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Metal-organic framework as nanocarriers for agricultural applications: a review

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Agrochemicals are substances used to promote plant growth and eradicate pests; and are essential to meeting the world food demand both qualitatively and quantitatively. Unfortunately, their extensive usage has resulted in certain significant problems, i.e., soil contamination, bioaccumulation, and the development of pest resistance. Consequently, there is a need for a reduced and more regulated usage of agrochemicals. The development of sustainable nanomaterials is encouraged in agricultural nanotechnology to minimize the usage of conventional fertilizers, herbicides, and pesticides. Metal-organic Framework (MOF)s are porous materials made of functional organic ligands and metal nodes or clusters connected by covalent bonds. Researchers are becoming increasingly interested in using MOF in a variety of sectors. The features of engineered MOFs have the potential to be an optimal strategy for producing impressively unique results in agricultural research. They also have advantages in terms of biocompatibility and biodegradability, which can be leveraged to boost the efficiency of traditional agrochemicals administration. Controlled release of agrochemicals offers an innovative solution to decrease the nutrient uptake rates and contamination of the environment. The application of MOF nanoparticles as nanocarriers to modify the loading and release properties of agrochemicals is particularly highlighted in this review.

KEYWORDS

sustainable agriculture, MOF, nano-carriers, insecticide, control release, slow release, agrochemicals

1 Introduction

One of the seventeen sustainable development goals set forward by the United Nations (UN) is "Zero hunger," and achieving this goal will require sustainable agriculture. The global food supply and distribution system is under extreme stress because of growing environmental pollution, population growth, climate change, and elevated water and energy requirements. Annual agricultural production worldwide nowadays utilizes an astounding amount of natural and industrial resources. The yearly requirement of agrochemicals all over the world is approximately 234.2 billion US dollars (USD) in 2019 and this value is expected to rise to more than 300 billion USD by 2025 (Devi et al., 2022). The global food production systems must undergo significant changes in response to this circumstance. Fertilizers and pesticides have become essential components of modern agricultural systems



to meet the enormous demand for food. Pesticides can be categorized according to several factors, including toxicity, target, chemical formula and composition, mode of entry or action, and source. However, between 10% and 75% of pesticides fail to reach their intended targets due to improper application techniques, and frequently contaminate both terrestrial and aquatic environments (Rojas Macías et al., 2022). It also promotes insect and disease resistance, reduces soil biodiversity, and increases the bioaccumulation of pesticides in the food. Even with the stringent governmental legislation, chemicals still find their way into the food chain. Concerns related to undeveloped countries are more severe (Rojas Macías et al., 2022). The impacts of conventional agrochemicals can be felt in the soil, where they are released by a variety of exposures. For instance, following application, the loss of fertilisers and pesticides results in significant environmental loss and higher cultivation expenses. This results in the overuse and frequent application of agrochemicals, which over time may contaminate plants, soil, and water. Despite these negative side effects, their use is necessary to increase agricultural output to fulfil the food demand of the growing population. The losses that occur during application can be attributed to several factors such as the weather, the formulations utilised, and the application techniques employed (Pretty and Pervez Bharucha, 2015).

A popular strategy for attaining regional and global sustainability is the circular economy (CE). As a result, certain behaviours and activities including eco-design, reuse, refurbishing, and remanufacturing define CE. To meet the 2030 Agenda for Sustainable Development objectives of combating climate change (SDG11), ending poverty and hunger (SDGs 1 and 2), and maintaining natural resources, food and agriculture play a critical role. This calls for radical adjustments in the direction of sustainability in agriculture as well as more sustainable usage and management of the current agricultural systems to maintain food production and the ecosystems (Nhemachena et al. 2018; Agrawal et al. 2023). Recent studies have demonstrated the promising

potential of nanotechnology to boost the efficiency of agricultural inputs, provide solutions for environmental and agricultural issues and enhance food security and productivity (Usman et al., 2020). "Smart" nanocarriers need to exhibit new and distinctive qualities in the delivery of agrochemicals to achieve improved agrochemical efficiency. Recent developments indicate that the extraordinary characteristics of nanoparticles can be used for this purpose (Goenka et al., 2014; Hsu et al., 2023). The term "precision farming," which aims to produce optimum crops without depleting water and soil resources, uses various carriers to carry and deliver agrochemicals in a more regulated and slow-release profile (Dutta et al., 2022).

After entering the intricate plant-soil system, these modern nanomaterials can change the physiology of plants, which may be easily used to comprehend the impacts. Furthermore, a fundamental understanding of these nanomaterials' positive or negative interactions with plants is necessary for the regulated delivery of active chemicals. Furthermore, it is thought that the concentration of nanomaterial in the natural environment is far lower than those deemed dangerous (Batley et al., 2013). High-level research is being done to create new delivery systems with nanoscale materials. It has been observed that when used as an agrochemical carrier, nanoparticles exhibit greater permeability, stiffness, thermal stability, and biodegradability than traditional agrochemicals (Nair et al., 2010; Deka et al., 2021). "Smart" nanocarriers provide specific formulation procedures that allow for a regulated release over an extended period of exposure and greatly prevent the early loss of the active components. One benefit of this strategy is that it uses fewer chemicals and requires less manual application (Fincheira et al., 2023). The active ingredients can be evenly distributed across the soil or plant surface due to the enormous surface area of nanoscale particles.

Currently, various nanocarriers for biological applications are being evaluated. They include the "soft" nanoparticles (NPs)- having an organic backbone (Shakiba et al., 2020), "hard" NPs-synthesized



from metallic elements (Li et al., 2007; Chandrakala et al., 2022), nanoclays (bin Hussein et al., 2005; Merino et al., 2020), carbon materials (Sarlak et al., 2014; Zare et al., 2021) and so on. However, the most significant issues present in these nanocarriers are longterm instability, initial burst of material entrapped, and bio-toxicity (Petrovic and Barbinta-Patrascu, 2023). These issues need to be addressed soon for sustainability in agriculture.

