

# The Antimicrobial Activity and Characterization of Bioactive Compounds in *Peganum harmala* L. Based on HPLC and HS-SPME-GC-MS

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Wang N, An J, Zhang Z, Liu Y, Fang J and Yang Z (2022) The Antimicrobial Activity and Characterization of Bioactive Compounds in Peganum harmala L. Based on HPLC and HS-SPME-GC-MS. Front. Microbiol. 13:916371. doi: 10.3389/fmicb.2022.916371 Peganum harmala L. is a perennial herb of the Tribulus family and its aerial parts and seeds can be used as medicine in the traditional medicine of China. However, the differences in chemical components and antibacterial activity between different parts have not been reported. In this study, the chemical composition of the different parts of P. harmala was characterized by high-performance liquid chromatography (HPLC) and headspace-solid phase microextraction-gas chromatography-mass spectrometry (HS-SPME-GC-MS). The antimicrobial activities of the different parts and some isolated components were also carried out on 12 bacterial strains and phytopathogenic fungi. The HPLC results revealed that the contents of harmine and harmaline in the seeds were higher than that in the aerial parts. A total of 94 volatile organic compounds (VOCs) were tentatively identified by HS-SPME-GC-MS for the first time. The major components were methyl hexadecanoate, p-xylene, octane, (Z)-9-octadecanoate, ethylbenzene, methyl octadecanoate, ethyl hexadecanoate, and methyl tetradecanoate. At the concentration of 800  $\mu$ g·mL<sup>-1</sup>, the methanol extracts of seeds showed stronger antimicrobial activities with a wide antimicrobial spectrum, inhibiting Escherichia coli (ATCC 24433), Xanthomonas oryzae (ACCC 11602), and Xanthomonas axonopodis with inhibitory rates of more than 90%. Furthermore, harmine and harmaline showed better antibacterial activities against all the bacteria. These findings indicated that alkaloids from P. harmala could account for antimicrobial activity, which could be used as lead molecules in the development of new antimicrobial drugs.

Keywords: alkaloids, HS-SPME-GC-MS, volatile organic compounds, antimicrobial activity, Peganum harmala L

# INTRODUCTION

*Peganum harmala* L. is widely distributed in the arid areas of Northwest China, and the aerial parts and seeds of *P. harmala* can be used as medicine in the traditional medicine of Uygur, Kazak, Mongolian, and Tibetan, which possesses neuroprotective, anticancer, antiviral, larvicidal, hepatoprotective activity, antioxidant and anti-inflammatory, antitussive, expectorant,

and bronchodilating effects (Li et al., 2017). It has been reported that the main compounds of P. harmala were alkaloids, such as  $\beta$ -carboline and quinazolone alkaloids and their derivatives (Jalali et al., 2021). However, the differences in chemical components between different parts have not been reported. High-performance liquid chromatography (HPLC) has the characteristics of high sensitivity, high separation efficiency, easy operation, and good repeatability (Sahu et al., 2018; Martinez-Navarro et al., 2021). Gas chromatography-mass spectrometry (GC-MS) combined with solid-phase microextraction (SPME) provides a more environmentally friendly, simple, and accurate method for the study of volatile components, which are widely used in the detection of volatile organic compounds (VOCs) in plants (Feng et al., 2022). The crushed leaves and stems of P. harmala have a disagreeable odor (Xu et al., 2008). However, the VOCs of P. harmala has not been reported using HS-SPME-GC-MS, and its volatile components in the aerial parts are still poorly understood. To explore the chemical constituents of P. harmala L., its aerial parts and seeds were analyzed by HPLC and HS-SPME-GC-MS.

Natural products derived from plants have significant efficacy with antibacterial activity, particularly against pathogens that could be used to overcome drug resistance (Ye et al., 2020). For example, it has been demonstrated that *P. harmala* has antimicrobial activities, effectively inhibiting the growth of *Staphylococcus aureus* (DSM 25693), *Bacillus cereus* (DSM 4313), *Bacillus cereus* (DSM4384), *Escherichia coli* (DMS 857), and *Pseudomonas aeruginosa* (ATCC 50071) (Azzazy et al., 2021). But *P. harmala* has not been reported for potential antifungal activities against phytopathogenic fungi.

In the present study, to clarify the chemicals of different parts of *P. harmala*, the HPLC method was established for the simultaneous determination of harmine, harmaline, vasicine, and vasicinone in its aerial parts and seeds. HS-SPME-GC-MS technology was also used to determine the differences of VOCs in different parts of *P. harmala*. Furthermore, extracts from different parts of *P. harmala* were assayed against 12 bacterial strains and phytopathogenic fungi. To our knowledge, this is the first time to report the HPLC, HS-SPME-GC-MS, and antimicrobial activity results of different parts of *P. harmala*.

# MATERIALS AND METHODS

## **Materials and Reagents**

Fresh aerial parts of *P. harmala* L. were collected and dried at room temperature in October 2020 from Pakistan and Gansu of China. Dried seeds were collected from Xinjiang of China. The plants were identified by Dr. Zhigang Yang, and the voucher specimen (No. 202010002) was stored in the School of Pharmacy at Lanzhou University.

Ethanol, methanol, ethyl acetate, dichloromethane, and chloroform were purchased from Tianjin Jiatong Chemical reagent Co., Ltd., Dimethylsulfoxide (DMSO) was obtained from Sigma-Aldrich (USA); Acetonitrile was purchased from Merck (Germany).

# **Extraction and Separation**

The seeds of *P. harmala* (9.0 kg) were extracted under reflux with 70% ethanol for three times. The extracts (1.6 kg) were suspended in 5% HCl, then the acidic mixture was successively extracted with dichloromethane and the aqueous layer was alkalinized to pH 9 with NaOH solution, followed by exhaustive extraction with dichloromethane to obtain crude alkaloids (273 g). The crude alkaloids were subjected to silica gel column chromatography, eluting with CH<sub>2</sub>Cl<sub>2</sub>:MeOH (100:0-0:100), to obtain nine fractions (Fr.A-I) (Wang et al., 2017). Fr.E was separated on Al<sub>2</sub>O<sub>3</sub> chromatography and eluted by chloroform and methanol to obtain compound 1 (1.4g). Fr.F was separated on silica gel chromatography, eluting with CH<sub>2</sub>Cl<sub>2</sub>:MeOH, and obtained compound 2 (0.1 g) by recrystallization in MeOH. Compound 3 (1.4 g) was obtained from Fr.B via recrystallization in MeOH. Fr.B was separated on silica gel chromatography, eluting with dichloromethane and acetone, and obtained compound 4 (0.1 g). Their structures were elucidated based on nuclear magnetic resonance (NMR). The NMR experiments were performed on a Bruker AVANCE AV III-400 instrument (Bruker, Switzerland, 100 MHz for <sup>13</sup>C and 400 MHz for <sup>1</sup>H) with tetramethylsilane (TMS) as an internal reference.

