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RECEIVED 11 July 2024 ACCEPTED 23 September 2024 PUBLISHED 05 November 2024

CITATION

Khan H, Mudgil P, Alkaabi SAS, AlRashdi YHS and Maqsood S (2024) Maillard reaction-based conjugation of pea protein and prebiotic polydextrose): optimization, characterization, and functional properties enhancement.

Front. Sustain. Food Syst. 8:1463058. doi: 10.3389/fsufs.2024.1463058

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Maillard reaction-based conjugation of pea protein and prebiotic (polydextrose): optimization, characterization, and functional properties enhancement

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There is a growing demand for plant-based protein ingredients with improved functionality for use in diverse food and nutraceutical applications. In line with this, the current study aimed to investigate the effect of plant protein-prebiotic (polydextrose) conjugation on the techno-functional properties (emulsification, solubility, fat absorption and foaming) of pea proteins through wet heating Maillard reaction. Pea protein (PeP) was conjugated with polydextrose by incubating the mixture at various process conditions (pea protein: polydextrose mass ratios, temperature, and time). Response surface methodology coupled with Box-Behnken design was used to optimize multiple responses, including conjugation efficiency (CE), emulsifying activity (EAI), and foaming capacity (FC). The pea protein conjugate (optimized value) showed improved solubility throughout a wide pH (2-10) range and higher emulsification activity than pea protein alone. The development of conjugates (PeC) was validated through ultraviolet-visible spectroscopy, FTIR, and o-Phthaldialdehyde (OPA) assay. Browning index, FT-IR spectra, thermal properties, and scanning electron microscopy (SEM) micrographs were analyzed for the conjugate (PeC) obtained at optimized values. The FTIR spectra of the conjugates showed new peaks at 3100-3480 cm⁻¹ and 1,000-1,166 cm⁻¹ indicating conjugation. The Maillard conjugation increased the proportion of β -turn, random coil, accompanied by a decrease in α -helix, and β -sheet. These conformational changes were associated to the improved techno-functional properties of the pea protein upon conjugation, offering potential applications in the formulation of plant-based foods and beverages.

KEYWORDS

pea proteins, polydextrose, conjugation, Maillard reaction, response surface methodology, techno-functional properties

1 Introduction

Due to the diminishing natural resources and the rising world population, it is hard to satisfy the consumer's demand for animal proteins using solely traditional techniques. A systematic approach involves exploring several alternative protein sources to guarantee a steady supply. Moreover, environmental problems regarding methane emission from livestock are also pushing the demand for more sustainable protein alternatives with similar techno-functional characteristics, and a lower carbon footprint (Kumar et al., 2022). With this growing scientific and commercial interest, plant proteins have recently gained the attention as a source of alternative proteins to partially substitute animal proteins (Bryant and Sanctorum, 2021). Numerous studies have emphasized the health advantages of consuming plant protein, which is typically thought to have hypoallergenic proteins (Estell et al., 2021). Plant proteins have various functional properties, such as water, and oil holding capacity, solubility, emulsifying, foaming, and gelling properties, and health benefits like antihypertensive, antioxidant, and modulating intestinal bacteria activities (Grossmann and Weiss, 2021). From a consumer perspective, plant-based products are more ethical, environmentally friendly, and nutritious than animal proteins.

Pea protein has gained much interest from the agro-food market as pea is cheap and environmentally sustainable. Pea (*Pisum sativum* L.) is widely grown worldwide in over 85 countries as a cool-season vegetable crop (Zha et al., 2019; Zha et al., 2021). Peas are used in various food-based applications as protein isolates, produced from pea powder by following specific processing steps such as solubilization and concentration (Kutzli et al., 2020).

Pea protein contains high levels of lysine, threonine, and tryptophan along with other beneficial characteristics like high digestibility, no transgenic origin, and low allergenicity. Irrespective of many advantages pea proteins have some limitations, such as their low solubility, low surface charge, and an abundance of hydrophobic residues on the surface of the molecules, these characteristics contribute to their poor functional properties, specifically in terms of solubility and emulsifying properties (Shanthakumar et al., 2022; Schneider et al., 2023). Consequently, the applications of pea protein isolates in the food industry are challenging and influence their market share (Zha et al., 2021; Ke and Li, 2023). However, different modification procedures have been implemented to improve the plant protein isolates' functionality (Saatchi et al., 2019; Sim et al., 2021; Ke and Li, 2023). Modification procedures include physical, chemical, enzymatic, and combination treatments to enhance their techno-functionality for increasing their application in different food systems (Saatchi et al., 2019). Among the listed approaches, the Maillard reaction is the most promising and non-enzymatic approach for protein conjugation. The Maillard reaction is a natural chemical process that occurs at high temperatures, involving the condensation of free α - and ε-amino groups of proteins with carbonyl groups of carbohydrates, resulting in the formation of glycoconjugates. This reaction takes place without adding any additives and plays an important role in the formation of color and flavor in the food. Recent studies on the functional characteristics of pea protein isolates and Gum Arabic conjugates produced through the Maillard reaction have shown improvement in techno functional properties of pea proteins (Zha et al., 2019; Zha et al., 2021).

