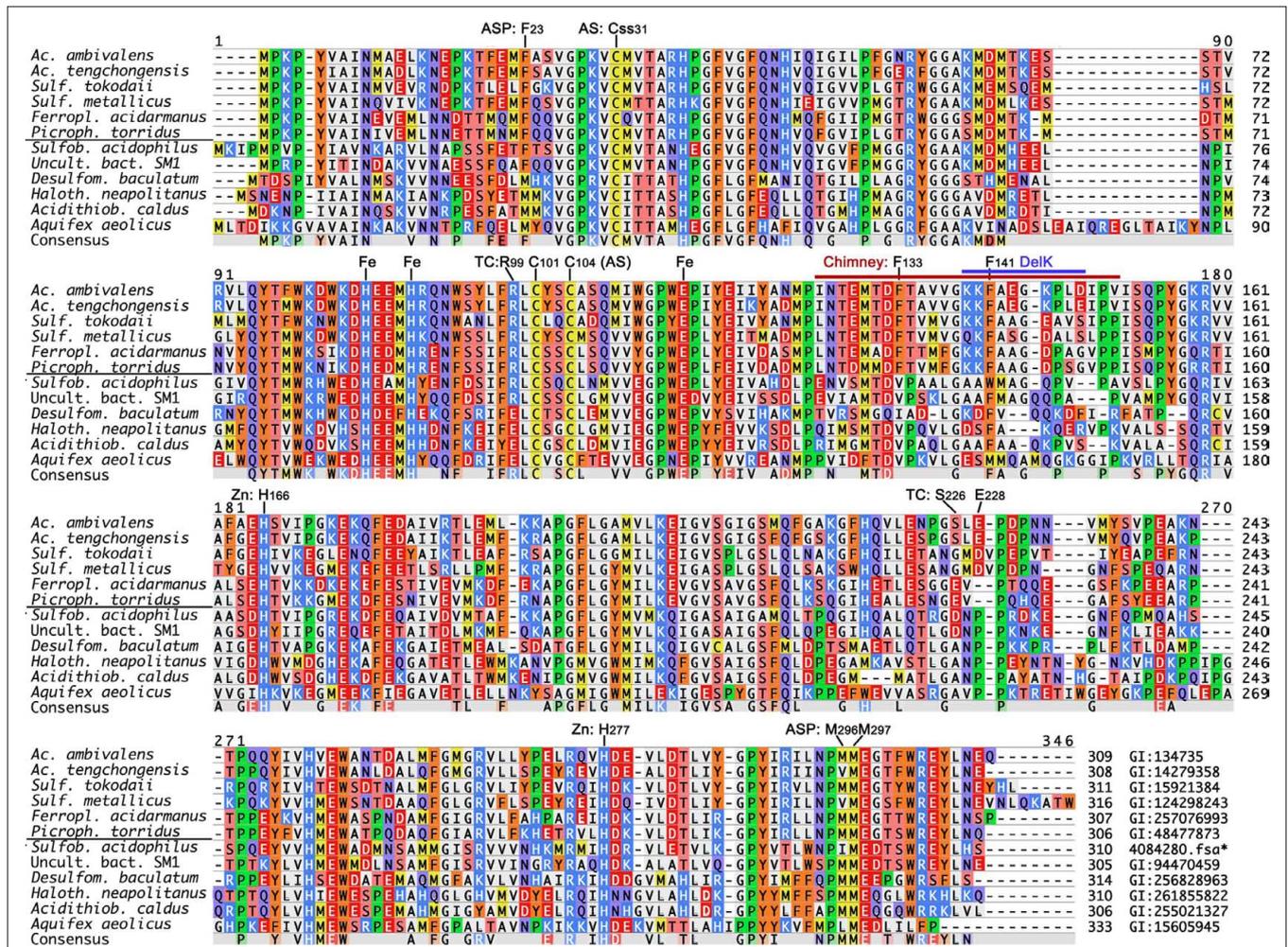
The best-known sulfur-oxidizing enzymes are sulfur oxygenase reductases (SOR), which were purified from two different thermoacidophilic *Acidianus* species (Emmel et al., 1986; Kletzin, 1989). In addition, SORs obtained by heterologous gene expression were studied from *Ac. ambivalens* and *Ac. tengchongensis*, from the hyperthermophilic bacterium *Aquifex aeolicus*, and from a moderately thermophilic bacterium from a bioleaching reactor (Sun et al., 2003; Ulrich et al., 2004; Chen et al., 2007; Pelletier et al., 2008). The SOR is the initial sulfur-oxidizing enzyme in the

Archaeon *Ac. ambivalens*, which is our model organism for sulfur metabolism, and which grows optimally at 80°C and pH 1–3. The SOR or *sor* genes do not occur frequently; so far they are restricted to some thermoacidophilic Archaea and to some mesophilic and thermophilic Bacteria (Figure 1). The SORs catalyze an oxygen-dependent sulfur disproportionation reaction with sulfite, thiosulfate and sulfide as products (Eqs 1–3; Kletzin, 1989; Sun et al., 2003; Pelletier et al., 2008).



External cofactors or electron donors are not required and the two enzyme activities could not be separated. Zn<sup>2+</sup>, Hg<sup>2+</sup> and thiol-modifying organic agents like iodoacetamide (IAA) and *N*-ethylmaleimide (NEM) inhibit activity (Kletzin, 1989). The SORs do not catalyze sulfur disproportionation in the absence of oxygen and they have neutral or slightly acidic pH optima (5–7.4; Emmel et al., 1986; Kletzin, 1989; Sun et al., 2003).

X-ray crystallographic structures were determined of two of these enzymes, from *Ac. ambivalens* and from *Ac. tengchongensis*. Both showed that the spherical, hollow oligomers are composed



**FIGURE 1 | Multiple alignments of known and complete SOR amino acid sequences from isolated proteins and from genome sequences.** PCR-generated fragments with high similarity to the *Acidianus* enzymes (Chen et al., 2007) were omitted. Genbank identification (GI) numbers are given at the end; \*derived from the genome sequence available at JGI<sup>1</sup>. The horizontal line separates

