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Functional foods have gained popularity in recent decades. They are exploited for their bioactive compounds like polyphenols, which are highly demanded in cosmetic, pharmaceutical and nutraceutical industries. However, extractive techniques and conditions used up to recently are almost obsolete and must be optimized for higher efficiency. The current study aimed to evaluate the antidiabetic potential of an optimized extract of Abelmoschus esculentus (okra) seeds. The optimal conditions for extracting polyphenolic compounds from okra seeds were determined using Microwave Assisted Extraction (MAE). A Face Center Composite Design (FCCD) was used for optimization. Solvent/dry matter ratio, wavelength and time were considered while the response studied was the polyphenolic content. The extract obtained at optimal conditions was characterized using Thin Layer Chromatography (TLC) and Fourier Transform Infra-Red (FTIR) spectroscopy, then tested for its antioxidant, alpha amylase inhibitory and antidiabetic activities. Response Surface Methodology (RSM) permitted the determination of the optimal conditions for phenols extraction as: microwave power 330 W, with a solvent ratio of 97.04/1 mL/g for 9.5 min of extraction time. The optimized extract showed a phenolic content up to $86.37 \pm 1.13 \,\mathrm{mg}$ GAE/g containing quercetin and catechin as revealed by the TLC. Functional groups characteristic of polyphenols were identified on FTIR spectra, and the extract exhibited good in vitro antioxidant capacities with DPPH (2, 2-diphenyl-1-picrylhydrazyl) radical scavenging capacity and FRAP (Ferric Reducing Antioxidant Power Assay). An IC $_{50}$ of 3.99 \pm 0.15 $\mu g/mL$ was obtained with the DPPH scavenging test. Alpha amylase inhibitory assay revealed that the optimized okra extract behaved as a non-competitive inhibitor of porcine pancreatic amylase with an IC₅₀ of 484.17 \pm 2.33 μ g/mL.

Antidiabetic activity of the extract was observed in streptozotocin-induced diabetic males Wistar rats, as shown by the fasting blood glucose levels, food intake, changes in body weight and serum lipid profile among others.

KEYWORDS

optimization, phenol, diabetes mellitus, okra, microwave assisted extraction

Introduction

Considerable adverse side effects of oral antidiabetics used up to date have reinforced the interest of scientists and patients in functional foods and their derivatives for the management of many chronic diseases (1). Abelmoschus esculentus (okra) fruits, for example, have long been investigated for their antidiabetic potential. Seeds, peels or the whole fruits have shown their efficacy in reducing blood glucose levels in experimental animals. Authors mostly attributed the antidiabetic activity of okra fruits and other plants to their polyphenolic content (2). However, the extraction techniques that had been used seem rudimentary compared to what is done nowadays (3). Content of bioactive compounds obtained by traditional extraction methods such as distillation, maceration and solvent extraction among others, used by the previous researchers can be highly improved by cutting-edge techniques like pressurized hot water (PHWE), microwave-assisted (MAE) and ultrasoundassisted (UAE) extractions, thus allowing a potentially great amelioration in the antidiabetic activity of okra fruits or parts. Also the extraction time is significantly reduced by these new techniques (4, 5). Meanwhile each extraction technique has its own advantages and limitations; MAE has demonstrated its high efficiency for a fast extraction of good quality bioactive compounds from natural sources when the solvent is well chosen (6). Okra seeds have been reported as the richest part of the fruits in polyphenols and flavonoids in general, and also in demonstrated antidiabetic compounds such as quercetin and its derivatives in particular (3).

Flavonoids, just like other phenolic compounds are also known to have antioxidants, anti-inflammatory, anti-cancer activities among others (7) and are highly demanded nowadays by cosmetic, pharmaceutical and nutraceutical industries. So, determination of experimental conditions for maximum extraction of polyphenolic compounds from okra seeds using a more efficient technique is quite urgent in order to reduce pollution, solvent and time wastage and above all improve the extraction yield, the quality and the bioactivity of the extracts (8–10). Geng et al. (11) have determined the optimal conditions for extracting polyphenols from okra flowers using conventional maceration technique while Amirabbasi et al. (6) established the conditions for a maximal extraction of polyphenols from okra stems using MAE and UAE. This study aimed to determine the optimal conditions for extraction of polyphenols from okra seeds by a face centered composite design, using MAE and to evaluate the antidiabetic activity of the extract obtained.

Materials and methods

Material

Plant material

Mature fruits of *Abelmoschus esculentus* were harvested from our botanical garden situated in Dschang ($5^{\circ}27^{/}$ nord, 10° 04/ east) West region of Cameroon. The fruits were shade dried and the pods opened for seed collection.

Chemicals

All reagents were purchased from local stores while streptozotocin was provided by Sigma-Aldrich (Hamburg; Germany).

Experimental animals

To evaluate the antidiabetic activity of the optimized extract, twenty four 24 Wistar rats were collected from the animal house of the department of Biochemistry (University of Dschang) and placed in the animal room of the research unit. For 1 week, rats were fed a normal diet and water for acclimatization. All the animals (after dividing into groups) were housed in individual cages in the animal room under optimal conditions of light (12/12 light and dark cycle), temperature ($27 \pm 2^{\circ}$ C), relative humidity ($60 \pm 10\%$) and a pathogen-free surrounding.

Method

Preparation of sample

Seeds of *Abelmoschus esculentus* were dried in an oven at 45° C until constant weight (24 h), and then finely ground using an electrical grinder (royalty line, 800 W, five cycles of 1 min each at full power). Powders were sieved (using a 500 μ m sieve) and immediately used for extraction of phenolic compounds.

Choosing factors affecting the phenolic content

Based on the literature, the following factors were considered: time of extraction, dry matter/solvent ratio and the power of the microwave oven. A 50:50 hydroethanolic solution was used as solvent system, based on previous works (12, 13). Also, literature indicates that the solvent's dielectric properties should be highly considered when planning to extract phenolic compounds using MAE. Compared to water, ethanol or its mixtures with water have a lower dielectric constant, and are more transparent to microwave, thus not well converting microwaves into heat, but have high capacity to dissolve and extract phenolic compounds (14). Preliminary studies permitted to determine the ranges used for each factors.