Plant proteins have been intensively explored for diverse applications and utilizations in the food system. Recent research has shown that protein-polysaccharide conjugation can play an important role in enhancing the technological functionality of underused plant proteins (Dai et al., 2017; Zha et al., 2019). In this context, using prebiotics-based molecules in conjugation with the plant-derived proteins, can effectively modify the structural conformation of the proteins and enhance their biological and technological functionality together (Van den Abbeele et al., 2013; Zha et al., 2019).

Prebiotics are mainly oligosaccharide-based molecules known to have multiple health-promoting benefits including positive effect on stool quality, low risk of gastroenteritis, and lowincidence of allergy symptoms such as atopic eczema (Paganini et al., 2017; Urango et al., 2024). A recent study on pea protein conjugation with inulin via Maillard reaction has greatly enhanced the emulsification (Jiang et al., 2022). The study showed that pea protein-inulin conjugates were even found to have better antioxidant properties than proteins alone. Similar results were also reported for whey protein hydrolysates conjugated with inulin, where significant improvements in emulsification and antioxidant activities were observed after inulin grafting with whey protein isolates (Wang et al., 2020).

Litesse[®] polydextrose is a carbohydrate known for its low-calorie and low-glycemic properties as well as prebiotic characteristics (Tiihonen et al., 2008). To the best of our knowledge, no such study exist on the impact of polydextrose conjugation with proteins on their techno functional properties. Therefore, this study aimed to explore the effects of conjugating prebiotics (Litesse[®] polydextrose) with plant protein (Pea Protein), via the Maillard reaction, a combination not extensively explored before to develop new protein-prebiotic conjugates with improved functional properties through safe and costeffective methods. Unlike previous studies, it focuses on enhancing both the techno-functional properties and health benefits of pea proteins, offering a novel, eco-friendly solution for developing superior plant-based protein products.

2 Materials and methods

Pea protein isolates used in this study were procured from ET-proteins headquartered in Suzhou, China. The prebiotic carbohydrate (Litesse[®] polydextrose) used in this study was procured from International Flavors & Fragrances Inc. (IFF, Dubai). All other analytical chemicals used such as sodium dodecyl sulfate (SDS), sodium borate, o-Phthaldialdehyde (OPA), methanol, β -mercaptoethanol and others were all procured from Sigma Aldrich (MO, United States). The protein concentration in the pea protein isolate sample (on a dry basis) was 86.0%.

2.1 Optimization of the pea protein isolates and polydextrose conjugates

For the optimization of pea protein (PeP) and polydextrose conjugation, the response surface methodology (RSM) model was developed based on the Box–Behnken design (BBD). The treatment temperature (X1) (60, 80, and 100°C), time (X2) (1, 2, and 3 days), and pea protein polydextrose ratio (X3) 0.5, 1, and 2 *w/w* were selected as the independent factors, meanwhile, conjugation efficiency (CE), emulsifying activity index (EAI) and foaming capacity (FC) of conjugates were the response factors. The experimental data were analyzed using Design-Expert software (version 13.0), and 15 sets of experimental design data were generated for Pea protein: polydextrose conjugate (PeC), as shown in Table 1. Actual and predicted values for foaming capacity and emulsion activity are also framed in Table 2. The proposed second-order polynomial model for the response surface analysis is presented in Equation 1:

$$Y = \beta_0 + \sum_{i=1}^{3} \beta_i X_i + \sum_{i=1}^{3} \beta_{ii} X_i^2 + \sum_{i=1}^{2} \sum_{j=1+1}^{3} \beta_{ij} X_i X_j$$
(1)

TABLE 1 Range and levels of independent variables for pea protein and polydextrose.

Variables	Labels	Ranges values of coded levels				
		-1	0	1		
Temperature (°C)	А	60	80	100		
Time (days)	В	1	2	3		
Protein: polydextrose ratio	С	0.5	1	2		

In this context, Y represents the predicted responses, where $\beta 0$ is the model constant, βi , $\beta i i$, and $\beta i j$ denote the linear, squared, and 2-way interaction effects, respectively. Meanwhile, Xi and Xj stand for the independent factors in the model.

2.2 Preparation of pea protein-polydextrose conjugates

The wet heating Maillard reaction (MR) method as described by Wang et al. (2020) was used to prepare conjugates based on pea protein isolate and polydextrose with some modifications. Initially, pea proteins were dissolved by adjusting the pH to 10.0 using 1 M NaOH and stirred at room temperature for 4 h, following which the pH was adjusted to 7.0 using 1 M HCl. Further, PeP and polydextrose were mixed with a mass ratio (0.5–2.0 w/w) using a homogenizer at 3200 rpm for 10 min. To attain a pH of 8.0 in the reaction mixture, a 1 M NaOH solution was utilized. The final reaction mixtures were subjected to heating in a water bath for varying lengths of time (1–3 days) and at various temperatures (60–100°C). The Maillard reaction was stopped by shifting the solutions to the ice to let the temperature to drop to 25°C. All the samples were then lyophilized for further analysis.

