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Unraveling the dynamic changes of volatile compounds during the rolling process of Congou black tea via GC-E-nose and GC–MS

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Rolling plays an important role in shaping the characteristic quality of black tea. However, the dynamic alterations in volatile compounds during the rolling process remain unclear, which hampers the quality control of black tea. In this study, the dynamic changes of volatile compounds in Congou black tea throughout the rolling process were explored by a combination of GC-E-Nose and GC-MS analysis. The results of PCA and PLS-DA by GC-E-Nose revealed significant alterations in the volatile profiles, particularly at the 15-min rolling stage. A total of 82 volatile compounds including eight categories were identified in Congou black tea during the rolling process. The most abundant volatiles were aldehydes, alcohols, and esters. Through the integration of variable importance in the projection >1 and p < 0.05, 21 volatile compounds were pinpointed as pivotal volatiles responsible for distinguishing diverse rolling processes. According to their pathways, glycoside-derived volatiles (GDVs) (such as linalool and geraniol) and fatty acid-derived volatiles (FADVs) (such as (Z)-2-hexen-1-ol and hexanal) highlighted the pivotal role in the formation of the aroma of black tea during rolling process. GDVs, characterized by floral, honeyed, and fruity aromas, reached the lowest point at 35-min rolling and subsequently rebounded. FADVs exhibited an ascending trajectory during the initial 15-min rolling, followed by a downward trend. These findings provide invaluable insights into the aroma evolution during the rolling stage, offering strategies for enhancing the aroma quality of Congou black tea.

KEYWORDS

rolling, black tea, volatile compounds, GC-E-nose, GC-MS.

Highlights

- Notable alterations in the volatile profiles occurred at 15-min rolling.
- A total of 82 volatiles were successfully identified during rolling process.
- Twenty-one volatiles played a key role in differentiating different rolling stages.
- GDVs and FADVs play an important role in aroma formation during rolling process.

1 Introduction

Tea, acknowledged as one of the world's most beloved beverages, is categorized into six major categories (green tea, white tea, yellow tea, oolong tea, black tea, and dark tea) in China based on the differences in processing techniques (Fang et al., 2023). Among these, Congou black tea, known as a fully fermented tea, has exceeded the others in global popularity due to its rich aroma, distinctive flavor, exquisite color, and perceived health benefits (Zheng et al., 2023). The post-harvest processing of Congou black tea typically involves withering, rolling, fermentation, and drying (Tu, 2015). As a crucial step, rolling involves curling tea leaves into fixed shapes either manually or mechanically, aiding in shaping their appearance and forming their internal qualities. Luo et al. (2016) have discovered that black tea with different modeling technology, could enhance the taste quality. In addition, the rolling process ensures the uniform mixing of tea leaves, facilitating various enzyme-catalyzed reactions and oxidation reactions during the rolling stage (Ozdemir et al., 2017). At present, relevant studies mainly focus on the effects of withering (Ye et al., 2020), fermentation (Chen et al., 2022a), and drying (Wang et al., 2023) on the aroma quality of black tea. However, the dynamic changes in volatile compounds during the rolling process are still unclear, which hampers the quality control of black tea, to some extent.

The comprehensive aroma profile of black tea is determined by an extensive array of volatile compounds with varying concentrations. The typical aromas of black tea include sweet, fruity, and floral fragrances (Li et al., 2021a). Traditional aroma quality assessment mainly relies on sensory evaluation, which is subjective and prone to variation. To overcome the limitations of human sensory perception, electronic nose (E-nose) technology, simulating the olfactory system of mammals, has been developed to provide more objective assessments (Peng et al., 2023). Gas chromatography electronic nose (GC-E-Nose), as a cutting-edge olfactory technology, offers advantages such as speediness, non-destructiveness, stability, and high sensitivity, and has been extensively employed in the tea field (Chen et al., 2022b; Wang et al., 2024a). GC-E-Nose surpasses traditional electronic nose in detecting a broader scope of volatile compounds. It is equipped with two chromatographic columns (a weak-polarity column of MXT-5 and a medium-polarity column of MXT-1701), effectively separating volatiles with diverse polarities (Liu et al., 2022). On the other hand, gas chromatography-mass spectrometry (GC-MS), renowned for its strong qualitative and quantitative capabilities, has emerged as the most prevalent and critical technology in tea aroma analysis (Zhang et al., 2023a). For instance, various storage periods of Pu-erh teas have been successfully distinguished by utilizing GC-E-Nose combined with GC-MS (Rong et al., 2023). Similarly, the profiles of volatile compounds in sweet and floral aroma black tea have been effectively characterized by using a comprehensive analysis strategy of GC-E-Nose and GC-MS (Yang et al., 2024a).