Determination of responses

For each trial, 1 g of seeds powder was mixed with the appropriate amount of solvent according to the experimental conditions as given by the chosen design. The mixture was stirred using a magnetic agitator, afterward, it was allowed to rest for 10 min at room temperature and put in a microwave oven (SAMSUNG M735) for extraction, under specified conditions. Samples were centrifuged (4000 rpm /5 min) and the supernatant was collected after filtration through Whatman paper n°1. Solvent was then evaporated in an air oven at 45°C until obtention of the dry extract. Dry extracts were immediately used for determination of polyphenolic content. Figure 1 depicts the global flowchart of the work.

Determination of total phenolic content

The total phenolic content was assessed according to the protocol described by (15). Briefly, 0.2 mL of Folin reagent (ten-fold diluted) was added to a tube containing 0.01 mL of plant extract (5 mg/mL) and 1.39 mL of distilled water. The mixture was allowed to stand for 3 min before addition of 0.4 mL of sodium carbonate (20% w/v), and then mixed using a vortex. The tube was then incubated at 40°C for 20 min in a water bath and absorbance was read at 760 nm against a blank using a BIOMATE spectrophotometer. Gallic acid (0.2 g/l) was used to draw a calibration curve. All experiments were carried out in triplicates and results were expressed as mg of gallic acid equivalent (GAE) per g of dry extract (mg GAE/g dry weight).

Optimization of the responses using the central composite design

A face Centered composite design was used to optimize the response, total phenolic content (Y_1) . Ranges different factors of were taken based on literature (6). Experiments randomized and were



responses evaluated in triplicates. The proposed model was:

$$Y = a_0 + a_1X_1 + a_2X_2 + a_3X_3 + a_{11}X_1^2 + a_{22}X_2^2 + a_{33}X_3^2 + a_{12}X_1X_2 + a_{13}X_1X_3 + a_{23}X_2X_3$$

Where Y is the response (phenolic content), X_1 , X_2 , X_3 are the studied factors, a_0 is the offset term while a_1 , a_2 , a_3 are linear effects, a_{11} , a_{22} , a_{33} the quadratic effects and a_{12} , a_{13} , a_{14} , a_{23} , a_{34} are interaction effects. Table 1 shows the experimental design in coded and real variables.

Characterization of the extract

Determination of the total flavonoid content

The Total Flavonoid Content (TFC) was obtained using the method described by (16). Sodium nitrite of 0.03 mL (5%) was added to a tube containing 1.49 mL of water and 0.1 mLof extract solution (5 g/mL). After 5 min, a volume (0.003 mL) of aluminum chloride (10%) was added to the tube and the mixture was allowed to rest for 6 min. Afterward, 0.3 mL of NaOH (1M), and 0.24 mL of distilled water were introduced into

TABLE 1	Experimental	design ir	coded an	d real variables.
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Trial	Matrix of real and coded variables					
	Time (min)	Solvent (mL/g)	Power (W)			
1	4.00 (-1)	30.00 (-1)	180.00 (-1)			
2	15.00 (+1)	30.00 (-1)	180.00 (-1)			
3	4.00 (-1)	80.00 (+1)	180.00 (-1)			
4	15.00 (+1)	80.00 (+1)	180.00 (-1)			
5	4.00 (-1)	30.00 (-1)	480.00 (+1)			
6	15.00 (+1)	30.00 (-1)	480.00 (+1)			
7	4.00 (-1)	80.00 (+1)	480.00 (+1)			
8	15.00 (+1)	80.00 (+1)	480.00 (+1)			
9	0.25 (-1.68)	55.00(0)	330.00(0)			
10	18.74 (+1.68)	55.00(0)	330.00(0)			
11	9.50(0)	12.95 (-1.68)	330.00(0)			
12	9.50(0)	97.04 (+1.68)	330.00(0)			
13	9.50(0)	55.00(0)	77.73 (-1.68)			
14	9.50(0)	55.00(0)	582.26 (+1.68)			
15	9.50(0)	55.00(0)	330.00(0)			
16	9.50(0)	55.00(0)	330.00(0)			
17	9.50(0)	55.00(0)	330.00(0)			
18	9.50(0)	55.00(0)	330.00(0)			
19	9.50(0)	55.00(0)	330.00(0)			
20	9.50(0)	55.00(0)	330.00(0)			

Bold values are replicates of the center points.

the tube and mixed with a vortex before absorbance was read at 510 nm against a blank. The calibration curved was made using catechin. All experiments were made in triplicates, and results were expressed as mg of catechin equivalent per g of dry extract (mg CE/g of dry weight).

Determination of crude fiber content

Crude fibers were quantified using the Ceramic Fiber Filter method as described by (17). These extracts and powders were previously treated to remove lipids using hexane (24 h soaking of 6 g of extracts and powders in 30 mL of hexane with gentle stirring). Briefly, 100 mL of 1.25% H₂SO₄ was added to 1 g of lipid-free powder in a round bottom flask and the mixture boiled under reflux for 30 min. The hot solution was quickly filtered under suction and the insoluble matter washed several times with hot distilled water until it was acid free. It was quantitatively transferred into the flask and 100 mL of hot 1.25% sodium hydroxide (NaOH) solution was added and the mixture boiled again under reflux for 30 min before it was quickly filtered under suction. The soluble residue was washed with boiling water until it was base free. Afterwards, it was dried to constant weight in the oven at 105°C, cooled in a desiccator and weighed. The weighed sample (C1) was incinerated in a muffle furnace at 300°C for about 2 h, cooled in the desiccator and weight measurement repeated (C2). The loss in weight of sample on incineration was given by C1–C2 while the crude fiber content was expressed as follows:

% Crude fiber =
$$\frac{C1 - C2}{Weight of original sample} \times 100$$

Saponin content

The saponin content was estimated as previously described by Koziol (18). Briefly, 0.5 g of the formulation was introduced in a graduated test tube, and 5 mL of distilled water was added.