2.3 Conjugation efficiency (CE)

Conjugation efficiency was measured using the method as described by Zhong et al. (2021). For this, o-Phthaldialdehyde (OPA) assay, which estimates total free amino groups in the conjugates after the Maillard reactions, was used to calculate the conjugation efficiency. Further, $20 \,\mu$ L of the samples were seeded onto 96-well microplates and mixed with $200 \,\mu$ L of OPA reagent. The absorbance (Abs) was measured at 340 nm using the SkanItTM Software (Multiskan SkyHigh microplate reader, Thermo Fisher Scientific, United States). CE % was measured according to Equation 2 as follows:

$$CE\% = 1 - \frac{Free \ amino \ groups \ after \ conjugation (M)}{Free \ amino \ groups \ before \ conjugation (M)} \times 100$$
(2)

2.4 Emulsifying activity index

EAI was calculated for the pea protein (PeP) and pea proteinpolydextrose conjugates (PeC) samples using a method validated in our laboratory and reported by Al-Shamsi et al. (2018). A 2 mL of PeC solution (5% w/v) at different pH ranging from 2 to 10 was mixed with a similar amount of sunflower oil in a 10 mL glass tube. The mixtures were homogenized for 1 min at 20,600 rpm using an ultra-turrax homogenizer (Janke & Kunkel, Ultra-Turrax T25) and carefully 50 μ L of the emulsion was removed from the bottom and combined with 5 mL of a 0.1% sodium dodecyl sulfate (SDS) solution. 300 μ L of this solution was then transferred to 96 well microplate and the absorbance of the samples was read at 500 nm in a Multiskan SkyHigh Microplate Spectrophotometer (ThermoFisher Scientific, Richmond, United States). EAI values were then calculated using the formula described by (Pearce and Kinsella, 1978) as follows in Equation 3:

Emulsifying Activity Index (EAI) $\left(\frac{m^2}{g} \right) = \frac{2 \times 2.303 \times Abs @500nm}{0.25 \times protein weight(g)}$ (3)

2.5 Foaming capacity

The analysis of foaming capacity was conducted following the method previously described by Baba et al. (2024) For this, the pH of the solutions was first adjusted to 2, 4, 6, 8, and 10 using a digital pH meter (Ohaus starter ST3100-B, NJ, USA) using 1 M HCl or 1 M NaOH. The solutions thus obtained were homogenized in a homogenizer for 3 min at a speed of 13,400 rpm in a measuring cylinder after which foaming capacity was determined using the Equation as follows in Equation 4:

Foaming capacity (%) = $\frac{\begin{pmatrix} \text{Volume after homogenization} - \\ \text{Volume before homogenization} \end{pmatrix}}{\begin{pmatrix} \text{Volume before homogenization} \end{pmatrix}} *100$ (4)

2.6 Fat absorption capacity

The fat absorption capacity of PeP and PeC was determined as per the protocol reported by Maqsood et al. (2019). Briefly, 1 mL of sunflower oil and 100 mg of powdered samples were mixed and vortexed vigorously for a few minutes. Thereafter, free oil was removed by centrifugation at 13,600 x g for 30 min at 25°C. Afterward, the oil was emptied from the tubes at a 45-degree angle. Samples' capacities for absorbing fat were determined using the formula shown in Equation 5:

Fat Absorption Capacity
$$(FAC)(\%) = \frac{Weight \ of \ absorbed \ oil}{Weight \ of \ protein \ sample} *100 \ (5)$$

2.7 Protein solubility in water

To find out the influence of pea protein and polydextrose conjugation on plant protein solubility method described earlier by Al-Shamsi et al. (2018), was used. For this, 200 mg of each sample, i.e., PeP as control and PeC, were dissolved in deionized water (20 mL) by vortexing and then centrifuged at 7500 g, 25°C, for 15 min. The

Source		CE			EAI				FC						
	Sum of squares	df	Mean square	F-value	p- value	Sum of squares	df	Mean square	F-value	p- value	Sum of squares	df	Mean square	F-value	p- value
Model	207.92	9	23.10	6.41	0.0273	422.09	9	46.90	53.54	0.0002	4197.39	9	466.38	146.70	< 0.0001
A-Temperature	27.58	1	27.58	7.66	0.0395	5.36	1	5.36	6.11	0.0563	2246.98	1	2246.98	706.78	< 0.0001
B-Time	17.88	1	17.88	4.96	0.0764	179.24	1	179.24	204.62	< 0.0001	598.96	1	598.96	188.40	< 0.0001
C-S:S	143.12	1	143.12	39.73	0.0015	2.44	1	2.44	2.79	0.1560	126.22	1	126.22	39.70	0.0015
AB	2.69	1	2.69	0.7464	0.4271	41.67	1	41.67	47.57	0.0010	424.98	1	424.98	133.68	< 0.0001
AC	1.73	1	1.73	0.4792	0.5196	96.90	1	96.90	110.63	0.0001	25.55	1	25.55	8.04	0.0365
ВС	11.18	1	11.18	3.10	0.1384	56.22	1	56.22	64.18	0.0005	1384.58	1	1384.58	435.52	< 0.0001
A ²	0.2516	1	0.2516	0.0699	0.8021	0.7131	1	0.7131	0.8141	0.4083	2014.28	1	2014.28	633.59	< 0.0001
B ²	0.0404	1	0.0404	0.0112	0.9198	22.89	1	22.89	26.13	0.0037	0.0539	1	0.0539	0.0170	0.9015
C ²	3.62	1	3.62	1.00	0.3623	13.81	1	13.81	15.77	0.0106	28.31	1	28.31	8.91	0.0306
Residual	18.01	5	3.60			4.38	5	0.8759			15.90	5	3.18		
Lack of Fit	16.20	3	5.40	5.95	0.1474	2.39	3	0.7970	0.8015	0.5966	8.09	3	2.70	0.6908	0.6370
Pure Error	1.82	2	0.9080			1.99	2	0.9944			7.81	2	3.90		
Cor Total	225.93	14				426.47	14				4213.29	14			
R^2	0.9203					0.9897					0.9962				
Adjusted R ²	0.7768					0.9712					0.9894				
Predicted R ²	-0.1650					0.8998					0.9651				

