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Study of the deposition and resuspension phase of aerosol particles in a straight test pipe

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The study of the transport and deposition characteristics of aerosol particles in test sections is a component of the probabilistic safety assessment of nuclear reactors under severe accident scenarios. The deposited particles may become resuspended under favorable conditions, thus affecting the source term estimates. The objective of the present study was to perform experiments on a straight test pipe section 4 m long under deposition and resuspension phases. Zinc oxide metal particles generated from a plasma torch aerosol generator (PTAG) were used as the test aerosols. Deposition phase experiments were performed at a total carrier gas flow rate of 180 Lmin⁻¹, whereas the flow was increased to 1265 Lmin⁻¹ for the resuspension phase. Thermophoresis as an effect of PTAG enthalpy-governed temperature gradients was seen to dominate the deposition phase. The effects of varying Reynold numbers in different volume sections were reflected in a higher resuspended-to-deposited-mass-ratio in the downstream direction. A profile of deposited and resuspended masses was interpreted for the resuspension time of 20 min. Experimentally obtained characteristics were also compared with numerical results from simulations performed with the SOPHAEROS module of the Accidental Source Term Evaluation Code (ASTEC). This study, performed at the National Aerosol Facility (NAF), Indian Institute of Technology, Kanpur, India, indicates the need of more research on aerosol resuspension effects as this impacts the estimation accuracy of the source term.

KEYWORDS

ASTEC, deposition, metal aerosol, plasma torch aerosol generator, resuspension

1 Introduction

Nuclear energy is a viable long-term energy source given its abundance, nil contribution to global warming, and reliability. The main safety concern of operating a nuclear reactor for electricity generation is the possibility of design-based and other accidents. Given the psychological impact of any nuclear accident, advanced technologies have been adopted to make reactor operation failsafe and mitigate the consequences in the worst case of a failure. The melting of a reactor core under severe accident conditions has been a priority in reactor accident research studies. Understanding the phenomenology of an accident, engineering safety features, and taking corrective action after an unforeseen scenario have always been

enhanced by information gained after accidents like Three Mile Island (1979), Chernobyl (1986), and most recently, Fukushima (2011), [Sehgal \(2011\)](#). In a core meltdown scenario, radioactive fission products and structural materials may be released into the atmosphere in aerosol form. These aerosols can then be transported by advective currents up to thousands of kilometers from the release point ([Pöllänen et al., 1995](#)). Concentrations of aerosol particles during a reactor accident can be as high as 100 g/m^3 (mass), corresponding to $10^{13}/\text{m}^3$ ([Sher et al., 1994](#)). An evaluation of radioactive surface contamination in the environment resulting from a theoretical nuclear research reactor incident indicates that radioactivity levels can reach $9.5 \times 10^2 \text{ kBq/m}^2$ 60 km from the site along the plume centerline ([Xoubi, 2020](#)). It is also known that radioactive aerosol particles from fallout can enter the food chain ([Shahidi and Brown, 1998](#)) and damage the DNA of cells, even leading to cancer at exceedingly high doses ([Kamiya et al., 2015](#)). The resuspension of radioactive particles in the post-accident phase and residual contamination on the accident site could lead to delayed hazard after a reactor accident ([Kottapalli and Novosselov, 2019](#)). Radioactive particles were found to be resuspended and transported over intercontinental distances due to forest fires in the Chernobyl region ([Ierko et al., 2021](#)). Following the Fukushima Daiichi Nuclear Power Plant (F1NPP) disaster, investigations measured radiocesium levels in animals, plants, and fungi in Svalbard ([Mezaki et al., 2019](#)). The deposited insoluble radiocesium-bearing microparticles (CsMPs) released from the F1NPP accident in March 2011 become secondary contamination sources due to resuspension by atmospheric migration or other environmental transferring processes. These particles were detected 25 kms from the reactor even late in 2019, providing evidence of natural resuspension processes ([Tang et al., 2022](#)).

Understanding deposition and resuspension phenomena are important for source term evaluation under a severe reactor accident scenario. The deposition and resuspension of aerosol particles in reactor component sections in such cases determines the release of radioactivity into the environment. Radionuclides are less likely to be retained in the circuit when they are resuspended; when they are discharged directly into the environment in bypass sequences, the phenomena are of particular importance. The deposition of aerosol particles depends on factors like temperature gradient, gravitational settling, nature of flow, and aerosol geometries. The nature of the deposited layer (mono- or multi-layer), aerodynamic forces, deposited particle properties, and surface features are the parameters that affect resuspension processes. During a turbulent flow, the deposited layer of particles loses its ability to adhere to surfaces and becomes resuspended in flow. This happens when the aerodynamics forces (lift and drag) acting on the particle become larger than cohesive and adhesive forces for particle–surface interactions. Depending upon its size, gravitational force, and aerodynamic force strength, peeled-off aerosol particles follow a rolling motion or a combination of lifting and rolling in the resuspension phase. These mechanisms are modeled in two categories: force-and-moment balance models and energy balance models ([Gradoń, 2009](#)). Micro-videography-based investigation by [Ibrahim et al. \(2003\)](#) showed the rolling of particles and subsequent particle-to-particle collisions as two main mechanisms of detachment from the deposited surface, with rolling

as the primary contributor. This phenomenon is particularly relevant in reactor components such as vacuum breaker lines, steam generator (SG) tubes, and decay heat removal piping, where deposited materials can re-entrain into the gas flow during system depressurization or emergency cooling operations. Aerosol retention during steam generator tube rupture (SGTR) severe accidents is significantly higher in a flooded steam generator than in a dry one ([Lind et al., 2020](#)). Mechanical resuspension within the reactor coolant system (RCS) is particularly important in bypass scenarios such as SGTR as it can directly contribute to the source term. It may occur due to events like core quenching from delayed emergency core cooling system (ECCS) activation, core collapse into residual water at the vessel bottom, or rapid RCS depressurization ([Parozzi \(1992\)](#)). Additionally, resuspension can happen under high-velocity turbulent gas flows, even under steady flow rates.

In the context of reactor accident aerosol research, specific studies have been conducted in different facilities around the world. Several large-scale experimental campaigns have played a foundational role in shaping our understanding of aerosol behavior in light-water reactor (LWR) severe accidents. The LWR Aerosol Containment Experiments (LACE), conducted at HEDL in Richland and coordinated by Oak Ridge National Laboratory, investigated aerosol retention under containment-scale thermal-hydraulic conditions and provided critical validation data for codes such as MELCOR and CONTAIN ([Wright et al., 1986](#)). The CODEX-Aerosol program, conducted at the KFKI Atomic Energy Research Institute, explored aerosol generation and resuspension during simulated reactor core degradation using small-scale fuel bundles. Aerosol emissions—monitored in oxidizing and steam environments—showed strong correlations with temperature rise and fuel cladding oxidation, highlighting the role of chemical speciation and morphology in transport behavior ([Pintér Csordás et al., 2000](#); [Hózer et al., 2006](#)). The Aerosol Code Evaluation/Radioactive Tracer Facility (ACE/RTF) experiments further focused on iodine volatility and resuspension under high radiation fields and varying containment pH. These studies demonstrated that iodine speciation becomes independent of initial chemical form due to radiolytic interconversion and that stainless steel versus epoxy surfaces significantly affect sorption behavior. The series also highlighted modeling gaps in organic iodide prediction, surface reactivity, and radiation-induced transformations. Additionally, the Mod-Advanced Corium Experiments (MACE) program under ACE Phase D provided data on heat transfer during molten core–concrete interaction and the generation of aerosols and gases during quenching events—relevant for containment pressure and source term evaluation ([Kupferschmidt et al., 1992](#); [Sehgal and Spencer, 1992](#)). Aerosol behavior under severe accident conditions has been studied by performing small and large-scale experimental programs such as PHEBUS FP ([Gonfotti and Paci, 2018](#); [Kissane and Drosik, 2006](#)), DEMONA ([Liljenzin et al., 1990](#)), WAVE ([Hidaka et al., 2000](#)), DRAGON ([Suckow and Guentay, 2008](#)), STORM ([De los Reyes et al., 1999](#)), THAI ([Gupta et al., 2015](#)), Falcon ([Beard et al., 1992](#)), and Wind ([Maruyama et al., 1999](#); [Freitag et al., 2018](#); [Freitag et al., 2020](#)). Such studies in context of the Indian Pressurized Water Reactor Program (IPHWR) have been conducted in the National Aerosol Test Facility (NATF) at the Bhabha Atomic Research Centre, India ([Sapra et al., 2008](#); [Modi et al., 2014](#); [Dwivedi et al., 2019](#)). To the best of our knowledge, there

has been no research into the resuspension phenomenon in such medium- or large-scale facilities under dry conditions. The Indian Institute of Technology (IIT) Kanpur, has recently established the National Aerosol Facility (NAF), which aims to investigate the spatial and temporal evolution of metal oxide aerosols in simple and complex geometries incorporating the study of the deposition and resuspension behavior of aerosols in dry and wet conditions for the context of the IPHWR program. Studies such as the hygroscopic nature of nuclear-accident-relevant cesium-bound compounds (Mishra et al., 2019a; Mishra et al., 2019b), characterization of aerosol generator system (Dwivedi et al., 2020), and the methodology to perform measurements of droplet (Kumar et al., 2022) have been recently performed at NAF.

This study focuses on the deposition and resuspension characteristics of zinc metal oxide aerosol particles in a circular cross section straight piping system of stainless steel at NAF. The choice of a plasma torch aerosol generator (PTAG) as an aerosol generation system ensures that the characteristics of the aerosol particles in the test sections remain as needed for simulating the accident conditions. The choice of zinc oxide (ZnO) aerosols as the test material was based on a combination of operational and physical factors. Metal powders are mostly preferred in PTAGs (Venkatramani, 2002), resulting in controlled and stable aerosol generation, and zinc metal powder fulfils this requirement. In past, studies at NATF (Sapra et al., 2008) and NAF (Dwivedi et al., 2020) have been helpful in optimizing PTAG parameters for experiments such as the current study. In addition, resuspension and thermophoresis are mostly physical phenomena, so they accurately simulate the experiential behavior of any chemical/radioactive species as long as the size and concentration is normalized with respect to the demand conditions.