The tube was closed and vigourously shaken for 30s, and the foam height was immediately measured. The saponin content is linked to the foam height by the following formula:

 $Saponin(mg) = \frac{[(0,432)(\text{Foam height after 5 -10 s}) + 0,008]}{\text{Sample weight (g)}}$

Zinc content

Zinc content was determined using the protocol established by Pauwels et al. (19). The sample (1 g) was carbonized at 450 $^{\circ}$ C for 2 h in an oven (Carbolite Eurotherm), then digested with 10 mL of nitric acid 1N during 30 min. after cooling down, the solution was filtered using Whatman N°1 filter paper in a 50 mL flat bottom flask. Distilled water is added to the filtrate to bring the final volume up to 50 mL. Filtrate (20 mL) + NH₄Cl (20 mL) + concentrated HCl (1 mL) + sodium sulfite (1 drop) + potassium ferrocyanate (1 mL; 0.5%) were mixed and the mixture allowed to rest for 5 min in dark before absorbance were read using UV-visible spectrophotometer at 650 nm.

Identification of chemical functions and the number of bioactive compounds

Fourier Transformed Infra-Red Spectroscopy (FTIR) was used to identify the main chemical groups present in the extract while TLC was used to estimates the number of bioactive compounds or groups of compounds present in the extract to identify some.

Fourier Transformed Infra-Red Spectroscopy

Spectra were collected at 4000–400 cm⁻¹ using a FT-IR Spectrometer (Alpha, Bruker, Germany) on a diamond plate at 4 cm⁻¹. Two replicates spectra of 50 scans each were recorded. Raw spectra were corrected.

Thin layer chromatography (TLC)

TLC was performed using a pre-coated plate with 60F250 silica gel (MERCK). Two standards were used: catechin and

quercetin (1 mg dissolved in 50 ml of ethanol, centrifuged, and the supernatant used). Development was done for 20 min in a pre-saturated (30 min) rectangular development chamber. The mobile phase was made of ethyl acetate/formic acid/ glacial acetic acid/water. The plate was dried at 45 $^{\circ}$ C in air oven and visualized under UV light (254 nm). Bands were circled and Rf calculated.

Antioxidant activities

Ferric reducing ability of plasma (FRAP)

Ability of extracts and formulation to reduce ferric iron was tested as described by Oyaizu (20).

Briefly, 75 μ L of extract/powder suspension was added to 2 mL of FRAP reagent (300 mM acetate buffer: pH 3.6; 10 mM TPTZ [(2, 4, 6-tris (2-pyridyl)-S-triazine)] in 400 mM of HCl; 10 mM ferric chloride). After 30 min of incubation at room temperature, the absorbance was read at 593 nm against a blank. BHT was used as standard.

DPPH inhibitory activity

DPPH inhibitory activity of the extract was assessed using a previously described method (21). Briefly, 100 mL of ethanol was introduced in tubes a_2-d_2 , a_3-d_3 , a_4-d_4 , a_5-d_5 , and a_6-d_6 . Afterward, the extract was added in tubes a_1-d_1 and a_2-d_2 . Dilutions were made starting from tubes a_2-d_2 to have final concentrations of 200; 100; 50; 25; 12,5 et 6,25 µg/mL. Finally, 900 µL of DPPH solution was added in tubes b_1-b_6 , c_1-c_6 et d_1-d_6 while the same volume of ethanol was added in tubes a_1-a_6 . The mixtures were allowed to stand for 30 min in dark before absorbance were read at 517 nm against blanks prepared in the same conditions (a_1-d_1).



DPPH inhibitory activities (%) were calculated using the formula:

$$\% = \frac{\text{DOc} - \text{DOt}}{\text{DOc}} \ge 100$$

Where Doc = absorbance of the control; Dot = absorbance of the test.

Values of IC_{50} (concentration of product inhibiting 50% of DPPH^o) were determined using percentages of antioxidant activities and were expressed in μ g/mL.

Alpha amylase inhibitory activity and mechanism of action

The alpha amylase inhibitory activity was determined as described by (22) with slight modifications. Decimal dilutions of the extract were made ranging from $12.5 \,\mu$ g/mL to $200 \,\mu$ g/mL. To $20 \,\mu$ L of extract, $20 \,\mu$ L of porcine pancreatic amylase (0.5 mg/mL) prepared in phosphate buffer (0.02 M, pH 6.9) was added. The mixture was pre-incubated at 25 °C for 10 min before introduction of 20 μ L of freshly prepared starch (1% w/v in distilled water). Tubes were incubated at 25° C for 10 min, afterward, 40 μ L of DNS was added and the tubes boiled for 15 min to stop the reaction and quantify reducing sugars. 600 μ L of distilled water was added to the tubes before absorbance were read at 540 nm. The control was made up of the same constituents with the extract replaced by the buffer.

Percentages of inhibition of each concentration were calculated as follow:

% inhibition =
$$\frac{DOc - DOt}{DOc} *100$$

Where D0c = absorbance of control and DOt = absorbance of the test.

The mechanism of alpha amylase inhibition was assessed using the same procedure as described before. The formulation was used at a concentration of $100 \,\mu$ g/mL with different concentrations of substrate (1.25, 2.5, 5 and 10 mg/mL). 1/V =f (1/S) (where V = velocity and S = substrate concentration) graph was plotted for determination of the mode of inhibition.

Acute toxicity

The acute toxicity of the extract was evaluated as per recommendations of OCDE, on evaluation of the acute toxicity of chemical products (23). Two groups of six female rats each, aged 8-12 weeks and weighing between 140 and 180 g, fed on normal chow and received tap water *ad libitum*. The treated group received a single dose of the okra seed extract (5000 mg /Kg of body weight) by oral gavage, while the control received a vehicle at the same dose. Animals were attentively observed for 2 h following administration of the extract, and after each six (06) h during the first day before a daily observation for14 days. Animals were sacrificed and theirs organs collected, observed, weighed and compared to those of the control.