TABLE 2 Quadratic model and Response foaming for pea protein- polydextrose conjugates as analyzed by ANOVA.

CE, Conjugation efficiency; EAI, Emulsification index; FC, foaming capacity; df: degree of freedom.

protein content in the supernatants was measured using bicinchoninic acid assay as described by Nongonierma et al. (2019), and analyzed as milligrams of protein per gram of sample.

2.8 Browning intensity

The Browning intensity of the samples was investigated, as reported by Chailangka et al. (2022) with some modifications. A UV-1800 spectrophotometer (Shimadzu UV-1601 PC, Shimadzu Corp., Kyoto, Japan) was used to measure the absorbance of the samples at 294 and 420 nm. In order to determine the Browning intensity of the samples, a modified version of the procedure described by Capar and Yalcin, (2021) was used.

2.9 Differential scanning calorimeter

DSC was used to examine the thermal behavior of PeP and PeC (TA Instruments, DSC Q100, United States). All dried samples (2–3 mg) were fed into the apparatus chamber at room temperature in hermetically sealed aluminum pans. To create a controlled atmosphere around the sample, a flow of nitrogen gas at a rate of 40 mL/min was used in the purge line. The temperature gradually increased from 60°C to 200°C at a rate of 10°C/min. The DSC instrument's universal analysis software was used to examine the data and Tg Peak of gelatinization was measured as characteristic temperatures.

2.10 Fourier transform infrared spectroscopy

FTIR spectra were recorded in Attenuated-total-reflectance (ATR) mode using a Spectrum-2 PerkinElmer FT-IR Spectrophotometer (model number 103146, United Kingdom). Samples were placed on the diamond detector and scans were then taken of each sample at a resolution of 4 cm⁻¹ and 32 scans in the 500–4,000 cm⁻¹ range. The spectral data were analyzed using Spectrum-2 Peak Analyzer software. Baseline corrections were carried out using the zeroes method of the second derivative to locate anchor points and identify the baseline. The second derivative method was used to find hidden peaks, and then the Savitsky-Golay function with 10 points and a polynomial order of 2 was used to smooth the data. All these procedures were performed using OriginPro 8.5 software.

2.11 Scanning electron microscopy

Microstructural features and particle morphology of spray-dried pea proteins and their polydextrose conjugates were observed using a Scanning Electron Microscope (JEOL JSM-6010LA) (SEM, Akishima, Tokyo, Japan). The samples were finely sprinkled on aluminum stubs containing double-sided carbon tape, sputtered with approximately 10 nm conductive layer of gold, and then scanned under vacuum by NeoScope JCM-5000 benchtop SEM from JEOL Ltd. in Tokyo, Japan. Micrographs of the samples were taken at different appropriate magnifications to find the best micrograph and 650x was selected and presented in the results.

2.12 Statistical analysis

All the conjugates (pea protein and polydextrose) were prepared in 3 batches and further analysis was done in triplicate. Mean differences were estimated using Tukey's multiple range test via one-way ANOVA by SPSS 28.0 (IBM, United States) Software. Data is presented as the mean and standard deviation (n=3) of three independent replicates.