Metal-organic frameworks (MOFs) are a unique class of porous material that is formed due to the self-assembly of infinite inorganic lattices (metal ions or clusters) and organic ligands via coordination bonds. They comprise of organic linkers, such as carboxylates, nitrogenated, or phosphonates, and inorganic nodes, such as atoms, chains, or clusters (Allendorf et al., 2021). The orientation of the metal and any associated linkers define its final structure. MOFs have been suggested for a wide range of industrial and social applications, including drug delivery (Anstoetz et al., 2016), energy (Srinivasan et al., 2020), adsorption (Ma et al., 2016; Bhattacharyya et al., 2023), separation (Qian et al., 2020), magnetism (Muratović et al., 2020), luminescence (Liu Yang et al., 2019c), conductivity (Su et al., 2020), electrochemical sensing (Zheng et al., 2018; Mohan et al., 2023c), catalysis (Chen et al., 2020a), heavy metal removal (Mohan et al. 2023a) and so many. We are totally dependent on agriculture and food industries for producing and processing food. Food products have been contaminated due to the growing use of hazardous chemicals in agriculture and packaging processes. MOFs in particular demonstrate great potential as agricultural materials because of their intriguing properties such as large specific surface areas and pore volumes, exceptional sorption capacities, large-scale synthesis and stability to carry out their function, preventing associated toxicity in animals/plants due to their accumulation after decomposition (Bazer-Bachi et al., 2014). MOFs usually have a higher porosity level and a more defined and adjustable pore size distribution when compared to other traditional porous materials. Furthermore, a significant deal of flexibility in the synthesis of MOFs with varying pore size, configuration, morphology, and chemical and physical properties is made possible by the many options for metal nodes and organic linkers, as well as the customizable synthesis parameters. These unique qualities have drawn a lot of interest in MOFs and shown a wide range of potential applications. From this perspective, metalorganic frameworks (MOFs) seem to be a more sustainable choice. In addition to bioavailability, the surface deposition of polar molecules can alter the water solubility of MOFs by improving their distribution in soil media. Furthermore, MOF provides stability by avoiding chemical modification of the encapsulated compounds in the empty spaces of the MOF architecture (Nadar and Rathod, 2018).

To increase agrochemicals' effectiveness and reduce their detrimental effects on the environment, their release must be slow and regulated from the formulation. Several studies on MOF-based carriers have previously shown how they can improve the effectiveness of nitrogen fertilizers. It is also possible to supplement the micronutrient for a specific crop by changing the metal centres of MOFs (Shan et al., 2020a). In this review, we outlined a brief discussion on the synthesis approaches and attributes of a MOF structure. We have also emphasized the application of MOFs and MOF derivatives in agriculture and highlighted different previously published MOF-based materials for their targeted delivery and sustained release of agrochemicals. Finally, we concluded by highlighting the prospects of MOF-based materials in agricultural applications.

2 Overview of MOF structures

MOFs are renowned for their wide variety of topologies and functionalities. When first considering molecular metal-organic hybrids, it is important to keep in mind that several factors, including design, binding affinity, and chelation of linker, open metal sites on the clusters, ionic strength between metal and organic component' and the symmetry of the moiety, influence the final structural design of the framework (Figure 1). For instance, a welldetailed description of an Ag-MOF framework with all the structural details was described (Figure 2) in a report by (Mohan et al. 2023b). The MOF possessed a monoclinic cc space group structure. Moreover, the unit consisted of a ligand, metal centre Ag (I), and one water molecule lattice. The nitrogen atoms in the pyridine and triazole groups of the organic ligand bond to the metal. Because of this, the core metal in Ag-MOFs has a distorted triangle geometry utilizing one pyridine and two triazole nitrogen from separate ligands.

When taken as a whole, these intricate elements have a big impact on the ultimate structural design, use, and application of a MOF (Kirchon et al., 2018). MOFs are known for having pore sizes



that may be adjusted, incredibly large surface areas, and flexible internal surface characteristics. These characteristics can be regulated by changing the combination of inorganic and organic components. There are countless options when the metal ion or cluster and the organic linker are combined to form a MOF. Interesting characteristics for an MOF are provided by the combination of the inorganic and organic components' physical attributes as well as any potential synergistic interactions between them. Dimensionality is a major factor that may be used to categorize MOFs and determine how intrinsically functional they are. For example, 3D-MOFs typically have substantial pore volumes and are frequently utilized for medicine delivery, gas storage, and water harvesting. Yet, 2D-MOFs can behave as gas separation membranes since they are created by van der Waals interactions. As a result, 2D-MOFs are advantageous in the fields of chemical sensing, supercapacitors, and gas separation. Thus, we can facilitate a new set of applications and properties by designing multidimensional MOFs (Dutta et al. 2021).

The precisely defined and adjustable crystalline structures of MOF provide them with a strong basis for the development of distinct structure–property relationships for making novel functional materials. MOFs are formed by combining metals or metal clusters, also known as secondary building units (SBUs), with organic linkers, primarily ligands that include nitrogen or carboxylic acid. It is possible to customise MOFs to some extent as anticipated by carefully choosing SBUs and linkers since the geometry of SBUs and the dimension and shape of the organic ligands determine the structures of MOFs. Yaghi and Li (1995) demonstrated that by methodically elongating the ligand structure, an isoreticular series of MOF-74 structures (IRMOF-74-n)