The precursors of tea aroma include lipids, glycosides, amino acids, and carotenoids (Chen et al., 2022a; El Hadi et al., 2013; Ho et al., 2015). According to the pathways, volatile compounds can be categorized into fatty acid-derived volatiles (FADVs), glycoside-derived volatiles (GDVs), amino acid-derived volatiles (AADVs), and carotenoid-derived volatiles (CDVs) (Xie et al., 2023). Considering the pivotal role of processing techniques in shaping the aroma, some scholars have endeavored to explore the mechanisms underlying

aroma formation of black tea. However, these studies have predominantly focused on exploring the aroma formation throughout the full-scope manufacturing process of black tea, overlooking the unique aspects of the rolling stage. The purpose of this study is to systematically investigate the dynamic changes of volatile compounds in Congou black tea during the rolling stage via GC-E-Nose combined with GC-MS. In addition, the potential formation mechanisms for key volatile compounds were also explored. The results will provide theoretical support for quality control and strategic processing of black tea.

2 Materials and methods

2.1 Materials and reagents

The 20-mL headspace vials with 18 mm magnetic PTFE/silicone caps were procured from Agilent Technologies Inc. (Palo Alto, California, United States). Ethyl decanoate was purchased from Shanghai Aladdin Biochemical Technology Co., Ltd. (Shanghai, China). Purified water was obtained from Hangzhou Wahaha Group Co., Ltd. (Hangzhou, China).

2.2 Sample preparation

The freshly plucked leaves of "Jiukeng" cultivar from one bud one leaf to one bud two leaves were harvested in September.

bud two leaves were harvested in September, 2023 from Shengzhou City, Zhejiang Province, China. The processing protocol encompassed the stages of withering and rolling. The parameters of the rolling process were based on previous studies with some minor adjustments (Yang et al., 2024b; Zhang et al., 2023b). The specific processing flow is depicted in Supplementary Figure S1. Tea samples were collected during the rolling process at different intervals (0 min, 15 min, 35 min, 55 min, and 75 min), and subsequently subjected to freeze-drying for further analysis.

2.3 GC-E-nose analysis

The tea samples were analyzed using GC-E-Nose (Alpha M.O.S., Toulouse, France), adhering to the method parameters detailed in a previous study (Xie et al., 2024). Initially, 0.5g of tea samples were weighed and transferred into a 20-mL hermetically sealed headspace vial. Subsequently, the headspace vial was placed in an incubator and subjected to incubation at 60°C for 20 min. Next, 5,000 µL of headspace gas was injected at a temperature of 200°C and a pressure of 10kPa. The programmed temperature was as follows: initially maintained at 50°C for 5 s, then increased from 50°C to 80°C at a rate of 0.1°C/s, and finally ramped from 80°C to 250°C at a rate of 0.4°C/s (holding for 100s). The two flame ionization detectors (FIDs) were maintained at a temperature of 260°C. The volatile profiles underwent detailed analysis via the built-in AlphaSoft software. Subsequently, a qualitative assessment was performed by employing a comparative analysis of the calculated retention indices (RIs) from C6-C16 against those within the AroChemBase database (Xie et al., 2023).