Despite extensive research on aerosol behavior in nuclear containments, a significant gap remains in understanding deposition and resuspension phenomena within reactor piping systems, especially under temperature gradients and dynamic flow conditions that arise during severe accidents. While existing models include resuspension modules, their validation is often limited to idealized geometries or small-scale test data. This study addresses that gap by conducting large-scale experimental simulations in a stainless-steel pipe representing a segment of reactor primary system piping using measured gas/wall temperatures and flow velocities as inputs. The central hypothesis is that thermal gradients and flow transitions directly govern deposition and resuspension dynamics and that models like SOPHAEROS can reproduce these effects when provided with realistic boundary conditions. The objective is twofold: first, to advance the physical understanding of aerosol behavior under accident-relevant thermal-hydraulic conditions, and second to evaluate the predictive capability of SOPHAEROS under such conditions. The study is designed to contribute to international efforts to refine source term predictions, improve safety code validation, and generate data relevant to severe accident management in light water reactors.

2 Materials and methods

2.1 Test section

The deposition and resuspension experiments were carried out in a test section of NAF. Various components of that

facility—powder feeder, plasma torch aerosol generator (PTAG) system, gas delivery system, chiller, and aerosol instrumentation—were utilized in these experiments. Zinc metal oxide particles, generated by PTAG by feeding a controlled supply of zinc powder from the powder feeder (model- MEC PF-3350) to the plasma zone, were used as “test aerosols” during these experiments. For deposition and resuspension analysis, a stainless steel pipe with an internal diameter of 7.62 cm and a length of 4 m was used as a test section. This was split into eight volumes, each 0.5 m long. The volumes were assigned names depending on their order of appearance as volumes 1 to 8. A pipe with a length of 1 m and diameter of 13.5 cm was used to transport the generated metal aerosol from PTAG to the test section. To prevent the overheating of this 1 m pipe section, chilled water was passed through the wrapped copper tubes on it; this section has been labeled “volume 0”. During the deposition phase of the experiment, eight thermocouples were placed on the outer walls of the pipes and eight in the center of the pipes to continuously monitor the wall and bulk gas temperature. The mass concentration (without size separation) was measured using a 47 mm stainless steel gross filter sampler, and the size-fractionated aerosol particles were collected using a Micro-Orifice Uniform Deposit Impactor (MOUDI, model:100 NR) for gravimetric analyses. Figure 1 shows a detailed schematic diagram of the experimental setup during the depositional phase.

For the resuspension phase of experiment, a 1 m connecting pipe was removed and sealed. Push-fit connectors were mounted at the beginning of the test section to facilitate the carrier's gas flow. Figure 2 shows the schematic diagram of the experimental setup during the resuspension phase.

The gross filter sampler and MOUDI were operated at a flow rate of 30 Lmin⁻¹. The other operating parameters are mentioned in Table 1 below.

2.2 Experimental procedure

The experiment was divided into two parts: the deposition phase and the resuspension phase. Pilot tests were carried out prior to the main experiments to optimize the aerosol generation parameters such as powder feed rate, plasma torch operating power, and carrier gas flow rate. The pilot experiments helped determine optimal duration and sampling flow rates for the experiment, ensuring that the aerosol instrumentation (MOUDI, PTAG flame nozzle, etc.) was not clogged while also ensuring adequate deposition on the impactor filter papers and in the test section. Aerodynamic diameter measurements using the impactor account for particle shape, density, and drag coefficient. Thus, while shape was not measured separately, it was incorporated in the aerodynamic behavior observed through the impactor. Table 1 lists the parameters used in all experiments. Before each experimental run, the entire test section was manually cleaned with a lint-free damp cloth, followed by flushing with HEPA-filtered air for 15 min using a centrifugal blower. This ensured a near-zero background aerosol concentration, confirmed by blank gross filter samples. Zinc oxide aerosols were generated continuously for 60 min at a controlled feed rate of 8.25 g/min. A stabilization period of 60 s after plasma torch ignition ensured a steady aerosol flow before particle injection into the test section. All measurement instruments

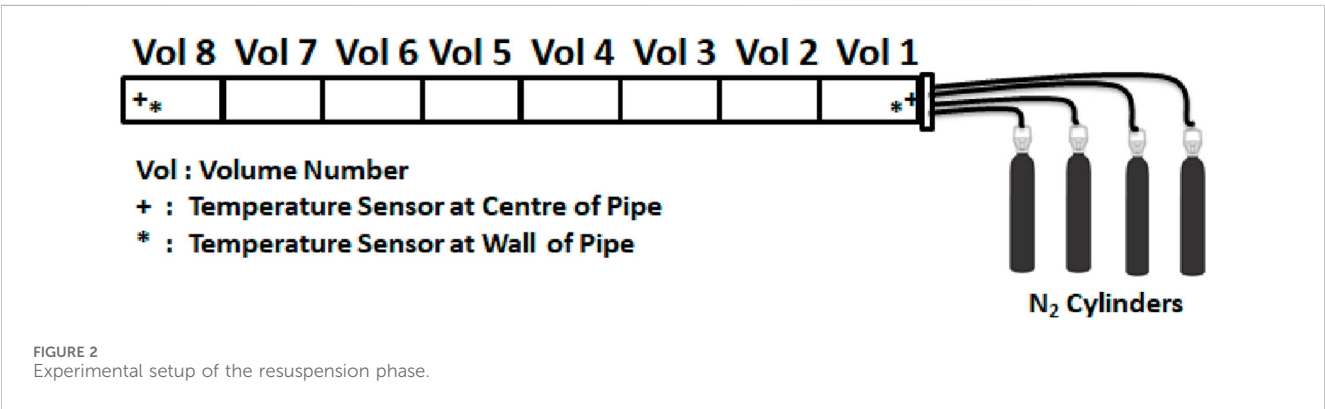
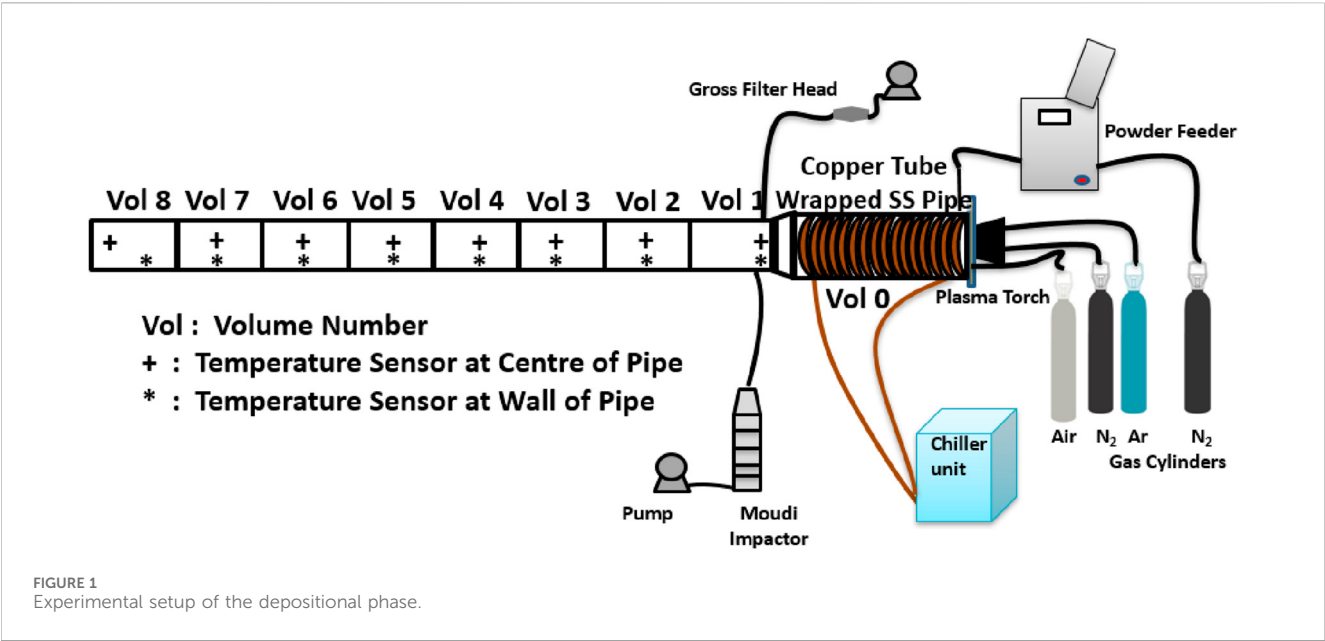


TABLE 1 Parameters used for experiment

Parameters During the Depositional Phase	
Depositional phase time	60min
Plasma torch aerosol generator (PTAG) power	25 kW
Carrier gas	Flow rate ($Lmin^{-1}$)
Argon flow rate	25
Nitrogen flow rate	128.75
Oxygen flow rate	26.25
Metal aerosol	Flow rate ($gmin^{-1}$)
ZnO	8.25
Parameters During the Resuspensional Phase	
Resuspension phase time	20min
Carrier gas	Flow rate
Nitrogen flow rate ($Lmin^{-1}$)	1265

used in the study were calibrated to ensure the accuracy and reliability of the results. The MOUDI was calibrated by the manufacturer using standard aerosol size reference particles. The gravimetric mass balance used for filter weighing was auto-calibrated and cross-checked using certified standard weights. The thermocouples employed for wall and gas temperature measurements were factory-calibrated against a certified mercury thermometer. Due to the large-scale and destructive nature of the experiment, full repeatability was not feasible. However, quality assurance was ensured through internal consistency checks and careful pre- and post-sampling filter weighing using a high-precision microbalance. The relative uncertainty in temperature measurements, as per the manufacturer's specification, ranged from ± 0.04 to ± 0.05 . The relative error in mass measurements ranged from ± 0.0004 to ± 0.0029 during the deposition phase and from ± 0.0013 to ± 0.0030 during the resuspension phase.