can be produced. However, different variations of IRMOF can be possible with different porous structures. Theoretically, it is possible to anticipate the MOF structures by considering the metals and organic linker building blocks. MOFs possess pore sizes of microporous to macroporous depending on their organic linker. MOFs with distinct electrical, magnetic, and optical properties can have their metal ions/ clusters and organic ligands precisely adjusted to the needs of the application. Furthermore, different functional groups can be placed in the pore space of MOFs to serve a variety of applications. Carboxylates or other organic anions, such as phosphonate, sulfonate, and heterocyclic compounds, are the organic units (linkers/bridging ligands). MOFs with preconfigured structural topologies can be built because of the orientation of the organic linkers. Zinc nitrate [Zn(NO₃)₂] and terephthalate (bdc) for instance react to form crystalline MOF-5 under solvothermal conditions. On the other hand, slow vapour diffusion of 10 triethylamine/toluene at room temperature into an N,N'-dimethylformamide (DMF)/toluene solution containing a mixture of Zn(NO₃)₂/H₂bdc results in colourless prism-shaped crystals of Zn(bdc)(DMF)(H₂O). Moreover, tetranuclear octahedral Zn₄O(CO₂)₆ SBU can be produced hydrothermally in basic zinc acetate. The ditopic carboxylate bdc, which is used in place of the acetate, has a stiff and divergent property that enables the clusters to articulate into a 3D framework with an attractively beautiful cubic shape (Lu et al., 2014). It has been claimed that a key factor in influencing the stability of MOF is the strength of the bond between the metal oxide cluster and the linker. It is discovered that the breakdown of MOF-5 is caused by the dissolution of carboxylic bridges between the benzene rings and Zn₄O clusters at 400°C or higher. On the contrary, MOF-177 has



low stability as it breaks down in water in 12 h and gradually changes from hexagonal to orthogonal phase and then to monoclinic phase in 5 weeks at room temperature (Kang et al., 2011). The relative stability of the MOFs depends on the metal ions which can be explained by the chemical inertness of the metal ions. Moreover, the bond strength between the metal cluster and the linker material has been used to explain the relative hydrothermal stability of MOFs. For example, CrIII is a well-known inert ion which explains the excellent stability of MIL-100 and MIL-101 in water (Haque et al., 2010; Kang et al., 2011). Another kind of fascinating 3D structure can be obtained by chelation between two bridged Ce³⁺ centres and eight carboxylate groups (Lu et al., 2014).

A new field of study in MOF is the manipulation of crystal size and form since this allows for the optimization of the physical characteristics for particular uses. Furthermore, thin films or membranes can be fabricated using nanoscale MOF crystals. The literature has documented a variety of techniques for creating MOF thin films or membranes. There are four different ways that growth can occur: direct, layer-by-layer, secondary, and chemical solution deposition (Stock and Biswas, 2012).

However, among different application areas of MOF, agriculture is an evolving one. The intriguing properties of MOFs make them promising materials for agriculture: i) large surface areas and pore volumes leading to exceptional sorption capabilities; ii) Different hybrid compositions; iii) Versatile cavities having specific host-guest interactions; iv) Proposition for large scale synthesis and commercialization v) Adequate stability profile before and after performing their function. These specifically prevent further toxicity in the environment due to accumulation.

3 Synthesis strategies of metal-organic frameworks

The remarkable physicochemical properties of metal-organic frameworks (MOFs) have attracted a lot of interest nowadays. These materials sometimes are called as porous coordination polymers (PCPs), that are built by connecting inorganic nodes with organic ligands. The term MOFs was first defined in the 1990s by Yaghi and Li (Yaghi and Li, 1995). Since then, there has been an accelerated expansion in the field of MOF research in recent decades. By merely adjusting the synthesis settings, the characteristics and structures of MOF can be easily tailored to suit the application that calls for them. These MOFs are two-component systems made up of organic binder and metal or metal clusters, also known as secondary building units (SBUs). During synthesis or afterwards, it is simple to add functional groups into the pores of MOF structures. Because of their typically large porosity and internal surface area, MOF-based materials are used for the majority of applications (Stock and Biswas, 2012). The most remarkable aspects of them are their tunable structures and functionalized pores, which are obtainable by combining metal ions and certain ligands with functional groups. Functionalized linkers with groups like, -NH2, -SH, -OH,-F and -COOH are introduced to create functionalized MOFs (Kumar et al. 2023). Different types of synthesis strategies are illustrated in Figure 3 and discussed below.

Researchers have historically employed two of the most common techniques to synthesise a wide range of MOF-based structures: solvothermal and room temperature synthesis. with Solvents higher boiling points like water, dimethylformamide (DMF), and dimethyl sulfoxide (DMSO) can be used in the solvothermal approach. The heating of the reaction mixture in the closed vessel causes the internal pressure of the container to rise proportionately, changing the structural characteristics of the MOF. The highest yield per batch of synthesis is guaranteed by carefully selecting the reactants, pH, temperature, and reaction duration, and then allowing the container to cool gradually. For instance, fumaric acid and ZrO Cl₂·8 H₂O were dissolved in a solvent mixture of DMF and of formic acid under constant stirring. The resulting solution was then heated at 130 °C for 10 h. The white product MOF-801 was filtrated, washed and dried overnight at ambient temperature (Aghajani Hashjin et al. 2023).

On the other hand, room temperature synthesis is accomplished in open flasks with air pressure (Tranchemontagne et al., 2008; McKinstry et al., 2016). Defect cracks can occur on a MOF's surface as a result of solvent evaporation by heat treatment in the open atmosphere. These cracks are crucial for the introduction of sizable meso and macropores without the drawbacks of microporosity. Avoiding the process of direct heat evaporation of a non-volatile solvent trapped in a MOF's micropores is common since it tends to marginally decrease the porosity overall. Yang and team (2014) described a template-free method for synthesizing porous hierarchical structures MOF-1 (IRMOF-1). This method features a template-free method for introducing large-sized meso- and macropores with a uniform distribution inside the MOF structure (Yang et al., 2014).

At the industry level, two distinct electrochemical approaches, namely, anodic deposition and cathodic deposition are commonly used for MOF synthesis. Electrochemical synthesis of MOF has

opened a versatile route for environment-friendly materials (Khazalpour et al., 2015). Mueller and the team were the first to report MOF synthesis via this technique (Mueller et al., 2006). The quantity of the product obtained in this process is one of the biggest advantages. The reaction time is also subsequently reduced in this method, where an electrode is immersed in an appropriate electrolyte and organic ligand solution. This allows the metal ions from the electrode to dissolve into the solution. There are two ways to perform MOF electrochemical synthesis: batch mode and continuous flow operation (Khazalpour et al., 2015). Being able to run continuous, high-frequency production of MOF is a distinct advantage of electrochemical synthesis as compared to conventional batch reactions (Mueller et al., 2006). Many of the risks associated with large-scale batch reactions, such as mass transfer, mixing, and heat runaway, are lessened in such continuous flow processes. Van Assche et al. (2016) published an important study on the electrosynthesis of HKUST-1 (Hong Kong University of Science and Technology -1) by anodic dissolution. A three-electrode electrochemical setup was used to synthesize HKUST-1. The working, counter, and reference electrodes in this configuration were Ag/AgCl, Pt foil, and Cu-coated silicon wafers, respectively. The electrodes were submerged in a solution of a solvent mixture of EtOH: H₂O and BTC (Benzotrichloride). The four stages of this method are initial nucleation, island development, intergrowth, and MOF crystal detachment (Varsha and Nageswaran 2020).