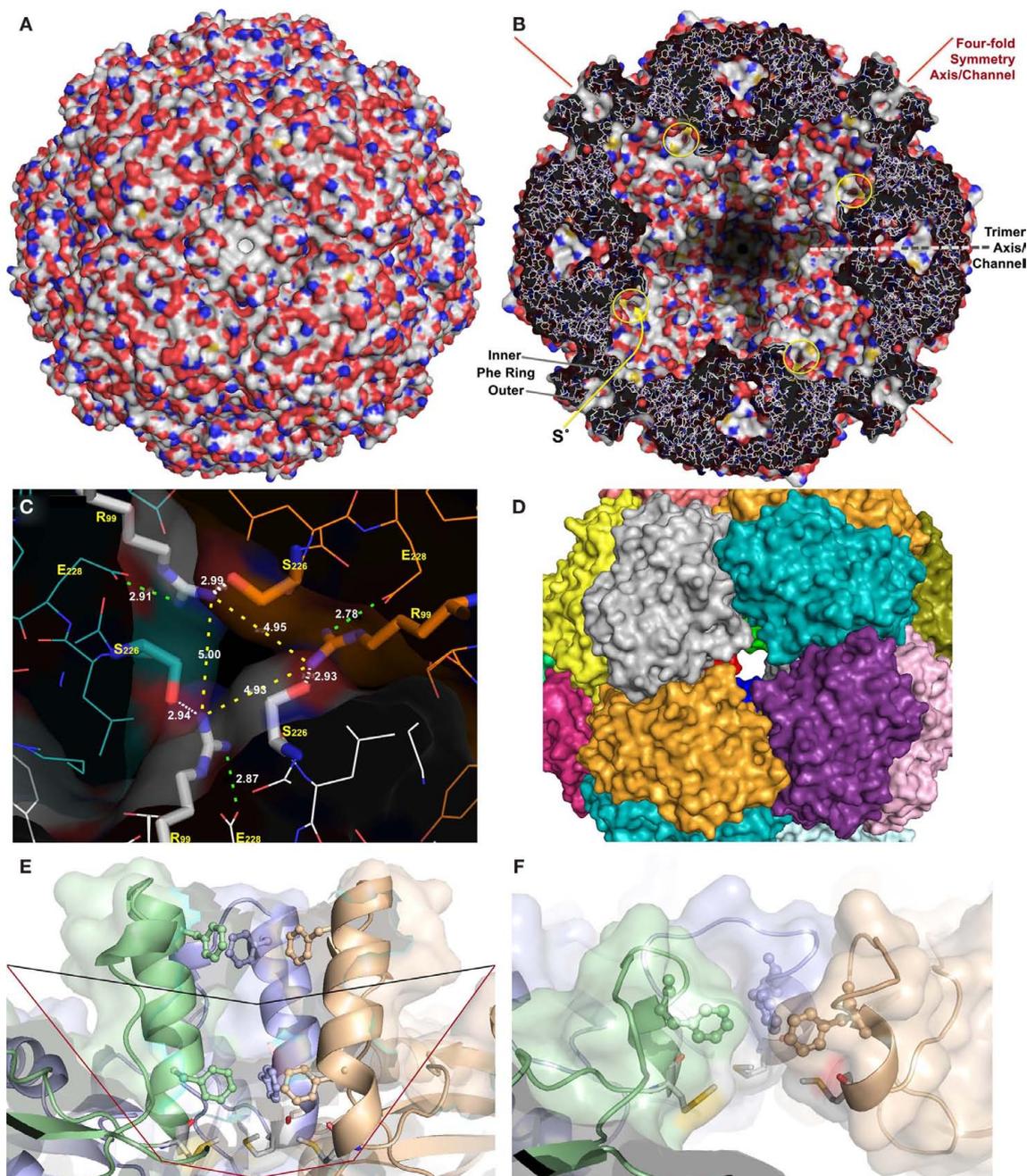
Archaea from Bacteria. Abbreviations: ASP, active site pore residues (Figure 5); AS, active site cysteines; Cys, cysteine persulfide; Fe, iron-coordinating residues; Chimney, chimney-like protrusions at the fourfold symmetry axes, identical to the DelK deletion (Figure 2); DelK, short deletion of the residues around the outer Phe ring (Figure 2); Zn, 2-His motif under the zinc-coordinating His<sub>277</sub>.

of 24 identical subunits arranged in a 432 point-group symmetry (Figure 2; Ulrich et al., 2006; Li et al., 2008). The outer surfaces of the spheres are apparently smooth and impervious with the exception of very narrow pores at the four- and threefold rotational axes of the oligomers (Figure 2). Each subunit contains a low-potential mononuclear non-heme iron site as the putative redox-active cofactor (Ulrich et al., 2004, 2006). We had shown using site-directed mutagenesis that the three Fe-coordinating residues (H<sub>86</sub>, H<sub>90</sub> and E<sub>114</sub> in *Ac. ambivalens* numbering) and a persulfurated cysteine (C<sub>31</sub>) are essential for catalysis. Most likely, the cysteine persulfide is involved in sulfur binding. Mutation of the other two cysteine residues did not abolish activity, not even in a double mutant (Ulrich et al., 2005b). Similar results had been obtained for the *Ac. tengchongensis* SOR (Chen et al., 2005). Our current hypothesis about the reaction mechanism of the SOR predicts that the catalytic cycle is initiated by covalent

sulfur binding to the active site C<sub>31</sub> as a polysulfide chain (R–S<sub>n</sub>–SH), followed by hydrolytic cleavage of the cysteine polysulfide to sulfide and a polysulfenyl moiety (R–S<sub>n</sub>–SOH). Either Fe<sup>2+</sup> or the sulfenyl group would subsequently activate oxygen (Kletzin, 2008).

The iron site and the three conserved cysteine residues (Figure 1) are located in an active site pocket that is connected to the inner cavity of the sphere by a narrow pore formed by two adjacent methionines and a phenylalanine (3–4 Å diameter; M<sub>296</sub>/M<sub>297</sub>, F<sub>23</sub>; Figures 1 and 2). In consequence, substrate and products must pass first the outer shell of the holoenzyme into the inner cavity and then the active site pore into the pocket or *vice versa*. We had postulated that only linear polysulfane species but not circular S<sub>8</sub> could pass both barriers (Ulrich et al., 2006). Therefore, S<sub>8</sub>-sulfur would require an initial activation step for catalysis, either by binding to a thiol group or by sulfide. The polysulfide sulfur formed in this reaction should be stable at the near-neutral pH of the cellular cytoplasm (6.5; Moll and Schäfer, 1988). In

<sup>1</sup>http://www.jgi.doe.gov



**FIGURE 2 | X-ray crystallography and modeling of the SOR and its pores. (A)**

Surface representation of the holoenzyme centered at the chimney-like structure at the fourfold symmetry axis. **(B)** Representation of the protein structure and the inner surface of the holoenzyme sliced at the center of the fourfold symmetry axes; the position of the inner and outer phenylalanine rings are indicated, also the active site pores (yellow circles), and the approximate position of the trimer symmetry axis, which is tilted out of plane. **(C)** Channel at the threefold symmetry axis formed by R<sub>99</sub> and S<sub>226</sub>; distances are given between the N $\eta$  atoms of the arginines (yellow dashes), for the salt bridges to E<sub>228</sub> (green dashes),

and for the putative hydrogen bond to the O $\gamma$  of the S<sub>226</sub> of the neighboring subunit (white dashes). **(D)** Subunit representation of the large deletion mutant at the fourfold symmetry axis DelL (Figure 1) modeled at the SwissModel server (Arnold et al., 2006). **(E)** Side view of the channel at the fourfold symmetry axis with outer (F<sub>141</sub>) and inner phenylalanine rings (F<sub>133'</sub>; from top, showing three out of four subunits; wild type) and the methionine ring at its base (M<sub>130</sub>); black line, approximate position of small deletion (DelK); red line, large deletion of entire chimney (DelL). **(F)** Model of the same three subunits as in panel E of the short DelK deletion, modeled at the Phyre server (Kelley and Sternberg, 2009).

contrast, polysulfides are not stable in the outside medium (pH 1–3) and disproportionate into sulfur and sulfide (Schauder and Kröger, 1993).

The question now arises how the sulfur gets inside of the enzyme and how the products get out. Two successive rings of four phenylalanine residues each form the hydrophobic pore at the

fourfold symmetry axis, followed at the inside by four methionines (Figure 2). In the crystallized protein, the pore is not fully open with C–C distances down to 5.0 Å in the inner Phe ring (Figure 2; Urich et al., 2006). Nevertheless, it is feasible that apolar molecules like the linear polysulfane species should be able to pass the pore when the protein is heated. In contrast, the polar reaction products hydrogen sulfide, sulfite and thiosulfate were proposed to exit the sphere via channels located at the threefold symmetry axes (Li et al., 2008). The narrow channel outlets are lined with hydrophilic residues such as R<sub>99</sub> and S<sub>226</sub>. The shortest distances between the Nη of the three arginines are ≈5 Å, or ≈6.3 Å between the Oγ of the Ser and Nη of the Arg (Figure 2). In addition, the arginines form salt bridges to the gamma-carboxylate group of a glutamate within the same subunit and potential hydrogen bonds to the S<sub>226</sub> residues of the neighboring subunit. Nevertheless, the arginines should be able to bind transiently the anionic reaction products sulfide, sulfite and thiosulfate allowing their exit.

Here, we show by site-directed mutagenesis that opening the putative substrate and product pathways in the outer shell leads to a significant increase in specific activity and to a shift in the stoichiometry of the products. In contrast, the integrity of the inner pore seems to be important. We also show by mutagenesis studies and crystallographic analysis of inhibitor derivatives that Hg<sup>2+</sup> and IAA bind in the active site as expected, whereas Zn<sup>2+</sup> does not and could interfere with the movement of substrates and products.