2.4 GC–MS analysis

The volatile compounds were analyzed using an Agilent 7890B gas chromatograph coupled to a 7000C mass spectrometer (Agilent Technologies, Palo Alto, CA, USA), following the method parameters described by Wang et al. (2024b). For each analysis, 0.5 g of tea samples, 2μ L of ethyl decanoate (internal standard, 50 mg/L), and 5 mL of pure water were introduced into a 20-mL headspace vial. Subsequently, the samples were extracted at 60°C for 60 min employing a divinylbenzene/ carboxen/polydimethylsiloxane fiber (DVB/CAR/PDMS, 50/30µm, 2 cm). After extraction, the DVB/CAR/PDMS fiber was thermally desorbed for 5 min in the GC-MS injector. Volatile compounds were separated utilizing a DB-5MS capillary column (Agilent Technologies, Palo Alto, CA, USA). The heating program of the oven temperature was as follows: initially maintained at 40°C for 5 min, then increased to 160°C at a rate of 4°C/min (maintaining for 2 min), and finally increased to 270°C at a rate of 10°C/min (maintaining for 2 min). Mass spectrometry analysis was performed in electron ionization mode at 70 eV, with a mass scan range of 40–450 m/z. The temperatures of the inlet, transfer line, and ion source were set to be 250°C, 270°C, and 230°C, respectively.

The qualitative analysis of volatile compounds was conducted using the Agilent MassHunter workstation software, via searching against the NIST 11 library. The RIs were determined using the homologous series of normal alkanes (C7 ~ C40) and compared with data obtained from scientific literatures (https://webbook.nist.gov/ chemistry/ and https://www.flavornet.org/flavornet.html). The volatile compounds were quantitatively determined using the internal standard (ethyl decanoate) method, with the following calculation formula:

$$\operatorname{Ci} = (\operatorname{Cis}^*\operatorname{Ai}) / \operatorname{Ais}.$$

whereas *Ci* represents the concentration of the targeted volatile component in μ g/L, *Ai* denotes the peak area of the targeted volatile component, *Cis* represents the concentration of ethyl decanoate (internal standard) in μ g/L, and *Ais* denotes the peak area of ethyl decanoate.

2.5 Statistical analysis

All the experiments were conducted in triplicate, and the results were presented as mean values \pm standard deviations. Principal component analysis (PCA) and partial least squares-discriminant analysis (PLS-DA) were performed using the SIMCA-P 14.1 software (Umetrics, Umea, Sweden). The significance analysis was conducted using the SPSS Statistics 20.0 software (SPSS Inc., Chicago, IL, United States). Heat maps, bar charts, sunburst charts and change trajectories were generated using the Origin 10.1 software (OriginLab Corporation, Northampton, MA).

3 Results and discussion

3.1 Volatile profiles analyzed by GC-E-nose

In this study, GC-E-Nose was employed to analyze the volatile profiles of black tea during the rolling stage. The volatile compounds

were qualitatively analyzed by comparing the retention index calculated by C6-C16 with those in the AroChemBase database, and the detailed information were listed in Supplementary Table S1. The parallel radar graphs for MXT-5 and MXT-1701 were illustrated in Supplementary Figure S2. Each peak represented a specific volatile compound, with the peak area indicating its content. Notably, the alterations in peak intensity with extended rolling time demonstrated the distinct variations in volatile compounds during the rolling stage. Initially, unsupervised PCA was conducted for analysis (Fu et al., 2024). As depicted in Supplementary Figure S3, the first two principal components explained 63.3% of the total variance of R²X, with the contribution rate of PC1 achieving 44.1%, and that of PC2 reaching 19.2%. Notable spatial distribution was observed, and the scatter plots of the 15-min rolling samples were distributed far away from the scatters of other groups, indicating that volatile compounds exhibited substantial discrepancies during the rolling process, particularly at the 15-min rolling stage. Subsequently, PLS-DA was conducted on the volatile profiles obtained from GC-E-Nose. As a supervised statistical analysis technique, PLS-DA integrates partial least squares regression and linear discriminant analysis to minimize intra-group disparities and maximize inter-group differences, assisting in the classification of distinct categories or groups. The R²Y and Q² parameters are crucial indicators for evaluating the model performance. R²Y represents the proportion of variance in the observed variables (Y variables), and Q² represents the model's predictive capability. The closer the values of R²Y and Q² are to 1, the stronger the model's explanatory and predictive power (Li et al., 2023). As illustrated in Figure 1A, a robust model with favorable parameters ($R^2Y=0.910$, $Q^2=0.745$) was obtained, indicating its considerable explanatory and predictive power. The score plots intuitively revealed that the distances between 0-min and 15-min rolling, as well as 55-min and 75-min rolling, were larger compared to other rolling periods, suggesting that notable alterations in the volatile profiles occurred at 15-min and 75-min rolling. To further validate the model's robustness, 200 permutation tests were performed, yielding $R^2 = 0.373$ and $Q^2 = -0.469$ (Figure 1B). In summary, our results demonstrate that GC-E-Nose is a promising technology capable of quickly and efficiently distinguishing the volatile profiles across various rolling stages of black tea.