Sonochemical synthesis offers homogeneous nucleation and reduced crystallization time compared to oven heating. Acoustic cavitation, another name for sonochemical synthesis, is the process of forming a bubble, growing it, and then watching it burst and disintegrate in the liquid. As the bubble collapses it generates extremely high temperatures and pressures locally. This is achieved *via* the application of high-frequency ultrasound. Son and the team have reported the use of this technique to synthesize high-quality MOF-5 crystals with 8 min of sonication (Son et al., 2008). Lower pH values resulted in the formation of monodispersed particles, however yield of this process is lower than 5% which was a major drawback (Son et al., 2008).

Microwave synthesis of MOF is another popular method because of its shorter reaction time. Electrical and magnetic components of electromagnetic radiation travel simultaneously and are perpendicular to each other. Electromagnetic waves, or microwaves, have energies ranging from 300 to approximately 300,000 MHz. The wave's energy instantly heats the solvent, causing rapid crystallization, the production of nanoscale particles, and the elimination of impurities all at once. While the quality of the crystals produced by the microwave-aided approach is comparable to that of classic solvothermal methods, the procedure is significantly faster (Son et al., 2008). On the other hand, for industrial operations, this microwave synthesis method seems more appealing due to its shorter reaction times. Couzon et al. (2022) reported a novel synthesis method of MOF-textile composites UiO-66-NH₂ by a simple microwave-assisted method. Excellent fiber-like samples have been produced that are further employed for nuclear and radiological protection tests. Because the crystals nucleate over the majority of the solution as a result of the localized solvent heating, this method is very effective in creating particles with tiny dimensions (Couzon et al., 2022).

A solvent-free method of synthesis or mechanical synthesis of MOF has attracted much attention in recent times. The procedure of grinding the solid reactants is quicker, more convenient, and uses little or no solvent. Grinding is one of the mechanochemical techniques used to create coordination polymers with open porosity. For example, to synthesize $[Cu(INA)_2]$ [INA = Isonicotinic acid] porous frameworks, Pichon and the team used a ball mill to crush isonicotinic acid and copper acetate for a short period. The successful formation of the product was signified by a change in the colour of the sample from green to blue (Pichon et al., 2006).

Another potent method for creating sophisticated materials with complex shapes, hierarchical organization, and controlled size, shape, and polymorphism in aquatic settings at ambient temperatures is biomimetic mineralization. The potential of biomimetic mineralization techniques has significantly increased with the growing comprehension of biomineralization principles (Xu et al., 2007). This process uses biomolecules like proteins, lipids, and polysaccharides to help form biominerals. The resulting biominerals frequently have complex morphological features and superior material qualities that far outweigh those of their synthetic counterparts. Consequently, the development of specialized organic-inorganic hybrid materials has recently required the application of biomimetic techniques meant to imitate the processes involved in biomineralization (Liang et al., 2017). Typical processes that produce metal-organic frameworks (MOFs) often result in poor conductivity and poor water stability, which may hinder their use in electrochemical sensing applications. Using a biomimetic approach, one-dimensional Ce-MOF nanowires with ultrafine structures have been produced. The produced mixed-valence Ce-MOF nanowires were found to have strong conductivity and good water stability, with a typical size of about 50 nm (Ma et al., 2021).

Despite advancements in MOF synthesis techniques, there is still a strong need for more rapid and dependable methods to achieve controlled synthesis of the targeted molecules. Microfluidic synthesis is one of these methods; it allows the synthesis of monodisperse nanomaterials with specific characteristics because of its small liquid volumes, quick mass transfer, and great mixing efficiency. The ability to precisely modify the physico-chemical properties of nanomaterials through structural deviations achievable by adjusting the flow reagents in microchannels is a remarkable aspect of the microfluidic synthesis approach. Microfluidics has also been used to realize the synthesis of various MOFs through droplet or flow-focusing regimes. The synthesis of MOF-enzyme composites was also accomplished using a continuous-flow microfluidic synthesis system. It was shown that the induction of defects (mesopores) in a MOF structure was caused by a change in the MOFs reactants in the microchannels. Koryakina et al. (2023) presented a continuous flow regime microfluidic production of HKUST-1 microcrystals with interesting structural properties. To accomplish the necessary reagent mixing, the reagent flows via the microchannels in a microfluidic chip (MFC) have been optimized through numerical simulation. Moreover, additions (metallic salts) and changes in reaction temperature provide control over the shape of HKUST-1 (Koryakina et al., 2023).

There are some alternative synthesis methods also available for synthesizing MOFs. For instance, in the diffusion method of synthesis, chemical solutions are layered on top of one another, divided by a solvent layer, or dispersed gradually through physical barriers. Gels are utilized occasionally as diffusion and crystallization media. Following the progressive diffusion of the precipitate solvent into the distinct layer, crystals develop at the interface between the layers. For example, Chen and team developed three-dimensional Zn(II) complex Zn(INA)₂(H₂O)₄ (INA = isonicotinate) by a unique ammonia vapour diffusion method (Chen et al., 2016). Along with this, the Table 1 summarizes the advantages and disadvantages of different MOF synthesis methods.