## MATERIALS AND METHODS

### CONSTRUCTION OF SITE-DIRECTED MUTANTS AND HETEROLOGOUS GENE EXPRESSION IN *ESCHERICHIA COLI*

The *sor* gene (EMBL accession number X56616) was expressed heterologously using the pASK75 vector and a C-terminal Strep-tag fusion as described elsewhere (pASK-SOR.05 plasmid; Skerra, 1994; Urich et al., 2004). The site-directed mutants of several codons were constructed by using the Quikchange method with

pASK-SOR.05 as a template (Stratagene; now Agilent Technologies, Böblingen, Germany; see Table 1 for a list of the mutants and oligonucleotides used).

The PCR product was digested for 4 h with 10 U *DpnI* (Fermentas; St. Leon-Rot, Germany), subsequently purified via the PCR Clean-Up Kit (Sigma-Aldrich; Steinheim, Germany) and eluted with 25 μl of elution buffer. After transformation of *E. coli* TOP 10' cells (Invitrogen, Darmstadt, Germany) with 6–7 μl of the purified PCR product, the resulting constructs were analyzed by restriction digestion and by sequencing. Plasmid minipreparations of 25 colonies were sequenced using the degenerated MM<sub>296/297</sub> primers, which resulted in the identification of the four mutants M<sub>297</sub>A, M<sub>296</sub>V, MM<sub>296/7</sub>VT, and MM<sub>296/7</sub>TT. The double mutant F<sub>133</sub>A/F<sub>141</sub> was constructed using the F<sub>141</sub>A oligonucleotides with the previously constructed F<sub>133</sub>A mutant plasmid. In the DelL mutant (deletion, l = long), 23 chimney-forming amino acid residues were replaced by three glycines (Figure 1; Table 1). In the DelK derivative (deletion, K = kurz; German for short), 10 residues were replaced by two glycines.

*Escherichia coli* BL 21 Codon plus (DE3) RIL cells (Stratagene) were transformed with the mutant plasmids and the original pASK-SOR.05. The expression of the *sor* genes was induced by addition of anhydrotetracycline (200 μg/l of culture; IBA; Göttingen, Germany) to either 0.5 or 15 l cultures growing at 37°C in 2× LB medium at an OD<sub>600</sub> between 0.6 and 0.8. The cultures were incubated for 20 h after induction with either vigorous shaking (0.5 l) or with vigorous aeration and stirring (15 l). In order to ensure sufficient iron incorporation, 100 μM ferric citrate was added to the media at the time of induction.

### PROTEIN PURIFICATION

The harvested cells were washed once in approximately 10 volumes of 100 mM Tris–HCl/150 mM NaCl buffer pH 8 and then resuspended in five volumes of the same buffer. Cells were disrupted with a High Pressure Homogenizer (Constant Systems; 0.18 mm

**Table 1 | Forward oligonucleotides used in this study for the mutagenesis of the *sor* gene; the corresponding reverse-complimentary oligonucleotides required for the Quikchange method (Stratagene) are not shown.**

Oligonucleotide name	Oligonucleotide sequence*	Comment*
DelK fwd	CCGACTTCACTGCAGTTGTAggaggtggaATTC CAGTTATTTACAACC	Replacement of aa 139-148 with two Gly residues; Addition of <i>EcoRI</i> site
DelL fwd	CAAACATGCCTATAAACACTGGGggtggaATTT CACAACCATATGGAAA	Replacement of aa 129-151 with two additional Gly residues; Removal of <i>PstI</i> site
R <sub>99</sub> A fwd	GGAGTTACTTATTCgcgCTATGCTATTCATGCGC	Addition of <i>Bsh1236I</i> site
R <sub>99</sub> IL fwd	AACTGGAGcTACTTATTCmtcCTATGCTATTCATG	Addition of <i>AluI</i> site
S <sub>226</sub> A fwd	AACCCTGGAGcACTTGAGCCcGATCCAAAT	Removal of <i>PsuI</i> site
S <sub>226</sub> T fwd	AACCCTGGAAcACTTGAGCCcGATCCAAAT	Removal of <i>PsuI</i> site
S <sub>226</sub> IL fwd	AACCCTGGamtACTTGAGCCcGATCCAAAT	Removal of <i>PsuI</i> site
MM <sub>296/297</sub> fwd**	ATTAAATCCArYgryGGAAGGCACcTTCGGAG- TGAAATGACCCGACgcaAtTGCGGTTGTA	Addition of <i>BanI</i> site
F <sub>133</sub> A fwd	GTAGGAAAGAAgcaGCAGAAGGAAAGCCT	Addition of <i>MunI</i> site, double mutant; T134I
F <sub>141</sub> A fwd	GCCTTTGCAGAGcgcGTCAGTAATTC	Removal of <i>PstI</i> site
H <sub>166</sub> A fwd	TGCTTTGTCAGAGcgcGTCAGTAATTC	Removal of <i>Bsp1286I</i> site
H <sub>277</sub> A fwd	TAAGACAAGTAgctGACGAAGTTTT	Removal of <i>TatI</i> site

\*Underlined, restriction sites specified in the comments column.

\*\*Multiple mutants: M<sub>297</sub>A, M<sub>296</sub>V, MM<sub>296/7</sub>VT, and MM<sub>296/7</sub>TT.

nozzle and 1.35 MPa pressure). After a first centrifugation step (10,000 × *g* for 30 min, Sorvall, SLA-3000; Thermo Fisher Scientific, Schwerte, Germany), the soluble protein-containing supernatant was centrifuged in an ultracentrifuge (100,000 × *g* for 45 min, Beckman Instruments, 45Ti). The particle-free protein extracts from 5 to 50 g of cells (wet mass) were applied to an 8 ml Strep-Tactin super-flow column (IBA, Göttingen, Germany) connected to an ÄKTApurifier 10 (GE Healthcare Bio-Sciences AB, Uppsala, Sweden). The elution step was performed with three column volumes of washing buffer containing 2.5 mM desthiobiotin (IBA). The column was washed and regenerated according to the manufacturer's recommendations. Alternatively, the column was regenerated with three column volumes each of ddH<sub>2</sub>O, 0.5 M NaOH, and ddH<sub>2</sub>O instead of the regular HABA solution (IBA).

### ANALYTICAL PROCEDURES

Specific activities of the wild type and mutant proteins were determined by incubation of 2–5 µg of purified enzyme/ml of Tris–HCl buffer pH 7.2 containing 2% sulfur and 0.1% of Tween 20 as described previously (Kletzin, 1989; Urich et al., 2004, 2005b). Samples were taken at appropriate time points (usually after 0, 2, 4, 6, 8, and 10 min) and the concentration of the products was determined colorimetrically. Specific activities were calculated from the linear increase of product concentrations (Kletzin, 1989). At least two activity measurements were performed for each mutant using independently purified protein preparations (see below, Table 2).

Iron quantification was performed with pure protein preparations using the 2,4,6-tripyridyl-1,3,5-triazine method (TPTZ; Fischer and Price, 1964). Protein concentrations were determined by the Coomassie Blue method (Bradford, 1976) or using the BCA Protein Assay Kit (Novagen/Merck, Darmstadt, Germany) according to the manufacturer's instructions. Samples of purified proteins were separated by SDS-PAGE (Schägger and von Jagow, 1987) and visualized with colloidal Coomassie Blue staining (Roti-Blue; Roth, Karlsruhe, Germany).

### INHIBITION ASSAYS

Zinc inhibition assays of the wild type and mutant proteins were performed by addition of a freshly prepared zinc chloride solution to the reaction buffer in concentrations ranging from 0.01 to 1 mM (Kletzin, 1989). After addition of the enzyme, the mixture was incubated for 30 min at room temperature before starting the reaction by heating to 85°C and measurement of the specific activities. A zinc/buffer mixture without protein was used as a negative control.