3.2 Dynamic change of volatile compounds in black tea during the rolling stage analyzed by GC–MS

3.2.1 Volatile compounds analyzed by GC-MS

The volatile compounds were analyzed by GC–MS, and the total ion chromatograms were shown in Supplementary Figure S4. A total of 82 volatile compounds were successfully identified during the rolling stage. These volatiles could be categorized into eight major classes, encompassing 15 aldehydes, 14 alcohols, 14 esters, 11 ketones, 10 alkanes, 9 aromatic compounds, 6 alkenes, and 3 heterocyclic compounds. The detailed information were listed in Supplementary Table S2. Notably, aldehydes, alcohols, and esters occupied the predominant proportions, accounting for 18.3%, 17.1%, and 17.1%, respectively (Figure 2A).

Additionally, obvious alterations in volatile compounds were observed during the rolling stage (Figure 2B). Aldehydes exhibited a noticeable increasing trend, whereas esters, alkanes, aromatic



compounds, and heterocyclic compounds displayed distinct decreasing trends. Interestingly, alkanes $(1.40 \,\mu\text{g/L} \sim 5.80 \,\mu\text{g/L})$ and heterocyclic compounds $(1.50 \,\mu\text{g/L} \sim 4.95 \,\mu\text{g/L})$ not only demonstrated uniform changing trends during the rolling stage but also possessed comparable concentration levels. Alcohols, alkenes, and ketones displayed minor fluctuations throughout the rolling process.

3.2.2 Multivariate statistical analysis

To further investigate the alterations in volatile compounds during the rolling stage, PLS-DA was conducted on the acquired volatile compounds. Favorable model was achieved with parameters of $R^2Y=0.892$ and $Q^2=0.613$, indicating the robust explanatory and predictive capabilities. As depicted in the Figure 3A, considerable changes occurred during the rolling stage, particularly at the 15-min rolling. In addition, 200 permutation tests were implemented, yielding parameters of $R^2=0.530$, $Q^2=-0.539$. All Q^2 points were lower than the rightmost point, confirming the validity of the model (Figure 3B). Moreover, utilizing the criteria of variable importance in the projection (VIP) > 1 and p < 0.05, 21 key volatile compounds were pinpointed, such as 1,2-dimethyl-4-vinylbenzene (VIP = 1.38), *o*-cymene (VIP = 1.37), β -myrcene (VIP = 1.18), linalool (VIP = 1.11), and (+)-4-carene (VIP = 1.03), which functioned as pivotal differentiators among diverse rolling stages (Figure 3C).

3.2.3 Analysis of dynamic changes in volatile compounds classified by different pathways

Based on the pathways, the aforementioned 21 key differentiating volatile compounds were classified into five categories: 8 GDVs, 6 FADVs, 3 AADVs, 1 CDV, and 3 others (Figure 4A). This categorization highlighted the significance of GDVs and FADVs in the aroma formation of black tea during the rolling stage. As shown in Fig. S5, the level of GDVs exhibited a descending from the initial 0-min rolling (107.90 µg/L) to the 35-min rolling (75.36 µg/L), subsequently exhibiting an ascending trend. As far as FADVs is concerned, the concentration of FADVs increased from the 0-min rolling (11.97 µg/L) to the 15-min rolling (31.24 µg/L), followed by a downward trend. As for the AADVs, the concentration progressively escalated throughout the rolling process (from $20.14 \mu g/L$ to $38.09 \mu g/L$). The concentration of CDV was several orders of magnitude inferior compared to other three categories. However, owing to the low threshold characteristics of CDV, its influence could not be overlooked (Xu et al., 2023).