The log-normal particle size distribution, with a mass mean aerodynamic diameter (MMAD) of $0.5 \mu m$ and a geometric standard deviation of 2.01, was estimated using measurements taken with the impactor positioned upstream of the test pipe. Following the firing of the plasma torch, the powder feeder was

turned on until the plasma flame was stabilized for 60 sec. The plasma torch was switched off at the end of the deposition phase, and the test section was left to cool with both ends closed.

The resuspension phase of the experiment was performed the next day. This phase of the experiment was carried out with a carrier gas flow rate of 1265 Lmin^{-1} . During the depositional phase, measurements were performed 1 min after starting the experiment. Aerosol sampling was done in the deposition phase, upstream of the test section, with the MOUDI for mass size distribution estimation and a gross filter sampler for concentration measurement. On both measures, the overall sample period was 1 min.

As it was not possible to measure the deposited mass of aerosols in the test section and resuspended in the same aerosol, the experiment was divided into two phases: deposition and resuspension. In the deposition phase, aerosols were deposited in the test sections, scrubbed manually, and measured gravimetrically. In the resuspension phase, the aerosols were first deposited under same input parameters and run time that were used in the depositional phase, and then these aerosols were resuspended under the mentioned flow condition. This method was verified from the pilot experiments and found to be suitable.

3 Numerical approach

This work used numerical simulations performed with Aerosol Source Term Evaluation Code (ASTEC) to compare experimental and theoretical results. ASTEC, jointly developed by the French Institut de Radioprotection et de Sûreté Nucléaire (IRSN) and the German Gesellschaft für Anlagen und Reaktorsicherheit mbH (GRS), is used for simulating an entire severe accident sequence in a nuclear water-cooled reactor from an initiating event. It has been used for several studies/applications in varied contexts, such as source term estimation, probabilistic safety assessment, accident management, and phenomenological interpretations (Chatelard et al., 2014). The modular flexibility of ASTEC helps researchers design experiments for studying standalone as well as integral mechanisms that target specific reactor component systems (RCS) and/or a certain phase of the accident sequence. The SOPHAEROS module of ASTEC is intended to simulate major fission product vapor and aerosol phenomena in the RCS, composed of a 1D series of control volumes (Cousin et al., 2008).

The SOPHAEROS module solves aerosol and vapor transport equations using a volume-based iterative approach. Each control volume represents a segment of the physical pipe and is solved sequentially at every step using an implicit Newton–Raphson scheme. In each volume, SOPHAEROS first determines the vapor-phase chemical equilibrium based on the mass of elements and species transferred from the previous iteration. It then constructs a matrix system that includes fluxes and source terms related to key physical phenomena: homogeneous and heterogeneous nucleation, condensation/evaporation, coagulation, sorption, deposition, and mechanical resuspension. These fluxes are evaluated based on both volatile and non-volatile species. The model tracks aerosol and vapor mass in five physical states: (1) volatile species in the vapor phase; (2) suspended non-volatile aerosols; (3) volatile species condensed on walls; (4) deposited aerosols; (5)

sorbed vapor species on surfaces. Aerosols are also divided into multiple particle size classes, allowing for dynamic updates due to coagulation and deposition processes. At each step, the solver balances source terms (from aerosol input, inter-volume transport, and surface interactions), linear transport and deposition rates, and nonlinear terms (such as coagulation and condensation). The resulting nonlinear system is solved using an iterative Newton update, with convergence achieved based on changes in mass across all tracked states and size bins. This approach allows SOPHAEROS to capture the evolving aerosol size distribution and deposition behavior along the pipe under varying thermal and flow conditions using experimentally measured temperatures and flow rates as fixed boundary conditions.

In this study, only the SOPHAEROS module of ASTEC was available and utilized in the standalone mode. As a result, no thermal-hydraulic calculations were performed using CESAR (the module responsible for two-phase thermal hydraulics in full ASTEC simulations). Instead, the wall and gas temperature profiles (Figure 4) obtained experimentally during the deposition phase were directly supplied as input boundary conditions in each control volume. These fixed temperatures influenced deposition and resuspension phenomena through thermophoresis and vapor equilibrium models but were not dynamically computed. The key aerosol parameters given as input to the code are MMAD = 0.5μ , GSD = 2.01, and density = 5.6 g/cm^3 . The carrier gas flow rates used as initial conditions in the code are given in Table 1 along with the aerosol mass flow rate.

SOPHAEROS is a lumped-parameter code that employs a one-dimensional (1D) axial representation of the system in which the physical test pipe is modeled as a sequence of control volumes. While this approach allows for the simulation of axial aerosol transport and thermophoretically driven deposition, it inherently lacks the ability to capture three-dimensional (3D) flow characteristics, such as radial velocity gradients, parabolic profiles, and circumferential non-uniformities that naturally arise in pipe flow. Unlike full 3D computational fluid dynamics (CFD) codes such as RELAP5-3D or ANSYS Fluent, SOPHAEROS does not resolve detailed boundary layer development or wall shear stress distributions. As a result, while it cannot predict localized hot spots or asymmetrical deposition patterns, it can reasonably approximate overall deposition trends and axially averaged mass distributions. This dimensional simplification may introduce some uncertainties, particularly in entrance regions with steep gradients. However, when supplied with accurate thermal and flow boundary conditions, SOPHAEROS is capable of providing good agreement with experimental data and valuable insights into aerosol transport behavior.

Coagulation is an important phenomenon for the interaction and growth of aerosol particles. The coagulation of particles increases their size and hence enhances their deposition. The SOPHAEROS module integrates these phenomena in the aerosol dynamic equation. These depositional mechanisms are discussed in the subsequent sections. The mass balance equation for intra-volume mechanisms and inter-volume transport develops a nonlinear system which is solved by adopting an implicit numerical approach. All the phenomena are computed in each control volume. An implicit method of solution allows coupling between the condensation/evaporation of vapor on/from wall and

aerosol dynamical processes (nucleation, agglomeration/coagulation, and drifts) which are solved in each control volume. The aerosol distribution is modeled using a sectional approach. In all aerosol size classes, the same species composition is taken into account. This section discusses the mathematical framework of the SOPHAEROS module adopted for performing theoretical simulation.

3.1 Coagulation

Coagulation/agglomeration is the process by which particles are chemically or physically bonded together, resulting in coagulated particles. These coagulated particles tend to settle down more quickly than their parent particles since they are larger in size.

There are three basic mechanisms for aerosol coagulation to consider the following:

- Brownian coagulation
- gravitational coagulation
- turbulent coagulation.

These mechanisms are discussed below.

3.1.1 Brownian coagulation

The model for the calculations of Brownian coagulation kernels in continuum flow regime is discussed in Loyalka (1976) and Williams and Loyalka (1991).

In this model, coagulation kernel between particle size classes i and j can be written as follows (Equation 1):

$$K_{i,j}^b = 4\pi \frac{\gamma}{\chi} (r_i + r_j) (D_B^i + D_B^j) \quad (1)$$

—where

$K_{i,j}^b$ is the Brownian coagulation kernel between particle size classes i and j (m^3/s)

γ is particle collision shape factor (=1)

χ is particle dynamic shape factor (=1)

$r_{i,j}$ is the radius of particles i and j (m)

$D_B^{i,j}$ is the Brownian diffusivity of particles i and j (m^2/s) and is given by (Equation 2):

$$D_B = \frac{kT_f}{6\pi\mu_f} \frac{C_n}{r} \quad (2)$$

μ_f is carrier-fluid dynamic viscosity ($kg/m/s$)

C_n is the particle Cunningham factor

T_f is the carrier fluid temperature in Kelvin

k is the thermal conductivity of gas ($W/m/K$).

3.1.2 Gravitational coagulation

The coagulation kernel between particle size classes i and j is calculated using the below Equation 3 (Pruppacher and Klett, 2012):

$$K_{i,j}^g = \pi \gamma^2 \xi_{ij} (r_i + r_j)^2 |V_{d,i}^s - V_{d,j}^s| \quad (3)$$

—where:

$V_{d,i}^s$ is the gravitational-settling velocity of particle size class i (m/s)

$V_{d,j}^s$ is the gravitational-settling velocity of particle size class j (m/s)

$K_{i,j}^g$ is the gravitational-coagulation kernel between particle size classes i and j (m^3/s)

ξ_{ij} is the coagulation efficiency between particle size classes i and j and is given thus (Equation 4)

$$\xi_{ij} = \text{Min} \left[0.5 \left\{ \frac{\text{Min}(r_i, r_j)}{(r_i + r_j)} \right\}^2, \frac{1}{18} \right] \quad (4)$$

3.1.3 Turbulent coagulation

The turbulent coagulation kernel (Equation 7) is based on a quadratic combination of the shear turbulent (sh) (Equation 5) and inertial turbulent (in) kernels (Equation 6) and follows the Saffman–Turner approach ((Levich and Tobias, 1963) Williams and Loyalka, 1991). For shear turbulence, the kernel can be written as

$$K_{i,j}^{sh} = \frac{2}{3} \left\{ \gamma (r_i + r_j) \right\}^3 \sqrt{\frac{2\pi E_T}{\nu_f}} \quad (5)$$

—where

E_T is carrier fluid turbulent dissipation energy density (m^2/s^3)

ν_f is carrier fluid kinematic viscosity (m^2/s)

$K_{i,j}^{sh}$ is the shear turbulent coagulation kernel between particle size classes i and j (m^3/s).