### CRYSTALLIZATION, DATA COLLECTION, ALIGNMENT AND MODELING

Crystallization of the SOR protein was performed as described previously (Urich et al., 2005a). The mercury derivative was prepared by addition of 5 mM sodium 4-(hydroxymercury)-benzoic acid (or *p*-chloromercuribenzoic acid, *p*-CMB) for 5 days at 22°C to crystallization drops containing grown crystals. The zinc-complexed SOR was prepared by co-crystallization of the SOR in presence of 10 mM

**Table 2 | Specific activities of wild type and mutant SOR, iron content, numbers (#) of preparations and assays per preparation.**

Mutant	Oxygenase spec. activity (U/mg protein)	Reductase spec. activity (U/mg protein)	Fe content absolute (nmol/2.8 nmol protein)	Fe content relative (nmol/nmol subunit)	# Preps   # Assays
Wild type	3.03 ± 0.31	1.69 ± 0.44	2.8	1	4   2
<b>TETRAMER CHANNEL MUTANTS</b>					
delL	12.74	12.02	3.0	1.1	3   3
delK	9.89 ± 1.48	8.05 ± 3.14	3.1	1.1	4   3
F133A	3.89	1.48	nd*	nd*	2   2
F141A	4.43	1.96	4.3	1.6	2   2
F133A/F141A	5.88	5.87	4.8	1.72	2   2
<b>ACTIVE SITE PORE MUTANTS</b>					
M296V	3.03 ± 0.31	2.74 ± 0.71	3.9	1.4	3   3
M297A	1.39 ± 0.27	0.74 ± 0.05	4.3	1.5	3   3
MM296/297VT	1.89 ± 0.11	0.94 ± 0.09	4.0	1.4	3   3
MM296/297TT	2.26 ± 0.57	0.98 ± 0.05	4.4	1.6	3   3
<b>TRIMER CHANNEL MUTANTS</b>					
R99A	5.53 ± 1.21	2.55 ± 0.57	4.0	1.4	4   3
R99I	4.25 ± 0.72	3.82 ± 1.07	2.9	1.0	1   3
S226A	4.12 ± 0.92	2.21 ± 0.47	3.4	1.2	4   3
S226T	5.54 ± 1.34	2.64 ± 0.26	3.4	1.2	4   3
S226I	3.1 ± 0.98	5.15 ± 0.68	3.1	1.1	1   3
S226L	2.68 ± 0.35	8.39 ± 1.19	2.9	1.0	1   3
<b>Zn BINDING SITE MUTANTS</b>					
H277A	2.73 ± 0.46	1.71 ± 0.76	3.3	1.2	2   2
H166A	2.9 ± 0.22	1.38 ± 0.21	6.9	2.4	2   2

\*Not determined.

zinc acetate for 2 days at 32°C. SOR was also co-crystallized with 10 mM IAA for 2 days. After washing in a cryoprotectant solution without the reagents used for soaking, the crystals were flash-frozen in liquid nitrogen (Urich et al., 2005a). The diffraction data sets of the SOR co-crystallized with zinc and IAA were collected with a resolution of 1.7 and 1.9 Å, respectively, at the ESRF beam line 14-1. The Hg<sup>2+</sup> dataset was collected with a resolution of 2.5 Å at the ESRF beam line 14-3.

Images from the three diffraction experiments were processed with DENZO and the observed intensities merged and scaled with SCALEPACK of the HKL Suite (Otwinowski and Minor, 1997). Structure factor amplitudes were then calculated with TRUNCATE and the three sets scaled together against the native data (PDB entry 2CB2-SF; Urich et al., 2006) with SCALEIT of the CCP4.2.2 suite (Collaborative Computational Project, 1994). The final model of the native structure including cysteine persulfide, Fe ions and waters molecules was used as template for structure refinement with REFMAC (Murshudov et al., 1997) or PHENIX (Adams et al., 2010). The stereochemistry of the Zn and Hg centers was refined without target geometrical restraints. After an initial rigid body refinement the models were iteratively refined, analyzed and edited with XTALVIEW-XFIT (McRee, 1999) against electron density maps, until convergence of *R*-values was achieved (Urich et al., 2006).

The sequences used for the multiple alignment (Figure 1) were obtained following a BLAST search at NCBI<sup>2</sup> with the *Ac. ambivalens* SOR sequence as input. The *Sulfobacillus acidophilus* SOR sequence was identified in the almost complete genome sequence available at the JGI genome server<sup>3</sup> (January 2011). The alignment was made with MAFFT using the default parameter at the Kyushu server<sup>4</sup>. The DelL and DelK deletion mutants were modeled using the Phyre<sup>5</sup> (Kelley and Sternberg, 2009) and SwissModel servers<sup>6</sup> (Arnold et al., 2006) with the wild type 3D structure as template. The figures were prepared in Pymol, including the qualitative surface electrostatic distribution (DeLano, 2002).

#### DATA DEPOSITION

The atomic coordinates of the SOR derivatives were deposited at the Protein Data Bank with identification numbers 2yav (Zn derivative), 2yaw (Hg) and 2yax (IAA).

## RESULTS

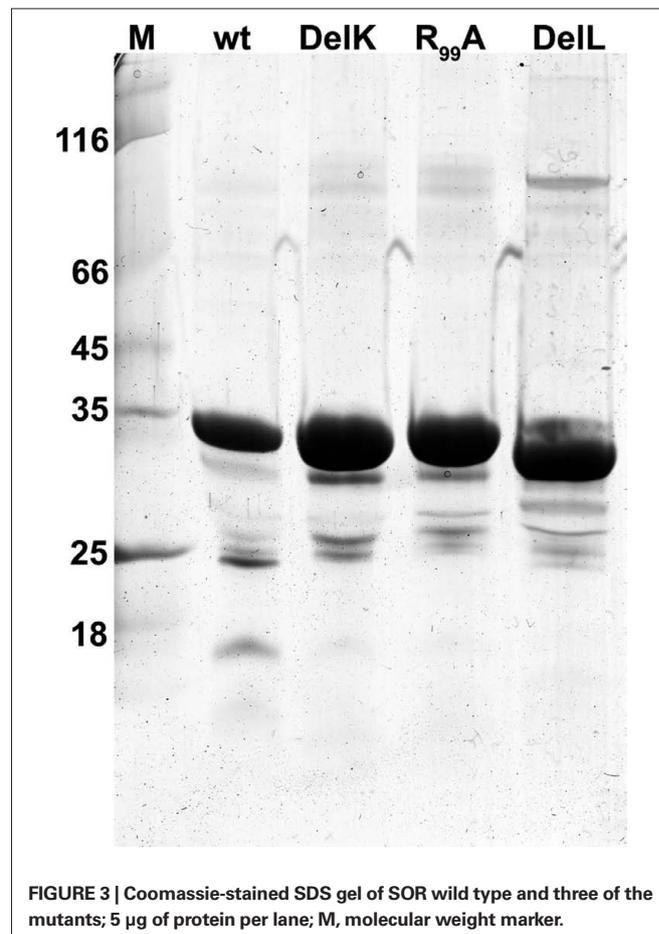
### PROPERTIES OF THE SOR MUTANTS

SOR mutant plasmids generated via site-directed mutagenesis were sequenced and were introduced, if correct, into *E. coli* BL21 Codon Plus cells. Wild type or mutant SOR protein was obtained after overnight incubation of the induced cultures. After breaking of the cells, 10–30% of the protein was present in soluble form while ≥70% precipitated in inclusion bodies as observed previously (Urich et al., 2004). The recombinant wild type and mutant SOR proteins were purified from the soluble *E. coli* fractions in a one-step procedure using the attached *Strep*-tag and *Strep*-Tactin columns. The yields

of purified protein ranged from 0.6 to 1.2 mg/l of culture volume. When analyzed with SDS-PAGE, all SOR preparations were more than 90% homogeneous. In each protein preparation one larger band with a molecular mass of 70 kDa was seen and several smaller bands (Figure 3), the latter of which had been shown to represent cleavage products (Urich et al., 2004). Wild type and mutant proteins were analyzed for specific enzyme activity and iron content in order to verify that altered enzyme activities reflect true mutagenesis effects and not low iron incorporation into the active site. The iron content of the purified proteins was in the range of 1–3 Fe/subunit (Table 3). The average values of specific wild type SOR activities were 3.03 ± 0.31 U/mg (oxygenase) and 1.69 ± 0.44 U/mg (reductase).