## HPLC Chromatographic Conditions Preparation of Standard Compound Solutions

The vasicine (3.53 mg), vasicinone (1.54 mg), harmine (6.66 mg), and harmaline (5.90 mg) were dissolved in methanol and configured at different gradient concentrations.

## **Preparation of Samples Solutions**

In total, 0.25 g powders of different parts from *P. harmala* were extracted by ultrasonic treatment with 25 ml methanol for 30 min at 30°C using an ultrasonic cleaner (Model KQ5200E, Kunshan ultrasonic instrument Co. LTD, China). Then, the sample was filtered by a membrane filter ( $0.22 \,\mu$ m) for HPLC analysis.

#### **HPLC Analysis**

EasySep-3030 liquid chromatography (Shanghai Tongwei Company) included an EasySep3030 UV detector, AS-2000 automatic sampler, and EasySep3030 binary pump. The column was C18 (4.6  $\times$  250 mm, 5  $\mu$ m, Shanghai Tongwei), and the mobile phase A was acetonitrile and mobile phase B was 0.6% glacial acetic acid solution containing 0.088 mol/L ammonium acetate. The HPLC elution conditions were operated as follows: 0–7 min, 10% A; 7.1–25 min, 19% A; 25.1–37 min, 27% A; 37.1–45 min, 27–40% A; and 45.1–60 min, 100–10% A. The injection volume was 10  $\mu$ l and eluted at 1.0 ml/min, the column temperature was kept at 30°C, and the detection wavelength was 265 nm (Wen et al., 2012).

# HS-SPME-GC-MS Conditions

#### Isolation and Concentration of GC-MS

In total, 1 g powder of samples was transferred immediately to a 20-ml head-space vial (Agilent, Palo Alto, CA, USA), containing NaCl saturated solution. The vials were sealed using crimp-top caps with TFE-silicone headspace septa (Agilent). At the time of SPME analysis, each vial was placed at  $100^{\circ}$ C for 5 min, then



TABLE 1 | Linear regression equations, linear range, and correlation coefficients for the four alkaloids.

Components	Equation of linear regression	The linear range ( $\mu g \cdot L^{-1}$ )	R <sup>2</sup>
Harmine	y = 14.757x + 260.16	60.4–3330.0	0.9998
Harmaline	y = 25.23x + 643.43	22.9–2950.0	0.9990
Vasicine	y = 18.952x + 1.4676	1.2–235.3	0.9992
Vasicinone	y = 20.104x - 14.036	0.5–102.7	0.9992

a 120-µm divinylbenzene/carboxen/polydimethylsilioxane fiber (Agilent) was exposed to the headspace of the sample for 15 min at 250°C.

## **GC-MS** Conditions

After sampling, desorption of the VOCs from the fiber coating was carried out in the injection port of the GC-MS (Agilent; Model 8890, 5977B) at 250°C for 5 min in the splitless mode. The GC was equipped with a  $30 \text{ m} \times 0.25 \text{ mm} \times 0.25 \mu \text{m}$  DB-5MS (5% phenyl-polymethylsiloxane) capillary column. Helium was used as the carrier gas at a linear velocity of 1.2 ml/min. The injector temperature was kept at 250°C and the detector at 280°C. The oven temperature was programmed from 40°C (3.5 min), increasing at  $10^{\circ}$  C/min to  $100^{\circ}$  C, at  $7^{\circ}$  C/min to  $180^{\circ}$  C, and at 25°C/min to 280°C, and held for 5 min. Mass spectra were recorded in electron impact (EI) ionization mode at 70 eV, using the full scan mode. The quadrupole mass detector, ion source, and transfer line temperatures were set at 150, 230, and 280°C, respectively. Mass spectra were scanned in the range m/z 50 to 500 amu at 1 s intervals. Identification of volatile compounds was achieved by comparing the mass spectra with the data system library (NIST) and linear retention index. The concentration of each compound in the *P. harmala* was calculated by the internal standard method.

# **Determination of Antibacterial Activity**

The petroleum ether and methanol extracts were prepared as follows: 0.50 g powder of seeds and the aerial parts were refluxed with 100 ml petroleum ether and methanol three times. After filtration, concentration, and evaporation, the petroleum ether and methanol extract were dried and then obtained for antibacterial activity tests. E. coli (ATCC 25922), Staphylococcus aureus (Newman), X. oryzae (ACCC 11602), X. axonopodis, and Pectobacterium atroseptica (ACCC 19901) were provided by the Institute of Plant Protection, Gansu Academy of Agricultural Science. The strains were removed and underlined on NB solid medium and incubated at 28°C until single colonies grew. Colonies on solid medium were picked for agricultural bacterial NB liquid medium at 28°C and 180 rpm and shaken to grow. Strains in the log growth phase were diluted to about 10<sup>6</sup> CFU/ml. The samples were separately dissolved in dimethyl sulfoxide (DMSO) and then added to the liquid medium with a





concentration of 1,600 µg/ml (0.5% DMSO). Before inoculating each well with 50 µl of bacterial suspension, a 2-fold series of NB dilutions of 50 µl of extracts and alkaloids were prepared in a 96-well plate (final concentrations of extracts were 800, 400, 200, 100, and 50 µg/ml; final concentrations of alkaloids were 100, 50, 25, and 12.5 µg/ml). A 100-µl suspension of the same concentration containing the same volume of DMSO was used as a control. The plates were then incubated at 28°C for 24 to 48 h and the optical density at 600 nm (OD<sub>600</sub>) was measured. The OD<sub>600</sub> of 100 µl of NB solution containing the same concentration of DMSO or compounds was measured for correction. The inhibition rate was calculated using the following formula (Ferreira-Santos et al., 2021):

The corrected values = 
$$OD_{bacterial wilt} - OD_{no bacterial wilt}(1)$$
  
The inhibition rate =  $(C - T)/C \times 100\%$  (2)

Where C is the corrected turbidity values of control and T is the corrected turbidity values of bacterial growth treated with compounds. The MIC values were defined as the concentration of the compound that inhibited bacterial growth by, at least, 90% (the  $MIC_{90}$ ).