The following relationship determines the inertial turbulent coagulation kernel ($K_{i,j}^{in}$) between size classes i and j (m^3/s)

$$K_{i,j}^{in} = 2\sqrt{2\pi} \frac{\gamma^2}{g} \xi_{ij} (r_i + r_j)^2 |V_{d,i}^s - V_{d,j}^s| \left(\frac{8\pi E_T^3}{15\nu_f} \right) \quad (6)$$

A quadratic combination yields the resulting turbulent coagulation kernel and is given by

$$K_{i,j}^t = \sqrt{(K_{i,j}^{sh})^2 + (K_{i,j}^{in})^2} \quad (7)$$

$K_{i,j}^t$ is the turbulent coagulation kernel between particle size classes i and j (m^3/s).

3.1.4 Resultant coagulation

In the case of simultaneously active mechanisms, all the above four coagulation kernels must be examined in order to calculate the resultant coagulation kernels (Equation 8).

A linear and quadratic combination produces the resulting coagulation kernel:

$$K_{i,j} = K_{i,j}^b + \sqrt{\frac{8}{3\pi} (K_{i,j}^g)^2 + (K_{i,j}^{sh})^2 + (K_{i,j}^{in})^2} \quad (8)$$

The coagulation kernels ($K_{i,j}$) for $1 \leq i \leq N_c$ and $j \leq i$ were calculated for symmetry reasons. When coagulation occurred between two particles of the same size class ($i = j$), the value of kernel $K_{i,i}$ was divided by 2.

3.2 Deposition mechanism

Gravitational settling, laminar diffusion deposition, and thermophoresis are the primary mechanisms responsible for

aerosol deposition. The models for the deposition process used by the SOPHAEROS module of the ASTEC are discussed below.

3.2.1 Gravitational settling

Gravitational settling is the accumulation of particles under the influence of gravity. Since the mass of an aerosol particle is small, the corresponding settling velocity can be termed the “Stokes velocity” (Equation 9).

When the drag force of the fluid on the particle is precisely equal and contrary to the force of gravity, the Stokes velocity is calculated using the equation below.

$$V_{st} = \frac{2}{9} \frac{g \rho_p r^2}{\mu_f \chi} \quad (9)$$

The settling velocity is given by Equation 10 below

$$V_d^s = V_{st} f_{st} C_n \quad (10)$$

The corresponding settling rate is given in below Equation 11

$$T_d^s = \frac{A_t}{\Omega} V_d^s \quad (11)$$

—where:

V_d^s is the settling velocity (m/s)

V_{st} is the Stokes velocity (m/s)

g is acceleration due to gravity (m/s²)

ρ_p is particle density (kg/m³)

r is particle radius (m)

μ_f is carrier fluid dynamic viscosity (kg/m/s)

χ is the particle shape factor (= 1)

f_{st} is the Stokes velocity correction factor and is given by Equation 12 for Reynolds Number less than equal to 0.001, Equation 13 for Reynolds Number 0.01 to 29500 and Equation 14 for Reynolds Number greater than 29500.

$$f_{st} = 1, R_{ep} \leq 0.001 \quad (12)$$

$$f_{st} = \log_{10}(24R_{ep}) \quad 0.01 < R_{ep} < 29500 \quad (13)$$

$$f_{st} = \frac{7.385}{\sqrt{R_{ep}}} \quad R_{ep} \geq 29500 \quad (14)$$

R_{ep} is the particle Reynolds number

C_n is Cunningham's slip correction factor

T_d^s is the settling rate (/s)

A_t is the wall settling area (m²)

Ω is the control volume (m³).

3.2.2 Laminar diffusion

For $Re \leq 2300$ —that is, for the laminar flow regime with no boundary layer—laminar or Brownian diffusion occurs. The particle diffusion coefficient, also known as particle “Brownian diffusivity,” describes this mechanism. The Brownian motion becomes more vigorous as diffusivity increases. When the aerosol concentration at the surface is zero, the concentration gradient allows aerosol particles to diffuse to the surface. Aerosol particles, unlike gas molecules, bind to the surface and deposit on it. For measuring diffusive deposition, analytical correlation is used (Williams and Loyalka, 1991; Gormley and Kennedy, 1948). To begin with, a dimensionless parameter is introduced given by 1, Equation 15:

$$h = 2 \frac{D_B L_W}{V_f D_h^2} \quad (15)$$

Then, two ratios (Equation 16 and 17) based on the values of h are considered:

$$\frac{n_{out}}{n_{in}} = 1 - 4.07h^{2/3} + 2.4h + 0.446h^{4/3}, \text{ for } h \leq 0.0156 \quad (16)$$

$$\frac{n_{out}}{n_{in}} = \frac{0.819}{\exp(7.31h)} + \frac{0.0975}{\exp(44.6h)} + \frac{0.0325}{\exp(144h)}, \text{ for } h \geq 0.0156 \quad (17)$$

The laminar diffusion deposition velocity is given by Equation 18

$$V_d^{bd} = \frac{\pi}{4} \frac{D_h^2 V_f}{A_t} \left(\frac{n_{inl}}{n_{out}} - 1 \right) \quad (18)$$

Its corresponding deposition rate is given by Equation 19

$$\tau_d^{bd} = \frac{A_t}{\Omega} V_d^{bd} \quad (19)$$

—where:

L_W is the wall length (m)

V_f is carrier fluid velocity (m/s)

D_h is the wall hydraulic diameter (m)

n_{out} is the inlet particle concentration (/m³)

n_{in} is the outlet particle concentration (/m³)

V_d^{bd} is laminar diffusion deposition velocity (m/s)

τ_d^{bd} is the laminar diffusion deposition rate (/s)

T_f is carrier fluid temperature (Kelvin)

D_B is particle Brownian diffusivity (m²/s).

3.2.3 Thermophoretic deposition

An aerosol particle in a gas experiences a force in the direction of reducing temperature if a temperature gradient occurs in that gas. The aerosol particles migrate toward the wall surface and bind to it due to the higher temperature of the gas relative to the wall surface temperature; this is known as thermophoretic deposition. The Talbot formulation (Talbot et al., 1980) is used to calculate thermophoretic deposition velocity (Equation 20):

$$V_d^{th} = \frac{\nu_f}{T_f} \frac{dT_w}{dr} \frac{2C_s \left(\frac{k_f}{k_p} + C_t K_n \right) C_n}{(1 + 3C_m K_n) \left\{ 1 + 2 \left(\frac{k_f}{k_p} + C_t K_n \right) \right\}} \quad (20)$$

The corresponding deposition rate is given in Equation 21

$$\tau_d^{th} = \frac{A_t}{\Omega} V_d^{th} \quad (21)$$

—where:

V_d^{th} is thermophoretic deposition velocity (m/s)

ν_f is carrier fluid kinematic viscosity (m²/s)

T_f is carrier fluid temperature (K)

$\frac{dT_w}{dr}$ is the wall-fluid temperature gradient (K/m)

C_s is the thermal slip coefficient (1.17)

k_f is carrier fluid thermal conductivity (W/m/s)

k_p is particle thermal conductivity (W/m/s)

C_t is the temperature jumping coefficient (2.18)

K_n is the Knudsen number

C_m is moment accommodation coefficient (1.14)
 τ_d^{th} is the thermophoretic deposition rate (/s).

3.3 Resuspension

Aerosol transport is a function of carrier fluid, aerosol deposition, and aerosol resuspension. Resuspension restores deposited aerosols to the fluid system as resuspended aerosols are again available to transport with the fluid and may contribute to an increase in the source term. Dry deposited aerosol is easier to resuspend than wet. The Reynolds number plays a vital role in the resuspension of aerosol. Numerically defined, the resuspension force (F_{res}) is defined as the difference of the aerodynamic lift force and total adhesive force and is different for wall/ceiling (Equation 22) and bottom surfaces (Equation 23).

For wall/ceiling surfaces, it is given by

$$F_{res} = F_d + F_b - (F_c + f\gamma F_c) \quad (22)$$

For bottom surfaces, it is given by

$$F_{res} = F_d + F_b - F_g + F_c + f\gamma(F_g + F_c) \quad (23)$$

For resuspension forces $F_{res} > 0$ and the rate of resuspension, (Λ) is given as a function of the resuspension force which is represented in Equation 24 (Parozzi et al., 1995):

$$\Lambda = E_1 [10^6 F_{res}]^{E_2} \quad (24)$$

The empirical constants E_1 and E_2 used for resuspension rate were estimated using data from several experiments (Wurelingen, STORM, and Oak Ridge); their default values in the SOPHAEROS module were 6.9 and 0.89, respectively. F_b is burst force (N), F_c is cohesive force (N), F_d is drag force (N), and F_g is gravitational force (N), which are discussed in subsequent sections.

ASTEC's SOPHAEROS module calculates resuspension using the force balance model by default. The first statistical force balance model considering the effect of various forces on a deposited particle in a turbulent boundary layer was proposed by Cleaver and Yates (1973). If the vector sum of aerodynamic forces acting on a surface is greater than the adhesive forces detaching the particle from the surface, the aerosol is resuspended from the surface. The term "adhesive forces" refers to any of the forces that cause particles to adhere to depositing surfaces, such as gravitational forces, inter-molecular forces such as van der Waals interactions such as cohesive and adhesive force (Katainen et al., 2006), and chemical bonds such as hydrogen bonds (Krupp and Sperling, 1966). The resuspension rate formula is based on several results (Parozzi, 1992; Wright et al., 1984; Fromentin, 1989). The resuspension rate is measured at the bottom and wall surfaces of each wall where particle resuspension is to be considered for each particle species and particle size class.

3.3.1 Adhesive forces

Adhesive forces are categorized into gravitational, cohesive, and frictional forces. These forces cause aerosol to stick and form a layer. Gravitational settling occurs only in the direction of gravity; thus, gravitational force (F_g) on the top surface can be neglected. This

force is an important component of the adhesion force for larger particles ($>1\mu m$) and is given by Equation 25.

$$F_g = \frac{4}{3}\pi r^3 \rho_p g \quad (25)$$

—where:

r is the particle radius (m)

ρ_p is particle density (kg/m^3)

g is gravitational acceleration (m/s^2).