### OPENING OF THE OUTER SPHERE INCREASES ACTIVITY

Two mutants were constructed, which feature truncated versions of the chimney-like structures located at the fourfold symmetry axes. In the DelL mutant (deletion, L = long) the 23 amino acid residues that form the protrusions including both phenylalanine rings (Figures 1 and 2D) are replaced by three glycines. In the DelK derivative (deletion, K = kurz; German for short) 10 residues including the outer phenylalanine ring were replaced by two glycines. When modeled into the SOR holoenzyme structure, the pore opened to a diameter of 9–10 Å in the DelL mutant. The atomic distances of the inner phenylalanine ring did not change significantly in the model of the DelK mutant, so that it does not



**FIGURE 3 |** Coomassie-stained SDS gel of SOR wild type and three of the mutants; 5 µg of protein per lane; M, molecular weight marker.

<sup>2</sup><http://www.ncbi.nih.gov>

<sup>3</sup><http://mafft.cbrc.jp/alignment/server/>

<sup>4</sup><http://www.sbg.bio.ic.ac.uk/~phyre/>

<sup>5</sup><http://swissmodel.expasy.org/>

<sup>6</sup><http://www.jgi.doe.gov>

**Table 3 | Diffraction data processing and refinement statistics.**

SOR derivative	SOR-Hg <sup>2+</sup>	Zn <sup>2+</sup>	Iodoacetamide
Source	ESRF ID14-3	ESRF ID14-3	ESRF ID14-3
Space group	I4 (79)	I4 (79)	I4 (79)
Unit cell parameters (Å) a	a = b = 161.88 c = 154.37	a = b = 162.07 c = 154.24	a = b = 161.90 c = 154.27
Wavelength (Å)	0.934	0.934	0.934
No unique intensities	68,600	213,677	172,862
Redundancy	15.0	1.9	2.8
Resolution (outer shell) (Å)	49.04–2.50 (2.53–2.50)	38.20–1.70 (1.76–1.70)	38.05–1.80 (1.85–1.80)
Completeness (outer shell) (%)	99.9 (98.9)	98.3 (94.1)	94.4
R <sub>merge</sub> * (outer shell) (%)	8.1 (48.7)	6.4 (45.9)	5.5 (26.2)
I/s(I) (outer shell)	33.3 (5.0)	12.0 (1.8)	16.7 (1.8)
Wilson B (Å <sup>2</sup> )	43	19	19
<b>REFINEMENT</b>			
Refined structure	1842 aa 526 waters	1842 aa 1382 waters	1842 aa 477 waters
R <sub>work</sub> (%)	16.2	16.3	17.0
R <sub>free</sub> (%)	18.2	19.3	19.0
R (%)	16.7	16.2	
Average ADP (Å <sup>2</sup> )	29	21	29
Bonds RMSD (Å)	0.017	0.017	0.022
Angles RMSD (°)	0.974	1.220	1.710

\*R<sub>merge</sub> =  $\sum |I_o - \langle I \rangle| / \sum I_o$ , where  $\langle I \rangle$  is the average of symmetry equivalent reflections and the summation extends over all observations  $I_o$  for all unique reflections.

display an open pore (from 5.0 Å in the wild type protein to 5.4 Å in the mutant; **Figure 2**). As expected, the apparent molecular masses of the DelL and DelK mutants were slightly smaller in SDS gels compared to the wild type enzyme (**Figure 3**).

Several-fold increased enzyme activities were observed in both cases. DelL showed 420% of the oxygenase and up to 771% of the reductase activities, while DelK showed an increase up to 326% (oxygenase) and 476% (reductase), respectively (**Figure 4; Table 2**). The phenylalanine residues were mutated into alanine independently (F<sub>133</sub>A, F<sub>141</sub>A) and as a double mutant (F<sub>133</sub>A/F<sub>141</sub>A). All three different mutants showed increased activities. The double mutant showed the highest activities (194% of the oxygenase and 347% of the reductase). Mutation of M<sub>130</sub>A located at the base of the channel (**Figure 2**) did not alter the catalytic properties of the enzyme (not shown).

R<sub>99</sub> and S<sub>226</sub>, both located at the postulated channel outlet at the threefold symmetry axis, were substituted for alanines independently. Together with an S<sub>226</sub>T variant, all three mutants showed elevated enzyme activities. R<sub>99</sub>A and S<sub>226</sub>T were comparable having both about 182% oxygenase and 156% reductase activities. Isoleucine and leucine variants of the pore-forming channel outlet residues were comparable to the wild type in oxygenase activity but showed a significantly increased reductase activity of up to 496% in case of S<sub>226</sub>L (**Figure 2; Table 2**).