# **Determination of Antifungal Activity**

Seven pathogenic fungi, *Candida albicans* (ATCC 24433), *Rhizoctonia solani, Sclerotinia sclerotiorum, Botrytis cinerea, Fusarium graminearum, Magnaporthe oryzae*, and *Phytophthora capsici* were obtained from the Institute of Plant Protection, Gansu Academy of Agricultural Science. The fungi were cultured in potato dextrose agar medium (PDA) at  $25^{\circ}$ C for 3–6 days to obtain new mycelia for the antifungal assay. The mycelium growth inhibition method was used to evaluate the antifungal activity of *P. harmala*. First, the petroleum ether, methanol



extracts, and alkaloids were dissolved in DMSO and then added to the PDA medium to obtain sterile media of different concentrations. Then, a 5-mm agar plug of each fungal strain was obtained from a 3-day-old PDA culture, which was inoculated in the middle of PDA plates containing samples. The same amount of DMSO (0.5%) was added to the sterile medium as a blank control. Petri dishes were incubated at 25°C, and the mycelial growth diameter was measured using the crossbracketing method until the fungal growth of the blank control had completely covered the Petri dishes. The inhibitory rates were calculated according to the following formula (Yan et al., 2021):

Mycelial growth inhibition rate (%) = 
$$[(C - T)/(C - 5 mm)]$$
  
× 100 (3)

Where C and T are the average colony diameters of the mycelium of the blank control and treatment, respectively.

# **Statistical Analysis**

All test data were calculated for the means and SDs using the SPSS 22.0 software, one-way ANOVA was employed for the significant differences test, and the means were compared by Duncan's tests at P < 0.05 and P < 0.01.

# **RESULTS AND DISCUSSION**

## NMR Data of the Isolated Alkaloids

Compound **1** was obtained as pale yellow prismatic crystallization and showed purple fluorescence at 365 nm. <sup>1</sup>H-NMR (400 MHz DMSO- $d_6$ )  $\delta$ ppm: 11.45 (1H, s, NH), 8.17 (1H, d, *J* =5.3 Hz), 8.06 (1H, d, *J* =8.6 Hz), 7.82 (1H, d, *J* =5.3 Hz), 7.03 (1H, d, *J* =2.0 Hz), 6.87 (1H, dd, *J* =8.6, 2.0 Hz), 3.88 (3H, s, OCH<sub>3</sub>), and 2.74 (3H, s, CH<sub>3</sub>); <sup>13</sup>C-NMR (100 MHz DMSO- $d_6$ )  $\delta$ ppm: 142.4 (C-1), 138.1 (C-3), 112.4 (C-4), 123.1 (C-5), 109.5 (C-6), 160.6 (C-7), 95.1 (C-8), 135.0 (C-10), 127.7 (C-11), 115.3 (C-12), 141.7 (C-13), 55.8 (OCH<sub>3</sub>), and 20.8 (CH<sub>3</sub>). Thus, the structure of compound **1** was determined as harmaline by comparing the NMR data with the literature (Benarous et al., 2015).

Compound **2** was obtained as yellow sheet crystallization and showed bright blue fluorescence at 365 nm. <sup>1</sup>H-NMR (400 MHz DMSO-*d*<sub>6</sub>)  $\delta$ ppm: 11.16 (1H, s, NH), 7.43 (1H, d, *J* = 8.7 Hz), 6.86 (1H, d, *J* = 1.8 Hz), 6.71 (1H, dd, *J* = 8.7, 2.2 Hz), 3.79 (3H, s, OCH<sub>3</sub>), 3.65 (2H, t, *J* = 8.8 Hz), 2.70 (3H, t, 8.4 Hz), and 2.25 (3H, s, CH<sub>3</sub>); <sup>13</sup>C-NMR (100 MHz DMSO-*d*<sub>6</sub>)  $\delta$ ppm: 157.7 (C-1), 48.0 (C-3), 19.6 (C-4), 119.9 (C-5), 110.7 (C-6), 157.3 (C-7), 95.1 (C-8), 129.0 (C-10), 120.8 (C-11), 115.2 (C-12), 138.1 (C-13), 55.6 (OCH<sub>3</sub>), and 22.4 (CH<sub>3</sub>). Therefore, the structure of compound **2** was elucidated as harmine by comparing the NMR data with the literature (Mukhtar et al., 2002).

TABLE 2 | Information of volatile chemical constituents of different samples of *P. harmala* L.