Cohesive forces (F_c) are due to the inter-molecular attraction force of molecules and are expressed as by below Equation 26

$$F_c = 2rH_c\gamma \quad (26)$$

—where:

H_c is a constant which depends on material properties (N/m), the SOPHAEROS module uses the default value of 10^{-6} (N/m), γ is the coagulation form factor.

Frictional forces (F_f) are due to the resistance experienced by particles while gliding on surfaces. Frictional forces are due to all the adhesive forces and include the gravitational force in the direction of gravity and is given by below Equation 27.

$$\begin{aligned} F_f &= f\gamma(F_g + F_c), \text{ for bottom surface;} \\ F_f &= f\gamma F_c \text{ for wall/ceiling surface;} \end{aligned} \quad (27)$$

where f is the friction factor having a default value of 0.2 in the SOPHAEROS module.

3.3.2 Aerodynamic forces

The aerodynamic forces comprise the drag force (F_d) and the lift force due to turbulent burst (F_b) in the fluid. The drag force is associated with the shear stress that the turbulent fluid flow imparts on the deposited particles and is given by Equation 28.

$$F_d = \tau \pi r^2 \chi^2 \quad (28)$$

—where the shear stress τ is calculated as a function of fluid velocity using the relation given by Equation 29

$$V_f^w = \sqrt{\frac{\tau}{\rho_f}} = V_f \sqrt{\frac{f_w}{2}} \quad (29)$$

—where:

τ is shear stress (N/m^2)

χ is the particle dynamic form factor

V_f^w is wall friction fluid velocity (m/s)

ρ_f is carrier-fluid density (kg/m^3)

V_f is carrier-fluid velocity (m/s).

f_w is the wall Fanning friction factor.

Turbulent burst randomly contributes to the resuspension of particles. The model is based on the measurement of the frequency distribution of its occurrence and is given in Equation 30. The data have been estimated from experiments for clean fluid flow conditions as a function of friction velocity V_f^w and kinematic viscosity ν

$$F_b = A_b \nu^2 \rho_f \left[\frac{2rV_f^w}{\nu} \right]^{B_b} \chi \quad (30)$$

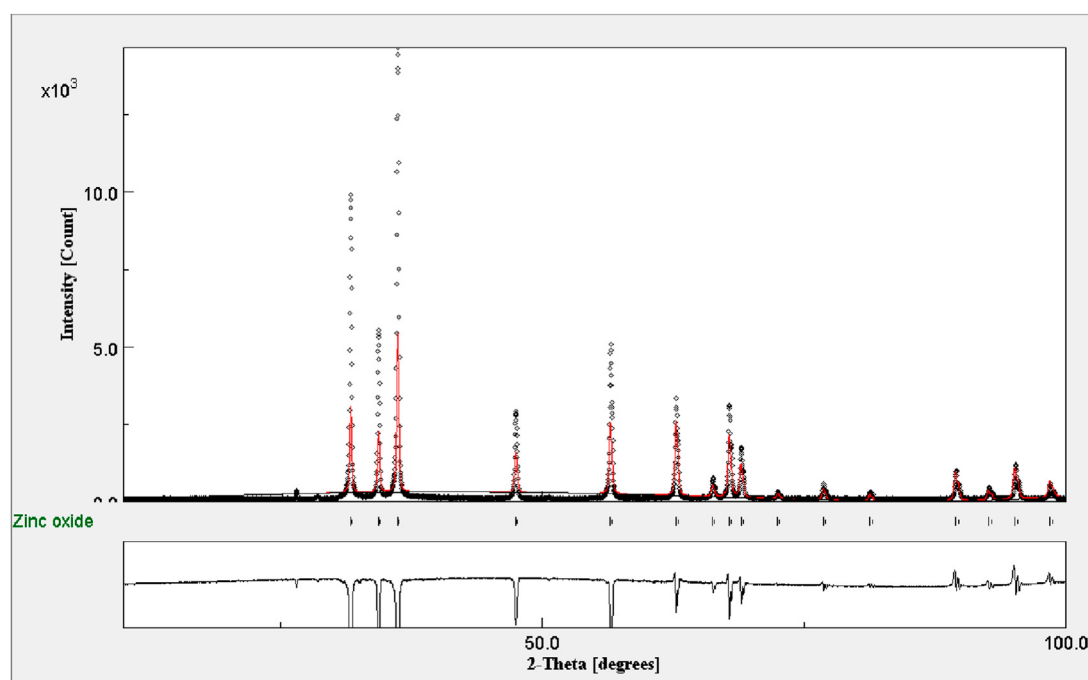


FIGURE 3
XRD plot of deposited aerosol particle.

—where:

ν is kinematic viscosity (m^2/s).

A_b and B_b are model constants.

The default value model constant in the SOPHAEROS module is taken from Parozzi et al. (1995).

4 Results and discussion

The experiments were conducted in two stages: depositional and resuspensional. Pilot experiments were used to establish the inlet concentration during the depositional phase and carrier gas flow rate during the resuspension phase. The mass concentration of inlet particles was kept at approximately 5 g/m^3 (measured using the gross filter sampler) so as to lead to sufficient particle deposition in the test volumes. It was also observed that no resuspension happened at low carrier gas flow rates ($<500 \text{ Lmin}^{-1}$). Table 1 lists the operating parameters for the deposition and resuspension processes.

4.1 XRD analysis

The crystalline phase of the deposited metal oxide particles on the inner wall of the test volumes was determined using X-ray diffraction (XRD). A Cu K α radiation source (40 kv, 40 mA) and a Lynx Eye 1D detector with a 0.18–0.25 V discrimination voltage range were used. The International Center for Diffraction Data (ICDD) diffraction database was used to identify the phases. Figure 3 shows the X-ray diffraction pattern of the metal oxide particle deposited in volume 8, with angle (2θ) on the x-axis and

intensity on the y-axis. Complete matching of the observed diffraction peaks for ZnO and absence of peaks for Zn indicated that the metal particles were completely oxidized and converted into metal oxide aerosol during the generation process.

4.2 Temperature profiles

The temperature of the plasma torch's carrier gas is determined by the plasma torch's operating power: the greater the operating power, the higher the carrier gas temperature. The operating power of PTAG used in this experiment was 25 kW. Figure 4 shows the temperature profile of the test sections during the depositional phase. Thermophoretic deposition is driven by the thermal gradient between the carrier gas and the inner wall of the test sections, so the wall temperature and carrier gas temperature were measured in each test section and are shown in Figures 4a,b, respectively. The continuous recirculation of cold water through the copper-wrapped tube significantly reduced the temperature of volume 0. Through threading, the volume 1 pipe was assembled to volume 0. The optimum heat transmission by conduction was possible with this metal-to-metal link. As the wall temperature of volume 0 was low, so the temperature of the wall of volume 1 was also low, but the convection mode of heat transfer from high-temperature carrier gas significantly increased the temperature of the volume 1 wall. In the experiment, it was observed that the wall temperature of volume 0 was significantly lower than the wall temperature of volume 1. Furthermore, volumes 2 and 1 were assembled through the thread, allowing optimum conduction heat transfer between the wall. The temperature difference between the carrier gas and the wall caused heat dissipation

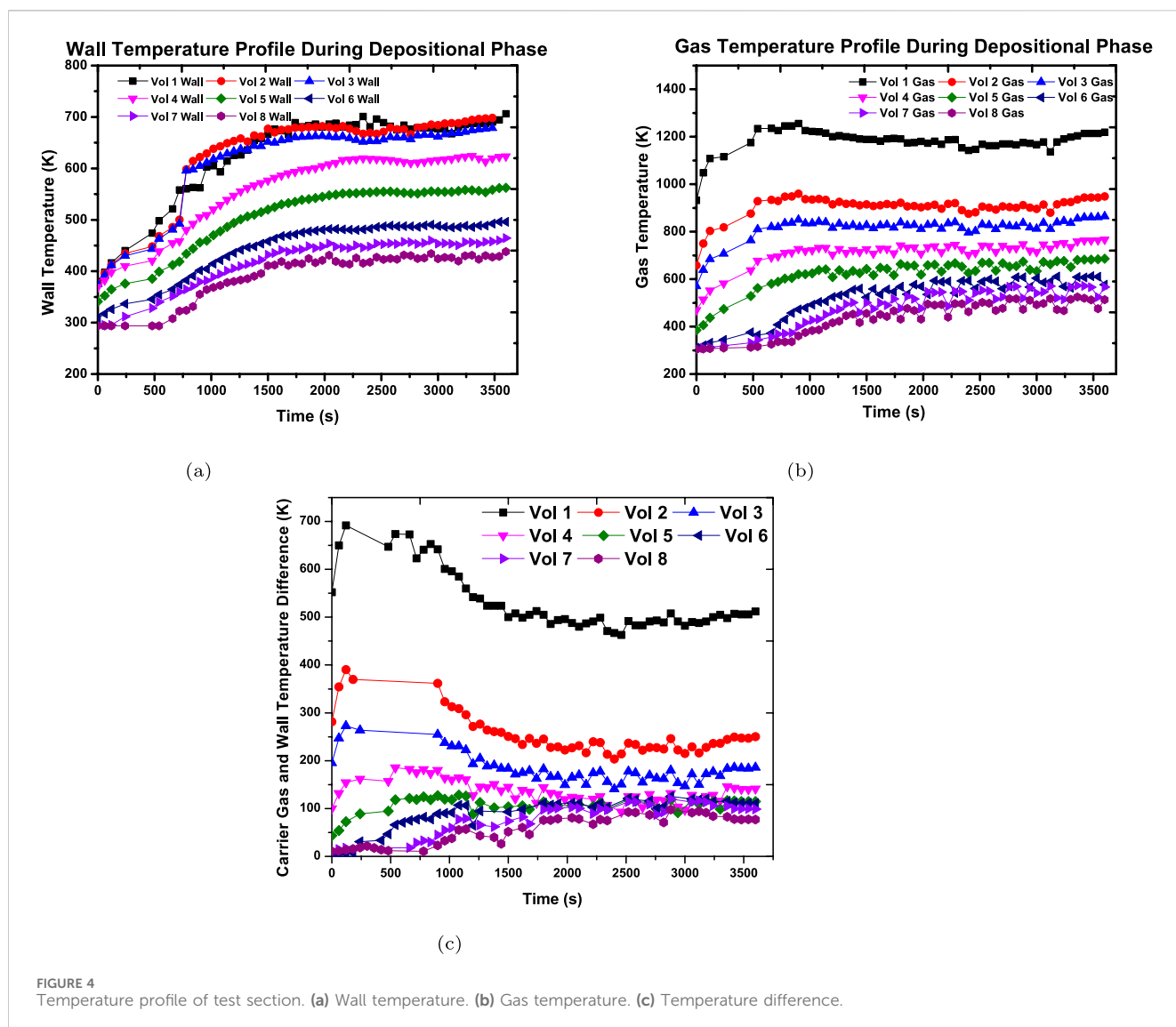


FIGURE 4
Temperature profile of test section. (a) Wall temperature. (b) Gas temperature. (c) Temperature difference.