3.2.3.1 Changes in FADVs

FADVs are prevalent and essential volatile compounds in tea (Li et al., 2021b). In this study, characteristic FADVs included hexanal, (Z)-2-hexen-1-ol, 2-butylfuran, (E)-4-heptenal, (E, E)-2,4-heptadienal and (E)-2-nonenal. FADVs were mainly derived from lipid oxidation, and there existed two primary pathways for generating FADVs (Ho et al., 2015). The first pathway involved free radical-induced autooxidation reaction, with (E, E)-2,4-heptadienal generated from the auto-oxidation of α -linolenic acid (Figure 4B) (El Hadi et al., 2013). (E, E)-2,4-Heptadienal could enhance the freshness of the aroma (Yang et al., 2024b; Zhang et al., 2023c). It was evident that the concentration of (E, E)-2,4-heptadienal was consistently increased during the rolling stage (Figure 5A). On the other hand, the applied force could potentially cause the breakage of tea leaves, leading to the release of cell sap that accelerated the auto-oxidation of α -linolenic acid and resulted in a higher concentration of (E,E)-2,4-heptadienal.

The second pathway was the lipoxygenase-mediated lipid oxidation pathway, which was the main pathway for tea aroma generation (Ho et al., 2015). Of the six FADVs identified in this study, five FADVs such as hexanal, (E)-4-heptenal, and (E)-2-nonenal were produced through this pathway (Figure 4B). Specifically, hexanal stood as one of the key aroma compounds in sweet-floral aroma black tea, contributing to the unique green, leafy, and grassy aromas (Yang et al., 2022a). In this study, hexanal exhibited a progressively increasing pattern throughout the rolling process (Figure 5A). It was produced from the oxidation of linoleic acid by lipoxygenase (LOX), which was subsequently further cleaved by hydroperoxide lyases (HPL). Similarly, (E)-4-heptenal and (E)-2-nonenal were the oxidation products of palmitoleic acid and α -linolenic acid, respectively (Caipo et al., 2021). (E)-4-Heptenal was recognized as a crucial aroma compound in Dianhong Congou tea, and it exhibited a continuous ascend during the rolling stage (Chen et al., 2022b). Additionally, (E)-2-nonenal was reported to exhibit a strong and intense aroma in Assam black tea, and it attained the lowest concentration at 35-min



rolling (Kang et al., 2019). It is worth mentioning that these aldehydes could be transformed into the corresponding alcohols under the action of alcohol dehydrogenases (ADH) or undergo isomerization reactions. For instance, (*Z*)-2-hexen-1-ol was predominantly formed through the oxidation of α -linolenic acid to 13S-hydroperoxylinolenic acid, which subsequently gave rise to (*Z*)-2-hexenal under the action

of HPL and further underwent isomerization (Chen et al., 2022a). (*Z*)-2-Hexen-1-ol was generally associated with a banana-like fruity aroma, and it achieved the maximum concentration at 15-min rolling (Ríos-Reina et al., 2023). As for 2-butylfuran, the formation primarily involved linoleic acid via LOX and ADH reactions to yield (*E*)-2-octenal, which was further transformed under the catalysts of amino



acids (Adams et al., 2011). As an important *N*,O-heterocyclic compound in tea, 2-butylfuran exhibited a decreased trend from 0-min rolling to 35-min rolling, subsequently increasing progressively (Yang et al., 2022b).

3.2.3.2 Changes in GDVs

GDVs in fresh tea leaves generally exist in glycosidic forms, and are easily hydrolyzed by endogenous glycosidases during processing,

which generally exhibit floral, honey, and fruity aromas (Yang et al., 2024a; Wang et al., 2022). In this study, representative GDVs such as benzyl alcohol, β -myrcene, (+)-4-carene, *o*-cymene, linalool, 1,3,8-*p*-menthatriene, geraniol, and β -citral were identified.