No.	Formula	RT (min)	Compound	Classify	CAS	Relative peak area		
						S1	S2	S3
1	C <sub>6</sub> H <sub>8</sub> O	3.467	2-Ethyl-furan	Heterocyclic compound	3208-16-0	9.00E+00	7.58E+05	9.00E+00
2	C <sub>5</sub> H <sub>12</sub> O	4.140	1-pentol	Alcohols	71-41-0	3.69E+04	4.89E+04	2.01E+04
3	C7H12O	4.140	2,2,3-trimethylcyclobutanone	Ketone	1449-49-6	4.32E+04	4.53E+04	2.90E+04
4	C <sub>8</sub> H <sub>18</sub>	5.450	Octane	Alkanes	111-65-9	5.23E+04	1.40E+05	4.78E+06
5	C <sub>6</sub> H <sub>12</sub> O	6.687	(E)-3-hexadienol	Alcohols	928-97-2	3.35E+04	6.40E+05	3.30E+04
6	C <sub>8</sub> H <sub>10</sub>	6.765	Acetyrene	Aromatic hydrocarbon	100-41-4	2.71E+05	3.94E+05	4.04E+06
7	C <sub>8</sub> O <sub>10</sub>	6.860	Anti-dixene isotopes	Aromatic hydrocarbon	41051-88- 1	6.80E+05	7.18E+05	4.67E+05
8	C <sub>8</sub> H <sub>10</sub>	6.972	p-xylene	Aromatic hydrocarbon	106-42-3	1.18E+06	2.12E+06	1.52E+07
9	C7H14O	7.654	Enanthal	Aldehyde	111-71-7	2.42E+04	2.07E+04	1.90E+04
10	C <sub>10</sub> H <sub>14</sub>	8.723	4-Methylene-1- (1-methylethyl) -biring [3.1.0] self-2-ene	Terpene	36262-09- 6	6.38E+05	2.39E+04	4.04E+04
11	C7H6O	8.877	Artificial almond oil	Aldehyde	100-52-7	1.76E+05	2.13E+05	2.19E+05
12	C <sub>10</sub> H <sub>16</sub>	9.184	β-phellandrene	Terpene	555-10-2	3.21E+04	9.00E+00	3.14E+04
13	$C_9H_{14}O$	9.394	The pentofuran	Heterocyclic compound	3777-69-3	9.52E+04	1.59E+05	1.38E+05
14	C <sub>10</sub> H <sub>14</sub>	10.052	1-Methyl-3- (1-methylethyl) -benzene	Aromatic hydrocarbon	535-77-3	4.61E+05	4.46E+04	7.91E+04
15	C <sub>10</sub> H <sub>16</sub>	10.141	(+)-cajuputene	Terpene	5989-27-5	5.41E+04	1.04E+04	1.13E+05
16	C <sub>8</sub> H <sub>8</sub> O	10.400	Phenylacetaldehyde	Aldehyde	122-78-1	4.53E+05	2.95E+05	1.66E+05
17	C <sub>10</sub> H <sub>16</sub>	11.191	2,6-dimethyl-2,4,6-octotriene	Terpene	673-84-7	5.47E+04	1.49E+04	3.83E+03
18	C <sub>10</sub> H <sub>12</sub>	11.277	1-methyl-3- (1-methyl vinyl) -benzene	Aromatic hydrocarbon	1124-20-5	1.61E+05	3.14E+04	8.87E+03
19	C <sub>9</sub> H <sub>18</sub> O	11.510	n-nonaldehyde	Aldehyde	124-19-6	4.93E+04	7.43E+04	2.38E+04
20	C <sub>8</sub> H <sub>16</sub> O	11.646	2,6-Dimethyl-cyclohexanol	Alcohols	5337-72-4	1.93E+04	1.08E+05	4.34E+03
21	C <sub>8</sub> H <sub>10</sub> O	11.702	2-Phenyl ethanol	Alcohols	1960/12/8	8.31E+04	5.77E+04	5.71E+04
22	C9H18O2	11.846	Methyl caprylate	Ester	111-11-5	2.11E+04	8.04E+04	7.71E+04
23	C <sub>10</sub> H <sub>16</sub>	11.952	(E, Z)—2,6-dimethyl-2,4,6-octotriene	Alkanes	7216-56-0	4.96E+04	1.15E+05	1.44E+05
24	C <sub>10</sub> H <sub>16</sub> O	12.261	Trans-rosin phenols	Terpene	5947-36-4	9.60E+04	1.48E+03	9.00E+00
25	C <sub>10</sub> H <sub>16</sub> O	12.371	(+)-camphor	Terpene	464-49-3	3.13E+05	1.50E+04	4.98E+03
26	C <sub>10</sub> H <sub>16</sub> O	12.800	Right-menthol 1 (7), 2-dienol	Terpene	65293-09- 6	6.11E+04	9.00E+00	9.00E+00
27	C <sub>10</sub> H <sub>20</sub> O	12.922	Mentha camphor	Terpene	2216-51-5	7.39E+04	1.72E+05	9.86E+03
28	C <sub>10</sub> H <sub>18</sub> O	12.988	(R)–4-methyl-1- (1-methylethyl)–3-cyclohexene- 1-alcohol	Terpene	20126-76- 5	2.62E+05	2.35E+04	3.83E+03
29	C <sub>10</sub> H <sub>8</sub>	13.098	Naphthalene	Aromatic hydrocarbon	91-20-3	1.58E+06	1.08E+05	2.27E+04
30	C <sub>8</sub> H <sub>8</sub> O <sub>3</sub>	13.193	Menthyl salicylate	Ester	119-36-8	3.00E+06	6.73E+05	1.58E+04
31	C <sub>10</sub> H <sub>20</sub> O	13.403	Capraldehyde	Aldehyde	112-31-2	1.56E+04	2.42E+04	1.87E+03
32	C <sub>10</sub> H <sub>14</sub> O	13.491	The whip said ketone	Terpene	80-57-9	2.21E+05	6.35E+03	1.27E+03
33	C <sub>7</sub> H <sub>7</sub> NO	13.666	N-phenyl, -formamide	Aromatic hydrocarbon	103-70-8	4.21E+05	1.22E+05	9.42E+03
34	C <sub>7</sub> H <sub>9</sub> NO <sub>2</sub>	13.857	3-Ethyl-4-methyl-1H-pyrrol-2,5- diketone	Ketone	20189-42- 8	5.66E+04	1.86E+05	4.28E+03
35	$C_9H_7N$	14.084	Isoquinoline	Heterocyclic compound	119-65-3	3.78E+05	6.50E+05	1.68E+04
36	C10H12O	14.141	1- (4-ethylphenyl) -ethylketone	Ketone	937-30-4	1.49E+05	1.79E+03	9.00E+00
37	C9H18O2	14.472	n-nonanoic acid	Acid	112-05-0	5.77E+04	5.64E+04	9.93E+03
38	C <sub>13</sub> H <sub>28</sub>	14.685	3,8-Dimethyl-dodecane	Alkanes	17301-30-	1.35E+04	2.87E+04	1.39E+04
					3			

(Continued)