Oral glucose tolerance test (OGTT)

After an overnight fasting (8 h), animals were given water or the extract at 200 mg / kg of body weight before receiving (5 min after) a D-glucose solution (2 g / kg of body weight). The blood glucose(expressed in mg/dL) was then measured(5-10 μ L from tail tip)after 0 min (T₀), 15 min (T₁), 30 min (T₂), 60 min (T₃), 90 min (T₄) and 120 min (T₅) using a portable glucometer (Accu-Chek).

Antidiabetic effect in high fat high sucrose + streptozotocin induced diabetes

Induction of diabetes

Human type 2 close diabetes mellitus was induced on obese albinos Wistar rats by single intra peritoneal administration of 40 mg/Kg body weight of a freshly prepared streptozotocin solution. Streptozotocin was prepared in citrate buffer (0.1 M, pH 4.5). After streptozotocin administration, blood obtained from tail puncture was used to assess fasting blood glucose (FBG) of the animals 3 days later and those with a FBG \geq 1.26 dg/dL were considered diabetic and used for the experiments.

Rats were divided into 4 groups of six rats each as follow:

T-: Normal control (healthy rats).

T+: Diabetic control (untreated obese and diabetic rats). Okra: Obese and diabetic rats treated with okra seeds' extract at 250 mg/Kg of body weight once per day. Met: Obese and diabetic rats treated with metformin at 0.25 mg/Kg body weight once per day.

Fasting blood glucose, food intake and body weight

Blood was collected by cardiac puncture for estimation of plasma biomarkers, while samples of key organs like liver, kidneys, heart, pancreas and lungs were collected for evaluation of specific markers (ALT, AST and ALP) and estimated oxidative stress markers at their levels.

Blood lipid profile and atherogenic index

Serum Triglyceride (TG), Total cholesterol (T Chol) and HDL cholesterol (HDL Chol) was determined according to the procedure describe on the commercial kit used (MONLAB). LDL cholesterol was estimated using the equation of Friedewald et al. (24):

n = 2, if results are expressed in mmol/L and n = 5, when results are expressed in mg/dl. The atherogenic index was calculated as follow:

 $A the rogenic index = \frac{Total cholesterol}{HDL cholesterol}$

Renal, hepatotoxicity and oxidative stress markers

Renal and hepatoprotective activity of the okra seeds extract was assessed by evaluating key biomarkers. Alanine Amino-Transferase (ALT), Aspartate Amino-Transferase (AST) and Alcaline Phosphatase (ALP) were tested to investigate liver functions, while plasmatic and urinary creatine levels were studied to monitor kidneys functions. All measurements were made as described in the commercial kits used (Teco Diagnostics, USA).

Malondialdehyde (MDA) level and reduced glutathione (GSH) were evaluated as oxidative stress markers in the plasma and at the level of the key organs up mentioned. MDA was determined according to the method described by Yagi (25) while GSH was determined as per the method of Ellman (26).

Statistical analysis

Designing and analysis of the results were done using Minitab 18. Experiments were carried out in triplicates. Statistical significance of the variables was determined at 5% probability level. Main effects and contour plots were plotted using Sigma Plot v11.0 (c) Systat. Data on phenol and flavonoid contents, as well as those on biochemical parameters were expressed as mean \pm SD and analyzed by One way Analysis of variance (ANOVA) using SPSS version 22 (IBM). Comparison were made using Bonferroni test at 5% significance.

Results and discussion

Optimization of the extraction of polyphenols from okra seeds using the central composite design

Three factors were studied for extraction of polyphenols from okra seeds, namely the microwave wavelength, the time of extraction and the solvent/dry matter ratio. Ethanol proportion was decided to be 50% based on previous work. Table 2 represents the experimental and predicted response values in different variable conditions given in real and coded values.

Analysis of main effects

The entire experimental plan consisted of 20 trials. The highest polyphenolic content (87.66 \pm 3.33 mg of GAE/g) was obtained at 330 W for 9.5 min of treatment time with 97.04 mL of solvent. The lowest content (45.45 \pm 2.33 mg GAE/g) was observed at 330 W of microwave power with 55.00 mL of extracting solvent and a heating time of 0.25 min. These values are greater than those of Peter et al. (27) and Hu et al. (28) who obtained a total phenolic content of 20.2 and 21.1 mg GAE/g from okra seeds by water and methanol extraction respectively. The highest phenolic content was also greater than what obtained by Geng et al. (11) from okra flowers by ethanol extraction (40.77 \pm 0.83 mg GAE /g material). Such differences could be related to the extraction method. Microwaves induce a quick elevation of the temperature, thus leading to a rapid breakdown of cell walls and liberation of polyphenols out of the matrix (5).

Effect of time

The effect of time on the total phenolic content is illustrated in Figure 2. Increase in the exposition time from 2 to 9.5 min led to an increase in the phenolic content, probably due to the breakdown of cell walls under the heat generated by the microwave, thus leading to a progressive liberation of polyphenols in the solvent system. From 9.5 min, any increase in the extracting time leads to a progressive reduction in the polyphenolic content of the extract obtained. This may be the result of progressive destruction of these thermo-sensitive compounds under long exposure to heat. Previously, the similar effect has also been noticed by Sanja et al. (29) and Xuan et al. (30).

Effect of solvent ratio

An almost linear increase in the phenolic content of the extracts was observed for any increase in the solvent/dry matter ratio (Figure 2). The highest content was obtained with the ratio 97.04 mL/g thus suggesting that high solvent ratio increases mobility of compounds (mass transfer) from plant matrix to the solvent system, as already reported by (31).

Effect of wavelength

It can be seen from Figure 2 that any increase in the extracting power from 77.73 to 330 W induced an increased in the polyphenolic content of the okra seed

TABLE 2 Experimental and predicted responses.