Geraniol and linalool were formed from geranyl pyrophosphate (geranyl-PP) precursor, liberated by geraniol synthase and linalool synthase, respectively. β -Citral and β -myrcene were the derivatives of geraniol and linalool, respectively (Figure 4B) (Ho et al., 2015; Yang





et al., 2024a). These compounds were typical glycosidically bound terpene alcohols, primarily exhibiting floral, sweet, and fruity aromas (Liu et al., 2023; Ouyang et al., 2022). Rolling process influences the levels and compositions of GDVs in tea by causing cellular damage, which releases various enzymes and substrates that interact to form or

alter GDVs. For example, increased interaction between enzyme and substrate during rolling can accelerate β -glucosidase reactions, promoting the release of GDVs. β -Citral exhibited an upward trend during the rolling process, and was recognized as an essential aroma-active compound in sweet aroma black tea (Yang et al., 2024a).

Geraniol, offering a rose-like aroma, was one of the primary aroma components in green tea, black tea, and oolong tea, and exhibited an increasing trend during the rolling process (Yu et al., 2021; Zhang et al., 2020). Linalool with citrus and floral notes, exhibited a descending trend during rolling (Figure 5B) (Chen et al., 2022b; Yang et al., 2022b). β -Myrcene, possessing geranium-like, carrot-like, and hop-like aromas, was a pivotal volatile component in sweet floral black tea. It attained the maximize level at 15-min rolling (Chen et al., 2022c; Yang et al., 2024a).

O-cymene was reported to play an essential role in the floral aroma of Wuyi rock tea (Peng et al., 2023). It demonstrated similar trend to β -myrcene during rolling, reaching the highest level at 15-min rolling prior to stabilization. Benzyl alcohol, a typical GDV and aromatic alcohol derived from glycosidic substances, contributed to the sweetness and roasty notes (Ho et al., 2015; Ouyang et al., 2022; Xie et al., 2024). It displayed a progressive upward trend during the rolling stage, which corroborated with the finding of Chen et al. (2022a). (+)-4-Carene was identified as one of the key aroma distinctions between shaken black tea and traditional black tea (Wang et al., 2023). It declined from 0-min rolling to 35-min rolling, and then remained stable unaltered. 1,3,8-*p*-Menthatriene attained its lowest concentration at 35-min rolling (Figure 5B).

3.2.3.3 Changes in AADVs

Amino acids perform a dual role in tea, serving not merely as crucial taste factors but also as the precursors of volatiles (Yang et al., 2024a). In this study, the representative AADVs including styrene, 1,2-dimethyl-4vinylbenzene, and benzeneacetaldehyde are highlighted. The primary pathway of styrene in plants was from linolenic acid, which was derived from phenylalanine (Schwab et al., 2008; Warhurst and Fewson, 1994). Benzeneacetaldehyde is derived from phenylalanine via Strecker degradation (Figure 4B). The generally mechanism of Strecker degradation of amino acids involves the nucleophilic addition of an amine group to a carbonyl group, generating an unstable aldimine, which subsequently eliminates a molecule of water to form a Schiff base. Subsequent decarboxylation yields an imine polymer, which then undergoes the addition of a molecule of water to form an unstable amino acid. Ultimately, this amino acid decomposes into a ketoamine and Strecker aldehyde, resulting in the formation of the Strecker aldehyde possessing one less carbon atom than its precursor amino acid (Ho et al., 2015). As depicted in Figure 5C, styrene exhibited a "V" pattern throughout the rolling stage, with the lowest concentration occurring at 35-min rolling. Styrene was characterized by sweet and floral aromas, exhibiting notable manifestation during the fermentation phase of black tea (Xiao et al., 2022). 1,2-Dimethyl-4-vinylbenzene exhibited an inverted "V" trend throughout the rolling stage, with the highest concentration occurring at 15-min rolling. Benzeneacetaldehyde exhibited a progressive upward trajectory during the rolling stage, corresponding with previous findings (Chen et al., 2022a). It is noteworthy that benzeneacetaldehyde possessed rose-like and sweet aromas, and was the dominant aromatic compound in Dajiling black tea (Xie et al., 2024).