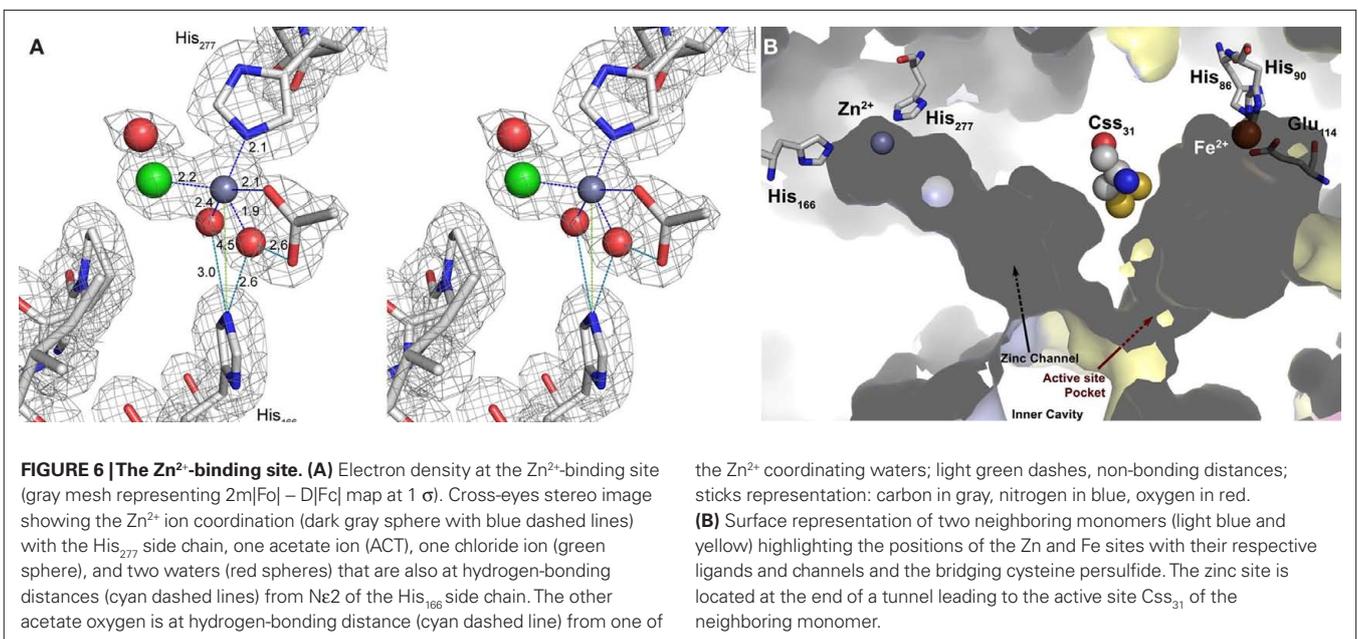
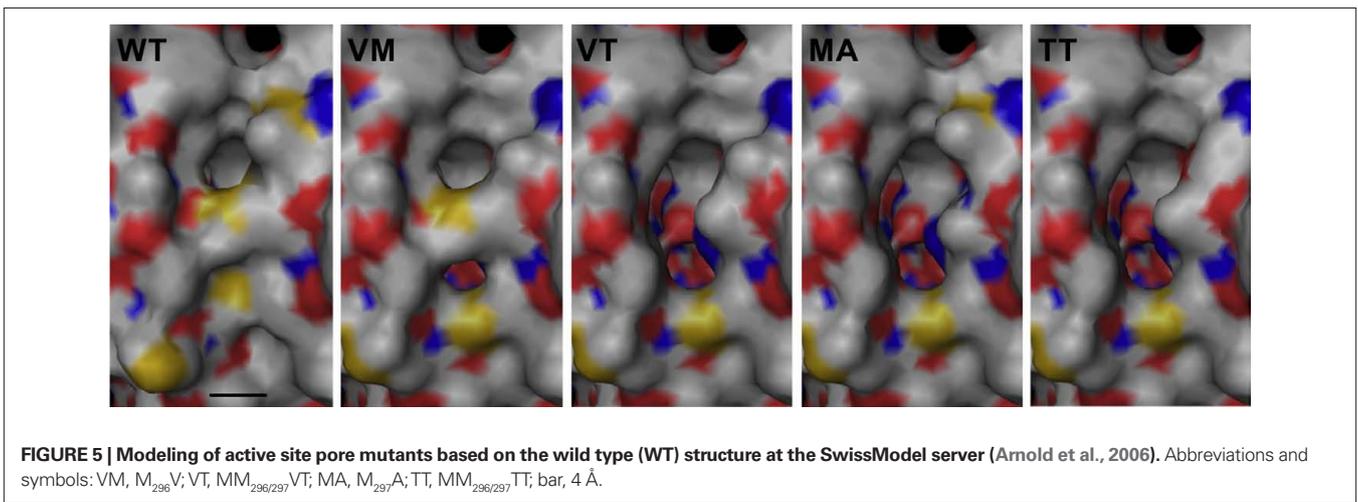
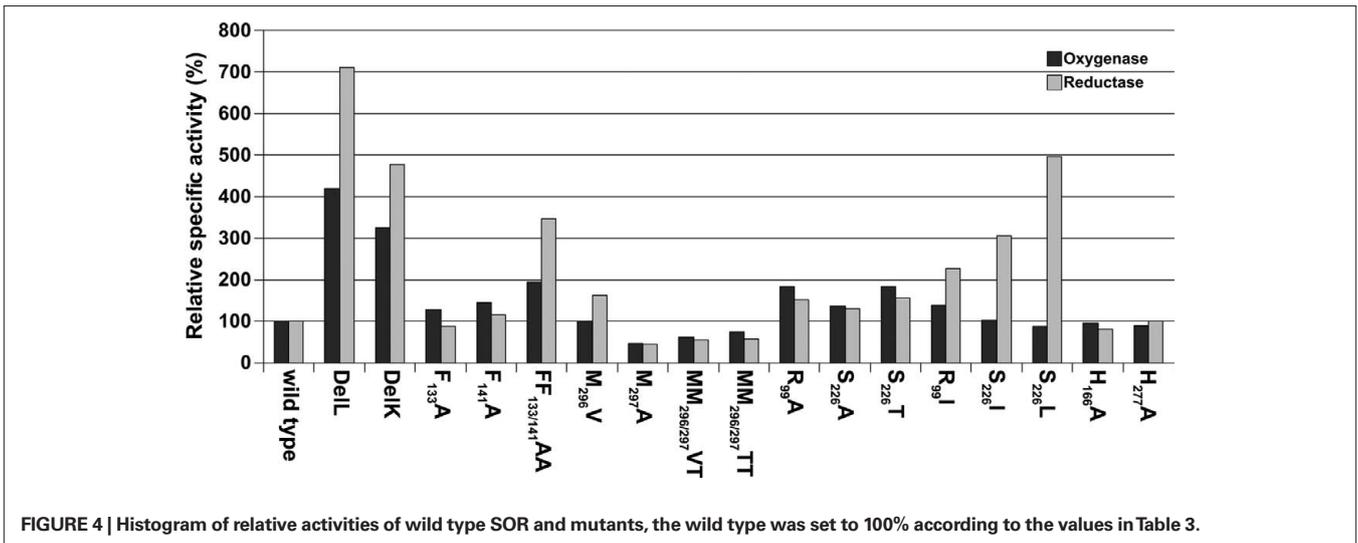
### THE INTEGRITY OF THE ACTIVE SITE PORE IS ESSENTIAL

The active site pore entrance, which provides access to the reaction center, is formed by two adjacent methionines (M<sub>296</sub>/M<sub>297</sub>) and one phenylalanine F<sub>23</sub> (**Figure 5**). We substituted the two methionines via site-directed mutagenesis using degenerated primers that

allowed for 16 variations. Twenty-five different plasmids were screened and four different mutants were obtained, two double mutants, MM<sub>296/297</sub>VT, MM<sub>296/297</sub>TT and two single mutants, M<sub>297</sub>A and M<sub>296</sub>V. Mutagenesis led to an opening of the active site pore as compared to the wild type with the exception of the M<sub>296</sub>V mutant (**Figure 5**). Both double mutants MM<sub>296/297</sub>VT, MM<sub>296/297</sub>TT and also M<sub>297</sub>A showed a decrease to approximately 50% of wild type activity (**Figure 4; Table 2**). The M<sub>296</sub>V mutant showed an increase in reductase activity (162%) but not in oxygenase activity. The latter residue is also present in several of the naturally occurring SORs from other species (**Figure 1**).

### ZINC BINDS FAR FROM THE ACTIVE SITE

Zn<sup>2+</sup> had been shown to be a potent inhibitor of SOR activity as long as it is free in solution and not complexed by ligands such as EDTA (Kletzin, 1989; Urich et al., 2004). Crystals were soaked with Zn<sup>2+</sup> in order to determine its binding site within the SOR. One Zn<sup>2+</sup> ion was detected in the Zn-complexed structure (with 0.8 occupancy) refined at 1.7 Å resolution (**Table 3**). It was located on the opposite side of the beta barrel core of the monomer (not shown; see Urich et al., 2006) but not in close vicinity to the active site (**Figure 6**). The zinc ion was bound in a distance of 2.1 Å to His<sub>277</sub> imidazole (interatomic distances from the here-presented structures are averages over of the six crystallographically independent monomers; **Figure 6**). An acetate and a chloride ion together with two water molecules completed the coordination sphere of Zn<sup>2+</sup> (**Figure 6**). His<sub>166</sub>, which constitutes a conserved 2-His motif together with His<sub>277</sub> (**Figure 1**), had its Ne2 in hydrogen-bonding distance to the two water molecules that coordinate the zinc (**Figure 6**). Both histidines are located at the



bottom of a long, semi-closed channel that opens to the inner cavity of the holoenzyme next to the active site entrance of a neighboring monomer. The Fe–Zn distances were about 27 Å, both within the same subunit and with the neighboring subunit. The side chains of Cys<sub>31</sub> and Met<sub>297</sub> separate the entrances to the active site pocket and the zinc-binding channel. We had postulated that Cys<sub>31</sub> is the sulfur-binding residue in the active site pocket (Urich et al., 2006). The distance between Zn<sup>2+</sup> and the Sδ atom of the cysteine persulfide Cys<sub>31</sub> was 18 Å, so that direct interference of zinc in the catalysis is improbable.

When His<sub>166</sub> and His<sub>277</sub> were substituted independently for alanine, the specific activities were similar to the wild type enzyme (Figure 4; Table 2). *K<sub>i</sub>*-values (half-maximal inhibitory concentration) of the wild type enzyme were 45 μM zinc chloride for the oxygenase activity and 39 μM for the reductase activity. *K<sub>i</sub>*-values for H<sub>166</sub>A were 121 μM (oxygenase) and 150 μM (reductase). The mutant H<sub>277</sub>A showed comparable *K<sub>i</sub>*-values of 157 μM (oxygenase) and 144 μM (reductase).

### MERCURY AND IODOACETAMIDE BIND AT ACTIVE SITE CYSTEINES

Analysis of the crystallographic model of the *p*-CMB treated crystal refined at 2.5 Å resolution (Table 3) showed in the active site pocket two partially occupied Hg<sup>2+</sup> ions per monomer. One Hg<sup>2+</sup>, with 0.5 occupancy, was bound at a distance of 2.1 Å to the Sγ atom of the cysteine persulfide (Cys<sub>31</sub>; Figure 7). A putative acetate ion refined reasonably well, also with 0.5 occupancy, with one of its carboxylic oxygens coordinating Hg<sup>2+</sup> at 2.0 Å distance, although the limited resolution maps also show some density noise in its neighborhood. The second mercury ion, with 0.3 occupancy, refined at 2.6 Å distance from Sγ of non-essential cysteine Cys<sub>101</sub>, with the side chain in *trans* conformation for torsional angle chi1, and with occupancy 0.3. The other side chain conformation, chi1 *gauche*(–) (with occupancy 0.7) corresponds to that of the native structure. This Hg<sup>2+</sup> is not too far away from the Sδ atom of Met<sub>108</sub>, at 3.7 Å distance. Thus, the inhibition of the enzyme by mercury compounds (Kletzin, 1989) is caused by modification of the active site cysteine(s).