Trials	Ma	Responses			
				TPC (mg	GAE/g)
	Time (min)	Solvent (mL/g)	Power (W)	Exp	Pre
1	4.00 (-1)	30.00 (-1)	180.00 (-1)	57.27 ± 1.09	51.97
2	15.00 (+1)	30.00 (-1)	180.00 (-1)	55.27 ± 2.02	52.77
3	4.00 (-1)	80.00 (+1)	180.00 (-1)	71.31 ± 2.00	69.05
4	15.00 (+1)	80.00 (+1)	180.00 (-1)	64.65 ± 1.60	67.76
5	4.00 (-1)	30.00 (-1)	480.00 (+1)	56.69 ± 3.33	52.95
6	15.00 (+1)	30.00 (-1)	480.00 (+1)	47.65 ± 2.21	49.28
7	4.00 (-1)	80.00 (+1)	480.00 (+1)	58.84 ± 1.09	60.71
8	15.00 (+1)	80.00 (+1)	480.00 (+1)	50.28 ± 1.13	54.95
9	0.25 (-1.68)	55.00 (0)	330.00 (0)	45.45 ± 2.33	50.75
10	18.74 (+1.68)	55.00 (0)	330.00 (0)	51.00 ± 1.59	46.58
11	9.50 (0)	12.95 (-1.68)	330.00 (0)	58.24 ± 0.90	63.83
12	9.50 (0)	97.04 (+1.68)	330.00 (0)	87.66 ± 3.33	82.96
13	9.50 (0)	55.00 (0)	77.73 (-1.68)	53.22 ± 1.09	57.04
14	9.50 (0)	55.00 (0)	582.26 (+1.68)	50.03 ± 3.45	47.09
15	9.50 (0)	55.00 (0)	330.00 (0)	68.36 ± 1.30	68.12
16	9.50 (0)	55.00 (0)	330.00 (0)	68.24 ± 1.90	68.12
17	9.50 (0)	55.00 (0)	330.00 (0)	67.89 ± 2.00	68.12
18	9.50 (0)	55.00 (0)	330.00 (0)	68.54 ± 3.10	68.12
19	9.50 (0)	55.00 (0)	330.00 (0)	68.01 ± 2.89	68.12
20	9.50 (0)	55.00 (0)	330.00 (0)	67.84 ± 1.09	68.12

Exp, experimental; Pre, predicted; Bold values are replicates of the center points.

extracts as a result of more break down of cell walls under the increased heat in the operating system alongside with the power increase, which led to more liberation of polyphenols in the solvent. But any increased in the power of the microwave apparatus above 330 W caused a progressive diminution of the polyphenolic content of the extracts as a consequence of degradation of these compounds exposed to high temperature, since usage of high power in microwave apparatus induce a quick and high elevation of the solvent temperature even when exposure is for a short duration (29).

ANOVA, regression equations for the responses

Table 3 below shows the ANOVA and the influence of each independent factor. Taken individually, solvent/dry matter ratio and the power of the microwave apparatus significantly influenced (p < 0.05) the polyphenolic content of the extracts. The quadratic effect of the time (X₁X₁) and power of extraction (X₃X₃) significantly influenced the total phenolic content of



the extracts, with the greatest contribution in the response (30.86%) for the time, followed by the quadratic effect of the microwave power (22.87%). The mathematical model predicting the effect of the factors on the response is given below (Eq. 1):

$$\begin{split} TPC &= 7.8 + \ 4.75 \ X_1 + \ 0.141 \ X_2 + \ 0.1938 \ X_3 - \ 0.2274 \ X_1 X_1 \\ &+ \ 0.00298 \ X_3 X_3 - \ 0.000252 X_2 X_2 - \ 0.0038 \ X_1 X_2 \\ &- \ 0.00135 \ X_1 X_3 - \ 0.000621 X_2 X_3 \end{split}$$

 $\begin{array}{cccc} TPC: & total & phenolic & content; \\ X_1: & time; & X_2: & power; & X_3: & solvent/dry \\ matter ratio. \end{array}$

Assessment of model quality and optimum conditions

Experimental values showed that these mathematical models can well explain the observed results. According to (32), a good mathematical model should predict at least 75% of the responses; R^2 should then range 0.75 to 1. Based on the determination coefficient for phenols (0.89) given in Table 3, it was concluded that the postulated second-order polynomial equations, truly represented the experimental data. Also, obtaining values of AADM (Analysis of the Absolute Average Deviation) and Bf (Bial factor) respectively equal to 0 and 1 confirmed the suitability of the models since values were in the normal range (0 for AADM and 0.75 < Bf < 1.25 for Bf).

TABLE 3 Evaluation of quadratic model: *P*-value, F value, RC, CF (Contribution Factor) (%), AADM and Bf for phenols.

Source	<i>p</i> -value	F value	CF (%)	
Time (X ₁)	0.347	0.98	1.04	
Solvent (X ₂)	0.001	20.50	21.77	
Power (X ₃)	0.040	5.55	5.89	
$X_1^* X_1$	0.000	31.63	30.86	
$X_2^*X_2$	0.158	2.32	4.22	
$X_3^* X_3$	0.001	21.53	22.87	
$X_1^* X_2$	0.757	0.10	0.11	
$X_1^* X_3$	0.512	0.46	0.49	
$X_2^* X_3$	0.186	2.01	2.14	
Validation of the m	nodel			
R ²		0.89		
AADM		0.00		
Bf		1.00		

Bold, Factors that significantly (p < 0.05) influenced the polyphenolic content.

Optimization of the process

After validation of the model, the optimal extraction conditions for extracting phenols from okra seeds were determined using response surface curves. Figures 3A–C illustrates the variation in the polyphenolic content of okra seed extracts under the influence of different factors taken two by two, drawn with Minitab 18. These figures show that maximum content of phenolic content (87.66 \pm 3.33 mg of GAE/g) is obtained at 330 W, with a solvent ratio of 97.04/1 for 9.5 min.



Confirmation experiments

Replications of the optimum conditions proposed by the model were made in order to confirm the quality of our model to predict the optimal conditions for TPC. No significant difference was noticed between optimal predicted and experimental values obtained, thus confirming the validity of the predicted optimal value given by the software as shown in Table 4. The optimization process permitted to determine the conditions of maximum extraction of polyphenols from okra seeds' as a TPC of 86.77 \pm 1.52 mg GAE/g were obtained, a value two-fold greater than the maximum content of 39.39 \pm 7.46 mg GAE/g obtained by Graham et al. (33) on different okra seeds cultivars.