### TABLE 2 | Continued

No.	Formula	RT (min)	Compound	Classify	CAS	Relative peak area		
						S1	S2	S3
39	C <sub>10</sub> H <sub>14</sub> O	14.954	3-Methyl-4-isopropylphenol	Phenol	3228/2/2	2.71E+06	1.34E+05	9.26E+03
40	$C_8H_7N$	15.062	Interaminophen acetylene	Amine	54060-30- 9	4.56E+04	3.60E+04	1.86E+04
41	C <sub>10</sub> H <sub>14</sub> O	15.122	2-Methyl-5- (1-methylethyl) -phenol	Phenol	499-75-2	1.56E+06	1.74E+05	4.63E+03
42	C <sub>13</sub> H <sub>28</sub>	15.155	Tridecane	Alkanes	629-50-5	4.20E+04	6.10E+04	8.69E+04
43	$C_9H_{10}O_2$	15.355	1- (2-hydroxyl-5-methphenyl) -acetone	Ketone	1450-72-2	2.60E+05	3.15E+05	1.10E+05
44	$C_{11}H_{22}O_2$	15.550	Methyl caprate	Ester	110-42-9	4.34E+04	4.42E+04	8.18E+04
45	C <sub>12</sub> H <sub>18</sub> O	16.193	Diprivan	Phenol	2078-54-8	2.19E+06	1.37E+06	4.41E+04
46	$C_{12}H_{25}Br$	16.439	2-Brododecane	Halogenated hydrocarbon	13187-99- 0	1.27E+04	1.77E+04	1.95E+04
47	$C_{15}H_{24}$	16.539	Admidia-butene	Terpene	30021-74- 0	7.88E+03	1.83E+04	2.23E+05
48	C <sub>14</sub> H <sub>30</sub>	16.974	Tetradecane	Alkanes	629-59-4	1.14E+05	1.68E+05	1.24E+05
49	C <sub>12</sub> H <sub>24</sub> O	17.131	Lauraldehyde	Aldehyde	112-54-9	1.79E+04	1.14E+04	2.12E+04
50	C <sub>13</sub> H <sub>22</sub> O	17.798	Cis-leaf yl acetone	Ketone	3879-26-3	3.51E+05	6.94E+05	2.85E+04
51	C <sub>15</sub> H <sub>24</sub>	17.901	β-farnesene	Terpene	18794-84- 8	6.37E+04	4.01E+04	5.17E+04
52	C <sub>16</sub> H <sub>34</sub>	18.040	2,6,10-trimethyldodecane	Alkanes	3891-99-4	2.19E+04	1.75E+05	7.40E+03
53	C15H22	18.438	α-curcumin	Terpene	644-30-4	1.87E+06	6.39E+04	1.29E+04
54	C <sub>13</sub> H <sub>28</sub> O	18.597	Tridecyl alcohol	Alcohols	112-70-9	2.23E+04	3.52E+04	1.04E+04
55	C <sub>15</sub> H <sub>24</sub>	18.663	(-)-α-Cyperylene	Terpene	469-61-4	7.64E+04	1.19E+04	7.59E+03
56	C15H32	18.732	Pentadecane	Alkanes	629-62-9	5.45E+04	6.96E+04	4.31E+04
57	C <sub>15</sub> H <sub>24</sub> O	18.796	Butylated hydroxytoluene	Aromatic hydrocarbon	128-37-0	3.00E+05	9.60E+04	1.32E+05
58	C13H26O2	19.102	Methyl laurate	Ester	111-82-0	2.99E+05	4.06E+05	3.77E+05
59	C <sub>16</sub> H <sub>34</sub>	19.930	3-Methylpentane	Alkanes	2882-96-4	4.12E+04	4.57E+04	2.82E+04
60	C <sub>12</sub> H <sub>14</sub> O <sub>4</sub>	20.190	Diethyl phthalate	Ester	84-66-2	1.46E+05	3.19E+04	3.38E+03
61	C <sub>16</sub> H <sub>30</sub> O <sub>4</sub>	20.207	2,2,4-trimethyl-1,3-glutarol diisobutyrate	Ester	6846-50-0	1.13E+05	6.90E+04	1.03E+04
62	C14H28O2	20.282	Ethyl laurate	Ester	106-33-2	4.92E+04	8.49E+04	1.95E+04
63	C <sub>16</sub> H <sub>34</sub>	20.416	Hexadecane	Alkanes	544-76-3	1.57E+05	1.51E+05	9.56E+04
64	$C_{15}H_{18}O_2$	20.469	l cerofuran ketone	Terpene	20493-56- 5	3.46E+05	5.65E+03	9.00E+00
65	C <sub>15</sub> H <sub>26</sub> O	20.703	Deodar	Terpene	77-53-2	6.45E+04	1.19E+05	7.49E+03
66	$C_{19}H_{33}F_5O_2$	21.317	Pentafluoropropionlic acid, 4-hexadyl ester	Ester	1000283- 04-1	2.34E+04	2.70E+04	1.16E+04
67	$C_{12}H_{12}O_2$	21.546	Z-butadiene	Alkanes	72917-31- 8	1.10E+05	4.99E+04	9.00E+00
68	C <sub>17</sub> H <sub>36</sub>	21.841	Dioctylmethane	Alkanes	629-78-7	1.22E+05	5.54E+04	2.82E+04
69	$C_{15}H_{22}O_2$	21.942	Benzoic acid 2-ethylhexanide	Ester	5444-75-7	5.26E+04	5.05E+04	4.23E+03
70	C <sub>15</sub> H <sub>30</sub> O	22.000	Is the glutenal	Aldehyde	2765/11/9	8.03E+05	7.21E+04	7.38E+03
71	$C_{15}H_{30}O_2$	22.083	Methyl myristate	Ester	124-10-7	3.42E+05	1.44E+06	1.77E+06
72	$C_{14}H_{30}O_3S$	22.130	Nrenyl ulfite	Ester	1000309- 14-2	2.47E+04	4.09E+04	9.00E+00
73	$C_{12}H_{14}O_2$	22.205	Artemicen lactone	Ester	81944-09- 4	6.20E+04	2.43E+04	9.00E+00
74	C <sub>15</sub> H <sub>22</sub> O	22.330	(R)—5- (1,5-dimethyl-4- hexenenyl)—2-methyl-phenol	Phenol	30199-26- 9	3.24E+06	6.01E+04	4.76E+03
75	$C_{14}H_{28}O_2$	22.419	Texanic acid	Acid	544-63-8	1.27E+04	4.82E+05	3.29E+04
76	C <sub>16</sub> H <sub>32</sub> O <sub>2</sub>	22.712	Ethyl tetraheatanate	Ester	124-06-1	5.34E+04	2.93E+05	8.77E+04

(Continued)