which further reduced the temperature of the carrier gas in volume 2. Thus, as shown in Figures 4a,b, the wall temperature of volumes 1 and 2 was almost the same, but the carrier gas temperature of volume 2 was significantly lower than that of volume 1. As flow moved downstream of volume 2, the wall temperature of the downstream volumes reduced significantly with the reduction in carrier gas temperature due to heat dissipation. The abrupt increase in wall temperature became gradual after 750 s from the start of the experiment, while stable gas temperature was observed after 500 s. Figure 4c shows the temperature difference between the wall and the carrier gas, with the temperature difference signifying thermophoretic deposition (Section 4.3).

4.3 Carrier gas properties

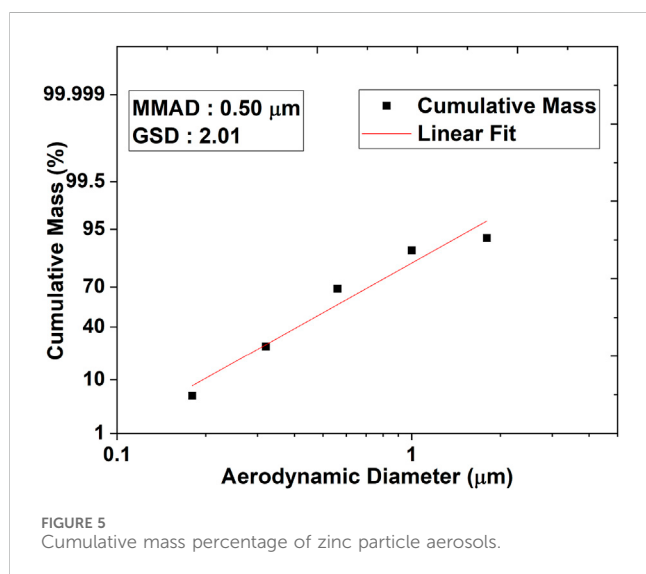
Carrier gas velocity was kept low during the depositional phase to sustain a laminar flow as the resuspension of deposited aerosol particles is negligible in a laminar flow field (Parozzi et al., 1995). Table 2 shows the carrier gas velocity and Reynolds number

calculated by ASTEC at the end of the depositional phase. As discussed in the previous section, gas temperature reduced from volume 1 to volume 8. Carrier gas obeys the perfect gas law; thus a reduction in gas temperature dominated the pressure drop phenomenon in the constant cross section pipe, ultimately causing a rise in gas density. For a pipe with a constant cross section, the law of mass conservation states that with no mass addition or subtraction, velocity is inversely proportional to the density of gas. In this experiment, the rise in density led to a reduction in the velocity of the carrier gas. Moreover, the temperature drop indicated a rise in viscosity, as per the kinetic theory of ideal gases. The velocity effect was dominated by the combination of density and viscosity; because of this, the Reynolds number went up on moving down the axial length of the test section, from volume 1 to volume 8.

Since resuspension begins to at a high gas velocity during a turbulent flow, the inlet velocity of carrier gas is critically important. As mentioned in Parozzi et al. (1995), gas mass flow rate increases to more than 30 kg/s as a result of the core fall into the bottom plenum of the vessel, corresponding to hot leg and surge line velocities of

TABLE 2 Carrier gas properties during depositional phase.

Volume no.	1	2	3	4	5	6	7	8
Gas velocity (<i>m/s</i>)	1.81	1.40	1.28	1.13	1.00	0.90	0.83	0.75
Gas density (<i>kg/m³</i>)	0.30	0.39	0.43	0.49	0.55	0.62	0.68	0.74
Gas viscosity (<i>kg/m/s</i>) $\times 10^{-5}$	4.70	4.09	3.85	3.55	3.28	3.03	2.85	2.66
Reynolds no.	886.2	1022	1087	1183	1284	1397	1494	1599



over 3 m/s and 20 m/s, respectively. For this experiment during the resuspension phase, the inlet velocity of carrier gas was set at 4.58 m/s (1265 $Lmin^{-1}$), corresponding to a Reynolds number of 23,510.

4.4 Aerosol characteristics during deposition and resuspension

4.4.1 Depositional phase

Figure 5 shows the cumulative mass percentage of the zinc oxide aerosol particles deposited at different stages of the MOUDI measured at the inlet of the test section during the deposition phase. The MMAD was found to be 0.5 μm and a GSD of 2.01. Table 3 and Figure 6 compare the mass of test particles deposited in all test sections obtained from experimental observations and from ASTEC simulations during the depositional phase. ASTEC's SOPHAEROS module, which deals with the transport and retention of fission products in a reactor coolant system, was used in stand-alone mode for these simulations (Cousin et al., 2008).

The initial aerosol size distribution was characterized by an MMAD of 0.5 μm and GSD of 2.01. However, in the downstream volumes, particularly in volume 8, the computed MMAD increased to approximately 0.9 μm . This shift is attributable to coagulation, where particle-particle collisions result in larger aggregates. SOPHAEROS dynamically accounts for such effects, and the size-dependent deposition processes (gravitational settling, Brownian diffusion, and thermophoresis) responded accordingly.

The maximum difference in deposited masses of 12.86% was observed in volume 8 of the test section, followed by volume 3 (10.2%), with the rest of the volumes having differences of less than 10%. For complete deposition in the test section, the difference between the experimental value and the SOPHAEROS module predictions was 3.35%. Both the computational and experimental data indicate a declining pattern in the deposited aerosol mass over the length of the test section. However, the trend changed from experimental overestimation to underestimation when moving from volume 1 to rest of the volumes. This could be due to the constraint of code to accurately use the actual developed parabolic velocity profile as the input conditions for different volumes. The contribution of various depositional mechanisms in each volume of the test section as estimated by the ASTEC's SOPHAEROS module during the depositional phase is shown in Figure 7. The thermophoretic deposition is clearly the dominant depositional phenomenon as an effect of a high temperature gradient between the carrier gas and wall temperature of the test section. Figure 4c shows that the temperature gradient decreases as it moves downstream of the test section, resulting in the reduction of the thermophoretic deposition of the aerosols. Gravitational settling is the second-most important phenomenon, and it increases along the downstream of test section. Downstream, the temperature reduces, leading to a decrease in kinematic viscosity of the carrier gas which enhances the coagulation (Equation 10) that directly leads to the enhancement of gravitational settling. Laminar diffusion makes a much smaller contribution; since the flow was laminar, there was no deposition due to turbulent diffusion and eddy impaction.