3 Results and discussion

3.1 Optimization of Maillard reaction-based conjugation preparation

This study has investigated the most important factors affecting the Maillard reaction, i.e., incubation time, temperature, and protein & prebiotic ratio. The range for all three independent parameters, i.e., time, temperature, and solid-to-solid ratio, was selected based on a one-factor-at-a-time (OFAT) experiment performed in our lab (data not shown here). RSM was used to design the 15 experimental sets using three indented factors, as shown in Table 3. All the responses were assessed in relation to the quadratic response and interaction terms of the independent variables, which included plant protein and prebiotic ratio (S:S), temperature, and time. Regression analysis and ANOVA were employed to fit the model and examine the statistical significance of the terms. Table 2 displays the coefficients of determination (R^2) for each dependent variable and the regression coefficients for the quadratic polynomial models. The R^2 values for the conjugation efficiency, foaming, and emulsifying activities were 0.92, 0.99, and 0.98, respectively. The effect of solid-to-solid ratio, temperature, and time on the formation of plant protein and polydextrose conjugates is statistically analyzed in Table 2. The F value of the selected model for CE, EAI, and FC was found to be 6.41, 53.54, and 146.70, respectively. The probability of the F value for the model is found to be less than 0.05. The ANOVA results indicate that the selected model is a suitable fit for the experimental data in practical terms. The probability of the selected quadratic model was found to be significant for each response. The effect of coded independent process variables on CE, EAI, and FC are shown in Figure 1, respectively. For plant protein-based conjugates, the independent variables such as time, temperature, and solid: solid ratio directly affect the dependent responses, including CE, EAI, and FC. According to the desirability framework, utilizing 0.50 protein-to-prebiotic ratios and heating at 100°C for 1.0 day would result in the greatest desirability level of 0.78 on a scale of 0 to 1. The anticipated values for CE, FC, and EAI at these derived conditions were 31.26, 67.25, and 20.92%, respectively. When the experiment was conducted, the observed values were 30.24, 66.61, and 20.49% for CE, FC, and EAI, respectively.

3.1.1 Conjugation efficiency

An OPA assay was conducted using 15 experimental data sets to measure the Conjugation efficiency (CE). The CE was reflected in reduced free amino groups per gram of protein throughout the Maillard reaction (Capar and Yalcin, 2021). The CE values for all the conjugates ranged between 22.59 and 37.24%. Significant factors

(p < 0.05) in the model equation were temperature and proteinprebiotic ratio. In Table 2, the model shows 92.03% accuracy for CE. The maximum CE (37.24%.) was recorded with the 1:1 proteinprebiotic ratio at 80°C temperature for 3 days. Visual analysis was conducted to explore how the interaction between two design variables influences the response variables when one design variable is held constant at its central value. Changes in the CE values with respect to time, temperature, and protein-prebiotic ratio are depicted in Figures 1a-c. Throughout all design variable interactions, the interaction between temperature and time has the most significant effect on CE. There is a noticeable increase in CE with increasing temperature and time. The most notable influence on CE is shown in the relationship between temperature and protein-prebiotic ratios, where a decrease in protein-prebiotic ratios and an increase in temperature have a substantial additive effect on CE. The interaction between time and protein-prebiotic ratios has the most significant impact on CE. Increasing the time while decreasing protein-prebiotic ratios resulted in a substantial improvement in CE.

The Maillard reaction occurs between reducing carbonyl groups in sugars and the amino groups in proteins or peptides through covalent bonding, which leads to the production of the Schiff base (Capar and Yalcin, 2021). Consequently, the reduction of free amino groups can be related to protein engagement during the glycosylation reaction. These authors also reported an improvement in the functional properties of proteins while conjugating between protein -polysaccharides conjugation through Maillard reactions in an aqueous media. Using optimization via Response Surface Methodology (RSM), the most effective conjugation was achieved for faba bean protein and Carrageenan under specific conditions. This included a protein-polysaccharide ratio of 1:2, a reaction temperature of 90°C, and a reaction time of 60 min, which resulted in the best outcome.

3.1.2 Emulsification activity index

A range of 3.61 to 22.09 m²/g was observed for the EAI values of the samples after the Maillard reaction. The model equation indicated that the protein-prebiotic ratio was a significant factor (p < 0.05). The model expresses the EAI with 98.97% accuracy can be seen in Table 2. At 60°C for 3 days, the maximum EAI $(22.09 \text{ m}^2/\text{g})$ was recorded with a protein-prebiotic ratio of 1.25. Figures 1d-f shows the changes in the EAI values depending on the time, temperature, and protein-prebiotic ratio. Among all design variable interactions, the combination of temperature and time has the most significant impact on EAI. Lowering the temperature while increasing the time results in a notable enhancement of EAI. It is apparent that the interaction between temperature and protein-prebiotic ratios has exhibited a minimal effect on EAI. The most important influence on EAI is seen by the connection between temperature and protein-prebiotic ratios, where increasing protein-prebiotic ratios and temperature significantly enhanced EAI.

The expanded secondary structure of the molecules may be primarily responsible for improving the EAI of conjugated proteins. The molecules' altered shape, which has more extended secondary structures, could help them adhere to the oil-water interface more quickly, which would improve the protein's ability to emulsify. Moreover, it is expected that heating proteins might have altered the secondary structures of proteins, causing additional rearrangement of modified proteins at the oil droplet interfaces. It is well-known that glycoproteins possess excellent interfacial stability, primarily attributed to their superior interfacial arrangement during emulsification. In glycoproteins, the hydrophobic protein fractions are embedded within the oil phase, whereas the amphiphilic carbohydrate fractions have an affinity for the water phase, leading to the creation of a dense layer at the oilwater interface (Li et al., 2023).