3.2.3.4 Changes in CDVs

Carotenoids, a category of tetraterpenoid pigments, are commonly stored in the plastids of plant leaves, flowers, and fruits. They play an indispensable and crucial role as accessory pigments and colorants in photosynthesis. Additionally, carotenoids act as vital precursors influencing the aroma quality of tea (Schwab et al., 2008). Typically, CDVs are generated through the action of carotenoid cleavage dioxygenase (CCD), resulting in the formation of carotenoids and their degradation products (El Hadi et al., 2013). In this study, typical CDVs such as geranyl acetone was identified in the rolling stage of black tea. Geranyl acetone possessed a low threshold, contributing to the unique sweet and floral aromas of black tea (Xu et al., 2023). It was derived from phytoene via CCD catalysis (Figure 4B). The concentration of geranyl acetone exhibited a distinctive "V" trend during the rolling stage. Specifically, it commenced from the highest concentration (0.31 μ g/L) at 0-min rolling, gradually declined to 0.15 μ g/L at 35-min rolling, and subsequently ascended to 0.29 μ g/L at 75-min rolling (Figure 5D).

3.2.3.5 Changes in other types of volatile compounds

We also identified three other types of volatile compounds such as *cis*-3-hexenyl benzoate, decanoic acid, methyl ester, and *m*-xylene that markedly influenced the aroma during the rolling stage. *Cis*-3hexenyl benzoate reached its peak at 0-min rolling, with the lowest point at 35-min rolling (Figure 5E). Decanoic acid, methyl ester was reported to play an important role in the formation of fruity and floral aromas (Tchouakeu Betnga et al., 2020). It reached the lowest concentrations at 35-min rolling. *m*-Xylene was reported as a predominant volatile component in aloe tea, exhibiting floral and sweet aromas (Zhang et al., 2024). In this study, *m*-xylene reached its peak concentration at 15-min rolling, followed by a primarily decreasing trend, reaching its lowest concentration at 75-min rolling.

4 Conclusion

This study employed a combination of GC-E-Nose and GC-MS analysis for exploring the dynamic alterations in volatile compounds throughout the rolling process of black tea. The PLS-DA results of GC-E-Nose and GC-MS demonstrated the capability to rapidly distinguish black teas at different rolling stages. Notable alterations in the volatile profiles occurred at 15-min rolling. A total of 82 volatile compounds were identified by GC-MS. Through integration of VIP>1 and p < 0.05, 21 volatile compounds such as β -myrcene, (+)-4-carene, o-cymene, 1,2-dimethyl-4-vinylbenzene, and linalool were pinpointed as pivotal volatiles responsible for distinguishing diverse stages of the rolling process. Notably, AADVs such as benzeneacetaldehyde, primarily exhibiting sweet and floral fragrances, exhibited an upward trend during the rolling stage. GDVs with floral, honeyed, and fruity notes, attained their lowest contents at 35-min rolling and subsequently rebounded. In conclusion, this study provides valuable insights into the aroma evolution during the rolling stage, offering strategies for enhancing the aroma quality of black tea. Despite the promising results, this study does possess certain limitations. Although the key differential volatile compounds were classified based on their formation pathways, the investigation did not quantify the precursors of these volatiles or the specific environmental conditions influencing their formation. As future innovations permit, we will monitor the rolling temperature to better understand the impact of temperature on the formation and/or degradation of volatile compounds during the rolling process. In addition, we intend to explore the changes of precursor substances during the rolling process, and trace the specific formation pathway of volatile substances by using stable isotope tracer technology.

Data availability statement

The original contributions presented in the study are included in the article/Supplementary material, further inquiries can be directed to the corresponding authors.

Author contributions

QW: Data curation, Formal analysis, Investigation, Methodology, Resources, Software, Validation, Visualization, Writing – original draft. DS: Conceptualization, Resources, Software, Writing – original draft. JH: Formal analysis, Software, Writing – original draft. JT: Formal analysis, Software, Writing – original draft. XZ: Software, Validation, Writing – original draft. LW: Resources, Software, Writing – original draft. JX: Formal analysis, Resources, Writing – original draft. YJ: Funding acquisition, Project administration, Writing – review & editing. HY: Funding acquisition, Project administration, Writing – review & editing. YY: Conceptualization, Investigation, Writing – review & editing.

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Conflict of interest

The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

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Supplementary material

The Supplementary material for this article can be found online at: https://www.frontiersin.org/articles/10.3389/fsufs.2024.1436542/ full#supplementary-material

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