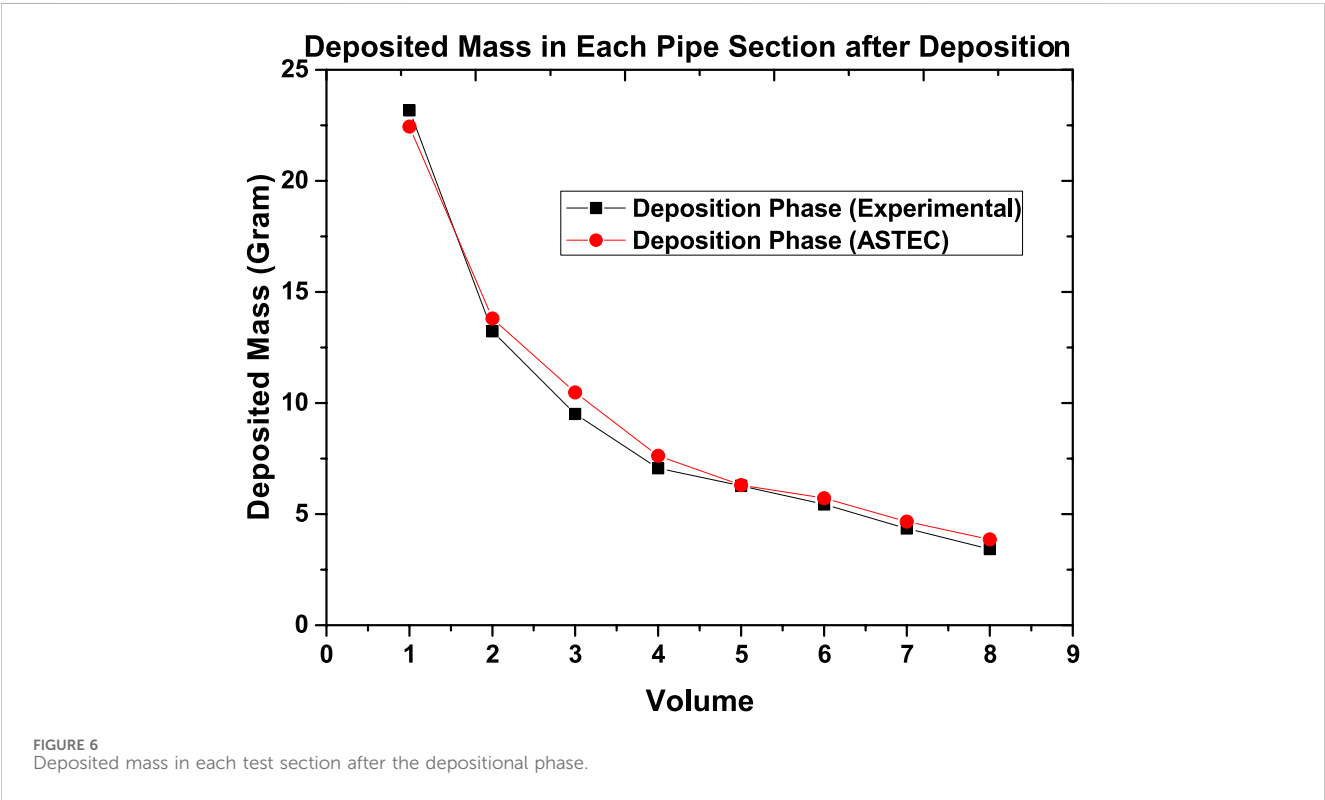
4.4.2 Resuspension phase

The mass of the aerosols deposited in each volume of the test section, both experimentally and as computed by ASTEC after the resuspension phase, is shown in Table 4. During the resuspension phase, 46.51% of the total deposited aerosol mass in the test sections were resuspended under the given turbulent flow conditions (Figure 8). Both the computed and experimental results show a decreasing trend in the deposited mass after resuspension. Volume 4 has the highest difference of 14.38%. The overall difference, defined as the difference in total deposited mass across the entire test section after resuspension, is 5.6%.

In overcoming the adhesive forces of the deposited aerosols, the carrier gas lost its kinetic energy while stripping the deposited aerosol from the surface of volume 1. Furthermore, the kinetic energy of the carrier gas decreased as it transported the resuspended aerosols, resulting in a decrease in carrier gas velocity. As the carrier gas flowed downstream, the stripping effect diminished, resulting in a reduction in the mass of resuspension of deposited aerosol. This

TABLE 3 Mass deposited after the depositional phase.

Volume no.	Deposited mass In grams (Experimental)	Deposited mass In grams (ASTEC)	Percentage difference
1	23.18	22.44	3.12
2	13.23	13.81	4.38
3	9.51	10.48	10.20
4	7.06	7.62	7.93
5	6.28	6.31	0.48
6	5.43	5.72	5.34
7	4.36	4.66	6.88
8	3.42	3.86	12.86
Total	72.47	74.90	3.35



diminishing striping effect caused a larger mass of deposited aerosol to remain deposited after volume 1, as shown in 8.

ASTEC underpredicted the deposited aerosol mass after resuspension (overpredicting the mass of resuspended aerosols) in all volumes of the test section except volume 1 (Figure 9). The carrier gas was fed directly into test volume 1 through seven separate tubes. As already mentioned, ASTEC is a one-dimensional code, so the source of difference can be the absence of an actual velocity profile as the initial condition.

Figure 10 shows the ratio of resuspended aerosol mass to that of deposited aerosol mass. It can be clearly observed that the

mass of resuspended aerosol decreased from volumes 1 to 8 in the experiment as well as computation. The trends of the ratio are clearly in good agreement. However, there are slight quantitative differences between the ratio of mass obtained from the experiment and the ASTEC code. This may be because of the input of the actual velocity profile during the resuspension phase. Figure 10 shows that the mass of the resuspended aerosol is more than that of the deposited aerosol at the beginning of the test section. On moving downstream along the test section, these resuspended aerosol masses decrease and become less than that of the deposited mass

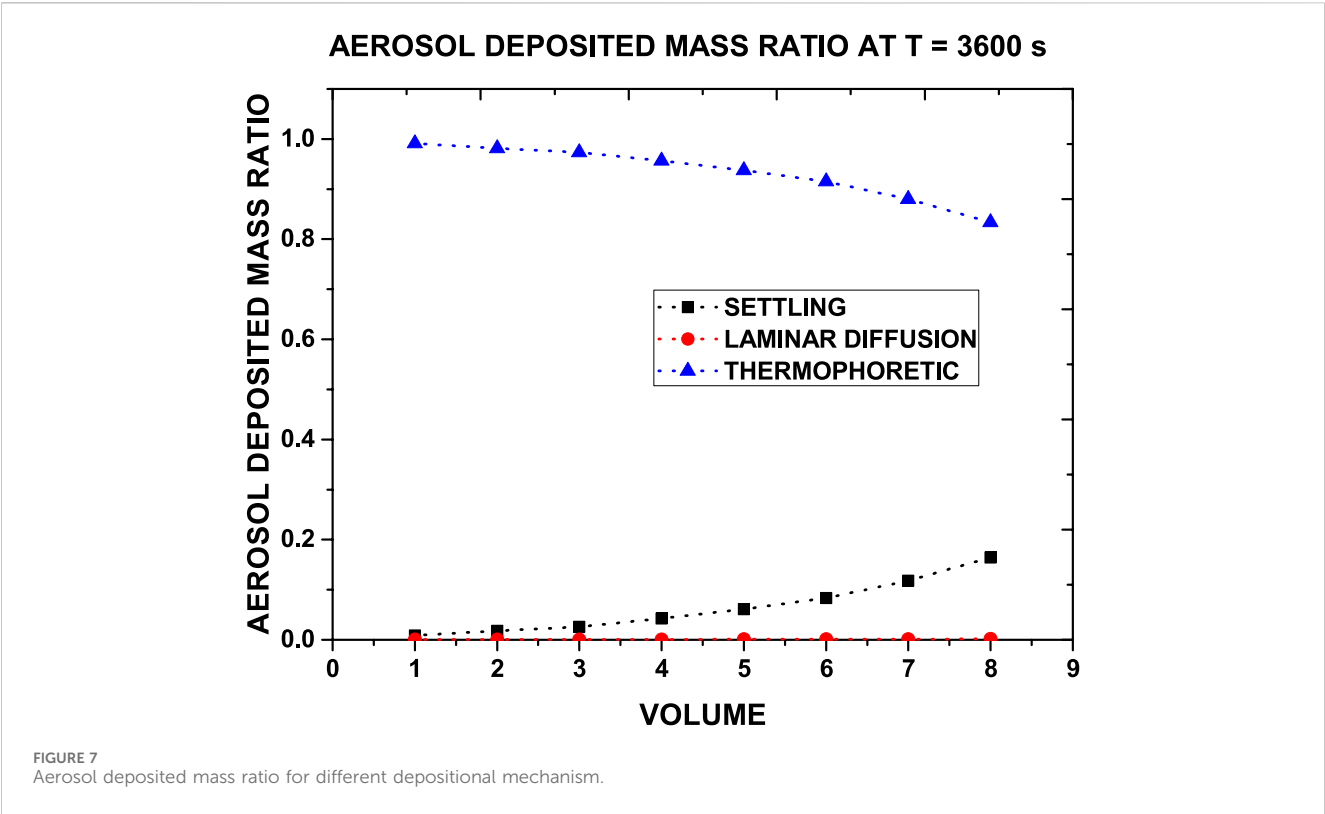


TABLE 4 Mass deposited after the resuspension phase.

Volume no.	Deposited mass In <i>grams</i> (Experimental)	Deposited mass In <i>grams</i> (ASTEC)	Percentage difference
1	7.45	8.22	10.33
2	6.32	5.69	9.97
3	5.57	4.84	13.10
4	4.73	4.05	14.38
5	4.07	3.73	8.35
6	3.83	3.63	5.22
7	3.48	3.34	4.02
8	3.31	3.10	6.34
Total	38.76	36.59	5.60

of aerosol. This phenomenon is attributed to loss of kinetic energy of the carrier gases (as mentioned above).

4.5 Evolution of deposited aerosol mass during deposition and resuspension phase

The deposited mass in the test section with respect to time, as estimated by ASTEC, is shown in Figure 11. In Figure 11a, aerosol deposition occurred at a higher rate within 750 s of the experiment. This is consistent with Figure 4, wherein a higher gas-to-wall

temperature gradient is evident within this timeframe. Once the gradient reduced and saturated, the slope of the curve decreased due to lesser thermophoresis deposition and became constant. Within 3,600 s of this experimental phase, the deposited mass increased to 72.47 g. In contrast, the mass of deposited particles reduced constantly during the resuspension phase, becoming 38.76 g from 72.47 g during an experimental duration of 20 minutes as can be seen in Figure 11b. It can thus be concluded that the default force balance model for resuspension in the SOPHAEROS module (Cousin et al., 2008) is highly dependent on the duration of the resuspension process.

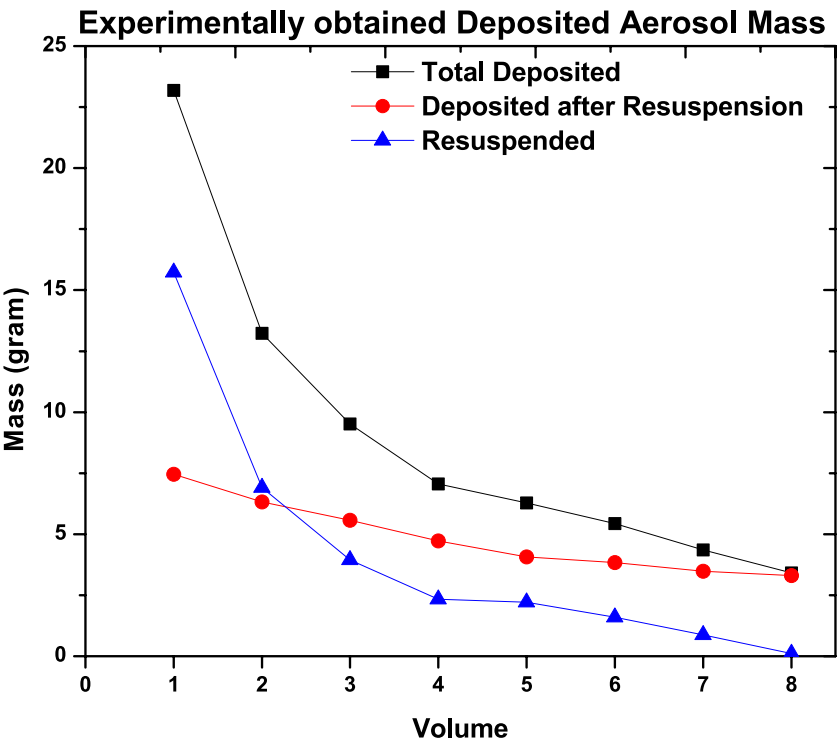


FIGURE 8
Experimentally obtained deposited and resuspend mass of aerosol.