Characterization of the extract

Phytochemical characterization of the extract

Table 5 show the phytochemical characterization of the extract obtained with the optimal conditions.

Antioxidant activity

Tables 6, 7 show the antioxidant capacities of the okra extract measured by the FRAP and DPPH scavenging methods. The extract performed better than BHT in both tests. It showed a very high DPPH scavenging ability in a concentration dependent manner with an IC₅₀ of $3.99 \pm 0.15 \,\mu$ g/mL, at least 7 times smaller than the minimum value obtained by 32. Optimization permitted to produce high antioxidant extract compared to what have been previously reported (28, 33). The great antioxidant capacity of the extract should be related to its high phenolic content, since polyphenols have been reported to possess antioxidant capacities (33).

Thin layer chromatography

Figure 4 shows the different spots observed on TLC plate. The presence of five (05) spots with Rf ranging from 0.17 to 0.97, led to the conclusion that the optimized extract of okra seeds' contained at least 5 compounds or groups of compounds. Among these compounds, quercetin and catechin were identified with Rf of 0.94 and 0.97 respectively. These observations are in accordance with those reported by Peter et al. (27) and Ong et al. (3) who showed the presence of quercetin and its derivatives in okra seeds' extract. The Rf values of the different spots observed and a tentative identification is given in Table 8.

FTIR spectral analysis

The functional groups of the bioactive compounds present in the extract were tentatively identified using spectral analysis. Figure 5 is the FTIR spectra of the extract while Table 9 summarizes the different bands obtained and their assignation. Ten (11) bands were observed among which eight were characteristic of molecules possessing antioxidant and antidiabetic activities, including phenols and flavonoids.

Acute toxicity

No visible change on the behavior or the macroscopic aspect of the main organs of the animals, were noted after administration of the extract to normal rats up to 14 days after

TABLE 4 Experimental, predicted values and desirability for polyphenolic content in optimal conditions.

	Predicted value	Experimental value	Desirability
Phenol (mg GAE/g)	85.12 ^a	86.77 ± 1.52^a	0.93

On the same line, values with different letters significantly differ (p > 0.05).

administration. The optimized extract of okra seeds was then said not to be toxic at unique dosage intake of up to 5000 mg/kg body weight. These results are in accordance with those of (42) who reported that okra hydroalcoholic extract did not show any toxicity or death up to a dose of 5000 mg/kg in Wistar rats. Uddin et al. (43) had also found okra mucilage powder and peel– seed mixture to be safe at a dose level of up to 1000 mg/kg of body weight in mice.

Oral glucose tolerance test

Table 10 shows the blood glucose concentrations of animals at different time during the oral glucose tolerance test. The okra seeds' extract limited the increase in the glucose level by 24.20 % compare to the normal control. Also, a quick drop down was noticed in the blood glucose level of rats treated with the optimized okra extract. This could be due the presence of soluble fibers in the extract which adsorbed glucose in the intestine, thus preventing it absorption into the blood (13).

Alpha amylase inhibitory activity and mechanism

The optimized okra seeds' extract was able to inhibit porcine pancreatic alpha amylase for up to 24.80 % at concentration of 200 μ g/mL as shown in Table 11. Lineaweaver-Buck plot permitted to classify the okra extract as a non-competitive inhibitor of alpha amylase (Figure 6). Similar observations were made by Quan et al. (44) who explained that the inhibitory activity could be due to the presence of phenolic compounds in the extract. Also, a possible synergistic interaction between polyphenolic and the terpenoid compounds could justify the great alpha amylase inhibitory activity of certain extracts compared to others (44). The observed result could also be

TABLE 6 Absorbance of the at different concentration during the FRAP assay.

	Concentration (µg/mL)							
	12,5	25	50	100	200			
AB	$1.51\pm0.17^{\rm b}$	3.02 ± 0.06^{b}	$3.20\pm0.06^{\rm b}$	$3.15\pm0.01^{\rm b}$	3.14 ± 0.03^{a}			
BHT	0.18 ± 0.02^a	0.40 ± 0.15^a	0.45 ± 0.09^a	0.97 ± 0.01^a	2.70 ± 0.10^a			

AB, Abelmoschus esculentus extract; BHT, Butyl hydroxyl–toluene; ^{a,b} in the same column, values with different letters differ significantly (p < 0.05).

TABLE 5	Phytochemical	composition of	the	optimized	okra extract.	
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Parameters	TPC (mg GAE/g)	TFC (mg CE/g	Fiber (g/100g)	Saponin (mg/g)	Proteins (mg/g)	Zinc (mg/g)
	86.77 ± 1.52	2.62 ± 0.27	1.84 ± 0.06	0.43 ± 0.05	0.175 ± 0.020	0.003 ± 0.00

TPC, total phenolic content; TFC, total flavonoid content; EC, equivalent catechin.

		Concentration (µg/mL)							
	12.5	25	50	100	200	IC ₅₀			
AB	$50.57\pm0.18^{\rm b}$	$62.22\pm0.04^{\text{b}}$	$87.18\pm0.08^{\rm b}$	$91.44\pm0.01^{\rm b}$	$91.29\pm0.00^{\rm b}$	3.99 ± 0.15^{a}			
BHT	$15.84\pm0.00^{\text{a}}$	20.16 ± 0.09^{a}	24.91 ± 0.08^a	38.63 ± 0.22^a	53.31 ± 0.15^a	$4.4 x 10^{11} \pm 15.28^{b}$			

TABLE 7 Percentage of DPPH inhibition at different concentrations.

AB, *Abelmoschus esculentus* (okra) extract; BHT, Butyl hydroxyl-toluene. ^{a,b} in the same column, values with different letters differ significantly (p < 0.05).



TABLE 8 Rf and identification of the different spots obtained on TLC with okra seeds' extract.