Run	Temp* (A)	Time (B)	S:S * (C)	CE %	EAI (m²/g)	FC (%)
1	80	1	2	22.59	3.61	26.76
2	80	2	1.25	26.19	12.47	34.78
3	80	2	1.25	27.16	11.19	38.71
4	80	2	1.25	28.10	10.51	36.39
5	60	3	1.25	26.59	22.09	50.33
6	100	2	0.50	34.77	16.70	55.56
7	80	3	0.50	37.24	12.78	14.58
8	60	1	1.25	26.20	5.35	75.90
9	100	3	1.25	30.47	16.83	64.44
10	80	3	2	23.20	19.75	59.78
11	100	2	2	27.23	5.17	59.67
12	100	1	1.25	26.80	13.00	48.78
13	60	2	0.50	28.26	4.77	60.78
14	60	2	2	23.35	12.93	75.00
15	80	1	0.50	29.95	11.63	55.98

TABLE 3 Box-Behnken design matrix with experimental results and predicted values of CE, EAI and FC.

CE, Conjugation efficiency; EAI, Emulsification index; FC, foaming capacity.



3.1.3 Foaming capacity

The challenges of natural proteins in food industry applications, such as cosmetics and foam food, can be resolved by improving the foaming capabilities of proteins. The FC value ranged between 14.58 and 75.9% in all the experimental data sets. Notably, the model equation identified time, temperature, and protein-prebiotic ratio as significant factors (p < 0.05) in predicting the FC value. The maximum FC (75.9%) was recorded with the 1.25 protein-prebiotic ratio at 60°C for 1 day. Figures 1g-i illustrates the variations in the FC value resulting from differences in time, temperature, and protein-prebiotic ratio. Among all design variable interactions, the combination of temperature and time has the most significant impact on FC. Reducing both temperature and time leads to a significant enhancement in FC. The interaction between temperature and protein-prebiotic ratios exhibits the most significant effect on FC, with increasing protein-prebiotic ratios and decreasing temperature exhibiting a significant enhancement effect on FC. The interaction between time and protein-prebiotic ratios exhibited a minimal effect on FC. The most effective explanatory models for CE, EAI, and FC were identified through regression analysis and are presented in Equations 6–8, respectively.

$$CE = +27.74 - 0.038 * A + 0.58 * B - 2.07 * C + 0.040$$

* AB - 0.043 * AC - 2.229 * BC + 0.006 A²
+ 0.104B² + 1.759 C² (6)

$$EAI = -34.94 + 0.59 A + 1.43B + 24.11C - 0.161A$$
$$*B - 0.328 A * C + 4.998 B * C + 0.001 A^{2}$$
$$+ 2.48 B^{2} - 3.43C^{2}$$
(7)

$$FC = +599.37 - 10.37 \text{ A} - 74.04 \text{ B} - 42.72 \text{ C} + 0.51 \text{A}$$
$$*\text{B} - 0.16 \text{ A} *\text{C} + 24.80 \text{ B} *\text{C} + 0.05 \text{ A}^{2}$$
$$+1.20 \text{ B}^{2} + 4.92 \text{ C}^{2}$$
(8)

Where, CE, Conjugation efficiency; EAI, Emulsification index; FC, Foaming capacity; A, Temperature; B, Time; and C, solid to solid ratio

3.2 Effect of pH on emulsification activity index (m²/g)

Additionally, the emulsifying activity of pea proteins before and after conjugation with polydextrose was investigated against the various pH (2-10) using the optimized value as shown in Figure 2A. Overall, at all pH levels, it is evident that the emulsifying activities of the PeC samples were much better than those of pea protein (PeP) alone. The lowest EAI (m^2/g) for all the samples was detected at pH 4.0, the isoelectric range of protein, and is attributed to the reduction of solubility of proteins at this pH. Overall, these improved emulsification activities at the neutral pH and alkaline ranges are like those obtained by other researchers. For instance, Karbasi and Askari (2021), also signified improved EAI (m²/g) activity of whey proteins when conjugated with various mono-oligo and polysaccharides by the Maillard reaction. The higher level of emulsification activity among conjugates might be ascribed to the possible covalent coupling between polyols groups of prebiotic carbohydrates that played a significant role in changing the hydrophobic nature of plant proteins to more amphiphilic making them more surface active and absorbable at the O/W interface (Maqsood et al., 2019).

3.3 Effect of pH on foaming capacity

Additionally, FC of pea proteins and conjugation with polydextrose was investigated against the various pH (2-10) using the optimized value as displayed in Figure 2B. Many factors, including protein unfolding, pH, protein concentration, temperature, mixing duration, and foaming technique, have an impact on the foaming properties of proteins. The changes in the foaming capacity of protein upon conjugation with carbohydrates are ascribed to the possible development of new functional groups on conjugation. The decline in foaming capacity at pH 4.0 is usually due to their isoelectric point of the protein. Proteins exhibit less solubility at an isoelectric point and foam is usually formed by soluble proteins; hence at the isoelectric point where soluble proteins are in less number leads to lower foaming capacity (Magsood et al., 2019). It is also noteworthy to notice that there was a possible shift in the isoelectric point of protein conjugates. As for native proteins, no decline in foaming capacity was observed as the foaming properties of proteins are influenced by numerous parameters, including concentration of proteins, method of isolation, pH, temperature, speed, and time of mixing. Analogous findings were also reported by Wang et al. (2020), who found that conjugating whey proteins with inulin enhanced forming capacity.