The goal of MOF synthesis was initially to create novel chemicals and structures with intriguing properties, but the discipline has been expanding since then. The domains of mechano-, sono-, and electrochemical synthesis as well as microwave-assisted synthesis are emerging extensively. They have proven to be useful for a few chemicals, frequently with milder reaction conditions, producing materials with various particle sizes and characteristics. This may have implications for upscaling synthesis processes for different types of MOFs. These alternate methods have, however, mostly been used to synthesize popular MOF compounds, thus its reproducibility is still questionable. Therefore, systematic studies under well-defined conditions need to be performed before opting for a method for synthesizing a particular MOF.

4 Application of MOF in agrochemical delivery

MOFs, more specifically, nanoscale MOFs - have emerged as promising carriers for drug delivery and sustained release in recent years because of their superior encapsulating capabilities (He et al., 2021). In agriculture also MOFs are being used for the slow release of agrochemicals, and a few examples are discussed here.

4.1 Application of MOFs for sustained release of fertilizers

Anstoetz and team (2017) have reported on the use of oxalatephosphate-amine metal-organic frameworks (OPA-MOFs) as growth regulators and nutrients to boost grain yield in wheat plants. The result of their study suggested that the synthetic MOF could serve as a productive source of N fertilizer (Anstoetz et al., 2017). Abdelhameed and the team conducted a study that demonstrated the potential of Fe-MOF with ethylenediamine tetra-acetic acid (EDTA) as a source of iron nutrition in Phaseolus vulgaris. The study found that compared to other iron resources, there was a 9.6% improvement in plant weight, protein, enzyme content, and chlorophyll II content (Abdelhameed et al., 2019). The results of the study indicated that the mechanism of Fe-MOF-EDTA-based growth promotion is the result of an increase in the chlorophyll content that ultimately resulted in better photosynthesis and later in more absorption of the nutrients by plants to grow.

A new MOF was synthesized using ferric chloride (FeCl₃·6H₂O), phosphoric acid (H₃PO₄), oxalic acid

(H₂C₂O₄·2H₂O), urea (CO(NH₂)₂), and de-ionized water (H₂O) under mild hydrothermal conditions (Wu et al., 2019). After adding in the field, the chemical structures of MOF changed, and oxalate-consuming bacteria in the soil caused its degradation. This ultimately resulted in the release of N, P, and K from the elemental components (Wu et al., 2019). A later study reported a Fe-MOF on a pilot scale under hydrothermal reaction conditions, and its degradation behavior and nutrient release were analyzed in field conditions (Wu et al., 2022). The N, P, and Fe content in their MOF was around 6.03, 14.48, and 14.69%, respectively. When applied in rice fields it was found that the MOF has supported the plant growth and can be used as an innovative fertilizer with sustained release. This large-scale synthesized Fe-MOFs displayed faster nutrient release when added in soil, which may be due to the consumption of oxalate by native oxalotrophic bacteria (Figure 4) (Wu et al., 2022). In an earlier study, Fe based MOFs were synthesized by a solvent-free mechanochemical method and used as slow release fertilizer (Du et al. 2021). The MOF were affirmed as oxalate phosphate amine frameworks (OPA-MOF), and its elemental composition was as- N (4.91%), P (15.71%), and Fe (18.60%). The potential application of these MOFs in agriculture as a novel slow-release fertilizer was indicated by the slow release rate of N, P, and K in soil (Du et al., 2021).

Liu and team documented a multipurpose porous carbon by employing β -cyclodextrin MOF (β -CD MOF) as a potassium source and precursor (Liu et al., 2019a). Comprehensive characterization results indicated that the produced carbon material (β -CD MOF-NPC) has abundant potassium element content, large surface areas, and a microporous structure. The results of the batch adsorption experiments showed that β -CD MOF-NPC could effectively remove four amide herbicides from water by adsorptive means, possibly by hydrogen bonding, electrostatic contact, or π - π interaction (Figure 5). Furthermore, by simultaneously adsorbing herbicides and releasing potassium ions in the pot trials, the β -CD MOF-NPC could efficiently support the growth of rice seedlings cultivated in wastewater contaminated with herbicides. The potassium released from MOF increased the chlorophyll content of rice leaves. So the results indicated that β -CD MOF-NPC combined the functions of fertilizer and adsorbent, making it a potential substance for costeffective and sustainable agriculture (Liu et al., 2019a).

In another study, an easy technique of ion cross-linking was used to create a novel MOF(Fe)@NaAlg aerogels composite, which was then characterized using different techniques (Wu et al., 2020). The ammonium (NH⁴⁺)-loaded MOF(Fe)@NaAlg aerogels were ready as the slow-release fertilizer (SRF). MOF(Fe)@NaAlg (2:10) has an adsorption capacity of 29.4 mg/g and a swelling of 73 g/g, respectively. They looked at the swelling behaviors of MOF(Fe) @NaAlg(2:10) in various saline solutions. The produced formulation's slow-release property was validated by the release assessments, indicating that MOF(Fe) plays an effective function in it. Additionally, the produced fertilizer compound showed outstanding soil water-retention capabilities; therefore have a wide range of possible uses in the agricultural sector (Wu et al., 2020). These earlier studies point to MOF's possible use as fertilizers carrier in agriculture to achieve sustainability.

An innovative method was utilized to synthesize ureatemplated Fe- OPA-MOF(I) / (II). The components P, N, and





Schematic diagram indicating the MOF degradation in soil. "Reprinted with permission from (Wu et al. 2022) Copyright (2022) American Chemical Society."



Possible interactions between β -cyclodextrin, MOF-NPC, and four amide herbicides. "Reprinted with permission from (Liu et al. 2019a). Copyright (2019) American Chemical Society."