No.	Formula	RT (min)	Compound	Classify	CAS	Re	lative peak ar	ea
						S1	S2	S3
77	C <sub>18</sub> H <sub>38</sub>	22.783	2,6,10-trimethyl-pentane	Alkanes	3892-00-0	8.21E+04	5.45E+04	3.42E+04
78	C <sub>15</sub> H <sub>28</sub> O <sub>2</sub>	22.837	2-Hydroxycyclohexanone	Ketone	4727-18-8	3.21E+04	6.22E+04	4.84E+04
79	C <sub>16</sub> H <sub>32</sub> O <sub>2</sub>	22.962	Methyl sevenate	Ester	7132-64-1	1.57E+05	1.34E+05	3.26E+05
80	$C_8H_{10}N_4O_2$	23.133	Caffeine	Heterocyclic compound	1958/8/2	4.59E+05	5.45E+05	3.99E+05
81	$C_{16}H_{22}O_4$	23.232	1,2-phenylic acid double (2-methylpropyl) ester	Ester	84-69-5	1.19E+07	1.43E+07	8.53E+05
82	C <sub>20</sub> H <sub>40</sub> O	23.361	3,7,11,15-Tetraethyl-2- dohexadecene-1-alcohol	Terpene	102608- 53-7	1.47E+05	2.26E+05	1.38E+04
83	C <sub>18</sub> H <sub>32</sub> O <sub>2</sub>	23.432	Linoleic acid	Acid	506-21-8	9.78E+05	4.12E+05	1.14E+05
84	C <sub>17</sub> H <sub>32</sub> O <sub>2</sub>	23.516	(Z)–9-methyl dohexaenoacrylate	Ester	1120-25-8	6.90E+04	2.68E+05	8.00E+05
85	$C_{17}H_{34}O_2$	23.688	Methyl palmitate	Ester	112-39-0	4.02E+06	9.42E+06	2.61E+07
86	$C_{19}H_{28}O_4$	23.845	Butane-4-butylester phthalate	Ester	1000356- 78-4	1.28E+06	1.66E+06	8.64E+04
87	C <sub>18</sub> H <sub>36</sub> O <sub>2</sub>	24.035	Acetate hexadecanate	Ester	628-97-7	5.53E+05	2.00E+06	2.89E+06
88	$C_{19}H_{36}O_2$	24.084	(Z) -Ethyl hexadienylate	Ester	1089325- 29-0	4.85E+04	6.77E+04	1.20E+05
89	C <sub>18</sub> H <sub>36</sub> O <sub>2</sub>	24.203	Xamyl hexaecate	Ester	6929-04-0	7.97E+04	1.86E+05	3.10E+05
90	C <sub>19</sub> H <sub>36</sub> O <sub>2</sub>	24.659	(Z)–9-methyl octacaracene	Ester	112-62-9	3.87E+05	4.67E+05	4.52E+06
91	C <sub>19</sub> H <sub>38</sub> O <sub>2</sub>	24.726	Methyl octaenate	Ester	112-61-8	3.26E+05	6.01E+05	3.29E+06
92	C <sub>18</sub> H <sub>34</sub> O <sub>2</sub>	24.781	Cis oxknee acid	Acid	506-17-2	9.00E+00	5.09E+05	9.00E+00
93	C <sub>20</sub> H <sub>38</sub> O <sub>2</sub>	25.439	Cis-11-eicosaric acid	Acid	5561-99-9	7.32E+03	1.46E+04	1.92E+05
94	$C_{19}H_{34}O_2$	26.159	(Z, Z) 9,12-18 endocardienylate-methyl ester	Ester	112-63-0	9.00E+00	5.52E+03	1.16E+04

Compound **3** was obtained as white amorphous powder. <sup>1</sup>H-NMR (400 MHz CDCl<sub>3</sub>)  $\delta$ ppm: 11.42 (1H, s, NH), 8.16 (1H, d, *J* = 5.2 Hz), 7.80 (1H, d, *J* = 5.2 Hz), 8.05 (1H, d, *J* = 8.8 Hz), 6.84 (1H, dd, *J* = 8.8, 2.0 Hz), 7.01 (1H, d, *J* = 2.0 Hz), 3.87 (3H, s, -OCH<sub>3</sub>), and 2.72 (3H, s, -CH<sub>3</sub>); <sup>13</sup>C-NMR (100 MHz CdCl<sub>3</sub>)  $\delta$  ppm: 160.0 (C-1), 137.7 (C-3), 111.9 (C-4), 122.6 (C-5), 109.0 (C-6), 141.9 (C-7), 94.6(C-8), 141.2 (C-10), 127.2 (C-11), 114.8 (C-12), 134.5 (C-13), 55.3 (OCH<sub>3</sub>), and 20.3 (CH<sub>3</sub>). Therefore, the structure of compound **3** was determined as vasicine by comparing the NMR data with the literature (Wang et al., 2015).

Compound **4** was obtained as a white amorphous powder. <sup>1</sup>H-NMR (400 MHz DMSO- $d_6$ ) &ppm: 7.78 (1H, dd, J = 8.2, 1.1 Hz), 7.84 (1H, m), 7.53 (1H, m), 8. 16 (1H, dd, J = 8.1, 1.5 Hz), 4.99 (1H, t, J = 4.0 Hz), 4.11 (1H, m), 3.89 (1H, m), and 1.98 (2H, m); <sup>13</sup>C-NMR (100 MHz DMSO- $d_6$ ) &ppm: 159.9 (C-2), 160.6 (C-4), 134.2 (C-5), 120.6 (C-5'), 125.8 (C-6), 127.0 (C-7), 126.35 (C-8), 149.1 (C-8'), 71.3 (C-9), 29.6 (C-10), and 43.0 (C-11). Therefore, the structure of compound **4** was determined as vasicinone by comparing the NMR data with the literature (Wang et al., 2015). Their structures are shown in **Figure 1**.

# HPLC Method Validation

#### Precision

The sample (No. S1) was injected six times according to the chromatographic conditions mentioned in the section HPLC

analysis. The RSD values of the relative retention time of vasicine, vasicinone, harmaline, and harmine were 0.26, 0.13, 0.33, and 0.35%, respectively; the RSD of relative peak area were 1.12, 1.34, 0.82, and 0.86%, respectively.

#### Stability

The sample (No. S1) was analyzed by the aforementioned chromatography conditions after 0, 2, 4, 6, 12, and 24 h of storage. The RSD value of the relative retention time of vasicine, vasicinone, harmaline, and harmine were 0.33, 0.16, 0.33, and 0.35%, respectively; The RSD values of the relative peak area were 1.03, 2.13, 0.90, and 1.02%, respectively.

#### Repeatability

The sample (No. S1) was extracted five times in parallel by the method mentioned in Section Preparation of Samples Solutions and measured by chromatography in the aforementioned Section HPLC Analysis. The RSD values of relative retention time of vasicine, vasicinone, harmaline, and harmine were 0.22, 0.09, 0.15, and 0.15%, respectively; the RSD of relative peak area were 0.15, 0.50, 0.22, and 0.30%, respectively.

#### Linear Relationship

According to the chromatographic conditions mentioned under Section HPLC Analysis, the different mass concentrations were the transverse coordinate (X) and the peak area as the vertical ordinate (Y), and the results are shown in **Table 1**.