3.4 Fat absorption capacity: analysis and measurement

The potential of proteins to bind fats is essential for food applications, primarily because it improves mouthfeel and flavor



retention (Wang and Arntfield, 2017). The fat absorption capacity percentage of pea protein and pea protein-polydextrose conjugate (optimized value) are shown in Figure 3A. The results did not display any significant difference between all samples before and after conjugation (p > 0.05). FAC of approximately 19% was observed for the PeP sample which did not show any significant improvement after conjugation and reached a value of only 20%. It has been stated that the FAC is higher in protein powders with lower density, as smaller particles allow for a greater entrapment of oil compared to protein powders with higher density (Maqsood et al., 2019). The number of non-polar sites, protein-lipid-carbohydrate interactions, and protein concentration can influence FAC. Proteins that are insoluble and hydrophobic possess a strong ability to bind with oils (Nakai, 1983; Chan et al., 2024).

3.5 Protein solubility in water

Protein solubility is one of the most crucial characteristics of proteins, which impacts their the texture, color, and sensory qualities of products (Boonlao et al., 2023). The composition of proteins has a direct impact on their utilization in food formulations and is closely linked to their ability to exhibit desirable physical, chemical, and functional properties, such as water retention, emulsification, and gelling (Kutzli et al., 2021). The outcomes showed that protein conjugation dramatically altered the samples' solubility pattern. As seen in Figure 3B, the conjugated pea proteins' (PeC) (optimized value) solubility (460.17 mg/g) was more than that of pea protein alone (PeP), which was only 271.73 mg/g. These findings showed that while conjugation might



significantly increase the protein's solubility, pea proteins were weakly soluble in water. Comparable results were also observed by Boonlao et al. (2023), where the conjugation of soy proteins with maltodextrin led to enhanced protein solubility and foaming properties through conjugates. The increased solubility of protein upon conjugation could be attributed to exposed hydrophilic hydroxyl groups that may be aligned on the surface of conjugated proteins, thus enhancing the interaction of the protein with water (Zha et al., 2021).

3.6 Browning intensity

The browning intensity was also used to analyze the extent of the Maillard reaction. Additionally, UV–visible absorption measurements were performed at 294 and 420 nm on pea protein and optimized value of conjugate to check the possible formation of Amadori, Heyns, and melanoidin compounds during both the initial and final stages of the Maillard reaction process. Amadori and Heyns compounds might be formed at the early stage of MR and are not supposed to show a brown color at 294 nm. However, the presence of melanoidin compounds was confirmed by measuring the absorption at 420 nm during the final stage of Maillard reaction (Chailangka et al., 2022). The brown color development could be visually observed at different stages of the Maillard reaction process. After being heated for 3 days, the browning index of PeC conjugate was recorded as significantly higher than PeP (p < 0.05). The browning index for PeP was recorded at 0.147, whereas the index for the conjugate was found to be 0.546. As per the previous studies, it is crucial to maintain a low browning index because a high index indicates the degradation of Amadori products, which in turn can be headed to the dissociation of the protein-polysaccharides conjugates (Chailangka et al., 2022).

3.7 Differential scanning calorimetry

The thermal transition of the PeP and PeC sample (optimized value) was investigated by DSC and analyzed thermographs are depicted in Figure 3C. The denaturation temperature (Td) of pea protein isolates was reported to be 82.1°C (Liang and Tang, 2013). As depicted in Figure 3C, the denaturation peak for pea protein conjugate was observed to be higher at 100.89°C, in contrast to the pea protein isolates which exhibited a peak at approximately 80.18°C. The denaturation temperature is directly associated with the thermal stability of proteins and reflects the breakdown of hydrogen bonds that uphold the tertiary and quaternary structures of proteins, principally tertiary ones (Ladjal-Ettoumi et al., 2016). Pea protein conjugate showed significant improvement in thermal stability through the Maillard reaction, which might be more likely due to glycosylation restricting the unfolding of the proteins by means of steric effects (Boye et al., 2010; Zhang et al., 2024).

3.8 Fourier transform infrared spectroscopy

FTIR analysis captured the structural changes during the conjugation between the pea protein and polydextrose, as shown in Figure 4A. The protein composition is characterized by the presence of peptide bonds, which contribute to the amide I, II, and III spectral regions, which are assigned to C=O stretching, N–H deformations,