Fe found in plant nutrients were all present in both compounds $(C_8Fe_8N_{16}O_{52}P_8)$. These elements are arranged in corner-shared FeO₆ and PO₄ units, which link to oxalate units in both a and c

directions to form the framework. Although the N and P levels of both compounds are appropriate for their use as fertilizers, compound I) exhibits the superior oxalate solubility needed to

Synthesis methods for MOF	Details	Advantages	Disadvantages	References
Mechanochemical synthesis	In general, ball milling is used in this synthesis to efficiently combine reactants, either without or with very little solvent	Environment friendly, Safe	Yield is low	Gao et al. (2022); Rautenberg et al. (2022); Dubadi et al. (2023)
Microwave-assisted synthesis	Microwave-assisted synthesis typically operates on the principle of aligning the material's dipoles in an external field via microwave excitation	Short reaction time	Limited reaction solvent requirements	Inamuddin and Mohammad, (2018); Couzon et al. (2022)
		High-quality small-size MOF produced		
		No by-products		
Ultrasonic-assisted synthesis	It involves an ultrasonic cleaner that, at normal temperature and atmospheric pressure, could execute several processes at once	The process is fast by the quick dissolution of solutes	Low yield	Mirzaie et al. (2017); Liu et al. (2019b)
Microfluidic synthesis	Microfluidic systems have the ability to accurately control and confine molecules or particles inside a very small volume of liquid, as well as operate numerous functional units concurrently in programmed manner	Organic and inorganic phases are well mixed	Adaption is limited	Bailey et al. (2021); Xie et al. (2022)
Biomimetic mineralization	In this biosynthesis process the organisms create hierarchical organic-inorganic composites for maintenance, growth, and biological evolution	Efficient biocompatible particles can be produced	The method is limited to the biomacromolecules comparable to the pore size of the MOFs	Nie et al. (2022); Singh et al. (2022)
Hydrothermal/solvothermal synthesis	It represents a wet chemical approach employed where the precursors are dissolved in a controlled aqueous solution with high vapor pressure and elevated temperature condition	Low cost and compatible with bulk production	Specific temperature requirements and long reaction time	Chen et al. (2020b); Zeng et al. (2021)
Room-temperature synthesis	This synthesis procedure is carried out at room temperature, with instantaneous crystallization, and rapid growth time	Simple, cost-effective, and compatible with bulk production	Limited usage	Liu et al. (2019b); Wang et al. (2022)

TABLE 1 MOF synthesis methods and their advantages and disadvantages.

start the bacterial mineralization of the structural oxalate, which leads to structural collapse and release the plant nutrients in the soil. Compound II) has an excessively high solubility in oxalate, which may be due to a stronger surface connectivity of bigger macroscopic pores (Anstoetz et al. 2016). In addition to offering a route for more effective nutrient supply control than conventional fertilizers, the use of such MOFs in agriculture may assist address decreased soil fertility caused by acidification. Some more studies are listed in Table 2.

4.2 Pesticides

To protect the crops, yield, and quality, the application of pesticides is an indispensable part of modern-day agriculture. Their applications have assured the increasing the food demand of worlds' growing population to some extent. However, the contamination of pesticides in food and environment is a serious concern. Earlier studies have indicated that the application of nanopesticides has many advantages like more scattering of the active ingredient, more bioavailability, longer durability, effective control, and sustainability (Balah and Pudake, 2019; Ray et al., 2023). Some of the studies using the MOF for sustainable release of pesticides are discussed below.

The exceptional chemical and thermal stability, non-toxicity, and mass production of nanoscale UiO-66 made it a promising candidate for use as a pesticide carrier (Meng et al., 2020). The nanoscale UiO-66 (LC@UiO-66) was used in this study to encapsulate lambda-cyhalothrin (LC) molecules, which had excellent sustained-release properties and an extremely high loading (87.71%). There had never been a loading of the active ingredient of this high for LC@UiO-66. Using nitrogen adsorption-desorption isotherms, thermogravimetric analysis measurements (TGA), Fourier transform infrared (FT-IR) spectroscopy, transmission electron microscopy (TEM), and X-ray diffraction (XRD), LC@UiO-66 nanoparticles were uniquely identified. The

Sr	MOF/MOF composite	Fertilizer/ Nutrient	Loading efficiency	Results	References
1	Fe-MOF	N, P, and Fe	6.03, 14.48, and 14.69%	Faster nutrient release and large-scale synthesis	Wu et al. (2022)
2	OPA-MOF	NH ⁴⁺ and K+	>40%	Good, controlled release over time	Du et al. (2021)
				Lower synthesis cost	
3	β-CD MOF- NPC	K ⁺		$\beta\text{-}\mathrm{CD}$ MOF-NPC can accommodate the function of both adsorbent and fertilizer	Liu et al. (2019a)
4	MOF(Fe)@NaAlg aerogels composite	NH ⁴ ⁺	29.4 mg/g	Sustained release and excellent soil water retention capability	Wu et al. (2020)
5	urea-templated Fe-OPA	N, P	N- 3.2% and 5.8%	The release mechanism has good oxalate solubility	Anstoetz et al. (2017)
			P- 11.3% and 15.6%		
6	Fe-carbon nanofibers (CNF)- modified Mo-MOF	Fe and Mo	NA	A low dose (0.22 g kg-1 soil) of Fe-CNF/Mo-MOF to soil was sufficient for plant growth promotion	Pandey et al. (2024)
7	Zeolitic imidazolate framework-8	Zn	NA	Along with pesticide, it provided the Zn micronutrient to plant	Ma et al. (2022)

TABLE 2 List of MOF-based materials associated with the slow release of nutrients in agriculture.

aggregate particle size of LC@UiO-66 nanoparticles was around 270 nm, and they exhibited a cubic form. LC@UiO-66 demonstrated a longer-lasting insecticidal impact in comparison to free LC and commercially available microcapsules, as demonstrated by the bioassay against *Musca domestica* (hygiene pests). In comparison to free LC and commercially available microcapsules, the bioassay demonstrated that LC@UiO-66 had a long-term insecticidal effect. Furthermore, the insecticidal activity of LC@UiO-66 was shown to be superior to that of commercially available microcapsules. Moreover, LC@UiO-66 effectively inhibited the activity of *Aphis craccivora* Koch, a pest of many crops. The aforementioned findings all showed that LC@UiO-66 has outstanding insecticidal properties in both agricultural and public hygiene (Meng et al., 2020).