## **Results of the HPLC Analysis**

The sample solution was prepared according to Section HPLC Chromatographic Conditions and analyzed according to the chromatographic conditions of Section HPLC Chromatographic Conditions. The chromatogram of the standard compounds is shown in **Figure 2** and the HPLC chromatogram of batch 10 samples is shown in **Figure 3**. The contents of each index component were calculated, and the results are shown in **Figure 4**. The contents of vasicine, vasicinone, harmaline, and harmine in seeds were 21.36–35.27  $\mu$ g·g<sup>-1</sup>, 5.18–13.66  $\mu$ g·g<sup>-1</sup>, 88.05–472.23  $\mu$ g·g<sup>-1</sup>, and 136.41–650.12  $\mu$ g·g<sup>-1</sup>, respectively. The contents of vasicine and vasicinone in the aerial parts were 12.99–60.46  $\mu$ g·g<sup>-1</sup> and 2.15–8.18  $\mu$ g·g<sup>-1</sup>, respectively, and the harmaline and harmine were not detected in the aerial parts.

## **Results of the HS-SPME-GC-MS Analysis**

Based on the NIST database, the metabolites of the samples were qualitatively quantified by mass spectrometry, and the compounds with a matching degree above 80 points were selected. The mass spectrometry files were opened with MassHunter quantitative software for the integration and correction of chromatographic peaks. A total of 94 VOCs were detected by HS-SPME-GC-MS, and the metabolite numbers, values, and corresponding metabolites names of some metabolites detected in this experiment are shown in Table 2. These consisted of five alcohols, eight aromatic hydrocarbons, seven aldehydes, five acids, 17 terpenoids, 12 hydrocarbons, and 24 esters. The clustering heat map of all metabolites by classification and content is shown in Figure 5. The main volatile components in the whole herbs from Pakistan were 1,2-Benzenedicarboxylic acid, bis(2-methylpropyl) ester, Hexadecanoic acid, methyl ester, methyl salicylate, 3methyl-4-isopropyl phenol, and propofol. The main volatile components in the whole herbs from Gansu were bis (2methylpropyl) 1, 2-phthalate, methyl hexadecate, (Z, Z)-1,8,11, 14-heptadiene, p-xylene, ethyl hexadecanoate, butyl heptadiate-4-butyl phthalate, methyl tetradecanoate, and propofol. The main volatile components in the seed samples from Xinjiang were methyl hexadecanoate, p-xylene, octane, (Z)-9-octadecanoate, ethylbenzene, methyl octadecanoate, ethyl hexadecanoate, and methyl tetradecanoate (Table 2). Four alkaloids, caffeine, pentyl furan, isoquinoline, and 2-ethyl-furan were also detected by GC-MS.

Sample	Concentration ( $\mu$ g·mL <sup>-1</sup> )	Average inhibition rate $\pm$ SD (%) ( $n = 3$ )						
		E. coli	S. aureus	X. oryzae	P. atroseptica	X. axonopodis		
E1	800	99.46 ± 0.11**	76.95 ± 3.38**	92.61 ± 0.25**	$47.34 \pm 4.28^{*}$	100 ± 0.00**		
	400	$92.62 \pm 2.00^{**}$	$15.80\pm3.25$	$88.14 \pm 2.29^{**}$	-	$97.74 \pm 0.97^{**}$		
	200	$21.24 \pm 2.47$	-	64.07 ± 1.21**	-	$89.68 \pm 1.81^{**}$		
	100	$5.50 \pm 2.58$	_	$37.56 \pm 3.62$	-	$67.42 \pm 0.00^{**}$		
	50	-	_	$15.84 \pm 2.90$	-	$67.06 \pm 0.97^{**}$		
E2	800	-	$46.68 \pm 1.07^{*}$	$48.05 \pm 4.33^{*}$	$32.15\pm1.58$	$27.86\pm1.47$		
E3	800	-	-	$59.51 \pm 4.84^{*}$	$31.25\pm8.21$	-		
E4	800	-	-	$68.68 \pm 3.57^{*}$	$24.49 \pm 2.81$	-		
C1	100	$50.11 \pm 2.50^{*}$	$13.92\pm0.98$	$2.65\pm0.88$	$27.44\pm0.61$	-		
	50	$32.01 \pm 1.18$	$7.14 \pm 1.10$	$2.65 \pm 1.77$	$3.66 \pm 1.83$	-		
C2	100	$90.68 \pm 1.18^{**}$	$12.64 \pm 1.47$	$0.44 \pm 0.44$	77.74 ± 2.13**	-		
	50	$61.37 \pm 1.47^{**}$	$9.16\pm0.61$	-	$40.55 \pm 1.52^{*}$	-		
	25	$38.59 \pm 0.86^{*}$	-	-	-	-		
	12.5	$9.93 \pm 1.33$	-	-	-	-		
C3	100	$32.26 \pm 0.18^{*}$	$15.32\pm2.56$	-	$19.50\pm0.00$	-		
	50	$12.34\pm1.32$	$7.51 \pm 1.59$	-	-	-		
C4	100	$25.34\pm0.37$	-	$1.33\pm0.44$	$22.13\pm0.58$	-		
	50	$5.29 \pm 1.25$	-	_	-	-		

E1, The methanol extract of seed; E2, The petroleum ether extract of seed; E3, The methanol extract of the aerial parts; E4, The petroleum ether extract of the aerial parts; C1, Harmine; C2, Harmaline; C3, Vasicine; C4, Vasicinone. \*P < 0.05, \*\*P < 0.01 compared with the control group.

TABLE 4 | Antifungal activity of extracts and alkaloids from P. harmala L.

Sample	Average inhibition rate $\pm$ SD (%) ( $n = 3$ )									
	R. solani	S. sclerotiorum	F. graminearum	M. oryzae	P. racapsici	C. albicans	B. cinerea			
E1	50.33 ± 0.11**	64.32 ± 0.98**	52.98 ± 0.89**	42.57 ± 0.9**	55.58 ± 0.44**	47.47 ± 1.18*	76.32 ± 0.88**			
E2	$10.88\pm0.16$	-	-	-	$13.57\pm0.64$	$25.32\pm2.16$	-			
E3	_	-	$10.89 \pm 1.21$	-	$10.70\pm0.49$	$20.08\pm4.33$	-			
E4	-	-	$12.08\pm0.34$	-	$16.57\pm1.78$	$15.07 \pm 2.65$	-			
C1	$21.89 \pm 1.33$	$18.67 \pm 1.02$	$15.23\pm0.54$	$21.93\pm1.67$	$20.37\pm1.56$	-	-			
C2	$13.32\pm2.16$	$23.45\pm2.32$	$24.12\pm1.48$	$15.23 \pm 1.15$	$29.05 \pm 2.27^{*}$	-	-			
C3	$16.12 \pm 1.78$	$19.87\pm0.59$	$19.26\pm0.68$	$16.38\pm2.56$	$13.56\pm1.33$	-	-			
C4	$24.36 \pm 1.36$	$15.38 \pm 1.38$	$13.63 \pm 1.12$	$10.81 \pm 0.26$	$14.80 \pm 0.68$	$7.15 \pm 0.56$	-			

E1, The methanol extract of seed; E2, The petroleum ether extract of seed; E3, The methanol extract of the aerial parts; E4, The petroleum ether extract of the aerial parts; C1, Harmine; C2, Harmaline; C3, Vasicine; C4, Vasicinone. \*P < 0.05, \*\*P < 0.01 compared with the control group.