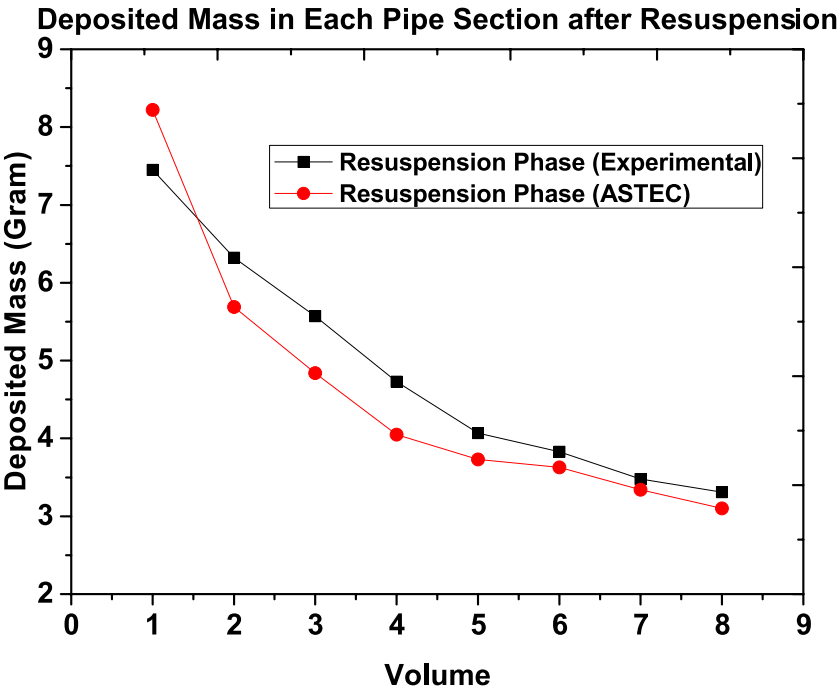


FIGURE 9
Deposited mass in each test section after the resuspension phase.

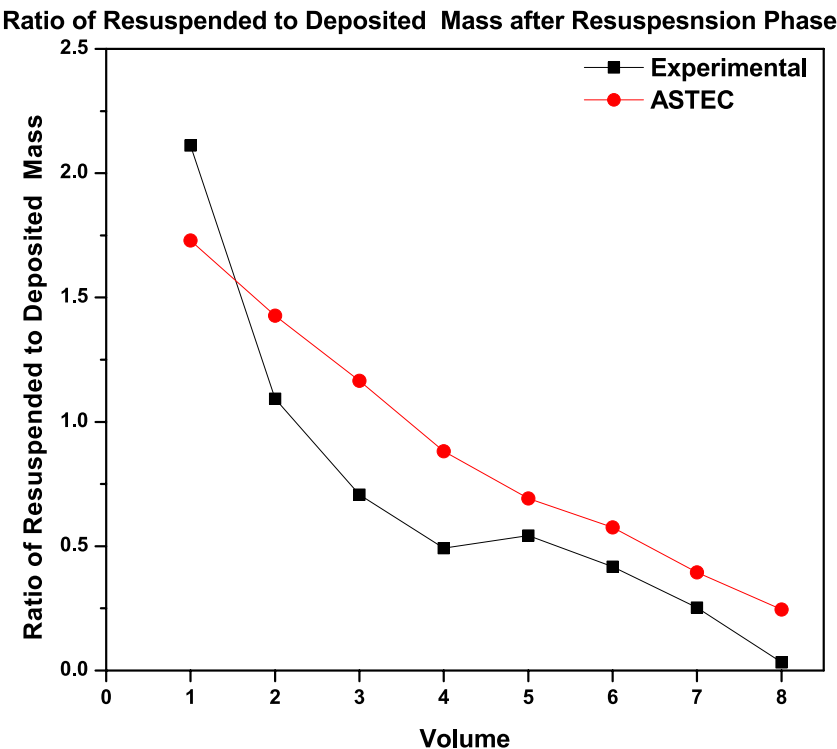
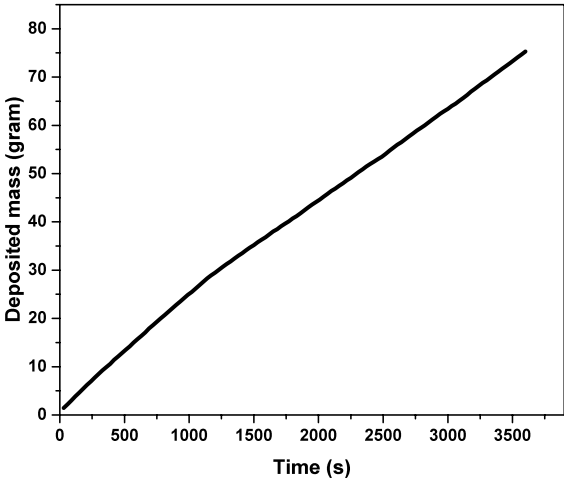


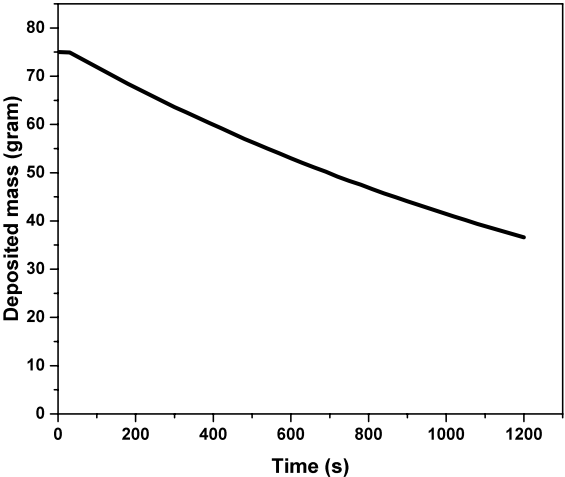
FIGURE 10
Ratio of resuspended to deposited aerosol mass after the resuspension phase.

Time Evolution of Deposited mass during Deposition Phase



(a)

Time Evolution of Deposited mass during Resuspension Phase



(b)

FIGURE 11
Computationally obtained time evolution of deposited aerosol mass. (a) Deposition phase. (b) Resuspension phase.

5 Summary and conclusion

At the National Aerosol Facility of the Indian Institute of Technology, Kanpur, India, an experiment was conducted to

investigate the deposition and resuspension of zinc oxide aerosol under dry environment conditions in a circular cross-section straight stainless steel pipe. The test aerosols were generated using a plasma torch aerosol generator by introducing zinc metal

powder into the plasma flame, wherein aerosolization occurred via the evaporation–condensation technique. The XRD analysis confirmed the formation of zinc oxide particles with no traces of zinc metal powder. The deposition phase of the experiment took place at a high temperature gradient laminar flow, while resuspension took place at a zero thermal gradient turbulent flow. The experimental results are also compared with the computational results from the SOPHAEROS module of the ASTEC code.

During the deposition phase of the experiments, the temperature gradient between the bulk gas and wall temperature was measured in all eight test volumes of the test section. It was seen that the temperature gradient increased initially and then reduced after approximately 750 s, finally saturating to an equilibrium value. Higher temperature gradient results in higher thermophoresis, which was also captured by code simulation and thermophoresis, was found to be the major deposition process. Overall, the experimentally measured reduction of deposited mass in subsequent volumes was seen to match the ASTEC predicted results. In the resuspension phase, the stripping of deposited aerosol and the transportation of resuspended aerosol caused mitigation in the kinetic energy of the carrier gas. The reduction in kinetic energy caused less stripping of deposited aerosol in the downstream test section. However, visual observation from the open end of the test section during the experiment revealed that the majority of the resuspension took place within a few seconds of the start of the resuspension process. Thus, it can be concluded that the default force balance model for resuspension in the SOPHAEROS module is highly dependent on the duration of the resuspension process. The trends obtained experimentally and computationally for the deposition and resuspension phases are in good agreement. However, quantitatively, the ASTEC result shows slight deviation from the experimental results. The main reason for these deviations could be due to the 1D-coded ASTEC limiting the use of the actual flow profile as the initial condition of the simulation. These results provide a good foundation for large-scale experiments and the development of severe accident simulation codes.

This study provides insights into aerosol behavior in nuclear-relevant pipe geometry under thermal gradient-driven deposition and high-flow resuspension. The experimental results serve not only to elucidate the dominant transport and interaction mechanisms but also to support the evaluation of SOPHAEROS predictions. While the model performs well in capturing overall trends when supplied with accurate thermal and velocity inputs, further development and validation, especially in resolving 3D effects, remain areas for future research.

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Data availability statement

The original contributions presented in the study are included in the article/supplementary material, further inquiries can be directed to the corresponding author.

Author contributions

MK: Writing – original draft, Writing – review and editing. MJ: Writing – review and editing. AD: Writing – review and editing. SK: Writing – review and editing. TS: Writing – review and editing. AK: Writing – review and editing. GM: Writing – review and editing. NS: Writing – review and editing. SG: Writing – review and editing. ST: Writing – review and editing. BS: Writing – review and editing.

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Conflict of interest

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