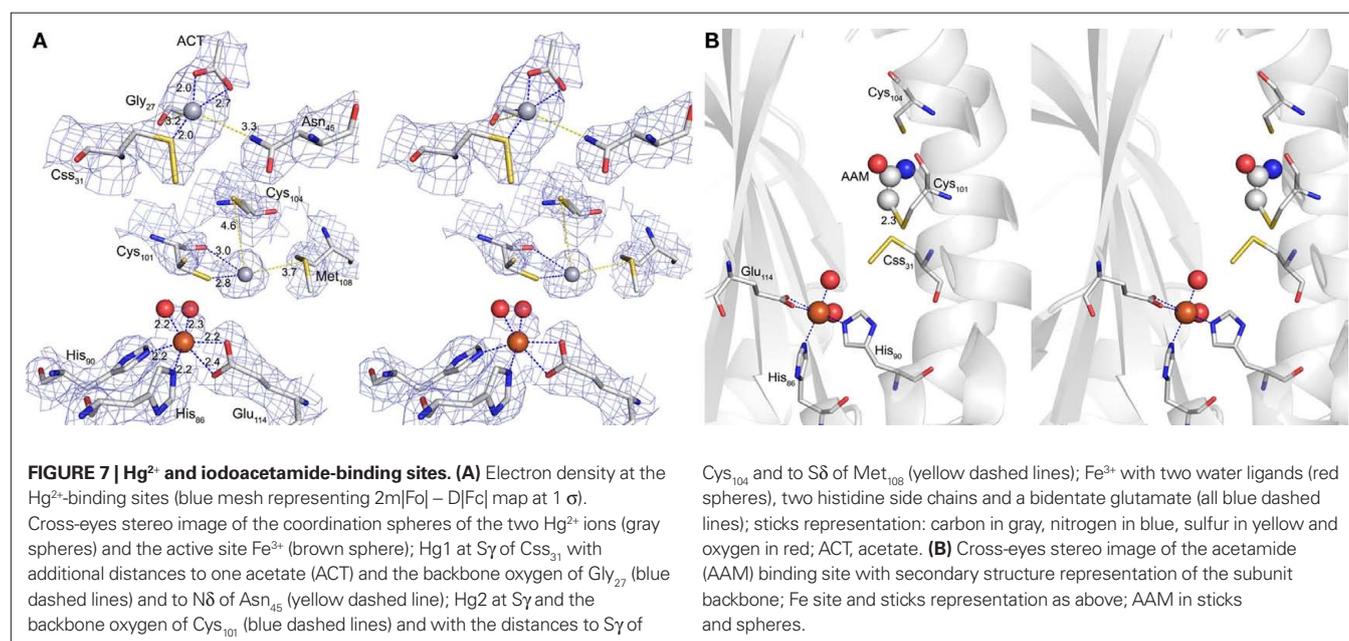
The co-crystallization with the thiol-modifying reagent IAA resulted in additional electron density resembling acetamide in covalent distance to Cys<sub>101</sub> (Figure 7). The electron density was present with low occupancy in three out of six monomers in the asymmetric unit. No additional density was observed at Cys<sub>31</sub> or Cys<sub>104</sub>. In consequence, it appears that Cys<sub>101</sub> is the primary target for alkylation with IAA, although this conclusion is tentative due to the low reactivity between enzyme and inhibitor in this experiment.

### DISCUSSION

Sulfur oxygenase reductases are uncommon sulfur-disproportionating enzymes restricted to a few species of sulfur-oxidizing Archaea and Bacteria, which are either facultative or obligatory chemolithoautotrophs (Figure 1). Determination the 3D structure of two of these enzymes showed that they form large hollow homomultimers with more or less secluded inner chambers. Each subunit contains a reaction pocket with conserved sites consisting of a mononuclear iron and a cysteine residue (Urich et al., 2004, 2005b, 2006; Chen et al., 2005; Li et al., 2008). The SORs thus provide enclosed reaction and/or storage compartments physically separated from the cytoplasm (Figure 2). Despite previous biochemical and mutagenesis studies it is still unclear how the proteins work in detail. Here, we focus on the pores in the protein and on the mode of action of important inhibitors of enzyme activity.

### THE PORES AT THE FOURFOLD SYMMETRY AXIS RESTRICT ENZYME ACTIVITY

Narrow pores padded with apolar amino acids are localized at the chimney-like structures at the fourfold symmetry axes (Figure 2). They were already considered to be the substrate entrance points to the inner cavity of SOR (Urich et al., 2006; Li et al., 2008). Two rings of four phenylalanine side chains each define these pores in the present 3D model (Figure 2). Here we show that mutation of the Phe residues into Ala, which induces enlargement of the pores, gave a moderate, less than twofold increase in specific enzyme



activity (Figure 4). A full deletion of the loop building the chimneys (Figures 1 and 2) resulted in a more than sevenfold increase in enzyme activity showing that substrate access to the active site and/or product exit is indeed limited by the outer shell of the protein. Therefore, it can be concluded that the specific activity of the wild type SOR is curbed to a lower level than optimally possible. The reasons for making the enzyme slow might be speculated upon. One option (among others) would predict that the reactive reaction products are not released uncontrollably into the cytoplasm but that they are delivered directly to the downstream oxidoreductases (Kletzin, 2008). One experimental indication for this interpretation – in the absence of known interaction partners – came from antibody/immunogold electron microscopy results of the *Ac. tengchongensis* SOR: the enzyme seemed to be attached to the inside of the cytoplasmic membrane (Chen et al., 2005). In addition, these pores might provide highly controlled access points to the active sites, preventing the oxidation of “unwanted substrates.”

### MUTAGENESIS AT THE THREEFOLD SYMMETRY AXIS AND REACTION MECHANISM

The amino acids Arg<sub>99</sub> and Ser<sub>226</sub> are central elements of the subunit interface at the threefold symmetry axis. Arg<sub>99</sub> forms an intra-subunit salt bridge to Glu<sub>228</sub> (Figure 2). The other η-nitrogen atom is in hydrogen-bonding distance (2.8 Å) to the Oγ of Ser<sub>226</sub> of the neighboring subunit. Alternative hydrogen-bonding networks are also possible: Arg<sub>99</sub> might link to the α-carbonyl oxygen atoms of Ser<sub>226</sub> of two subunits. In addition, a hypothetical salt bridge between the ε-nitrogen of the Arg<sub>99</sub> and Glu<sub>228</sub> is possible, provided that an extensive charge delocalization exists in the guanidinium group. We did not obtain protein of an E<sub>228</sub>A mutant from *E. coli* cells, so that we could not analyze its effects on salt bridge formation (data not shown). In contrast, mutation of Arg<sub>99</sub> and Ser<sub>226</sub> into Ala gave a modest, less than 1.5-fold increase in specific activity (Figure 4). Mutation into more hydrophobic residues changed the ratio between oxidized and reduced reaction products in favor of sulfide (Figure 4; Table 2). The same had happened in a less pronounced way in the chimney mutants DelK and DelL, suggesting that opening of the closed reaction chamber in the interior of the protein changes the ratio between the oxygenase and disproportionation partial reactions (Eqs 1 and 2).