In another study, the smart core-shell carrier was designed for the stimuli-based release of pesticides. It includes MIL-101(Fe) @silica for site-specific delivery of chlorantraniliprole (CAP) and also as a source of plant nutrition (Gao et al., 2021). The encapsulated CAP was 16.5 times more stable under the sunlight than the conventional formulation. The premature release of active ingredient was prevented by silica shell, and the release from MIL-101(Fe) core was triggered by basic environment. During the laboratory assay with Plutella xylostella larvae the LC50 values for the MOF based nanoformulation was found to be 0.389 mg/L. In green house experiment they found that CAP@MIL-101(Fe) @silica formulation was significantly more effective in achieving mortality rate and was safe for crops. The treatment enhanced the crop biomass by 18.6%-19.6% and 4.8%-17.7% in terms of fresh and dry weight, respectively. This growth promotion was attributed to the simultaneous action of MIL-101(Fe)@silica in site-specific pesticide delivery and acting as a fertilizer for crop (Gao et al., 2021).

In another study, zinc-based zeolitic imidazolate frameworks (ZIF-8) with encapsulated ortho-disulfides have been produced (Mejías et al., 2021). After 2-hydroxypropyl- β -cyclodextrin HP- β -CD surface functionalization, an *in-situ* approach was used to achieve encapsulation, which in turn improved the water solubility by more than eight times. The effectiveness of the

encapsulation process and the sample's crystallinity were confirmed further. Studies on the main weeds that infect rice, maize, and potato filed have shown that inhibition produces better results than the commercial herbicide- Logran. This discovery, in conjunction with a brief synthesis time of 2 hours at 25°C, represents a novel class of herbicides derived from natural products that have immediate agricultural applications (Mejías et al., 2021).

A degradable and intelligent MOF-based nanopesticide with MIL-101(Fe^{III}) that is gated using Fe^{III}-tannic acid (TA) networks was developed. In this formulation, the coordinatively unsaturated Fe^{III} sites on the surfaces of the MIL-101(Fe^{III}) nanocarriers naturally orient the capping of the Fe^{III}-TA network gatekeepers. In this study, the release from the nanocarrier was stimuli based; and was tested for seven stimuli including the pH, H₂O₂, glutathione, phosphates, ethylene diamine tetraacetate (EDTA), and sunlight. The release of pesticide was studied in field conditions and was nanocarrier was degradable. The reported nanocarrier improved the wettability of aqueous formulation due to the naturally occurring polyphenol in model crops. It increased the pesticide adhesion and retention on foliage and was found to be effective against *Rhizoctonia solani* (rice sheath blight) and *Fusarium graminearum* (wheat head blight) by stimuli release of tebuconazole (Dong et al., 2021).

In another study, Chen and team synthesized aluminum-based metal-organic framework material (NH₂-Al-MIL-101) with a "cauliflower-like" structure and a high surface area of 2,359.0 m²/g by using Al³⁺ as the metal node and 2-amino terephthalic acid as the organic chain (Chen et al., 2021). The fungicides diniconazole (Dini) and azoxystrobin (AZOX) were concurrently encapsulated into NH₂-Al-MIL-101, with loading contents of 29.72% and 6.71%, respectively. The AZOX@Dini@NH₂-Al-MIL-101 dual fungicide delivery system showed pH-responsive and prolonged release patterns. Dini and AZOX had respective release times of 46 and 136 h when their maximum cumulative release rates reached almost 90%. Moreover, AZOX@Dini@NH₂-Al-MIL-101 demonstrated improved germicidal efficiency against rice sheath blast (Chen et al., 2021).

Sr	MOF/MOF composite	Pesticide	Loading efficiency	Results	References	
1	Iron-based MOFs (NH ₂ -Fe-MIL-101)	Diniconazole (Dini)	28.1%	- The formulation displayed a better control of pathogenic fungus-causing wheat head scab (<i>Fusarium graminearum</i>)	Shan et al. (2020b)	
				- The modification of Dini@NH ₂ -Fe-MIL- 101 with polydopamine (PDA) made the formulation pH-sensitive, and the release of Dini was much faster in an acidic medium compared to that in neutral and basic media		
2	FeCl ₂ (H ₂ O) ₂ O(BDC)3@carboxymethyl chitosan (MIL-101@CMCS)	Dinotefuran (DFN)	24.5%	- Around 83.1% of DNF was citric acid based released from the inner layer of MOF	Feng et al. (2020)	
				- The photodegradation of DNF was avoided by MOF formulation	-	
3	UiO-66 with polydopamine-coated Fe ₃ O ₄ (Fe ₃ O ₄ @PDA@UiO-66)	Imidacloprid (IMI)	15.87%	- The insecticidal activity of formulation toward <i>Aphis craccivora</i> Koch was better as compared to free IMI	Meng et al. (2021)	
				- The magnetic carrier can be readily collected after pesticide release, which may induce sustainability by reducing the hazardous effects of the residual pesticide on the ecosystem	-	
4	pH-jump reagent-loaded zeolitic imidazolate framework-8 (PD@ZIF-8)	Prochloraz (Pro)	NA	Pro release by UV light was $63.4\% \pm 3.5\%$ as compared to $13.7\% \pm 0.8\%$ under the dark conditions	Liang et al. (2021)	
5	Sodium lignosulfonate (SL) with protonated UIO-66-NH $_2$	Thiamethoxam (TMX)	33.56% ± 2.82%	The formulation resulted in excellent biosafety and stability in an outdoor environment	Huang et al. (2021)	
6	Copper-doped ZIF-8 (Cu@ZIF-8)	Fludioxonil	23.9%	- Flu@Cu@ZIF-8 exhibited pH-responsive release	Xu et al. (2023)	
				- Aquatic toxicity of fludioxonil was reduced by up to 68%	-	
7	ZIF -8	Dinotefuran (DNF)	-	- Sustained DNF delivery was achieved over a period of month in soil	Ma et al. (2022)	
				- The stability of DNF in UV was about 10 times more		
				- Also the insecticidal efficiency was increased		
8	ZIF-8	Clothianidin	9.8%	- Acidic pH-based 'burst release effect' was	Ma et al. (2023)	

TABLE 3 List of MOF-based materials associated with the slow release of pesticides for sustainability in agriculture.