and C-N stretching & N-H bending vibrations, respectively (Xu et al., 2019). The FTIR graph of the PeC (optimized value) displayed a broad band in the amide A region (N-H stretching and O-H stretching) at around 3,276 cm⁻¹. The Maillard reaction modified the absorption band within the 3,200–3,700 cm⁻¹ range induced by the stretching vibration of OH, which was more substantial in PeC compared to PeP. A clear peak at around 2,923 cm⁻¹ (C-H stretching) in PeC has been observed. The signature bonds of amide I, II, and III in the PeP spectrum appeared at 1658 cm⁻¹, 1,562 cm⁻¹, and 1,398 & 1,248 cm⁻¹, respectively. The partially shifted bands with increased intensities were observed for amide I, II, and III in PeC's FTIR spectra. The intensity of absorption peaks around 1,050-1,150 cm⁻¹ represents C=O stretching and OH deformation vibration, which has increased in PeC, as can be seen in Figure 4A (Xu et al., 2019). The use of several functional groups, including NH₂, especially from lysine, that may be lost during the reaction could cause quite a few modifications in the FTIR spectrum due to the chemical changes that occur during the Maillard reaction in PeC during conjugation. Other than that, the aforementioned events were attributed to the Maillard reaction, shown in several earlier publications (Xu et al., 2019). Similar results were earlier demonstrated by Zhang et al. (2022), the Maillard reaction, which induces conjugation between pea proteins and maltodextrins also causes similar spectral changes in the FTIR spectrum.

Within the protein backbone, the amide I and amide II bands are the most prominent, with the amide I band (1700–1,600 cm⁻¹) being particularly sensitive to the secondary structural components of the



(A) FTIR spectrum (B) Second derivative (C,D) Curve fitted amide I region (1600–1700 cm⁻¹) of native pea protein and its conjugates as obtained by wet heating via Maillard reaction.

protein. The C=O stretch vibrations of the peptide bonds, which make up around 80% of the amide I band's structure, are the main cause of this sensitivity. In Table 4, the relative secondary structure content of PeP is shown, indicating approximately 31.96% α -helix, 21.14% β -sheet, 32.10% β -turns, and 11.91% random coil. The pea conjugate structure differed from PeP in that it contained more unordered secondary structures, such as β -turn and random coil, and less ordered secondary structures, such as α -helix and β -sheet. This indicates that the glycosylation process can potentially affect the intermolecular forces within protein molecules, leading to changes in their overall structure (Wen et al., 2020).

3.9 Scanning Electron microscopy

SEM was used to examine the morphological features and changes in the protein, PeP, and their conjugation with polydextrose, PeC (optimized value). The changes in particle morphology during conjugation are visible in the micrographs, as seen in Figure 5. Cleary pea protein particles were seen as circular in shape but were somehow

TABLE 4 Secondary structure estimations from the deconvolution of the amide I spectra.

Designation	Wavenumber (cm ^{_1})	PeP	PeC
β Sheet	1,610-1,640	21.14 ± 0.13	19.28 ± 0.32
Random coils	1,640-1,650	11.91 ± 0.65	20.94 ± 0.40
α-helix	1,650–1,660	31.96 ± 1.58	21.11 ± 0.38
β turns	1,660–1,695	32.10 ± 1.73	37.35 ± 1.2

Values are presented as mean \pm standard deviation (n = 3).

irregular in their morphology upon conjugation with polydextrose. The clear interaction of protein and prebiotic molecules was demonstrated through a dense network of protein and prebiotics. Compared to the pea protein sample (Figures 5A,B), PeC conjugate (Figures 5C,D) displayed distinct morphological characteristics and transitioned from smooth spherical particles to larger particle structures with wrinkled morphology. Zha et al. (2019), has reported similar morphological observations after the conjugation between pea protein and gum Arabic.

4 Conclusion

This study demonstrates that pea protein and polydextrose conjugates can be efficiently produced through the wet heating Maillard reaction, which is a cost-effective and straightforward approach for creating food-grade functional ingredients from underutilized plant proteins. The wet-heating conjugation process notably enhanced the functional properties of the conjugates, including solubility, emulsifying, and foaming activities, with the solid-to-solid ratio, temperature, and time identified as key factors influencing these improvements. The optimized conditions, specifically a 0.50 protein-polydextrose ratio and heating at 100°C for 1 day, yielded the maximum conjugation efficiency, foaming capacity, and emulsifying activity index. The findings of this research study contribute significantly to the area of food science by showcasing how underutilized plant proteins can be transformed into functional ingredients through controlled Maillard reactions. This advancement not only addresses the need for sustainable and plantbased food solutions but also opens avenues for further exploration in optimizing the conjugation process for various protein-prebiotic combinations. Future research could explore the application of these conjugates in different food systems, as well as their potential health benefits as functional food ingredients.



FIGURE 5

SEM micrographs of native pea protein (A,B) and their conjugate (C,D) as obtained by wet heating via Maillard reaction.

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Data availability statement

The raw data supporting the conclusions of this article will be made available by the authors, without undue reservation.

Author contributions

HK: Formal analysis, Investigation, Methodology, Validation, Writing – original draft, Writing – review & editing. PM: Conceptualization, Formal analysis, Investigation, Methodology, Supervision, Visualization, Writing – original draft, Writing – review & editing. SA: Investigation, Writing – original draft. YR: Data curation, Investigation, Methodology, Writing – original draft. SM: Conceptualization, Funding acquisition, Supervision, Visualization, Writing – original draft, Writing – review & editing.

Funding

The author(s) declare that financial support was received for the research, authorship, and/or publication of this article. The authors express their appreciation to the United Arab Emirates University for

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Conflict of interest

The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

The author(s) declared that they were an editorial board member of Frontiers, at the time of submission. This had no impact on the peer review process and the final decision.

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