The interpretation of the activity data is complicated by non-enzymatic reactions occurring with sulfur and ISC in aqueous solutions, which depend on the incubation temperature and the pH of the buffer. For example, sulfite reacts rapidly with excess sulfur to thiosulfate at pH ≥ 5 and 85°C so that it is still unclear whether thiosulfate is a primary or secondary reaction product of the SOR (Kletzin, 1989). The sulfur disproportionation depicted in Eq. 2 and with H<sub>2</sub>S, polysulfides and thiosulfate (instead of sulfite) as products occurs non-enzymatically at alkaline pH (detectable above pH 7.5 at 75°C; Roy and Trudinger, 1970; Kletzin, 1989). We had concluded from previous activity assays in the presence of chemically complexed zinc (to overcome zinc inhibition and to precipitate sulfide *in situ*) that an approximate 1:1 ratio between oxidized and reduced reaction products is maintained (Eq. 3) as opposed to the 1:2 ratio for the oxygen-independent disproportionation shown in Eq. 2. We also had assumed that the sub-stoichiometric hydrogen sulfide detection in the standard assay is due to the rapid

non-enzymatic re-oxidation under the aerobic assay conditions (see Table 2, wild type enzyme; Kletzin, 1989, 2008; Kletzin et al., 2004; Urich et al., 2004). This picture however changes with the channel mutants, which prove that the ratio between oxidized and reduced reaction products is not constant and that it depends on the integrity of the protein shell.

### THE INTEGRITY OF THE ACTIVE SITE PORE SEEMS TO BE IMPORTANT

A different story resulted from the mutations at the active site pore, made from two adjacent methionines, M<sub>296</sub>/M<sub>297</sub> and a phenylalanine, F<sub>23</sub>. M<sub>297</sub> is conserved in all SORs, while M<sub>296</sub> is exchanged for another hydrophobic amino acid in several naturally occurring SOR sequences (Figure 1). F<sub>23</sub> is conserved in Archaea while Bacteria mostly use methionine at this position. The pore supposedly represents the entrance of substrate and the exit of the products; at least, we were unable to find a “back exit” in the active site pocket of the *Ac. ambivalens* enzyme as suggested for the *Ac. tengchongensis* SOR (Li et al., 2008). Expansion of the active site pore did not increase the enzyme activity, which is contrasting our observations on the tetramer and trimer channels (Figures 4 and 5). Replacement of the hydrophobic Met residues with the smaller and in the case of the threonine more hydrophilic residues diminished the specific activity by half, suggesting that the hydrophobic barrier and/or at least one of the methionines are essential. One could speculate that non-covalent nucleophilic interaction between the atoms of the sulfur substrate and the Sδ atom of the gate-keeping methionine residues could direct the sulfur substrate toward the active site cavity.

### DIFFERENT MECHANISMS OF INHIBITOR ACTION

Zinc and the thiol-binding agents *p*-CMB, IAA and maleimide were previously shown to act as inhibitors (Kletzin, 1989). No activity had been observed in the presence of 1 mM Hg<sup>2+</sup> or Zn<sup>2+</sup> ions, whereas 0.1 mM resulted in 37 and 74% residual activity, respectively, for the oxidized reaction products (Kletzin, 1989). Inhibition by Zn<sup>2+</sup> was neutralized by addition of at least equimolar concentrations of EDTA (Urich et al., 2004). Both Hg<sup>2+</sup> and Zn<sup>2+</sup> ions can theoretically bind to sulfur atoms of cysteine(s), to histidine and/or carboxylate ligands. The alkylating agents IAA and NEM were less potent inhibitors: 27% of oxygenase activity remained in the presence of 1 mM IAA (Kletzin, 1989), while 47% remained with 1 mM NEM. In order to determine the mechanisms of inhibition, we resolved the 3D structures of wild type SOR crystals soaked in *p*-CMB, IAA, and Zn<sup>2+</sup>.

Two Hg<sup>2+</sup> ions were found in the active site pocket of *p*-CMB treated crystals, one of them in bonding distance (2.1 Å) to the Sy atom of the essential cysteine persulfide Cys<sub>31</sub>, thus explaining the inhibitory effect of Hg<sup>2+</sup> (Figure 7). A second ligand was provided by the oxygen (at 2.0 Å distance) of a putative acetate from the crystallization media. The second Hg<sup>2+</sup> ion is bound (2.6 Å) to Sy of conserved cysteine Cys<sub>101</sub>. Although unresolved, water molecules might provide additional ligands as both mercury ions face the lumen of the water-filled active site pocket. To conclude, the main mechanism of inhibition seems to be the result of mercury binding to Cys<sub>31</sub>, an amino acid that cannot be mutated without total loss of enzyme activity (Urich et al., 2005b). Mercury binding to Cys<sub>101</sub> seems to play a less important role.

In contrast, the effect of IAA is more difficult to explain. Acetamide bound to Cys<sub>101</sub> was observed only in three out of six subunits of the asymmetric unit and only with low occupancy (Figure 7). This observation has two implications. First it shows that the 24 subunits are not equal, minor distortions could be the result of inhibitor and/or substrate binding. Distortions were so far not seen at all because of the sheer number of subunits and the rotational symmetry of the holoenzyme. Both would average out minor distortions in the electron density and make subtle changes undetectable. We had tried to soak crystals with many different ISC without seeing additional electron density anywhere in the molecule (Urich, 2005), presumably because substrate or product binding occur only in a minority of subunits at a given time. Bound sulfur species would be averaged out within the holoenzyme. This situation is slightly different with IAA, where the low occupancy points into the same direction, supported by the observation that not all of the potential binding sites are occupied. However, the stoichiometry of inhibitor binding is high enough to facilitate the positive identification of the bound molecule. A second implication came from the binding site: Cys<sub>101</sub> could be mutated to Ala with partial loss of activity ( $\approx 20\%$ ; Urich et al., 2005b). A Cys<sub>101</sub>S mutant had very low residual activity ( $\approx 1\%$ ) concomitant with a very low iron content ( $< 0.1$ /subunit; Urich et al., 2005b). This observation showed that the chemical nature of the amino acid side chain at this position directly affects enzyme activity and active site integrity, although the cysteine itself is not essential.

Inhibition by zinc seems to follow a completely different mechanism. The Zn<sup>2+</sup> ion was found in a dead-end channel, which opens next to the active site pore of the adjacent subunit. The active site Cys<sub>31</sub> separates the lumina of both channels (Figure 6). The question arises how zinc inhibits the enzyme activity over a distance of approximately 27 Å to the next iron. The zinc ion is coordinated by histidine residue H<sub>277</sub>, one acetate, one chloride and two water molecules that in turn are hydrogen-bonding His<sub>166</sub> (Figure 6). Mutation of either histidine into alanine did not alter enzyme activity significantly. However, their  $K_i$ -values for Zn<sup>2+</sup> increased two- to threefold, compared to the wild type. The 2-His motif is conserved in the SOR sequences (Figure 1). It cannot be answered

at present which role the obviously conserved zinc channel plays. In contrast, several options exist about the mechanism of inhibition at a distance. The Zn<sup>2+</sup> ion might make the protein less flexible and thus block the substrate entry or product exit from the active site. In addition, it might block the important Cys<sub>31</sub> residue, which is located at the interface of both channels, in its movements during the catalytic cycle. Stiffening of the protein seems to be the most probable mechanism of inhibition because no further connection between the zinc channel and the active site pocket is present in the enzyme.

## CONCLUSION

From these and previous findings we can propose a hypothetical model of the sulfur pathway in the SOR and the different modes of enzyme inhibition. The chimney-like structures at the four-fold symmetry axes formed by two phenylalanine rings are not essential for activity, but presumably act as restrictive elements for the access of the hydrophobic sulfur substrate to the inner hollow. Product exit might occur via hydrophilic channels, which have their outlets at the threefold symmetry axes. Enlargement of both openings increased the enzyme activity several-fold and also affected the formation of H<sub>2</sub>S and the stoichiometry of reaction products. In contrast, the integrity of the active site pore, which provides a passage to the catalytic center, cannot be opened significantly without decreasing the specific activity of the enzyme. The inhibition of the SOR activity by Hg<sup>2+</sup> and IAA occurs by binding of the compounds to different cysteine residues within the active site. In contrast, Zn<sup>2+</sup> does not bind anywhere in the active site but in a separate channel. It might restrict protein flexibility and/or substrate and product movement within the protein subunits. It might also effect movements of the active site-cysteine persulfide (Cys<sub>31</sub>) side chain.

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