# **Antimicrobial Activity**

#### **Results of Antibacterial Activity**

The petroleum ether extract and methanol extract of *P. harmala* showed broad-spectrum antibacterial activity against all the tested bacteria, as shown in **Table 3**. At the concentration of 800  $\mu$ g·ml<sup>-1</sup>, the methanol extract of seeds displayed the highest bioactivity against *E. coli* ATCC 25922 with an inhibition rate of more than 90% and the petroleum ether extract of seeds showed slight inhibition against *Staphylococcus aureus*. The methanol and petroleum ether extracts of the aerial parts were only slightly against *Candida albicans* ATCC 24433 with inhibition rates of 20.08 and 15.07%, respectively.

The methanol extract of seeds displayed the highest bioactivity against *X. oryzae*, and the inhibition rate was 92.61% at 800  $\mu$ g·ml<sup>-1</sup>. The petroleum ether extract of seeds and the aerial parts and the methanol extract of seeds displayed moderate bioactivity against *X. oryzae*, and the inhibition rates were from 48.05 to 68.68% at 800  $\mu$ g·ml<sup>-1</sup>.

Citrus canker is the most serious Citrus bacterial disease caused by the gram-positive *X. axonopodis;* it can affect plant growth, fruit quality, and yield, which caused great economic losses to the citrus industry (Sun et al., 2019). In the present study, the methanol extract of seeds was effective against *X. axonopodis*, the inhibition rates were 67.42 and 100% at 50 and

800  $\mu$ g·ml<sup>-1</sup>, respectively. The petroleum ether extract of seeds was slightly against *X. axonopodis* with an inhibition rate of 27.86% at 800  $\mu$ g·ml<sup>-1</sup>.

As shown in **Table 3**, methanol extract of seeds exhibited moderate antimicrobial activity against *P. atroseptica* ACCC 19901, and the inhibition rate was 47.34% at 800  $\mu$ g·ml<sup>-1</sup>. Petroleum ether extract of seeds and the aerial parts and methanol extract of seeds also exhibited slight antimicrobial activity against *P. atroseptica* ACCC 19901, and the inhibition rates were 32.15, 31.25, and 24.49%, respectively.

#### Antifungal Activity of P. harmala Extracts

The antifungal activities of *P. harmala* extract against pathogenic fungi are shown in Table 4. At the concentration of 500 µg·ml<sup>-1</sup>, the methanol extract of seeds effectively inhibited seven fungi of Rhizoctonia solani, Sclerotinia sclerotiorum, Botrytis cinerea, Fusarium graminearum, Magnaporthe oryzae, Phytophthora capsici, and Candida albicans ATCC 24433, with inhibition rates of 50.33, 64.32, 76.32, 52.98, 42.57, 55.58, and 47.47%, respectively. The petroleum ether extract of seeds weakly inhibited Rhizoctonia solani and Magnaporthe oryzae, with inhibition rates of 10.88 and 10.88%, respectively. The methanol and parts petroleum extract of aerial parts only weakly suppressed Fusarium graminearum and Magnaporthe oryzae, with inhibition rates of 10.89% and 10.7%, and 12.08% and 16.57%, respectively. The methanol extracts of seeds showed stronger antifungal activities with a wide antifungal spectrum. Our results showed that the aerial parts and seed extracts have broad-spectrum antibacterial activities against agricultural pathogenic bacteria, which may be related to its alkaloids (Adnan et al., 2019; Zhang et al., 2020), but its specific active components and its antibacterial mechanism need to be further studied.

Furthermore, we evaluated the antimicrobial activity of four alkaloids isolated from *P. harmala*, including *Rhizoctonia* solani, *Sclerotinia sclerotiorum*, *Magnaporthe oryzae*, *Fusarium* graminearum, and *Phytophthora capsici*, and found that the growth of all five strains was inhibited in the presence of alkaloids (**Tables 3**, **4**), while harmine and harmaline showed broad-spectrum antibacterial activities against *X. oryzae* (ACCC 11602), *E. coli* ATCC 25922, and *Staphylococcus aureus* (Newman). These findings indicated that alkaloids could account for antimicrobial activity. Furthermore, harmine and harmaline showed better antibacterial activities, which could be used as lead molecules in the development of new antimicrobial drugs. In the future study, we are going to select harmine and harmaline as lead compounds, and a series of derivatives are supposed to

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design and synthesize to evaluate their antifungal activities and preliminary antifungal mechanism.

# CONCLUSION

In conclusion, the aerial parts and seeds of *Peganum harmala* L showed a large difference in the content of four alkaloids. Harmine and harmaline were mainly found in seeds, while the aerial parts mainly contained vasicine and vasicinone. A total of 94 metabolites were detected by HS-SPME-GC-MS technology for the first time. The VOCs of *P. harmala* mainly consisted of hydrocarbons, acids, ketones, aromatic hydrocarbons, esters, and heterocyclic compounds. In addition, the extracts of the aerial parts and seeds showed antibacterial activities against 12 bacterial strains and fungi. This study demonstrates that the alkaloids from *P. harmala* are a potentially non-toxic and ecofriendly botanical fungicide for the management of fungi.

# DATA AVAILABILITY STATEMENT

The original contributions presented in the study are included in the article/**Supplementary Material**, further inquiries can be directed to the corresponding author/s.

# **AUTHOR CONTRIBUTIONS**

Methodology, conceptualization, investigation, and writing original draft preparation: NW. Supervision, writing—review and editing, and project administration: ZY. Investigation and formal analysis: JA and ZZ. Writing—review and visualization: YL and JF. All authors have read and agreed to the published version of the manuscript.

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## SUPPLEMENTARY MATERIAL

The Supplementary Material for this article can be found online at: https://www.frontiersin.org/articles/10.3389/fmicb. 2022.916371/full#supplementary-material

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