Bands	1	2	3	4	5
Rf	0.18	0.32	0.59	0.94	0.97
Identification	/	/	/	Quercetin	Catechin

due to the fiber content of the extract since (45) and Nsor-Atindana et al. (46) previously demonstrated that cellulose in a concentration and particles size-dependent manner can inhibit alpha amylase and alpha glucosidase.

Antidiabetic activity of the optimized okra extract

Effect of the treatment on the fasting blood glucose (FBG), the body weight and the food intake

A significant decrease in the FBG was observed in the group treated with the okra extract compared to the negative control. The optimized extract exhibited better hypoglycemic activity than metformin. Such activity could be explained by the presence of polyphenols which are able to stimulate glucose absorption at the muscular level, or to increase insulin production by the pancreas. Flavonoids like quercetin that were found in the okra extract can induce the expression of the glucose transporter GLUT 4 or inhibit PPAR γ 1, which end up in an increased absorption of glucose in muscles (47).

At the end of the treatment, a significant weight loss was notice in untreated diabetic rats, as a consequence of diabetes (48). The okra extract was then able to alleviate muscular weight loss in treated diabetic rats, probably by ameliorating the assimilation and utilization of glucose at the levels of cells, the weight loss in diabetics being a direct consequence of the inability of the body to use circulating glucose. Similar observations were made by Gupta et al. (49, 50), with plant extracts on diabetic rats.

A decrease of 2.57 mg in the food intake of the rats treated with the okra extract was observed, thus suggesting the extract had been able to alleviate polyphagia in animals. The observed activity can be related to the soluble fibers contained in the extract, which may limit the production of orexigenic compounds like ghrelin by enhancing the production of appetite suppressors like cholecystikinin, Glucagon-like Peptide 1 (GPL-1) and peptide YY (50–52). Table 12 summarizes the variations in the different parameters.

Effect of the treatment on the lipid profile and the atherogenic index

The treatment with okra extract significantly lowered blood triglyceride (TAG), total cholesterol (T Chol) and LDL cholesterol concentrations in diabetics rats. HDL cholesterol concentration was significantly elevated in the AB group, compared to diabetic and normal controls. The atherogenic risk was significantly reduced in okra treated diabetic controls and metformine treated rats, thus leading to the conclusion that treatment of diabetic patients with okra extract could reduce the risk of cardiovascular accidents. Presence of flavonoids like quercetin (53) or alkaloids (54) in the extract may explain these activities. Esmaeilzadeh et al. (55) reported that okra extracts could stimulate the production of alpha hydroxylase, the enzyme converting cholesterol into bile acids in the



TABLE 9 Peak wave numbers and tentative identification.

Bands (cm ⁻¹)	Vibration	Assignation	Reference	
3269	-C-OH (stretching)	Water, cabohydrates	(34, 35)	
2926	-C-H (stretching)	Aliphatic portion of lipids	(35)	
		Isoflavones	(36)	
629	-C=C		(37)	
400	–O–H (bending)	Phenols or tertiary alcohol	(38)	
237	unidentified			
030	Ester –C–O (stretching)	Glycosidic groups	(39)	
26	-C-C (stretching)	Alkane: lipids, amino acids, proteins		
07	Phenolic ring (torsion)	Phenols	(40)	
46	unidentified			
13	-C-OH in plane (bending)	Phenols	(41)	

TABLE 10 Blood glucose levels and increment during the OGTT.

	0 min	15 min	30 min	60 min	90 min	120 min	Increment
AB	$100\pm7.54^{\text{b}}$	155.33 ± 4.5^{a}	$134\pm7^{\text{a}}$	124.33 ± 2.08^a	113.33 ± 4.93^{a}	98 ± 8^a	55.33±3.05ª
Т	85 ± 4.35^a	158 ± 6.2^{a}	$141\pm\!8.12^a$	126 ± 3.46^a	116.33 ± 6.65^a	107.33 ± 7.09^a	73 ± 2.64^{b}

AB, Abelmoschus esculentus (okra) extract; T, normal control. Increment: highest blood glucose value (at $15 \min$)—start value ($0\min$); ^{a,b} in the same column, values with different letters differ significantly (p < 0.05).

		Concentration (µg/mL)			IC ₅₀ (µg/mL)
	25	50	100	200	
AB	10.04 ± 0.30	10.66 ± 0.76	17.13 ± 2.16	24.80 ± 2.36	484.17 ± 2.33



AB, Abelmoschus esculentus (okra) seeds' extract.



liver, or inhibit the HMG-COA (3-hydroxy-3-methyl-glutarylcoenzyme A) reductase; resulting in a decrease in the circulating cholesterol. The extract could exert its LDL cholesterol lowering activity by limiting the intestinal absorption of lipids implicated in their formation in the liver (56). Also, Brijyog et al. (57) stated that flavonoids present in the extract could induce the transcription of lipoprotein lipase genes. The enzyme is then produced and breakdown lipoproteins like LDL cholesterol with an end result of its blood concentration reduction. Similar observations were made by Abd El Latif et al. (58) who noticed a normalization in the lipid profile of diabetic rats treated with soybean isoflavones. Table 13 gives the lipid profile and atherogenic index of the animals at the end of the treatment.

Renal and hepato-protective activities of the optimized okra extract

Serum Aspartate amino transferase (AST), Alanine amino transferase (ALT) and Alcaline Phosphatase (ALP) concentrations were measured to assess the integrity of the liver. The liver plays a key role in the glucose homeostasis, and diabetes could disturb its functioning, which is marked by elevated concentrations of AST, ALT and ALP in the blood, resulting from an inflammatory state induced by hyperglycemia

TABLE 12 Variation of some parameters between the start and the end of the experiment.