Tang and team reported the construction of a porphyrin metalorganic framework (MOF) nanocomposite by introducing 5,10,15,20-tetrakis(1-methyl-4-pyridinio) porphyrin tetra(p-toluenesulfonate) (TMPyP) as a photosensitizer (PS) within the cage of MOF (HKUST-1) to effectively produce singlet oxygen under light irradiation (Tang et al., 2021). As a consequence of its good and broad-spectrum photodynamic antimicrobial activity against two pathogenic bacteria and three plant pathogenic fungi, the produced PS@MOF was shown to have a loading rate of roughly 12% (w/w). In the pot experiment, PS@MOF additionally showed exceptional control efficiency against *Sclerotinia sclerotiorum* on cucumber. A safety test on Chinese cabbage and cucumber along with *Allium cepa* chromosome aberration assay showed that PS@MOF was safe for plants and did not cause any genotoxicity. Porphyrin MOF thus showed significant promise as a new and effective alternative pesticide for sustainably managing plant diseases (Tang et al., 2021).

-43% pest control efficacy against *Nilaparvata lugens* after 10 days of application

observed

Yang and team developed two Ca^{2+} ions and l-lactate $[CH_3CH(OH)COO^-]$ enriched porous, chiral metal-organic frameworks (MOFs): MOF-1201 and MOF-1203 (Yang et al., 2017). The carboxylate and hydroxyl groups of lactate and the carboxylate group of acetate bridge Ca^{2+} ions to produce a three-dimensional arrangement of Ca(-COO, -OH) polyhedra that support one-dimensional pores with apertures and internal diameters of 7.8 and 9.6 Å (MOF-1201) and 4.6 and 5.6 Å (MOF-1203), respectively. These MOFs provide lactate and Ca^{2+-} based porous structures. They used it to encapsulate an important fumigant cis-1,3-dichloropropene in agriculture. The

hydrolysis of MOF-1201 released the fumigant and indicated the development of biodegradable porous solid carrier for agrochemicals (Yang et al., 2017).

In a recent study, an intelligent stimuli-triggered release formulation of pyraclostrobin (PYR) was developed using the binding of pectin with FeMOF nanoparticles (Liang et al., 2022). The loading of PYR in FeMOF-pectin nanoparticles was achieved up to 20.6%, and release from the MOF was achieved with the action of dual stimuli of redox and pectinase. The MOF and pectin protected PYR from photodegradation and improved the spreading on the rice leaves. This resulted in achieving better efficiency in controlling Magnaporthe oryzae as compared to the control. The formulation also promoted the growth of rice without any phytotoxicity effects (Liang et al., 2022). In a similar study, the Fe-based MOF was used as a multifunctional carrier of Azoxystrobin (Shan et al., 2020a). A simple octahedral Fe-MOF achieved the loading of 16.2% fungicide active ingredient. Along with this, it also acted as a source of Fe nutrients and promoted wheat growth. The initial release from this Fe-MOFs was pHresponsive and then the fungicide was sustained released. It effectively controlled the two pathogens - wheat head scab (F. graminearum) and tomato late blight (Phytophthora infestans) (Shan et al., 2020a).

To explore the mechanism of plant growth promotion due to the MOF-based fungicide carrier, a study was done by Zhao and team (2022). In their study, the physiological, biochemical, and metabolic levels were estimated in wheat, and it was found that many metabolites were upregulated by the application of Fe-MOF. In the comparative study, the team also found that the negative effects of fungicides were reduced in wheat seedlings when treated with tebuconazole-loaded Fe-MOF. The result of this study indicated that MOF regulates the metabolic pathways, and results in mitigation of the negative effects (Table 3) on MOF-based nanopesticides have indicated their potential in the field of modern agriculture for achieving high quality and yield.

4.3 MOF-based nanoplatforms for delivery of plant hormones

Plant growth hormones are naturally occurring or synthetic organic compounds that can regulate plant growth and development. They mainly include auxins, cytokinins, abscisic acid, salicylic acid, nitric oxide, gibberellins (GAs), jasmonates, ethylene, and brassinosteroids. When applied judiciously these chemicals have the potential to enhance crop resistance and increase crop production and quality. So, it is needed to have a delivery method for the slow release and protection of plant hormones for sustainability. For overcoming the defects of traditional drug delivery methods MOF can be the alternative. For GA a smart supramolecular delivery system was prepared with CLT6@ PCN-Q and found to have a loading capacity of 25.96%. The delivery system possesses multiple stimuli release based on pH, temperature, and competitive agent spermine (SPM). In plant growth studies, it was found that formulation was efficient in achieving stable GA bioactivity, promotion of wheat seeds germination, and stem growth of Chinese cabbage (Yang et al., 2021).

5 Conclusion and prospects

In conventional farming, overuse of agrochemicals, such as pesticides and fertilizers, is a dire problem since it can contaminate water and worsen the condition of the soil. In these circumstances, MOFs have emerged as a perfect candidate for the controlled release of agrochemicals and the liquid-phase adsorption of several dangerous compounds comprising sulfur and nitrogen. This review emphasizes the application of MOF as a novel carrier for agrochemical delivery and paves a novel path for the modern agriculture economy. They can be used for stimuli-based release, and the trigger can be achieved due to the changes happening in the crop microenvironment because of their interaction with pests. Along with these advantages of MOF for sustainability in crop production, they can also function as the source of some macro and micro-nutrients. Although this integrated strategy has many advantages, the transport and distribution in plants need further investigation on toxicology and safety. To effectively use agrochemicals, future research must concentrate on enhancing the loading capacity of MOF materials. MOF nanocarriers for agrochemicals provide enormous potential for lowering chemical needs and advancing sustainability.

Author contributions

SB: Writing-review and editing, Writing-original draft. PB: Writing-review and editing, Writing-original draft. RP: Writing-review and editing, Writing-original draft, Supervision, Conceptualization. PL: Writing-review and editing, Writing-original draft. UR: Writing-review and editing, Writing-original draft. SC: Writing-review and editing, Writing-original draft, Supervision.

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Conflict of interest

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