	FBG (g/L)	Body weight (g)	Food intake (g)
AB	-102 ± 7.81^{a}	$-2.33\pm9.01^{\text{b}}$	-2.57 ± 0.04^a
Met	$-83.66\pm9.23^{\text{b}}$	-7.66 ± 15.94^{b}	$0.32\pm0.01^{\text{b}}$
T+	26 ± 2.64^{c}	-45.33 ± 11.06^{a}	$8.46\pm0.04^{\rm d}$
T-	$2\pm1.41^{\rm d}$	39.5 ± 8.54^{c}	4.54 ± 0.02^{c}

AB, *Abelmoschus esculentus* (okra) seeds' extract; Met, Metformine; T+, Diabetic control; T-, Negative control; FBG, Fasting blood glucose. Variation = final value-start value; a, b, c, and d, in the same column, values with different letters differ significantly (p < 0.05).

and oxidative stress (59, 60). The okra extract exhibited hepato-protective capacities by significantly reducing these enzymes concentrations in the blood, compared to diabetic control. Similar observations were made by (58, 61, 62). The hepato-protective activity noticed with the optimized okra extract could be due to its antioxidant capacity, as (63) hypothesized that plant extracts could exert their hepato protective activity by combating the oxidative stress which is responsible of the inflammation and necrosis of the liver, resulting in the elevation of the above mentioned liver

TABLE 13 Lipid profile and atherogenic index of animals at the end of the treatment.

	TAG	T CHOL	HDL	LDL	AI
AB	33.41 ± 5.03^a	47.73 ± 0.23^a	36.73 ± 1.93^{b}	4.32 ± 0.89^{a}	1.30 ± 0.06^{a}
Met	64.81 ± 1.56^{c}	$68.40 \pm 2.11^{\mathrm{b}}$	36.08 ± 2.60^{b}	19.35 ± 3.69^{c}	1.90 ± 0.15^{b}
T+	42.67 ± 3.61^{b}	76.53 ± 4.73^{c}	30.94 ± 3.23^a	37.05 ± 5.83^d	2.49 ± 0.33^{c}
T–	47.43 ± 3.50^{b}	64.25 ± 4.59^{b}	38.98 ± 4.07^{b}	15.78 ± 2.31^{b}	1.65 ± 0.08^{b}

AB, Abelmoschus esculentus; F, Formulation; Met, Metformin; T+, Diabetic control T–, Normal control. TAG, Triglycerides; T CHOL, Total cholesterol; HDL, HDL cholesterol; LDL, LDL cholesterol; AI, Atherogenic index; a, b, c, and d, in the same column; values with different letters differ significantly (p < 0.05).

TABLE 14 Effect of the treatment on liver biomarkers.

	AST (UI/l)	ALT (UI/l)	ALP (UI/l)
AB	38.11 ± 6.71^{a}	8.75 ± 1.54^{a}	$214.65\pm5.18^{\text{b}}$
Met	74.08 ± 5.92^{b}	21.19 ± 3.75^{b}	209.15 ± 4.85^{b}
T+	89.83 ± 5.05^{c}	$28.58\pm4.04^{\text{c}}$	231.17 ± 2.59^{c}
T-	40.83 ± 6.06^a	22.45 ± 3.94^{b}	137.60 ± 7.78^a

AB, Abelmoschus esculentus; Met, Metformine; T+, Diabetic control; T-, Normal control; AST, Aspartate amino transferase; ALT, Alanine amino transferase; ALP, Alcaline Phosphatase; a, b, and c, in the same column; values with different letters differ significantly (p < 0.05).

TABLE 15 Serum and urine concentrations of creatinin of the treated rats.

Serum		Urine	
Creatinin concentrations (µmole/L)			
AB	2.53 ± 0.10^{a}	$172.61 \pm 1.17^{\rm d}$	
Met	2.47 ± 0.30^{a}	156.11 ± 9.25^{c}	
T+	$2.88\pm0^{\rm b}$	41.96 ± 2.40^a	
T-	2.38 ± 0.10^a	122.69 ± 7.82^b	

AB, Abelmoschus esculentus; Met, Metformin; T+, Diabetic control; T-, Normal control. a, b, c, and d: in the same column, values with different letters differ significantly (p < 0.05).

enzymes and precisely the ALP. Table 14 summarizes the values obtained.

Among the most severe complications of diabetes, is renal failure. A good antidiabetic management should then prevent or delay it appearance. Creatinin clearance is the commonest way to assess the renal status since a high concentration of creatinin in the blood circulation can be indicative of a renal failure. The okra extract significantly reduced the blood concentration of creatinin in comparison to diabetic control (Table 15). This was in accordance with previous works (48, 56). Danish et al. (56) reported that daily administration of *Albizzia lebbeck* stem bark extracts to diabetic rats for 45 days induced a significant decrease in the serum creatinin concentration.

Conclusion

The study aimed at evaluating the antidiabetic activity of an optimized polyphenolic rich extract obtained from okra seeds. RSM was used with MAE to determine the optimal conditions for extracting polyphenols from okra seeds. It was found out that the solvent/dry matter ratio, the power of the microwave apparatus, the quadratic effect of the time (X_1X_1) and the interaction between the solvent/dry matter ratio and the operating power (X_3X_2) significantly influenced (p < 0.05) the polyphenolic content of the extracts. RSM permitted to define the conditions for maximum extraction of polyphenols from okra (87.66 \pm 3.33 mg of GAE/g) as: microwave power of 330 W, with a solvent ratio of 97.04/1 for 9.5 min. Optimization thus permitted to determine the conditions for extraction of at least two-fold the average maximum TPC reported up to date from okra seeds. The optimized extract exhibited powerful antioxidant capacities with an IC₅₀ of 3.99 \pm 0.15 µg/mL in DPPH scavenging assay. It also acted as a non-competitive inhibitor of porcine pancreatic amylase, and showed good antidiabetic capacities on streptozotocin induced diabetic rats.

Data availability statement

The original contributions presented in the study are included in the article/supplementary material, further inquiries can be directed to the corresponding author.

Author contributions

CW and DK conceived the work, collected seeds, carried out experimentations, analyzed and interpreted data, and wrote the article. DM collected the seeds, assisted in experimentations, and read the article. HW supervised the work and read the article. All authors have approved the final article.

Conflict of interest

The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

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